

Carbene-Mediated Quaternarization of the Anomeric Position of Carbohydrates: Synthesis of Allylic Ketopyranosides, Access to the Missing #-Gluco and #-Manno Stereoisomers, and Preparation of Quaternary 2-Deoxy-2-Acetamido Sugars

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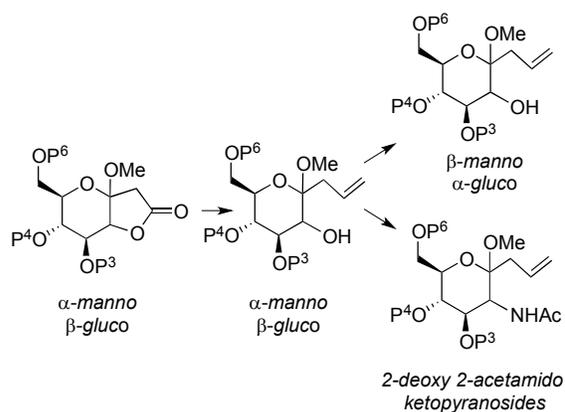
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3 **Carbene-Mediated Quaternarization of the Anomeric Position of**
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6 **Carbohydrates: Synthesis of Allylic Ketopyranosides, Access to**
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9 **the Missing α -Gluco and β -Manno Stereoisomers, and Preparation**
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12 **of Quaternary 2-Deoxy-2-Acetamido Sugars**
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Graphical Abstract:



Abstract:

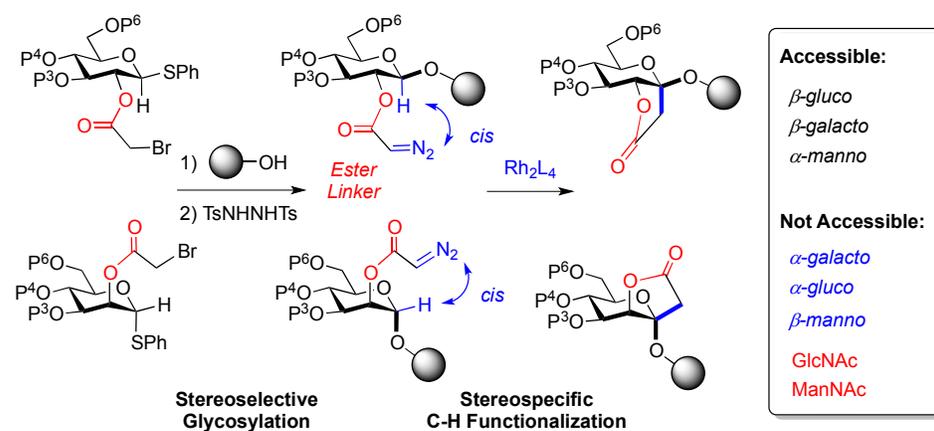
Following our work on the C-H functionalization of carbohydrates by 1,5 insertion of metal-carbenes, we report herein the robust and scalable conversion of sugar γ -lactones into highly valuable glycosides having a quaternary anomeric position substituted by an

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2
3 allyl chain ready for further functionalization. A divergent synthetic approach
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5 furthermore provided a straightforward access to ketopyranosides with a large chemo-
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7 and stereodiversity at position 2.
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10 11 12 **Introduction:**

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16 The field of glycobiology has made significant progress in recent years, in particular as a
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18 result of the emergence of new synthetic tools in carbohydrate chemistry. Efficient and
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20 innovative synthetic methods have contributed to a better understanding of the role
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22 played by complex oligosaccharides in numerous biological events, and paved the way
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24 to the new era of chemical glycobiology.¹ In this context, we recently reported a new
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26 approach toward ketopyranosides, in which quaternarization of the anomeric position
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28 was performed in a late stage of the synthetic process.² After glycosylation with
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30 anchimeric assistance of a 2-*O*-bromoacetyl, functionalization of the anomeric C-H bond
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32 with retention of configuration was promoted by 1,5 insertion of a Rh(II)-carbene.³
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34 Apart from allowing a more efficient access to ketopyranosides having an axial aglycone,
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36 this strategy also opened the way to the unprecedented β -anomers (Scheme 1).
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38 Although critical to ensure a clean C-H functionalization process, anchoring the metal-
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40 carbene at position 2 also had some limitations. First, only ketopyranosides in the α -
41
42 *manno*, β -*gluco* and β -*galacto* series could be obtained because insertion into the
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44 anomeric C-H bond took place after formation of a 1,2-*trans* glycosidic linkage.
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46 Moreover, 2-deoxy-2-amino sugars with a quaternary anomeric position, which are
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48 highly desirable since GlcNAc, GalNAc and ManNAc are major components of
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50 glycoconjugates,^{4,5} could not be prepared by direct functionalization of the anomeric C-H
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52 bond of 2-deoxy 2-amino sugars. In fact, only complex mixtures were obtained, if an
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amide tether between the metal-carbene and the sugar moiety replaced the ester linkage.⁶



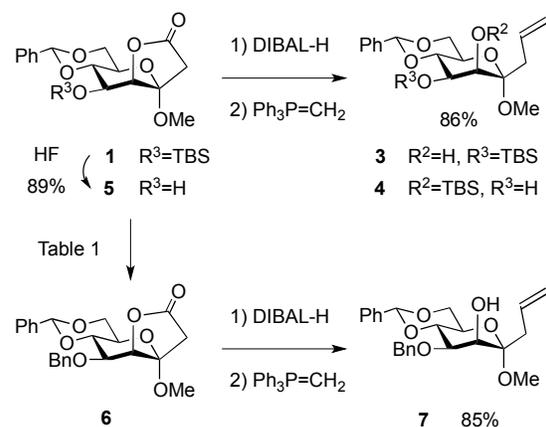
Scheme 1. Carbene-Mediated Quaternarization of the Anomeric Position: Scope and Limitations

The design of new chemical tools for glycobiology with a ketopyranosidic central core would require a straightforward and robust access toward compounds covering a larger chemical space than the α -manno, β -gluco and β -galacto series. In this context, we would like to report herein transformation of α -manno and β -gluco γ -lactones into ketopyranosides bearing an allyl pending chain, and the efficient chemical and stereochemical diversification at position 2 following a diversity-oriented approach.

Results and Discussion:

Sugar lactones **1** (Scheme 2) and **2** (Scheme 3) with a *tert*-butyldimethylsilyl ether protecting group at position 3 were routinely prepared on gram scale by decomposition of diazo sugars under $\text{Rh}_2(\text{OAc})_4$ catalysis in refluxing 1,2 dichloroethane.⁷ Since partial

reduction of **1** with DIBAL-H in DCM at -78 °C and Wittig olefination gave a mixture of *C,O*-glycosides **3** and **4** by migration of the *tert*-butyldimethylsilyl group (Scheme 2), switching toward a robust benzyl ether was thus considered. Although deprotection of lactone **1** with TBAF resulted in massive degradation, HF/Pyridine gave alcohol **5** in 89% yield after purification by silica gel flash chromatography.

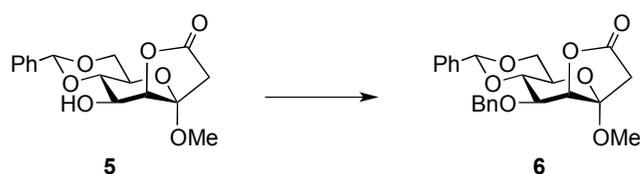


Scheme 2. Preparation of ketopyranoside **7** in the α -manno series.

Not surprisingly, benzylation under Williamson's conditions induced degradation of lactone **5**.⁸ Benzyl 2,2,2-trichloroacetimidate (BnTCA)⁹ gave **6** in moderate yield and low purity because Lewis or Brønsted acid activation induced competitive cleavage of the 4,6-*O*-benzylidene function (Table 1, entries 1-3). Benzylation of **5** was next attempted with 2-benzyloxy-1-methylpyridinium triflate (BnPYR) and MgO.¹⁰ Even if conversion remained partial, pure lactone **6** was obtained in 66% yield after prolonged heating in trifluorotoluene (entries 4 and 5). Under microwave irradiation,¹¹ the reaction time was reduced to 9 h, and the yield improved to 74%, without affecting the purity of **6** (entry 6). Finally, clean and full conversion of **5** was achieved with benzyl bromide (BnBr) and silver(I) oxide (Ag₂O) in refluxing dichloromethane to give pure 3-*O*-benzyl lactone **6** in

88% isolated yield after silica gel flash chromatography (entries 7 and 8).¹² Partial reduction of **6** with DIBAL-H at -78 °C in DCM and Wittig olefination gave α -ketopyranoside **7** having a complete set of orthogonal protecting groups in 85% yield over two steps (Scheme 2).

Table 1. Benzylation at position 3 in the α -manno series.

