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A Convenient Synthesis of 2H-3,4-Dihydro-1,4-Benzothiazines

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Abstract: Aminothioalkenols react with paratoluenesulfonic acid in benzene or with phosphoric acid in toluene, and provide mainly heterocyclization in 2H-3,4-dihydro-1,4- benzothiazines; formation of some indenes and diene (in one case) was observed too.

Introduction

Dihydro-1,4-benzothiazines are compounds of pharmacological interest¹. In the last two decades, steady progress in their chemistry has been made along with the discovery of this structural unit in pheomelanins and trichochromes (characteristic pigments of red human hair)².

These compounds are usually prepared from acyclic precursors¹⁻³ or by ring expansion of benzothiazoles and benzothiazolines⁴. One drawback of such

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a synthetic strategy is that, each time, different precursors are required for the construction of the thiazine ring (yields are seldom high).

A more convenient method involves the functionalisation of simple and easily available precursors containing the thiazine ring⁵.

The literature provides only scanty information on 2H-3,4dihydro-1,4-benzothiazines; it has been proposed that benzothiazinones could partake, to some extent, in the building up of the 1,4- benzothiazines units^{1.6}.

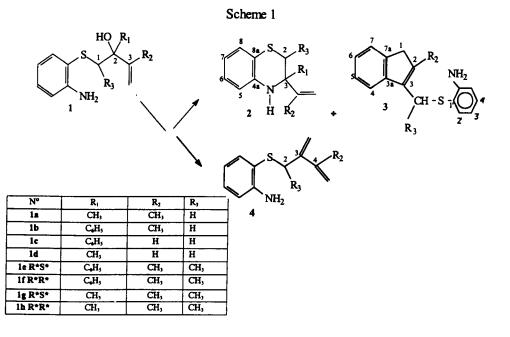
In connection with our studies on the chemistry of arylsulfides⁷, we wish to report a direct new route to produce various 3- alkylidene-2H-3,4-dihydro-1,4- benzothiazines 2. The method illustrates the successful utilization of aminothioalkenols 1 and represents a novel intramolecular approach to substituted benzothiazines.

Moreover, the reaction used permitted the introduction of various substituents in the 2 and 3 positions without requiring construction of the benzothiazine ring from acyclic precursors each time a different substituent was desired.

As far as we know, there have been no previous reports on the generation of compounds 2 with an alkene substituent in the 3 position.

Results and discussion

The synthetic steps leading to the formation of compounds 2a-h are outlined in Scheme 1.



The desired aminothioalkenols **1a-h** required for the present study were readily prepared from the corresponding ethylenic epoxides by nucleophilic attack of 2-amino thiophenol in basic medium with satisfactory yields^{7,8}. Synthesis of the new compounds **1e**, **1g R*S***and **1f**, **1h R*R*** was achieved from the corresponding diastereoisomeric vinyloxiranes; the stereochemical assignments for **2e - h** were derived from their ¹H , ¹³C NMR spectra and from our own findings on aminoalcohols⁹. In a typical experiment, when **1** was heated in the presence of paratoluenesulfonic acid (TsOH) in benzene (method A) or with phosphoric acid in xylene (method B), the reaction mixture after work-up afforded predominantly benzothiazines **2** and sometimes a mixture of

Substrate	Method	Total %	Time	Products % (ratio)	
		yield	(h)	2	3
1a	А	77	2.5	87(1)	-
	В	75	0.5	80(1)	-
1b	Α	55	5	55	45
	В	53	1.5	66	34
1¢	A	45	2	86	14
	В	65	0.75	80	20
1d	A	70	10.5	100	-
	В	60	1	100	-
1e R*S*	Α	50	10	87	13
	В	70	1	50	50
1f R*R*	A	73	10	82	18
	В	60	1	87	13
1g R*S*	Α	70	8	100	-
	В	65	1.5	100	
1h R*R*	Α	60	7	100	-
	В	50	1.5	100	[*

Table 1: reaction of Aminothioalkenols 1 with TsOH (method A) or H₃PO₄ (method B)

(1) Compound 4 was obtained only with 1a. Product 4a, yield%, A/B: 13/20.

2 with indenes 3 and diene 4a (Table 1). The products 2, 3, 4a were purified by column chromatography on silica gel and identified through their analytical and spectral data. All compounds were investigated by ¹H, ¹³C NMR and IRFT spectra.

