ARTICLE IN PRESS

Tetrahedron Letters xxx (2012) xxx-xxx

Contents lists available at SciVerse ScienceDirect



Tetrahedron Letters

journal homepage: www.elsevier.com/locate/tetlet

Nano copper oxide mediated ligand-free C–S cross-coupling and concomitant oxidative aromatization of 4-aryl-3,4-dihydropyrimidin-2(1*H*)-thione with diaryliodonium salts

Bhagyashree Y. Bhong, Amol V. Shelke, Nandkishor N. Karade*

Department of Chemistry, Rashtrasant Tukadoji Maharaj Nagpur University, Nagpur, Maharashtra 440 033, India

ARTICLE INFO

Article history: Received 13 September 2012 Revised 27 October 2012 Accepted 30 October 2012 Available online xxxx

Keywords: Nanoparticles Biginelli reaction C-S cross-coupling Diaryliodonium salts Oxidative aromatization

ABSTRACT

A wide range of Biginelli 4-aryl-3,4-dihydropyrimidin-2(1*H*)-thiones undergo ligand-free C–S crosscoupling with diaryliodonium triflates in the presence of CuO nanoparticles with the concomitant oxidative aromatization to form highly substituted 2-(thioaryl)pyrimidine. The nano CuO catalyst can be recycled and reused three times without any significant loss of catalytic activity.

© 2012 Elsevier Ltd. All rights reserved.

etrahedro

Aryl sulfides are structural motifs found in many biological and pharmaceutically active compounds and have been used as important intermediates in several organic transformations.¹ The transition metal (Pd, Cu, Ni, Co, and Fe) catalyzed cross-coupling reactions of aryl halides with appropriate thiols are generally a method of choice for the construction of a C(aryl)-S bond.² Recently, the advent of nanostructured transition metals has been considered as a sustainable and competitive alternative to conventional catalysis.³ This is due to the high surface areas and rich exposed active sites of the nano-catalysts which enhance catalytic efficiency, thereby requiring lower loading of catalyst. Additionally, the insolubility of nano-catalysts in reaction solvents imparts the advantages of heterogeneous catalysis such as simplified isolation of the product along with easy recovery and recyclability of the catalysts. As a result, intensive work has recently been reported for the C(aryl)–S cross-coupling reactions of iodoarenes with thiols to form diaryl sulfides using various nanoparticle size catalysts such as CuI,⁴ CuFe₂O₄,⁵ and CuO⁶ and In₂O₃.⁷

Biginelli 3,4-dihydropyrimidine-2(1*H*)-one (DHPM) is a privileged heterocyclic scaffold exhibiting a wide range of pharmacological properties such as calcium channel blockers, hepatitis B virus replication inhibitors, mitotic kinesin inhibitors, and α_{1a} -adrenergic receptor antagonists.⁸ Owing to this remarkable pharmacological profile, the recent decade has witnessed a

* Corresponding author. Fax: +91 712 2532841. E-mail address: nnkarade@gmail.com (N.N. Karade).

0040-4039/\$ - see front matter @ 2012 Elsevier Ltd. All rights reserved. http://dx.doi.org/10.1016/j.tetlet.2012.10.131 considerable growth in the decoration of all the six positions of DHPMs (N1, C2, N3, C4, C5, and C6) to produce low molecular weight 'drug like' compounds.⁹ As the original Biginelli reaction is carried out by a three-component condensation of ethyl acetoacetate, aldehyde, and urea or thiourea, the C2 position of DHPMs is always either with carbonyl or thiocarbonyl functional groups. The C2 position of DHPMs with thiocarbonyl functional group has been synthetically manipulated (Fig. 1) by (i) reaction with alkyl halide to produce 1,4-dihydropyrimidine¹⁰ followed by oxidative aromatization using (diacetoxyiodo)benzene, CAN, Mn(OAc)₃, and MnO₂ to form pyrimidine,¹¹ (ii) reaction of alkyne or doubly electrophilic building block like α -halo ketone to furnish fused pyrimidines,¹² (iii) oxidative aromatization to form disulfide,¹³ and (iv) desulfitative Liebeskind-Srogl coupling of arylboronic acid or aryltributyltin to form 2-arylation product.¹⁴ The C-S cross-coupling of 3,4-dihydropyrimidin-2(1*H*)-thione with aryl halide is a relatively less investigated reaction in the literature.

