The Stereoselective Synthesis of Alkenyl-β-lactams by Palladium-Catalyzed [2+2] Carbonylative Cycloaddition

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Allyl halides of different structures, under CO pressure, in the presence of Et_3N , a catalytic amount of $Pd(OAc)_2$, and triphenylphosphane as ligand, undergo a [2+2] cycloaddition reaction with various imines. The reaction is highly regioand stereoselective: β -lactams are formed in good yields and

Introduction

Among the catalyzed carbonylation reactions, the cyclocarbonylation catalyzed by transition metals has recently proved to be an extremely helpful technique for the production of interesting transformations in organic chemistry.^[1–3] In particular, a large number of papers has appeared in which palladium complexes have been used to prepare many new and useful molecules. Particularly interesting seems to be the cyclocarbonylation of unsaturated alcohols,^[4–12] which leads to the formation of lactones.

The analogous cyclocarbonylation of unsaturated amines, leading to the formation of lactams, seems to have been studied less in the literature.^[13,14] Of all lactams, the β -lactams^[15-22] are of special biological and pharmaceutical interest, and these compounds have also been obtained by different synthetic pathways, summarized below:

a) Carbonylation of the aziridines **1** in the presence of a catalytic amount of $[Rh(CO)_2Cl]_2$.^[23–28] The carbonyl insertion is regio- and stereospecific, occurring at the most substituted carbon-nitrogen bond in the aziridine ring, and proceeding with retention of stereochemistry of the substituents linked to the aziridinic carbon atoms (Scheme 1).



Scheme 1

organic bapers has en used to $Pd(OAc)_2/PPh_3$ O R

lactams^[29-31] (Scheme 2).

Scheme 2

Germany, 2004)

c) Carbonylative coupling and cyclization reaction of the 1,3 thiazines **3** with allyl phosphates, catalyzed by bis(benzonitrile)palladium dichloride.^[32] Several rhodium complexes are also effective catalysts for this process (Scheme 3).

with *trans* diastereoselectivity in both the β -lactam ring and

the vinylic moiety. New and important information is sug-

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b) Palladium-catalyzed carbonylation of the aminovinyl bromide **2**. This is a convenient synthesis of vinyl β -

gested regarding the known reaction mechanism.



Scheme 3

d) Palladium-catalyzed carbonylation of the allyl phosphate **4** in the presence of imines, under CO pressure.^[33,34] The reaction is stereoselective since the formation of the *trans*-**7** or the *cis*-**8** β -lactam depends on the imine used for the coupling. An imine conjugated **6** with a carbonyl group gives the *cis*- β -lactam, whereas the unconjugated imine **5** gives the *trans* isomer (Scheme 4).

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Scheme 4

No reaction products, or just traces of β -lactams, were reportedly formed when allyl bromide,^[33-37] allylacetate,^[38,39] allyl phenyl ether,^[40] allyl carbonate^[41,42] or allylsulfone^[43] were used under similar reaction conditions. In contrast with these observations, we found that simple allyl halides of various structures react with imines under slightly different reaction conditions, leading to β lactams in a stereoselective way.

Results and Discussion

Several imines 9a-d, which were to be used as substrates (Scheme 5), were prepared as reported in the Exp. Sect.

Table 1. Cyclocarbonylation of imines 9a-d with allyl halides 10a-g

[2+2] Cycloaddition reactions of allyl halides of different structures 10a-1 with the simple imines 9a-d, under CO pressure, in the presence of Et₃N, triphenylphosphane and a catalytic amount of Pd(OAc)₂, led to β -lactams in good yields and with high diastereoselectivity (Table 1). Slightly different reaction conditions to those reported in the literature, such as the use of Et₃N instead of *i*Pr₂NEt, allyl halides as substrates instead of the less easily accessible allyl phosphates, and a lower pressure of CO, make this methodology much simpler than those reported in the literature.

The *trans* relative configuration was found to predominate both in the β -lactam ring and in the C–C double bond of the products. The stereochemistry of the C–C double bond in the products was assigned on the basis of the ${}^{3}J_{\rm H,H}$ coupling constants across the double bond. For the β -lactam moiety itself, the stereochemistry was assigned based on of the coupling constants between the protons linked to C-3 and C-4, according to the literature report that for small rings, $J_{cis} > J_{trans}$.^[44] The diastereoisomers **11n,11n'** and **110,110'** showed *trans*-type coupling constants $J_{\rm H,H}$ for the β -lactam ring, similar to those measured for the other compounds. The protons of the R substituent on the nitrogen atom showed a difference in chemical shift (δ) between compounds **11n** and **11n'**, and between **110** and **110'**. This suggests two possible configurations of the β -lactamic nitrogen.

According to the mechanism proposed by Torii^[33] (Scheme 6), the reaction starts by the formation of the π -allyl palladium complex **16**. In the acyl complex **17**, derived from the insertion of CO, the protons alpha to the carbonyl group show appreciable acidity.