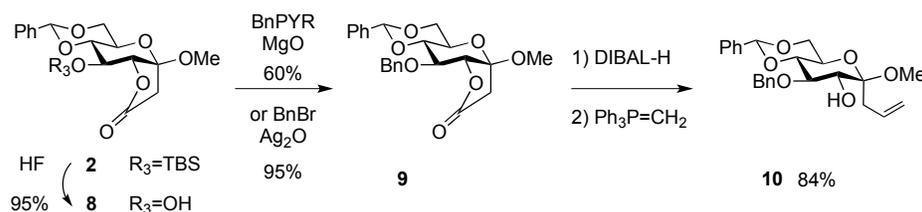


Entry	Reagent (equiv.)	Additive (equiv.)	Temp. (°C)	Time (h.)	Conv. (%) ^a	Yield (%) ^b
1 ^c	BnTCA (3)	TfOH (0.2)	0	1	70	57
2 ^c	BnTCA (3)	TMSOTf (0.2)	0	1.5	100	31
3 ^c	BnTCA (3)	BF ₃ .OEt ₂ (0.2)	0	1.5	50	12
4 ^d	BnPYR (2)	MgO (2)	85	48	74	66
5 ^d	BnPYR (2)	MgO (2)	100	48	80	64
6 ^{d,e}	BnPYR (2)	MgO (2)	80	10	80	74
7 ^c	BnBr (1.2)	Ag ₂ O (1.2)	R.T.	4	0	-
8 ^c	BnBr (3)	Ag ₂ O (2.4)	40	4	100	88

^a: evaluated by ¹H NMR analysis of the crude; ^b: isolated after flash chromatography

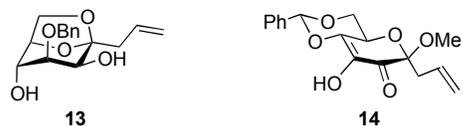
^c: CH₂Cl₂ as solvent; ^d: trifluorotoluene as solvent; ^e: microwave irradiation.

The same sequence was next performed in the β -*gluco* series. Desilylation of **2** with HF/pyridine followed by benzylation either with BnPYR/MgO or BnBr/Ag₂O gave lactone **9** with a 3-*O*-benzyl ether in 57 and 90% yield respectively over two steps. After partial reduction with DIBAL-H and olefination, β -ketopyranoside **10** was finally obtained in 84% yield (Scheme 3).

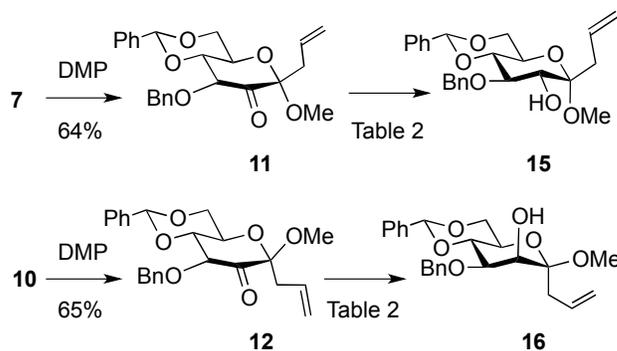


Scheme 3. Preparation of ketopyranoside **10** in the β -*gluco* series.

Having developed a robust and scalable access to α -*manno* and β -*gluco* *C,O*-glycosides **7** and **10**, ketopyranosides **15** and **16** in the α -*gluco* and β -*manno* series, as well as 2-deoxy 2-acetamido sugars with a quaternary anomeric position **17-20**, were next prepared following a diversity-oriented approach (scheme 4). To our surprise, oxidation of **7** under Swern conditions gave a complex mixture from which 1,6-anhydro derivative **13**, resulting from cleavage of the glycosidic bond, was identified. Methyl 2-ulosides **11** and **12** were finally prepared in 64 and 65% yield respectively by oxidation of **7** and **10** with Dess-Martin Periodinane (DMP). It is worth of note that pyridine should be used as a cosolvent,¹³ and the amount of DMP carefully controlled, in order to prevent formation of **14**, presumably obtained by radical hydrogen abstraction at position 3 followed by β -fragmentation of the benzyl ether.



Reduction of **11** and **12** into the missing α -*gluco* and β -*manno* ketopyranosides **15** and **16** was next investigated with NaBH₄, DIBAL-H and L-Selectride.¹⁴



Scheme 4. Preparation and reduction of methyl 2-ulosides **11** and **12**.

From methyl α -uloside **11**, ketopyranoside **15** in the α -*gluco* series was obtained as the major diastereoisomer with NaBH₄ or DIBAL-H, (table 2, entries 1 and 2), whereas L-Selectride mainly gave back the quaternary mannoside **7**. Similarly to previous reports from the literature,¹⁵ reduction of **11** by NaBH₄ mainly occurred by axial attack of the hydride, thus showing that the quaternary anomeric centre had little influence on the stereochemical outcome of the reaction. However, reduction of uloside **12** with NaBH₄ did not yield β -mannoside **16** as expected,¹⁶ but rather a 1:1 mixture of epimers. The quaternary anomeric centre, in addition to the cyclic 4,6-*O*-benzylidene,¹⁷ thus had a dramatic impact on the stereochemical outcome of the reduction in the β series. The desired methyl β -ketopyranoside **16** in the *manno* series was finally obtained with complete stereocontrol by reduction with L-selectride.¹⁸

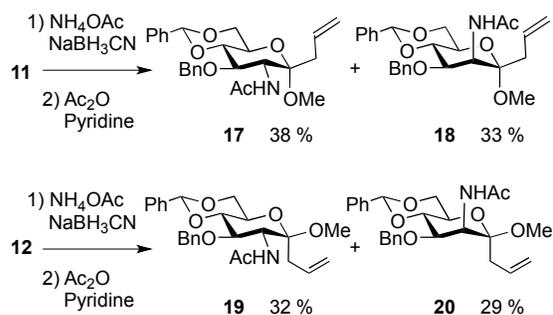
Table 2. Reduction of ulosides **11** and **12**.

Entry	2-uloside	Reagent	Alcohols (d.r.) ^a	Yield (%) ^b
1	11	NaBH ₄	7/15 (3:7)	91
2	11	DIBAL-H	7/15 (1:3)	51
3	11	L-Selectride	7/15 (4:1)	56
4	12	NaBH ₄	10/16 (1:1)	60
5	12	DIBAL-H	10/16 (7:1)	51
6	12	L-Selectride	10/16 (0:1)	61

^a: evaluated by ¹H NMR analysis of the crude; ^b: isolated after flash chromatography.

With suitable conditions for accessing the missing α -gluco and β -manno series in hands, we next turned our attention toward the preparation of quaternary 2-deoxy 2-acetamido sugars from methyl 2-ulosides **11** and **12** (scheme 5). Since classical approaches relying on substitution of 2-*O*-sulfonates with azides,¹⁹ or on formation of oximes followed by reductive cleavage of the N-O bond,²⁰ all failed despite extensive

efforts, direct reductive amination of **11** and **12** with NH_4OAc and NaBH_3CN was thus considered.²¹ Under microwave irradiation,²² amination was performed at 80 °C without degradation of the starting materials, but competitive reduction of **11** and **12** into alcohols **7/15** and **10/16** gave a complex mixture. By delaying introduction of the reducing agent after 10 min heating with NH_4OAc alone, the desired 2-deoxy 2-acetamido sugars **17-20** were finally obtained in good yield after acetylation. Configuration of **17-20** was determined on the basis of vicinal coupling constants between H_2 and H_3 ($J = 4.5$ Hz for **18** in the *manno* series, and $J = 9.2$ Hz for **19** with a *gluco* configuration).



Scheme 5. Preparation of 2-deoxy 2-acetamido ketopyranosides **17-20**.

Overall, we reported herein the preparation of eight non-natural sugars derivatives with a quaternary anomeric position substituted by an allyl pending chain. From lactones **1** and **2**, a robust and scalable synthetic sequence first gave α -*manno* and β -*gluco* ketopyranosides **7** and **10** in 67 and 76% yield respectively over 4 steps. After oxidation into key ulosides **11** and **12**, a diversity-oriented approach provided a straightforward access to ketopyranosides **15** and **16** in the missing α -*gluco* and β -*manno* series, as well as to the four stereoisomers **17-20** of unprecedented quaternary 2-deoxy 2-acetamido sugars. Having in hands a large diversity of non-natural carbohydrate derivatives having a quaternary anomeric position substituted by an allyl chain ready for further

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3 functionalization, new chemical tools for glycobiology are currently under preparation
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5 in our laboratory.
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8 9 10 **Experimental Section**

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14 **General Information and Method.** For reactions, solvents were purchased anhydrous
15 (dichloromethane, and pyridine) or distilled (tetrahydrofuran over
16 sodium/benzophenone). All reactions were conducted under an argon atmosphere. All
17 reagents were used as received unless otherwise indicated. Reactions were monitored
18 by thin-layer chromatography with silica gel 60 F254 pre-coated aluminium plate (0.25
19 mm). Visualization was performed under UV light and phosphomolybdic acid oxidation.
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21 ^1H NMR spectra were recorded at 300 MHz, and ^{13}C NMR spectra at 75 MHz.
22
23 Abbreviations used for peak multiplicities are s: singlet, d: doublet, t: triplet, q:
24 quadruplet and m: multiplet. Coupling constants J are in Hz and chemical shifts are given
25 in ppm and calibrated with CDCl_3 (residual solvent signals). Carbon multiplicities were
26 assigned by distortionless enhancement by polarization transfer (DEPT) experiments.
27
28 The ^1H and ^{13}C signals were assigned by COSY and HSQC experiments. Accurate mass
29 measurements (HRMS) were performed with a Q-TOF analyser. Infrared spectra (IR)
30 were recorded by application on a Single Reflection Attenuated Total Reflectance (ATR)
31 Accessories, and data are reported in cm^{-1} . Optical rotations were determined with a
32 water-jacketed 10 cm cell. Specific rotations are reported in $10^{-1} \text{ deg cm}^2 \text{ g}^{-1}$ and
33 concentrations in g per 100 mL. Melting points are uncorrected. Microwave experiments
34 were conducted in a Biotage Initiator equipped with a monomode cavity with a
35 microwave power delivery system ranging from 0 to 850 W allowing pressurized
36 reactions (0 to 30 bars) to be carried out in sealed glass vials (0.5 to 20 mL) equipped
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with a snap cap and a silicon septum. The temperature (0 to 300 °C) was monitored via a contact-less infrared sensor and was calibrated with a Ruby Thermometer.