Among the various cross-coupling reactions, there is a great synthetic challenge for C–S bond formation due to the susceptibility of sulfur for oxidative dimerization and its tendency to bind with metal which causes the modification or deactivation of the catalyst.¹⁵ In contrast to aryl sulfides, the methods for the synthesis of heterocyclic sulfides including pyrimidine moiety are very limited. The 3,4-dihydropyrimidine-2(1*H*)-thione can be considered as the structural analog of cyclic thiourea and therefore, we were intrigued with the possibility of C–S cross-coupling reaction with a suitable arylating agent. Moreover, only a few reports have



Figure 1. Decoration of 3,4-dihydropyrimidine-2(1*H*)-thione via functionalization at the C-2 position.

appeared describing the coupling of thiocarbonyl functional groups with aryl halides.¹⁶ To date, there is only one example in the literature for C–S cross-coupling of 3,4-dihydropyrimidine-2(1*H*)-thione with phenylboronic acid using stoichiometric Cu(OAc)₂ under microwave conditions (45 min) to form 2-thiophenyl-1,4dihydropyrimidine.^{14a} Recently, a diaryliodonium salt has gained a renewed interest in various C–C, C–N, C–O, and C–S coupling reactions as a more reactive version of iodobenzene.¹⁷ It has been explored in the C–S cross-coupling reactions of aromatic and aliphatic thiols using $Pd(PPh_3)_4$.¹⁸ Here, we report our studies on the C–S cross-coupling reaction of 3,4-dihydropyrimidine-2(1*H*)-thione with a diaryliodonium salt using CuO nano-catalyst under ligand-free conditions. It was observed that 3,4-dihydropyrimidine-2(1*H*)-thione undergoes C–S cross-coupling with a symmetrical diaryliodonium salt with concomitant oxidative aromatization to form highly functionalized 2-thioarylpyrimidine derivatives (Scheme 1).

The optimum conditions for the S-arylation of 4-phenyl-3,4dihydropyrimidine-2(1H)-thione (1a) were investigated by screening of appropriate arylating agents and copper catalysts in DMF using K₂CO₃ as the base. The results are summarized in Table 1. The initial attempts of utilizing iodobenzene for arylation of 1a using various copper catalysts such as Cu(OAc)₂, CuCl₂, CuI, and nano CuO (entries 1–6) resulted in poor vields (0–27%) of 2-phenylthio-1,4-dihydropyrimidine 3a. Therefore, we used diphenyliodonium salts as arylating agents for the tentative C-S cross-coupling using different copper catalysts (entries 7-15). In the absence of copper catalyst, C(aryl)-S coupling of **1a** with diphenyliodonium salts did not take place (entries 7 and 12). When the same coupling reaction was examined with diphenyliodonium chloride (entry 8) using CuI (10 mol %), the formation of **3a** and aromatized 2-(phenylthio)pyrimidine 4a, respectively, took place in 17% and 46% yields. Other catalysts such as Cu(OAc)₂ and CuCl₂ were found to be inferior for the C-S cross-coupling of diphenyliodonium chloride with 1a (entries 9 and 10). Screening of nano CuO as a catalyst revealed that the yield of aromatized pyrimidine 4a was much higher than CuI with negligible formation of 3a (entry 11). Further



Scheme 1. C-S cross-coupling and concomitant oxidative aromatization of 4-aryl-3,4-dihydropyrimidine-2(1H)-thione.

Table 1

Optimization of reaction conditions for C-S cross-coupling of 4-phenyl-3,4-dihydropyrimidine-2(1H)-thione with arylating agents^a



Entry	Arylating agent	Copper catalyst (mol %)	Yield 3a (%)	Yield 4a (%)	Time (h)
1	PhI	Cu(OAc) ₂ (10 mol %)	00	00	24
2	PhI	CuCl ₂ (10 mol %)	00	00	24
3	PhI	CuI (10 mol %)	14	00	24
4	PhI	CuI (20 mol %)	23	00	24
5	PhI	CuO nano (10 mol %)	18	00	24
6	PhI	CuO nano (20 mol %)	27	00	24
7	Ph ₂ ICl	_	00	00	24
8	Ph ₂ ICl	Cul (10 mol %)	17	46	12
9	Ph ₂ ICl	Cu(OAc) ₂ (10 mol %)	21	00	24
10	Ph ₂ ICl	CuCl ₂ (10 mol %)	00	00	24
11	Ph ₂ ICl	CuO nano (10 mol %)	14	58	24
12	Ph ₂ IOTf	_			08
13	Ph ₂ IOTf	CuO nano (10 mol %)	12	67	12
14 ^b	Ph ₂ IOTf	CuO nano (10 mol %)	00	84	08
15	Ph ₂ IBF ₄	CuO nano (10 mol %)	16	63	12

^a Reaction conditions unless otherwise stated: Arylating agent (1 equiv), K₂CO₃ (1.5 equiv), catalyst, DMF (10 mL).

 $^{\rm b}~$ Ar_2IOTf (1.5 equiv), K_2CO_3 (2.5 equiv), catalyst, DMF (10 mL).

