Discovery of Pyrazine-Carboxamide-Diphenyl-Ethers as Novel Succinate Dehydrogenase Inhibitors via Fragment Recombination

Hua Li, Meng-Qi Gao, Yan Chen, Yu-Xia Wang, Xiao-Lei Zhu,* and Guang-Fu Yang*

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ABSTRACT: The discovery of novel succinate dehydrogenase inhibitors (SDHIs) has attracted great attention worldwide. Herein, a fragment recombination strategy was proposed to design new SDHIs by understanding the ligand—receptor interaction mechanism of SDHIs. Three fragments, pyrazine from pyraziflumid, diphenyl-ether from flubeneteram, and a prolonged amide linker from pydiflumetofen and fluopyram, were identified and recombined to produce a pyrazine-carboxamide-diphenyl-ether scaffold as a new SDHI. After substituent optimization, compound **6**y was successfully identified with good inhibitory activity against porcine SDH, which was about 2-fold more potent than pyraziflumid. Furthermore, compound **6**y exhibited 95% and 80% inhibitory rates against soybean gray mold and wheat powdery mildew at a dosage of 100 mg/L *in vivo* assay, respectively. The results of the present work showed that the pyrazine-carboxamide-diphenyl-ether scaffold could be used as a new starting point for the discovery of new SDHIs.

KEYWORDS: succinate dehydrogenase, fragment, molecular docking, fungicide, diphenyl-ether

INTRODUCTION

Succinate dehydrogenase (SDH, EC 1.3.5.1, also known as succinate-ubiquinone oxidoreductase or complex II) is one of the most important fungicidal targets and is a functional part of the tricarboxylic acid cycle and linked to the mitochondrial electron transport chain.^{1,2} SDH catalyzes the oxidation of succinate to fumarate followed by reduction of ubiquinone to ubiquinol. Inhibiting SDH will cause the organism not to synthesize ATP normally and then die.³⁻⁵ Due to its crucial role in the life cycle, 23 commercial carboxamide fungicides targeting SDH have been launched to date. Among them, the sale of pyrazole carboxamide fungicides significantly promoted market development of SDH inhibitors (SDHIs) due to high fungicidal activity and broad-spectrum properties.⁶ However, according to the results from the Fungicide Resistance Action Committee (FRAC), many plant pathogens have developed resistance toward existing SDHI fungicides due to their longterm abuse and high usage rate.⁷⁻¹⁰ Therefore, it is particularly important and urgent to design new SDHI fungicides.

Over the past several decades, some new techniques, such as high-performance computation, structure biology, and artificial intelligence, have been widely applied in structure-based drug and pesticide discovery.^{11–14} Fragment-based drug discovery (FBDD) has been recognized as a successful method for hit identification and lead conception. Then, FBDD has been successfully used in many systems and yielded very promising results.^{15,16} Some articles have summarized fragment-to-lead (F2L) success stories published during 2017 and 2018.^{17,18} The core step in the FBDD process is to identify fragments with a low molecular weight (<300 g/mol) that weakly bind the target protein.¹⁹ Due to the low binding affinity of fragments, they are usually difficult to detect using bioassay-based screening methods, such as high-throughput X-ray crystallography, nuclear magnetic resonance (NMR), and

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surface plasmon resonance (SPR).^{20,21} Therefore, a variety of alternative biophysical methods have been used to detect the binding of such fragments.²² In our laboratories, we developed a methodology for screening fragments, known as pharmacophore-linked fragment virtual screening (PFVS), which identifies high potent inhibitors for the bc_1 complex and SDH.^{23,24} Recently, our group also identified potent and bioselective inhibitors for protoporphyrinogen oxidase using a fragment deconstruction method.²² The common characteristic of PFVS and fragment deconstruction methods is fragment identification and linking. The hypothesis for fragment linking is that the binding mode of the fragments is conserved upon modification of the fragment, and when two fragments are linked together, their orientation in their respective binding district should remain the same.^{22,25}

