

C–C (alkynylation) vs C–O (ether) bond formation under Pd/C–Cu catalysis: synthesis and pharmacological evaluation of 4-alkynylthieno[2,3-*d*]pyrimidines

Dhilli Rao Gorja^{1,2}, K. Shiva Kumar¹, K. Mukkanti² and Manojit Pal^{*1}

Full Research Paper

Open Access

Address:

¹Institute of Life Sciences, University of Hyderabad Campus, Gachibowli, Hyderabad-500 046, India and ²Chemistry Division, Institute of Science and Technology, JNT University, Kukatpally, Hyderabad 500 072, India

Email:

Manojit Pal* - manojitpal@rediffmail.com

* Corresponding author

Keywords:

catalysis; C–C bond; copper; palladium; thieno[2,3-*d*]pyrimidine

Beilstein J. Org. Chem. **2011**, *7*, 338–345.

doi:10.3762/bjoc.7.44

Received: 05 December 2010

Accepted: 08 March 2011

Published: 21 March 2011

Associate Editor: J. Aubé

© 2011 Gorja et al; licensee Beilstein-Institut.

License and terms: see end of document.

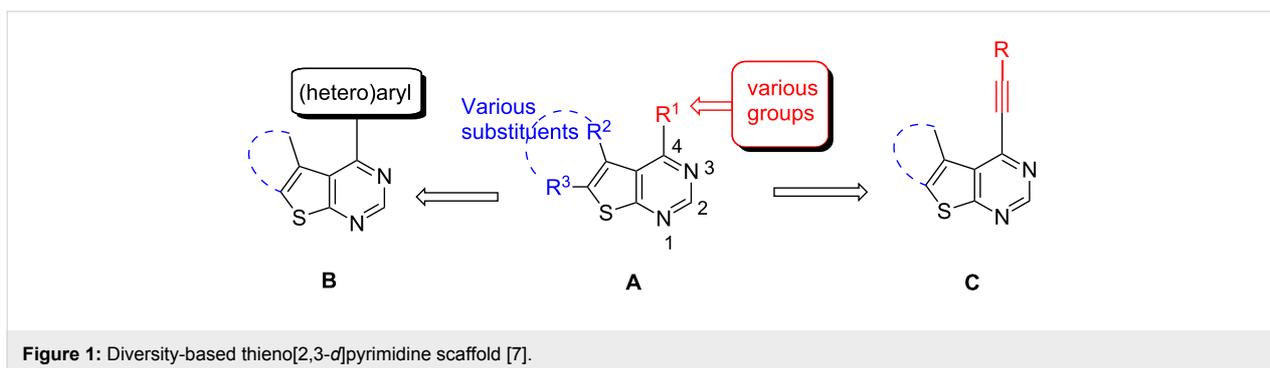
Abstract

The Pd/C–CuI–PPh₃ catalytic system facilitated C–C bond formation between 4-chlorothieno[2,3-*d*]pyrimidines and terminal alkynes in methanol with high selectivity without generating any significant side products arising from C–O bond formation between the chloro compounds and methanol. A variety of novel 4-alkynylthieno[2,3-*d*]pyrimidines were prepared via alkynylation of 4-chlorothieno[2,3-*d*]pyrimidines in good to excellent yields. Some of the compounds synthesized were tested for cytotoxic activity in vitro.

Introduction

Alkynyl substituted pyrimidines are of considerable pharmacological interest because of their notable biological activities [1], in particular, adenosine kinase inhibitory activity in the treatment of pain and inflammatory diseases [2] and thymidylate synthase inhibitory properties in cancer therapy [3]. On the other hand, the thiophene moiety is a common feature in many bioactive agents and drugs [4] and is considered as a bioisostere of the benzene ring [5]. Thus, one can anticipate that combining the pyrimidine ring of an alkynyl substituted pyrimidine moiety

with a thiophene ring might afford compounds, i.e., alkynyl substituted thienopyrimidines of potential pharmacological interest. Notably, 6-ethynylthieno[3,2-*d*]- and 6-ethynylthieno[2,3-*d*]pyrimidin-4-aniline derivatives were found to be potent inhibitors of ErbB family receptor tyrosine kinases (EGFR, ErbB-2) and the proliferation of tumor cells that highly express these kinases [6]. In continuation of our research program into new drug discovery, we became interested in the generation of a small-molecule library A (Figure 1) based on

