Synthesis and Structural Analysis of 5-Deoxy-3-O-methyl-5-C-[(R)- and (S)-phenylphosphinothioyl]- α - and β -D-xylopyranoses

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Treatment of 5-deoxy-5-iodo-1,2-*O*-isopropylidene-3-*O*-methyl-α-D-xylofuranose with ethyl phenylphosphinothioate in the presence of NaH in DMF gave a 1:1 mixture of the 5-deoxy-5-*C*-[(*R*)- and (*S*)-(ethoxy)-phenylphosphinothioyl] derivatives. Reduction of these D-xylofuranoses with sodium dihydrobis(2-methoxy)aluminate, followed by the acid hydrolysis, provided the title compounds, which are the first sugar analogues having a phosphinothioyl group in the hemiacetal ring. These compounds were converted into four, separable 1,2,4-tri-*O*-acetates, the structures and conformations of which were established by spectroscopy. The corresponding per-*O*-acetyl-5-deoxy-5-*C*-[(*R*)- and (*S*)-phenylphosphinyl] analogues were also prepared and their previously presented structures were revised. Complete ¹H NMR (500-MHz) parameters of these two types of compounds are given for a comparative, structural study.

A large number of sugar analogues having a phosphorus atom in place of oxygen in the hemiacetal ring have been prepared in recent years:1) e.g., Dglucopyranose type 12) and D-ribofuranose type 2.3) Such compounds are of interest from the viewpoint of their physicochemical properties and potential biological activity.¹⁾ All of these analogues possess an alkyl-, aryl-, or hydroxyphosphinyl group in the hemiacetal ring, but no derivatives containing a phosphinothioyl group in the ring have been prepared so far. On the other hand, it is well-known that many organophosphorus insecticides of P=S type, such as fenithion4) (3) and EPN5) (4), are in wide use and the P=S functional group of these compounds is effective in the relatively facile permeability through the skin of insects and then the slow oxidative desulfurization to generate the active P=O form by the action of a microsomal oxidase in vivo.6) In this paper we wish to describe a detailed study on the two synthetic routes to the first sugar analogue having a phosphinothioyl group in the hemiacetal ring, as well as a comparative study with the corresponding 5-C-

(phenylphosphinyl) derivatives, taking a 3-O-methylp-xylopyranose analogue as a model compound.⁷⁾

Results and Discussion

After examination of various reaction conditions, introduction of a phosphinothioyl group to the model compound was found to be best achieved by the reaction of 5-deoxy-5-iodo-1,2-O-isopropylidene-3-O-methyl- α -D-xylofuranose⁸⁾ (5) with ethyl phenylphos-

Scheme 1. Synthesis of 5-deoxy-3-O-methyl-5-C-(phenylphosphinothioyl)-p-xylopyranoses.

phinothioate⁹⁾ in dry N,N-dimethylformamide (DMF) containing an equivalent of sodium hydride at 0°C for 3 h under argon (see Scheme 1). Thus the 5-C-[(R)and (S)-(ethoxy)phenylphosphinothioyl] derivatives 6a,b were obtained in 84% yield as a 1:1 mixture which were chromatographically inseparable. Compound 6 was obtained in less satisfactory yields when the reaction was carried out at room temp in DMF (3 h, 40% yield) or at elevated temps in benzene or xylene (19 h, ca. 9% yield). Complete ¹H NMR spectral parameters of both 6a and 6b were obtained at 500-MHz and are summarized in Table 1, by taking into consideration the similar spectral data of the 5-C-[(ethoxy)phenylphosphinyl] analogues 7a and 7b which were separable by chromatography (see later). On inspecting the molecular models of these P-epimers (see formulas 6a and 6b), the characteristic upfield shift of the δ value for MeO-3 of **6a** (δ 2.99) from the normal value of 6b (δ 3.41) strongly suggests the most likely configurations of the P atom of 6a and **6b** to be (R) and (S), respectively, because a stronger shielding effect on the MeO-3 group apparently exists in **6a** by the *P*-phenyl group in the vicinity.

Reduction of 6 with sodium dihydrobis(2-methoxyethoxy)aluminate (SDMA) in dry benzene or tetrahydrofuran (THF) was found to proceed much slower than that of the alkoxyphosphinyl analogues which normally completes within 30 min at 0 °C.¹⁻³⁾ Thus, the reduction was carried out at 20 °C for 17 h to give the unstable 5-C-(phenylphosphinothioyl) compound 8. Other conditions such as with SDMA in benzene (at 50 °C for 30 h or 20 °C for 35-44 h), SDMA in THF (0 °C, 26 h) or LiAlH₄ in ether (20 °C, 60 h) have been found so far to result in less satisfactory yields of 8 from 6; in most cases the starting material 6 remains unchanged. The acid-catalyzed deprotection of 1,2-Oisopropylidene group of 8 and the subsequent ringtransposition in oxygen-free 0.5 M HCl-EtOH (1:1 v/v) (1 M=1 mol·dm⁻³) at 80 °C for 2 h provided 5deoxy-3-O-methyl-5-C-(phenylphosphinothioyl)-D-xylopyranoses (10).