In this study, we report the molecular design of new SDHIs using a fragment identification and recombination strategy, which includes five steps: (1) to uncover the ligand-receptor interaction mechanisms of four representative SDHIs, including pyraziflumid, flubeneteram, pydiflumetofen, and fluopyram (Figure 1), by integrating molecular docking, molecular dynamics, and binding energy calculations; (2) to identify three active fragments, pyrazine from pyraziflumid, diphenyl-ether from flubeneteram, and prolong amide linker from pydiflumetofen and fluopyram, which occupied three different sub-pockets in the SDH binding site; (3) to combine three active fragments and produce four virtual compounds

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Figure 1. Design of virtual compounds. Pyraziflumid, pydiflumetofen, and fluopyram are commercial SDH fungicides. Flubeneteram is a candidate SDH fungicide and is not currently on the market. Virtual-1, virtual-2, virtual-3, and virtual-4 were designed based on the above compounds with different amine bond lengths.

(virtual-1, virtual-2, virtual-3, and virtual-4); (4) to synthesize a series of pyrazine-carboxamide-diphenyl-ether scaffold based on virtual-3 and virtual-4 binding modes and evaluate their structural-activity relationship *in vitro* and *in vivo*; and (5) one hit compound **6y** was identified with an IC₅₀ value of 0.83 μ M against porcine SDH, which was about 2-fold more potent than pyraziflumid (IC₅₀ = 1.52 μ M). Further biological assays showed that compound **6y** had excellent fungicidal potency toward *Botryotinia fuckeliana in vitro* and soybean gray mold *in vivo*. Computational simulations revealed that compared to the commercial fungicide pyraziflumid, compound **6y** showed a more favorable VDW interaction with SDH, which led to its higher activity. The present study provides an example of an application of computational design of pesticide molecules.

MATERIALS AND METHODS

Chemistry. All reagents and solvents were commercially available and used directly without further purification. Reactions were monitored by thin-layer chromatography (TLC). Target compounds were purified by column chromatography using silica. The ¹H NMR (600, 500, or 400 MHz) and ¹³C NMR (125 MHz) spectra were recorded on a Bruker Avance 500 spectrometer (Bruker Inc., Billerica, MA), Mercury Plus 400 MHz, or 600 MHz spectrometer (Varian, Palo Alto, CA) in DMSO- d_6 or CDCl₃ solution, with SiMe₄ (TMS) as the internal standard. Mass spectrometry (MS) data were obtained with a DSQII GC-MS (Thermo Fisher, Austin, TX) instrument with an electrospray ionization (ESI) source. High-resolution mass spectra (HRMS) were determined with an Agilent 6224 time-of-flight liquid chromatograph mass spectrometer (Agilent Technologies, Santa Clara, CA), which was equipped with a 250 mm \times 4.6 mm i.d., 5 µm, Eclipse XDB-C18 (Agilent Technologies, Santa Clara, CA) column. The melting points were determined on a BüCHI B-545 melting point apparatus and were uncorrected. Detailed synthetic routes and characterization data for all synthesized compounds are given in the Supporting Information.

Molecular Model. The 3D structures of the newly synthesized compounds were built using SYBYL and then minimized with the steepest descent method and conjugate gradient method, both with 2000 steps with a convergence criterion of 0.001 kcal/mol/Å.²⁶ The crystal structure of porcine SDH (PDB ID: 1ZOY)² was chosen as the receptor.

Molecular docking was performed using Autodock4.2.²⁷ The grid center was set according to the UBIQUINONE ligand in 1ZOY. The grid box was set as $42 \times 34 \times 50$, and the grid space was set at 0.375 Å. The default values were used for all other parameters. The 256 conformations were acquired with the conformation search method of the Lamarckian Genetic Algorithm (LGA)²⁷ and then subjected to further energy minimization and short molecular dynamics (MD) simulation using Amber16.^{28,29} First, the ligand was minimized with the protein fixed. Second, only the backbone atoms of the protein were fixed. Last, minimization was carried out with all the atoms without any constraint. Each step of minimization was performed by means of 1000 cycle steepest descent and 2000 cycle conjugate gradient method, with a convergence criterion of 0.01 kcal/mol/Å. The energy minimization was followed by an additional 20 ps MD simulation, which was carried out at 300 k by applying the Langevin thermostat. The coordinate file was recorded every 1 ps. The last frame of the MD simulation was minimized with the backbone atoms of proteins fixed. Finally, the minimized complex structure was used to calculate the binding energy, making use of the molecular mechanics Possion-Boltzmann surface area (MM/PBSA) method.²⁴ The final structure was selected based on the binding energy and binding mode of commercial carboxamide fungicides obtained from our previous study.³

