



A common and stereoselective strategy for the synthesis of *N,O,O,O-tetra-acetyl D-ribo-(2S,3S,4R)-phytosphingosine and 2-epi-jaspine B*

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ARTICLE INFO

Article history:

Received 10 June 2011

Revised 5 July 2011

Accepted 9 July 2011

Available online 22 July 2011

ABSTRACT

A common and stereoselective strategy for the synthesis of *N,O,O,O-tetra-acetyl D-ribo-(2S,3S,4R)-phytosphingosine and 2-epi-jaspine B* was achieved by using Grignard addition on *N*-benzyl sugar lactamine and Wittig olefination as key steps.

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Sphingoids are long-chain amino-diol and -triol bases that form the backbone and the characteristic structural unit of sphingolipids, which are important membrane constituents and play a vital role in cell regulation as well as signal transduction.¹ Furthermore, glycosphingolipids show important biological activities, such as antitumor,^{2a} antiviral,^{2b} antifungal,^{2c} and cytotoxic properties.^{2d} Phytosphingosines, one of the major classes of sphingoids, have been isolated and identified either separately or as parts of sphingolipids from plants,^{3a} marine organisms,^{3b,c} fungi,^{3d} yeasts^{3e} and even mammalian tissues^{3f–k} of kidney,^{3g} liver,^{3h} uterus,³ⁱ intestine,^{3j} and skin.^{3k} In addition to being base components of sphingolipids in membranes, phytosphingosines themselves are found to be bioactive lipids. For example, *ribo*-phytosphingosine **1** is a potential heat stress signal in yeast cells.⁴ It has also been established that various diastereomers of sphingosines exhibit different activities and metabolism.⁵ Some of its derivatives exhibit important physiological activities.⁶ For instance KRN 7000 (Fig. 1) which is a glucosphingolipid exhibits natural killer activity and strongly inhibits tumor metastasis in mice.^{6b} Since it is costly to isolate these lipids from the natural sources, and also not possible to obtain homogeneous material for biological studies, a great work has been carried out for the synthesis of these compounds. Among all the eight possible stereoisomers of phytosphingosine (2S,3S,4R)-2-aminoctadecane-1,3,4-triol **1** with *D*-ribo-stereochemistry is the most common one. Owing to their interesting biological activities many synthetic approaches to enantiomerically pure *D*-ribo-phytosphingosine **1** have been reported in the literature.⁷

Jaspine B, also known as pachastrissamine (**2a**, Fig. 1), is a cyclic anhydro-phytosphingosine derivative, which was initially isolated from a marine sponge *Pachastrissa* sp. (family Calthropellidae) in 2002 by Higa and co-workers,⁸ and found to have cytotoxicity at a level of IC₅₀ 0.01 µg/mL against P388, A549, HT29, and MEL28

cell lines. Later, Debitus and co-workers⁹ isolated the same compound from a Vanuatu marine sponge genus *Jaspis* and named it as jaspine B **2a**. This compound displayed remarkable bioactivity (IC₅₀ 0.24 µM) against the A549 human lung carcinoma cell line using the ATPlite assay and represented as the most potent anti-cancer agent on this cell line. Delgado and co-workers reported that the potency of cytotoxicity is dependent on the stereochemistry of the tetrahydropuran moiety.¹⁰

The novel structural features and interesting biological activity of pachastrissamine have prompted chemists to develop several syntheses for jaspine B **2a** and its isomers.¹¹ Our group published the first total synthesis of jaspine B **2a** and its C2 epimer **2b**.^{12a} We also published the synthesis of the C2 and C3 epimer **2c** and an enantiomer of jaspine B **2d** and their biological activity in comparison to jaspine B **2a**^{12b} and 3-*epi* jaspine **2e**.^{12c}

In continuation of our efforts in this area, we envisaged a concise, efficient, and common method for the synthesis of *N,O,O,O-tetra-acetyl D-ribo-(2S,3S,4R)-phytosphingosine* **3** and 2-*epi*-jaspine **2b**. The starting material for our approach is *D*-ribose, which is an inexpensive sugar and possesses desired chirality. The retrosynthetic analysis of *N,O,O,O-tetra-acetyl D-ribo-(2S,3S,4R)-phytosphingosine* **3** and 2-*epi*-jaspine **2b** is depicted in Scheme 1. Compound **4** is a common intermediate for the synthesis of **3** and **2b** and it can be obtained from compound **5** after appropriate manipulations. The amino group can be introduced by nucleophilic addition on ribosylamine **6** and the required lipid chain can be introduced by Wittig olefination.

