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A Novel Synthesis of Chain-Extended Amino Sugar Derivatives through Aza-Cope Rearrangement of N-Galactosyl-N-Homoallylamines

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Abstract: Chain-extended amino sugars are synthesized via stereocontrolled Lewis acid-catalyzed aza-Cope rearrangement of *N*-glycosyl homoallylamines in high yield and stereoselectivity. © 1998 Elsevier Science Ltd. All rights reserved.

In recent years, so-called "higher sugars" have been receiving increasing attention, because several of these 7-11 carbon carbohydrates play prominent roles in biological processes. Important representatives of these higher sugars are KDO¹ (3-deoxy-D-manno-2-octulosonic acid) and L-glycero-D-manno-heptopyranose,² essential components of bacterial lipopolysaccharides, and the amino sugar *N*-acetylneu-raminic acid,³ an important component of human and animal glycoconjugates. A number of strategies for the carbon chain extension of appropriate aldoses have been developed,⁴ including syntheses of C-glycosides and C-nucleosides.⁵ The interest in higher sugars was further enforced by the observation that a 4-guanidino-*N*-acetylneuraminic acid derivative proved to be an effective inhibitor of the influenza virus sialidase and therewith a promising antiviral agent.⁶ Only a few chain-extending reactions on carbohydrates have been described in which amino-substituted stereogenic centers are formed. These are the Strecker syntheses on aldoses⁷ and the recently reported reactions of Grignard reagents with N-alkyl glycosylamines.⁸

We here describe an alternative stereoselective synthesis of chain-extended amino sugars achieved by an aza-Cope rearrangement of N-galactosyl-N-homoallylamines. This method avoids basic conditions. Simple O-acyl protecting groups can therefore be applied. N-Galactosyl-N-homoallylamines 2^9 were obtained with high asymmetric induction by SnCl₄-induced addition of allyltrimethylsilane or allyltributylstannane to aromatic and aliphatic Schiff bases 1 of 2,3,4,6-tetra-O-pivaloyl- β -D-galactopyranosylamine (Scheme 1). We were initially interested in a Lewis acid-induced Cope rearrangement of the (S)-unsaturated homoallylamine 2a obtained from the (E)-N-cinnamylidene derivative 1a with a high diastereomeric ratio (d.r. = 16-19 :1). This rearrangement could not be achieved under thermal conditions or by Lewis base- or Lewis acid-catalysis. Notwithstanding, with two of the investigated Lewis acid catalysts, BF₃ etherate and TiCl₄, an immaculate conversion of 1a occurred yielding quantitatively the unexpected chain-extended imino derivative 3a (Scheme 1). In a simple procedure, BF₃ etherate was added to a solution of the homoallylamine 2a in xylene at room temperature. After completion of the reaction (t.l.c. control) and hydrolysis, the imine $3a^{10}$ was obtained with a high diastereoselectivity (d.r. > 15:1, Table 1).

Various Lewis acids were tested on the N-galactosyl-N-homoallylamine 2a under different conditions. Titanium tetrachloride also produced the imine 3a in quantitative yield but with lower diastereoselectivity. The tin tetrachloride-induced reaction proceeded only at high temperature (refluxing tetrahydrofuran) with concomitant decrease in yield and diastereoselectivity. Zinc chloride was unable to catalyze the reaction. The obtained results are summarized in Table 1.



Scheme 1. a: Allyltrimethylsilane or allyltributylstannane, SnCl₄, THF. Piv = (CH₃)₃C-CO; b: 1.1 eq. Lewis acid (conditions see Table 1).

Single crystals of the N-cinnamylidene imine 3a were obtained by recrystallization from n-pentane. The X-ray analysis revealed that the major diastereomer 3a possessed (R)-configuration at the new chiral center.¹¹ The stereoselective formation of the chain-extended imino derivatives can be explained by a two step reaction: Coordination of the Lewis acid to the ring-oxygen atom of the carbohydrate moiety results in ring opening and is followed by a 2-azonia-[3,3]-sigmatropic rearrangement (cationic aza-Cope rearrangement¹²) of the intermediate iminium ion¹³ (Scheme 2).