The structural assignment of **10** was made by converting them into the 1,2,4-tri-O-acetates **12** by treatment with acetic anhydride-pyridine at 20 °C for 20 h as in the case of the previous phosphorus-containing sugar analogues.¹⁻³⁾ Purification of the

crude mixture by column chromatography on silica gel using ethyl acetate-hexane as eluant and then recrystallization gave four, crystalline diastereomers: 5-deoxy-3-*O*-methyl-5-*C*-[(*S*)-phenylphosphinothioyl]- α -D-xylopyranose (12a, 1.8% overall yield from 6), the 5-C-[(S)-P]- β - (12b, 1.8% yield), the 5-C-[(R)-P]- β -(12c, 3.4% yield), and the 5-C-[(R)-P]- α -isomer (12d, 2.3% yield). The identical molecular composition (C₁₈H₂₃O₇PS) of **12a**—**d** was confirmed by the highresolution MS. The precise configurations and the ${}^4C_1(D)$ conformation of 12a—d were established on the basis of the 500-MHz ¹H NMR spectra by taking into account the known parameters of structurally similar compounds obtained before; e.g., the four diastereomers (with respect to the C-1 and ring-P atom) of per-O-acetyl-5-deoxy-5-C-(methoxyphosphinyl)-3-O-methylp-xylopyranoses^{8,10)} (14) and other related compounds such as 5-C-[(S)-(phenylphosphinyl)]- α -L-idopyranoses¹¹⁾ (15) and L-iditol¹¹⁾ (16). The complete NMR assignments of 12a-d are listed in Table 1 and will be discussed later.

Meanwhile, the corresponding four 5-C-phenylphosphinyl derivatives 13a-d of the above thioyl compounds have been prepared by Seo, 12) employing 1,2-O-isopropylidene-3-O-methyl- α -D-xylo-pentodialdo-1,4-furanose¹³⁾ (17) as the starting material by sequence $17 \rightarrow 19 \rightarrow 20 \rightarrow 21 \rightarrow 9 \rightarrow 11 \rightarrow 13$; structural assignments of the final products were made on the evidence of the insufficiently resolved 60-MHz ¹H NMR data and the values of optical rotations. ¹²⁾ Having found some ambiguity in a part of the reported NMR assignments for 13a-d and felt necessity of a comparative study of the exact parameters of 12a-d with those of 13a-d for further structural proof for the former compounds, we decided to prepare 13a-d, by using 7 as the starting material. Thus, 7 was readily obtained from 514) by the Michaelis-Arbusov reaction with diethyl phenylphosphonite in 91% yield as a 1:1 mixture of the two

Table 1. ¹H NMR (500-MHz) Parameters for **6**, **7**, **12**, and **13** in CDCl₃^{a)}