Entry	Imine	Halide	Products (yield, %) ^[a]			
1	9a	10a	11a (84)	12a (5)	_	_
2	9a	10b	11a (81)	12a (6)	_	_
3	9a	10c	11b (81)	12b (15)	_	_
4	9a	10d	11c (58)	12c (traces)	13c (21)	14c (traces)
5	9a	10f	11d (50) ^[b]	-	13d (15) ^[b]	_
6	9a	10g	_	_	13e (6) ^[b]	14e (14) ^[b]
7	9b	10a	11f (85)	_	_	_ ``
8	9b	10c	11g (78)	_	-	_
9	9b	10d	11h (60) ^[b]	12h (traces)	13h (20) ^[b]	14h (traces)
10	9b	10e	11i (98)	- ,	-	- ` `
11	9b	10f	111 (50) ^[b]	_	13l (15) ^[b]	_
12	9b	10g	_	_	13m (4) ^[b]	14m (11) ^[b]
13	9c	10a	11n (60) + 11n' (32)	_	-	-
14	9d	10a	110(53) + 110'(40)	_	_	_

^[a] Isolated yields. ^[b] Inseparable mixture of diastereoisomers (ratio measured by GC and ¹H NMR spectroscopy).



Scheme 5



Scheme 6

Table 2. Cyclocarbonylation of imines 9a-b with allyl halides 10h-l





Entry 1	Imine 9a	Halide	Products (yield, %) ^[a]				
			11c (60)	12c (traces)	13c (20)	14c (traces)	
2	9b	10h	11h (60) ^[b]	12h (traces)	13h (25) ^[b]	14h (traces)	
3	9a	10i	11c (55)	12c (traces)	13c (23)	14c (traces)	
4	9b	10i	11h (65) ^[b]	12h (traces)	13h (23) ^[b]	14h (traces)	
5	9a	101	11d (52) ^[b]	_ `	13d (14) ^[b]	- ` `	
6	9b	10 l	111 (53) ^[b]	_	13l (15) ^[b]	—	

^[a] Isolated yields. ^[b] Inseparable mixture of diastereoisomers (ratio measured by GC and ¹H NMR spectroscopy).

Consequently, deprotonation by Et_3N could lead to the formation of a carbanion, or even a ketene,^[45] which, through an addition reaction with the imine, could form a four-membered ring **18**.

According to previously reported data,^[33,34] as well as the data in Table 1, the CO insertion would always occur between the carbon (C-3) directly attached to the halogen and the palladium (structure **17**). When the halogen is not in the terminal position of the allylic chain, we found, instead, that the insertion occurs at the less hindered carbon atom (C-1) of structure **15**. In particular, when the halosubstrate is 3-chloro-1-butene (**10h**) or 3-bromo-1-butene (**10i**) ($\mathbb{R}^2 = \mathbb{R}^3 = \mathbb{H}$, $\mathbb{R}^1 = \mathbb{CH}_3$), or 3-bromo-1-hexene (**10l**) [$\mathbb{R}^2 = \mathbb{R}^3 = \mathbb{H}$, $\mathbb{R}^1 = (\mathbb{CH}_2)_2\mathbb{CH}_3$], carbonylation occurs at the C-1 position, and not at C-3 (Table 2).

In fact, in these cases, we isolated the same reaction products as were obtained when crotyl bromide (10d) or 1bromo-2-hexene (10f) were used to form the intermediate π -allyl palladium complex (Entries 4, 5, 9 and 11, Table 1). The low yield obtained with 3-bromocyclohexene (10g) (Entries 6 and 12, Table 1) could be caused by steric effects. The formation of the π -allyl palladium complex, and the subsequent terminal double-bond shift, could not proceed easily for steric reasons, resulting in a considerably lower yield for this reaction.

Conclusion

This protocol can be used to synthesize a large number of new alkenyl- β -lactams stereoselectively. The main advantages of our method, compared to those reported in the literature, are that Et₃N may be used as base, and that simple allyl halides may be used as substrates instead of the less easily accessible allyl phosphates.

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Moreover, we have shown a new aspect of the suggested mechanism: the CO insertion can occur on either of the carbon atoms of the intermediate π -allyl palladium complex, depending on the steric environment.

Experimental Section

General Remarks: THF, allyl bromide, allyl chloride, crotyl bromide, 1-chloro-3-methyl-2-butene, cinnamyl chloride, 3-bromocyclohexene, 3-chloro-1-butene, (+/-)- α -methylbenzylamine, aniline, benzaldehyde, triethylamine, palladium(II) acetate, triphenylphosphane and all other chemicals were of commercial grade (Aldrich), and they were used without further purification. Imines were prepared starting from the corresponding carbonyl compounds and amines.^[46] The allylic halides 10f, 10i and 10l were prepared starting from commercially available 1-butene, 2-hexene and 1-hexene, respectively, by bromination with N-bromosuccinimide. Petroleum ether refers to the fraction boiling in the range 40-60 °C. ¹H and ¹³C NMR spectra were recorded with a Bruker Avance 400 spectrometer (400.13 MHz and 100.62 MHz, for ¹H and ¹³C, respectively), with CDCl₃ as the solvent ($\delta = 7.24$ ppm for ¹H spectra; $\delta = 77.0$ ppm for ¹³C spectra) with TMS as an internal standard. IR spectra were recorded with a Perkin-Elmer spectrometer Model 283. GC-MS analyses were performed with a Shimadzu-17A gas chromatograph (5% diphenyl/95% dimethylpolysiloxane capillary column, 30 m, 0.25 mm i.d.), equipped with a Shimadzu GCMS-QP5050A mass-selective detector operating at 70 eV (EI). Microanalyses were performed on a Carlo-Erba C,H,N analyser. Melting points were determined using an electrothermal melting point apparatus and are uncorrected. TLC was performed on Merck silica gel plates with F-254 indicator; viewing was by UV light (254 nm). Column chromatography was performed on silica gel (63-200 mm) using petroleum ether/diethyl ether (Et₂O) mixtures as eluents. All reactions involving air-sensitive reagents were performed under nitrogen in oven-dried glassware, using syringe/septum cap techniques.