(2*R*,4*aR*,5*aS*,8*aS*,9*S*,9*aS*)-9-hydroxy-5*a*-methoxy-2-

phenylhexahydrofuro[2',3':5,6]pyrano[3,2-*d*][1,3]dioxin-7(8*aH*)-one 5. To a solution of lactone **1** (1.456 g, 3.334 mmol) in THF (50 mL), was added a solution of HF.pyridine (dry THF/dry pyridine/HF.pyridine, 2:2:1, 65.5 mL, 333 mmol) at room temperature. After 18 h, ethyl acetate (150 mL) and a saturated solution of NaHCO₃ (600 mL) were carefully added. After 30 min, the organic layer was separated and the aqueous layer was extracted with ethyl acetate (3 x 150 mL). The combined organic layers were dried (MgSO₄), filtrated, and concentrated under reduce pressure. Purification by silica gel chromatography (cyclohexane/ethyl acetate 1.5:1) gave **5** as a white foam (946 mg, 89%):

$R_f = 0.22$ (Silica, cyclohexane/ethyl acetate 3:1); ¹H NMR (300 MHz, CDCl₃): δ (ppm) 7.57-7.26 (m, 5H, H_{Ar}), 5.56 (s, 1H, H7), 4.61 (d, $J = 4.2$ Hz, 1H, H2), 4.28-4.24 (m, 1H, H6), 4.09 (dd, $J = 9.3, 4.2$ Hz, 1H, H3), 3.86-3.70 (m, 3H, H4, H5, H6), 3.37 (s, 3H, OMe), 2.82 (d, $J = 16.3$ Hz, 1H, CH₂-C=O), 2.61 (d, $J = 16.3$ Hz, 1H, CH₂-C=O), 2.50 (br s, 1H, OH); ¹³C NMR (75 MHz, CDCl₃): δ (ppm) 171.4 (C=O), 136.9 (C_{qAr}), 129.4 (C_{Ar}), 128.4 (C_{Ar}), 126.3 (C_{Ar}), 103.5 (C1), 102.1 (C7), 81.0 (C2), 77.6 (C4), 68.1 (C6), 67.7 (C3), 64.3 (C5), 51.9 (OMe), 40.4 (CH₂-C=O); FT-IR (film): 3467, 2922, 2859, 1778, 1455, 1393, 1300, 1254, 1226, 1165, 1191, 1088, 1025, 999, 954, 917, 774, 702, 670, 568, 535 cm⁻¹; HRMS (ESI): m/z Calcd for C₁₆H₂₂NO₇ [M + NH₄]⁺ 340.1396, found 340.1395; [α]_D²⁰ = - 58 (CHCl₃, $c = 1.85$).

(2*R*,4*aR*,5*aR*,8*aR*,9*S*,9*aS*)-9-hydroxy-5*a*-methoxy-2-

phenylhexahydrofuro[2',3':5,6]pyrano[3,2-*d*][1,3]dioxin-7(8*aH*)-one 8. To a

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3 solution of lactone **2** (800 mg, 1.83 mmol) in THF (30 mL), was added a solution of
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5 HF.pyridine (dry THF/dry pyridine/HF.pyridine, 2:2:1, 21 mL, 183 mmol) at room
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7 temperature. After 18 h, ethyl acetate (100 mL) and a saturated solution of NaHCO₃
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9 (400 mL) were carefully added. After 30 min, the organic layer was separated and the
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11 aqueous layer was extracted with ethyl acetate (3 x 150 mL). The combined organic
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13 layers were dried (MgSO₄), filtrated, and concentrated under reduce pressure.
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15 Purification by silica gel chromatography (cyclohexane / ethyl acetate 1.5:1) gave **8** as a
16
17 white foam (560 mg, 95%). R_f = 0.38 (Silica, dichloromethane/ethyl acetate 15:1); ¹H
18
19 NMR (300 MHz, CDCl₃): δ (ppm) 7.50-7.38 (m, 5H, H_{Ar}), 5.57 (s, 1H, H7), 4.43 (d, J = 6.4
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21 Hz, 1H, H2), 4.35 (dd, J = 10.0, 4.9 Hz, 1H, H6), 3.91 (dd, J = 10.2, 6.4 Hz, 1H, H3), 3.78-
22
23 3.73 (m, 2H, H6, H4), 3.62 (ddd, J = 10.0, 9.1, 4.9 Hz, 1H, H5), 3.38 (s, 3H, OMe), 2.93 (br
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25 s, 1H, OH), 2.82 (d, J = 17.1 Hz, 1H, CH₂-C=O), 2.67 (d, J = 17.1 Hz, 1H, CH₂-C=O); ¹³C NMR
26
27 (75 MHz, CDCl₃): δ (ppm) 172.0 (C=O), 136.7 (C_{qAr}), 129.6 (C_{Ar}), 128.6 (C_{Ar}), 126.4 (C_{Ar}),
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29 103.8 (C1), 102.0 (C7), 83.4 (C2), 78.1 (C4), 72.8 (C3), 68.7 (C6), 66.0 (C5), 50.4 (OMe),
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31 35.9 (CH₂-C=O); FT-IR (film): 3454, 2922, 2857, 1788, 1456, 1380, 1212, 1167, 1079,
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33 1007, 918, 835, 754, 699, 542 cm⁻¹; HRMS (ESI): m/z Calcd for C₁₆H₂₂NO₇ [M + NH₄]⁺
34
35 340.1396, found 340.1396; [α]_D²⁰ = + 3 (CHCl₃, c = 1.0).

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42 **(2*R*,4*aR*,5*aS*,8*aS*,9*S*,9*aR*)-9-(benzyloxy)-5*a*-methoxy-2-**

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44 **phenylhexahydrofuro[2',3':5,6]pyrano[3,2-*d*][1,3]dioxin-7(8*aH*)-one **6**. Procedure**

