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Introduction

Hexavalent thiofucosides to probe the role of the *Aspergillus fumigatus* lectin FleA in fungal pathogenicity[†]

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Aspergillus fumigatus is a pathogenic fungus infecting the respiratory system and responsible for a variety of life-threatening lung diseases. A fucose-binding lectin named FleA which has a controversial role in *A. fumigatus* pathogenesis was recently identified. New chemical probes with high affinity and enzymatic stability are needed to explore the role of FleA in the infection process. In this study, we developed potent FleA antagonists based on optimized and non-hydrolysable thiofucoside ligands. We first synthesized a set of monovalent sugars showing micromolar affinity for FleA by isothermal titration calorimetry. The most potent derivative was co-crystallized with FleA to gain insights into the binding mode in operation. Its chemical multimerization on a cyclodextrin scaffold led to an hexavalent compound with a significantly enhanced binding affinity ($K_d = 223 \pm 21$ nM) thanks to a chelate binding mode. The compound could probe the role of bronchial epithelial cells in a FleA-mediated response to tissue invasion.

Aspergillus fumigatus is one of the most ubiquitous saprophytic fungi, with an ecological niche in the soil. *A. fumigatus* abundantly disseminates conidia into the atmosphere, which are easily inhaled into the lungs where they reach the pulmonary alveoli due to their small size. Immunocompetent hosts efficiently clear the conidia out from their respiratory system by mechanical mucus elimination or innate immunity. However, *A. fumigatus* is a severe opportunistic pathogen for immunocompromised patients in whom may cause generally fatal lung infections such as invasive aspergillosis or allergic bronchopulmonary aspergillosis.¹ Increasing resistances to antifungal drug treatments such as the azole class of antifungal agents has fostered research to identify the virulence factors of *A. fumigatus*.

Carbohydrate-binding proteins (lectins) from microbial pathogens are one of the most common group of receptors involved in the primary steps of cell infection. Lectins expressed by bacteria or fungi and mediating host adhesion are therefore promising targets for the development of therapies based on antiadhesive molecules. Such a strategy is less prone to antimicrobial resistance than existing treatments as selective pressure is not exerted on the microorganisms in the same way. This promising antiadhesive strategy has been particularly explored in recent decades,2-4 with several studies reporting significant in vivo effects. Striking decolonization and antibiofilm effects have been reported on bacterial lectins such as the Shiga-like and cholera toxins,^{5,6} FimH expressed by pathogenic Escherichia coli strains implicated in urinary tract infections⁷⁻⁹ or inflammatory bowel diseases,¹⁰⁻¹² LecA and LecB expressed by Pseudomonas aeruginosa in the context of lung infections.^{13–15} This promising concept demonstrated on bacteria, has now reached the clinical stage, and may be potentially applied to other classes of pathogens such as fungi. Recently, Armstrong and coll. identified a lectin named FleA (or AFL) from A. fumigatus which shares homologies with a fucose-binding lectin from the orange peel mushroom Aleuria aurianta.¹⁶ The role of the lectin is still unclear and it was initially suggested to help the fungus to thrive in decaying matter, to be an important virulence factor involved in the early stage of colonization and to contribute to the inflammatory response.¹⁷ However, studies have also shown that FleA recognition by bronchial epithelial cells attenuates conidial germination,¹⁸ and that mucin binding and macrophage killing may prevent A. fumigatus pneumonia.¹⁹ Thus, there is a strong interest in developing molecular antagonists of FleA to fully probe the role of this lectin in the fungal pathogenicity and to assess antagonist therapeutic potential.

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X-ray diffraction of FleA co-crystallized with methyl-α-L-selenofucoside gave structural insights into this first characterized lectin from pathogenic fungi.¹⁷ The lectin forms homodimers of six-bladed β -propellers. Although the six sugar binding sites (one per blade) are non-equivalent and have specific oligosaccharide preferences, FleA shows a marked preference for α -anomeric fucosides.²⁰ We and others have exploited these specific features to design synthetic multivalent fucosides with the potential ability to interact simultaneously in several binding sites of FleA by a chelate binding mode (Fig. 1A).^{21–23} In particular, we recently showed that hexavalent fucosides based on a cyclodextrin (CD) core and optimized oligo(ethyleneglycol) spacers (Fig. 1B) are potent FleA antagonists and can decrease spore adhesion to pneumocytes at low micromolar concentrations.²¹ In the present work, we designed a second generation of hexavalent fucosides. A small library of monovalent fucosides 1-10 was first designed (Fig. 1C, Step 1) and their binding affinity for FleA assessed by isothermal microcalorimetry (ITC) (Step 2). The most promising monovalent ligand could be cocrystallized in complex with FleA and was selected to design the corresponding hexavalent fucoside (Step 3).