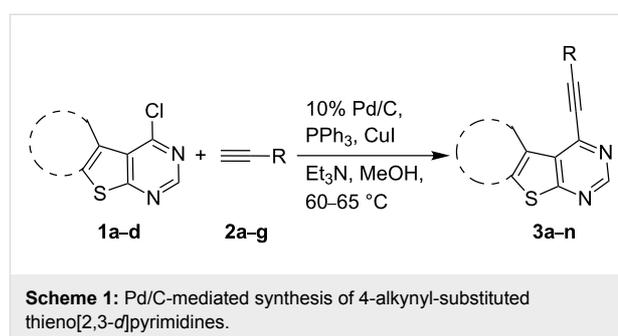


thieno[2,3-*d*]pyrimidine for in-house pharmacological evaluation. Accordingly, we recently reported the synthesis of 4-(hetero)aryl substituted thieno[2,3-*d*]pyrimidines **B** [7]. As a further extension of this research and in view of possible pharmacological value of compounds containing alkyne, thiophene and pyrimidine moieties, we now wish to report the synthesis and in vitro cytotoxicity of novel 4-alkynylthieno[2,3-*d*]pyrimidines **C** (Figure 1). These derivatives are attractive due to the synthetic potential of C-4 alkyne fragments for further use in library construction.

A number of methods have been reported for the synthesis of alkynyl substituted pyrimidines and most of which involve the use of Sonogashira coupling of halopyrimidines with terminal alkynes [1-3] (for a review see [8]). While alkylation of the thiophene ring of thienopyrimidines under Sonogashira conditions [9] has previously been reported [6], a similar coupling reaction on the pyrimidine ring of thieno[2,3-*d*]pyrimidines is uncommon in the literature [10]. The use of Pd/C–CuI–PPh₃ as a less expensive catalyst system for efficient Sonogashira coupling has been explored earlier. Due to our continuing interest in Pd/C-mediated alkylation of aryl and heteroaryl halides we decided to investigate the Pd/C-based methodology for the synthesis of our target compounds, i.e., 4-alkynylthieno[2,3-*d*]pyrimidines as shown in Scheme 1.

Results and Discussion

The key starting materials, i.e., 4-chlorothieno[2,3-*d*]pyrimidines **1a–c** required for our synthesis were prepared according



to a known method reported earlier [7]. The other chloro compound 4-chloro-6,7,8,9-tetrahydro-5*H*-cyclohepta[4,5]thieno[2,3-*d*]pyrimidine (**1d**) was prepared from cycloheptanone and ethyl cyanoacetate following a similar procedure as shown in Scheme 2.

Condensation of resulting amino ester, i.e., ethyl 2-amino-5,6,7,8-tetrahydro-4*H*-cyclohepta[*b*]thiophene-3-carboxylate (**4**) with formamide gave 6,7,8,9-tetrahydro-5*H*-cyclohepta[4,5]thieno[2,3-*d*]pyrimidin-4-one (**5**) which on treatment with POCl₃ under refluxing conditions provided the desired 4-chloro derivative **1d**. All the terminal alkynes used were commercially available. Initially, we chose to examine the coupling reaction of 4-chloro-5,6,7,8-tetrahydrobenzothieno[2,3-*d*]pyrimidine (**1a**) with phenylacetylene (**2a**) in the presence of 10% Pd/C (0.023 mmol), PPh₃ (0.17 mmol), CuI (0.04 mmol), and triethylamine (2.67 mmol) in various solvents. The results of this study are summarized in Table 1. The reaction was initially

