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Indium-promoted Barbier-type allylations in aqueous media: a convenient approach to 4-C-branched monosaccharides and $(1\rightarrow 4)$ -C-disaccharides

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Abstract—Starting from methyl 6-bromo-4,6-dideoxy- α -D-*threo*-4-enopyranoside, 4-*C*-branched sugars have been prepared through indium-promoted Barbier-type allylation of various aldehydes in aqueous media followed by hydroboration of the resulting double bond. The intermediate unsaturated monosaccharides were shown to rearrange in acidic media to give 4-*C*-acetyl-5-*C*-alkyl pyranose compounds. From β -1-formyl sugars the corresponding β -(1 \rightarrow 4)-*C*-disaccharides were obtained. © 2004 Elsevier Ltd. All rights reserved.

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1. Introduction

The stereospecific formation of carbon–carbon bonds is currently the subject of intense research. In carbohydrate chemistry, the regio- and stereoselective synthesis of branched sugars has mainly been stimulated by their occurrence in a large variety of natural products.¹ Moreover, relatively few methods of general preparative significance are available for the synthesis of such compounds.²

Recently, we reported a very convenient protocol for the preparation of C-branched monosaccharides and C-disaccharides under indium promoted Barbier-type allylations in aqueous media.³ Indeed because indiummediated reactions can be carried out in such media, they were shown to be particularly suited for carbohydrates and their analogues.⁴

In order to extend our preliminary approach, which started from 2-bromo-3-enopyranoside and 4-bromo-2enopyranoside derivatives, we now apply this methodology to 6-bromo-4-enopyranoside, which allows access to various 4-C-branched monosaccharides and $(1\rightarrow 4)$ -C-disaccharides.

2. Results and discussion

The synthesis of the key compound for the allylation reaction, 6-bromo-4,6-dideoxy- α -D-threo-4-enopyranoside 7 is summarized in Scheme 1. We started from commercial methyl α -D-glucopyranoside 1. After quantitative acetvlation under standard conditions (acetic anhydride in pyridine, overnight), the peracetylated derivative 2 was treated with lipase from Candida cylindracea in a mixture of phosphate buffer (0.1 M; pH7.0) and di-n-butyl ether, to give methyl 2,3,4-tri-O-acetyl- α -D-glucopyranoside **3** in 97% yield.⁵ In the latter reaction, we observed, as reported, complete and regioselective hydrolysis of the primary acetyl ester. Then, PCC oxidation⁶ gave directly the unsaturated aldehyde 4^7 in 78% yield provided that the reaction was performed on a small scale. On a larger scale, we preferred the Swern oxidation (dimethyl sulfoxide/oxalyl chloride/triethylamine), which gave 4 in 91% reproducible yield.⁸

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Scheme 1. Reagents and conditions: (a) Ac₂O, pyridine (quantitative yield); (b) lipase from *C. cylindracea*, phosphate buffer (0.1 M, pH7.0)/Bu₂O (97%); (c) Me₂SO, (CO)₂Cl₂, Et₃N, CH₂Cl₂ (91%); (d) NaBH₄, MeOH (91%); (e) CBr₄, (C₆H₅)₃P, CH₂Cl₂ (93%); (f) LiOH, MeOH/water (97%, overall yield 72%).

Then, the unsaturated aldehyde **4** was reduced with NaBH₄ in MeOH to provide the allyl alcohol **5** in 91% yield. Finally, under standard bromination conditions (triphenylphosphine, carbon tetrabromide) followed by deacetylation with lithium hydroxide in a 3:1 mixture of MeOH–water (97%), we obtained the desired 6-bromo-4-enopyranoside **7** (72% overall yield for the six steps from **1**).

Then we tested the allylation reactions with various aldehydes under indium-promoted Barbier-type conditions. All the results are summarized in Table 1.

First, we tried the reaction with benzaldehyde. It took place in a mixture of 2:1 THF-phosphate buffer (0.11 M; pH7.0) at rt for 1.5 h to give the adduct **8** in 86% yield as a unique stereoisomer resulting from complete regioand diastereoselectivity (Table 1, entry 1). The structure of **8** was deduced from NMR analysis. The ¹H NMR spectrum notably exhibits a doublet of doublet ($J_{3,4}$ 8.5 Hz and $J_{4,7}$ 4.0 Hz) for H-4, indicating equatorial alkylation at C-4. In ¹³C NMR spectroscopy, the presence of a quaternary carbon atom and a secondary carbon atom in the olefinic shift area are in good agreement with an exocyclic double bond in C-5,6 position. We obtained along with compound **8**, a small amount (9%) of **26**, which resulted from the reduction in water of the starting bromide with indium. Indeed, when **7** was treated with indium(0), alone without any aldehyde in a 1:1 THF–phosphate buffer (0.05 M; pH7.0)/THF mixture, we obtained in 1 h **26** in 73% yield. The ¹³C NMR spectrum of **26** shows clearly an exocyclic double bond at C-5 (97.2 and 153.1 ppm) and a methylene group at C-4 (36.2 ppm). It is noteworthy that this reduction, which occurred to a small extent in the case of the primary bromide **7**, did not occur at all when using a secondary bromide.³

If the reaction is conducted in pure water, in the absence of the phosphate buffer, the 4-C-adduct 8 is slowly transformed (30% after 1.5h) into two new products $9\alpha,\beta$ as a 1:3 mixture of the two α and β anomers. These compounds $9\alpha,\beta$ were fully characterized by spectral analysis. For both compounds, ¹H and ¹³C NMR spectroscopy indicates the presence of the 5-C-phenyl moiety in equatorial orientation instead of the H-6 and H-6' protons and a 4-C-acetyl group in axial orientation. Acetylation of $9\alpha,\beta$ led to $10\alpha,\beta$ whose NMR data confirmed the structure. Indeed, we observed a

Table 1.	Reactions	of 7	with	various	aldehydes
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Entry	RCHO	Products	Cond ^b	Yield ^c (%)	Eq./ax. ^d	Dr: S/R ^d
1	PhCHO	8	1.5	86	1/0	~100/0
2	HCHO	11a,b	3	73	0.66/0.34	
3	13	14	1.5	57	1/0	$\sim \! 100/0$
4	18	19a,b	2	45	0.64/0.36	~100/0

^a RCHO (3equiv), In (2equiv), 2:1 THF-phosphate buffer (0.11 M, pH7.0).

^b Conditions: reaction time (h); rt.

^c Isolated yields.

^d Diastereomeric ratio determined within the accuracy of ¹H and ¹³C NMR spectra.



Scheme 2. Reagents and conditions: (a) water, PhCHO (3equiv), In (2equiv); 8, 5%; 9α , β (α / β : 1:3, 30%) or 2:1 THF–phosphate buffer (0.11 M, pH7), PhCHO (3equiv), In (2equiv); 8, 86%; 9α , β , 0%; (b) Ac₂O, py.

doublet of doublet for H-3 ($J_{3,4}$ 6Hz), which confirmed the 4-*C*-acetyl group in axial orientation and a doublet for H-5 ($J_{3,5}$ 3.5Hz) indicating the presence of the phenyl group in equatorial orientation.

In fact, without buffering the reaction mixture, $9\alpha,\beta$ arise from the acid hydrolysis of the labile enol ether derivative **8** as shown in Scheme 2. After electrophilic addition of water followed by opening of the ring with elimination of the anomeric methoxy group, the cyclization of the C-7 hydroxyl group onto the aldehyde led to the keto-derivatives $9\alpha,\beta$. In order to confirm these results, compound **8** was treated by 0.1 M aq HCl. Effectively, in these conditions, $9\alpha,\beta$ was obtained in 90% yield in 5 min as a mixture of anomers (α/β , 1:3).

Interestingly, the 4-*C*-5-*C*-substituted D-galacto-like structure of $9\alpha,\beta$ confirms the alkylation at C-4 in equatorial orientation and the configuration of the second new stereogenic center at C-7 in 8. Indeed, C-4 in $9\alpha,\beta$ (or $10\alpha,\beta$) arises from C-7 in 8. On the basis of these results, configuration at C-7 was assigned as (*S*).

From a mechanistic point of view, our results can be explained by the formation of the currently accepted six-membered cyclic transition state between the carbonyl compound and the allylindium sugar moiety. More-



Chart 1. Transition states between the carbonyl compound and the allylindium sugar moiety.

over, in this case, two six-membered transition states seem to be possible, a 'trans-decalin' state A and a 'cisdecalin' state B (Chart 1). Generally, trans-decalin is more stable than its cis-isomer; accordingly, the alkylation at C-4 in equatorial orientation in **8** is in favor of the formation of the 'trans-decalin' state A. Moreover, the C-7 (S) configuration indicates that the phenyl group is in the more favorable equatorial orientation in the transition state as depicted in Chart 1A.