Enzymatic Assay. The preparation of SDH from a porcine heart was the same as previously reported.³¹ The enzymatic activities of SDH were analyzed in a reaction mixture as reported previously.³² The inhibition rates of 26 samples were measured at a 10 μ M concentration. To test the IC₅₀ value, reactions were carried out in the presence of varying concentrations of the inhibitor.^{33,34} The commercial fungicide pyraziflumid was chosen as a positive control.

In Vitro Fungicidal Activity. Compounds were evaluated in mycelia growth inhibition tests against *Phytophthora capsici, Pythium dissimile, B. fuckeliana, Gibberella zeae,* and *Zymoseptoria tritici* (rich media) in artificial media and *Uromyces viciae-fabae* on bean leaf disks.³⁵ All testing was done by Syngenta AG and undertaken on 96-well microtiter plates. Each testing was carried out in triplicate.³⁶ Three commercial fungicides were chosen as positive controls.

In Vivo Fungicidal Activity. According to the published pesticide bioassay testing method in Shenyang Sinochem Agrochemicals R&D Co., Ltd. (Shenyang, China),³⁷ 13 target compounds were evaluated for protective activities in the greenhouse against wheat powdery mildew and soybean gray mold. The commercial fungicide pyraziflumid was chosen as a positive control.



Figure 2. (A and B) Binding mode of pyraziflumid with SDH. Projection_1 and projection_2 in B are the projections of the inner phenyl in pyraziflumid. (C) Relative positions of pyraziflumid (green line), pydiflumetofen (magenta stick), and fluopyram (yellow stick) when they bound with the SDH binding site. (D) Relative positions of pyraziflumid, pydiflumetofen, fluopyram, and flubeneteram bound with the SDH binding site. For clarity, the pyrazine ring (green stick) in pyraziflumid, prolonged amide bond (gray and yellow stick) in pydiflumetofen and fluopyram, and diphenyl-ether (magenta stick) in flubeneteram. (E) Binding mode of compound **6**y with SDH. (F) Overlay of pyraziflumid (yellow stick) and **6**y (magenta stick) when bound with SDH. Only the key residues are shown. The red and green lines represent an Hbond. All these figures were built by a Pymol program.

RESULTS AND DISCUSSION

Fragment Identification and Recombination. Finding novel fragments as starting points for optimization is a major challenge in FBDD. The structures of the fragments binding to the protein can be used to design new compounds with increased affinity and novelty. The chemical structure of commercial SDH fungicides always consists of three fragments, including an acid moiety, hydrophobic side moiety, and amide linker, indicating three fragment directions for designing novel SDHI.^{21,38} According to our previous study, an acid moiety bound to the bottom of the SDH binding site formed a cation- π interaction with C R46, an amide linker located at the mouth of the SDH binding site formed a hydrogen bond (Hbond) with B_W173 and D_Y91, and a hydrophobic side moiety extended outside of the SDH binding site formed hydrophobic interaction with some residues, such as C W35, D_Y91, and C_I30, among others (Figure 1S, Supporting Information).³⁰

Some active fragments from commercial fungicides are good choices for designing SDHIs. The pyrazine-carboxamide fungicide pyraziflumid with structural novelty was developed by Nihon Nohyaku Co., Ltd. in 2017 with excellent biological performance and no cross resistance with the existing fungicides.³⁹ The main difference between pyraziflumid and other SDH fungicides is its pyrazine ring in the acid moiety. Understanding the binding mechanism of these highly potent inhibitors can help to uncover the binding "hot spots" and identify the regions contributing to activity. Hence, molecular