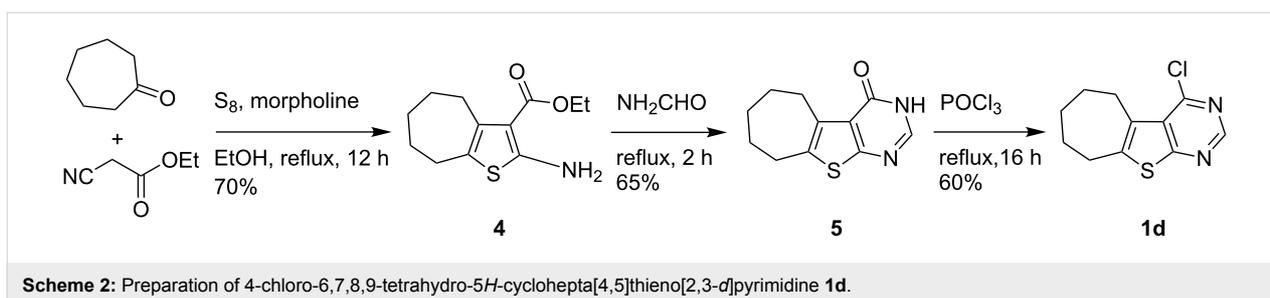
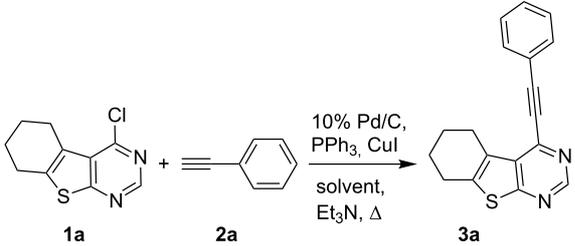


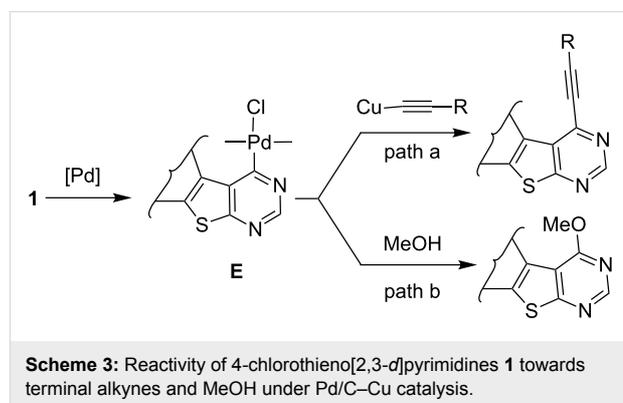
Table 1: Effect of solvent on the coupling reaction of 4-chloro-5,6,7,8-tetrahydrobenzothieno[2,3-*d*]pyrimidine (**1a**) with phenylacetylene (**2a**).^a


Entry	Solvent	Time (h)	Yield ^b (%)
1	MeOH	5	67
2	MeOH	10	90
3	THF	14	80
4	MeCN	12	57 (80 ^c)
5	1,4-dioxane	10	59 (80 ^c)
6	DMF	10	49 (75 ^c)
7	MeOH	10	83 ^d

^aAll reactions were carried out by using **1** (0.89 mmol), **2a** (1.33 mmol), 10% Pd/C (0.023 mmol), PPh₃ (0.17 mmol), CuI (0.04 mmol), and Et₃N (2.67 mmol) at 60–65 °C under a nitrogen atmosphere. ^bIsolated yields. ^cThe reaction was carried out at 80 °C. ^d(PPh₃)₂PdCl₂-CuI was used instead of 10% Pd/C-PPh₃-CuI.

carried out in MeOH for 5 h and the desired product, i.e., 4-(phenylethynyl)-5,6,7,8-tetrahydrobenzothieno[2,3-*d*]pyrimidine (**3a**) was isolated in 67% yield (Table 1, entry 1). The yield of **3a** however was significantly improved when the reaction time was increased to 10 h (Table 1, entry 2). Thus the reaction proceeded well in MeOH to give the expected product via a C–C bond forming reaction (Scheme 3, path a) and no side product as a result of C–O bond formation [11] due to participation of MeOH (Scheme 3, path b) was detected in the reaction mixture. The use of MeOH as a nucleophile in Pd-catalyzed reactions has been well documented in the literature, see for example [12]. Moreover, when the addition of alkyne was omitted without changing the other reaction conditions, compound **1a** reacted with MeOH to give the corresponding ether, 4-methoxy-5,6,7,8-tetrahydrobenzothieno[2,3-*d*]pyrimidine, albeit in low yield. The use of other solvents such as THF, acetonitrile, 1,4-dioxane and DMF (Table 1, entries 3–6) also gave good yields of product but MeOH was found to be superior. Moreover, the reaction temperature (and duration in some cases) was higher, i.e., 80 °C in case of acetonitrile, 1,4-dioxane and DMF compared to 60–65 °C in case of MeOH. While triethylamine was used as a base in all these cases, the use of a secondary amine, e.g., pyrrolidine was also examined. A side product was observed in this case due to the C–N bond forming reaction between **1a** and pyrrolidine and was identified as 4-(*N*-pyrrolidinyl)-5,6,7,8-tetrahydrobenzothieno[2,3-*d*]pyrimidine.