Then in a second step, the reaction was extended to others aldehydes. In the case of formaldehyde (Table 1, entry 2), we obtained at rt in 3h, a mixture of axial and equatorial C-4 adducts **11a,b** (73% yield in a 2:1 ratio in favor of the equatorial adduct) and the reduced compound **26** (7%). Compound **11a,b** were fully characterized after peracetylation to **12a,b**. The presence of the two epimers at C-4 in **11a,b** can be explained by the formation of the two six-membered cyclic transition states A and B (Chart 1) and certainly reflects the low energy difference between these two states in the case of formal-dehyde, resulting from a decrease in the steric hindrance compared with the reaction with benzaldehyde.

Next, we examined the reaction between 7 and 3-O-benzyl-1,2-O-isopropylidene- α -D-xylo-dialdose 13.⁹ In this case, the reaction was less efficient than with benzaldehyde and the C-4 equatorial adduct was obtained at rt in 1.5h as a single diastereoisomer 14 in 57% yield (Table 1, entry 3) along with the reduced derivative 26 (18%). Compound 14 was fully characterized after peracetylation. Treatment of 14 by aq HCl followed by acetylation gave a 11:9 mixture of the two anomers $17\alpha,\beta$ in 81% yield, which confirmed that the configuration of the new stereogenic center C-7 in the starting compound 14 was (S). These results indicate clearly that in this case, as previously with benzaldehyde, the bulky substituent (furanose ring) is in the more favorable equatorial orientation in a 'trans-decalin'-like six-membered transition state.

Finally, to obtain analogues of *C*-oligosaccharides, we turned our attention to the reaction between the



6-bromo-enopyranoside derivative 7 and the β -C-linked 2,3,4,6-tetra-O-benzyl- β -D-glucosyl aldehyde **18**.¹⁰ The reaction gave an inseparable 16:9 mixture of the two adducts 19a,b in 45% yield along with 26 (30%). It is difficult at this stage to deduce the stereochemistry at C-4 from the ¹H NMR spectrum of the **19a,b** mixture as the coupling constants $J_{3,4}$ (6 Hz) were the same for both compounds. Therefore, the mixture was acetylated (16h, Ac₂O-pyridine) at rt to give the fully acetylated compound 20a and 20b in which the tertiary hydroxyl group remained unchanged, which allowed the separation at this stage. Then, **20b** was fully acetylated at 40°C for 24h to give 21b. However, it is still difficult to deduce from the ¹H NMR spectrum the stereochemistry at C-4 in 20a $(J_{3,4} 5.5 \text{ Hz})$ and 21b $(J_{3,4} 6 \text{ Hz})$. Compounds 20a and 21b were separately treated with 0.1 N aq HCl to give the rearranged products 22α , β and 24α , β , respectively, which were acetylated to $23\alpha,\beta$ and $25\alpha,\beta$ in which the perbenzyl glucosyl moiety was in equatorial orientation indicating a C-7 (S) configuration for both **20a** and **21b**. However, in 23α , β the coupling constant $J_{3,4}$ 6Hz for both anomers still did not allow to assign the configuration at C-4. However, a NOESY experiment showed clearly a strong NOE effect between H-4 and both H-3 and H-5 but not with H-2, which implied an equatorial H-4 and an axial methylketone. In 25α , β , the coupling constants $J_{3,4}$ 8 Hz for the α anomer and 10.5 Hz for the β anomer indicate an equatorial orientation for the methylketone substituent. This was confirmed through NOESY experiments, which showed strong effects between H-4 and H-2 but not with H-3 and H-5. From these correlations, it can be concluded that for the major stereoisomer **19a**, the alkylation operated at C-4 in equatorial orientation with C-7 (S)configuration. As previously, the bulkier substituent (pyranosidic ring) is in more favorable equatorial orientation with a 'trans-decalin' transition state. For the minor isomer 19b, the alkylation operated in C-4 axial position with C-7 (S) configuration, which implied a cis-decalin transition state in which the pyranosidic ring is in axial orientation. This could be explained by a chelation of the indium atom with an oxygen atom in 18



Chart 2. Possible transition states with chelation of indium atom in the reaction with 18.

(either O-3' or endocyclic oxygen), which cannot operate if the sugar is in equatorial orientation (Chart 2). We have in fact, already found such chelation in previous cases.³ This axial orientation of the bulkier substituent in the transition state can be correlated with the moderate yield (45%) obtained with the β -C-linked 2,3,4,6-tetra-*O*-benzyl- β -D-glucosyl aldehyde **18** in comparison to that of previous aldehydes.

In fact, all these allylation reactions led to unsaturated analogues of *C*-branched sugars or *C*-disaccharides, which need further elaboration to saccharides by hydroboration of the exocyclic double bond. As an example, we focused on the functionalization of **20a** and **21b** to disaccharides. Thus, hydroboration of **20a** with diborane–tetrahydrofuran followed by oxidation with hydrogen peroxide¹¹ in phosphate buffer (0.5 M; pH 7.0) and acetylation (Ac₂O–pyridine) led to a protected *C*-disaccharide derivative [D-Glc- β -(1 \rightarrow 4)-*C*-L-Ido] **27** as a single diastereoisomer, in 91% overall yield (Scheme 3).

The stereochemistry of the *C*-disaccharide **27** resulted from the attack of the electrophile onto the double bond on the opposite side of the C-1 and C-4 substituents. Moreover, steric hindrance of the 2,3,4,6-tetra-*O*-benzyl- β -D-glucosyl moiety could explain the complete diastereoselectivity of this reaction.

In the same way, hydroboration of compound **21b** with diborane-tetrahydrofuran followed by oxidation with hydrogen peroxide in phosphate buffer (0.5 M; pH 7.0) and then acetylation gave the protected *C*-disaccharide derivative [p-Glc- β -($1\rightarrow 4$)-*C*-p-Gal] **25** in 83% yield as a single diastereoisomer (Scheme 4). Here the formation of **25** could be explained by attack of the electrophile at C-5 on the opposite side of the 2,3,4,6-



Scheme 3. Reagents and conditions: (a) BH_3 , THF; (b) H_2O_2 , phosphate buffer (0.5 M, pH 7.0); (c) Ac_2O , pyridine; 91% overall yield.



Scheme 4. Reagents and conditions: (a) BH_3 , THF; (b) H_2O_2 , phosphate buffer (0.5 M, pH7.0); (c) Ac_2O , pyridine; 83% overall yield.

tetra-O-benzyl- β -D-glucosyl moiety located at the axial C-4 position.

In conclusion, we have reported an efficient synthesis of 4-C-branched monosaccharides and $(1\rightarrow 4)$ -C-disaccharides in aqueous media through indium-promoted Barbier-type allylations. Particularly, we have described an effective access to protected Glc- β - $(1\rightarrow 4)$ -C-Ido and Glc- β - $(1\rightarrow 4)$ -C-Gal derivatives from methyl α -D-glucopyranoside. The synthesis of these new compounds exemplifies the application of this recently developed method.

3. Experimental

3.1. General methods and materials

All moisture sensitive reactions were performed under argon using oven-dried glassware. If necessary, solvents were dried and distilled prior to use. Reactions were monitored on E. Merck Silica Gel 60 F254 plates. Detection was performed using UV light, iodine and/or 5% H₂SO₄ in EtOH, followed by heating. Flash chromatography was performed on Silica Gel 6–35 μ m. ¹H and ¹³C NMR spectra were recorded at rt with Bruker AC 200, 250, or AM 400 spectrometers. Chemical shifts are reported in δ relative to Me₄Si for ¹H and ¹³C NMR spectra (external reference for D₂O) and relative to the CDCl₃ resonance at 77.00 ppm for ¹³C NMR spectra in CDCl₃. Melting points were measured on a Reichert apparatus and were uncorrected. Optical rotations were measured on an Electronic Digital Jasco DIP-370 Polarimeter. Mass spectra were recorded in positive mode on a Finnigan MAT 95 S spectrometer using electrospray ionization. Elemental analyses were performed at the Service Central de Microanalyses du CNRS (Gif-sur-Yvette, France).