ARTICLE IN PRESS

B. Y. Bhong et al./Tetrahedron Letters xxx (2012) xxx-xxx

Table 2

Cross-coupling of 3,4-dihydropyrimidine-2(1H)-thione with diphenyl iodonium triflates^a



^a Reaction conditions: 3,4-Dihydropyrimidine-2(1*H*)-thione (1 mmol), diaryliodonium triflate (1.5 mmol), CuO nanoparticle (0.025 mmol), and K₂CO₃ (2.5 mmol) in 10 mL DMF under reflux for 12 h.

studies were also carried out about the influence of different counter anions of diphenyliodonium salts (entries 12–15), which showed that satisfactory results could be obtained with diphenyliodonium triflates, resulting in the formation of aromatized pyrimidine **4a** in good yields. However, 1.5 equiv of diphenyliodonium triflates with respect to **1a** were required for the exclusive formation of aromatized **4a** in 84% excellent yield (entry 14). Therefore, the optimized conditions for C–S cross-coupling of **1a** employed nano CuO (10 mol %), diphenyliodonium triflate (1.5 equiv), and K₂CO₃ (2.5 equiv) in DMF for the exclusive formation of aromatized **4a**.¹⁹

With the successful optimization results in hand, we moved on to investigate the scope of the process in terms of the reaction of various 4-aryl-3,4-dihydropyrimidine-2(1*H*)-thiones with diphenyliodonium triflates (Table 2). A wide variety of 4-aryl-3,4-dihydropyrimidine-2(1*H*)-thiones derived from ethyl acetoacetate (entries **4a**–**h** and **4m**–**p**) and acetyl acetone (entries **4i**–**l**) with differently substituted aromatic ring at the C4 position were employed for C–S cross-coupling reaction with diphenyliodonium triflates. As can be seen, in all the cases excellent yields of highly functionalized 2-(thioaryl)pyrimidine were obtained and all the products were characterized by IR, NMR and mass spectrometry.²⁰ To the best of our knowledge, this is the first kind of report on the one-pot synthesis of biologically and synthetically important 2-(thioaryl)pyrimidine from readily accessible 4-aryl-3,4-dihydropy-rimidine-2(1*H*)-thione.

Various symmetrical diaryliodonium salts were also examined in C–S cross-coupling with 4-aryl-3,4-dihydropyrimidine-2(1*H*)thione (entries **4k–p**). The protocol is tolerant to the presence of electron-donating and withdrawing functional groups of symmetrical diaryliodonium salts. The reaction of unsymmetrical iodonium salts with **1** resulted in a mixture of 2-(thioaryl)pyrimidines due to the transfer of both the aryl groups on sulfur (Supplementary data).

As nano CuO is a heterogeneous catalyst, the recyclability of the catalyst in this coupling reaction was examined. To facilitate the catalyst recovery, the model C–S cross-coupling reaction of diphe-nyliodonium triflate with 4-phenyl-3,4-dihydropyrimidine-2(1*H*)-thione was carried out on 5 mmol scale. After each cycle, the catalyst was recovered by simple centrifugation, washing with deionized water and ethanol, and then drying in vacuo. The separated catalyst was again reused for C–S cross-coupling and this was repeated three times. In these cases, the yield of **4a** was found to be 87%, 79%, and 73% in successive three times use.

4

B. Y. Bhong et al. / Tetrahedron Letters xxx (2012) xxx-xxx

In summary, we have developed a facile one-pot CuO nanoparticle catalyzed ligand-free C–S cross-coupling of diaryliodonium salts with 4-aryl-3,4-dihydropyrimidine-2(1*H*)-thione to furnish highly functionalized 2-(phenylthio)pyrimidine which are biologically and synthetically important scaffolds. It is a unique example of CuO nanoparticle catalysis for a C–S cross-coupling reaction of thiocarbonyl functional groups of DHPMs with concomitant oxidative aromatization to produce pyrimidine. This method also offers significant advantages such as operational simplicity with a recyclable catalytic system. This is also a first report of a C–S crosscoupling reaction of diaryliodonium salts using a copper based catalytic system.

Acknowledgments

The authors are thankful to the Department of Science and Technology, New Delhi, India (No. SR/S1/OC-72/2009) for the financial support. The authors are also grateful to SAIF, Punjab University, Chandigarh, India, for recording the NMR spectra.

Supplementary data

Supplementary data associated with this article can be found, in the online version, at http://dx.doi.org/10.1016/j.tetlet.2012. 10.131.