	T	able 1.	¹H NMR	(500-MH			for 6 , 7 ,	12 , and	13 in CD	Cl ₃ a)		
					Chem	ical s	hift					
Compd	H-l	H-2		H-3 H-4			H-5 (H _a -5	H-5' H _e -5)		Me ₂ C-O-1, 2 (AcO-1, 2, 4) ^{b)}		
6a	5.80	4.4		3.49	4.62		2.69	2.58	1.50,	1.3		
6 b	5.75	4.5		3.73	4.39		2.78	2.49	1.40,	1.2		
7a	5.80	4.5		3.52	4.53		2.47	2.38	1,47,	1.2		
7b	5.78	4.5		3.66	4.32		2.52	2.31	1.39,	1.2		0.00
12a	5.68	5.9		3.67	5.68 5.65		2.91	2.60)	(1.94,	2.1		2.03)
12b 12c	5.79 5.77	5.7 5.0		3.57 3.53			2.44 2.37	2.61) 3.23)	(1.91, (2.16,	2.0 2.1		2.07) 2.07)
12d	6.11	3.0 4.7		3.65	4.83	(2.56		3.16)	(2.10,	2.1		2.07)
13a	5.76	5.6		3.69	5.54		2.49	2.68)	(1.98,	2.1		2.04)
. 13b	5.60	5.7		3.58	5.53		2.05	2.70)	(1.93,	2.0		2.07)
13c	5.80	5.0	9 :	3.57	4.99	(2.33	3.01)	(2.15,	2.0	8,	2.05)
13d	6.10	4.7		3.70	4.86		2.55	3.00)	(2.25,	2.1		2.06)
Compd	MeO-3 HC-O-I						e-C-O-P			$\frac{\mathrm{Ph}(o, m, p)}{2}$		7.50
6a 6b	2.99	2.99 4.12 3.41 4.12		3.78 3.79		1.28 1.27		7.92 7.93		7.43, 7.48,		7.53 7.53
7a		3.06 4.09		3.86		1.29		7.81		7.48,		7.55 7.55
7b		3.39 4.11		3.89		1.31		7.81,		7.48,		7.55
12a	3.51				-			7.83	,	7.49,		7.57
12b	3.51							7.89		7.51,		7.58
12 c	3.47							7.95		7.56,		7.58
12d	3.47							7.86		7.59,		7.59
13a		3.52						7.72,		7.50,		7.60
13b	3.53							7.75		7.52,		7.60
13c 13d	3.49 3.48							7.95 7.89		7.57, 7.60,		7.6 4 7.65
	3.10							7.03	, 			
					Coupli	ng co		7	7		7	T_
Compd	$J_{1,2}$	$J_{1,P}$	J _{1,5e}	$J_{2,3}$	J _{2,P}	J _{3,4}	$J_{4,5a}$ $(J_{4,5}$	J _{4,5e} J _{4,5}	J _{4,P}	$J_{5a,5e} \ J_{5,5'}$	J _{5a, P} J _{5, P}	J _{5e, P} J _{5'P})
6a	3.0			0		3.0	(9.1	4.4	8.7	14.5	10.6	17.8)
6b	3.9			0		3.0	(8.7	4.6	9.5	14.2	15.5	14.2)
7a 7b	3.9 3.9			0 0		3.1 3.1	(9.5 (9.0	4.7 5.0	6.4 7.2	14.8 14.9	12.8 16.6	17.7)
76 12a	2.8	7.5	2.3	10.2	0	9.7	11.8	3.9	3.0	13.1	10.6	14.8) 15.9
12b	11.0	0.3	0	9.4	3.4	9.7	11.8	4.2	3.3	13.8	8.5	16.7
12c	10.9	6.6	0	8.8	5.2	9.4	11.9	4.1	3.9	15.3	13.8	15.3
12d	2.6	7.2	2.2	10.1	0	9.4	12.1	4.2	2.8	15.2	15.1	14.5
13a	2.8	10.8	2.3	10.1	0	9.7	12.0	4.3	2.2	14.0	6.4	18.3
13b	11.0	2.7	0	9.5	2.8	9.6	12.3	4.3	2.9	14.5	4.2	19.5
13c 13d	10.3 2.6	11.3 8.9	0 2.0	8.3 9.3	5.8 3.6	8.9 8.7	11.4 11.2	4.3 4.3	5.5 5.5	15.1 15.0	17.5 18.5	15.4 15.0
Compd	P-OEt								P-Ph			
<u> </u>	$J_{ m H,H'}$	$J_{ m P,H}$	$J_{ m H,Me}$	$J_{ m P,H'}$	$J_{ m H',N}$		$J_{P,2}$	J _{P,3}	J _{P,4}	$J_{2,3}$	$J_{3,4}$	$J_m^{c)}$
6a	10.2	9.8	7.1	8.5	7.0		13.1	3.3	1.8	7.0	7.5	1.5
6 b	10.2	9.8	7.1	8.7	7.0		13.1	3.3	1.8	7.0	7.5	1.5
7a 7b	10.1	7.1	7.1	7.1	7.1		12.1	3.3	1.4	8.3	7.5	1.5
76 12a	10.1	7.1	7.1	7.1	7.1		11.8 13.4	3.3 3.5	1.5 1.8	7.5 7.5	7.5 7.7	1.5 1.5
12b							13.4	3.3	2.0	7.3	7.7	1.5
12c							12.8	3.3	1.8	8.1	7.8	1.5
12d							12.8	3.2	2.0	8.0	7.5	1.5
13a							11.8	3.3	1.5	7.5	7.5	1.5
13b							12.5	3.3	1.7	7.5	7.5	1.5
13c 13d							11.8	3.3	1.7	7.7	7.7	1.5
							12.1	3.2	1.6	7.5	7.7	1.5

a) The assignments of all signals were made by employing a first-order analysis with the aid of decoupling technique and the parameters were confirmed by a computer-assisted simulation analysis. Very slight corrections were made on a few of the parameters for 12c, d that were reported in Ref. 7, because the spectra of pure 12c and 12d have now become available. b) The assignments of acetoxyl groups may have to be interchanged. c) Approximate J values for meta-coupling of the P-phenyl ring protons: $J_{2,4'}$, $J_{2,6'}$, and $J_{3,5}$.

diastereomers, which were separable by chromatography into pure components 7a and 7b. From the analysis of the complete 1H NMR data (see Table 1) the most likely configuration of the phosphorus atom of 7a and 7b were assigned to (R) and (S), respectively, on the same grounds taken into consideration in the case of 6a,b shown above.

