2010 Vol. 12, No. 1 1-3

Synthesis of 2-Hydroxymethyl Ketones by Ruthenium Hydride-Catalyzed Cross-Coupling Reaction of α,β -Unsaturated Aldehydes with Primary Alcohols

Aurélien Denichoux, Takahide Fukuyama, Takashi Doi, Jiro Horiguchi, and Ilhyong Ryu*

Department of Chemistry, Graduate School of Science, Osaka Prefecture University, Sakai, Osaka 599-8531, Japan

ryu@c.s.osakafu-u.ac.jp

Received October 3, 2009

ABSTRACT

The cross-coupling reaction of α , β -unsaturated aldehydes with primary alcohols to give 2-hydroxymethyl ketones was achieved using RuHCl(CO)(PPh₃)₃ as a catalyst. This atom-economical reaction is likely to proceed via the hydroruthenation of α , β -unsaturated aldehydes followed by an aldol reaction of the resultant enolates with aldehydes to give α -formylated ketones, which undergo transfer hydrogenation with primary alcohols leading to α -hydroxymethyl ketones. The reduction step can generate aldehydes, participating in the next catalytic cycle.

Development of new catalytic coupling processes accompanied with atom economy is highly desirable in organic synthesis.^{1,2} In this regard, we are interested in the potentials of ruthenium hydride-catalyzed bond forming reactions in conjunction with the use of readily available oxygenated substrates, such as alcohols, ketones, and aldehydes.^{3–5} Recently, we have reported that dimerization of primary

unsaturated alcohols^{4a} and reductive dimerization of α,β -unsaturated aldehydes⁵ were effectively catalyzed by RuH-Cl(CO)(PPh₃)₃. In the latter homocoupling reaction of enals, secondary alcohols such as isopropanol act as a hydrogen source, which is converted to acetone, an inert compound in the system (Scheme 1, eq 1). Since the pioneering work of Gregorio and co-workers on catalytic dimerization of primary alcohols,⁶ many researchers have pursued the potential of cross-coupling reactions of primary alcohols via transfer hydrogenation.⁷ In this regard, recent work by

⁽¹⁾ Beller, M.; Bolm, C., Eds. *Transition Metals for Organic Synthesis*; Wiley-VCH: Weinheim, 1998; Vols. *1 and 2*. (b) Cornils, B.; Herrmann, A. W., Eds. *Applied Homogeneous Catalysis with Organometallic Compounds*; VCH: Weinheim, 1996; Vol. 1 and 2.

^{(2) (}a) Trost, B. M. Science **1991**, 254, 1471. (b) Trost, B. M. Angew. Chem., Int. Ed. Engl. **1995**, 34, 259. (c) Trost, B. M.; Frederiksen, M. U.; Rudd, M. T. Angew. Chem., Int. Ed. **2005**, 44, 6630.

⁽³⁾ For selected reviews of ruthenium-catalyzed C—C coupling, see: (a) Bruneau, C.; Dixneuf, H. P., Eds. *Ruthenium Catalysts and Fine Chemisty*; Springer-Verlag: Berlin, Heidelberg, 2004. (b) Naota, T.; Takaya, H.; Murahashi, S.-I. *Chem. Rev.* **1998**, 98, 2599. (c) Trost, B. M.; Toste, F. D.; Pinkerton, A. B. *Chem. Rev.* **2001**, 101, 2067. (d) Kondo, T; Mitsudo, T.-a. *Curr. Org. Chem.* **2002**, 6, 1163. (e) Trost, B. M.; Frederiksen, M. U.; Rudd, M. T. *Angew. Chem., Int. Ed.* **2005**, 44, 6630. (f) Shibahara, F.; Krische, M. J. *Chem. Lett.* **2008**, 37, 1102.

⁽⁴⁾ For our previous work on the use of the ruthenium hydride catalyst, see: (a) Doi, T.; Fukuyama, T.; Minamino, S.; Husson, G.; Ryu, I. *Chem. Commun.* 2006, 1875. (b) Doi, T.; Fukuyama, T.; Horiguchi, J.; Okamura, T.; Ryu, I. *Synlett* 2006, 721. (c) Fukuyama, T.; Doi, T.; Minamino, S.; Omura, S.; Ryu, I. *Angew. Chem., Int. Ed.* 2007, 46, 5559. (d) Omura, S.; Fukuyama, T.; Horiguchi, J.; Murakami, Y.; Ryu, I. *J. Am. Chem. Soc.* 2008, 130, 14094. (e) Omura, S.; Fukuyama, T.; Murakami, Y.; Okamoto, H.; Ryu, I. *Chem. Commun.* 2009, 6741.