The use of another palladium catalyst (PPh₃)₂PdCl₂ was also examined and found to be effective (Table 1, entry 7). However, we preferred to use Pd/C because it is less expensive, easy to handle and separable from the product and has the potential for recyclability, for a review see [13]. To assess the recyclability of the recovered Pd/C-catalyst in the present coupling reaction, the reaction mixture of **1a** and **2a** (Table 1, entry 2) was allowed to cool to room temperature and filtered. The residue was washed with MeOH, acetone, and DCM. After drying under vacuum the recovered catalyst was used directly in the reaction of **1a** with **2a** in the presence of PPh₃, CuI, and Et₃N: A conversion of 85% was observed confirming the recyclability of the recovered Pd/C-catalyst.



Having established the optimum reaction conditions for the preparation of **3a** we decided to test the generality and scope of this protocol further. Thus, a variety of commercially available terminal alkynes was employed under the reaction conditions indicated in entry 2 of Table 1 and the results are summarized in Table 2.

As evident from Table 2, the reaction proceeded well with both aliphatic (Table 2, entries 2–6 and 8–14) and aromatic alkynes (Table 2, entry 1 and 7). The chloro compounds containing a five (**1b**), a six (**1a**) or a seven membered cycloalkane ring (**1d**) or without a ring (**1c**) were found to be equally effective under the reaction conditions employed. All the reactions were generally complete within 10 h irrespective of the nature of substituents present in the terminal alkynes **2** to afford the desired products **3a–n** in good to excellent yields.

A plausible mechanism for the Pd/C–Cu mediated alkylation of 4-chlorothieno[2,3-*d*]pyrimidines **1** is shown in Scheme 4. The alkylation proceeds via generation of an active Pd(0) species in situ that undergoes oxidative addition with **1** to give the organo-Pd(II) species **E**. The active Pd(0) species is generated from the minor portion of the bound palladium (Pd/C) via a Pd leaching process in the solution [13]. The leached Pd then

Table 2: Pd/C-mediated synthesis of 4-alkynylthieno[2,3-*d*]pyrimidine in methanol.^a

Entry	4-Chlorothieno[2,3- <i>d</i>]pyrimidine	R-≡ (R=)	Products ^b	Yield ^c (%)
1		-C ₆ H ₅ (2a)		90
2	1a	-CH ₂ OH (2b)		88
3	1a	-CH ₂ CH ₂ OH (2c)		85
4	1a	-(CH ₂) ₂ CH ₃ (2d)		75
5	1a	-C(CH ₃) ₂ OH (2e)		77
6		-C(CH ₃) ₂ OH (2e)		80

Table 2: Pd/C-mediated synthesis of 4-alkynylthieno[2,3-*d*]pyrimidine in methanol.^a (continued)