For the preparative purpose, however, the crude mixture 7 (without separation) can be subjected to the reduction with SDMA, because both 7a and 7b afford the identical 5-deoxy-5-C-phenylphosphinyl compound 912) (see Scheme 1). Thus, after following the usual procedures,1-3) 9 was led to 11 and then to 13, from which the four diastereomers, 13a (colorless needles, mp 202-203 °C, 11% overall yield from 7), 13b (colorless needles, mp 296—298 °C, 11% yield), 13c (colorless syrup, 5% yield), and 13d (colorless needles, mp 148-149 °C, 6% yield) were separated by silica-gel chromatography. As in the case of 12a-d shown above, these compounds were characterized by MS and 500-MHz ¹H NMR (see Table 1 for the summarized parameters) as 5-deoxy-3-O-methyl-5-C-[(S)-phenylphosphinyl]- α -D-xylopyranose (13a), 5-C-[(S)-P]- β -(13b), 5-C-[(R)-P]- β - (13c), and 5-C-[(R)-P]- α - isomer (13d), in contrast to the previously reported assignments¹²⁾ of the structures 5-C-[(R)-P]- β -, 5-C-[(R)-P]- α -, 5-C-[(S)-P]- β -, and 5-C-[(S)-P]- α - isomer, respectively.

The ¹H NMR data of 12 and 13 generally follow the characteristic features in the δ and J values for 5deoxy-5-C-(alkyl-, aryl-, and methoxyphosphinyl)-Daldopyranoses approximately in the ${}^4C_1(D)$ conformation. 1-3) Namely, the axial orientations (S) of the ring P=S and P=O groups in 12a,b and 13a,b are established by the downfield shift (0.7—1.1 ppm) of the H-2 and H-4 signals from those of 12c,d and 13c,d having equatorial P=S and P=O groups (see the δ values in Table 1). The α -orientation of C-1 is derived by considering the small magnitudes of $J_{1,2}$ (2.6—2.8 Hz) and $J_{1,5e}$ (2.0–2.3 Hz) values of 12a,d and 13a,d, whereas the β -anomers 12b,c and 13b,c show large $J_{1,2}$ (10.3-11.0 Hz) and negligible $J_{1,5e}$. It is noteworthy to mention some characteristic features in the spectral data of 12 and 13 more in detail for the purpose of comparison:

1) The proton signals of H-2 and H-4 of 12a and

12b appear at a slightly lower field (0.1-0.3 ppm) compared with those of 13a and 13b. The axial P=S group in the ${}^4C_1(D)$ conformation therefore seems to excert a larger down-field shift on the signals of protons at the 1,3-diaxial positions in comparison with the same effect by the axial P=O group.

- 2) Appreciable differences in the magnitudes of the $J_{1,P}$, $J_{5a,P}$, and $J_{5e,P}$ are observed between the corresponding diastereomers of 12a-d and 13a-d (e.g., compare the $J_{1,P}$ values of 12a and 13a).
- 3) All of the AcO-1 signals of 12a,b and 13a,b appear at a rather high field (δ 1.91—1.99), indicating that these acetoxyl groups are most likely to be situated at the positions above the plane of the vicinal, equatorial P-phenyl ring. Significant down-field shifts (ca. 0.4 ppm) are observed for the δ values of H_a-5 of 12a.b from those of the corresponding 13a.b which seem to have normal values. Taking into consideration of the known values of the torsion angles (determined by X-ray crystallography^{1,11)}) between the P=O bond and the plane of the P-phenyl ring of compounds 15 and 16 (see Fig. 1), the most probable torsion angle between P=S and the plane of the P-phenyl ring of 12a and 12b in CDCl₃ solution is illustrated in Fig. 1, which would account for the above unusual δ values of Ha-5 and AcO-1 of these compounds.

However, exact reasons for these interesting features of the NMR parameters of the P=S compounds compared with those of the P=O analogues have remained to be further clarified employing a larger number of similar sugar analogues; particularly, the relation to their ${}^4C_1(D)$ conformational deviations as well as intrinsic characters of P=S bonding affecting

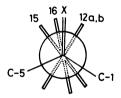


Fig. 1. "Newman" projection down the P-C(phenyl) bond illustrating the phenyl ring. [The digits mark the phenyl rings for compounds 15 (15, X=O),¹¹⁾ 16 (16, X=O),¹¹⁾ 12a, and 12b (12a, b, X=S)].