Scheme 2: a: Lewis acid L.A.; b: Hydrolysis.

The attack of the allyl group from the *Re*-side of the iminium ion is obviously controlled by the configuration of the homoallylamine in a chair-like transition state.

The Lewis acid-catalyzed chain extension was also applied to two other aromatic N-galactosyl-Nhomoallylamines 2b and 2c, prepared from the Schiff bases 1b and 1c, respectively. The reactions were conducted using boron trifluoride etherate or titanium tetrachloride as Lewis acids under different conditions (Table 1). In both cases, the expected chain-extended imino compounds 3b and 3c were formed in high yield and diastereoselectivity (Scheme 1). Enrichment of the major diastereomer 3b was achieved by a single recrystallization from n-hexane.

Educt 2	Lewis acid	Solvent	TPC	t	Product 3	Yield [%]	d.r.[a]
2a	BF ₃ ·Et ₂ O	xylene	r. t.	12h	3 a	99[ь]	>15:1
2a	BF3·Et2O	THF	reflux	9h	3a	40[c]	11:1
2a	Ti Cl 4	toluene	1) -30 to -10 2) r.t.	1)1h 2)5min	3a	97[b]	>9 :1
2a	TiCl4	CH ₂ Cl ₂	-30 to r. t.	5h	3 a	96[b]	11:1
2a	SnCl ₄	THF	reflux	3.5h	3a	69[c]	4:1
2a	ZnCl ₂	THF	-30 to reflux	24h			
2ь	BF3·Et2O	xylene	r. t.	12h	3b	74[c]	6:1
2b	TiCl4	toluene	-30 to r. t.	3h	3b	96[ь]	9 :1
2c	BF3·Et2O	xylene	r. t.	12h	3c	99 [ь]	>20:1

Table 1. Synthesis of the chain-extended amino sugar derivatives 3 according to Scheme 1.

[a] Diastereomeric ratio determined from the crude product by ¹H NMR. [b] Further purification not necessary. [c] Yield after preparative thin layer chromatography.

In order to prepare the chain-extended amino sugar derivative 4, the N-cinnamylidene derivative 3a was N-deprotected. In a not yet optimized procedure, the imine 3a was first O-acetylated with acetic anhydride in pyridine. Treatment of the obtained crude mixture with aqueous 1N HCl in the presence of 2,4-dinitrophenylhydrazine and subsequent deprotonation with sodium hydrogencarbonate furnished the free amine 4^{14} in an overall yield of 41% (scheme 3).



Scheme 3: a: Ac₂O, pyridine; b: 1N HCl, 2,4-dinitrophenylhydrazine, MeOH, CH₂Cl₂, r.t.; c: NaHCO₃

More detailed studies are required in order to improve the hydrolytic deprotection of the imines 3. However, the high efficiency and diastereoselectivity of the three carbon homologation on chiral N-glycosyl-Nhomoallylamines via an asymmetric cationic aza-Cope rearrangement should provide an interesting access to chain-extented amino sugars of varied substitution patterns by oxidative functionalization of the C=C double bond. Acknowledgement: This work was supported by the Deutsche Forschungsgemeinschaft. S. D. is grateful for a postdoctorate fellowship of the Alexander von Humboldt Foundation.