3.2. Methyl 2,3-di-*O*-acetyl-4-deoxy-α-D-*threo*-hex-4enopyranoside (5)

Sodium borohydride (0.120 g, 3.2 mmol) was added to a cooled (0 °C) soln of **4** (0.816 g, 3.2 mmol) in MeOH (20 mL). The mixture was stirred at 0 °C for 10 min, then

quenched with water, and extracted with CH₂Cl₂. The organic phase was dried (MgSO₄), then concentrated. Flash chromatography of the residue (1:1 petroleum ether–EtOAc) gave **5** (0.750g, 91%) as a colorless oil. $[\alpha]_D$ +277 (*c* 1.0, CH₂Cl₂); ¹H NMR (250 MHz, CDCl₃): δ 5.34 (m, 1H, H-3), 5.06 (dd, 1H, $J_{2,1}$ 2.5, $J_{2,3}$ 7 Hz, H-2), 4.98–4.92 (br s, 2H, H-1, H-4), 3.99 (m, 2H, H-6, H-6'), 3.47 (s, 3H, CH₃O), 2.07, 2.01 (2s, 6H, acetyl); ¹³C NMR (63 MHz, CDCl₃): δ 170.0 (C=O), 169.9, 152.6 (C-5), 97.2, 94.8 (C-1, C-4), 68.9, 66.6 (C-2, C-3), 60.8 (C-6), 56.0 (CH₃O), 20.5, 20.3 (acetyl). Anal. Calcd for C₁₁H₁₆O₇: C, 50.77; H, 6.20; O, 43.03. Found: C, 50.48; H, 6.17; O, 43.49.

3.3. Methyl 2,3-di-*O*-acetyl-6-bromo-4,6-dideoxy-α-D*threo*-hex-4-enopyranoside (6)

PPh₃ (0.906 g, 3.46 mmol) and CBr₄ (1.05 g, 3.17 mmol) were added to a cooled (-78°C) soln of 5 (0.750g, 2.9 mmol) in CH₂Cl₂ (30 mL). The suspension was stirred at 0 °C for 2h, then quenched with satd aq NaHCO₃, and extracted with CH₂Cl₂. The organic phase was dried (MgSO₄) then concentrated. Flash chromatography of the crude residue (1:4 petroleum ether-EtOAc) gave 6 (0.872 g, 93%) as white crystals: mp 89 °C (Et₂O); $[\alpha]_D$ +265 (c 1.1, CH₂Cl₂); ¹H NMR (250 MHz, CDCl₃): δ 5.45 (dd, 1H, J_{3,4} 2.9, J_{3,2} 7.5 Hz, H-3), 5.13 (m, 2H, H-4, H-2), 5.04 (d, 1H, J_{1.2} 2.5 Hz, H-1), 3.84 (m, 2H, H-6, H-6'), 3.58 (s, 3H, CH₃O), 2.12, 2.07, (2s, 6H, acetyl); ¹³C NMR (63 MHz, CDCl₃): δ 170.3 (C=O), 170.2, 148.9 (C-5), 99.8, 98.0 (C-1, C-4), 69.0, 66.9 (C-2, C-3), 56.8 (CH₃O), 29.5 (C-6), 21.0, 20.8 (acetyl). Anal. Calcd for C₁₁H₁₅BrO₆: C, 40.89; H, 4.68; O, 29.71. Found: C, 40.76; H, 4.55; O, 29.96.

3.4. Methyl 6-bromo-4,6-dideoxy-α-D-*threo*-hex-4-enopyranoside (7)

LiOH (0.453 g, 10.8 mmol) was added to a soln of 6(0.872g, 2.7mmol) in a 3:1 mixture of MeOH-water (24 mL). The suspension was stirred at rt for 5 min, then quenched with aq HCl (1.5M), and extracted with CH₂Cl₂. The organic phase was dried (MgSO₄) then concentrated. Flash chromatography of the residue (EtOAc) gave 7 (0.626 g, 97%) as a colorless oil, which decomposed slowly when kept at rt. ¹H NMR (250 MHz, CDCl₃): δ 5.09 (d, 1H, $J_{1,2}$ 2.5 Hz, H-1), 4.98 (d, 1H, $J_{4,3}$ 3.0Hz, H-4), 4.27 (dd, 1H, $J_{3,4}$ 3.0, J_{3,2} 7.5 Hz, H-3), 3.85 (m, 2H, H-6, H-6'), 3.71 (dd, 1H, J_{2,1} 2.5, J_{2,3} 7.5 Hz, H-2), 3.59 (s, 3H, CH₃O), 2.85 (br s, 2H, OH); ¹³C NMR (63 MHz, CDCl₃): δ 147.2 (C-5), 103.5, 100.1 (C-1, C-4), 71.3, 67.0 (C-2, C-3), 59.6 (CH₃O), 30.2 (C-6). ESIHRMS (positive ion mode): calcd for $C_7H_{11}O_4BrNa$: 260.97385, found: m/z260.97411.

3.5. Methyl 4-*C*-[(*S*)-1-phenyl-1-hydroxymethyl]-4,6dideoxy-α-D-*xylo*-hex-5-enopyranoside (8) and methyl 4,6-deoxy-α-L-*threo*-hex-5-enopyranoside (26)

Indium powder (115 mg, 1 mmol) was added to a soln of 7 (120 mg, 0.5 mmol) and benzaldehyde (153 μ L, 1.5 mmol) in a mixture of THF (1 mL) and phosphate buffer (pH7, 0.11 M, 0.5 mL). After stirring for 1.5 h, the mixture was neutralized with satd aq NaHCO₃. The suspension was filtered over Celite, washed with EtOH, and the filtrate was concentrated. Flash chromatography (1:4 petroleum ether–EtOAc) of the residue gave **8** (0.114 g, 86%) as white crystals followed by **26** (7 mg, 9%) as a colorless oil.

Compound **8**: mp 98 °C (Et₂O); $[\alpha]_D$ +99 (*c* 1.1, CH₂Cl₂); ¹H NMR (250 MHz, CDCl₃): δ 7.52–7.14 (m, 5H, Ar), 5.25 (br s, 1H, H-1), 4.81 (d, 1H, $J_{7,4}$ 4.0 Hz, H-7), 4.68 (s, 1H, H-6'), 4.39 (s, 1H, H-6), 4.20 (t, 1H, $J_{3,2} = J_{3,4}$ 8.5 Hz, H-3), 3.71 (dd, 1H, $J_{2,1}$ 3.5, $J_{2,3}$ 8.5 Hz, H-2), 3.44 (s, 3H, CH₃O), 2.82 (dd, 1H, $J_{4,7}$ 4.0, $J_{4,3}$ 8.5 Hz, H-4); ¹³C NMR (63 MHz, CDCl₃): δ 152.0 (C-5), 142.2, 128.2, 126.9, 125.9 (Ar), 100.2, 99.7 (C-1, C-6), 72.8, 70.0, 69.1 (C-2, C-3, C-7), 55.7 (CH₃O), 51.4 (C-4). Anal. Calcd for C₁₄H₁₈O₅: C, 63.15; H, 6.81; O, 30.04. Found: C, 63.21; H, 6.92; O, 30.25.

Compound **26**: $[\alpha]_D^{29}$ +84 (*c* 0.8, MeOH); ¹H NMR (250 MHz, CDCl₃): δ 4.86 (d, 1H, $J_{1,2}$ 3.5Hz, H-1), 4.54 (d, 1H, $J_{6',5}$ 1.0Hz, H-6'), 4.37 (d, 1H, $J_{6,5}$ 1.0Hz, H-6), 3.89 (ddd, 1H, $J_{3,4'}$ 5.5, $J_{3,2}$ 8.5, $J_{3,4}$ 10.5Hz, H-3), 3.55 (dd, 1H, $J_{2,1}$ 3.5, $J_{2,3}$ 8.5Hz, H-2), 3.47 (s, 3H, CH₃O), 2.69 (dd, 1H, $J_{4',3}$ 5.5, J_{gem} 13.5Hz, H-4'), 2.25 (tdd, 1H, $J_{4,6} = J_{4,6'}$ 1.0, $J_{4,3}$ 10.5, J_{gem} 13.5Hz, H-4); ¹³C NMR (62.9MHz, CDCl₃): δ 153.06 (C-5), 100.18 (C-1), 97.22 (C-6), 73.73, 68.22 (C-2, C-3), 55.56 (CH₃O), 36.17 (C-4); ESIMS (positive ion mode): calcd for C₇H₁₂O₄: (M+Na⁺), 183.0; found: *m*/*z* 183.0. Anal. Calcd for C₇H₁₂O₄: C, 51.49; H, 7.55. Found: C, 51.78; H, 7.53.

3.6. 4-C-Acetyl-5-(S)-C-phenyl-4-deoxy- α , β -L-arabinopyranose (9 α , β)

A soln of **8** (0.053 g, 0.199 mmol) was stirred at rt in aq HCl (0.1 M, 3 mL) for 5 min, then quenched with satd aq NaHCO₃, and extracted with CH₂Cl₂. The organic phase was dried (MgSO₄) and concentrated. Flash chromatography of the residue (EtOAc) gave a 1:3 mixture of the two adducts 9α , β (0.045 g, 90%) as a colorless oil.