TsNHNH

HC=NNHTs

HC
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Scheme 2. Synthesis of 5-deoxy-3-O-benzyl-5-C-(ethylphosphinothioyl)-p-xylopyranoses.

the δ and J values of the neighboring groups would be of interest. Nevertheless, these exact parameters for **12a**—**d** are certainly of value in determining other 5-deoxy-5-C-phosphinothioyl-D-aldopyranoses that are still unknown.

Another approach for the synthesis of the sugar analogues having a P=S group in the ring was also made (see Scheme 2) utilizing a similar scheme employed for the synthesis of D-ribofuranose analogues³⁾ 2. Namely, condensation of 3-O-benzyl-1,2-Oisopropylidene- α -D-xylo-pentodialdo-1,4-furanose¹⁵⁾ (18) with tosylhydrazine in methanol at 20 °C for 2 h gave the hydrazone 22 (85% yield), which was treated with ethyl ethylphosphinothioate9) in the presence of trifluoromethanesulfonic acid at 0-20°C for 7 h, thus affording the 5-C-[(ethoxy)ethylphosphinothioyl] derivative 23 (73%). Although reduction of 23 with sodium borohydride in THF was repeatedly examined at 25-50 °C for 7-48 h, the desired compound 24 was obtained in fluctuating yields (10-55%) besides a large amount of unidentified by-products. The same treatment of 24 (as in the case of 6 and 7) with SDMA, followed by the 0.5 M HCl-EtOH and then acetic anhydride-pyridine afforded a diastereomeric mixture of 1,2,4-tri-O-acetyl-3-O-benzyl-5-deoxy-5-C-[(R)- and (S)-ethylphosphinyl]- α , β -D-xylopyranoses (25, colorless syrup) in only 1.6% overall yield from 24. Compounds 25 were not separable into each diastereomers by chromatography. Therefore the synthetic approach shown in Scheme 2 seems to be less satisfactory than that described in Scheme 1.

Although improvement of the yields of some steps particularly of the reduction and the subsequent ringenlargement reactions of the D-xylofuranoses **6** and **24** remains to be done, present work demonstrates an effective way for the preparation of 5-deoxy-5-*C*-(phenylphosphinothioyl)-D-xylopyranoses from an appropriate precursor. This scheme is therefore expected to be applicable to the preparation of various other 5-deoxy-5-*C*-phosphinothioyl-D- and L-aldofuranoses and pyranoses, which is currently under investigation.

Experimental

Melting points were determined with a Yanagimoto MP-S3 instrument and were uncorrected. All reactions were monitored by TLC (Merck silica gel 60F, 0.25 mm) with an appropriate solvent. Column chromatography was performed with Wako C-200 silica gel [or Merck Lobar prepacked silica gel (Size A) for rechromatography of 12 and 13]. The ¹H NMR spectra were measured in CDCl₃ with a Hitachi R-600 (60-MHz, FT) or Varian VXR-500 instrument (500-MHz, the CS-NMR Lab., Okayama Univ.) at 27 °C. Chemical shifts (at 60-MHz unless otherwise specified) were recorded as δ values relative to tetramethylsilane as internal standard. The mass spectra were taken on a Shimadzu LKB-9000 low-resolution or a JEOL JMX-HX100 high-resolution instrument and were given in terms of m/z

(rel intensity) compared with the base peak.

5-Deoxy-5-C-[(R)- and (S)-(ethoxy)phenylphosphinothioyl]-1,2-O-isopropylidene-3-O-methyl-α-D-xylofuranoses (6a, b). In a 100 cm³ two-necked flask was placed sodium hydride (60% in oil, 688 mg, 1.2 equiv), which was then washed twice with dry hexane under argon and diluted with dry DMF (10 cm³). To this suspension was dropwise added a solution of ethyl phenylphosphinothioate9) (4.00 g, 1.5 equiv) in dry DMF (10 cm³) at room temp under argon. After the consumption of sodium hydride to form the sodium salt of the thioate, a solution of 58 (4.50 g) in dry DMF (10 cm³) was dropwise added at 0 °C, followed by stirring at this temp for 3 h. The solvent was evaporated in vacuo (pump). The residue was dissolved in chloroform and washed with water. The organic layer was dried (Na₂SO₄) and evaporated in vacuo. The residue was purified by column chromatography with 1:3 AcOEt-hexane, giving an inseparable, 1:1 (by ¹H NMR) mixture of 6a and 6b (presumably 5-C-[(R)-P] and (S)-P diastereomer, respectively): A colorless syrup (4.50 g, 84%); R_1 =0.65 (1:2) AcOEt-hexane); ¹H NMR (500-MHz), see Table 1; MS m/z 373 (M+1; 27.5), 372 (~0), 340 (42), 312 (12), 311 (15), 287 (17), 282 (25), 186 (100), 185 (18), 157 (39), 153 (41), and 142 (21).