References and notes

- (a) Wang, Y.; Chackalamannil, S.; Hu, Z.; Clader, J. W.; Greenlee, W.; Billard, W.; Binch, H.; Crosby, G.; Ruperto, V.; Duffy, R. A.; McQuade, R.; Lachowicz, J. E. *Bioorg. Med. Chem. Lett.* **2000**, *10*, 2247; (b) Nielsen, S. F.; Nielsen, E. O.; Olsen, G. M.; Liljefors, T.; Peters, D. J. Med. Chem. **2000**, *43*, 2217; (c) De Martino, G.; Edler, M. C.; LaRegina, G.; Coluccia, A.; Barbera, M. C.; Barrow, D.; Nicholson, R. I.; Chiosis, G.; Brancale, A.; Hamel, E.; Artico, M.; Silvestri, R. J. Med. Chem. **2006**, *49*, 947; (d) Hartwig, J. F. Angew. Chem., Int. Ed. **1998**, *37*, 2046; (e) Wolfe, J. P.; Wagaw, S.; Marcoux, J. F.; Buchwald, S. L. Acc. Chem. Res. **1998**, *31*, 805.
 (a) Beletskaya, I. P.; Ananikov, V. P. Chem. Rev. **2011**, *111*, 1596; (b) Fernández-
- (a) Beletskaya, I. P.; Ananikov, V. P. Chem. Rev. 2011, 111, 1596; (b) Fernández-Rodríguez, M. A.; Shen, Q.; Hartwig, J. F. Chem. Eur. J. 2006, 12, 7782; (c) Fernández-Rodríguez, M. A.; Shen, Q.; Hartwig, J. F. J. Am. Chem. Soc. 2006, 128, 2180; (d) Zhang, Y.; Ngeow, K. N.; Ying, J. Y. Org. Lett. 2007, 9, 3495; (e) Jammi, S.; Barua, P.; Rout, L.; Saha, P.; Punniyamurthy, T. Tetrahedron Lett. 2008, 49, 1484; (f) Wong, Y. C.; Jayanth, T. T.; Cheng, C. H. Org. Lett. 2006, 8, 5613; (g) Correa, A.; Carril, M.; Bolm, C. Angew. Chem., Int. Ed. 2008, 47, 2880; (h) Beletskaya, I. P.; Cheprakov, A. V. Coord. Chem. Rev. 2004, 248, 2337; (i) Liu, Y.; Wan, J.-P. Org. Biomol. Chem. 2011, 9, 6873.
- (a) Pacchioni, G. Surf. Rev. Lett. 2000, 7, 277; (b) Knight, W. D.; Clemenger, K.; de Heer, W. A.; Saunders, W. A. M.; Chou, Y.; Cohen, M. L. Phys. Rev. Lett. 1984, 52, 2141; (c) Kaldor, A.; Cox, D.; Zakin, M. R. Adv. Chem. Phys. 1988, 70, 211; (d) Pachon, L. D.; van Maarseveen, J. H.; Rothenberg, G. Adv. Synth. Catal. 2005, 347, 811.
- Xu, H.-J.; Liang, Y.-F.; Zhou, X.-F.; Feng, Y.-S. Org. Biomol. Chem. 2012, 10, 2562.
 Swapna, K.; Murthy, S. N.; Jyothi, M. T.; Nageswar, Y. V. D. Org. Biomol. Chem. 2011, 9, 5989.
- 6. (a) Rout, L.; Sen, T. K.; Punniyamurthy, T. Angew. Chem., Int. Ed. 2007, 46, 5583; (b) Zhang, J.; Zhang, Z.; Wang, Y.; Zheng, X.; Wang, Z. Eur. J. Org. Chem. 2008, 5112; (c) Jammi, S.; Sakthivel, S.; Rout, L.; Mukherjee, T.; Mandal, S.; Mitra, R.; Saha, P.; Punniamurthy, T. J. Org. Chem. 2009, 74, 1971; (d) Reddy, V. P.; Swapna, K.; Kumar, A. V.; Rao, K. R. Synlett 2009, 17, 2783; (e) Schwab, R. S.; Singh, D.; Alberto, E. E.; Piquini, P.; Rodrigues, O. E. D.; Braga, A. L. Catal. Sci. Technol. 2011, 1, 569; (f) Harsha Vardhan, Reddy K.; Prakash, Reddy V.; Ashwan, Kumar A.; Kranthi, G.; Nageswar, Y. V. D. Beilstein J. Org. Chem. 2011, 7, 886; (g) Harsha Vardhan, Reddy K.; Satish, G.; Ramesh, K.; Karnakar, K.; Nageswar, Y. V. D. Tetrahedron Lett. 2012, 53, 3061.
- 7. Reddy, V. P.; Kumar, A. V.; Swapna, K.; Rao, K. R. Org. Lett. 2009, 11, 1697.
- (a) Kappe, C. O. *Eur. J. Med. Chem.* **2000**, *35*, 1043; (b) Kaan, H. Y. K.; Ulaganathan, V.; Rath, O.; Prokopcova, H.; Dallinger, D.; Kappe, C. O.; Kozielski, F. J. Med. Chem. **2010**, *53*, 5676.
- 9. Dallinger, D.; Kappe, C. O. Pure Appl. Chem. 2005, 77, 155.
- (a) Atwal, K.; Rovnyak, G. C.; Schwartz, J.; Moreland, S.; Hedberg, A.; Gougoutas, J. Z.; Malley, M. F.; Floyd, D. M. J. Med. Chem **1990**, 33, 1510; (b) Pathak, P.; Kaur, R.; Kaur, B. Arkivoc **2006**, xvi, 160; (c) Singh, S.; Schober, A.; Gebinoga, M.; Grob, G. A. Tetrahedron Lett. **2009**, 50, 1838; (d) Kappe, C. O.; Roschger, P. J. Heterocycl. Chem. **1989**, 26, 55.
- Mn(OAc)₃: (a) Akhtar, M. S.; Seth, M.; Bhaduri, A. P. Indian J. Chem. **1987**, 26B, 556; Phl(OAc)₂: (b) Karade, N. N.; Gampawar, S. V.; Tale, N. P.; Kedar, S. B. J. Heterocycl. Chem. **2010**, 47, 740; (c) Matloobi, M.; Kappe, C. O. J. Comb. Chem. **2007**, 9, 275.