docking followed by molecular dynamics and MM/PBSA calculations was performed for pyraziflumid. As shown in Figure 2A, pyraziflumid showed a conserved binding mode with other carboxamide SDH fungicides. The pyrazine ring in pyraziflumid, similar to other acid moieties in SDHIs, bound to the bottom of the SDH binding site and formed a cation- π interaction with the residue of C R46. The carbonyl oxygen in pyraziflumid formed an Hbond with D_Y91 and B_W173, its biphenyl moiety extended toward the entrance of the SDH binding site, then its inner phenyl bounded by C I30, C I43, and C W35, and its tail phenyl formed the edge-to-edge $\pi - \pi$ interaction with D Y91. Here, we noticed that the ring plane of inner phenyl in pyraziflumid was misaligned with C I30 and C_W35 (Figure 2B), which could be further structurally optimized to increase its activity. Collectively, these results indicated that a pyrazine ring as an acid moiety would be a good starting point to design new SDHIs.

Moreover, in our previous study, the diphenyl-ether in flubeneteram was proven as a good fragment extending toward the entrance of the SDH binding site, forming a $T-\pi$ interaction with the indole ring of C_W35.^{24,40} To continue our research, the diphenyl-ether was selected as the hydrophobic side fragment in this work. Here, we noticed that the distance between the pyrazine ring and diphenyl-ether in the SDH binding site was about 5.1 Å, which was larger than one amide bond length (3.8 Å) (Figure 2S, Supporting Information).

For the linker between the pyrazine ring and diphenyl-ether, the prolonged amide bond with one or two methylene groups

Scheme 1. Synthetic Route for the Target Compounds $6a-6z^{a}$



"Reagents and conditions: (a) K₂CO₃, DMF, reflux; (b) NH₂OH·HCl, EtOH, reflux; Zn/HCl, room temperature; (c) **5**, HATU/DIPEA, CH₂Cl₂, room temperature.

Table	1. Chemical	Structures,	Inhibitory	Activities	against	Porcine	SDH,	and the	Binding	Energy	between	Compound	ls with
SDH (kcal/mol)												

no.	n	R	inhibitory rate (%)/IC ₅₀ $(\mu M)^a$	$\Delta E_{ m vdw}$	$\Delta E_{ m ele}$	$\Delta G_{ m pol}$	$\Delta G_{ m np}$	$\Delta G_{ m cal}$	ΔG_{exp}
6a	1	2-F	23.61%	ь					
6b	1	2-Br	48.43%						
6c	1	2-Me	29.27%						
6d	1	3-Cl	31.42%						
6e	1	4-Me	32.45%						
6f	1	4-Br	32.85%						
6g	1	2,3-F ₂	26.16%						
6h	1	2,3-Cl ₂	59.68%						
6i	1	2-Cl-4-F	54.83%						
6g	1	2,4-Cl ₂	43.27%						
6k	1	2-Cl-4-CF ₃	63.37%	-36.44	-24.27	45.12	-3.80	-19.39	
61	1	3,4-F ₂	31.34%						
6m	1	3,4-Cl ₂	52.45%						
6n	0	2-F	4.79 ± 0.12	-39.86	-19.45	37.98	-3.85	-25.19	-7.30
60	0	2-Cl	1.25 ± 0.12	-40.83	-20.55	36.78	-3.82	-28.42	-8.10
6p	0	2-Br	3.04 ± 0.01	-40.99	-18.91	37.77	-3.86	-25.99	-7.57
6q	0	4-F	5.40 ± 0.12	-39.92	-17.67	36.17	-3.60	-25.01	-7.23
6r	0	2,3-F ₂	3.40 ± 0.01	-40.62	-22.59	40.59	-3.84	-26.45	-7.50
6s	0	2-Cl-3-F	2.72 ± 0.12	-40.42	-20.94	37.63	-3.62	-27.34	-7.64
6t	0	2,3-Cl ₂	0.90 ± 0.01	-42.40	-17.70	34.12	-3.78	-29.76	-8.30
6u	0	3,4-F ₂	4.18 ± 0.11	-40.31	-20.25	38.49	-3.78	-25.85	-7.38
6v	0	2,4-Cl ₂	1.14 ± 0.12	-40.44	-20.63	36.02	-3.72	-28.77	-8.15
6w	0	2-Cl-4-F	2.67 ± 0.11	-38.30	-18.39	33.03	-3.43	-27.08	-7.65
6x	0	2-Cl-4-CF ₃	1.23 ± 0.12	-41.73	-16.59	33.65	-3.75	-28.42	-8.11
6y	0	2-Br-4-Cl	0.83 ± 0.01	-45.88	-19.80	38.80	-3.94	-30.83	-8.34
6z	0	2-Cl-5-F	3.49 ± 0.12	-40.01	-18.59	36.01	-3.57	-26.16	-7.49
pyrazifl	umid		1.52 ± 0.11	-41.68	-19.77	36.57	-3.49	-28.37	-7.98
lanı		1.1	6 1	76.1 . 1.1		=00(1	1 10	1 111	1 1.