References and Notes

- 1. a) F. M. Unger, Adv. Carbohydr. Chem. Biochem. 1981, 38, 323; b) A. Esswein, R. Betz, R. R. Schmidt, Helv. Chim. Acta 1989, 72, 213; c) A. Dondoni, G. Fantin, M. Fogagnolo, P. Merino, Tetrahedron Lett. 1990, 31, 4513; d) S. F. Martin, P. W. Zinke, J. Org. Chem. 1991, 56, 6600.
- 2. N. J. Philips, R. McLaughlin, T. J. Müller, M. A. Apicella, B. W. Gibson, Biochemistry 1996, 35, 5937.
- 3. a) R. Schauer, A. K. Shukla, C. Schröder, E. Müller, Pure Appl. Chem. 1984, 56, 907; b) M. P. DeNinno, Synthesis 1991, 583.
- 4. A. Zamojski, S. Jarosz, Pol. J. Chem. 1992, 66, 525-585, and references cited therein.
- 5. a)B. Aebischer, J. H. Bieri, R. Prewo, A. Vasella, Helv. Chim. Acta 1982, 65, 2251; b) J. M. Beau, P. Sinaÿ, Tetrahedron Lett. 1985, 26, 6185. c) B. Giese, H.-G. Zeitz, in Preparative Carbohydrate Chemistry (S. Hanessian, Ed.), Marcel Dekker, New York, 1997, S. 507. d) K. Suzuki, T. Matsumoto ibid., S. 527; e) M. Yokoyama, H. Toyoshima, M. Shimizu, H. Togo, J. Chem. Soc. Perkin Trans I 1997, 29.
- 6. M. von Itzstein, W.-Y. Wu, G. B. Kok, M. S. Pegg, J. C. Dyason, B. Jin, T. V. Phan. M. L. Smythe, H. F. White, S. W. Oliver, P. M. Colman, J. N. Varhese, D. M. Ryan, J. M. Woods, R. C. Bethell, V. J. Hothman, J. M. Cameron, Nature 1993, 363, 418.
- 7. R. Kuhn, W. Kirschenlohr, Liebigs Ann. Chem. 1956, 600, 115.
- L. Cipolla, L. Lay, F. Nicotra, C. Pangrazio, L. Panza, *Tetrahedron* 1995, 51, 4679.
 a) S. Laschat, H. Kunz, *Synlett* 1990, 51; b) S. Laschat, H. Kunz, *ibid*. 1990, 629; c) S. Laschat, H. Kunz, J. Org. Chem. 1991, 56, 5883.
- 10. 3a: M.p. 97-98°C; $[a]_D^{22} = -20.3$ (c = 0.95, CDCl₃); ¹H NMR (400 MHz, CDCl₃): d = 2.40 (m, 2H, 2H-3), 3.13 (dt, ${}^{3}J$ = 8.0 Hz, ${}^{3}J$ = 4.0 Hz, 1H, H-4), 3.71 (t, ${}^{3}J$ = 6.0 Hz, 1H, H-8), 3.88 (dd, ${}^{2}J$ = 11.3 Hz, ${}^{3}J = 5.8$ Hz, 1H, H-9a), 3.99 (dd, ${}^{2}J = 11.3$ Hz, ${}^{3}J = 7.5$ Hz, 1H, H-9b), 5.01 (m, 3H, 2H-1, H-7), 5.30 (dd, ${}^{3}J$ = 8.6 Hz, ${}^{3}J$ = 1.4 Hz, 1H, H-5), 5.62 (m, 1H, H-2), 5.64 (dd, ${}^{3}J$ = 7.5 Hz, ${}^{3}J$ = 1.8 Hz, 1H, H-6), 6.87 (m, 2H, H-1', H-2'), 7.78 (d, 1H, ${}^{3}J$ = 7.5 Hz, H-3'); ${}^{13}C$ NMR (100.6 MHz, ${}^{1}H$ -¹³C-COSY, CDCl₃): d = 36.12 (C-3), 64.18 (C-9), 67.85 (C-8), 68.14 (C-7), 68.70 (C-6), 71.03 (C-5), 71.53 (C-4), 118.02 (C-1), 129.28 (C-1'), 134.06 (C-2), 142.47 (C-2'), 163.74 (C-3'). 11. Monoclinic, space group P2₁ with a = 12.349(1)Å, b = 8.589(3)Å, c = 19.982(1)Å and b = 103.839(6).