- (a) Singh, S.; Schober, A.; Gebinoga, M.; Groß, G. A. *Tetrahedron Lett.* **2011**, *52*, 3814; (b) Xiao, D.; Han, L.; Sun, Q.; Chen, Q.; Gong, N.; Lv, Y.; Suzenet, F.; Guillaumet, G.; Cheng, T.; Li, R. *RSC Adv.* **2012**, *2*, 5054.
- Hayashi, M.; Okunaga, K.; Nishida, S.; Kawamura, K.; Eda, K. *Tetrahedron Lett.* 2010, 51, 6734.
- (a) Lengar, A.; Kappe, C. O. Org. Lett. 2004, 5, 771; (b) Sun, Q.; Suzenet, F.; Guillaumet, G. Tetrahedron Lett. 2012, 53, 2694.
- 15. Kondo, T.; Mitsudo, T. A. Chem. Rev. 2000, 100, 3205.
- 16. Niu, L.-F.; Cai, Y.; Liang, C.; Hui, X.-P.; Xu, P.-F. Tetrahedron 2011, 67, 2878.
- (a) Merritt, E. A.; Olofsson, B. Angew. Chem., Int. Ed. 2009, 48, 9052; (b) Yusubov, M. S.; Maskaev, A. V.; Zhdankin, V. V. Arkivoc 2011, 1, 370.
- 18. Wang, L.; Chen, Z. C. Synth. Commun. 2001, 1227.
- 19. General procedure for C-S cross-coupling of 4-aryl-3,4-dihydropyrimidin-2(1H)thione with symmetrical diaryliodonium triflates: CuO nano particles (mean particle size, 33 nm, surface area, 29 m²/g, and purity, 99.99%) were purchased from Sigma Aldrich. A mixture of 4-aryl-3,4-dihydropyrimidin-2(1H)-thione (1.5 mmol), diaryliodonium triflates (2.25 mmol), K₂CO₃ (2.5 mmol), nano CuO (10 mol %) was refluxed in DMF (10 mL) for the appropriate time (8-12 h) till the consumption of starting material took place. The progress of the reaction was monitored by TLC using ethyl acetate and hexane as eluent. After completion of reaction, water (10 mL) and EtOAc (10 mL) were added and the organic layers were separated. The combined organic layers were dried on anhydrous Na₂SO₄ followed by the evaporation of the solvent to obtain the crude product, which was purified by column chromatography on silica gel using petroleum ether/EtOAc 9:1 as eluent to give desired products in excellent purity.
- Spectral data for all the compounds: Ethyl 4-methyl-6-phenyl-2-(phenylthio)pyrimidine-5-carboxylate (4a): Yellow solid, mp 87-88 °C. ¹H NMR 400 MHz (CDCl₃): δ 1.06 (3H, t, *J* = 6.8 Hz, CH₃), 2.51 (3H, s, CH₃), 4.16 (2H, q, *J* = 7.2 Hz, CH₂), 7.38-7.43 (4H, m, ArH), 7.51 (2H, d, *J* = 8 Hz, ArH), 7.65 (2H, d, *J* = 7.2 Hz, ArH). ¹³C NMR 100 MHz (CDCl₃): δ 13.63, 22.62, 61.80, 121.49, 128.39, 128.42, 128.95, 129.10, 129.46, 130.21, 135.22, 137.34, 163.57, 165.95, 168.12, 172.10. IR (cm⁻¹): 3059, 2980, 1718, 1524, 1215, 686. HRMS (ESI) Calcd for C₂₀H₁₈N₂O₂S (M+H) 351.10890. Found: 351.19550. Ethyl 4-methyl-2-(phenylthio)-6-p-tolylpyrimidine-5-carboxylate (4b): White