General Procedure for the Preparation of β-Lactams: 9a-d (1.0 mmol), 10a-1 (1.5 mmol), PPh₃ (0.08 mmol), Pd(AcO)₂ (0.02 mmol), and Et₃N (2 mmol) were dissolved in THF (10 mL), and placed in a 45 mL autoclave. The autoclave was purged, pressurized (400 psi of CO), and then heated to 100 °C for 18 h. The reaction mixture was then cooled to room temperature, and worked up by the addition of water (5 mL) and extraction with Et₂O (3 × 5 mL). The combined organic layers were dried (Na₂SO₄) and concentrated in vacuo. The crude products were purified by column chromatography (silica gel, petroleum ether/Et₂O, 7:3) to afford the pure β-lactams; yields: 15-98%.

1,4-Diphenyl-3-vinylazetidin-2-one (*trans*-11a): From allylic halide **10a**, yield 209 mg (84%), solid. M.p. 97.2–99.2 °C (*n*-hexane). ¹H NMR (400.13 MHz, CDCl₃): δ = 3.74 (dd, J = 7.8, 2.4 Hz, 1 H), 4.81 (d, J = 2.4 Hz, 1 H), 5.35 (dd, J = 23.9, 17.2 Hz, 2 H), 6.00–6.10 (m, 1 H), 7.05 (t, J = 7.2 Hz, 1 H), 7.16–7.40 (m, 9 H) ppm. ¹³C NMR (100.62 MHz, CDCl₃): δ = 61.2, 64.0, 117.0, 119.9, 123.9, 125.8, 128.6, 129.1, 129.2, 130.5, 137.3, 138.1, 165.3 ppm. GC-MS (70 eV): *mlz* (%) = 249 (5) [M⁺], 181 (30), 180 (33), 130 (100), 129 (64), 115 (26), 77 (50). IR (CHCl₃): \tilde{v} = 3050 cm⁻¹, 3020, 3000, 2910, 1740, 1600, 1370. C₁₇H₁₅NO (249.31): calcd. C 81.90, H 6.06, N 5.62; found C 82.10, H 6.10, N 5.55. *cis*-12a: Yield 12 mg (5%), oil. ¹H NMR (400.13 MHz, CDCl₃): δ = 4.29 (dd, J = 6.5, 6.4 Hz, 1 H), 5.08–5.12 (m, 1 H), 5.25–5.39 (m, 3 H), 7.06 (t, J = 7.3 Hz, 1 H), 7.16–7.34 (m, 9 H) ppm. ¹³C NMR (100.62 MHz, CDCl₃): $\delta = 58.4$, 58.8, 117.2, 120.9, 123.9, 127.0, 128.3, 128.7, 128.8, 129.0, 134.7, 137.5, 165.4 ppm. GC-MS (70 eV): m/z (%) = 249 (5) [M⁺], 181 (30), 180 (33), 130 (100), 129 (64), 115 (26), 77 (50). IR (film): $\tilde{\nu} = 3050 \text{ cm}^{-1}$, 3020, 3000, 2910, 1740, 1600, 1370. (From allylic halide **10b**) *trans*-**11a**: Yield 202 mg (81%), *cis*-**12a**: 15 mg (6%).