Earlier we explored stereoselective Grignard addition on chiral-imines for the synthesis of polyhydroxylated pyrrolidines¹³ and carbasugars.¹⁴ In this report we expanded the utility of this sugarimine Grignard addition, for the synthesis of *N,O,O,O-tetra-acetyl D-ribo-phytosphingosine* **3** and 2-*epi*-jaspine **2b**.

The starting material 5-*tert*-butyl dimethylsilyl-2,3-O-isopropylidene-*D*-ribofuranose **7**, required for our synthesis was prepared from *D*-ribose using our earlier procedure (Scheme 2).¹⁵ Reaction on **7** with benzylamine gave ribosylamine **6**. Treatment of the crude ribosylamine **6** with vinylmagnesium bromide at –78 °C

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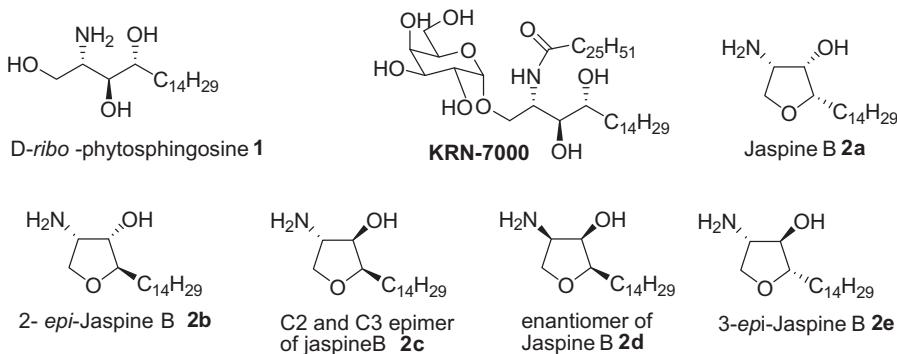
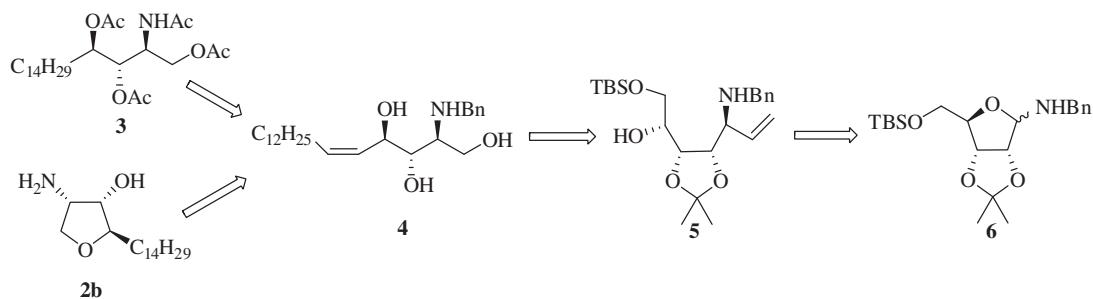
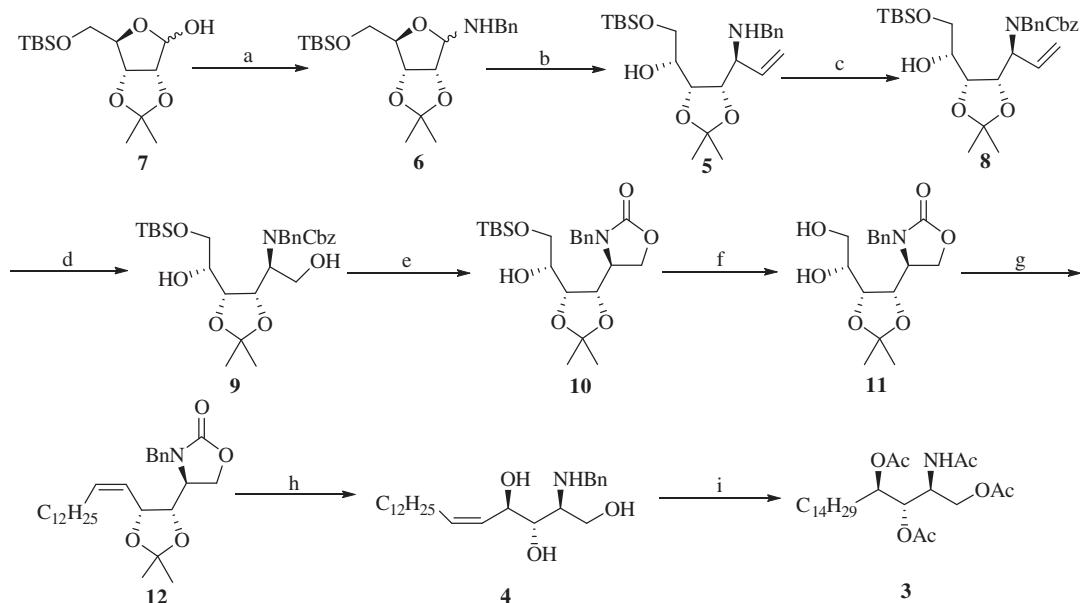