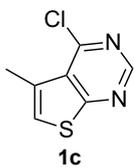
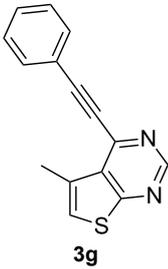
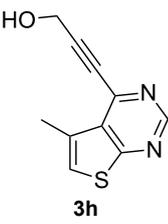
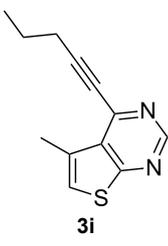
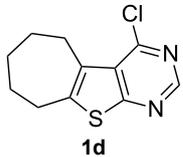
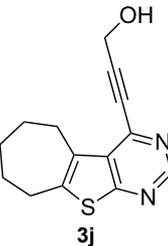
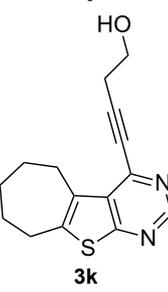
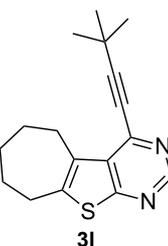
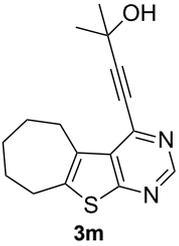
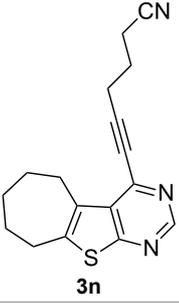
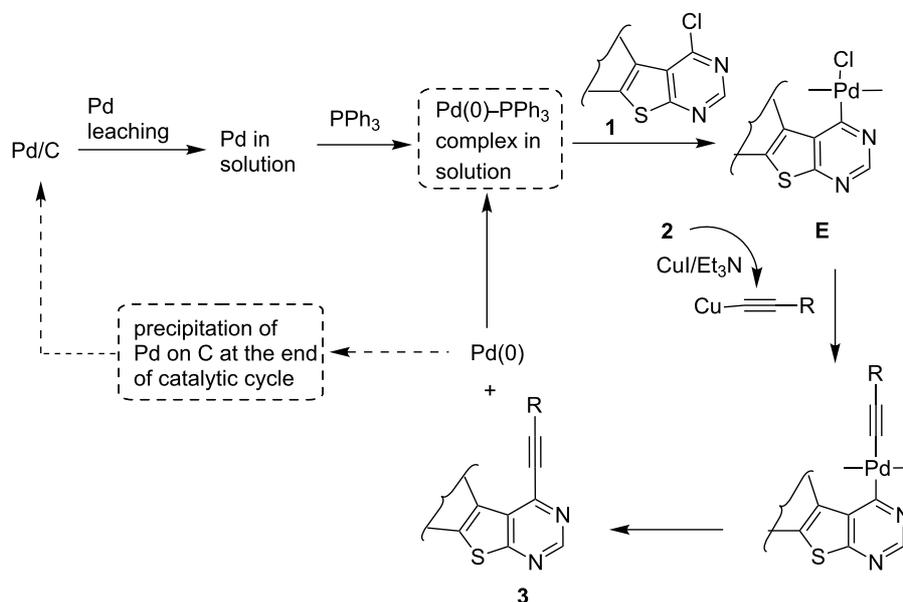
7		-C ₆ H ₅ (2a)		90
8	1c	-CH ₂ OH (2b)		85
9	1c	-(CH ₂) ₂ CH ₃ (2d)		79
10		-CH ₂ OH (2b)		87
11	1d	-CH ₂ CH ₂ OH (2c)		88
12	1d	-C(CH ₃) ₃ (2f)		85

Table 2: Pd/C-mediated synthesis of 4-alkynylthieno[2,3-*d*]pyrimidine in methanol.^a (continued)

13	1d	-C(CH ₃) ₂ OH (2e)		85
14	1d	-C(CH ₂) ₃ CN (2g)		80

^aAll the reactions were carried out with **1** (0.89 mmol), **2** (1.33 mmol), 10% Pd/C (0.023 mmol), PPh₃ (0.17 mmol), CuI (0.04 mmol), and Et₃N (2.67 mmol) in MeOH (5.0 mL) at 60–65 °C for 10–12 h. ^bIdentified by ¹H NMR, IR, and MS. ^cIsolated yields.

**Scheme 4:** Plausible mechanism of Pd/C-mediated alkylation of 4-chlorothieno[2,3-*d*]pyrimidines **1**.

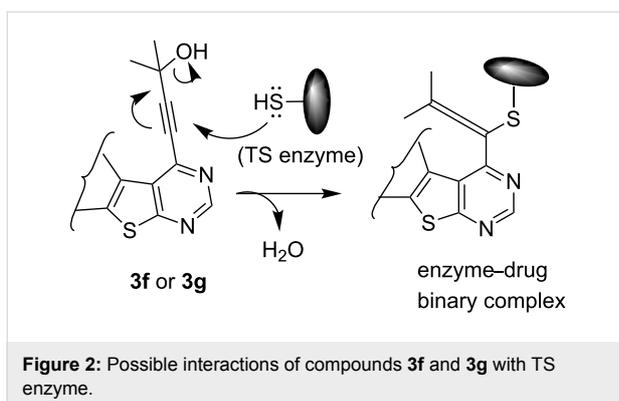
becomes an active species by interacting with phosphine ligands. Thus, a soluble Pd(0)–PPh₃ complex is the active species that actually catalyzes the C–C bond forming reaction in solution. The catalytic cycle therefore works in solution rather than on the surface, and at the end of the reaction, re-precipitation of Pd occurs on the surface of the charcoal. Once generated, the organo-Pd(II) species **E** then facilitates the stepwise

formation of C–C bond via (i) trans organometallation with copper acetylide generated in situ from CuI and the terminal alkyne followed by (ii) reductive elimination of Pd(0) to afford 4-alkynylthieno[2,3-*d*]pyrimidine (**3**). A C–O bond forming reaction between **1** and MeOH (Scheme 2, path b) was not observed perhaps due to the higher reactivity of copper acetylide over MeOH (even although present in excess) towards **E**.