Compound **9** α (25%): ¹H NMR (250 MHz, CDCl₃): δ 7.46–7.21 (m, 5H, Ar), 5.44 (d, 1H, $J_{5,4}$ 3.0Hz, H-5), 5.37 (d, 1H, $J_{1,2}$ 3.0Hz, H-1), 4.33–4.20 (m, 2H, H-2, H-3), 3.82 (m, 1H, H-4), 1.64 (s, 3H, CH₃CO); ¹³C NMR (63 MHz, CDCl₃): δ 217.3 (C=O), 140.2, 131.6, 131.0, 128.3 (Ar), 95.6 (C-1), 76.8, 74.8, 72.3 (C-2, C-3, C-5), 61.3 (C-4), 37.8 (CH₃CO). Compound **9** β (75%): ¹H NMR (250 MHz, CDCl₃): δ 7.46–7.21 (m, 5H, Ar), 4.91 (d, 1H, $J_{5,4}$ 2.9 Hz, H-5), 4.66 (d, 1H, $J_{1,2}$ 7.5 Hz, H-1), 4.10–3.94 (m, 2H, H-2, H-3), 3.74 (m, 1H, H-4), 1.61 (s, 3H, CH₃CO); ¹³C NMR (63 MHz, CDCl₃): δ 216.1 (C=O), 140.2, 131.6, 131.0, 128.3 (Ar), 100.0 (C-1), 75.0, 74.8, 71.2 (C-2, C-3, C-5), 61.4 (C-4), 37.3 (CH₃CO).

3.7. 1,**2**,**3**-Tri-*O*-acetyl-4-*C*-acetyl-5-(*S*)-*C*-(phenyl)-4-deoxy-α,β-L-arabinopyranose (10α,β)

A soln of $9\alpha,\beta$ (45mg, 0.18mmol) in a mixture of 1:2 Ac₂O-pyridine (1 mL) was kept overnight at rt and concentrated. Flash chromatography of the residue (1:4 petroleum ether-EtOAc) gave $10\alpha,\beta$ (66mg, 98%) as a 1:3 mixture of α and β anomers. Anal. Calcd for C₁₉H₂₂O₈: C, 60.31; H, 5.86; O, 33.83. Found: C, 60.29; H, 5.89; O, 33.17.

Compound 10a (25%): ¹H NMR (250 MHz, CDCl₃): δ 7.40–7.22 (m, 5H, Ar), 6.84 (d, 1H, J_{1,2} 4.0 Hz, H-1), 6.08 (dd, 1H, J_{2,1} 4.0, J_{2,3} 10.0 Hz, H-2), 5.54 (dd, 1H, J_{3.4} 6.0, J_{3.2} 10.0 Hz, H-3), 5.36 (d, 1H, J_{5.4} 3.5 Hz, H-5), 3.81 (dd, 1H, J_{4,5} 3.5, J_{4,3} 6.0 Hz, H-4), 2.35, 2.04, 2.03, 1.51 (4s, 12H, CH₃CO); ¹³C NMR (63 MHz, CDCl₃): δ 204.8, 170.3, 169.6, 168.8 (C=O), 136.3, 128.2, 125.5 (Ar), 90.1 (C-1), 71.4, 68.7, 66.6 (C-2, C-3, C-5), 55.8 (C-4), 34.0 (CH₃CO), 20.9, 20.7, 20.5 (CH₃COO). Compound **10**β (75%): ¹H NMR (250 MHz, CDCl₃): δ 7.40–7.22 (m, 5H, Ar), 6.00 (dd, 1H, J_{1.2} 8.5, J_{2.3} 9.5 Hz, H-2), 5.79 (d, 1H, J_{1.2} 8.5 Hz, H-1), 5.27 (dd, 1H, J_{3,4} 6.0, J_{3,2} 9.5Hz, H-3), 5.00 (d, 1H, $J_{5,4}$ 3.5Hz, H-5), 3.72 (dd, 1H, $J_{4,5}$ 3.5, $J_{4,3}$ 6.0 Hz, H-4), 2.15, 2.06, 2.02, 1.55 (4s, 12H, CH₃CO); ¹³C NMR (63 MHz, CDCl₃): δ 204.2, 170.2, 169.3 (C=O), 136.0, 128.6, 125.3 (Ar), 92.8 (C-1), 74.6, 72.2, 68.4 (C-2, C-3, C-5), 55.8 (C-4), 33.7, 20.8, 20.6 $(CH_3CO).$

3.8. Methyl 2,3-di-*O*-acetyl-4-*C*-[acetoxymethyl]-4,6dideoxy-α-D-*xylo*-hex-5-enopyranoside (12a) and methyl 2,3-di-*O*-acetyl-4-*C*-[acetoxymethyl]-4,6-dideoxy-β-L*arabino*-hex-5-enopyranoside (12b)

Indium powder (115 mg, 1 mmol) was added to a soln of 7 (120 mg, 0.5 mmol) and formaldehyde (37% in water, 120 μ L, 1.5 mmol) in a mixture of THF (1 mL) and phosphate buffer (pH7, 0.15 M, 0.38 mL). After stirring for 3h, the mixture was neutralized with satd aq NaHCO₃. The suspension was filtered over Celite, washed with EtOH and the filtrate was concentrated. Flash chromatography (EtOAc) of the residue gave a mixture of **11a,b** (65 mg, 73%) as a colorless oil, which was dissolved in 1:2 Ac₂O–pyridine (3 mL). After 15h, the solvents were coevaporated with toluene, then flash chromatography (3:1 petroleum ether–EtOAc) of the residue gave an inseparable 16:9 mixture of **12a** and **12b** (111 mg, 96%)

as a colorless oil followed by **26** (6 mg, 7%) as a colorless oil. Anal. Calcd for $C_{14}H_{20}O_8$: C, 53.16; H, 6.37; O, 40.47. Found: C, 52.86; H, 6.42; O, 40.48.

Compound **12a**: ¹H NMR (250 MHz, CDCl₃): δ 5.39 (dd, 1H, $J_{3,4}$ 9.0, $J_{3,2}$ 10.2 Hz, H-3), 5.05–4.94 (m, 2H, H-1, H-2), 4.79 (d, 1H, H-6'), 4.59 (m, 1H, H-6), 4.29 (dd, 1H, $J_{7',4}$ 6.0, J_{gem} 12.5 Hz, H-7'), 4.21 (dd, 1H, $J_{7,4}$ 4.0, J_{gem} 12.5 Hz, H-7), 3.45 (s, 3H, CH₃O), 2.72 (m, 1H, H-4), 2.11–2.07 (m, 9H, CH₃CO); ¹³C NMR (63 MHz, CDCl₃): δ 170.7, 170.2, 169.7, 152.0 (C-5), 98.4 (C-6), 97.6 (C-1), 71.8, 66.9 (C-2, C-3), 60.1 (C-7), 55.4 (CH₃O), 43.2 (C-4), 20.7 (CH₃CO).

Compound **12b**: ¹H NMR (250 MHz, CDCl₃): δ 5.43 (dd, 0.5H, $J_{3,4}$ 6.5, $J_{3,2}$ 10.0 Hz, H-3), 5.11 (dd, 0.5H, $J_{2,1}$ 6.0, $J_{2,3}$ 10.0 Hz, H-2), 5.05–4.94 (m, 2.5H, H-1), 4.70 (m, 1H, H-6'), 4.50 (m, 1H, H-6), 4.27 (m, 2H, H-7, H-7'), 3.45 (s, 3H, CH₃O), 3.22 (m, 1H, H-4), 2.11–2.07 (m, 12H, CH₃CO); ¹³C NMR (63 MHz, CDCl₃): δ 169.7, 170.1, 170.5 (C=O), 151.9 (C-5), 99.9 (C-6), 97.7 (C-1), 68.3, 67.6 (C-2, C-3), 61.3 (C-7), 55.5 (CH₃O), 42.0 (C-4), 20.7 (CH₃CO).