Entry 4-interlyi-2-(piterlinio)-6-p-100/py/intune-5-curboxylate (44), white solid, mp 62-63 °C. ¹H NMR 400 MHz (CDCl₃): δ 1.10 (3H, t, *J* = 7.12 Hz, CH₃), 2.35 (3H, s, CH₃), 2.49 (3H, s, CH₃), 4.20 (2H, q, *J* = 7.12 Hz, CH₂), 7.16 (2H, d, *J* = 8 Hz), 7.38-7.43 (5H, m, ArH), 7.63-7.65 (2H, m, ArH). ¹³C NMR 100 MHz (CDCl₃): δ 13.74, 21.41, 22.61, 61.80, 121.26, 128.42, 128.92, 129.05, 129.14, 129.54, 134.40, 135.26, 140.63, 163.30, 165.75, 168.36, 171.94. IR (cm⁻¹): 3058, 2922, 1715, 1609, 1524, 1222, 832, 687. HRMS (ESI) Calcd for C₂₁H₂₀N₂O₂S (M+H) 365.12455. Found: 365.24188.

Ethyl 4-(4-methoxyphenyl)-6-methyl-2-(phenylthio)pyrimidine-5-carboxylate (**4c**): White solid, mp 78–80 °C. ¹H NMR 400 MHz (CDCl₃): δ 1.05 (3H, t, J = 7.16 Hz, CH₃), 2.40 (3H, s, CH₃), 3.73 (3H, s, CH₃), 4.14 (2H, q, J = 7.16 Hz, CH₂), 6.78 (2H, d, J = 8 Hz, ArH), 7.32–7.34 (3H, m, ArH), 7.41 (2H, d, J = 8 Hz, ArH), 7.55–7.57 (2H, m, ArH). ¹³C NMR 100 MHz (CDCl₃): δ 13.83, 22.58, 29.72, 61.82, 113.84, 120.84, 128.93, 129.06, 129.52, 129.59, 130.17, 135.28, 161.49, 162.58, 165.66, 168.55, 171.80. IR (cm⁻¹): 3063, 2918, 1714, 1605, 1526, 1224, 841, 687. HRMS (ESI) Calcd for C₂₁H₂₀N₂O₃S (M+H) 381.11946. Found: 381.17743.

Ethyl 4-(3-methoxyphenyl)-6-methyl-2-(phenylthio)pyrimidine-5-carboxylate (4d): White solid, mp 69–70 °C. ¹H NMR 400 MHz (CDCl₃): δ 1.08 (3H, t, J = 7.2 Hz, CH₃), 2.51 (3H. s, CH₃), 3.72 (3H, s, OCH₃), 4.19 (2H, q, J = 7.2 Hz, CH₂), 6.93 (1H, d, J = 7.6 Hz, ArH), 7.07 (2H, d, J = 7.2 Hz, ArH), 7.23 (1H, s, ArH), 7.40 (3H, d, J = 6 Hz, ArH), 7.64 (2H, d, J = 7.6 Hz, ArH), 7.64 (2H, d, J = 7.6 Hz, ArH), 7.64 (2H, d, J = 7.6 Hz, ArH), 13C NMR 100 MHz (CDCl₃): δ 13.70, 22.59, 55.28, 61.86, 113.07, 117.02, 120.72, 121.49, 128.96, 129.09, 129.34, 129.53, 135.37, 138.52, 159.64, 163.10, 165.90, 168.13, 172.11. IR (cm⁻¹): 3074, 2922, 1714, 1597, 1524, 1223, 778, 686. HRMS (ESI) Calcd for C₂₀H₁₇N₃O₄S (M+H) 381.11946. Found: 381.22009.

Ethyl 4-(4-chlolophenyl)-6-methyl-2-(phenylthio)pyrimidine-5-carboxylate (**4e**): White Solid, mp 87–89 °C. ¹H NMR 400 MHz (CDCl₃): δ 1.04 (3H, t, *J* = 7.2 Hz, CH₃), 2.43 (3H, s, CH₃), 4.13 (2H, q, *J* = 7.48 Hz, CH₂), 7.21 (2H, d, *J* = 6.6 Hz), 7.33–7.38 (5H, m, ArH), 7.56 (2H, d, *J* = 6.84 Hz, ArH). ¹³C NMR 100 MHz (CDCl₃): δ 13.74, 22.67, 61.96, 99.99, 121.25, 128.67, 128.99, 129.22, 129.26, 129.79, 135.30, 135.72, 136.59, 162.17, 166.16, 167.94, 172.32. IR (cm⁻¹): 3072, 2922, 1719, 1605, 1523, 1219, 841, 687. HRMS (ESI) Calcd for C₂₀H₁₇ClN₂O₂S (M+H) 384.06993. Found: 384.17688.