The high resolution IR spectrum showed for 2, a peak about 3390 cm⁻¹ (NH stretching vibrations of non-associated species) and absence of hydroxyl (3300 cm⁻¹) absorption. For the conjugated system 4a two absorption bands were observed at 1650 cm⁻¹ and 1600 cm⁻¹, vinylidene groups were noted at 890 cm⁻¹. The ¹³C NMR data (experimental part) provided support for the structural assignments of 2 and 3: > in 2, substantial evidence for the intramolecular cyclization of 1:

- an upfield shift of about 19 ppm (C₃) and 14 ppm (C₂) in **2a** (compared with **1a**: C₁: 46.77 ppm, C₂: 74.87 ppm)⁷.

- the ¹JCH coupling constants for carbons of the Ar-S-CH₂ moiety in 2a ¹J=142 Hz (in 1a:145.2 Hz)⁷ were similar to methylene coupling constants (about140 Hz) in benzothiazines ring systems^{10,11}.

> in **3b** for example the absence of the vinylic carbons (C=CH₂), while two sp² quaternary carbon pairs at C₂/C₃ appear at about 145 / 132 ppm and two methylene sp³ CH₂-S and CH₂-1 absorbed respectively at 29.54 ppm ('J=132 Hz) and 42.59 ppm ('J=127 Hz).

We found that arylaminothioalcenols 1e R*S* and 1h R*R* underwent cyclization to give in each case two diastereoisomers : 2e (60%/40%) and 2h (55%/45%). In contrast, the reaction of 1f R*R* and 1g R*S* provided only *one* diastereoisomer, 2f (identical with 2e 60%), and 2g (identical with 2h 55%) respectively.

Apparently, if we take into consideration structures 2-4, they testify the involvement of a delocalized carbocation developed on C_2 in the presence of acids, which is significantly stabilized along the side chain of reactants 1; the presence of an aryl group in compounds 1 ($R_1=C_6H_5$) - a good stabilizing substituent- which increases the delocalization in the benzylic- allylic cationic

system, permits cyclization with ortho-aryl position (indenes 3b, 3c, 3e and 3f).

On the other hand, the existence of a carbocationic species allows us to understand the formation of a diene 4a which was isolated only in the case of 1a (not with 1d, 1g and 1h).

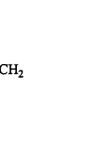
In the light of the results obtained with arylaminothioalkenols $1 \in \mathbb{R}^{*}S^{*}$, **1h** $\mathbb{R}^{*}\mathbb{R}^{*}$ we think that the cyclization into benzothiazines 2e, 2h is not only subject to charge control but also affected principally by steric hindrance. The selection observed, that is to say the formation of one diastereoisomer 2f or 2g(from 1f or 1g), and of two diastereoisomers 2e or 2h (from 1e or 1h), could be rationalized by assuming that carbon-2 undergoes the nucleophilic attack of the nitrogen lone pair from the less sterically hindered site of the *incipient* carbocation. If we suppose that a phenyl group (\mathbb{R}_1) cause less steric hindrance than an isopropenyl group, scheme 2A can depict the main conformation of $1g \mathbb{R}^*S^*$ and $1f \mathbb{R}^*\mathbb{R}^*$; in such a conformation the lone pair of the nitrogen atom might be able to attack rapidly the *incipient* carbocation.

The cis geometry of the CH₃ group attached at C_2 and C_3 of benzothiazine **2g** based upon NMR ¹³C spectral evidence (gamma gauche effect) is consistent with this mechanism.

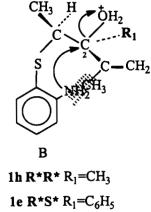
In contrast, scheme 2B represents the main conformation of 1e R*S* and 1h R*R* and shows steric hindrance between NH₂-Ar-S and the isopropenyl

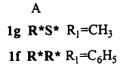
OH,

 $NH_2 \mathbf{R}_1 CH_3$



Scheme 2





CH

group. Thus the isolation of two diastereoisomers in each case, supports the total development of the carbocation intermediate at C_2 . Then the lone pair of the nitrogen atom is able to approach either side of the positive carbon to give a mixture of diastereoisomers : **2e** 60/40 (or **2h**: 55/45).

Conclusion.

In conclusion all arylaminothioalkenols are readily heterocyclized and the reaction used permits the introduction of various susbtituents in the 2 and 3 positions without requiring construction of the benzothiazine ring from acyclic precursors each time a different substituent is desired.