3-(2-Methylpropenyl)-1,4-diphenylazetidin-2-one, (trans-11b): Yield 224 mg (81%), solid. M.p. 101.0-103.0 °C (n-hexane). ¹H NMR $(400.13 \text{ MHz}, \text{CDCl}_3)$: $\delta = 1.60$ (s, 3 H), 1.79 (s, 3 H), 3.88 (dd, J = 7.8, 2.4 Hz, 1 H), 4.70 (d, J = 2.4 Hz, 1 H), 5.40 (d, J =7.6 Hz, 1 H), 7.05 (t, J = 7.2 Hz, 1 H), 7.20–7.37 (m, 9 H) ppm. ¹³C NMR (100.62 MHz, CDCl₃): $\delta = 18.7, 25.7, 60.3, 62.3, 116.9,$ 117.1, 123.7, 125.8, 128.3, 128.9, 129.1, 137.7, 137.8, 139.1, 166.9 ppm. GC-MS (70 eV): m/z (%) = 277 (1) [M⁺], 180 (17), 158 (72), 143 (100), 128 (20), 115 (10), 77 (32). IR (CHCl₃): $\tilde{v} = 3030$ cm⁻¹, 2970, 2910, 1740, 1600, 1370. C₁₉H₁₉NO (277.37): calcd. C 82.28, H 6.90, N 5.05; found C 82.38, H 7.08, N 5.01. cis-(12b): Yield 42 mg (15%), oil. ¹H NMR (400.13 MHz, CDCl₃): $\delta = 1.53$ (s, 3 H), 1.66 (s, 3 H), 4.48 (dd, J = 9.0, 6.0 Hz, 1 H), 4.76 (d, J =9.0 Hz, 1 H), 5.25 (d, J = 6.0 Hz, 1 H), 7.04 (t, J = 7.3 Hz, 1 H), 7.20–7.36 (m, 9 H) ppm. ¹³C NMR (100.62 MHz, CDCl₃): δ = 22.7, 29.7, 54.3, 59.3, 114.6, 117.2, 123.7, 127.0, 128.0, 128.6, 129.0, 135.3, 137.7, 139.1, 167.1 ppm. GC-MS (70 eV): m/z (%) = 277 (1) [M⁺], 180 (17), 158 (72), 143 (100), 128 (20), 115 (10), 77 (32). IR (film): $\tilde{v} = 3030 \text{ cm}^{-1}$, 2970, 2910, 1740, 1600, 1370.

1,4-Diphenyl-3-propenylazetidin-2-one (trans, trans-11c): (From allylic halide 10d). Yield 153 mg (58%), solid. M.p. 74.0-76.0 °C (nhexane). ¹H NMR (400.13 MHz, CDCl₃): $\delta = 1.74$ (d, J = 6.0 Hz, 3 H), 3.66 (dd, J = 8.1, 2.3 Hz, 1 H), 4.74 (d, J = 2.3 Hz, 1 H), 5.62-5.84 (m, 2 H), 7.01 (t, J = 7.5 Hz, 1 H), 7.19-7.50 (m, 9 H) ppm. ¹³C NMR (100.62 MHz, CDCl₃): $\delta = 18.0, 61.6, 63.4, 116.8,$ 123.3, 123.7, 125.6, 128.4, 128.9, 129.0, 130.4, 131.2, 137.4, 166.0 ppm. GC-MS (70 eV): m/z (%) = 263 (4) [M⁺], 181 (35), 180 (55), 144 (89), 129 (100), 128 (62), 115 (25), 77 (83). IR (CHCl₃): $\tilde{v} = 3020 \text{ cm}^{-1}$, 2920, 1740, 1600, 1490, 1370. C₁₈H₁₇NO (263.34): calcd. C 82.10, H 6.51, N 5.32; found C 81.86, H 6.42, N 5.15. trans, cis-13c: Yield 55 mg (15%), oil. ¹H NMR (400.13 MHz, CDCl₃): $\delta = 1.59$ (dd, J = 6.8, 1.1 Hz, 3 H), 3.99 (dd, J = 9.0, 2.3 Hz, 1 H), 4.74 (d, J = 2.3 Hz, 1 H), 5.62–5.84 (m, 2 H), 7.01 $(t, J = 7.5 \text{ Hz}, 1 \text{ H}), 7.19-7.50 \text{ (m, 9 H) ppm.}^{13}\text{C} \text{ NMR}$ $(100.62 \text{ MHz}, \text{ CDCl}_3)$: $\delta = 13.7, 58.9, 61.8, 116.8, 122.5, 123.7,$ 125.7, 128.3, 128.8, 129.0, 130.4, 131.3, 137.5, 166.0 ppm. GC-MS $(70 \text{ eV}): m/z \ (\%) = 263 \ (2) \ [M^+], 180 \ (18), 144 \ (64), 129 \ (100), 115$ (10), 77 (34). IR (film): $\tilde{\nu} = 3060 \text{ cm}^{-1}$, 3020, 2920, 1740, 1600, 1490, 1370. (From allylic halides 10h and 10i) Yield trans, trans-11c: 158 mg (60%) and 144 mg (55%), trans, cis-13c: 66 mg (25%) and 60 mg (23%) respectively.

3-(Pent-1-enyl)-1,4-diphenylazetidin-2-one (inseparable mixture of diastereoisomers, 50:15): (from allylic halide **10f**). Overall yield 190 mg (65%), oil. *trans,trans-***11d**: ¹H NMR (400.13 MHz, CDCl₃): $\delta = 0.91$ (t, J = 7.4 Hz, 3 H), 1.41 (sext, J = 7.4 Hz, 2 H), 2.06 (q, J = 7.4 Hz, 2 H), 3.67 (dd, J = 8.0, 2.2 Hz, 1 H), 4.74 (d, J = 2.2 Hz, 1 H), 5.63 (dd, J = 8.0, 15.2 Hz, 1 H), 5.73–5.80 (m, 1 H), 7.02 (t, J = 7.2 Hz, 1 H), 7.02–7.38 (m, 9 H) ppm. ¹³C NMR (100.62 MHz, CDCl₃): $\delta = 13.6$, 22.1, 34.6, 61.9, 63.5, 117.0, 122.3, 123.8, 125.8, 128.4, 128.5, 129.0, 129.1, 136.5, 137.6, 166.2 ppm. GC-MS (70 eV): *m/z* (%) = 291 (2) [M⁺], 181(15), (180 (22), 172 (100), 77 (50). *trans,cis-***13d**: ¹H NMR (400.13 MHz, CDCl₃): $\delta = 0.83$ (t, J = 7.4 Hz, 3 H), 1.33 (sext, J = 7.4 Hz, 2 H), 1.96 (q, J = 7.4 Hz, 2 H), 3.97 (dd, J = 8.0, 2.2 Hz, 1 H), 4.72 (d, J = 2.2 Hz, 1 H), 5.63 (dd, J = 8.0, 8.0 Hz, 1 H), 5.73–5.80 (m, 1 H), 7.02 (t, J = 7.2 Hz, 1 H), 7.20–7.38 (m, 9 H) ppm. ¹³C