^{*a*}The compound inhibitory rate was first tested at 10 μ M concentration. If the inhibitory rate was over 70%, then the IC₅₀ value would be calculated with eight different concentration gradients. ^{*b*}Not tested.

was taken into consideration. Here, the commercial fungicides, pydiflumetofen and fluopyram, with different novel prolonged strategies for the amide bond (Figure 1), were selected as representative compounds to study the interaction with the target SDH.^{41,42} As shown in Figure 3S (Supporting Information), both of them formed an Hbond with B_W173 and D_Y91 and a cation $-\pi$ interaction with C_R46. Compared to the classical amide bond linker, the N-alkyl

amide linker in pydiflumetofen is also located at the mouth of the SDH binding site, and then the difference was that the prolonged amide linker in pydiflumetofen resulted in its substituted phenyl ring extending further outside of the binding site to form $\pi - \pi$ interaction with C_W35 rather than pyraziflumid (Figure 2C). The same thing occurred to the ethanamine linker in fluopyram (Figure 2C). These results indicated that the prolonged amide bond could adjust the interaction of the hydrophobic side moiety as an inhibitor with SDH. Our previous study showed that as more interaction occurs between the hydrophobic side moiety in the inhibitor and SDH, the stronger the activity of the inhibitor.³²

Subsequently, four virtual compounds (virtual-1, virtual-2, virtual-3, and virtual-4) were designed based on the relative position of the pyrazine ring, diphenyl-ether, and prolonged amide bond in the SDH binding site (Figure 2D). Then, four of them were docked to the SDH binding site. As shown in Figure 4S, the ring plane of the pyrazine fragment in virtual-1 did not form a good cation– π interaction with C_R46, and virtual-2 just formed one Hbond with B_W173. However, virtual-3 and virtual-4 showed the conserved binding modes with that of the commercial fungicides, indicating that they could be selected as the new starting points for SDHIs and subjected to subsequent synthesis. Meanwhile, the synthesis processes of virtual-3 and virtual-4 were simpler than those of virtual-1 and virtual-2 as described in the following Chemistry section.

Chemistry. To enrich the structural-activity relationship (SAR) of virtual-3 (equal to 6k) and virtual-4 (equal to 6x), a series of pyrazine-carboxamide-diphenyl-ether compounds (6a-6z) were synthesized. According to Scheme 1, compounds 6a-6m were synthesized with yields of 37-66% by only three steps. The substituted phenol 2 and 2'fluoroacetophenone 1 were selected as the starting materials and then employed under the presence of K₂CO₃ in DMF to produce intermediate 3.43 Intermediate 3 reacted with NH₂OH·HCl to afford oxime and then was restored to key intermediate amine 4 under a zinc duct.⁴⁴ Finally, compound 5, 3-(trifluoromethyl)pyrazine-2-carboxylic acid, was obtained according to reported methods³⁹ and then reacted with amine 4 to afford the target compounds 6a-6m by the condensing agent HATU/DIPEA catalyst. The compounds 6n-6z were synthesized with 2-fluorobenzaldehyde as a starting material, and their yields were about 32-62%.

Structural–Activity Relationship. As shown in Table 1, **6a–6m** showed weak enzymatic inhibition activity with an inhibitory rate ranging from 23.61 to 63.37% against porcine SDH at 10 μ M. To understand its low activity, compound **6k** was redocked followed by MM/PBSA calculations. The binding energy with SDH was –19.39 kcal/mol, which was larger than the binding energy of pyraziflumid with SDH (-28.37 kcal/mol). Here, we noticed that the VDW interaction energy of **6k** with SDH was –36.44 kcal/mol (Table 1), which was lower than pyraziflumid (-41.68 kcal/ mol) and may explain its low activity.