⁽⁵⁾ Doi, T.; Fukuyama, T.; Minamino, S.; Ryu, I. Synlett 2006, 3013.
(6) Gregorio, G.; Pregaglia, G. F.; Ugo, R. J. Orgnomet. Chem. 1972, 37, 385.

Krische and co-workers that demonstrates Ru- and Ircatalyzed cross-coupling of alcohols with unsaturated C–C double and triple bonds is noteworthy. We thought that if primary alcohols, such as benzyl alcohol, are employed as a hydrogen source, the resulting aldehydes would act as a coupling partner to give cross-coupling products (Scheme 1, eq 2). We report herein the Ru–H-catalyzed atomeconomical cross-coupling reaction of α , β -unsaturated aldehydes with primary alcohols leading to α -hydroxymethyl ketones.

Scheme 1. Ru-H Catalyzed Coupling Reactions of Enals

Reductive homocoupling of enals (ref 5)

$$\frac{\text{hydrogen source}}{\text{Me}_2\text{CHOH}} + \frac{\text{O}}{\text{H}} + \frac{\text{Ru-H}}{\text{R}} + \frac{\text{Ne}_2\text{CO}}{\text{I}}$$

Cross-coupling of enals with primary alcohols (this work)

When a benzene solution of benzyl alcohol (1a, 1.2 equiv) and 2-hexenal (2a) in the presence of RuHCl(CO)(PPh₃)₃ (10 mol %) was heated under reflux for 2 h (Procedure A, Scheme 2), 2-hydroxymethyl-1-phenyl-1-hexanone (3a) was obtained in 40% yield. In this case, a significant amount of dimer derived by reductive coupling of 2a was formed as byproduct (3a/dimer = 5/1). To suppress the undesired dimerization course, 2a was added slowly using a syringe pump over the period of 1 h, then the resulting mixture was stirred for another 1 h (Procedure B). In this case, the desired 3a was obtained in 61% yield. Since in the initial stage of the reaction 2a was consumed by transfer hydrogenation to produce benzaldehyde, we decided to add a small amount of aldehyde beforehand (Procedure C). This caused further improvement of the yield of 3a up to 72%.

Scheme 2. Optimization of the Reaction Conditions

Having an optimized procedure C in hand, we then examined the generality of the present cross-coupling reaction with various alcohols and enals (Table 1). The reactions of 2-hexenal (2a) with several benzylic alcohols, having an

Table 1. Cross-Coupling Reaction of α,β -Unsaturated Aldehydes with Benzylic Alcohols^a

Aldehydes with Benzylic Alcohols ^a				
entry	benzyl alcohol 1	enal 2	β-hydroxymethyl ketone 3	yield ^b
1	ОН	2a	За ОН	72%
2	OMe 1b	2a	О ОМе 3b	53%
3	OH CF ₃ 1c	2a	О СF ₃	66%
4	OMe 1d	2a	OMe 3d	59%
5	CI 1e	2a	CI OH	74%
6	OH 1f	2 a	он 3f	67%
7°	S OH	2a	S ОН	64%
8	O OH	2a	о он он за вышения за вышения вышени вышения вышени вышения вышения вышения вышения вышения вышения вышения в	66%
9	1a	H	ОН	66%
10	1a	2b O 2c	3i O O O	45%
11	1a	O H	3k	60%
12°	1e [O H 2e	OH OH	69%
13	1і	2a	О ОН 3m	30%

^a Conditions: **1** (1.2 equiv), **2** (0.8 mmol), corresponding aldehyde (10 mol %), RuHCl(CO)(PPh₃)₃ (0.08 mmol), benzene (4 mL). A benzene solution of **1** (5 mL) was added using a syringe pump over the period of 1 h, then the mixture was heated at reflux for 1 h. ^b Isolated yield after flash chromatography on SiO₂. ^c A benzene solution of **1g** and **2a** was added over the period of 1 h.