NMR (100.62 MHz, CDCl₃): $\delta = 13.6, 22.6, 29.7, 59.4, 62.1, 116.9, 121.8, 125.8, 127.3, 127.4, 128.6, 129.0, 129.2, 136.3, 137.7, 166.2 ppm. GC-MS (70 eV):$ *m/z* $(%) = 291 (2) [M⁺], 181 (15), 180 (22), 172 (100), 77 (50). IR (film) measured on the mixture: <math>\tilde{v} = 3060 \text{ cm}^{-1}$, 3000, 2910, 1740, 1595, 1490, 1370. (From allylic halide **10**) Overall yield 192 mg (66%), *trans,trans*-11d and *trans,cis*-13d in a ratio 52:14.

2,3-Diphenyl-2-azaspiro[3,5]non-5-en-1-one (inseparable mixture of diastereoisomers, 6:14): Overall yield 57 mg (20%), oil. *trans*-13e: ¹H NMR (400.13 MHz, CDCl₃): $\delta = 1.50-2.20$ (m, 6 H), 4.92 (s, 1 H), 5.83 (d, J = 10.0 Hz, 1 H), 6.06 (dt, J = 10.0, 6.0 Hz, 1 H); 7.15–7.35 (m, 10 H) ppm. ¹³C NMR (100.62 MHz, CDCl₃): $\delta = 19.6$, 25.1, 29.7, 58.8, 67.0, 117.3, 126.6, 127.6, 128.0, 128.6, 128.7, 128.9, 129.7, 130.1, 134.5, 170.2 ppm. GC-MS (70 eV): *m/z* (%) = 289 (2) [M⁺], 180 (22), 170 (100), 155 (17). *trans*-14e: ¹H NMR (400.13 MHz, CDCl₃): $\delta = 1.50-2.20$ (m, 6 H), 4.88 (s, 1 H), 5.19 (d, J = 10.0 Hz, 1 H), 5.77 (dt, J = 10.0, 6.0 Hz, 1 H), 7.15–7.35 (m, 10 H) ppm. ¹³C NMR (100.62 MHz, CDCl₃): $\delta = 19.9$, 24.6, 30.6, 60.4, 67.8, 117.3, 122.1, 123.7, 126.6, 127.9, 128.7, 129.0, 133.0, 135.5, 137.8, 170.2 ppm. GC-MS (70 eV): *m/z* (%) = 289 (2) [M⁺], 180 (22), 170 (100), 155 (17). IR (film) measured on the mixture: $\tilde{v} = 3050$ cm⁻¹, 3000, 2905, 1735, 1595, 1490, 1370.

trans-1-Benzyl-4-phenyl-3-vinylazetidin-2-one (11f): Yield 223 mg (85%), oil. ¹H NMR (400.13 MHz, CDCl₃): δ = 3.64 (dd, *J* = 7.7, 1.8 Hz, 1 H), 3.76 (d, *J* = 15.0 Hz, 1 H), 4.19 (d, *J* = 1.8 Hz, 1 H), 4.83 (d, *J* = 15.0 Hz, 1 H), 5.23 (dd, *J* = 23.0, 17.1 Hz, 2 H), 5.86-5.94 (m, 1 H), 7.11 (d, *J* = 6.5 Hz, 1 H), 7.15-7.35 (m, 9 H) ppm. ¹³C NMR (100.62 MHz, CDCl₃): δ = 44.1, 60.3, 63.6, 118.7, 126.1, 127.3, 128.0, 128.2, 128.4, 128.7, 130.6, 135.1, 136.8, 167.5 ppm. GC-MS (70 eV): *m/z* (%) = 263 (1) [M⁺], 194 (5), 130 (100), 129 (90), 115 (55), 91 (75). IR (film): \tilde{v} = 3060 cm⁻¹, 3020, 2900, 1740, 1490, 1390.

trans-1-Benzyl-3-(2-methylpropenyl)-4-phenylazetidin-2-one (11g): Yield 227 mg (78%), oil. ¹H NMR (400.13 MHz, CDCl₃): $\delta = 1.51$ (d, J = 1.0 Hz, 3 H), 1.72 (s, 3 H), 3.76 (d, J = 15.0 Hz, 1 H), 3.83 (dd, J = 7.9, 2.1 Hz, 1 H), 4.10 (d, J = 2.1 Hz, 1 H), 4.83 (d, J = 15.0 Hz, 1 H), 5.30 (dq, J = 7.9, 1.0 Hz, 1 H), 7.11–7.37 (m, 10 H) ppm. ¹³C NMR (100.62 MHz, CDCl₃): $\delta = 18.3$, 25.3, 44.1, 59.9, 61.3, 117.1, 126.0, 126.1, 127.3, 128.0, 128.4, 128.6, 135.3, 137.2, 137.8, 169.0 ppm. GC-MS (70 eV): m/z (%) = 291 (0) [M⁺], 195 (30), 194 (30), 91 (100). IR (film): $\tilde{v} = 3060$ cm⁻¹, 3030, 2920, 1740, 1600, 1500, 1390.