Figure 1. Structures of D-ribo-(2S,3S,4R)-phytosphingosine, KRN 7000, jaspine B, and its isomers.



Scheme 1. Retrosynthetic analysis of N,O,O,O-tetra-acetyl D-ribo-phytosphingosine **3** and 2-*epi*-jaspine B **2b**.



Scheme 2. Synthesis of N,O,O,O-tetra-acetyl D-ribo-phytosphingosine **3**. Reagents and conditions: (a) BnNH_2 , MeOH, reflux; (b) vinylmagnesium bromide, THF, -78°C , 2 h, 72%; (c) Cbz-Cl , NaHCO_3 , MeOH, 1 h, 98%; (d) (i) O_3 , CH_2Cl_2 , -78°C , 30 min and DMS, (ii) NaBH_4 , MeOH, 1 h (85% for two steps); (e) NaH , THF, 30 min, 92%; (f) TBAF, THF, 2 h, 96%; (g) (i) NaIO_4 , CH_2Cl_2 , H_2O , rt, 4 h, (ii) $\text{C}_{13}\text{H}_{27}\text{Ph}_3\text{P}^+\text{Br}^-$, $n\text{-BuLi}$, THF, 1 h (90% for two steps); (h) 6 N HCl , EtOH, reflux, 6 h, 94%; (i) (i) H_2 , Pd/C , MeOH, 6 h, (ii) $(\text{Ac})_2\text{O}$, $\text{N}(\text{Et})_3$, CH_2Cl_2 , 6 h (85% for two steps).

gave exclusively **5** as a single isomer. The absolute configuration of the newly generated stereo center was not confirmed at this stage, but as per our earlier observation it should give the *erythro*-isomer.¹⁴ The formation of *erythro*-isomer can be explained via seven membered transition state **A**¹⁶ or Felkin-Anh model **B** (Fig. 2). In fact the formation of *erythro*-isomer shows that chelation between sugarimine and isopropylidene group has not taken place due to steric strain.

The amino functionality in **5** was converted into carbamate with CbzCl to give **8**. Oxidative degradation of the alkene functionality in **8** with ozone afforded the aldehyde, which on treatment with NaBH_4 in MeOH gave alcohol **9**. In order to protect the primary hydroxyl group, compound **9** was treated with sodium hydride to give the cyclic carbamate **10**. Deprotection of silyl ether in **10** with TBAF gave diol **11**. The diol functionality of compound **11** was oxidatively cleaved to aldehyde using NaIO_4 in CH_2Cl_2 /

