## Palladium-Catalyzed Asymmetric Alkylation of 2,3-Alkadienyl Phosphates. Synthesis of Optically Active 2-(2,3-Alkadienyl)malonates

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Asymmetric alkylation of 2,3-alkadienyl phosphates with soft carbon nucleophiles proceeds efficiently in the presence of palladium complex catalyst bearing MeOBIPHEP or BINAP ligand to give optically active functionalized allenes up to 90% ee.

Optically active allenes bearing axial chirality are important synthetic intermediates for various naturally occurring biologically active compounds.1 Therefore, much attention has been paid to the asymmetric synthesis of this class of compounds. Optically active allenes have been prepared by either optical resolution of racemic allenes<sup>1</sup> or chirality transfer reactions of optically active precursors.<sup>2,3</sup> Preparation of optically active allenes using chiral catalyst is an attractive method; however, the reported methods give moderate enantioselectivity (20-60% ee)<sup>4</sup> except kinetic resolution by asymmetric oxidative degradation of allenes.<sup>5</sup> Recently, Hayashi et al. reported that palladium-catalyzed reaction of 2-bromo-1,3butadienes gives optically active allenes with enantioselectivity up to 89% ee.<sup>6</sup> We wish to report that palladium-catalyzed asymmetric reaction of racemic 2,3-alkadienyl phosphates 1 with soft carbon nucleophiles 2 gives optically active functionalized allenes 3 (Scheme 1).<sup>7</sup> This is based on the non-asymmetric reaction reported by Gore et al.8

Alkylations of 2,3-alkadienyl phosphates **1** with soft carbon nucleophiles **2** were examined in the presence of *N*, *O*-bis(trimethylsilyl)acetamide (BSA) and the palladium catalyst prepared from  $Pd_2(dba)_3$ •CHCl<sub>3</sub> and MeOBIPHEP<sup>9</sup> or BINAP<sup>10</sup> in THF at room temperature (Scheme 1). The representative results of the reactions are summarized in Table 1.

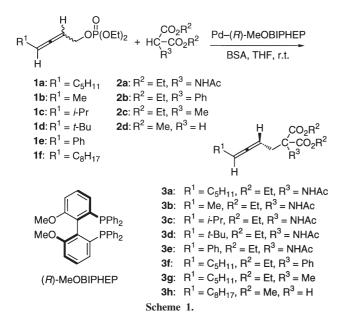


Table 1. Asymmetric alkylation of diethyl 2,3-alkadienyl phosphates 1 with malonates  $2^{\rm a}$ 

Entry	Phosphate	Malonate	Time/h	Product	Yield/% <sup>b</sup>	Ee/% <sup>c</sup>
1	1a	2a	8	3a	76	73
2	1b	2a	6	3b	72	77
3	1c	2a	6	3c	89	81
4	1d	2a	12	3d	69	90
5	1e	2a	8	3e	80	60
6	1a	2b	2	3f	85	76
7	1a	2c	3	3g	74	69
8	1f	2d	8	3h	90	70

<sup>a</sup>All reactions were carried out in THF (0.1 M) at room temperature. The ratio of **1/2**/BSA/Pd<sub>2</sub>(dba)<sub>3</sub> •CHCl<sub>3</sub>/(R)-MeOBIPHEP was 100/110/120/ 1/4. <sup>b</sup>Isolated yield based on **1**. <sup>c</sup>Determined by HPLC analysis with chiral stationary phase column.

The palladium-catalyzed reactions of phosphates **1a–e** with diethyl 2-acetamidomalonate (**2a**) give the corresponding functionalized allenes **3a–e** with 60–90% ee in good yields (entries 1–5). The catalytic reactions of the phosphate **1a** with diethyl 2-phenylmalonate (**2b**) and diethyl 2-methylmalonate (**2c**) gave **3f** and **3g**, respectively, with similar enantio-selectivities (entries 6 and 7), indicating that the acetamido group is not essential for getting high level of enantioselectivity. Noteworthy is that the reaction of the phosphate **1f** with dimethyl malonate (**2d**) afforded monosubstituted malonate **3h** as a sole product. Disubstituted malonate could not be detected among the products (entry 8).

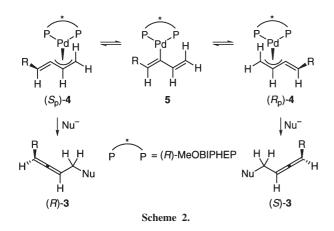
Typically, asymmetric reaction of diethyl 5,5-dimethyl-2,3hexadienyl phosphate (1d) with acetamidomalonate 2a was carried out in THF at room temperature for 12 h in the presence of BSA and the palladium catalyst prepared *in situ* from Pd<sub>2</sub>(dba)<sub>3</sub>•CHCl<sub>3</sub> (0.01 equiv) and (*R*)-MeOBIPHEP (0.04 equiv). Ethyl 2-acetamido-2ethoxycarbonyl-7,7-dimethyl-4,5-octadienoate [(-)-3d] was obtained with 90% ee in 69% isolated yield. The absolute configuration of (-)-3d was determined to be (*R*) on the basis of the Lowe's rule.<sup>11</sup>

A survey of chiral bidentate phosphine ligands revealed that the  $C_2$  ligands bearing binaphthyl and biphenyl backbone, such as MeOBIPHEP and BINAP are effective.