1-Benzyl-4-phenyl-3-propenylazetidin-2-one (inseparable mixture of diastereoisomers, 60:20): (From allylic halide 10d). Overall yield 222 mg (80%), oil. trans, trans-11h: ¹H NMR (400.13 MHz, CDCl₃): $\delta = 1.70$ (d, J = 6.4 Hz, 3 H), 3.60 (dd, J = 8.0, 2.0 Hz, 1 H), 3.77 (d, J = 15.0 Hz, 1 H), 4.14 (d, J = 2.0 Hz, 1 H), 4.85 $(d, J = 15.0 \text{ Hz}, 1 \text{ H}), 5.53 (dd, J = 16.8, 8.0 \text{ Hz}, 1 \text{ H}), 5.70 (dq, J = 16.8, 8.0 \text{ Hz}, 1 \text{ H})), 5.70 (dq, J = 16.8, 8.0 \text{ Hz}, 1 \text{ H$ J = 16.8, 6.4 Hz, 1 H), 7.13–7.37 (m, 10 H) ppm. *trans,cis*-13h: ¹H NMR (400.13 MHz, CDCl₃): $\delta = 1.55$ (d, J = 6.8 Hz, 3 H), 3.77 (d, J = 15.0 Hz, 1 H), 3.93 (dd, J = 8.0, 2.0 Hz, 1 H), 4.12(d, J = 2.0 Hz, 1 H), 4.85 (d, J = 15.0 Hz, 1 H), 5.53 (dd, J = 8.0,8.0 Hz, 1 H), 5.70 (dq, J = 8.0, 6.8 Hz, 1 H), 7.13–7.37 (m, 10 H) ppm. ¹³C NMR, GC-MS and IR data were measured on the mixture. ¹³C NMR (100.62 MHz, CDCl₃): $\delta = 13.6, 17.9, 44.3, 44.4,$ 58.9, 61.1, 61.3, 63.5, 121.7, 122.8, 123.7, 126.38, 126.4, 127.4, 127.6, 128.3, 128.4, 128.41, 128.7, 128.9, 128.91, 129.8, 130.7, 135.4, 135.5, 137.2, 137.3, 138.0, 168.8, 168.9 ppm. GC-MS $(70 \text{ eV}): m/z \ (\%) = 277 \ (0) \ [M^+], \ 196 \ (7), \ 144 \ (68), \ 129 \ (100), \ 91$ (43), 65 (20). IR (film): $\tilde{v} = 3060 \text{ cm}^{-1}$, 3000, 2910, 1740, 1600, 1490, 1380, 1100. (From allylic halide 10h) Overall yield 235 mg (85%), *trans,trans*-11h and *trans,cis*-13h in a ratio 65:25; (from allylic halide 10i) overall yield 244 mg (88%), *trans,trans*-11h and *trans,cis*-13h in a ratio 65:25.

trans,trans-1-Benzyl-4-phenyl-3-styrylazetidin-2-one (11i): Yield 332 mg (98%), oil. ¹H NMR (400.13 MHz, CDCl₃): δ = 3.83 (d, J = 15.0 Hz, 1 H), 3.83–3.86 (m, 1 H), 4.27 (d, J = 2.1 Hz, 1 H), 4.90 (d, J = 15.0 Hz, 1 H), 6.26 (dd, J = 15.8, 8.4 Hz, 1 H), 6.58 (d, J = 15.8 Hz, 1 H), 7.20–7.40 (m, 15 H) ppm. ¹³C NMR (100.62 MHz, CDCl₃): δ = 44.6, 61.3, 63.7, 122.1, 126.4, 126.5, 127.7, 127.8, 128.4, 128.5, 128.6, 128.8, 129.1, 134.1, 135.4, 136.4, 137.1, 168.1 ppm. GC-MS (70 eV): m/z (%) = 339 (0) [M⁺], 195 (30), 194 (28), 144 (27), 115 (35), 91 (100). IR (film): \tilde{v} = 3060 cm⁻¹, 3020, 2900, 1750, 1600, 1590, 1450, 1350, 730, 690.