Org. Lett., Vol. 12, No. 1, 2010

electron-donating substituent or an electron-withdrawing substituent, gave the 2-hydroxymethyl ketone **3** in good yields (entries 1–6). The reaction also worked well with alcohols having heteroaromatic rings such as 2-thiophenemethanol (**1g**) and 2-furanmethanol (**1h**) (entries 7 and 8). In the case of the alcohol **1g**, transfer hydrogenation of **1g** and **2a** leading to 2-thiophenecarboxyaldehyde and 1-hexanol was fast with the procedure C. Thus, syringe pump addition of a mixture solution of **1g** and **2a** was examined, giving the desired coupling product **3g** in good yield. Disubstituted enal **2c** also reacted with **1a**, whereas the yield of **3j** was rather modest (entry 10). Enals having an aromatic substitu-

(8) (a) Bower, J. F.; Skucas, E.; Patma, R. L.; Krische, M. J. *J. Am. Chem. Soc.* **2007**, *129*, 15134. (b) Shibahara, F.; Bower, J. F.; Krische, M. J. *J. Am. Chem. Soc.* **2008**, *130*, 6338. Also see recent reviews: (c) Bower, J. F.; Kim, I. S.; Patman, R. L; Krische, M. J. *Angew. Chem., Int. Ed* **2009**, *48*, 34, and ref 3f and references therein.

(9) For studies on ruthenium enolate complexes, see: (a) Hartwig, J. F.; Andersen, R. A.; Bergman, G. B. *J. Am. Chem. Soc.* **1990**, *112*, 5670. (b) Hartwig, J. F.; Bergman, R. G.; Andersen, R. A. *Organometallics* **1991**, *10*, 3326. (c) Tasley, B. T.; Rapta, M.; Kulawiec, R. J. *Organometallics* **1996**, *15*, 2852.

(10) For examples of catalytic transformations based on ruthenium enolates, see: (a) Matsuda, I.; Shibata, M.; Sato, S. *J. Organomet. Chem.* **1988**, *340*, C5. (b) Sato, S.; Matsuda, I.; Shibata, M. *J. Organomet. Chem.* **1989**, *377*, 347. (c) Murahashi, S.-I.; Naota, T.; Taki, H.; Mizuno, M.; Takaya, H.; Komiya, S.; Mizuho, Y.; Oyasato, N.; Hiraoka, M.; Hiraoka, M.; Fukuoka, A. *J. Am. Chem. Soc.* **1995**, *117*, 12436. (d) Trost, B. M.; Pinkerton, A. B. *J. Am. Chem. Soc.* **2000**, *122*, 8081. (e) Uma, R.; Davies, M.; Crévisy, C.; Grée, R. *Tetrahedron Lett.* **2001**, *42*, 3069.

(11) For Ru-catalyzed aldol type reactions, see: (a) Naota, T.; Taki, T.; Mizuno, M.; Murahashi, S.-I. J. Am. Chem. Soc. 1989, 111, 5954. (b) Mizuho, Y.; Kasuga, N.; Komiya, S. Chem. Lett. 1991, 2127. (c) Wang, M.; Yang, X.-F.; Li, C.-J. Eur. J. Org. Chem. 2003, 998. (d) Yang, X.-F.; Wang, M.; Varma, R. S.; Li, C.-J. Org. Lett. 2003, 5, 657. (e) Yang, X.-F.; Wang, M.; Varma, R. S.; Li, C.-J. J. Mol. Catal. A: Chem. 2004, 214, 147. (f) Martín-Matute, B.; Bogár, K.; Edin, M.; Kaynak, F. B.; Bäckvall, J.-E. Chem.—Eur. J. 2005, 11, 5832. (g) Bartoszewicz, A.; Livendahl, M.; Martín-Matute, B. Chem.—Eur. J. 2008, 14, 10547.

(12) For reviews on catalytic reductive aldol coupling, see: (a) Motherwell, W. B. *Pure Appl. Chem.* **2002**, *74*, 135. (b) Huddleston, R. R.; Krische, M. J. *Synlett* **2003**, 12. (c) Jang, H.-Y.; Krische, M. J. *Acc. Chem. Res.* **2004**, *37*, 653. (e) Chiu, P. *Synthesis* **2004**, 2210.

