Preliminary communication

Synthesis of luminescent probe—sugar conjugates of either protected or unprotected sugars

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Fluorescent carbohydrate conjugates have recently received considerable attention in a variety of biological fields ranging from cell-surface topography¹, membrane transport², polysaccharide mobility³, and gelling mechanism⁴, to the analysis of glycoproteins⁵. Concurrently, the need has arisen for a broader spectrum of labelling methods⁶, and, in this report, we focus attention on reactions whereby an amine-containing luminescent probe (L-NH₂) can be covalently attached to a partially protected or unprotected sugar, and also to polysaccharides.

Reaction of the amines^{7,8} 1 and 2 with ClCH₂COCl in ethyl acetate or ether under reflux (~2 h), followed by concentration by use of a stream of dry N₂ at room temperature, affords the pure chloroacetamide derivative[†], in high yield, directly from the reaction mixture: 3a, m.p. 133.5° (from toluene) and 3b, m.p. 131.0–131.5° (dec., from toluene). Such derivatives, which constitute a novel, and versatile, type of luminescent reagent can be used to alkylate thiol, amino, or hydroxyl functionalities under appropriate reaction-conditions; being less reactive than the equivalent iodoacetamides⁹, they offer the advantage of greater selectivity. For example, 3a reacts with the thio sugar tetraacetate 4 in CHCl₃ in the presence of solid NaHCO₃ during 18 h to yield 5 in 67% yield, m.p. 130° (from ethanol-H₂O), $[\alpha]_D^{22}$ -30° (c 1.6, CHCl₃); in contrast, no reaction between 3a and the D-galacto derivative 6 was observed under similar conditions. However, with NaH-HCONMe₂¹⁰, the ether 7 was produced in 39% yield, m.p. 128° (from isopropyl alcohol), $[\alpha]_D^{22}$ -12° (c 1.1, C₆H₆).

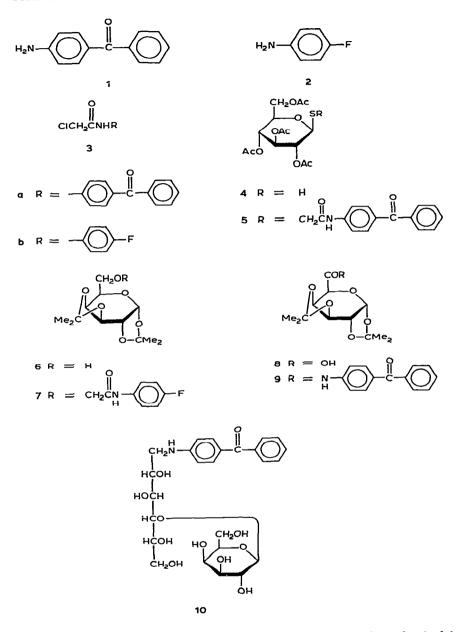
Luminescent amide derivatives may be obtained *via* carbodiimide-mediated coupling; for example, the D-galacturonic acid derivative 8 reacts with 1 and dicyclohexyl-carbodiimide¹¹ in CH₂Cl₂ solution during 35 h at room temperature to afford, after chromatography on a column of silica gel, compound 9 in 48% yield, m.p. 117° (dec., from isopropyl alcohol), $[\alpha]_D^{22} - 122.5^\circ$ (c 0.4, CHCl₃).

Complementary to the foregoing procedures are methods for selectively labelling, in polar media, unprotected sugars. Reductive amination¹² of oligosaccharides using sodium cyanoborohydride in methanol under reflux proceeds in high yields¹³. For

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[†]All derivatives reported here have n.m.r. spectra, elemental analyses, and/or high-resolution, mass spectra in accord with the structures assigned.

PRELIMINARY COMMUNICATION



example, 1 and lactose reacted during 23 h to afford the 1-deoxylactitol-1-yl derivative 10, in 75% yield after chromatography on a column of silica gel, m.p. 157–159° (from ethanol), $[\alpha]_D^{22}$ –15.8° (c 1.1, H₂O). This compound was further characterised as the nona-acetate, m.p. 61–63° (from EtOH–H₂O), $[\alpha]_D^{22}$ +25.9° (c 1.1, CHCl₃).

The same, reductive-amination procedure can be equally well applied to polysaccharides. Thus, sodium alginate that had been periodate-oxidized¹⁴ (26%) to the dialdehyde derivative reacted with *p*-fluoroaniline and sodium cyanoborohydride to yield, after extensive dialysis and lyophilization, the luminescent alginate derivative. Conversely, chitosan reacted¹⁵ with 9-anthraldehyde in 1:49 aqueous acetic acid—MeOH to form a stiff gel which was repeatedly washed (MeOH and ether) to afford the luminescent Schiffbase derivative. This derivative could be stabilized against hydrolysis by cyanoboro-hydride reduction, either simultaneous with its formation or following its isolation.

Insofar as the yields reported here are typical of those we have obtained for other similar reactions, we consider that thes.⁺ three methods will find widespread application in covalent labelling (luminescent, e.s.r., and n.m.r.) of many sugar-containing systems. In the case of the polysaccharide reactions where conventional, chemical characterization is not easy, we draw attention to the advantage of using as models the reactions of analogous, spin-labelled compounds, which can be accurately assayed by e.s.r. spectroscopy¹⁶.

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