Table 3: Cytotoxic activity of 4-alkynylthieno[2,3-*d*]pyrimidines (**3**) against chronic myelogenous leukemia (CML) cell line.

Compound	% of cell death at two concentrations	
	10 μ M	20 μ M
3a	20	26
3b	10	24
3c	25	26
3d	28	36
3f	30	54
3g	32	60
3h	24	26
10% DMSO	5.4	10
Untreated	nil	nil

We have shown that an alkynyl moiety can be introduced efficiently at the C-4 position of a thieno[2,3-*d*]pyrimidine ring via C–C coupling under Pd/C–Cu catalysis. To the best of our knowledge, only two examples of similar coupling have been reported using the system $(\text{PPh}_3)_2\text{PdCl}_2/\text{CuI}$ as catalyst in the presence of Et_3N [10]. The alkynyl substituent of compound **3** could be utilized for further structural elaboration leading to the functionalized derivatives of thieno[2,3-*d*]pyrimidine preparation which may be difficult to access by other methods. Some of the 4-alkynyl-derivatives synthesized were screened for their cytotoxic activity against chronic myelogenous leukemia (CML) cell line in vitro. The percentage of cell death measured for each compound at two different concentrations, i.e., 10 and 20 μM is shown in Table 3. As evident from Table 3, most of the compounds showed moderate to low activity against CML under the in vitro condition employed. The compounds **3f** and **3g** however, showed reasonable activity when tested at 10 μM which doubled at 20 μM . Comparing the structural features of compounds **3f** and **3g** with the other compounds tested, it may be noted that the presence of CMe_2OH moiety played an important role in the cytotoxic activities of these compounds. It is known that alkynyl substituted pyrimidine derivatives exhibit



their anticancer properties by inhibiting a key enzyme, i.e., thymidylate synthase (TS) which is essential for cellular growth [14]. A possible explanation for observed cytotoxic activities of present series of compounds **3** therefore could be their potential inhibition of TS [15]. The CMe_2OH group next to the alkynyl moiety may facilitate the binding of the TS enzyme through its sulfhydryl ($-\text{SH}$) moiety with compounds **3f** and **3g** thereby generating the corresponding drug–enzyme allene intermediates (Figure 2). Nevertheless, the present study indicated that 4-alkynylthieno[2,3-*d*]pyrimidine can be used as a template for the identification of novel and potential anticancer agents.

Conclusion

In conclusion, the present study demonstrates the first efficient synthesis 4-alkynylthieno[2,3-*d*]pyrimidines in good to excellent yields. The combination of Pd/C–CuI– PPh_3 proved to be an efficient and selective catalytic system for the C–C bond formation between 4-chlorothieno[2,3-*d*]pyrimidine and terminal alkynes in methanol providing a general and practical method for the preparation of these novel compounds. The reaction proceeds well with both hydrophobic and hydrophilic terminal alkynes without generating any significant side products arising from C–O bond formation or dimerization of the terminal alkynes. The reaction does not involve the use of expensive catalysts or reagents and is easy to perform. Some of the compounds synthesized were tested for cytotoxic activities in vitro. The methodology is amenable to the diversity-oriented synthesis of thieno[2,3-*d*]pyrimidine derivatives of potential pharmacological significance and therefore may find use in organic and medicinal chemistry.

Supporting Information

Supporting Information features details on experimental procedures and spectral data as well as NMR spectra of compounds **3a–n**.

Supporting Information File 1

Experimental procedures and spectral data.

[<http://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-7-44-S1.pdf>]

Supporting Information File 2

NMR spectra of compounds **3a–n**.

[<http://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-7-44-S2.pdf>]

Acknowledgements

The authors sincerely thank the management of Institute of Life Sciences for continuous encouragement and support.