However, the activity of compounds 6n-6z showed significant improvement over 6a-6m. Moreover, the IC₅₀ values of compounds 6o, 6t, 6v, and 6x were 1.25, 0.90, 1.14, and 1.23 μ M, respectively, which showed higher activities than pyraziflumid (IC₅₀ = 1.52 μ M). Remarkably, compound 6y, bearing 2-Br-4-Cl, displayed the highest activity with an IC₅₀ value of 0.83 μ M, with approximately 2-fold improved potency compared with pyraziflumid. The SARs of compounds 6n-6z are summarized as follows: (1) Monosubstituted compounds, a 2-position substituent, had a positive effect on the activities compared to a 4-position substituent. For example, 6n (2-F, IC₅₀ = 4.79 μ M) showed a higher activity than 6q (4-F, IC₅₀ = 5.40 μ M). (2) Disubstituted compounds always had better activity than monosubstituted compounds. (3) For 2,4-disubstituted compounds, 6v showed a higher

activity (2,4-Cl₂, IC₅₀ = 1.14 μ M) than **6x** (2-Cl-4-CF₃, IC₅₀ = 1.23 μ M) and **6w** (2-Cl-4-F, IC₅₀ = 2.67 μ M).

To further understand SAR, molecular docking was used to explore the inhibition mechanism of compounds 6n-6z. As shown in Table 1, the correlation coefficient (R^2) between ΔG_{cal} (calculated binding energies) and ΔG_{exp} ($\Delta G_{exp} = -RT$ \times ln IC₅₀) was high, up to 0.95, which indicated that the binding mode was reasonable and reliable. The binding mode of the representative compound 6y (Figure 2E) was similar to pyraziflumid, forming a cation $-\pi$ interaction with C R46 and a Hbond with B W173 and D Y91. The pyrazine ring in 6y showed nearly the same position as that in pyraziflumid. Different from pyraziflumid (Figure 2F), the amide bond extension in 6y led to its diphenyl-ether moiety extending toward the binding site entrance of SDH, forming a stronger sandwich hydrophobic interaction with C_I30 and C_I43 than pyraziflumid (Figure 2F). This resulted in a more favorable VDW interaction with SDH ($\Delta E_{\rm vdw}$ = -45.88 kcal/mol for compound **6y** vs -41.68 kcal/mol for pyraziflumid, Table 1). Then, we also noticed that the $\Delta E_{\rm vdw}$ values for compounds 6n-6z were around -38.30 to -42.40 kcal/mol, lower than 6k(-36.44 kcal/mol), indicating that one methyl introduction (*n* = 1) into the scaffold of the target was unfavorable for VDW interaction between the ligand and SDH.

As shown in Table 1, one interesting thing was that the increase in $\Delta E_{\rm ele}$ was not able to enhance the activity of the compound. For example, compound **6r** had the highest $\Delta E_{\rm ele}$ (-22.59 kcal/mol) but its IC₅₀ value (3.40 μ M) was not good. However, the influence of $\Delta E_{\rm vdw}$ was remarkable. In general, the stronger the $\Delta E_{\rm vdw}$ between the compound and SDH, the higher the activity of the compound, such as in compounds **6t**, **6k**, and **6y**. These results indicated that $\Delta E_{\rm vdw}$ is very important for designing new SDH inhibitors.

We note that the activity of compound **6y** was still lower than pydiflumetofen (0.13 μ M, data was not shown) and flubeneteram (0.11 μ M). Nevertheless, the scaffold of compound **6y**, especially for its amide bond with one methylene extension, provides new insight into designing new SDHIs.

In Vitro Fungicidal Activity. To determine whether the target compounds had fungicidal activity, compounds 6n-6z were assessed on six plant pathogens, namely, *Ph. capsici*, *Py. dissimile*, *B. fuckeliana*, *G. zeae*, *Z. tritici*, and *U. viciae-fabae*. The results are summarized in Table 2 and expressed as general assessment criteria for biological assays.