3.9. Methyl 2,3-di-*O*-acetyl-4-*C*-[1-(*S*)-1-acetoxymethyl-1-(3'-*O*-benzyl-1',2'-di-*O*-isopropylidene-4'-(*S*)-*C*- α -Dthreofuranosyl)]-4,6-dideoxy- α -D-*xylo*-hex-5-enopyranoside (15)

Indium powder (69 mg, 2 mmol) was added to a soln of 7 (72mg, 0.3mmol) and aldehyde 13 (0.25g, 0.9mmol) in a mixture of THF (0.6 mL) and phosphate buffer (pH7, 0.11 M, 0.3 mL). After stirring for 1.5 h, the mixture was neutralized with satd aq NaHCO₃. The suspension was filtered over Celite, washed with EtOH and the filtrate was concentrated. Flash chromatography (3:1-1:3 petroleum ether-EtOAc) gave starting aldehyde 13 (170 mg) along with 14 (75 mg, 57%) as a colorless oil followed by 26 (9mg, 18%) as a colorless oil. Compound 14 was dissolved in 1:2 Ac₂O-pyridine (3mL) and stirred overnight. The solvents were coevaporated with toluene, then flash chromatography (3:1 petroleum ether-EtOAc) of the residue gave 15 (96mg, 99%) as a colorless oil; $[\alpha]_D$ +52 (c 1.3, CH₂Cl₂); ¹H NMR (250 MHz, CDCl₃): δ 7.48–7.32 (m, 5H, Ar), 5.97 (d, 1H, $J_{1',2'}$ 4.0 Hz, H-1'), 5.69 (dd, 1H, J_{7,4} 3.0, J_{7,4'} 9.0 Hz, H-7), 5.20 (t, 1H, $J_{3,2} = J_{3,4}$ 10.0 Hz, H-3), 4.93 (br s, 1H, H-6'), 4.87 (d, 1H, J_{1.2} 3.5Hz, H-1), 4.81 (br s, 1H, H-6), 4.79 (dd, 1H, J_{2.1} 3.5, J_{2.3} 10.0 Hz, H-2), 4.74 (d, 1H, J_{gem} 11.5 Hz, PhCH₂), 4.69 (d, 1H, J_{2',1'} 4.0 Hz, H-2'), 4.62 (dd, 1H, $J_{4',3'}$ 3.5, $J_{4',7}$ 9.0 Hz, H-4'), 4.41 (d, 1H, J_{gem} 11.5Hz, PhCH₂), 3.89 (d, 1H, J_{3',4'} 3.5Hz, H-3'), 3.42 (s, 3H, CH₃O), 2.38 (dd, 1H, J_{4.7} 3.0, J_{4.3} 10.0 Hz, H-4), 2.07, 2.03, 1.99 (3s, 9H, CH₃CO), 1.48, 1.33 (2s, 6H, C(CH₃)₂); ¹³C NMR (63 MHz, CDCl₃): δ 170.3, 169.8, 169.5 (CH₃CO), 149.9 (C-5), 136.3, 128.6, 128.3 (Ar), 111.6 (C(CH₃)₂), 104.8 (C-1'), 102.0 (C-1), 98.0 (C-6), 81.7, 80.6, 78.5 (C-2', C-3', C-4'), 72.7 (PhCH₂),

71.4, 66.5, 65.7 (C-2, C-3, C-7), 55.6 (CH₃O), 44.9 (C-4), 26.6, 26.1 (C(CH₃)₂), 20.9, 20.7, 20.6 (CH₃CO). Anal. Calcd for $C_{28}H_{36}O_{12}$: C, 59.57; H, 6.43; O, 34.00. Found: C, 59.32; H, 6.53; O, 34.28.

3.10. 4-C-Acetyl-5-(S)-C-[3'-O-benzyl-1',2'-di-O-isopropylidene-4'-(S)-C- α -D-threofuranosyl]-4-deoxy- α , β -L-arabinopyranose (17 α , β)

A soln of **14** (0.092 g, 0.21 mmol) was stirred at rt in aq HCl (0.1 M, 2mL) for 30 min, then quenched with a satd aq NaHCO₃ and extracted with CH₂Cl₂. The organic phase was then dried (MgSO₄) and concentrated. Flash chromatography (1:4 petroleum ether–EtOAc) of the residue gave **16** α , β (11:9, 77 mg, 86%) as a colorless oil. This oil was then dissolved in 1:2 Ac₂O–pyridine (1.5 mL), stirred overnight and coevaporated with toluene. Flash chromatography (4:1–3:2 petroleum ether–EtOAc) of the residue gave a 11:9 mixture of α and β anomers **17** α , β (0.093 g, 94%) as a colorless oil. Anal. Calcd for C₂₇H₃₄O₁₂: C, 58.90; H, 6.22; O, 34.88. Found: C, 58.96; H, 6.41; O, 34.73.

Compound 17 α : ¹H NMR (250 MHz, CDCl₃): δ 7.48– 7.34 (m, 5H, H-Ar), 6.45 (d, 1H, $J_{1,2}$ 4.0 Hz, H-1), 5.90 (d, 1H, $J_{1',2'}$ 4.0 Hz, H-1'), 5.61 (dd, 1H, $J_{2,1}$ 4.0, $J_{2,3}$ 10.5 Hz, H-2), 5.38 (dd, 1H, J_{3,4} 6.0, J_{3,2} 10.5 Hz, H-3), 4.73 (d, 1H, J_{gem} 11.5Hz, CH₂C₆H₅), 4.69 (dd, 1H, $J_{2',3'}$ 1.0, $J_{2',1'}$ 4.0 Hz, 2'-H), 4.44 (d, 1H, J_{gem} 11.5 Hz, CH₂C₆H₅), 4.34 (dd, 1H, J_{5.4} 2.5, J_{5.4}, 7.0 Hz, H-5), 4.19 (dd, 1H, $J_{4',3'}$ 4.5, $J_{4',5}$ 7.0 Hz, H-4'), 3.89 (dd, 1H, $J_{3',2'}$ 1.0, $J_{3',4'}$ 4.5 Hz, H-3'), 3.28 (dd, 1H, $J_{4,5}$ 2.5, J_{4,3} 6.0 Hz, H-4), 2.06, 1.98, 1.91 (3s, 12H, CH₃ acetyl), 1.45 (s, 3H, CH₃), 1.33 (s, 3H, CH₃). ¹³C NMR (62.9 MHz, CDCl₃): δ 205.3 (CO, ketone), 169.8, 169.6, 168.9 (CO, acetyl), 136.3, 129.1-128.3 (Ar), 112.7 ((CH₃)₂C), 105.0 (C-1'), 89.8 (C-1), 82.3, 81.0, 79.1 (C-2', C-3', C-4'), 71.5 (CH₂C₆H₅), 70.2, 68.0, 66.7 (C-2, C-3, C-5), 50.8 (C-4), 33.3 (CH₃, ketone), 27.1, 26.7 (isopropyl), 20.8, 20.7, 20.5 (acetyl).

Compound 17 β : ¹H NMR (250 MHz, CDCl₃): δ 7.48–7.34 (m, 5H, Ar), 5.94 (d, 1H, $J_{1',2'}$ 4.0 Hz, H-1'), 5.72 (d, 1H, J_{1,2} 8.5 Hz, H-1), 5.55 (dd, 1H, J_{2,1} 8.5, J_{2,3} 10.0 Hz, H-2), 5.17 (dd, 1H, J_{3,4} 6.0, J_{3,2} 10.0 Hz, H-3), 4.72 (d, 1H, J_{gem} 11.5 Hz, CH₂C₆H₅), 4.64 (d, 1H, $J_{2',1'}$ 4.0 Hz, H-2'), 4.43 (d, 1H, J_{gem} 11.5 Hz, CH₂C₆H₅), 4.13 (dd, 1H, J_{3',4'} 3.5, J_{4',5} 8.5Hz, H-4'), 4.01 (dd, 1H, J_{5.4} 2.5, J_{5.4}' 8.5 Hz, H-5), 3.92 (d, 1H, $J_{4',3'}$ 3.5 Hz, H-3'), 3.08 (dd, 1H, $J_{4,5}$ 2.5, $J_{4,3}$ 6.0 Hz, H-4), 2.10, 2.09, 2.03, 2.02 (4s, 12H, CH₃CO), 1.44 (s, 3H, CH₃), 1.31 (s, 3H, CH₃); ¹³C NMR (63 MHz, CDCl₃): δ 205.0 (CO ketone), 169.5, 168.8 (acetyl), 136.5, 128.4–128.7 (Ar), 112.2 ((CH₃)₂C), 105.2 (C-1'), 92.7 (C-1), 81.8, 81.0, 79.0 (C-2', C-3', C-4'), 73.0 (CH₂C₆H₅), 71.7, 71.2, 69.0 (C-2, C-3, C-5), 51.8 (C-4), 32.9 (CH₃ ketone), 26.9, 26.4 (CH₃), 20.9, 20.6 (acetyl).

3.11. Methyl 2,3-di-O-acetyl-4-C-[(S)-1-O-acetyl-1-(2',3',4',6'-tetra-O-benzyl-β-D-glucopyranosyl)-methyl]-4,6-dideoxy-α-D-*xylo*-hex-5-enopyranoside (20a), methyl 2,3-di-O-acetyl-4-C-[(S)-1-hydroxy-1-(2',3',4',6'-tetra-Obenzyl-β-D-glucopyranosyl)-methyl]-4,6-dideoxy-α-D-*arabino*-hex-5-enopyranoside (21b)

Indium powder (53 mg, 0.46 mmol) was added to a soln of 7 (55 mg, 0.23 mmol) and aldehyde **18** (0.38 g, 0.69 mmol) in a mixture of THF (0.46 mL) and phosphate buffer (pH7, 0.11 M, 0.23 mL). After stirring for 2h, the mixture was neutralized with satd aq NaHCO₃. The suspension was filtered over Celite, washed with EtOH, and the filtrate was concentrated. Flash chromatography (3:1–1:3 petroleum ether–EtOAc) of the residue gave the starting aldehyde (248 mg, 0.45 mmol) followed by a 16:9 mixture of **19a** and **19b** (74 mg, 45%), and finally by **26** (11 mg, 30%) as a colorless oil.