1-Benzyl-3-(pent-1-enyl)-4-phenylazetidin-2-one (inseparable mixture of diastereoisomers, 50:15): (From allylic halide 10f). Overall yield 229 mg (75%), mixture of solid. M.p. 60-63 °C (n-hexane). *trans*,*trans*-111: ¹H NMR (400.13 MHz, CDCl₃): $\delta = 0.87$ (t, J =7.3 Hz, 3 H), 1.37 (sext, J = 7.3 Hz, 2 H), 2.00 (q, J = 7.3 Hz, 2 H), 3.61 (dd, J = 8.0, 1.6 Hz, 1 H), 3.76 (d, J = 15.0 Hz, 1 H), 4.14 (d, J = 1.6 Hz, 1 H), 4.86 (d, J = 15.0 Hz, 1 H), 5.50 (dd, J = 16.0, 8.0 Hz, 1 H), 5.66-5.70 (m, 1 H), 7.13 (d, J = 6.5 Hz, 2 H), 7.22 (d, J = 6.5 Hz, 2 H), 7.27–7.37 (m, 6 H) ppm. ¹³C NMR $(100.62 \text{ MHz}, \text{CDCl}_3): \delta = 13.5, 22.0, 34.5, 44.3, 61.2, 63.4, 122.5,$ 126.3, 127.5, 128.26, 128.3, 128.31, 128.6, 128.8, 135.6, 137.3, 168.7 ppm. GC-MS (70 eV): m/z (%) = 305 (1) [M⁺], 196 (10), 172 (62), 143 (45), 129 (100), 115 (17), 91 (59). *trans,cis*-13l: ¹H NMR $(400.13 \text{ MHz}, \text{CDCl}_3)$: $\delta = 0.77$ (t, J = 7.3 Hz, 3 H), 1.33 (sext, J = 7.3 Hz, 2 H), 1.90 (q, J = 7.3 Hz, 2 H), 3.77 (d, J = 15.0 Hz, 1 H),3.90 (dd, J = 8.0, 1.6 Hz, 1 H), 4.14 (d, J = 1.6 Hz, 1 H), 4.84 (d, J = 15.0 Hz, 1 H), 5.50 (dd, J = 8.0, 8.0 Hz, 1 H), 5.66-5.70 (m, 1 H), 7.13 (d, J = 6.5 Hz, 2 H), 7.22 (d, J = 6.5 Hz, 2 H), 7.27-7.37 (m, 6 H) ppm. ¹³C NMR (100.62 MHz, CDCl₃): $\delta = 13.5, 22.4, 29.8, 44.4, 59.3, 61.4, 122.0, 126.3, 127.5, 128.2,$ 128.3, 128.4, 128.6, 128.9, 135.5, 137.3, 168.7 ppm. GC-MS $(70 \text{ eV}): m/z \ (\%) = 305 \ (1) \ [M^+], \ 196 \ (10), \ 172 \ (62), \ 143 \ (45), \ 129$ (100), 115 (17), 91 (59). IR (film) measured on the mixture: $\tilde{v} =$ 3060 cm⁻¹, 3000, 2910, 1740, 1600, 1490, 1380, 1110. C₂₁H₂₃NO (305.42) measured on the mixture: calcd. C 82.58, H 7.59, N 4.59; found C 82.10, H 7.72, N 4.50. (From allylic halide 101). Overall yield 207 mg (68%), trans, trans-111 and trans, cis-131 in a ratio 53:15.

2-Benzyl-3-phenyl-2-azaspiro[3.5]non-5-en-1-one (inseparable mixture of diastereoisomers, 4:11): Overall yield 45 mg (15%), oil. trans-**13m:** ¹H NMR (400.13 MHz, CDCl₃): $\delta = 1.50 - 2.10$ (m, 6 H), 3.90 (d, J = 14.8 Hz, 1 H), 4.28 (s, 1 H), 4.96 (d, J = 14.8 Hz, 1 H)H), 5.67 (d, J = 10.2 Hz, 1 H), 5.96 (dt, J = 10.2, 6.3 Hz, 1 H), 7.15-7.39 (m, 10 H). *cis*-14m: ¹H NMR (400.13 MHz, CDCl₃): $\delta = 1.5 \ 0 - 2.10 \ (m, 6 \ H), 3.88 \ (d, J = 14.8 \ Hz, 1 \ H), 4.22 \ (s, 1 \ H),$ 5.00 (d, J = 14.8 Hz, 1 H), 5.23 (d, J = 10.2 Hz, 1 H), 5.71 (dt, J = 10.2, 6.3 Hz, 1 H), 7.15–7.39 (m, 10 H) ppm. ¹³C NMR, GC-MS and IR data were measured on the mixture. ¹³C NMR $(100.62 \text{ MHz}, \text{CDCl}_3): \delta = 19.1, 19.9, 24.4, 24.6, 24.7, 30.0, 44.1,$ 44.3, 60.9, 61.2, 65.8, 67.3, 122.4, 125.7, 126.9, 127.2, 127.6, 127.7, 127.8, 128.1, 128.3, 128.4, 128.5, 128.6, 128.7, 128.8, 132.1, 132.5, 135.6, 135.8, 136.0, 138.1, 172.6, 172.8 ppm. GC-MS (70 eV): m/z $(\%) = 303 (2) [M^+], 196 (7), 170 (100), 155 (17), 144 (17), 142 (17),$ 141 (16), 115 (10), 108 (22), 91 (67). IR (film): $\tilde{v} = 3060 \text{ cm}^{-1}$, 3020, 2900, 1740, 1490, 1450.