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46 **A:** A suspension of alcohol **5** (100 mg, 0.310 mmol), MgO (25 mg, 0.620 mmol) and 2-
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48 benzyloxy-methylpyridinium triflate (217 mg, 0.620 mmol) in trifluorotoluene (0.6 mL)
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50 under an argon atmosphere was heated under microwave irradiation at 85 °C. After 10
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52 h, dichloromethane (3 mL) and a saturated solution of NaHCO₃ (3 mL) were added, and
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54 the mixture was stirred until it became homogenous. After separation, the aqueous layer
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56 was extracted with dichloromethane, the combined organic phases were dried (MgSO₄),
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3 filtered, and concentrated under reduce pressure. After purification by silica gel
4 chromatography (cyclohexane/ethyl acetate 3:1), **6** was obtained as a colourless oil (85
5 mg, 66%). Procedure B: A suspension of **5** (300 mg, 0.93 mmol), Ag₂O (646.5 mg, 2.79
6 mmol) and benzyl bromide (265 μ L, 2.23 mmol) in dichloromethane (2.1 mL) under an
7 argon atmosphere was heated at 40 °C in a sealed tube (dark). After 4h30, the reaction
8 mixture was diluted with dichloromethane (20 mL), filtered through a Celite pad, and
9 added with a saturated solution of NaHCO₃ (20 mL). After separation of the organic
10 layer, the aqueous layer was extracted with dichloromethane (2 x 20 mL). The combined
11 organic layers were dried (MgSO₄), filtered and concentrated under reduce pressure.
12 After purification by silica gel chromatography (cyclohexane/ethyl acetate 2:1) **6** was
13 obtained as a colourless oil (340 mg, 88%).
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30 R_f = 0.28 (Silica, cyclohexane/ethyl acetate 3:1); ¹H NMR (300 MHz, CDCl₃): δ (ppm)
31 7.53-7.25 (m, 10H, H_{Ar}), 5.61 (s, 1H, H7), 4.88 (d, *J* = 12.3 Hz, 1H, O-CH₂-Ph), 4.78 (d, *J* =
32 12.3 Hz, 1H, O-CH₂-Ph), 4.58 (d, *J* = 3.6 Hz, 1H, H2), 4.28 (dd, *J* = 9.8, 3.6 Hz, 1H, H6), 4.06
33 (t, *J* = 9.5 Hz, 1H, H4), 3.99 (dd, *J* = 9.5, 3.6 Hz, 1H, H3), 3.77 (t, *J* = 9.8 Hz, 1H, H6), 3.76-
34 3.74 (m, 1H, H5), 3.30 (s, 3H, OMe), 2.80 (d, *J* = 16.5 Hz, 1H, CH₂-C=O), 2.66 (d, *J* = 16.5
35 Hz, 1H, CH₂-C=O); ¹³C NMR (75 MHz, CDCl₃): δ (ppm) 171.6 (C=O), 137.8 (C_{qAr}), 137.2
36 (C_{qAr}), 129.2 (C_{Ar}), 128.5 (C_{Ar}), 128.4 (C_{Ar}), 127.9 (C_{Ar}), 126.1 (C_{Ar}), 103.7 (C1), 101.6
37 (C7), 80.1 (C2), 77.6 (C4), 73.2 (C3), 73.1 (O-CH₂-Ph), 68.3 (C6), 64.6 (C5), 51.9 (OMe),
38 40.3 (CH₂-C=O); FT-IR (film): 2933, 2871, 1797, 1493, 1454, 1377, 1302, 1282, 1217,
39 1177, 1155, 1096, 1077, 1004, 962, 750, 698 cm⁻¹; HRMS (ESI): *m/z* Calcd for C₂₃H₂₅O₇
40 [M + H]⁺ 413.1600, found 413.1613; [α]_D²⁰ = - 40 (CHCl₃, *c* = 1.0).
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53 **(2R,4aR,5aR,8aR,9S,9aR)-9-(benzyloxy)-5a-methoxy-2-**
54 **phenylhexahydrofuro[2',3':5,6]pyrano[3,2-*d*][1,3]dioxin-7(8aH)-one 9.** Procedure
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2
3 A: A suspension of alcohol **8** (100 mg, 0.310 mmol), MgO (25 mg, 0.620 mmol) and 2-
4 benzyloxy-methylpyridinium triflate (217 mg, 0.620 mmol) in trifluorotoluene (0.6 mL)
5
6 under an argon atmosphere was heated under microwave irradiation at 85 °C. After 10
7
8 h, dichloromethane (3 mL) and a saturated solution of NaHCO₃ (3 mL) were added, and
9
10 the mixture was stirred until it became homogenous. After separation, the aqueous layer
11
12 was extracted with dichloromethane, the combined organic phases were dried (MgSO₄),
13
14 filtered, and concentrated under reduce pressure. After purification by silica gel
15
16 chromatography (cyclohexane/ethyl acetate 3:1), **9** was obtained as a colourless oil (77
17
18 mg, 60%). Procedure B: A suspension of **8** (300 mg, 0.93 mmol), Ag₂O (646.5 mg, 2.79
19
20 mmol) and benzyl bromide (265 μL, 2.23 mmol) in dichloromethane (2.1 mL) under an
21
22 argon atmosphere was heated at 40 °C in a sealed tube (dark). After 3h30, the reaction
23
24 mixture was diluted with dichloromethane (20 mL), filtered through a Celite pad, and
25
26 added with a saturated solution of NaHCO₃ (20 mL). After separation, the aqueous layer
27
28 was extracted with dichloromethane (2 x 20 mL), and the combined organic layers were
29
30 dried (MgSO₄), filtered and concentrated under reduce pressure. After purification by
31
32 silica gel chromatography (cyclohexane/ethyl acetate 2:1) **9** was obtained as a
33
34 colourless oil (365 mg, 95%). R_f = 0.34 (Silica, cyclohexane/ethyl acetate 3:1); ¹H NMR
35
36 (300 MHz, CDCl₃): δ (ppm) 7.52-7.31 (m, 10H, H_{Ar}), 5.63 (s, 1H, H7), 4.85 (s, 2H, O-CH₂-
37
38 Ph), 4.54 (d, J = 5.6 Hz, 1H, H2), 4.34 (dd, J = 10.3, 4.8 Hz, 1H, H6), 3.99 (t, J = 9.7 Hz, 1H,
39
40 H4), 3.87-3.74 (m, 2H, H3, H6), 3.67 (dt, J = 9.7, 4.8 Hz, 1H, H5), 3.40 (s, 3H, OMe), 2.86
41
42 (d, J = 17.2 Hz, 1H, CH₂-C=O), 2.69 (d, J = 17.2 Hz, 1H, CH₂-C=O); ¹³C NMR (75 MHz,
43
44 CDCl₃): δ (ppm) 171.6 (C=O), 137.4 (C_{qAr}), 136.9 (C_{qAr}), 129.2 (C_{Ar}), 128.5 (C_{Ar}), 128.4
45
46 (C_{Ar}), 128.0 (C_{Ar}), 126.1 (C_{Ar}), 103.9 (C1), 101.6 (C7), 83.9 (C2), 78.8 (C3), 78.2 (C4), 73.7
47
48 (O-CH₂-Ph), 68.9 (C6), 66.4 (C5), 50.7 (OMe), 36.8 (CH₂-C=O); FT-IR (film): 2933, 1793,
49
50 1605, 1497, 1454, 1372, 1331, 1268, 1245, 1213, 1170, 1091, 1028, 916, 890, 752, 698,
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601 cm^{-1} ; HRMS (ESI): m/z Calcd for $\text{C}_{23}\text{H}_{24}\text{O}_7\text{Na}$ $[\text{M} + \text{Na}]^+$ 435.1420, found 435.1415; $[\alpha]_{\text{D}}^{20} = +5$ (CHCl_3 , $c = 1.0$).

(2R,4aR,6S,7S,8R,8aR)-6-allyl-8-(benzyloxy)-6-methoxy-2-

phenylhexahydropyrano[3,2-d][1,3]dioxin-7-ol 7. To a solution of **6** (520 mg, 1.26 mmol) in CH_2Cl_2 (10 mL) at -78 °C was added drop-wise a 1 M solution of DIBAL-H in hexanes (1.388 mL, 1.388 mmol). After 15 min at -78 °C, the mixture was quenched with methanol, and added with CH_2Cl_2 (20 mL) and Rochelle salts (1 M, 20 mL). After 12 h of vigorous stirring at room temperature, the organic layer was separated, and the aqueous layer was extracted with dichloromethane (3 x 15 mL). The combined organic layers were dried (MgSO_4), filtered and concentrated under reduce pressure. Filtration over a silica plug delivered the lactol as a mixture of diastereoisomers. A suspension of methyltriphenylphosphonium bromide (678 mg, 1.89 mmol) in THF (8 mL) at 0 °C was added with potassium *tert*-butoxide (213 mg, 1.89 mmol) and a solution of lactol in THF (2 mL). After 15 min stirring at room temperature, CH_2Cl_2 (25 mL) and water (15 mL) were added. After separation, the aqueous layer was extracted with dichloromethane (3 x 25 mL), and the combined organic layers were dried (MgSO_4), filtered and concentrated under reduce pressure. Purification by silica gel chromatography (dichloromethane/ethyl acetate from 100:0 to 90:10) gave **7** as a colorless oil (442 mg, 85% over 2 steps). $R_f = 0.53$ (silica, cyclohexane / ethyl acetate 2.5:1); ^1H NMR (300 MHz, CDCl_3): δ (ppm) 7.54-7.29 (m, 10H, H_{Ar}), 5.77 (dddd, $J = 17.1, 10.3, 7.3, 6.7$ Hz, 1H, $\text{CH}_2\text{-}\underline{\text{CH}}=\text{CH}_2$), 5.63 (s, 1H, H7), 5.28-5.18 (m, 2H, $\text{CH}_2\text{-CH}=\underline{\text{CH}}_2$), 4.86 (d, $J = 11.9$ Hz, 1H, O- $\underline{\text{CH}}_2$ -Ph), 4.74 (d, $J = 11.9$ Hz, 1H, O- $\underline{\text{CH}}_2$ -Ph), 4.30 (dd, $J = 10.4, 4.8$ Hz, 1H, H6), 4.11 (t, $J = 9.1$ Hz, 1H, H4), 4.04-3.99 (m, 2H, H2, H3), 3.88 (t, $J = 10.4$ Hz, 1H, H6), 3.70 (ddd, $J = 10.4, 9.1, 4.8$ Hz, 1H, H5), 3.27 (s, 3H, OMe), 2.73 (m, 2H, $\underline{\text{CH}}_2\text{-CH}=\text{CH}_2$, OH), 2.54 (dd, $J = 14.7, 6.7$ Hz, 1H, $\underline{\text{CH}}_2\text{-CH}=\text{CH}_2$); ^{13}C NMR (75 MHz, CDCl_3): δ (ppm) 138.1 (C_{qAr}), 137.6

1
2
3 (C_{qAr}), 131.6 (CH₂-CH=CH₂), 128.9 (C_{Ar}), 128.5 (C_{Ar}), 128.2 (C_{Ar}), 127.85 (C_{Ar}), 127.83
4
5 (C_{Ar}), 126.1 (C_{Ar}), 119.1 (CH₂-CH=CH₂), 102.6 (C1), 101.5 (C7), 78.6 (C4), 76.1 (C3), 72.8
6
7 (O-CH₂-Ph), 69.5 (C2), 68.9 (C6), 64.1 (C5), 47.4 (OMe), 35.2 (CH₂-CH=CH₂); FT-IR (film):
8
9 3482, 3033, 2912, 1643, 1497, 1454, 1374, 1211, 1094, 1059, 1026, 1004, 915, 848, 817,
10
11 745, 695, 649 cm⁻¹; HRMS (ESI): m/z Calcd for C₂₄H₂₉O₆ [M + H]⁺ 413.1964, found
12
13 413.1963; [α]_D²⁰ = - 26 (CHCl₃, c = 1.0).