Found: m/z 373.1190. Calcd for $C_{17}H_{26}O_5PS$: M+1, 373.1239.

1,2,4-Tri-O-acetyl-5-deoxy-3-O-methyl-5-C-[(R)- and (S)-phenylphosphinothioyl]-α,β-D-xylopyranoses (12a—d). A solution of SDMA (490 mg, 70% in toluene, 3.0 equiv) in dry benzene (3 cm³) was dropwise added into a solution of 6 (300 mg) in dry benzene (5 cm³) at 5 °C under argon, followed by stirring at 20 °C for 18 h. Cold water (ca. 0.5 cm³) was slowly added at 0 °C under stirring to decompose excess SDMA. The resulting white ppt was centrifuged and washed with benzene. Combined benzene solutions were concentrated in vacuo to give crude 5-deoxy-1,2-O-isopropylidene-3-O-methyl-5-C-[(R)- and (S)-phenyl-phosphinothioyl]-α-D-xylofuranoses (8) as a colorless syrup.

This was immediately dissolved in oxygen-free ethanol $(5\,\mathrm{cm^3})$ and $0.5\,\mathrm{M}$ hydrochloric acid $(5\,\mathrm{cm^3})$. The mixture (pH ca. 1) was degassed with argon and then refluxed for 2 h. The cooled mixture was neutralized by adding enough Amberlite IRA-45. The resin was filtered off and washed with aq ethanol. The filtrate was evaporated in vacuo to give crude 5-deoxy-3-O-methyl-5-C-[(R)- and (S)-phenyl-phosphinothioyl]- α,β -D-xylopyranoses (10) as a colorless syrup.

This was dissolved in pyridine (2 cm^3) and acetic anhydride (1 cm^3) at 5 °C. The mixture was stirred at 20 °C overnight, diluted with a small amount of cold water, and concentrated in vacuo. The residue was dissolved in chloroform and washed with water. The organic layer was dried (Na_2SO_4) and evaporated in vacuo. The residue was chromatographed with 1:2 AcOEt-hexane to give 12a-d; 12d was separated from 12c by recrystallization (three times).

12a (5-*C*-[(*S*)-Phosphinothioyl]-α-isomer): Colorless needles (6.0 mg, 1.8% overall yield from **6**); mp 177—178 °C (from 1:1 AcOEt-hexane); R_i =0.24 (1:1 AcOEt-hexane); ¹H NMR (500-MHz), see Table 1; MS m/z 414 (M+; 12.5), 382 (12), 356 (16), 355 (38), 324 (29), 323 (56), 322 (57), 313 (20), 294 (19), 281 (52), 280 (100), 265 (27), 264 (28), 239 (43), 238 (28), 222 (31), 171 (22), 87 (23), 71 (31), and 43 (68).

Found: m/z 414.0930. Calcd for C₁₈H₂₃O₇PS: M, 414.0902.

12b (5-*C*-[(*S*)-Phosphinothioyl]-*β*-isomer): Colorless needles (6.0 mg, 1.8% yield from **6**); mp 166.5—167 °C (from 1:1 AcOEt-hexane); R_i =0.31 (1:1 AcOEt-hexane); ¹H NMR (500-MHz), see Table 1; MS m/z 414 (M+; 8.3), 355 (14), 323 (14), 322 (56), 281 (35), 280 (100), 239 (29), 238 (29), and 43 (51).

Found: m/z 414.0825. Calcd for C₁₈H₂₃O₇PS: M, 414.0902. **12c** (5-C-[(R)-Phosphinothioyl]- β -isomer): Colorless needles (7.7 mg, 2.3% yield from **6**); mp 231—232 °C (from 1:1 AcOEt-hexane); R_1 =0.44 (1:1 AcOEt-hexane); ¹H NMR (500-MHz), see Table 1; MS m/z 414 (M+; 2.5), 355 (13), 354 (38), 295 (72), 294 (45), 254 (19), 253 (100), 252 (66), 157 (20), 155 (66), 113 (68), and 43 (72).

Found: m/z 414.0914. Calcd for C₁₈H₂₃O₇PS: M, 414.0902. **12d** (5-C-[(R)-Phosphinothioyl]- α -isomer): Colorless needles (11.3 mg, 3.4% yield from **6**); mp 181—182 °C (from 1:1 AcOEt-hexane); R_1 =0.44 (1:1 AcOEt-hexane); ¹H NMR (500-MHz), see Table 1; MS m/z 414 (M+; 30.7), 354 (16), 296 (24), 295 (54), 294 (49), 281 (23), 253 (94), 252 (79), 237 (53), 128 (40), 127 (45), 113 (96), 71 (84), and 43 (100).