As shown in Table 2, all tested compounds did not show good fungicidal activity against *Ph. capsici*, *Py. dissimile*, *G. zeae*, *Z. tritici*, and *U. viciae-fabae*. In contrast, most compounds showed an over 80% inhibitory rate activity against *B. fuckeliana*, except **6x**, with a 62% inhibitory rate at 20 mg/L. Furthermore, compounds **60–6r**, **6u**, **6y**, and **6z** had a 100% control effect against *B. fuckeliana*, which was better than pyraziflumid (90%). Compound **6y** also showed an 80% inhibitory rate against *Z. tritici* at a concentration of 20 mg/L, indicating its potential broad-spectrum property.

In Vivo Fungicidal Activities. It is known that gray mold caused by *B. fuckeliana* can cause heavy losses in many crop yields worldwide, such as grapes, soybean, cucumber, and strawberry.⁴⁵ Here, soybean gray mold (SGM) and wheat powdery mildew (WPM) were selected as the target diseases to evaluate the biological activity of the tested compounds in the greenhouse, and the commercial fungicide pyraziflumid was selected as a positive control. As shown in Table 3, most

Table 2. Fungicidal Activity of Target Compounds In Vill	Table	2. Fungicida	Activity	of Target	Compounds	In	Vitr
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	Uvf ¹²	Pd ^a	Bf	Gz ^a	Pc ^a	Zť
no.	100 ^b	2 ^b	20 ^b	20 ^b	20 ^b	20 ^b
6n	0 ^c	0	95 ± 1	0	0	0
60	0	0	100	0	0	0
6p	0	0	100	0	0	0
6q	0	0	100	0	0	0
6r	0	0	100	0	0	0
6s	0	0	90 ± 1	0	0	0
6t	0	0	85 ± 2	0	0	0
6u	0	0	100	0	0	27 ± 2
6v	0	0	90 ± 1	0	0	0
6w	0	0	80 ± 2	0	0	0
6x	0	0	62 ± 1	0	0	0
6у	0	0	100	0	0	80 ± 1
6z	0	0	100	0	0	0
pyraziflumid	100	0	90 ± 1	0	0	0
azoxystrobin	100	100	100	0	100	100
prochloraz	0	0	100	100	0	100

^{*a*}Uvf, *U. viciae-fabae*; Pd, *Py. dissimile*; Bf, *B. fuckeliana*; Gz, *G. zeae*; Pc, *Ph. capsici*; Zt, *Z. tritici.* ^{*b*}Dose in mg/L. ^{*c*}The data are the mean of three replicates.

compounds did not show good activity against WPM. As expected, for SGM at 100 mg/L, compounds **6p**, **6r**, **6u**, and **6z** showed over a 80% control effect, and compounds **6o**, **6q**, and **6y** showed over a 90% control effect, all of which were better than pyraziflumid (70%). Notably, compound **6y** had a 95% control effect against SGM and an 80% control effect against WPM at a concentration of 100 mg/L. Compound **6q** had nearly the same control effect against SGM and WPM as **6y**. Then, the EC₅₀ and EC₉₀ were tested against SGM for some potential compounds. The results are shown in Table 4 (Table 1S, Supporting Information), and we can see that compounds **6y** and **6q** exhibited better EC₅₀ and EC₉₀ values than pyraziflumid. These results indicated that compounds **6y** and **6q** have potential broad-spectrum features and thus may be used as candidates for further development.

In summary, we implemented a fragment recombination strategy for new SDHI design. Computational analysis demonstrates that three active fragments, pyrazine ring, diphenyl-ether, and prolonged amide bond, occupied three different sub-pockets in the SDH binding site and could be combined to a novel scaffold of SDHI. After substitute optimizations, a hit compound **6y**, N-(2-(2-bromo-4-chlorophenoxy)benzyl)-3-(trifluoromethyl)pyrazine-2-carboxamide, was identified with an IC₅₀ value 0.83 μ M, which was about 2-fold more potent than pyraziflumid. Biological experiments showed that compounds **6q** and **6y** had 100%