4-Phenyl-1-(1-phenyl-ethyl)-3-vinylazetidin-2-one (*trans***-11n**): Yield 166 mg (60%), oil. ¹H NMR (400.13 MHz, CDCl₃): δ = 1.80 (d, J = 7.1 Hz, 3 H), 3.59 (dd, J = 7.7, 1.9 Hz, 1 H), 4.13 (d, J =

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1.9 Hz, 1 H), 4.28 (q, J = 7.1 Hz, 1 H), 5.24 (dd, J = 17.7, 10.3 Hz, 2 H), 5.87–5.96 (m, 1 H), 7.16–7.32 (m, 10 H) ppm. ¹³C NMR (100.62 MHz, CDCl₃): $\delta = 20.1$, 54.6, 60.5, 63.1, 118.9, 126.5, 126.7, 127.4, 128.3, 128.5, 128.8, 131.0, 137.6, 141.3, 168.0 ppm. GC-MS (70 eV): *mlz* (%) = 264 (0) [M⁺], 130 (100), 129 (50), 115 (20), 105 (22), 77 (21). IR (film): $\tilde{v} = 3060$ cm⁻¹, 2980, 1735, 1450. *trans*-11n': Yield 89 mg (32%), oil. ¹H NMR (400.13 MHz, CDCl₃): $\delta = 1.31$ (d, J = 7.2 Hz, 3 H), 3.62 (dd, J = 7.6, 2.2 Hz, 1 H), 4.05 (d, J = 2.2 Hz, 1 H), 5.06 (q, J = 7.2 Hz, 1 H), 5.15–5.26 (m, 2 H), 5.77–5.86 (m, 1 H), 7.20–7.33 (m, 10 H) ppm. ¹³C NMR (100.62 MHz, CDCl₃): $\delta = 18.6$, 52.0, 60.7, 63.2, 119.0, 126.7, 127.2, 127.7, 128.5, 128.6, 128.7, 130.7, 138.9, 139.8, 168.2 ppm. GC-MS (70 eV): *m/z* (%) = 264 (0) [M⁺], 130 (100), 129 (50), 115 (20), 105 (22), 77 (21). IR (film): $\tilde{v} = 3060$ cm⁻¹, 2980, 1735, 1450.

1-(2-Methoxy-1-phenylethyl)-4-phenyl-3-vinylazetidin-2-one (trans-110): Yield 160 mg (52%), solid. M.p. 97.2-99.2 °C (n-hexane). ¹H NMR (400.13 MHz, CDCl₃): $\delta = 3.40$ (s, 3 H), 3.61 (dd, J = 7.9, 2.0 Hz, 1 H), 3.69 (dd, J = 6.9, 2.4 Hz, 1 H), 4.22 (d, J = 2.0 Hz, 1 H), 4.28-4.33 (m, 2 H), 5.25 (dd, J = 19.0, 10.4 Hz, 2 H), 5.90-5.98 (m, 1 H), 7.19-7.36 (m, 10 H) ppm. ¹³C NMR $(100.62 \text{ MHz}, \text{CDCl}_3)$: $\delta = 58.7, 59.0, 60.4, 63.4, 73.0, 119.0, 126.6,$ 127.4, 127.8, 128.3, 128.6, 128.7, 131.0, 137.5, 137.6, 168.1 ppm. GC-MS (70 eV): m/z (%) = 307 (0) [M⁺], 262 (10), 240 (21), 194 (27), 130 (100), 129 (81), 115 (50), 91 (28). IR (CHCl₃): $\tilde{v} = 3060$ cm⁻¹, 3030, 2920, 2850, 1600, 1520, 1490, 1430, 1340, 1310, 1110, 960, 760, 730, 690. C₂₀H₂₁NO₂ (307.39): calcd. C 78.14, H 6.89, N 4.55; found C 78.10, H 7.02, N 4.50. trans-11o': Yield 123 mg (40%), oil. ¹H NMR (400.13 MHz, CDCl₃): $\delta = 3.25$ (s, 3 H), 3.43 (dd, J = 9.3, 5.4 Hz, 1 H), 3.60 (dd, J = 8.0, 2.0 Hz, 1 H), 3.75 (t, t)J = 9.3 Hz, 1 H), 4.24 (d, J = 2.0 Hz, 1 H), 4.81 (dd, J = 9.3, 5.4 Hz, 1 H), 5.18-5.23 (m, 2 H), 5.83-5.97 (m, 1 H), 7.22-7.34 (m, 10 H) ppm. ¹³C NMR (100.62 MHz, CDCl₃): $\delta = 57.8, 58.5,$ 62.3, 63.2, 72.0, 119.0, 126.6, 126.8, 127.8, 127.9, 128.3, 128.6, 130.9, 136.7, 138.6, 168.7 ppm. GC-MS (70 eV): *m*/*z* (%) = 307 (0) [M⁺], 262 (10), 240 (21), 194 (27), 130 (100), 129 (81), 115 (50), 91 (28). IR (film): $\tilde{v} = 3060 \text{ cm}^{-1}$, 3030, 2920, 2850, 1600, 1520, 1490, 1430, 1340, 1310, 1110, 960, 760, 730, 690.

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