It is worth noting that the outcome of the reaction depends on the degree of substitution of the NH_2 moiety in compounds 1. We have reported earlier⁷ that if a tosyl moiety (NH-Ts) is substituted to the NH_2 group in

compounds 1, heterocyclization at C_1 is predominant and the balance is tipped in favor of benzothiazoles.

Experimental

IR spectra were recorded on a 16 PC FTIR Perkin-Elmer Spectrometer. Solids were examined with a diffuse reflectance accessory. For liquids, a horizontal attenuated total reflectance (HATR) accessory with a ZnSe crystal was used.¹H NMR and ¹³C NMR spectra were recorded using a Brucker DMX (at 500 MHz and 125 MHz respectively). CDCl₃ was used as solvent with TMS as the internal standard; chemical shifts (δ) were in ppm. Mass spectrometric analyses were performed using a Varian MAT 311 instrument. All compounds were purified by column chromatography with silica gel 60 (70-230 mesh) purchased from Merck.Arylaminothioalkenols **1a-d** are described by us in⁷.

Synthesis of new arylaminothioalkenols 1e-h

To a solution of KOH (0.05M) in absolute ethanol was added freshly distilled orthoaminothiophenol (0.05M). After cooling at room temperature, 0.05M of each diastereoisomer of the corresponding epoxide was added. The reaction occurred at room temperature (16 h) when R_3 =CH₃ but was heated at 80°C for 15 h when R_3 =C₆H₅. Water was added to the mixture and after extraction with ether, the organic layer was dried with sodium sulfate and evaporated. Each product was purified by chromatography on a silica gel column. **2-hydroxy-2-phenyl-1,3-dimethyl-1-(2-aminophenylthio)-3 butene** \mathbb{R}^*S^* :1e Yield 50% (chromatography in CH₂Cl₂/CH₃COOC₂H₅ 90/10);mp 86-88°C. IR: υ cm⁻¹:NH2 3409-3340, C=CH2: 902. ¹H NMR δ ppm: CH₃-1(d: 1.23); CH₃-3(m:1.60); CH-1. (m: 4.0-4.05); C=CH2(m: 4.84-5.30); Ar (m: 6.63-7.53). ¹³C NMR δ ppm:CH₃-1 (15.88¹J= 128.37); CH₃-3 (19.35 ¹J=127.15); C₁ (49.51 ¹J=139.26); C₂ (81.17 ¹J=161.07); C₄ (111.15 ¹J=157.43) C3' (115.32 ¹J= 157.43); C_{1'} (116.41); C_{5'} (118.98); C_{2''}(126.08 ¹J=156.22); C_{4''} (126.86 ¹J=159.85); C_{3''} (127.78 ¹J=159.85); C_{4'} (130.40 ¹J=157.43); C_{6'} (138.41 ¹J=161.07);C_{1''} (144.71); C_{2'}(147.11) C₃ (149.05).HR-SM for C₁₈H₂₁NO,cal.267.3396,found267.3399

2-hydroxy-2-phenyl-1,3-dimethyl-1-(2-aminophenylthio)-3 buteneR*R*:1f Yield 56% (chromatography with CH₂Cl₂). IR \cup cm⁻¹:NH2: 3409-3340, C=CH₂: 900. ¹H NMR δ ppm:CH₃-1(d: 0.95); CH₃-3(m:1.69); CH-1. (m: 4.0-4.07); C=CH2 (m: 4.90-5.37); Ar (m: 6.63-7.53). ¹³C NMR δ ppm:CH₃-1 (15.89¹J= 129.58); CH₃-3 (19.44 ¹J=127.16); C₁ (49.15 ¹J=139.27); C₂ (81.00); C₄ (111.34 1J=156.22);C3' (115.37 ¹J= 157.43);C₁· (115.96); C₅· (118.99); C₂··(125.60 ¹J=157.43); C₄·· (126.68 ¹J=159.86); C₃·· (127.86 ¹J=158.65); C₄· (130.54 ¹J=158.65); C₆· (138.64 ¹J=161.06); C₁·· (143.03); C₂·(149.30); C₃ (149.40). Anal. Calcd. for C₁₈H₂₁NOS: C, 74.24; H,7.02; N,4.68. Found: C, 75.0; H,6.80; N,4.70.