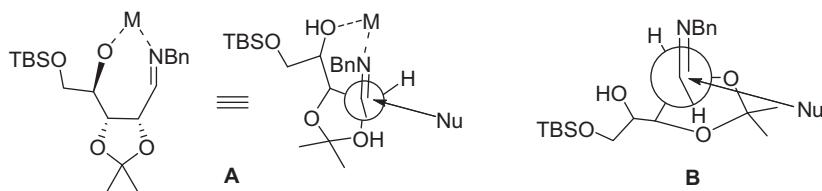
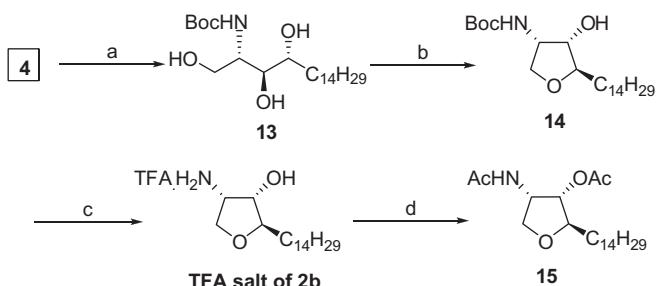


Figure 2. Seven-membered transition state (A) and Felkin Anh Model (B).

Scheme 3. Synthesis of TFA salt of 2-epi-jaspine B 2b. Reagents and conditions: (a) (i) H_2 , Pd/C, MeOH, 6 h; (ii) $(Boc)_2O$, Et₃N, CH₂Cl₂, 2 h (95% over two steps); (b) (i) TsCl, Et₃N, DMAP, CH₂Cl₂, 86%; (c) TFA, CH₂Cl₂, 2 h; 89%; (d) $(Ac)_2O$, Et₃N, CH₂Cl₂, 4 h, 94%.

H_2O , which was carried to the next step without any purification. Wittig reaction on crude aldehyde with $C_{13}H_{27}PPh_3^+$ gave **12**. Global deprotection of carbamate and acetonide in **12** with 6 N HCl in EtOH under reflux conditions gave **4**. Hydrogenation of **4** gave the D -ribo-phytosphingosine **1**. For the sake of characterization, it was converted as such into *N,O,O,O-tetra-acetyl D-ribo-phytosphingosine* **3**,¹⁷ by treating with acetic anhydride and Et₃N, whose physical properties are in good agreement with the reported values.¹⁸

To convert the compound **4** into 2-epi-jaspine **2b** the following reactions were carried out (Scheme 3). Hydrogenation of compound **4** using Pd/C in MeOH followed by treatment of the resultant amino functionality with $(Boc)_2O$ afforded carbamate **13**. Regioselective tosylation of the primary hydroxy group of **13** prompted spontaneous cyclization to give the tetrahydrofuran derivative **14**.^{11b} Deprotection of the Boc group in **14** with TFA/CH₂Cl₂ provided the desired 2-epi-jaspine **2b** as TFA salt^{19a} and TFA salt of **2b** on acetylation with acetic anhydride in the presence of excess Et₃N gave the acetyl derivative **15**.^{19b} The physical properties of the TFA salt of 2-epi-jaspine **2b**^{13a} and its diacetate derivative **15**^{12b} were in good agreement with the reported values.

In conclusion we have successfully demonstrated a general strategy for the synthesis of *N,O,O,O-tetra-acetyl D-ribo-phytosphingosine* **3** and 2-epi-jaspine **2b** and its diacetate derivative **15** by using stereoselective vinyl Grignard on ribosylamine. Application of this strategy to make some more analogs of phytosphingosine and jaspine B are in progress.

Acknowledgments

G.S.R. thanks CSIR, New Delhi for research fellowship. The authors also thank Dr. J. S. Yadav for his constant support and encouragement. We also thank DST (SR/S1/OC-14/2007), New Delhi, for financial assistance.

Supplementary data

Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.tetlet.2011.07.032.