14
15
16 **(2R,4aR,6R,7R,8R,8aR)-6-allyl-8-(benzyloxy)-6-methoxy-2-**

17 **phenylhexahydropyrano[3,2-d][1,3]dioxin-7-ol 10.** To a solution of **9** (450 mg, 1.08
18
19 mmol) in CH₂Cl₂ (3 mL) at -78 °C was added drop wise a 1 M solution of DIBAL-H in
20
21 hexanes (1.42 mL, 1.42 mmol). After 15 min at -78 °C, the mixture was quenched with
22
23 methanol, and added with CH₂Cl₂ (5 mL) and Rochelle salts (1 M, 5 mL). After 12 h of
24
25 vigorous stirring at room temperature, the organic layer was separated, and the aqueous
26
27 layer was extracted with dichloromethane (3 x 15 mL). The combined organic layers
28
29 were dried (MgSO₄), filtered and concentrated under reduce pressure. Filtration over a
30
31 silica plug delivered the lactol as a mixture of diastereoisomers. A suspension of
32
33 methyltriphenylphosphonium bromide (1.02 g, 2.86 mmol) in THF (2.5 mL) at 0 °C was
34
35 added with potassium *tert*-butoxide (321 mg, 2.86 mmol) and a solution of lactol in THF
36
37 (2.5 mL). After 15 min stirring at room temperature, CH₂Cl₂ (10 mL) and water (10 mL)
38
39 were added. After separation, the aqueous layer was extracted with dichloromethane (3
40
41 x 15 mL), and the combined organic layers were dried (MgSO₄), filtered and
42
43 concentrated under reduce pressure. Purification by silica gel chromatography
44
45 (cyclohexane/ethyl acetate from 3:1) gave **10** as a colorless oil (374 mg, 84% over 2
46
47 steps). R_f = 0.28 (silica, cyclohexane/ethyl acetate 3:1); ¹H NMR (300 MHz, CDCl₃): δ
48
49 (ppm) 7.51-7.29 (m, 10H, H_{Ar}), 5.91 (dddd, J = 18.0, 11.8, 7.8, 6.3 Hz, 1H, CH₂-CH=CH₂),
50
51 5.59 (s, 1H, H7), 5.19-5.14 (m, 2H, CH₂-CH=CH₂), 5.02 (d, J = 11.4 Hz, 1H, O-CH₂-Ph), 4.76
52
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2
3 (d, $J = 11.4$ Hz, 1H, O-CH₂-Ph), 4.31 (dd, $J = 10.4, 4.5$ Hz, 1H, H6), 3.93-3.90 (m, 1H, H3),
4
5 3.80-3.72 (m, 3H, H2, H4, H6), 3.60 (td, $J = 9.4, 4.5$ Hz, 1H, H5), 3.37 (s, 3H, OMe), 2.72
6
7 (dd, $J = 15.2, 6.3$ Hz, 1H, CH₂-CH=CH₂), 2.59 (dd, $J = 15.2, 7.8$, Hz, 1H, CH₂-CH=CH₂), 2.37
8
9 (d, 1H, OH, $J = 2.4$ Hz); ¹³C NMR (75 MHz, CDCl₃): δ (ppm) 138.2 (C_{qAr}), 137.3 (C_{qAr}),
10
11 131.6 (CH₂-CH=CH₂), 129.0 (C_{Ar}), 128.5 (C_{Ar}), 128.3 (C_{Ar}), 128.1 (C_{Ar}), 127.9 (C_{Ar}), 125.9
12
13 (C_{Ar}), 118.2 (CH₂-CH=CH₂), 102.4 (C1), 101.2 (C7), 82.2 (C2), 80.1 (C4), 74.6 (O-CH₂-Ph),
14
15 71.6 (C3), 69.2 (C6), 64.7 (C5), 48.9 (OMe), 35.3 (CH₂-CH=CH₂); FT-IR (film): 3479, 3362,
16
17 2922, 2854, 1454, 1375, 1212, 1177, 1076, 1059, 1026, 1006, 962, 915, 865, 762, 745,
18
19 697, 660, 488 cm⁻¹; HRMS (ESI): m/z Calcd for C₂₄H₃₂NO₆ [M + NH₄]⁺ 430.2230, found
20
21 430.2228; $[\alpha]_D^{20} = +10$ (CHCl₃, $c = 1.0$).

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26 **(2R,4aR,6S,8R,8aR)-6-allyl-8-(benzyloxy)-6-methoxy-2-**

27
28 **phenyltetrahydropyrano[3,2-*d*][1,3]dioxin-7(6H)-one 11.** A solution of Dess-Martin
29
30 periodinane (144 mg, 0.338 mmol) in pyridine (1 mL) was stirred at room temperature
31
32 for 30 min, and added with a solution of alcohol **7** (140 mg, 0.338 mmol) in
33
34 dichloromethane (1 mL). After 10 h at room temperature, additional Dess-Martin
35
36 periodinane (144 mg, 0.338 mmol) was added to the reaction mixture. After 10 h, TLC
37
38 revealed complete consumption of the starting material, and dichloromethane (8 mL), a
39
40 saturated solution of NaHCO₃ (4 mL) and a saturated solution of Na₂S₂O₃ (4 mL) were
41
42 added. After 4 h, the organic layer was separated and the aqueous layer was extracted
43
44 with dichloromethane (3 x 10 mL). The combined organic layers were dried (MgSO₄),
45
46 filtered and concentrated under reduce pressure. After purification by silica gel
47
48 chromatography (cyclohexane/ethyl acetate 5:1), ketone **11** was obtained as a colorless
49
50 film (90 mg, 64%). $R_f = 0.85$ (silica, cyclohexane / ethyl acetate 3:1); ¹H NMR (300 MHz,
51
52 CDCl₃): δ (ppm) 7.53-7.27 (m, 10H, H_{Ar}), 5.81 (dddd, $J = 17.6, 10.4, 7.4, 6.2$ Hz, 1H, CH₂-
53
54 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
55
56 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
57
58 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
59
60 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
61 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
62 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
63 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
64 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
65 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
66 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
67 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
68 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
69 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
70 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
71 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
72 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
73 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
74 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
75 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
76 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
77 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
78 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
79 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
80 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
81 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
82 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
83 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
84 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
85 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
86 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
87 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
88 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
89 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
90 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
91 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
92 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
93 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
94 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
95 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
96 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
97 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
98 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
99 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
100 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
101 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
102 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
103 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
104 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
105 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
106 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
107 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
108 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
109 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
110 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
111 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
112 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
113 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
114 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
115 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
116 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
117 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
118 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
119 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
120 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
121 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
122 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
123 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
124 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
125 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
126 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
127 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
128 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
129 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
130 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
131 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
132 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
133 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
134 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
135 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
136 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
137 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
138 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
139 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
140 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
141 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
142 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
143 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
144 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
145 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
146 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
147 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
148 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
149 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
150 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
151 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
152 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
153 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
154 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
155 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
156 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
157 CH=CH₂), 5.56 (s, 1H, H7), 5.21-5.12 (m, 2H, CH₂-CH=CH₂), 4.93 (d, $J = 12.0$ Hz, 1H, O-
158 CH=CH

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3 $\underline{\text{CH}_2\text{-Ph}}$), 4.69 (d, $J = 12.0$ Hz, 1H, O- $\underline{\text{CH}_2\text{-Ph}}$), 4.61 (d, $J = 10.4$ Hz, 1H, H3), 4.39 (dd, $J =$
4 10.3, 5.3 Hz, 1H, H6), 4.09 (td, $J = 10.3, 5.3$ Hz, 1H, H5), 3.86-3.78 (m, 2H, H4, H6), 3.31 (s,
5 3H, OMe), 2.70-2.56 (m, 2H, $\underline{\text{CH}_2\text{-CH=CH}_2}$); ^{13}C NMR (75 MHz, CDCl_3): δ (ppm) 198.2
6 (C2), 137.7 (C_{qAr}), 137.0 (C_{qAr}), 131.5 ($\text{CH}_2\text{-}\underline{\text{CH}}\text{=CH}_2$), 129.2 (C_{Ar}), 128.4 (C_{Ar}), 128.3
7 (C_{Ar}), 127.9 (C_{Ar}), 127.8 (C_{Ar}), 126.2 (C_{Ar}), 118.9 ($\text{CH}_2\text{-CH=}\underline{\text{CH}_2}$), 102.2 (C1), 101.1 (C7),
8 82.4 (C4), 80.5 (C3), 73.4 (O- $\underline{\text{CH}_2\text{-Ph}}$), 68.8 (C6), 63.9 (C5), 48.6 (OMe), 35.4 ($\underline{\text{CH}_2\text{-}$
9 CH=CH_2); FT-IR (film): 3063, 3030, 2924, 2854, 1720, 1602, 1495, 1454, 1361, 1313,
10 1270, 1207, 1176, 1095, 1069, 1027, 910, 733, 712, 695, 604, 461 cm^{-1} ; HRMS (ESI): m/z
11 Calcd for $\text{C}_{24}\text{H}_{30}\text{NO}_6$ [$\text{M} + \text{NH}_4$] $^+$ 428.2073, found 428.2071; $[\alpha]_{\text{D}}^{20} = -7$ (CHCl_3 , $c = 2.5$).