Results and discussion

FleA displays a slight preference for the α -fucosides compared with the corresponding β -anomers.²⁰ Thus, a Fischer glycosylation protocol was employed in the first step of the chemical

synthesis of the monovalent α -O-fucosides **1**, **3**, **4** and **6** (Scheme 1). Sulfuric acid immobilized on silica²⁴ catalysed α -addition of propargyl alcohols (synthesis in ESI†) to fucose, leading to the expected compounds **15–18** after acetate protection. The fucosides were engaged in a copper-catalysed azide-alkyne cyclization in the presence of sodium ascorbate to form exclusively 1,4-regioisomers **19–22** as shown by the large $\Delta(\delta_{c-4}-\delta_{c-5})$ observed by ¹³C NMR.²⁵



Scheme 1 Chemical synthesis of the monovalent α -O-fucosides 1, 3, 4, 6.



Fig. 1 (A) Multivalent fucosides binding the FleA target in a chelate fashion. (B) In a previous study, simple fucose sugar was appended to a cyclodextrin core.²¹ (C) In the present study, we designed a new generation of fucosides with varied hydrophobic groups at the anomeric position. A three-step process was used to obtain optimized and non-hydrolysable hexavalent fucosides.

Excess copper was removed from the crude product after overnight stirring with a chelex resin prior to chromatography on silica gel. The acetate groups were removed using a basic resin to form the monovalent α -*O*-fucosides **1**, **3**, **4** and **6**.

Thiofucosides 2, 5 and 7-10 were obtained from the common intermediate 1-thio-a-L-fucopyranose tetraacetate 23^{26} (Scheme 2). The thioacetate group was first deprotected as a thiolate 24 using potassium carbonate and sodium methanethiolate in methanol. Surprisingly, using a mixture of the two bases proved very efficient at forming 24. The thiolate was directly engaged in a thiol-ene click reaction with commercially available vinylic synthon 29 and 30 to form 31 and 32 with 57 and 50% yields, respectively. Then, the aromatic hydroxyl groups were alkylated with propargyl bromide 25 in presence of potassium carbonate to form 37 and 38. Compounds 33-36 were also obtained from the crude thiolate 24, directly engaged in a nucleophilic substitution reaction of bromo- or tosylated alkynyl-armed synthons 25-28 (synthesis in ESI[†]). The CuAAc protocol to form 19-22 was repeated with 33-38 and led to good yields of pure cycloadducts 39-44 and the expected thio-fucosides 2, 5, 7-10 after acetates hydrolysis with IRN78.

Once we had the monovalent fucosides **1–10** in hand, we assessed their affinity towards FleA by isothermal titration calorimetry (ITC). We initially performed docking experiments with the monovalent fucosides. Unfortunatly, the results (not shown) poorly correlated with ITC ranking. The theoretical predictive efficiency of the compounds is challenging on the FleA target possessing six non-equivalent fucose binding pockets. Dissociations constants K_d and relative affinity values ($\beta = K_{dMF}/K_{dCpd}$) towards α -methylfucoside (MF, $K_d = 110 \ \mu$ M)²¹ are presented Table 1. All of the compounds showed improved potency compared with MF, with β values ranging to 1.1 to 5.9. Switching the anomeric oxygen for a sulfur atom was detrimental with thio-compounds 5 and 7 showing around two fold higher K_d values compared with the *O*-analogues 4 and 6,

i) K₂CO₃, NaSMe ii) DPAP, THF MeOH, 0°C, 40 mi UV. 2H 2OAc 2ZOAG ACO OAC ÓAc 29 or 30 24 **31** (X₅) =57% **32** (X₆) =50% ii') K₂CO_{3,} CH₃CN, 25-28 (X1-X4) **31** (X₅) iii) K₂CO₃, 60°C, 2h ⊷о-н ACO OAC iv) 2ZOR ⊁он 32 (X₆) 25 CuSO₄, NaAsc R=Ac 39 (X₁) =50% 42 (X₄) =75% **33** (X₁) =63% **36** (X₄) =80% **34** (X₂) =60% **37** (X₅) =67% **40** $(X_2) = 67\%$ **43** $(X_5) = 56\%$ **41** $(X_3) = 78\%$ **44** $(X_6) = 66\%$ 35 (X₃) =64% 38 (X₆) =66% Building blocks v) IRN78, rt, 12H Br TsO R=H 26 (<mark>X</mark>2) 25 (X1) 29 (Xs) OH **2** (X₁) =93% **8** (X₄) =93% **5** (X₂) =90% **9** (X₅) =90% TsO 10 (X₆) =84% OH 27 (X₃) 30 (X₆) OMe 28 (X₄)

Scheme 2 Chemical synthesis of the monovalent α-S-fucosides.