Ethyl 4-(4-bromophenyl)-6-methyl-2-(phenylthio)pyrimidine-5-carboxylate (**4f**): White solid, mp 82–84 °C. ¹H NMR 400 MHz (CDCl₃): δ 1.11 (3H, t, *J* = 7.16 Hz, CH₃), 2.50 (3H, s, CH₃), 4.20 (2H, q, *J* = 7.12 Hz, CH₂), 7.36–7.42 (5H, m, ArH), 7.50 (2H, d, *J* = 8.48 Hz), 7.60 (2H, d, *J* = 8.68 Hz). ¹³C NMR 100 MHz (CDCl₃): δ 13.75, 22.68, 61.98, 121.23, 125, 129, 129.23, 130.01, 130.06, 131.64, 135.30, 136.19, 137.59, 162.24, 166.19, 167.91, 172.36. IR (cm⁻¹): 3071, 2920, 1718, 1588, 1523, 1224, 839, 698. HRMS (ESI) Calcd for C₂₀H₁₇BrN₂O₂S (M+H) 430.01941. Found: 431.16266.

Ethyl 4-(3-bromophenyl)-6-methyl-2-(phenylthio)pyrimidine-5-carboxylate (**4g**): White solid, mp 88–90 °C. ¹H NMR 400 MHz (CDCl₃): δ 1.04 (3H, t, CH₃, J = 7.16 Hz), 2.47 (3H, s, CH₃), 4.14 (2H, q, CH₂, J = 7.2 Hz), 7.33–7.37 (5H, m, ArH), 7.45 (1H, d, ArH, J = 7 Hz), 7.54–7.59 (3H, m, ArH). ¹³C NMR 100 MHz (CDCl₃): δ 13.80, 22.68, 62.00, 121.29, 122.50, 129.02, 129.25, 129.27, 129.85, 129.92, 131.59, 133.12, 135.30, 139.03, 139.18, 161.57, 161.73, 166.31, 172.42. IR (cm⁻¹): 3061, 2980, 1712, 1604, 1519, 1225, 780, 687. HRMS (ESI) Calcd for C₂₀H₁₇BrN₂O₂S (M+H) 428.01941. Found: 429.11279.

Ethyl 4-methyl-6-(3-nitrophenyl)-2-(phenylthio)pyrimidine-5-carboxylate (**4h**): White Solid, mp 98–99 °C. ¹H NMR 400 MHz (CDCl₃): δ 1.17 (3H, t, J = 7.2 Hz,

CH₃), 2.55 (3H, s, CH₃), 4.26 (2H, q, *J* = 7.6 Hz, CH₂), 7.44 (3H, d, *J* = 6.4 Hz, ArH), 7.54 (1H, t, *J* = 8 Hz, ArH), 7.65 (2H, d, *J* = 7.2 Hz, ArH), 7.84 (1H, d, *J* = 7.6 Hz, ArH), 8.27 (1H, d, *J* = 7.6 Hz, ArH), 8.41 (1H, s, ArH). ¹³C NMR 100 MHz (CDCl₃): δ 13.77, 22.81, 62.24, 121.35, 123.73, 124.81, 128.91, 129.15, 129.44, 129.49, 134.21, 135.32, 138.82, 148.23, 160.66, 166.71, 167.43, 172.89. IR (cm⁻¹):3080, 2923, 1724, 1598, 1519, 1218, 772, 688. HRMS (ESI) Calcd for C₂₀H₁₇N₃O₄S (M+H) 396.0933. Found: 396.10890.

 $1\mathcal{i}$ -(4-methyl-2-(phenylthio)-6-p-tolylpyrimidin-5-yl)ethanone (4i): White solid, mp 114-116 °C. ¹H NMR 400 MHz (CDCl₃): δ 2.03 (3H, s, CH₃), 2.36 (3H, s, CH₃), 2.41 (3H, s, CH₃), 7.19 (2H, d, J = 7.6 Hz, ArH), 7.38–7.42 (5H, m, ArH), 7.64–7.66 (2H, m, ArH). ¹³C NMR 100 MHz (CDCl₃): δ 21.44, 22.49, 32.06, 128.72, 128.93, 129.06, 129.50, 129.59, 134.20, 135.26, 141.22, 162.21, 164.59, 171.51, 204.55. IR (cm⁻¹): 3032, 2919, 1696, 1607, 1518, 1206, 833, 689. HRMS (ESI) Calcd for C₂₀H₁₈N₂OS (M+H) 335.11398. Found: 335.26208.