Compounds **19a,b** (64:36): ¹H NMR (250 MHz, CDCl₃): δ 3.07 (t, 0.55H, $J_{4,3} = J_{4,7}$ 6.0 Hz, H-4(b)), 2.67 (t, 1H, $J_{4,3} = J_{4,7} = 6.0$ Hz, H-4(a)); ¹³C NMR (62.9 MHz, CDCl₃): δ 154.5 (C-5(b)), 153.5 (C-5(a)), 138.5–137.0, 128.9–127.2 (Ar), 100.6 (C-1(a)), 100.2–100.0 (C-1(b), C-6(b)), 98.7 (C-6(a)), 87.2 (C-1'(b)), 87.1 (C-1'(a)), 75.5, 75.4, 75.3, 75.0, 74.9, 73.4, 73.3 (CH₂ benzyl (a,b)), 72.15, 72.0, 70.5, 69.3, 69.2, 68.65 (C-2(a,b), C-3(a,b), C-7(a,b), C-6'(a,b)), 56.0 (OCH₃ (a)), 55.8 (OCH₃ (b)), 48.3 (C-4(a)), 48.0 (C-4(b)).

An aliquot of the 19:9 mixture of **19a,b** (50 mg, 0.07 mmol) was added to a mixture of Ac_2O (0.5 mL) and pyridine (1 mL). After 16h at rt, the reaction mixture was concentrated and flash chromatography of the residue (4:1–3:2 petroleum ether–EtOAc) gave first **20a** (51 mg, 89%) followed by the partially acetylated compound **20b** (25 mg, 81%). This latter was then fully acetylated in 2:1 pyridine– Ac_2O (1.5 mL) at 40 °C for 12h. The mixture was concentrated, and flash chromatography of the residue (3:1 petroleum ether–EtOAc) gave **21b** (25 mg, 95%) as a colorless oil.

Compound **20a**: $[\alpha]_D$ +33 (*c* 1.2, CH₂Cl₂); ¹H NMR (250 MHz, CDCl₃): δ 7.45–7.15 (m, 20H, H-Ar), 5.63 (dd, 1H, $J_{7,1}$ 3.0, $J_{7,4}$ 8.5Hz, H-7), 5.37 (dd, 1H, $J_{3,4}$ 5.5, $J_{3,2}$ 9.0Hz, H-3), 5.05–4.50 (m, 11H, 1-H, 2-H, H-6b, CH₂ benzyl), 4.37 (s, 1H, H-6a), 3.85–3.37 (m, 10H, CH₃O, H-1', H-2', H-3', H-4', H-5', 2H-6'), 3.04 (dd, 1H, $J_{4,3}$ 5.5, $J_{4,7}$ 8.5Hz, H-4), 2.06, 1.91, 1.82 (3s, 9H, Ac); ¹³C NMR (63 MHz, CDCl₃): δ 170.2, 169.5 (CO), 151.5 (C-5), 138.4–138.1, 128.3–127.5 (Ar), 98.5 (C-6), 97.7 (C-1), 87.4 (C-1'), 79.3, 79.2, 78.3, 76.5 (C-2', C-3', C-4', C-5'), 75.6, 75.0, 74.4, 73.2 (CH₂ benzyl), 71.8, 70.9, 66.9 (C-2, C-3, C-7), 68.6 (C-6'), 56.0 (CH₃O), 46.2 (C-4), 20.9, 20.7, 20.6 (CH₃ acetyl). Anal. Calcd for C₄₈H₅₄O₁₃: C, 68.72; H, 6.49; O, 24.79. Found: C, 68.37; H, 6.55; O, 24.16.

Compound **20b**: ¹H NMR (250 MHz, CDCl₃): δ 5.49 (dd, 1H, $J_{2,3}$ 10.0, $J_{2,1}$ 3.0Hz, H-2), 5.43 (dd, 1H, $J_{3,2}$

10.0, $J_{3,4}$ 5.0 Hz, H-3), 5.04 (d, 1H, $J_{1,2}$ 3.0 Hz, H-1), 3.15 (t, 1H, $J_{4,7} = J_{4,3}$ 5.0 Hz, H-4), 2.08 (s, 3H, CH₃), 1.88 (s, 3H, CH₃). ESIMS (positive ion mode): calcd for C₄₆H₅₂O₁₂: (M+Na⁺), 819.3; found: *m*/*z* 819.4.

Compound **21b**: $[\alpha]_D^{29}$ +63 (*c* 0.9, CH₂Cl₂); ¹H NMR (250 MHz, CDCl₃): δ 7.45–7.15 (m, 20H, Ar), 5.78 (d, 1H, J 8.5Hz, H-7), 5.48 (dd, 1H, J_{3,4} 6.0, J_{3,2} 10.5 Hz, H-3), 5.35 (dd, 1H, J_{2,1} 3.5, J_{2,3} 10.5 Hz, H-2), 5.01 (d, 1H, J_{1,2} 3.5Hz, H-1), 4.99–4.51 (m, 10H, H-6a, H-6b, CH₂ benzyl), 3.80-3.32 (m, 11H, CH₃O, H-1', H-2', H-3', H-4', H-5', H-6a', H-6b', H-4), 2.11, 2.08, 1.90 (3s, 9H, acetyl); ¹³C NMR (62.9 MHz, CDCl₃): δ 170.2–169.8 (acetyl), 151.4 (C-5), 138.4– 138.0, 128.5–127.5 (Ar), 102.3 (C-6), 98.0 (C-1), 87.7 (C-1'), 79.5, 78.35, 78.1 (C-2', C-3', C-4', C-5'), 75.6, 75.0, 74.9, 73.6 (CH₂ benzyl), 69.0, 68.15 (C-2, C-3), 68.5 (C-6'), 66.5 (C-7), 55.5 (CH₃O), 44.7 (C-4), 21.05, 20.75, 20.7 (CH₃CO). ESIMS (positive ion mode): calcd for $C_{48}H_{54}O_{13}$: (M+Na⁺), 861.3; found: *m*/*z* 861.0.

3.12. 1,2,3-Tri-*O*-acetyl-4-*C*-acetyl-(*S*)-5-*C*-[2',3',4',6'-tetra-*O*-benzyl- β -D-glucopyranosyl]-4-deoxy- α , β -L-arabinopyranose (23 α , β)

LiOH (16mg, 0.4mmol) was added to a soln of compound 20a (55mg, 65µmol) in 2:1 MeOH-water (1.5 mL). After 30 min at rt, aq HCl (0.5 M) was added until neutrality and the mixture was extracted with CH₂Cl₂ then concentrated. Aq HCl (0.1 M, 1 mL) was added and the mixture was stirred for 2h at rt. After neutralization with sat aq NaHCO₃, the mixture was extracted with CH₂Cl₂, dried (MgSO₄), and concentrated. A mixture of pyridine-Ac₂O (2:1, 3mL) was then added and after 16h at rt, the reaction mixture was concentrated. Flash chromatography of the residue (4:1-2:1 petroleum ether–EtOAc) gave a 12:8 mixture of α and β anomers 23 α , β (44 mg, 82%) as a colorless oil. ¹H NMR (250 MHz, CDCl₃): δ 7.42–7.10 (m, 32.8H, Ar), 6.46 (d, 1H, $J_{1,2}$ 4.0 Hz, H-1(α)), 5.76 (dd, 1H, $J_{2,1}$ 4.0, $J_{2,3}$ 10.5 Hz, H-2(α)), 5.62 (dd, 0.64 H, $J_{2,1}$ 8.5, $J_{2,3}$ 10.0 Hz, H-2(β)), 5.52 (d, 0.64H, $J_{1,2}$ 8.5 Hz, H-1(β)), 5.22 (dd, 1H, J_{3,4} 6.0, J_{3,2} 10.5 Hz, H-3(α)), 4.93–4.41 (m, 13.8H, H-3(β), CH₂Ph), 4.39 (t, 1H, $J_{5,4} = J_{5,1'}$ 2.5 Hz, H-5(α)), 3.78 (t, 0.64H, $J_{5,4} = J_{5,1'}$ 2.5 Hz, H-5(β)), 3.74–3.33 (m, 11.4H, H-1'(α , β), H-2'(α , β), H- $3'(\alpha,\beta)$, H-4'(α,β), H-5'(α,β), H-6a'(α,β), H-6b'(α,β)), 3.21 (dd, 1H, $J_{4,5}$ 2.5, $J_{4,3}$ 6.0Hz, H-4(α)), 2.91 (dd, $0.64H, J_{4,5}$ 2.5, $J_{4,3}$ 6.0 Hz, H-4(β)), 2.12, 2.11, 2.03, 2.00, 1.97, 1.96 (8s, 20H, CH₃COO and CH₃CO); ¹³C NMR (62.9 MHz, CDCl₃): δ 205.15 (CO ketone (α)), 204.73 (CO ketone (β)), 169.87, 169.54, 169.25 (CO ester (β)), 169.69, 169.39, 168.77 (CO ester(α)), 138.16– 137.55, 128.70-127.62 (Ar), 92.64 (C-1(β)), 90.33 (C- $1(\alpha)$), 86,65 (C-1'(β)), 86.43 (C-1'(α)), 78.81, 78.45, 78.39, 78.41, 78.37, 77.68, 77.55, 77.46 (C-2'(α,β), C- 3'(α,β), C-4'(α,β), C-5'(α,β)), 75.43, 74.83, 74.41, 74.27, 73.24 (CH₂Ph(α,β)), 74.22, 71.89, 71.08, 68.90, 68.77, 66.96, (C-2(α,β), C-3(α,β), C-5(α,β)), 68.60 (C-6'(β)), 68.43 (C-6'(α)), 49.17 (C-4(β)), 49.00 (C-4(α)), 34.08 (CH₃CO(α)), 33.69 (CH₃CO(β)), 20.74, 20.71, 20.66 (CH₃COO(β)), 20.79, 20.63, 20.51 (CH₃COO(α)). Anal. Calcd for C₄₇H₅₂O₁₃ (434.54): C, 68.43; H, 6.35; O, 25.21. Found: C, 68.58; H, 6.59; O, 25.21.