Table 1 Binding affinity of compounds 1–10 towards FleA as determined by ITC

| R-2 N=N HO OH | | | | |
|---------------------|---|----------------------------------|-----|--|
| Cpd | R= | $K_{\rm d}$ (μ M) | β | |
| MF | _ | 110 ± 2 | _ | |
| 1 | 0 ~ 22° | 55.8 ± 1.8 | 2.0 | |
| 2 | S | $\textbf{48.8} \pm \textbf{4.1}$ | 2.2 | |
| 3 | 0~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~ | 53.8 ± 3.3 | 2.0 | |
| 4 | 0~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~ | 59.5 ± 0.0 | 1.8 | |
| 5 | s~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~ | 100.7 ± 0.7 | 1.1 | |
| 6 | 0-/0-}- | 46.7 ± 1.9 | 2.4 | |
| 7 | s | 90.6 ± 3.7 | 1.2 | |
| 8 | S O Start | 18.5 ± 0.6 | 5.9 | |
| 9 | S Const | 89.4 ± 3.2 | 1.2 | |
| 10 | S OMe | 103.9 ± 4.9 | 1.1 | |

respectively. This is however not a general rule, as 1 and 2 with $O-CH_2$ and $S-CH_2$ R groups displayed virtually identical affinity constants. The impact of methylene homologation was shown to be insignificant in the aliphatic series of compounds 1, 3 and 4 but dramatically reduced FleA affinity when comparing 8 and 9 with phenyl rings. Thus, subtly different interactions operate depending on the anomeric substituents and a larger library of fucosides may allow the identification of more potent submicromolar FleA antagonists.

Ligand 8 (K_d = 18 μ M), showed a significantly higher affinity for FleA compared with MF or compounds 1-7, 9 and 10. In consequence, we decided to study its binding mode with FleA in greater depth by X-ray crystallography. 8 is bound to five out of the six binding sites of FleA (Fig. 2). In site 4, we find a glycerol or a PEG molecule bound, since it is more deeply buried. The binding of compound 8 will be inhibited here by steric clashes of the aglycone moiety with the protein and in particular with the side chain of Tyr168. In molecule B, a glycerol moiety is also observed in site 6 that could result in binding difficulties due to the crystal contacts. In site 5, only electron density for the fucosyl moiety is visible. In sites 1-3, the phenyl moiety of ligand 8 could be modelled in the electron density. In site 1 of molecule A, we observed electron density up to the triazole ring of compound 8 (Fig. 2D, section 1A). No electron density was visible for the oligo(ethylene glycol) spacer as it is exposed to the solvent which leads to disorder. The interactions with the fucose moiety are the same as previously described.20 The phenyl moiety does not interact directly with the protein but is visible in sites 1-3 where some hydrophobic interactions would stabilize its conformation. In site 1, the aglycone displays two conformations: one where it makes hydrophobic contact with Leu39 and the side chain of



Fig. 2 Crystal structure of FleA in complex with ligand 8. (A) Representation of FleA dimer coloured by protomer. (B) FleA beta propeller coloured by blade represented in cartoon representation for protomer B, and (C) in surface representation for protomer A. (D) Zoom on FleA binding sites with 2Fo-DFc electron density represented at 1σ (0.4 eÅ³). Panels are labelled by site and protomer. Ligands are represented in balls and sticks.

Glu41 and one where the triazole ring stacks against the indole ring of His23 (Fig. 2B).

Based on the ITC results obtained with monovalent derivatives **1–10**, we decided to synthesize the CD fucosides **48** and **49**, which are hexavalent clusters of ligands **2** and **8** displaying the highest FleA affinity of the thiofucosides serie. Several scaffolds may be used for the efficient design of multivalent fucosides.²² α CD scaffolds are good candidates due to their hydrophilic nature and rigid core limiting nonspecific interactions with proteins and entropic penalty upon ligand

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binding. They are easily functionalized at the primary rim to present the sugar ligands with an angular spatial restriction.²⁷ In a previous study,²¹ we showed that azido- α CD analogues could be easily functionalized by simple alkyne *O*-fucosides using click chemistry techniques to form water-soluble hexavalent CD. We also optimized the spacer arm length, and the best affinity for FleA was obtained with a tetraethyleneglycol spacer. For these reasons, functionalized scaffold **45** was selected in the present study. Only *S*-fucosides were clusterized due to their potential inertness to glycosidase hydrolysis. The ligands were grafted by CuAAc on azido-functionalized CD **45**, obtained by a previously described protocole.⁹ Acetate deprotection on a basic resin followed by purification on a Sephadex column led to the expected hexavalent compounds **48** and **49** (Scheme 3).