 $\begin{array}{l} 1-(4-(3-methoxyphenyl)-6-methyl-2-(phenylthio)pyrimidin-5-yl) \ ethanone \ (4j): \\ \mbox{White solid, mp } 62-63 \ ^{\circ}C. \ ^{1}H \ NMR \ 400 \ MHz \ (CDCl_3): \ \delta \ 2.05 \ (3H, s, CH_3), \\ 2.42 \ (3H, s, CH_3), \ 3.74 \ (3H, s, CH_3), \ 6.97-7.07 \ (3H, m, ArH), \ 7.21-7.29 \ (2H, m, ArH), \ 7.42 \ (2H, d, l = 6 \ Hz, \ ArH), \ 7.53 \ (1H, d, l = 6.8 \ Hz, \ ArH), \ 7.67 \ (1H, d, l = 7.6 \ Hz, \ ArH), \ 7.47 \ (2H, d, l = 6 \ Hz, \ ArH), \ 7.57 \ (1H, d, l = 6.8 \ Hz, \ ArH), \ 7.67 \ (1H, d, l = 7.6 \ Hz, \ ArH), \ 7.67 \ (1H, d, l = 6.8 \ Hz, \ A$

1-(4-(4-chlorophenyl)-6-methyl-2-(phenylthio)pyrimidin-5-yl) ethanone (**4k**): White solid, mp 95–96 °C. ¹H NMR 400 MHz (CDCl₃): δ 1.13 (3H, s, CH₃), 2.50 (3H, s, CH₃), 7.34 (2H, d, *J* = 8.5 Hz, ArH), 7.41–7.46 (5H, m, ArH), 7.63 (2H, d, *J* = 6.52 Hz, ArH). ¹³C NMR 100 MHz (CDCl₃): δ 22.66, 29.72, 121.26, 128.67, 128.98, 129.21, 129.79, 135.29, 135.73, 136.59, 162.16, 166.15, 167.93, 172.32, 199.87. IR (cm⁻¹): 3059, 2922, 1708, 1579, 1475, 1237, 830, 685. HRMS (ESI) Calcd for C₁₉H₁₅ClN₂OS (M+H) 355.05936. Found: 355.26031.

 $\begin{array}{l} 1-(2-(4-chlorophenylthio)-4-methyl-6-phenylpyrimidin-5-yl)ethanone \quad \textbf{(41)}: \\ \text{White solid, mp } 92-94~\text{C}. \ ^{1}\text{H}\ \text{NMR}\ 400\ \text{MHz}\ (\text{CDCl}_3): \ \delta\ 2.02\ (3\text{H}, \ \text{s}, \ \text{CH}_3), \\ 2.42\ (3\text{H}, \ \text{s}, \ \text{CH}_3), \ 7.38-7.51\ (7\text{H}, \ \text{m}, \ \text{ArH}), \ 7.58\ (2\text{H}, \ \text{d}, \ J=6.8\ \text{Hz}, \ \text{ArH}), \ ^{1}3C\ \text{NMR} \\ 100\ \text{MHz}\ (\text{CDCl}_3): \ \delta\ 2.2.51\ , \ 32.04, \ 127.92, \ 128.94, \ 129.01, \ 129.17, \ 130.83, \\ 135.41, \ 136.92, \ 162.39, \ 164.87, \ 171.04, \ 204.23.\ \text{IR}\ (\text{cm}^{-1}): \ 3057, \ 2923, \ 1696, \end{array}$

1578, 1519, 1225, 842, 688. HRMS (ESI) Calcd for $C_{19}H_{15}ClN_2OS~(M\text{+}H)$ 355.05936. Found: 355.28076.

Solution 2010 Section 2.5 (4) Solution 2.5 (5) Solution

Ethyl 4-methyl-6-(3-nitrophenyl)-2-(p-tolylthio)pyrimidine-5-carboxylate (**4p**): Yellow solid, mp 82–84 °C. ¹H NMR 400 MHz (CDCl₃): δ 1.08 (3H, t, *J* = 7.2 Hz), 2.33 (3H, s, CH₃), 2.47 (3H, s, CH₃), 4.18 (2H, q, *J* = 7.2 Hz, CH₃), 7.18 (2H, d, *J* = 7.16 Hz, ArH), 7.43–7.52 (3H, m, ArH), 7.75 (2H, d, *J* = 6.28 Hz, ArH), 8.21 (2H, d, *J* = 6 Hz, ArH), 8.33 (1H, s, ArH). ¹³C NMR 100 MHz (CDCl₃): δ 12.72, 20.36, 21.77, 61.16, 120.17, 122.78, 123.75, 124.31, 128.34, 128.93, 133.14, 134.21, 137.87, 138.76, 140.44, 147.24, 159.55, 165.60, 172.23. IR (cm⁻¹): 3057, 2922,2852, 1703, 1565, 1515, 1217, 690, 805. HRMS (ESI) CalCd for C₂₁H₁₉N₃O₄S (M+H) 410.10963. Found: 410.19470.