3.13. 1,2,3-Tri-O-acetyl-4-C-acetyl-(S)-5-C-[2',3',4',6'-tetra-O-benzyl- β -D-glucopyranosyl]-4-deoxy- α , β -L-arabinopyranose (25 α , β)

LiOH (11 mg, 0.27 mmol) was added to a soln of compound **21b** (37 mg, 44 μ mol) in 2:1 MeOH–water (1.5 mL). After 30 min at rt, aq HCl (0.5 M) was added until neutrality and the mixture extracted with CH₂Cl₂ then concentrated. Aq HCl (0.1 M, 1 mL) was added and the mixture was stirred for 2 h at rt. After neutralization with satd aq NaHCO₃, the mixture was extracted with CH₂Cl₂, dried (MgSO₄), and concentrated. Pyridine–Ac₂O (2:1, 3 mL) was then added and after 16 h at rt, the reaction mixture was concentrated. Flash chromatography of the residue (4:1–2:1 petroleum ether– EtOAc) gave a 4:5 mixture of α and β anomers **25** α , β (24 mg, 77%) as a colorless oil. Anal. Calcd for C₄₇H₅₂O₁₃: C, 68.43; H, 6.35; O, 25.21. Found: C, 68.46; H, 6.61; O, 24.89.

Compound **25***a*: ¹H NMR (250 MHz, CDCl₃): δ 7.42– 7.15 (m, 20H, Ar), 6.21 (d, 1H, $J_{1,2}$ 3.5Hz, H-1), 6.00 (t, 1H, $J_{3,2} = J_{3,4}$ 8.0Hz, H-3), 4.98 (dd, 1H, $J_{2,1}$ 3.5, $J_{2,3} = 8.0$ Hz, H-2), 4.94.52 (m, 9H, H-5, CH₂Ph), 3.81–3.51, 3.31–3.21 (m, 7H, H-1', H-2', H-3', H-4', H-5', H-6a', H-6b'), 3.08 (dd, 1H, $J_{4,5}$ 6.5, $J_{4,3}$ 8.0Hz, H-4), 2.16, 2.09, 2.05, 1.91 (4s, 12H, acetyl); ¹³C NMR (62.9 MHz, CDCl₃): δ 202.71 (CO ketone), 170.19, 169.74, 169.41 (CO acetyl), 138.79, 138.66, 138.26, 138.19, 128.35, 127.19 (Ar), 91.07 (C-1), 86.75 (C-1'), 79.90, 79.25, 77.96, 77.19 (C-2', C-3', C-4', C-5'), 75.40, 75.10, 74.97, 73.49 (CH₂Ph), 72.18, 71.03, 69.29 (C-2, C-3, C-5), 68.78 (C-6'), 55.10 (C-4), 29.71 (CH₃ ketone), 20.84, 20.79, 20.30 (acetyl).

Compound **25**β: ¹H NMR (250 MHz, CDCl₃): δ 7.40–7.12 (m, 20H, Ar), 6.32 (dd, 1H, $J_{3,4}$ 10.5, $J_{3,2}$ 9.5Hz, H-3), 4.95 (dd, 1H, $J_{2,1}$ 8.5, $J_{2,3}$ 9.5Hz, H-2), 4.91–4.58 (m, 9H, H-5, CH₂Ph), 4.12 (t, 1H, J 9.5Hz), 3.92–3.77 (m, 3H), 3.63 (t, 1H, J 9.5Hz), 3.25 (dd, 1H, $J_{4,3}$ 10.5, $J_{4,5}$ 7.5Hz H-4), 3.39–3.17 (m, 2H), 2.18, 2.12, 1.97, 1.69 (4s, 12H, acetyl); ¹³C NMR (62.9MHz, CDCl₃): δ 203.98 (CO ketone), 168.86, 168.11, 167.64 (acetyl), 138.28, 138.11, 137.66, 137.50, 128.45–127.25 (Ar), 90.07 (C-1), 85.96 (C-1'), 79.53, 79.34, 78.88, 78.44 (C-2', C-3', C-4', C-5'), 74.96, 74.46, 74.04, 73.79 (CH₂Ph), 72.15, 71.18, 69.38, (C-2, C-3, C-5), 68.68 (C-6'), 55.05 (C-4), 29.78 (CH₃ ketone), 20.65, 20.59, 20.30 (acetyl).

3.14. Methyl 2,3,6-tri-*O*-acetyl-4-*C*-[(*S*)-1-*O*-acetyl-1-(2',3',4',6'-tetra-*O*-benzyl- β -D-glucopyranosyl)-methyl]- α -L-idopyranoside (27)

A 1M soln of BH₃-THF complex in THF (0.20mL, 0.2 mmol) was added to a cooled (0°C) soln of 20a (34 mg, 40 µmol) in THF (0.5 mL). The temperature was then raised to rt and after 1h, a mixture of H_2O_2 (37% v/v, 0.2 mL) and phosphate buffer (0.5 M, pH7, mH7)1mL) was added and the resulting mixture stirred for 1h at rt. CH₂Cl₂ (10mL) was then added and the mixture was washed with water, dried (MgSO₄), and concentrated. Then, 1:2 Ac₂O-pyridine (3mL) was added and after stirring for 12h, the reaction mixture was concentrated. Flash chromatography of the residue (4:1 petroleum ether-EtOAc) gave 27 (33mg, 91%) as a colorless oil. $[\alpha]_{D}^{29}$ +36 (c 1.1, CH₂Cl₂); ¹H NMR (250 MHz, CDCl₃): δ 7.41–7.08 (m, 20H, Ar), 5.54 (dd, 1H, $J_{7,1'}$ 4.0, J_{7,4} 3.5 Hz, H-7), 5.41 (dd, 1H, J_{3,2} 10.5, J_{3,4} 10.0 Hz, H-3), 5.03-4.36 (m, 13H, H-1, H-2, H-5, 2H-6, CH₂Ph), 3.44 (s, 3H, OMe), 3.78-3.37 (m, 7H, H-1', H-2', H-3', H-4', H-5', 2H-6'), 2.53 (ddd, 1H, $J_{4,3}$ 10.0, $J_{4,5}$ 5.0, J_{4,7} 3.5 Hz, H-4), 2.03, 1.98, 1.88, 1.87 (4s, 12H, acetyl), ¹³C NMR (62.9 MHz, CDCl₃): δ 170.75, 170.38, 170.18, 169.92 (CO), 138.17-137.92, 128.44-127.44 (Ar), 98.90 (C-1), 87.06 (C-1'), 79.05, 78.73, 78.36, 77.90 (C-2', C-3', C-4', C-5'), 75.42, 74.93, 74.17, 73.21, 72.69 (CH₂Ph, C-5), 68.77 (C-6'), 67.27 (C-7), 65.85, 64.73, 63.73 (C-2, C-3, C-6), 56.76 (OMe), 43.21 (C-4), 20.94, 20.73, 20.57 (CH₃ acetyl). Anal. Calcd for C₅₀H₅₈O₁₅: C, 66.80; H, 6.50; O, 26.70. Found: C, 66.82; H, 6.66; O, 26.61.