- Crystal dimensions : 0.064 x 0.128 x 0.800 mm³. Density amounts to 1.084gcm⁻³ with a volume of 2057.8(7)Å³, z = 2. $2q_{max} = 150^{\circ}$. Cu-Ka Graphitmonochromator. Wavelength : 1.54061. Scan mode : w/2q. Temp. : 296 K. 8962 measured and independent reflections, 8309 included in the refinement. All reflections are used in refinement. Lorentzian polarization corrections applied. No absorption corrections were performed. Direct methods of structure solution with SIR92. Method of refinement: full least squares with SHELXL-93. 459 parameters. Except hydrogens of pivaloyl groups all atoms found in difference Fourier maps, isotropic refinement assumed a riding motion model. Refined against F². Residual electron density : max. 0.16, min. -0.11eÅ⁻³. Flack parameter : -0.06(20). Further details are available on request from the Fachinformationszentrum Karlsruhe GmbH, D-76344 Eggenstein-Leopoldshafen (Germany), on quoting the deposit number CSD-406759, the names of the authors and journal citation. We thank Dr. D. Schollmeyer, Institut für Organische Chemie, Universität Mainz, for the X-ray analysis and calculations.
- 12. a) E. J. Jacobsen, J. Levin, L. E. Overman, J. Am. Chem. Soc. 1988, 110, 4329; b) S. D. Knight, L. E. Overman, G. Pairaudeau, J. Am. Chem. Soc. 1995, 117, 5776; c) C. Agami, F. Couty, J. Lin, A. Mikaeloff, M. Poursoulis, Tetrahedron 1993, 49, 7239.
- 13. [3,3]-Sigmatropic rearrangements accelerated by charged atoms: R. P. Lutz, Chem. Rev. 1984, 84, 205.
- 14. 4: $[a]_D^{22} = +51.3$ (c = 0.61, CHCl₃); ¹H NMR (400 MHz, CDCl₃): d = 2.03 (s, 3H, CH₃CO), 2.19 $(dd, {}^{2}J = 14.0 \text{ Hz}, {}^{3}J = 7.2 \text{ Hz}, 1\text{H}, \text{H-3a}), 2.33 \text{ (m, } {}^{2}J = 14.0 \text{ Hz}, 1\text{H}, \text{H-3b}), 3.68 \text{ (m, } 1\text{H}, \text{H-4}), 3.86 \text{ (m, } 1\text{H}, \text{H-4}), 3.8$ $(dd, ^{2}J = 11.3 Hz, ^{3}J = 7.8 Hz, 1H, H-9a), 4.06 (dd, ^{2}J = 11.3 Hz, ^{3}J = 5.8 Hz, 1H, H-9b), 4.15 (dd, ^{2}J = 11.3 Hz, ^{3}J = 5.8 Hz, 1H, H-9b), 4.15 (dd, ^{3}J = 5.8 Hz, 1H, H-9b),$ ${}^{3}J = 5.5$ Hz, ${}^{3}J = 1.4$ Hz, 1H, H-5), 5.10 (dd, ${}^{3}J = 9.6$ Hz, ${}^{3}J = 1.4$ Hz, 1H, H-6), 5.18 (dt, ${}^{3}J = 7.5$ Hz, ${}^{3}J = 1.7$ Hz, 1H, H-8), 5.43 (dd, ${}^{3}J = 9.6$ Hz, ${}^{3}J = 2.0$ Hz, 1H, H-7); ${}^{13}C$ NMR (100.6 MHz, $CDCl_3$): d = 20.71 (CH₃CO), 39.41 (C-3), 61.81 (C-9), 66.18, 68.18, 68.33, 69.47 (C-5, C-6, C-7, C-8), 80.09 (C-4).