Next, the binding affinity of 48 and 49 for FleA was assessed by ITC and compared to MF. In this assay, the reference compound MF showed a similar dissociation constant value as previously published²¹ (K_d = 98 vs. 110 μ M). Surprisingly, the differences in binding affinity of monovalent 2 and 8 for FleA disappeared when the compounds were multimerized on CD. Indeed, hexavalent derivatives 48 and 49 showed rather similar $K_{\rm ds}$ measured by ITC (Table 2). The fact that the relative inhibitory potency of the fucosides is not conserved when 2 and 8 are multimerized may be related to the non-equivalence of the six binding sites of FleA. Nevertheless, the hexavalent compounds were much more potent than their monovalent analogues, with K_{ds} reaching the nanomolar inhibition level. Each clustered fucoside was more than 400-fold more potent compared to MF. This sygnergistic effect may be explained by a partial chelation of the FleA binding sites as suggested by the lower than 1 average stoechiometry (n) values obtained for 48



Scheme 3 Chemical synthesis of the hexavalent α -S-fucosides 48 and 49.

Table 2 Binding affinity of compounds 48 and 49 towards FleA as determined by ITC

| Cpd | R= | $K_{\rm d}$ (nM) | β |
|-----|-------|------------------|-----|
| MF | _ | 98200 ± 0 | _ |
| 48 | S | 238.5 ± 28.5 | 412 |
| 49 | S O C | 223 ± 21 | 440 |

 $(n = 0.814 \pm 0.155)$ and **49** $(n = 0.784 \pm 0.006)$ from three independant ITC measurements.

Next we evaluated the potency of 49 to probe the respective role of FleA in A. fumigatus pathogenesis. In a previous study, Balloy and co-workers investigated the mechanisms of A. fumigatus conidia clearance from the lung.¹⁸ Human bronchoepithelial cells were shown to inhibit the filament formation of extracellular A. fumigatus (conidia germination into hyphae) via FleA recognition. FleA from conidia bound at cell surfaces led to cell receptors clustering with enhanced avidity, triggering a fungistatic process that involves the phosphoinositide 3-kinase pathway. This antifungal activity could be abolished by cell-receptors saturation, with recombinant FleA. Here, we assessed if a similar effect could occur after specific binding of 49 to FleA from conidia. The level of conidia germination into hyphae (filamentation) was followed by the release of galactomannan in the medium (Fig. 3). Pre-incubation of BEAS-2B epithelial cells with FleA (1 µM) before (1 hour) and during infection (15 hours) with A. fumigatus conidia led to a much higher level of filamentation, as previously reported,¹⁸ and shown by a higher galactomannan level (Fig. 3, Af + FleA). Interestingly, a similar trend was observed when 49 (100 μ M), instead of FleA, was co-incubated with the conidia and the cells (Af + 49). This inhibition of the fungistatic process was however not observed with the less potent FleA antagonist MF at 100 μ M (Af + MF). Thus, these preliminary results suggest that optimized multivalent fucosides such as 49 are valuable probes to study A. fumigatus pathogenesis.



Fig. 3 Effect of FleA inhibition on galactomannan release. Values, expressed in optical density (O.D.), are presented as mean \pm SEM ****p < 0.0001, *p < 0.05, ns: not significant (ANOVA test, followed by Bonferroni's multiple comparison test).

Conclusions

Fucose-binding lectins expressed by life-threatening pathogens such as B. cenocepacia or A. fumigatus have attracted attention in view of their potential implications in respiratory infections.^{19,21,22,28,29} The role of the recently identified FleA lectin in A. fumigatus pathogenesis is however still unclear. Published studies suggest that FleA-promoted binding may actually attenuates A. fumigatus virulence, by improving mucocilliary clearance, macrophage killing, and/or inhibition of conidia germination to hyphae.^{18,19} Thus new chemical probes are required to better understand the complex role(s) of FleA during A. fumigatus infections. In this work, we developed nonhydrolysable hexavalent thiofucosides with nanomolar affinity for FleA. Insights in the binding mode of the fucoside ligands were provided by X-ray crystallography and ITC experiment. Coincubation of an optimized hexavalent fucoside (49) with A. fumigatus conidia was shown to restore hyphae development, further suggesting a role of the lectin in host response to tissue invasion.

Author contributions

Conceptualization, supervision and writing: S.G. and A.V. revising: D. D. acquisition and formal analysis: C.D.; P-A. L. and V. L. (chemistry), A.V. crystallography, A.C. and E.G. (ITC), V.B. (cell assay).

Conflicts of interest

There are no conflicts of interest to declare.

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