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24 **(2R,4aR,6R,8R,8aR)-6-allyl-8-(benzyloxy)-6-methoxy-2-**

25 **phenyltetrahydropyrano[3,2-*d*][1,3]dioxin-7(6H)-one 12.** A solution of Dess-Martin
26 periodinane (236 mg, 0.557 mmol) in pyridine (2 mL) was stirred at room temperature
27 for 30 min, and added with a solution of alcohol **10** (230 mg, 0.557 mmol) in
28 dichloromethane (2 mL). After 10 h at room temperature, additional Dess-Martin
29 periodinane (236 mg, 0.557 mmol) was added to the reaction mixture. After 10 h, TLC
30 revealed complete consumption of the starting material, and dichloromethane (20 mL),
31 a saturated solution of NaHCO_3 (10 mL) and a saturated solution of $\text{Na}_2\text{S}_2\text{O}_3$ (10 mL)
32 were added. After 4 h, the organic layer was separated and the aqueous layer was
33 extracted with dichloromethane (3 x 20 mL). The combined organic layers were dried
34 (MgSO_4), filtered and concentrated under reduce pressure. After purification by silica gel
35 chromatography (cyclohexane/ethyl acetate 5:1), ketone **12** was obtained as a colorless
36 film (150 mg, 65%). $R_f = 0.82$ (silica, cyclohexane / ethyl acetate 3:1); ^1H NMR (300 MHz,
37 CDCl_3): δ (ppm) 7.53-7.29 (m, 10H, H_{Ar}), 5.76 (ddt, $J = 17.3, 10.2, 7.0$ Hz, 1H, $\text{CH}_2\text{-}$
38 $\underline{\text{CH}}\text{=CH}_2$), 5.60 (s, 1H, H7), 5.17-5.02 (m, 2H, $\text{CH}_2\text{-CH=}\underline{\text{CH}_2}$), 4.98 (d, $J = 12.2$ Hz, 1H, O-
39 $\underline{\text{CH}_2\text{-Ph}}$), 4.78 (d, $J = 12.2$ Hz, 1H, O- $\underline{\text{CH}_2\text{-Ph}}$), 4.43-4.40 (m, 1H, H6), 4.22 (d, $J = 10.2$ Hz,
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3 1H, H3), 3.98-3.76 (m, 3H, H4, H5, H6), 3.33 (s, 3H, OMe), 2.55-2.53 (m, 2H, CH₂-
4 CH=CH₂); ¹³C NMR (75 MHz, CDCl₃): δ (ppm) 201.4 (C2), 137.3 (C_{qAr}), 136.9 (C_{qAr}), 129.8
5 (CH₂-CH=CH₂), 129.3 (C_{Ar}), 128.6 (C_{Ar}), 128.4 (C_{Ar}), 128.20 (C_{Ar}), 128.17 (C_{Ar}), 126.2
6 (C_{Ar}), 119.8 (CH₂-CH=CH₂), 104.9 (C1), 101.1 (C7), 81.4 (C4), 81.3 (C3), 74.0 (O-CH₂-Ph),
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12 69.1 (C6), 65.1 (C5), 51.9 (OMe), 38.9 (CH₂-CH=CH₂); FT-IR (film): 3040, 2921, 2857,
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14 1735, 1640, 1497, 1455, 1441, 1370, 1243, 1214, 1132, 1076, 999, 876, 741, 699, 459
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16 cm⁻¹; HRMS (ESI): m/z Calcd for C₂₄H₃₀N₆O₆ [M + NH₄]⁺ 428.2073, found 428.2066; [α]_D²⁰
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18 = - 10 (CHCl₃, c = 2.1).

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21 **(2R,4aR,6S,7R,8R,8aR)-6-allyl-8-(benzyloxy)-6-methoxy-2-**

22 **phenylhexahydropyrano[3,2-d][1,3]dioxin-7-ol 15.** A solution of **11** (53mg,
23
24 0.129mmol) in dichloromethane/methanol 1:1 (1.3 mL) was cooled to 0 °C and added
25
26 with NaBH₄ (20 mg, 0.516 mmol). After 15 min, dichloromethane (2.5 mL) and a 2%
27
28 aqueous solution of AcOH (2.5 mL) were added, and the mixture was stirred at room
29
30 temperature for 30 min. After separation, the aqueous layer was extracted with
31
32 dichloromethane (3 x 10 mL), and the combined organic phases were dried (MgSO₄),
33
34 filtered and concentrated under reduce pressure. Purification by silica gel
35
36 chromatography (cyclohexane/ethyl acetate 4:1) gave alcohols **7** (18 mg, 34%) and **15**
37
38 (30 mg, 57%) as colorless films. R_f = 0.44 (silica, cyclohexane/ethyl acetate 3:1); ¹H NMR
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40 (300 MHz, CDCl₃): δ (ppm) 7.52-7.29 (m, 10H, H_{Ar}), 5.79 (dddd, J = 16.8, 10.4, 8.2, 6.3 Hz,
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42 1H, CH₂-CH=CH₂), 5.57 (s, 1H, H7), 5.19-5.11 (m, 2H, CH₂-CH=CH₂), 4.96 (d, J = 11.4 Hz,
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44 1H, O-CH₂-Ph), 4.78 (d, J = 11.4 Hz, 1H, O-CH₂-Ph), 4.33-4.28 (m, 1H, H6), 3.85 (t, J = 9.1
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46 Hz, 1H, H3), 3.80-3.64 (m, 3H, H6, H5, H2), 3.59 (t, J = 9.1 Hz, 1H, H4), 3.30 (s, 3H, OMe),
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48 2.63-2.49 (m, 2H, CH₂-CH=CH₂), 2.26 (d, 1H, OH, J = 6.9Hz); ¹³C NMR (75 MHz, CDCl₃): δ
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50 (ppm) 138.6 (C_{qAr}), 137.5 (C_{qAr}), 132.5 (CH₂-CH=CH₂), 129.1 (C_{Ar}), 128.5 (C_{Ar}), 128.4
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52 (C_{Ar}), 128.2 (C_{Ar}), 127.9 (C_{Ar}), 126.1 (C_{Ar}), 119.2 (CH₂-CH=CH₂), 101.8 (C1), 101.3 (C7),
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3 82.1 (C4), 79.5 (C3), 75.1 (O-CH₂-Ph), 73.6 (C2), 69.1 (C6), 63.6 (C5), 47.9 (OMe), 37.2
4 (CH₂-CH=CH₂); FT-IR (film): 3556, 2921, 2853, 1643, 1495, 1454, 1373, 1200, 1179,
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7 1084, 986, 919, 763, 736, 697, 463 cm⁻¹; HRMS (ESI): m/z Calcd for C₂₄H₃₂NO₆ [M +
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9 NH₄]⁺ 430.2230, found 430.2213; [α]_D²⁰ = +56 (CHCl₃, c = 1.0);

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11 **(2R,4aR,6R,7S,8R,8aR)-6-allyl-8-(benzyloxy)-6-methoxy-2-**
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14 **phenylhexahydropyrano[3,2-d][1,3]dioxin-7-ol 16.** A solution of **12** (26 mg, 0.06
15 mmol) in THF (0.6 mL) was cooled to -78 °C and added drop-wise with a 1M solution of
16 L-selectride in THF (0.3 mL, 0.30 mmol). After 20 min at -78 °C the mixture was
17 quenched with methanol. Dichloromethane (5 mL) and Rochelle salts (1 M, 5 mL) were
18 added, and after 18 h stirring at room temperature, the organic layer was separated and
19 the aqueous layer was extracted with dichloromethane (3 x 10 mL). The combined
20 organic phases were dried (MgSO₄), filtered, and concentrated under reduce pressure.
21 After purification by silica gel chromatography (dichloromethane/acetonitrile 97:3),
22 alcohol **16** was obtained as a colorless film (16 mg, 61%). R_f = 0.39 (silica, cyclohexane /
23 ethyl acetate 3:1); ¹H NMR (300 MHz, CDCl₃): δ (ppm) 7.52-7.28 (m, 10H, H_{Ar}), 5.70 (ddt,
24 J = 17.1, 10.2, 7.1 Hz, 1H, CH₂-CH=CH₂), 5.59 (s, 1H, H7), 5.09-4.93 (m, 2H, CH₂-CH=CH₂),
25 4.86 (d, J = 12.3 Hz, 1H, O-CH₂-Ph), 4.77 (d, J = 12.3 Hz, 1H, O-CH₂-Ph), 4.32-4.23 (m, 2H,
26 H6, H4), 4.03-4.01 (m, 1H, H2), 3.85-3.76 (m, 2H, H3, H6), 3.52 (ddd, J = 10.5, 9.3, 4.7 Hz,
27 1H, H5), 3.41 (s, 3H, OMe), 2.90 (d, J = 2.7Hz, 1H, OH), 2.56-2.54 (m, 2H, CH₂-CH=CH₂);
28 ¹³C NMR (75 MHz, CDCl₃): δ (ppm) 138.1 (C_{qAr}), 137.6 (C_{qAr}), 131.8 (CH₂-CH=CH₂),
29 129.1 (C_{Ar}), 128.6 (C_{Ar}), 128.4 (C_{Ar}), 128.2 (C_{Ar}), 127.9 (C_{Ar}), 126.2 (C_{Ar}), 119.2 (CH₂-
30 CH=CH₂), 101.7 (C7), 101.0 (C1), 79.4 (C4), 74.9 (C3), 73.1 (O-CH₂-Ph), 69.69 (C6), 69.67
31 (C2), 65.6 (C5), 49.4 (OMe), 35.2 (CH₂-CH=CH₂); FT-IR (film): 3527, 3070, 3031, 2922,
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60 2853, 1728, 1640, 1496, 1454, 1373, 1314, 1274, 1210, 1176, 1093, 1028, 999, 918, 743,

696 cm^{-1} ; HRMS (ESI): m/z Calcd for $\text{C}_{24}\text{H}_{32}\text{NO}_6$ $[\text{M} + \text{NH}_4]^+$ 430.2230, found 430.2240; $[\alpha]_{\text{D}}^{20} = +15$ (CHCl_3 , $c = 3.6$).