Found: *m/z* 414.0902. Calcd for C₁₈H₂₃O₇PS: M, 414.0902. 5-Deoxy-5-*C*-[(*R*)- and (*S*)-(ethoxy)phenylphosphinyl]-1,2-*O*-isopropylidene-3-*O*-methyl-α-D-xylofuranoses (7a,b). A mixture of 5 (500 mg) and diethyl phenylphosphonite (1.58 g, 5.0 equiv) was heated at 160 °C for 4 h under nitrogen and then concentrated below ca. 80 °C in vacuo. The residue was dissolved in ether and washed with water. The organic layer was dried (Na₂SO₄) and evaporated in vacuo. The residue was chromatographed with 1:1 AcOEt-hexane to give a 1:1 (by ¹H NMR) mixture of 7a and 7b as a colorless syrup (516 mg, 91% yield). The mixture was separated by careful rechromatography into the pure diastereomers.

7a (5-*C*-[(*R*)-P] isomer): Colorless syrup (170 mg, 30%); R_f =0.28 (2:1 AcOEt-hexane); ¹H NMR (500-MHz), see Table 1; MS m/z 356 (M⁺).

7b (5-*C*-[(*S*)-P] isomer): Colorless syrup (190 mg, 34%); R_1 =0.23 (2:1 AcOEt-hexane); ¹H NMR (500-MHz), see Table 1; MS m/z 356 (M⁺).

1,2,4-Tri-O-acetyl-5-deoxy-3-O-methyl-5-C-[(R)- and (S)-phenylphosphinyl]- α , β -p-xylopyranoses¹²⁾ (13a—d). Similar procedures to those for 12 from 6 were employed. Thus, 7 (320 mg) was treated with SDMA (218 mg, 1.2 equiv) at 5 °C for 30 min, giving 9 as a colorless syrup. This was refluxed with 1:1 ethanol-0.5 M HCl (10 cm³, oxygen-free) to afford 11 as colorless syrup. This was acetylated with Ac₂O-pyridine to give a mixture of 13, which was separated by column chromatography with a gradient solution of 2:1 AcOEt-hexane \rightarrow AcOEt \rightarrow 19:1 AcOEt-EtOH.

13a (5-*C*-[(*S*)-Phenylphosphinyl]-α-isomer): Colorless needles (39 mg, 11% yield from 7); mp 202—203 °C (from 1:1 AcOEt-hexane, Ref. 12, mp 295—296 °C); R_t =0.44 (AcOEt); ¹H NMR (500-MHz), see Table 1; MS m/z 398 (M+).

13b (5-*C*-[(*S*)-P]-*β*-isomer): Colorless needles (40 mg, 11% yield from 7); mp 296—298 °C (from 1:1 AcOEt-hexane, Ref. 12, mp 200—200.5 °C); R_t =0.44 (AcOEt); ¹H NMR (500-MHz), see Table 1; MS m/z 398 (M+).

13c (5-C-[(R)-P]-β-isomer): Colorless syrup (17 mg, 5% yield from 7); R_i =0.33 (AcOEt); ¹H NMR (500-MHz), see Table 1; MS m/z 398 (M⁺).

13d (5-C-[(R)-P]- α -isomer): Colorless needles (20 mg, 6% yield from 7); mp 148—149 °C (from AcOEt-hexane, Ref.

12, syrup); R_i =0.22 (AcOEt); ¹H NMR (500-MHz), see Table 1; MS m/z 398 (M⁺).

3-*O*-Benzyl-1,2-*O*-isopropylidene-α-D-xylo-pentodialdo-1,4-furanose 5-Tosylhydrazone (22). To a solution of 18^{15} (300 mg) in dry methanol (15 cm³) was added tosylhydrazine (200 mg, 1.07 equiv) at 0 °C. The mixture was stirred at room temp for 2 h and then evaporated in vacuo. The residue was dissolved in chloroform, dried (Na₂SO₄), concentrated in vacuo, and then purified by column chromatography, giving 22 as colorless needles (406 mg, 85% yield): mp 129—132 °C (from AcOEt-hexane); R_f =0.68 (1:1 AcOEt-hexane); 1 H NMR δ=1.31, 1.48 (6H, 2s, Me₂C), 2.42 (3H, s, Me-C₆-S), 4.02—4.80 (5H, m, H-2,3,4, CH₂-O-3), 5.94 (1H, d, $J_{1,2}$ =3.4 Hz, H-1), 7.15, 7.82 (2H, each, 2brd, J=8 Hz, C_6H_4 -S), 7.25 (6H, brs, C_6H_5 -C-O-3, NH), and 8.13 (1H, s, H-5); MS m/z 446 (M+).