The palladium-catalyzed reaction of 2,3-alkadienyl phosphates **1** with soft carbon nucleophiles<sup>8</sup> can be rationalized by assuming the mechanism shown in Scheme 2. The first step is facile formation of  $1-3-\eta^3-1,3$ -alkadienylpalladium **4** from allyl esters bearing highly reactive leaving group OP(O)(OEt)<sub>2</sub>.<sup>12</sup> Soft carbon nucleophiles attack at the less hindered  $\alpha$ -position of the palladium complex **4**. It is in contrast to the  $\gamma$ -attack of hard carbon nucleophiles,<sup>8,13</sup> such as organozinc and Grignard reagents, and carbon monoxide<sup>14</sup> via  $\eta^1$ -1,3-alkadien-3-yl-palladium **5** to give conjugated dienes.

An equilibrium is present between two diastereomeric  $\eta^3$ alkadienylpalladium complexes ( $S_p$ )-4 and ( $R_p$ )-4 through  $\eta^1$ alkadienylpalladium complex 5. Preferential nucleophilic attack of malonate anion to ( $S_p$ )-4 at the less hindered allylic terminus from the opposite side to the palladium would give the optically active

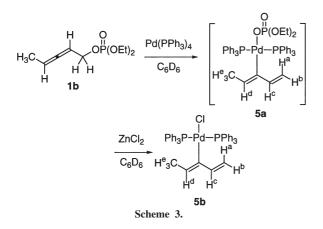
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functionalized allene (*R*)-3.

Fast equilibrium between two diastereomeric  $\eta^3$ -alkadienylpalladium complexes (*S*<sub>p</sub>)-**4** and (*R*<sub>p</sub>)-**4** was confirmed by the following control experiment. When enantiomerically pure (*S*)-**1a**<sup>15</sup> was allowed to react with malonate **2a** in the presence of BSA and the palladium catalyst, prepared from Pd<sub>2</sub>(dba)<sub>3</sub>•CHCl<sub>3</sub> and 1,4bisdiphenylphosphinobutane, in THF at room temperature for 2 h, the racemic allene **3a** was obtained in 76% yield.

The participation of  $\eta^1$ -alkadienylpalladium complex **5** was confirmed from the reaction of phosphate **1b** with Pd(PPh<sub>3</sub>)<sub>4</sub>. The <sup>1</sup>H and <sup>31</sup>P NMR spectra of the product derived from the reaction of the phosphate **1b** with an equivalent of Pd(PPh<sub>3</sub>)<sub>4</sub> in C<sub>6</sub>D<sub>6</sub> in an NMR tube showed a facile, exclusive formation of  $\eta^1$ -pentadienylpalladium **5a**.<sup>16</sup> The chloro complex **5b** was obtained by ligand exchange reaction upon treatment with zinc chloride as shown in Scheme 3. The structure of **5b** was confirmed by single-crystal Xray analysis.<sup>17</sup>



In summary, we succeeded in asymmetric synthesis of functionalized allenes **3** with high enantioselectivity by the palladium-catalyzed asymmetric alkylation of racemic 2,3-alkadienyl phosphates **1** using palladium–MeOBIPHEP catalyst. Further mechanistic study and application are in progress.

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- 15 Enantiomerically pure (*S*)-**1a** { $[\alpha]_D$  48.1 ° (*c* 1.1, CHCl<sub>3</sub>)} was obtained by the optical resolution of racemic **1a** by HPLC with a chiral stationary phase (Chiralcel OD).
- 16 **5a**: <sup>1</sup>H NMR (500 MHz, C<sub>6</sub>D<sub>6</sub>, selected data)  $\delta$  0.60 (t, J = 7 Hz, 6 H), 1.57 (d, J = 7 Hz, 3 H, H<sup>e</sup>), 3.25 (dq, J = 7, 7 Hz, 4 H), 4.51 (br, 1 H, H<sup>d</sup>), 4.60 (d, J = 10 Hz, H<sup>b</sup>), 5.27 (dd, J = 10, 17 Hz, H<sup>c</sup>), 6,13 (d, J = 17 Hz, H<sup>a</sup>). **5b**: <sup>1</sup>H NMR (500 MHz, C<sub>6</sub>D<sub>6</sub>, selected data)  $\delta$  1.72 (d, J = 7 Hz, 3 H, H<sup>e</sup>), 4.71 (dd, J = 2, 11 Hz, H<sup>b</sup>), 4.97 (br, 1 H, H<sup>d</sup>), 5,68 (ddd, J = 2, 10, 16 Hz, H<sup>e</sup>), 6,07 (dd, J = 2, 10 Hz, H<sup>a</sup>), <sup>31</sup>P{<sup>1</sup>H} NMR (202 MHz, C<sub>6</sub>D<sub>6</sub>)  $\delta$  25.73. For related ( $\eta^{1}$ -1,3-butadien-2-yl)palladium complexes generated from Pd(PPh<sub>3</sub>)<sub>4</sub> and 2,3-butadienyl chlorides, see: S. A. Benyunes, L. Brandt, A. Fries, M. Green, M. F. Mahon, and T. M. T. Papworth, *J. Chem. Soc., Dalton Trans.*, **1993**, 3785.
- 17 Crystal data for **5b**•3C<sub>6</sub>H<sub>6</sub>: C<sub>59</sub>H<sub>55</sub>ClP<sub>2</sub>Pd, fw = 967.82, Rigaku RAXIS-RAPID, monoclinic, *Cc* (No. 9), *a* = 20.1149(10) A, *b* = 17.3353(11) A, *c* = 13.9252(8) A,  $\beta$  = 94.843(2)°, *V* = 4838.4(5) A<sup>3</sup> *Z* = 4, *T* = 101(2) K, *D*<sub>calcd</sub> = 1.329 Mg m<sup>-3</sup>,  $\lambda$  = 0.71069 A, *F*(000) = 2008, *R*<sub>1</sub> = 0.0314; *wR*<sub>2</sub> = 0.0676 for all 5532 reflections, GOF = 1.040.