Гable 3. Fungicida	l Activity	v of Tar	get Com	pounds In	Vivo
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Table 4.	. EC ₅₀	and EC ₉	0 Values	(mg/L)	for Target
Compoi	inds a	gainst So	vbean G	rav Mole	đ

no.	EC ₅₀	EC ₉₀	regression equation	r
6p	76.51	139.74	-4.2278 + 4.8987x	0.94
60	72.04	128.97	-4.4124 + 5.0671x	0.95
6у	52.28	87.61	-4.8227 + 5.7163x	0.93
6q	57.24	96.90	-4.8523 + 5.6052x	0.94
pyraziflumid	60.96	107.06	-4.3515 + 5.2389x	0.91

control effects against *B. fuckeliana* at a dosage of 20 mg/L *in vitro* and over 90% control effects against SGM at 100 mg/L *in vivo*, which were also better than that of pyraziflumid. Further EC_{50} and EC_{90} values also indicated that both of them were superior to pyraziflumid. The computational simulations revealed that the van der Waals interactions between compounds and SDH played an important role in adjusting its activity, which may provide insight into designing new SDHIs. The above results also indicated that the amide bond with one methylene extension is a novel active fragment for designing SDHIs.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.jafc.0c05646.

Relative position, binding mode, decomposed VDW interaction energy, control effect picture, enzymatic assay, *in vitro* and *in vivo* fungicidal assay, leaf-piece assay, synthetic routes, and ¹H and ¹³C NMR spectrograms (PDF)

AUTHOR INFORMATION

Corresponding Authors

- Xiao-Lei Zhu Key Laboratory of Pesticide & Chemical Biology of Ministry of Education, International Joint Research Center for Intelligent Biosensor Technology and Health of Ministry of Science and Technology, Central China Normal University, Wuhan 430079, People's Republic of China;
 orcid.org/0000-0002-5672-5209; Email: xlzhu@ mail.ccnu.edu.cn
- Guang-Fu Yang Key Laboratory of Pesticide & Chemical Biology of Ministry of Education, International Joint Research Center for Intelligent Biosensor Technology and Health of Ministry of Science and Technology, Central China Normal University, Wuhan 430079, People's Republic of China; Collaborative Innovation Center of Chemical Science and Engineering, Tianjin 300071, People's Republic of China; Phone: 86-27-67867800; Email: gfyang@ mail.ccnu.edu.cn; Fax: 86-27-67867141

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no.	SGM^a (100 mg/L)	WPM ^{a} (100 mg/L)	no.	SGM^a (100 mg/L)	WPM ^{a} (100 mg/L)
6n	0	NT^{b}	6u	80	20
60	90	30	6v	0	NT
6p	85	40	6w	20	NT
6q	90	85	6x	10	NT
6r	85	30	бу	95	80
6s	20	NT	6z	80	0
6t	0	NT	pyraziflumid	70	98

^aSGM, soybean gray mold; WPM, wheat powdery mildew. ^bNot tested.

Authors

- Hua Li Key Laboratory of Pesticide & Chemical Biology of Ministry of Education, International Joint Research Center for Intelligent Biosensor Technology and Health of Ministry of Science and Technology, Central China Normal University, Wuhan 430079, People's Republic of China
- Meng-Qi Gao Key Laboratory of Pesticide & Chemical Biology of Ministry of Education, International Joint Research Center for Intelligent Biosensor Technology and Health of Ministry of Science and Technology, Central China Normal University, Wuhan 430079, People's Republic of China
- Yan Chen Key Laboratory of Pesticide & Chemical Biology of Ministry of Education, International Joint Research Center for Intelligent Biosensor Technology and Health of Ministry of Science and Technology, Central China Normal University, Wuhan 430079, People's Republic of China
- Yu-Xia Wang Key Laboratory of Pesticide & Chemical Biology of Ministry of Education, International Joint Research Center for Intelligent Biosensor Technology and Health of Ministry of Science and Technology, Central China Normal University, Wuhan 430079, People's Republic of China

Complete contact information is available at: https://pubs.acs.org/10.1021/acs.jafc.0c05646

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Notes

The authors declare no competing financial interest.

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