(R)- and (S)-3-O-Benzyl-5-deoxy-1,2-O-isopropylidene-5- $C-\lceil (R) - \text{and } (S) - (\text{ethoxy}) + \text{ethoxphinothiov} \rceil - S-C-\lceil (2-\text{tosy}) - S-C-\lceil (2-\text{tosy}) \rceil - S-C-\lceil$ hydrazino]- α -p-xylo-pentofuranose (23). A mixture of 22 (200 mg) and ethyl ethylphosphinothioate⁹⁾ (320 mg, 5 equiv) was degassed with argon. Trifluoromethanesulfonic acid (25 mg, 0.4 equiv) was dropwise added to the above mixture at 0°C with stirring under argon, followed by stirring at 20 °C for 7 h. The mixture was diluted with chloroform (ca. 30 cm³) and neutralized by washing with cold, saturated aq NaHCO3. The organic layer was dried (Na₂SO₄) and concentrated in vacuo. The residue was purified by column chromatography with 1:2 AcOEt-hexane, giving 23 as a yellow syrup (192 mg, 73%): R_1 =0.68 (1:1 AcOEt-hexane); ¹H NMR δ =0.65—1.65 (6H, m, MeC-P, MeC-O-P), 1.28, 1.48 (6H, 2s, Me₂C), 1.7-2.4 (2H, m, C-CH₂-P), 2.39 (3H, s, MeC₆-S), 3.4-5.0 (7H, m, H-2,3,4, CH₂-O-P, NH), 4.64 (2H, s, CH₂-O-3), 5.93 (1H, d, $I_{1.2}$ =3.5 Hz, H-5), and 6.8—7.9 (10H, m, H-5, C₆H₅-C-O-3, C_6H_4-S); MS m/z 584 (M⁺).

3-O-Benzyl-5-deoxy-1,2-O-isopropylidene-5-C-[(R)- and (S)-(ethoxy)ethylphosphiothioyl]- α -D-xylofuranose (24). To a solution of 23 (100 mg) in dry THF (10 cm³) was slowly added sodium borohydride (32 mg) at 5 °C. The mixture was degassed with argon, stirred at 50 °C for 13 h, and then refluxed for 9 h (under argon). It was diluted with ether and an aq saturated solution of Na₂SO₄ was added. separation, the aq layer was extracted with ether. The combined ethereal layer was dried (Na₂SO₄) and evaporated in vacuo. The residue was purified by column chromatography to give 24 as a yellow syrup (38 mg, 55% yield): $R_f = 0.6 (1:2 \text{ AcOEt-hexane}); ^1H \text{ NMR } \delta = 0.8 - 1.6 (6H, m,$ MeC-P, MeC-O-P), 1.34, 1.49 (6H, 2s, Me₂C), 1.6-2.4 (4H, m, 2H-5, CH₂-P), 3.5-4.9 (5H, m, H-2,3,4, CH₂-O-P), 4.58, 4.66 (2H, 2s, CH₂-O-3), 5.91 (1H, d, $J_{1,2}$ =3.8 Hz, H-1), and 7.34 (5H, s, $C_6H_5-C-O-3$); MS m/z 400 (M+).

1,2,4-Tri-*O*-acetyl-3-*O*-benzyl-5-deoxy-5-*C*-[(*R*)- and (*S*)-ethylphosphinothioyl]- α , β -D-xylopyranoses (25). The same procedures as those for 12 and 13 were employed. Thus, 24 (66 mg) was treated with SDMA (0.05 cm³) at 5 °C for 5.5 h, followed by the usual procedures, giving an inseparable mixture of 25 as a pale yellow syrup (1.2 mg, 1.6% yield from 24): R_t =0.5—0.7 (1:1 AcOEt-hexane); ¹H NMR δ =0.8—1.6 (3H, m, MeC-P), 1.6—2.1 (2H, m, MeCH₂-P), 1.88, 2.02, 2.09 (9H, 3brs, 3AcO), 2.2—3.5 (2H, m, 2H-5), 3.5—4.0 (1H, m, H-3), 4.60 (2H, brs, CH₂O-3), and 4.8—6.0 (3H, m, H-1,2,4); MS m/z 442 (M+; 28.1), 383 (9), 336 (14), 332 (19), 305

(13), 292 (14), 262 (30), 245 (59), 203 (29), 185 (100), 157 (20), 138 (23), 137 (21), 109 (38), 91 (81), and 43 (33).

Found: m/z 442.1243. Calcd for C₂₀H₂₇O₇PS: M, 442.1208.

We are grateful to Mr. Shin-ichi Takekuma (Kinki University) for the measurement of mass spectra.

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