Preparation of quaternary methyl- α -D GlcNAc and ManNAc derivatives 17 and 18. A solution of **11** (48 mg, 0.116 mmol) and ammonium acetate (360 mg, 4.665 mmol) in THF / methanol (0.2 mL / 0.9 mL) was heated at 80 °C under microwave irradiation. After 10 min, NaBH_3CN (74 mg, 1.165 mmol) was added and the solution was heated at 80 °C under microwave irradiation for 5 mn. After cooling to room temperature, dichloromethane (5 mL) and NaHCO_3 (5 mL) were added. After separation, the aqueous phase was extracted with dichloromethane (3 x 10 mL), and the combined organic layers were dried (MgSO_4), filtered, and concentrated under reduce pressure. The residue was dissolved in dichloromethane (3 mL) and added with pyridine (190 μL , 2.33 mmol) and acetic anhydride (110 μL , 1.16 mmol). After 1 h, methanol (0.1 mL), dichloromethane (5 mL) and NaHCO_3 (5 mL) were added, and the phases were separated. The aqueous phase was extracted with dichloromethane (3 x 10 mL), and the combined organic layers were dried (MgSO_4), filtered, and concentrated under reduce pressure. Purification by silica gel chromatography (chloroforme/methanol 99.5:0.5) gave **17** (20 mg, 38% over two steps) and **18** (17 mg, 33% over two steps) as colorless films.

N-((2R,4aR,6S,7R,8R,8aS)-6-allyl-8-(benzyloxy)-6-methoxy-2-phenylhexahydropyrano[3,2-d][1,3]dioxin-7-yl)acetamide 17: $R_f = 0.33$ (silica, chloroforme/methanol 99.5:0.5); ^1H NMR (300 MHz, CDCl_3): δ (ppm) 7.50-7.23 (m, 10H, H_{Ar}), 5.72 (ddt, $J = 17.6, 9.6, 7.3$ Hz, 1H, $\text{CH}_2\text{-CH}=\text{CH}_2$), 5.55 (s, 1H, H7), 5.41 (d, $J = 10.2$ Hz, 1H, NH), 5.19-5.04 (m, 2H, $\text{CH}_2\text{-CH}=\text{CH}_2$), 4.86 (d, $J = 12.3$ Hz, 1H, O- CH_2 -Ph), 4.61 (d, $J = 12.3$ Hz, 1H, O- CH_2 -Ph), 4.28-4.22 (m, 2H, H2, H6), 3.78-3.57 (m, 4H, H6, H4, H3, H5), 3.20 (s, 3H, OMe), 2.45 (dd, $J = 14.1, 7.3$ Hz, 1H, $\text{CH}_2\text{-CH}=\text{CH}_2$), 2.25 (dd, $J = 14.1, 7.3$ Hz, 1H, $\text{CH}_2\text{-CH}=\text{CH}_2$), 1.86 (s, 3H, $\text{CH}_3\text{-C=O}$); ^{13}C NMR (75 MHz, CDCl_3): δ (ppm) 169.4 (C=O),

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3 138.8 (C_{qAr}), 137.5 (C_{qAr}), 131.4 (CH₂-CH=CH₂), 129.1 (C_{Ar}), 128.38 (C_{Ar}), 128.36 (C_{Ar}),
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5 128.1 (C_{Ar}), 127.6 (C_{Ar}), 126.1 (C_{Ar}), 119.3 (CH₂-CH=CH₂), 102.2 (C1), 101.2 (C7), 82.5
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7 (C4), 77.5 (C3), 74.1 (O-CH₂-Ph), 69.0 (C6), 63.6 (C5), 54.1 (C2), 47.9 (OMe), 37.6 (CH₂-
8
9 CH=CH₂), 23.7 (CH₃-C=O); FT-IR (film): 3286, 3065, 3033, 2923, 2856, 1656, 1515, 1498,
10
11 1371, 1313, 1213, 1172, 1087, 997, 915, 748, 696, 598, 478 cm⁻¹; HRMS (ESI): m/z Calcd
12
13 for C₂₄H₃₂NO₆ [M + H]⁺ 454.2230, found 454.2216; [α]_D²⁰ = + 47 (CHCl₃, c =12.5). **N-**
14
15 **((2R,4aR,6S,7S,8R,8aS)-6-allyl-8-(benzyloxy)-6-methoxy-2-**
16
17 **phenylhexahydropyrano[3,2-d][1,3]dioxin-7-yl)acetamide 18**; R_f = 0.28 (silica,
18
19 chloroforme/methanol 99.5:0.5); ¹H NMR (300 MHz, CDCl₃): δ (ppm) 7.44-7.15 (m, 10H,
20
21 H_{Ar}), 5.68 (dddd, J = 17.0, 10.2, 7.6, 6.6 Hz, 1H, CH₂-CH=CH₂), 5.51 (s, 1H, H7), 5.47 (d, J =
22
23 10.8 Hz, 1H, NH), 5.10-4.99 (m, 2H, CH₂-CH=CH₂), 4.75 (dd, J = 10.8, 4.5 Hz, 1H, H2), 4.69
24
25 (d, J = 12.0 Hz, 1H, O-CH₂-Ph), 4.47 (d, J = 12.0 Hz, 1H, O-CH₂-Ph), 4.19 (dd, J = 9.0, 2.7 Hz,
26
27 1H, H6), 4.07 (dd, J = 9.6, 4.5 Hz, 1H, H3), 3.74-3.54 (m, 3H, H6, H4, H5), 3.15 (s, 3H,
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29 OMe), 2.46 (dd, J = 15.1, 7.6 Hz, 1H, CH₂-CH=CH₂), 2.30 (dd, J = 15.1, 6.6 Hz, 1H, CH₂-
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31 CH=CH₂), 1.98 (s, 3H, CH₃-C=O); ¹³C NMR (75 MHz, CDCl₃): δ (ppm) 169.7 (C=O), 138.3
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33 (C_{qAr}), 137.3 (C_{qAr}), 130.7 (CH₂-CH=CH₂) 128.9 (C_{Ar}), 128.3 (C_{Ar}), 128.2 (C_{Ar}), 127.9 (C_{Ar}),
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35 127.5 (C_{Ar}), 126.1 (C_{Ar}), 119.4 (CH₂-CH=CH₂), 103.4 (C1), 101.7 (C7), 78.5 (C4), 74.1
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37 (C3), 71.4 (O-CH₂-Ph), 68.9 (C6), 64.5 (C5), 50.6 (C2), 48.0 (OMe), 36.6 (CH₂-CH=CH₂),
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39 23.7 (CH₃-C=O); FT-IR (film): 3358, 2960, 2922, 2852, 1654, 1538, 1455, 1374, 1260,
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41 1091, 1016, 798, 696, 498 cm⁻¹; HRMS (ESI): m/z Calcd for C₂₄H₃₂NO₆ [M + H]⁺
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43 454.2230, found 454.2243; [α]_D²⁰ = 16 (CHCl₃, c = 2.3).

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49 *Preparation of quaternary methyl-β-D GlcNAc and ManNAc derivatives 19 and 20.*