References and notes

- Reviews: (a) Kolter, T.; Sandhoff, K. *Angew. Chem., Int. Ed.* **1999**, *38*, 1532; (b) Brodesser, S.; Sawatzki, P.; Kolter, T. *Eur. J. Org. Chem.* **2003**, *1*, 2021; (c) Kolter, T. Conformational Restriction of Sphingolipids. In *Highlights Bioorganic Chemistry Methods and Applications*; Schmuck, C., Wennemers, H., Eds.; Wiley-VCH: Weinheim, 2004; p 48. Chapter 1.4; (d) Liao, J.; Tao, J.; Lin, G.; Liu, D. *Tetrahedron* **2005**, *61*, 4715.
- (a) Naroti, T.; Morita, M.; Akimoto, K.; Koezuka, Y. *Tetrahedron* **1994**, *50*, 2771; (b) Kamitakahara, H.; Suzuki, T.; Nishigori, N.; Suzuki, Y.; Kamiie, O.; Wong, C. H. *Angew. Chem., Int. Ed.* **1998**, *37*, 1524; (c) Kobayashi, S.; Furuta, T.; Hayashi, T.; Nishijima; Anada, K. *J. Am. Chem. Soc.* **1998**, *120*, 908; (d) Matsunaga, H.; Li, S.; Fusetani, N. *Tetrahedron* **1995**, *51*, 2273.
- (a) Carter, H. E.; Clemer, W. D.; Lands, W. M.; Muller, K. L.; Tomizawa, H. H. *J. Biol. Chem.* **1954**, *206*, 613; (b) Kawano, Y.; Higushi, R.; Isobe, R.; Komori, T. *Liebigs Ann. Chem.* **1988**, *19*; (c) Li, Y. T.; Hirabayashi, Y.; DeGasperi, R.; Yu, R. K.; Ariga, T.; Koerner, T. A. W.; Li, S. C. *J. Biol. Chem.* **1984**, *259*, 8980; (d) Oda, T. *J. Pharm. Soc. Jpn.* **1952**, *72*, 142; (e) Thorpe, S. R.; Sweeley, C. *Biochemistry* **1967**, *6*, 887; (f) Karlsson, K. A.; Samuelsson, B. E.; Steen, G. O. *Acta Chem. Scand.* **1968**, *22*, 1361; (g) Barenholz, Y.; Gatt, S. *Biochem. Biophys. Res. Commun.* **1967**, *27*, 319; (h) Takamatsu, K.; Mikami, M.; Kiguschi, K.; Nozawa, S.; Iwamori, M. *Biochem. Biophys. Acta* **1992**, *1165*, 177; (i) Okabe, K.; Keenan, R. W.; Schmidt, G. *Biochem. Biophys. Res. Commun.* **1968**, *31*, 137; (j) Wertz, P. W.; Miethke, M. C.; Long, S. A.; Stauss, J. S.; Downing, D. T. *J. Invest. Dermatol.* **1985**, *84*, 410; (k) Vance, D. E.; Sweeley, C. C. *Lipid Res.* **1967**, *8*, 621.
- (a) Dickson, R. C.; Nagiec, E. E.; Skrzypek, M.; Tillman, P.; Wells, G. B.; Lester, R. L. *J. Biol. Chem.* **1997**, *272*, 30196; (b) Schneiter, R. *Bioessays* **1999**, *21*, 1004.
- (a) Brodesser, S.; Sawatzki, P.; Kolter, T. *Eur. J. Org. Chem.* **2003**, *2021*; (b) Vankar, Y. D.; Schmidt, R. R. *Chem. Soc. Rev.* **2000**, *29*, 201.
- (a) Kobayashi, E.; Motoki, K.; Yamaguchi, Y.; Uchida, T.; Fukushima, H.; Koezuka, Y. *Bioorg. Med. Chem.* **1996**, *4*, 615; (b) Kobayashi, E.; Motoki, K.; Uchida, T.; Fukushima, H.; Koezuka, Y. *Oncol. Res.* **1995**, *7*, 529.
- For a review of PHS synthesis prior to 2002, see: (a) Howell, A. R.; Ndakala, A. J. *Curr. Org. Chem.* **2002**, *6*, 365; For additional references on subsequent synthesis of **1**, see: (b) Lu, X.; Byun, H. S.; Bittman, R. J. *Org. Chem.* **2004**, *69*, 5433; (c) Singh, O. V.; Kampf, D. J.; Han, H. *Tetrahedron Lett.* **2004**, *45*, 7239; (d) Lu, X.; Bittman, R. *Tetrahedron Lett.* **2005**, *46*, 3165; (e) Lombardo, M.; Capdevila, M. G.; Pasi, F.; Trombini, C. *Org. Lett.