References

- Rai, D.; Johar, M.; Manning, T.; Agrawal, B.; Kunimoto, D. Y.; Kumar, R. *J. Med. Chem.* **2005**, *48*, 7012–7017. doi:10.1021/jm058167w
- Gomtsyan, A.; Didomenico, S.; Lee, C.-H.; Matulenko, M. A.; Kim, K.; Kowaluk, E. A.; Wismer, C. T.; Mikusa, J.; Yu, H.; Kohlhaas, K.; Jarvis, M. F.; Bhagwat, S. S. *J. Med. Chem.* **2002**, *45*, 3639–3648. doi:10.1021/jm020049a
- Spector, T.; Porter, D. J. T.; Nelson, D. J.; Baccanari, D. P.; Davis, S. T.; Almond, M. R.; Khor, S. P.; Amyx, H.; Cao, S.; Rustum, Y. M. *Drugs Future* **1994**, *19*, 565–571.
- Campaigne, E. In *Comprehensive Heterocyclic Chemistry*; Katritzky, A. R., Ed.; Pergamon Press: New York, 1984; Vol. 4, pp 911–913.
- Sanfilippo, P. J.; McNally, J. J.; Press, J. B.; Fitzpatrick, L.; Urbanski, M. J.; Katz, L. B.; Giardino, E.; Falotico, R.; Salata, J.; Moore, J. B., Jr.; Miller, W. J. *J. Med. Chem.* **1992**, *35*, 4425–4433. doi:10.1021/jm00101a020
- Wood, E. R.; Shewchuk, L. M.; Ellis, B.; Brignola, P.; Brashear, R. L.; Caferro, T. R.; Dickerson, S. H.; Dickson, H. D.; Donaldson, K. H.; Gaul, M.; Griffin, R. J.; Hassell, A. M.; Keith, B.; Mullin, R.; Petrov, K. G.; Reno, M. J.; Rusnak, D. W.; Tadepalli, S. M.; Ulrich, J. C.; Wagner, C. D.; Vanderwall, D. E.; Waterson, A. G.; Williams, J. D.; White, W. L.; Uehling, D. E. *Proc. Natl. Acad. Sci. U. S. A.* **2008**, *105*, 2773–2778. doi:10.1073/pnas.0708281105
- Kumar, K. S.; Chamakuri, S.; Vishweshwar, P.; Iqbal, J.; Pal, M. *Tetrahedron Lett.* **2010**, *51*, 3269–3273. doi:10.1016/j.tetlet.2010.04.057
- Schröter, S.; Stock, C.; Bach, T. *Tetrahedron* **2005**, *61*, 2245–2267. doi:10.1016/j.tet.2004.11.074
- Sonogashira, K.; Tohda, Y.; Hagihara, N. *Tetrahedron Lett.* **1975**, *16*, 4467–4470. doi:10.1016/S0040-4039(00)91094-3
- Konno, S.; Watanabe, R.; Yamanaka, H. *Yakugaku Zasshi* **1989**, *109*, 642–649.
- Snégaroff, K.; Lassagne, F.; Bentabed-Ababsa, G.; Nassar, E.; Ely, S. C. S.; Hesse, S.; Perspicace, E.; Derdour, A.; Mongin, F. *Org. Biomol. Chem.* **2009**, *7*, 4782–4788. doi:10.1039/b915274a
- van der Deen, H.; van Oeveren, A.; Kellogg, R. M.; Feringa, B. L. *Tetrahedron Lett.* **1999**, *40*, 1755–1758. doi:10.1016/S0040-4039(98)02683-5
- Pal, M. *Synlett* **2009**, 2896–2912. doi:10.1055/s-0029-1218021
- Kundu, N. G.; Das, B.; Spears, C. P.; Majumdar, A.; Kang, S. I. *J. Med. Chem.* **1990**, *33*, 1975–1979. doi:10.1021/jm00169a026
- Rao, K. N.; Bhattacharya, R. K.; Venkatachalam, S. R. *Cancer Lett.* **1998**, *128*, 183–188. doi:10.1016/S0304-3835(98)00061-5

License and Terms

This is an Open Access article under the terms of the Creative Commons Attribution License (<http://creativecommons.org/licenses/by/2.0>), which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

The license is subject to the *Beilstein Journal of Organic Chemistry* terms and conditions: (<http://www.beilstein-journals.org/bjoc>)

The definitive version of this article is the electronic one which can be found at: doi:10.3762/bjoc.7.44