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53 A solution of **12** (65 mg, 0.158 mmol) and ammonium acetate (488 mg, 6.331 mmol) in
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55 THF / methanol (0.3 mL / 1.4 mL) was heated at 80 °C under microwave irradiation.
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57 After 10 min, NaBH₃CN (98 mg, 1.58 mmol) was added and the solution was heated at
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80 °C under microwave irradiation for 5 mn. After cooling to room temperature, dichloromethane (5 mL) and NaHCO₃ (5 mL) were added. After separation, the aqueous phase was extracted with dichloromethane (3 x 10 mL), and the combined organic layers were dried (MgSO₄), filtered, and concentrated under reduce pressure. The residue was dissolved in dichloromethane (3 mL) and added with pyridine (256 µL, 3.166 mmol) and acetic anhydride (150 µL, 1.58 mmol). After 1 h, methanol (0.2 mL), dichloromethane (5 mL) and NaHCO₃ (5 mL) were added, and the phases were separated. The aqueous phase was extracted with dichloromethane (3 x 10 mL), and the combined organic layers were dried (MgSO₄), filtered, and concentrated under reduce pressure. Purification by silica gel chromatography (dichloromethane/methanol from 96:4 to 93:7) gave **19** (23 mg, 32% over two steps) and **20** (21 mg, 29% over two steps) as colorless films. **N-((2R,4aR,6R,7R,8R,8aS)-6-allyl-8-(benzyloxy)-6-methoxy-2-phenylhexahydropyrano[3,2-d][1,3]dioxin-7-yl)acetamide 19**: R_f = 0.33 (silica, dichloromethane/methanol 95:5); ¹H NMR (300 MHz, CDCl₃): δ (ppm) 7.45-7.19 (m, 10H, H_{Ar}), 5.76 (ddt, J = 17.1, 10.1, 7.2 Hz, 1H, CH₂-CH=CH₂), 5.53 (s, 1H, H7), 5.08-4.99 (m, 3H, CH₂-CH=CH₂, NH), 4.86 (d, J = 12.4 Hz, 1H, O-CH₂-Ph), 4.66 (d, J = 12.4 Hz, 1H, O-CH₂-Ph), 4.41 (t, J = 9.2 Hz, 1H, H2), 4.23 (dd, J = 10.2, 4.8 Hz, 1H, H6), 3.85 (t, J = 9.3 Hz, 1H, H4), 3.69 (t, J = 10.2 Hz, 1H, H6), 3.57-3.44 (m, 2H, H3, H5), 3.34 (s, 3H, OMe), 2.44 (dd, J = 14.7, 7.2 Hz, 1H, CH₂-CH=CH₂), 2.35 (dd, J = 14.7, 7.2 Hz, 1H, CH₂-CH=CH₂), 1.87 (s, 3H, CH₃-C=O); ¹³C NMR (75 MHz, CDCl₃): δ (ppm) 169.6 (C=O), 138.4 (C_{qAr}), 137.3 (C_{qAr}), 131.3 (CH₂-CH=CH₂), 129.1 (C_{Ar}), 128.5 (C_{Ar}), 128.4 (C_{Ar}), 128.3 (C_{Ar}), 127.9 (C_{Ar}), 126.1 (C_{Ar}), 118.6 (CH₂-CH=CH₂), 102.1 (C1), 101.3 (C7), 82.6 (C4), 77.9 (C3), 73.4 (O-CH₂-Ph), 69.4 (C6), 64.9 (C5), 52.5 (C2), 49.2 (OMe), 36.9 (CH₂-CH=CH₂), 23.7 (CH₃-C=O); FT-IR (film): 3285, 2961, 2924, 2853, 1651, 1514, 1454, 1371, 1312, 1260, 1216, 1092, 1053, 1016, 920, 800, 735, 696, 497 cm⁻¹;
HRMS (ESI): m/z Calcd for C₂₄H₃₂NO₆ [M + H]⁺ 454.2230, found 454.2227; [α]_D²⁰ = + 15 (CHCl₃, c = 3.7). **N-((2R,4aR,6R,7S,8R,8aS)-6-allyl-8-(benzyloxy)-6-methoxy-2-phenylhexahydropyrano[3,2-d][1,3]dioxin-7-yl)acetamide 20**: R_f = 0.30 (silica, dichloromethane/methanol 95:5); ¹H NMR (300 MHz, CDCl₃): δ (ppm) 7.45-7.28 (m, 10H, H_{Ar}), 5.98 (d, J = 9.9 Hz, 1H, NH), 5.75 (ddt, J = 17.2, 10.2, 7.5 Hz, 1H, CH₂-CH=CH₂), 5.54 (s, 1H, H7), 5.10-5.03 (m, 2H, CH₂-CH=CH₂), 4.75 (d, J = 12.0 Hz, 1H, O-CH₂-Ph), 4.69-4.62 (m, 2H, O-CH₂-Ph, H2), 4.39-4.30 (m, 1H, H6), 4.20-4.14 (m, 1H, H4), 3.95 (t, J = 6.9 Hz, 1H, H3), 3.69-3.66 (m, 2H, H6, H5), 3.32 (s, 3H, OMe), 2.57 (dd, J = 14.5, 7.5 Hz,

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3 1H, $\underline{\text{CH}}_2\text{-CH=CH}_2$), 2.39 (dd, $J = 14.5, 7.5$ Hz, 1H, $\underline{\text{CH}}_2\text{-CH=CH}_2$), 1.95 (s, 3H, $\underline{\text{CH}}_3\text{-C=O}$); ^{13}C
4 NMR (75 MHz, CDCl_3): δ (ppm) 168.8 (C=O), 137.1 (C_{qAr}), 136.4 (C_{qAr}), 130.9 ($\text{CH}_2\text{-}$
5 $\underline{\text{CH}}=\text{CH}_2$), 128.1 (C_{Ar}), 127.5 (C_{Ar}), 127.4 (C_{Ar}), 127.1 (C_{Ar}), 126.9 (C_{Ar}), 125.2 (C_{Ar}), 117.6
6 ($\text{CH}_2\text{-CH}=\underline{\text{CH}}$), 100.7 (C1), 100.3 (C7), 80.2 (C4), 74.1 (C3), 72.9 (O- $\underline{\text{CH}}_2\text{-Ph}$), 69.0 (C6),
7 (C5), 49.3 (C2), 47.2 (OMe), 36.9 ($\underline{\text{CH}}_2\text{-CH=CH}_2$), 22.4 ($\underline{\text{CH}}_3\text{-C=O}$); FT-IR (film): 2962,
8 2922, 2852, 1632, 1464, 1260, 1093, 1018, 798, 695, 458 cm^{-1} ; HRMS (ESI): m/z Calcd
9 for $\text{C}_{24}\text{H}_{32}\text{NO}_6$ [$\text{M} + \text{H}$] $^+$ 454.2230, found 454.2228; $[\alpha]_{\text{D}}^{20} = +11$ (CHCl_3 , $c = 4.5$);

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13 **(1R,2R,3S,4S,5R)-5-allyl-3-(benzyloxy)-6,8-dioxabicyclo[3.2.1]octane-2,4-diol 13.**

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15 $R_f = 0.17$ (silica, cyclohexane/ethyl acetate 5:1); ^1H NMR (300 MHz, CDCl_3): δ (ppm)
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17 7.39-7.31 (m, 5H, H_{Ar}), 5.87 (ddt, $J = 17.2, 10.1, 7.3$ Hz, 1H, $\text{CH}_2\text{-}\underline{\text{CH}}=\text{CH}_2$), 5.21-5.13 (m,
18 2H, $\text{CH}_2\text{-CH}=\underline{\text{CH}}_2$), 4.71 (d, $J = 11.4$ Hz, 1H, O- $\underline{\text{CH}}_2\text{-Ph}$), 4.59 (d, $J = 11.4$ Hz, 1H, O- $\underline{\text{CH}}_2\text{-Ph}$),
19 4.45 (d, $J = 5.7$ Hz, 1H, H2), 4.19 (d, $J = 7.5$ Hz, 1H, H6), 3.86-3.77 (m, 3H, H5, H6, H3),
20 3.67 (dd, $J = 11.4, 6.0$ Hz, 1H, H4), 3.04 (d, $J = 11.4$ Hz, 1H, OH), 2.73-2.59 (m, 2H, $\underline{\text{CH}}_2\text{-}$
21 CH=CH_2), 2.36 (br d, 9.9 Hz, 1H, OH); ^{13}C NMR (75 MHz, CDCl_3): δ (ppm) 137.3 (C_{qAr}),
22 131.9 ($\text{CH}_2\text{-}\underline{\text{CH}}=\text{CH}_2$), 128.8 (C_{Ar}), 128.3 (C_{Ar}), 127.9 (C_{Ar}), 118.8 ($\text{CH}_2\text{-CH}=\underline{\text{CH}}_2$), 108.8
23 (C1), 78.8 (C3), 77.1 (C2), 74.1 (O- $\underline{\text{CH}}_2\text{-Ph}$), 69.6 (C5), 67.5 (C4), 65.9 (C6), 37.7 ($\underline{\text{CH}}_2\text{-}$
24 CH=CH_2); FT-IR (film): 3427, 3073, 3034, 2897, 1718, 1642, 1454, 1400, 1207, 1063,
25 918, 796, 736, 696, 608, 460 cm^{-1} ; $[\alpha]_{\text{D}}^{20} = -10$ (CHCl_3 , $c = 4.9$).

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40 **(2R,4aR,6R)-6-allyl-8-hydroxy-6-methoxy-2-phenyl-4,4a-dihydropyrano[3,2-**

41 **d][1,3]dioxin-7(6H)-one 14.** $R_f = 0.11$ (silica, cyclohexane/ethyl acetate 5:1); ^1H NMR
42 (300 MHz, CDCl_3): δ (ppm) 7.57-7.29 (m, 5H, H_{Ar}), 5.97-5.83 (m, 2H, H7, $\text{CH}_2\text{-}\underline{\text{CH}}=\text{CH}_2$),
43 5.63 (br s, 1H, OH), 5.22-5.16 (m, 2H, $\text{CH}_2\text{-CH}=\underline{\text{CH}}_2$), 4.82 (dd, $J = 10.1, 6.8$ Hz, 1H, H5),
44 4.51 (dd, $J = 10.1, 6.8$ Hz, 1H, H6), 3.96 (t, $J = 10.1$ Hz, 1H, H6), 3.34 (s, 3H, OMe), 2.71-
45 2.55 (m, 2H, $\underline{\text{CH}}_2\text{-CH=CH}_2$); ^{13}C NMR (75 MHz, CDCl_3): δ (ppm) 190.2 (C2), 148.0 (C3),
46 135.2 (C_{qAr}), 131.7 (C4), 130.4 ($\text{CH}_2\text{-}\underline{\text{CH}}=\text{CH}_2$), 130.2 (C_{Ar}), 128.7 (C_{Ar}), 126.5 (C_{Ar}), 119.7
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3 (CH₂-CH=CH₂), 103.3 (C7), 102.7 (C1), 69.0 (C6), 62.3 (C5), 51.9 (OMe), 38.3 (CH₂-
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5 CH=CH₂).
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8 9 **Associated Content**

10 The supporting Information is available free of charge on the ACS publications website.

11 They include copy of the NMR spectra (¹H and ¹³C) of **5, 6, 7, 8, 9, 10, 11, 12, 13, 14, 15,**
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16 **16, 17, 18, 19, 20.**
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18 19 **Author Informations**

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22 The authors declare no competing financial interest.
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36 diazo sugars.
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