* **2006**, *8*, 3303; (f) Jeon, J.; Shin, M.; Yoo, J. W.; Oh, J. S.; Bae, J. G.; Jung, S. H.; Kim, Y. G. *Tetrahedron Lett.* **2007**, *48*, 1105; (g) Chang, C. W.; Chen, Y. N.; Adak, A. K.; Lin, K. H.; Tzou, D. L. M.; Lin, C. *Tetrahedron* **2007**, *63*, 4310; (h) Abraham, E.; Candela-Lena, J. I.; Davies, S. G.; Georgiou, M.; Nicholson, R. L.; Roberts, P. M.; Russell, A. J.; Sanchez-Fernandez, E. M.; Smith, A. D.; Thomson, J. E. *Tetrahedron: Asymmetry* **2007**, *18*, 2510; (i) Kim, S.; Lee, N.; Lee, S.; Lee, T.; Lee, Y. M. *J. Org. Chem.* **2008**, *73*, 1379; (j) Abraham, E.; Brock, E. A.; Candela-Lena, J. I.; Davies, S. G.; Georgiou, M.; Nicholson, R. L.; Perkins, J. H.; Roberts, P. M.; Russell, A. J.; Sanchez-Fernandez, E. M.; Scott, P. M.; Smith, A. D.; Thomson, J. E. *Org. Biomol. Chem.* **2008**, *6*, 1665; (k) Llaveria, J.; Diaz, Y.; Matheu, M. I.; Castillon, S. *Org. Lett.* **2009**, *11*, 205; (l) Liu, Z.; Byun, H. S.; Bittman, R. J. *Org. Chem.* **2010**, *75*, 4356.
- Kuroda, I.; Musuman, M.; Ohtani, I. I.; Ichiba, T.; Tanaka, J.; Gravalos, D. G.; Higa, T. *J. Nat. Prod.* **2002**, *65*, 1505.
- Ledroit, V.; Debitus, C.; Lavaud, C.; Massiot, G. *Tetrahedron Lett.* **2003**, *44*, 225.
- Canals, D.; Mormeneo, D.; Fabrias, G.; Llebaria, A.; Casas, J.; Delgado, A. *Bioorg. Med. Chem.* **2009**, *17*, 235.
- For previous syntheses, see: (a) Bhaket, P.; Morris, K.; Stauffer, C. S.; Datta, A. *Org. Lett.* **2005**, *7*, 875; (b) Van Den Berg, R. J. B. H. N.; Boltje, T. J.; Verhagen, C. P.; Litjens, R. E. J. N.; Vander Marel, G. A.; Overkleef, H. S. J. *Org. Chem.* **2006**, *71*, 836; (c) Liu, Y.; Du, J.; Linhardt, R. J. *J. Org. Chem.* **2006**, *71*, 1251; (d) Liu, J.; Du, Y.; Dong, X.; Meng, S.; Xiao, J.; Cheng, L. *Carbohydr. Res.* **2006**, *341*, 2653; (e) Ribes, C.; Falomir, E.; Carda, M.; Marco, J. A. *Tetrahedron* **2006**, *62*, 5421; (f) Lee, T.; Lee, S.; Kwak, Y. S.; Kim, D.; Kim, S. *Org. Lett.* **2007**, *9*, 429; (g) Reddy, L. V. R.; Reddy, P. V.; Shaw, A. K. *Tetrahedron: Asymmetry* **2007**, *18*, 542; (h) Ramana, C. V.; Giri, A. G.; Suryawanshi, S. B.; Gonnade, R. G. *Tetrahedron Lett.* **2007**, *48*, 265; (i) Prasad, K. R.; Chandrakumar, A. *J. Org. Chem.* **2007**, *72*, 6312; (j) Yakura, T.; Sato, S.; Yoshimoto, Y. *Chem. Pharm. Bull.* **2007**, *55*, 1284; (k) Passiniemi, M.; Koskinen, A. M. P. *Tetrahedron Lett.* **2008**, *49*, 980; (l) Venkatesan, K.; Srinivasan, K. V. *Tetrahedron: Asymmetry* **2008**, *19*, 209; (m) Enders, D.; Terteryan, V.; Palecek, J. *Synthesis* **2008**, 2278; (n) Ichikawa, Y.; Matsunaga, K.; Masuda, T.; Kotsuki, H.; Nakano, K. *Tetrahedron* **2008**, *64*, 11313; (o) Reddipalli, G. S.; Venkataiah, M.; Mishra, M. K.; Fadnavis, N. W. *Tetrahedron: Asymmetry* **2009**, *20*, 1802; (p) Inukai, S.; Yoshimitsu, Y.; Oishi, S.; Fujii, N.; Ohno, H. *Org. Lett.* **2009**, *11*, 4478; (q) Inukai, S.; Yoshimitsu, Y.; Oishi, S.; Fujii, N.; Ohno, H. *J. Org.*