(13) For reviews on transfer hydrogenation, see: (a) Brieger, G.; Nestrick, T. J. Chem. Rev. 1974, 74, 567. (b) Zassinovich, G.; Mestroni, G.; Gladiali, S. Chem. Rev. 1992, 92, 1051. (c) Noyori, R.; Hashiguchi, S. Acc. Chem. Res. 1997, 30, 97. (d) Bäckvall, J.-E. J. Organomet. Chem. 2002, 652, 105. (e) Everaere, K.; Mortreux, A.; Carpentier, J.-F. Adv. Synth. Catal. 2003, 345, 67. Transfer hydrogenation reactions catalyzed by RuHCl(CO)(PPh₃)₃: (f) Watanabe, Y.; Ohta, T.; Tsuji, Y. Bull. Chem. Soc. Jpn. 1982, 55, 2441. (g) Gordon, E. M.; Gaba, D. C.; Jebber, K. A.; Zacharias, D. M. Organometallics 1993, 12, 5020. (h) Hiraki, K.; Nonaka, A.; Matsugana, T.; Kawano, H. J. Organomet. Chem. 1999, 574, 121. Also see ref 8.

ent at the β -position, such as **2d** and **2e**, gave the corresponding 2-hydroxymethyl ketones in good yields (entries 11 and 12). The reaction of nonbenzylic type alcohol **1i** also gave the coupling **3m** in 30%, which requires further efforts for yield optimization.

A possible mechanism for this enal/alcohol coupling reaction is shown in Scheme 3 with an example of the reaction of **1a** and **2a**. The hydroruthenation of **2a** would give ruthenium enolate \mathbf{A} , which then undergoes aldol reaction with benzaldehyde. He resulting aldol adduct \mathbf{B} would lead to the keto aldehyde \mathbf{C} . Finally, transfer hydrogenation between the aldehyde moiety of \mathbf{C} and benzyl alcohol (**1a**) gives 2-hydroxymethyl ketone **3a** and benzaldehyde which then reacts with another molecule of ruthenium enolate \mathbf{A} , creating a catalytic cycle.

Scheme 3. Possible Mechanism for the Coupling of 1a and 2a

In summary, we have shown that a novel cross-coupling reaction of enals with primary alcohols is effectively catalyzed by RuHCl(CO)(PPh₃)₃, which leads to good yields of 2-hydroxymethyl ketones. The reaction is likely to proceed via an aldol reaction of a ruthenium enolate followed by a transfer hydrogenation, where primary alcohols act as both a hydrogen source and a latent aldehyde. The simplicity, atom efficiency, mild reaction conditions, and short reaction times make this an appealing methodology for accessing 2-hydroxymethyl ketones. Synthetic applications of the present reaction are currently underway in this laboratory.

Acknowledgment. A.D. thanks a Grant-in-Aid for JSPS Fellows. T.F. and I.R. thank JSPS and MEXT Japan for funding.

Supporting Information Available: Experimental procedure and compound characterization. This material is available free of charge via the Internet at http://pubs.acs.org. OL902289R

Org. Lett., Vol. 12, No. 1, 2010

⁽⁷⁾ For reviews, see: (a) Guillena, G.; Ramón, D. J.; Yus, M. Angew. Chem., Int. Ed. 2007, 46, 2358. (b) Hamid, M. H S. A.; Slatford, P. A.; Williams, J. M. Adv. Synth. Catal. 2007, 349, 1555 For recent examples. (c) Cho, S. K.; Kim, B. T.; Kim, T.-J.; Shim, S. C. J. Org. Chem. 2001, 66, 9020. (d) Taguchi, K.; Nakagawa, H.; Hirabayashi, T.; Sakaguchi, S.; Ishii, Y. J. Am. Chem. Soc. 2004, 126, 72. (e) Fujita, K.; Asai, C. Yamaguchi, T.; Hanasaka, F.; Yamaguchi, R. Org. Lett. 2005, 7, 4017. (f) Cho, C. S. J. Mol. Catal. A: Chem. 2005, 240, 55. (g) Martínez, R.; Ramón, D. J.; Yus, M. Tetrahedron 2006, 62, 8988. (h) Löfberg, C.; Grigg, R.; Whittaker, M. A.; Keep, A.; Derrick, A. J. Org. Chem. 2006, 71, 8023. (i) Hall, M. I.; Pridmore, S. J.; Williams, J. M. J. Adv. Synth. Catal 2008, 350, 1975. (j) Pridmore, S. J; Williams, J. M. J. Tetrahedron Lett. 2008, 49, 7413. (k) Jensen, T.; Madsen, R. J. Org. Chem. 2009, 74, 3990. (1) Grigg, R.; Lofberg, C.; Whitney, S.; Sridharan, V.; Keep, A.; Derrick, A. Tetrahedron 2009, 65, 849. (m) Shimizu, K.; Sato, R.; Satsuma, A. Angew. Chem., Int. Ed. 2009, 48, 3982.