3.15. Methyl 2,3,6-tri-*O*-acetyl-4-*C*-[(*S*)-1-*O*-acetyl-1-(2',3',4',6'-tetra-*O*-benzyl- β -D-glucopyranosyl)-methyl]- α -D-galactopyranoside (28)

A 1M soln of BH₃-THF complex in THF (0.1mL, 0.1 mmol) was added to a cooled (0°C) soln of 21b (18mg, 21µmol) in THF (0.5mL). The temperature was then raised to rt and after 1h, a mixture of H_2O_2 (37% v/v, 0.1 mL) and phosphate buffer (0.5 M, pH7, 1mL) was added and the resulting mixture stirred for 1h at rt. CH₂Cl₂ (10mL) was then added and the mixture was washed with water, dried (MgSO₄), and concentrated. Then, Ac₂O-pyridine (1:2, 3mL) was added and after stirring for 12h, the reaction mixture was concentrated. Flash chromatography of the residue (4:1 petroleum ether-EtOAc) gave 28 (16mg, 83%) as a colorless oil. $[\alpha]_{D}^{28}$ +88 (c 1.0, CHCl₃); ¹H NMR (250 MHz, CDCl₃): δ 7.45–7.08 (m, 20H, Ar), 5.87 (d, 1H, J_{7,4} 3.0 Hz, H-7), 5.49 (dd, 1H, J_{1,2} 4.0, J_{2,3} 10.5 Hz, H-2), 5.25 (dd, 1H, J_{3,2} 10.5, J_{3,4} 5.5 Hz, H-3), 5.01–4.36 (m, 12H, H-1, H-5, 2H-6, CH₂Ph), 3.34 (s, 3H, OCH₃), 3.75-3.29 (m, 6H, H-2', H-3', H-4', H-5', 2H-6'), 2.91 (d, 1H, $J_{1',2'}$ 9.5Hz, H-1'), 2.81 (m, 1H, H-4), 2.17, 2.10, 2.02, 1.74 (4s, 12H, acetyl); ¹³C NMR (62.9 MHz, CDCl₃): δ 170.47, 170.35, 170.13, 169.87 (CO), 138.42, 138.35, 138.05, 137.96, 128.61–127.45 (Ar), 97.11 (C-1), 87.31 (C-1'), 81.91, 79.08, 78.25 (C-2', C-3', C-4', C-5'), 75.66, 75.13, 75.00, 73.34 (CH₂Ph, C-5), 70.20, 69.46, 68.31, 65.88, 65.20 (C-2, C-3, C-6, C-7, C-6'), 54.82 (OCH₃), 42.30 (C-4), 21.26, 20.92, 20.83, 20.47 (CH₃ acetyl). Anal. Calcd for C₅₀H₅₈O₁₅: C, 66.80; H, 6.50; O, 26.70. Found: C, 66.81; H, 6.76; O, 26.66.

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References

- (a) Chapleur, Y.; Chrétien, F. In Preparative Carbohydrate Chemistry; Hanessian, S., Ed.; Marcel Dekker: New York, 1997; pp 207–262; (b) Yoshimura, J. Adv. Carbohydr. Chem. Biochem. 1984, 42, 69–134; (c) Celmer, W. D. Pure Appl. Chem. 1971, 28, 413–453; (d) Hanessian, S. Total Synthesis of Natural Products: The Chiron Approach; Pergamon: Oxford, 1983; (e) Fraser-Reid, B. Acc. Chem. Res. 1996, 29, 57–66; (f) Makoto, B.; Nishikawa, T.; Isobe, M. Tetrahedron 1998, 54, 6639–6650; (g) Chida, N.; Takeoka, J.; Ando, K.; Tsutsumi, N.; Ogawa, S. Tetrahedron 1997, 53, 16287–16298; (h) Sasaki, M.; Inoue, M.; Tachibana, K. J. Org. Chem. 1994, 59, 715–717; (i) Alonso, R. A.; Burgey, C. S.; Rao, B. V.; Vite, G. D.; Vollerthun, R.; Zottola, M. A.; Fraser-Reid, B. J. Am. Chem. Soc. 1993, 115, 6666–6672.
- 2. (a) Hanessian, S.; Pernet, A. G. Adv. Carbohydr. Chem. Biochem. 1976, 33, 111-188; (b) Postema, M. H. D. Tetrahedron 1992, 48, 8545-8599; (c) Levy, D. E.; Tang, C. The Chemistry of C-Glycosides; Pergamon: Oxford, 1995; (d) Postema, M. H. D. C-Glycoside Synthesis; CRC: London, 1995; (e) Beau, J. M.; Gallagher, T. Top. Curr. Chem. 1997, 187, 1-54; (f) Nicotra, F. Top. Curr. Chem. 1997, 187, 55-83; (g) Schmidt, R. R.; Kast, J. Tetrahedron Lett. 1986, 27, 4007-4010; (h) Fraser-Reid, B.; Magdzinski, L.; Molino, B. F.; Mootoo, D. R. J. Org. Chem. 1987, 52, 4495-4504; (i) Lawrence, A. J.; Pavey, J. B. J.; Cosstick, R.; O'Neil, I. A. J. Org. Chem. 1996, 61, 9213-9222; (j) Dondoni, A.; Kniezo, L.; Martinkova, M.; Imrich, J. Chem. Eur. J. 1997, 3, 424-430; (k) Sinay, P. Pure Appl. Chem. 1997, 69, 459-463; (1) Giese, B.; Gonzalez-Gomez, J. A.; Witzel, T. Angew. Chem., Int. Ed. Engl. 1984, 23, 69-70; (m) Giese, B.; Gröninger, K. Tetrahedron Lett. 1984, 25, 2743-2746; (n) Jung, M. E.; Choe, S. W. T. Tetrahedron Lett. 1993, 34, 6247-6250; (o) Linker, T.; Hartmann, K.; Sommermann, T.; Scheutzow, D.; Ruckdeschel, E. Angew. Chem., Int. Ed. Engl. 1996, 35, 1730–1732; (p) Linker, T.; Sommermann, T.; Kahlenberg, F. J. Am. Chem. Soc. 1997, 119, 9377-9384; (q) Beyer, J.; Madsen, R. J. Am. Chem. Soc. 1998, 120, 12137-12138; (r) Rauter, A.; Ferreira, M.; Borges, C.; Duarte, T.; Piedade, F.; Silava, M.; Santos, H. Carbohydr. Res. 2000, 325, 1-15.
- (a) Canac, Y.; Levoirier, E.; Lubineau, A. J. Org. Chem.
 2001, 66, 3206–3210; (b) Lubineau, A.; Canac, Y.; Le Goff, N. Adv. Synth. Catal. 2002, 344, 319–327.

- (a) Schmid, W.; Whitesides, G. M. J. Am. Chem. Soc. 1991, 113, 6674–6675; (b) Chan, T. H.; Li, C. J. J. Chem. Soc., Chem. Commun. 1992, 747–748; (c) Gao, J.; Härter, R.; Gordon, D. M.; Whitesides, G. M. J. Org. Chem. 1994, 59, 3714–3715; (d) Chan, T. H.; Lee, M. C. J. Org. Chem. 1995, 60, 4228–4232; (e) Chan, T. H.; Xin, Y. C.; VonItzstein, M. J. Org. Chem. 1997, 62, 3500–3504; (f) Warwel, M.; Fessner, W. D. Synlett 2000, 6, 865–867.
- Kloosterman, M.; Mosmuller, E. W. J.; Schoemaker, H. E.; Meijer, E. M. *Tetrahedron Lett.* 1987, 28, 2989– 2992.
- Mathad, V. T.; Shefali, V.; Raj, K.; Bhaduri, A. P. Indian J. Chem. Sect. B 1997, 36, 808–809.

- 7. Horton, D.; Liav, A. Carbohydr. Res. 1972, 24, 105-113.
- Sauerbrei, B.; Niggemann, J.; Gröger, S.; Lee, S.; Floss, H. G. Carbohydr. Res. 1996, 280, 223–235.
- (a) Wolfrom, M. L.; Hanessian, S. J. Org. Chem. 1962, 27, 1800–1804;
 (b) Horton, D.; Swanson, F. O. Carbohydr. Res. 1970, 14, 159–171.
- (a) Sanchez, M. E. L.; Michelet, V.; Besnier, I.; Genêt, J. P. Synlett 1994, 705–708; (b) Dondoni, A.; Scherrmann, M. C. J. Org. Chem. 1994, 59, 6404–6412.
- (a) Rochepeau-Jobron, L.; Jacquinet, J. C. Carbohydr. Res. 1997, 303, 395–406; (b) Hinou, H.; Kurosawa, H.; Matsuoka, D.; Turunama, D.; Kuzuhara, H. Tetrahedron Lett. 1999, 40, 1501–1504.