- Chem.* **2010**, *75*, 3831; (r) Yoshimitsu, Y.; Inuki, S.; Oishi, S.; Fujii, N.; Ohno, H. *J. Org. Chem.* **2010**, *75*, 3843; (s) Urano, H.; Enomoto, M.; Kuwahara, S. *Biosci., Biotechnol., Biochem.* **2010**, *74*, 152; (t) Salma, Y.; Ballereau, S.; Maaliki, C.; Ladeira, S.; Andrieu-Abadie, N.; Genison, Y. *Org. Biomol. Chem.* **2010**, *8*, 3227; (u) Passiniemi, M.; Koskinen, A. M. P. *Org. Biomol. Chem.* **2011**, *9*, 1774; (v) Llaveria, J.; Diaz, Y.; Matheu, M. I.; Castillon, S. *Eur. J. Org. Chem.* **2011**, *1514*; While preparing our manuscript the following publications appeared (w) Eleuterio, A. S.; Quintero, L.; Piscil, F. S. *J. Org. Chem.* **2011**, *76*, 5466; For a review, see: (x) Abraham, E.; Davies, S. G.; Roberts, P. M.; Russell, A. J.; Thomson, J. E. *Tetrahedron: Asymmetry* **2008**, *19*, 1027.
12. (a) Sudhakar, N.; Kumar, A. R.; Prabhakar, A.; Jagadeesh, B.; Rao, B. V. *Tetrahedron Lett.* **2005**, *46*, 325; (b) Chitra, C. J.; Sudhakar, N.; Kumar, A. R.; Rao, B. V.; Roy, S.; Benerjee, R. *Synthesis* **2010**, *115*; (c) Rao, G. S.; Sudhakar, N.; Rao, B. V.; Basha, S. J. *Tetrahedron: Asymmetry* **1963**, *2010*, *21*.
13. (a) Jagadeesh, Y.; Chandrasekhar, B.; Rao, B. V. *Tetrahedron: Asymmetry* **2010**, *21*, 2314; (b) Chandrasekhar, B.; Rao, B. V. *Tetrahedron: Asymmetry* **2009**, *20*, 1217; (c) Chandrasekhar, B.; Madhan, A.; Rao, B. V. *Tetrahedron* **2007**, *63*, 8746; (d) Madhan, A.; Rao, B. V. *Tetrahedron Lett.* **2003**, *44*, 5641.
14. Rao, J. P.; Rao, B. V.; Swarnalatha, J. L. *Tetrahedron Lett.* **2010**, *51*, 3083.
15. Kumar, D. N.; Rao, B. V.; Ramjaneyulu, G. S. *Tetrahedron: Asymmetry* **2005**, *6*, 1611.
16. Mekki, B.; Singh, G.; Wightman, R. H. *Tetrahedron Lett.* **1991**, *32*, 5143.
17. Spectral data of compound **3**: $[\alpha]_D^{26} + 23.1$ (*c* 0.62, CHCl_3); (lit. ¹⁸ $[\alpha]_D^{20} + 21.9$ (*c* 1.1, CHCl_3)); IR ν_{max} : 2924, 2854, 1741, 1659, 1543, 1370, 1221, 1044 cm^{-1} ; ¹H NMR (CDCl_3 , 300 MHz): δ 6.00 (d, 1H, *J* = 9.4 Hz), 5.04 (dd, 1H, *J* = 3.0 and 8.3 Hz), 4.83–4.91 (m, 1H), 4.40 (m, 1H), 4.22 (dd, 1H, *J* = 4.7 and 11.5 Hz), 3.94 (dd, 1H, *J* = 3.0 and 11.5 Hz), 2.01 (s, 3H), 1.99 (s, 6H), 1.96 (s, 3H), 1.76 (m, 1H), 1.51 (m, 1H), 1.05–1.31 (m, 24H), 0.80 (t, 3H, *J* = 6.6 Hz); ¹³C NMR (CDCl_3 , 75 MHz): δ 171.1, 170.8, 170.0, 169.7, 72.9, 71.9, 62.8, 47.6, 31.9, 29.6, 29.5, 29.4, 29.3, 29.3, 28.1, 25.5, 23.3, 22.6, 21.0, 20.7, 14.1; ESIMS *m/z*: 486 [M+H]⁺; HRMS (ESI): calcd for $\text{C}_{26}\text{H}_{48}\text{NO}_7$ [M+H]⁺ 486.3430, found 486.3441.
18. Shiota, O.; Nakanishi, K.; Berova, N. *Tetrahedron* **1999**, *55*, 13643.
19. (a) Spectral data of compound of **2a** as TFA salt: $[\alpha]_D^{26} + 16.2$ (*c* 1.1, EtOH); (lit. ^{12a} $[\alpha]_D^{25} + 14.46$ (*c* 1.5, EtOH)); IR ν_{max} : 2920, 2851, 1669, 1209, 1147 cm^{-1} ; ¹H NMR (CD_3OD , 300 MHz): δ 4.14 (m, 1H), 4.03 (t, 3H, *J* = 5.7 Hz), 3.65–3.80 (m, 3H), 1.17–1.70 (m, 26H), 0.89 (t, 3H, *J* = 6.8 Hz); ¹³C NMR (CD_3OD , 75 MHz): δ 85.3, 74.4, 69.4, 53.7, 47.9, 34.1, 33.1, 30.8, 30.7, 30.6, 30.5, 26.9, 23.7, 14.4; ESIMS *m/z*: 300 (M⁺–CF₃COOH); HRMS (ESI): calcd for $\text{C}_{18}\text{H}_{38}\text{NO}_2$ [M+H]⁺ 300.2902, found 300.2902.
- (b) Spectral data of compound **15**: $[\alpha]_D^{26} - 15.1$ (*c* 1.2, CHCl_3); (lit. ^{11b} $[\alpha]_D^{22} - 15.4$ (*c* 1.0, CHCl_3)); IR ν_{max} : 3298, 2920, 2851, 1740, 1658, 1557, 1376, 1232, 1072 cm^{-1} ; ¹H NMR (CDCl_3 , 300 MHz): δ 5.68 (br d, 1H, *J* = 7.7 Hz), 4.66 (m, 1H), 4.18 (t, 1H, *J* = 7.7 Hz), 3.84–3.90 (m, 1H), 3.52 (t, 1H, *J* = 8.5 Hz), 2.14 (s, 3H), 2.02 (s, 3H), 1.49–1.68 (m, 2H), 1.15–1.47 (m, 26H), 0.88 ((t, 1H, *J* = 6.6 Hz); ¹³C NMR (CDCl_3 , 75 MHz): δ 170.1, 169.9, 84.0, 76.7, 69.7, 49.9, 33.5, 31.9, 29.6, 29.5, 29.5, 29.4, 29.3, 25.5, 23.2, 22.7, 21.0, 14.1; ESIMS *m/z*: 406 (M+Na)⁺; HRMS (ESI): calcd for $\text{C}_{22}\text{H}_{41}\text{NO}_4\text{Na}$ [M+Na]⁺ 406.2933, found 406.2933.