2-hydroxy-1,2,3-trimethyl-1-(2-aminophenylthio)-3 butene R*S*:1g

50% (chromatography with CH₂Cl₂); mp 55-57°C. IR υ cm⁻¹: NH2: 3459-3354, C=CH₂: 902. ¹H NMR:CH₃-2(s: 1.27); CH₃-1(d: 1.29); CH₃-3 (m: 2.04); CH-1. (q: 3.89); C=CH₂ (m: 4.92-5.12) ; Ar (m: 6.65-7.25). ¹³C NMR δ ppm: CH₃-1 (13.20 ¹J= 125.50); CH₃-3 (21.10 ¹J=128.20);CH₃-2 (22.78 ¹J=126.0); C₁ (50.47 ¹J= 139.43); C₂ (80.10); C₄ (110.53 ¹J= 158.20); C_{3'} (114.55 ¹J= 156.0); C_{1'} (116.42); C_{5'} (119.99);C_{4'} (132.20 ¹J=157.20); C_{6'} (139.0 ¹J=163.06); C_{2'}(149.10); C₃ (150.0).

2-hydroxy-1,2,3-trimethyl-1-(2-aminophenylthio)-3 butene R*R*:1h

70% (chromatography with $CH_2Cl_2/CH_3COOC_2H_5$: 80/20). IR $\cup cm^{-1}$:NH2: 3459-3354, C=CH₂: 900. ¹H NMR δ ppm:CH₃-1 (d: 1.21); CH₃-2 (s: 1.36); CH₃-3(m: 1.76); CH-1. (q: 3.33); C=CH₂(m: 4.92-5.12) ; Ar (m: 6.65-7.25). ¹³C NMR δ ppm:CH₃-1 (15.26¹J= 124.50); CH₃-3 (19.10 ¹J=127.16);CH₃-2 (23.78 ¹J=126.82);C₁ (52.47 ¹J=140.43); C₂ (77.11); C₄ (111.531J=157.20); C₃·(115.55 ¹J= 156.45);C₁· (117.42); C₅· (118.99);C₄· (130.25 ¹J=157.65); C₆· (137.67 ¹J=163.06);C₂·(148.39); C₃ (149.45).Anal. Calcd. for C₁₃H₁₉NOS: C, 65.82; H,8.01; N,5.91. Found: C, 66.10; H,7.90; N,5.90.

Synthesis of 2*H*-3,4- dihydro-1,4-benzothiazines 2a-h: general procedure Method A (Method B): To a solution of the appropriate aminothioalkenol 1 (0.01 mol) in 50 mL of anhydrous benzene (anhydrous xylene),was added 0.001 mol of TsOH (H_3PO_4 85%). The reaction mixture was stirred at reflux for

the time determined in Table 1. After added water (20 mL) the solution was extracted with $(C_2H_5)_2O$. The extracts were dried (sodium sulfate), evaporated and purified by column chromatography (solvent for **2a-c**: $CH_2Cl_2/C_6H_5CH_3:50/50$, for **2d-f**: toluene and **2g-h**: CH_2Cl_2)

(3-propenyl-3-methyl)-2H-3,4-dihydro-1,4-benzothiazine: 2a

IR:0NH: 3390, CH=CH2: 1640. ¹H NMR δ ppm:CH₃-3 (s: 1.42); CH₃-C_(1') (s:1.83); CH₂-2 (2.7-3.03 J_{AB}= 12.8);CH₂-2'(m:4.88 à 5.13); Ar (m:7.14-7.46). ¹³C NMR δ ppm: CH₃-1' (18.0 ¹J=127); CH₃-3 (27.4 ¹J=127.8); CH₂-2 (33.9 ¹J=142.0 Hz); C₃ (55.2);C_{2'} (112.5 ¹J=156.1); C_{8a} (114.7) C₅ (115.0 ¹J=155.2); C₇ (117.5 ¹J=161.7); C₆ (125.6 ¹J=158.7); C₈ (127.3 ¹J=158.0; C4a (140. 9); C_{1'} (148.0). HR-SM for C₁₂H₁₅SN, cal. 205.09226, found 205.09252. Anal. Calcd. for C₁₂H₁₅SN:C, 70.24; H,7.32; N,6.83. Found: C, 70.14; H,7.30; N,6.80.

(3-propenyl-3-phenyl)-2H-3,4-dihydro-1,4-benzothiazine: 2b

IR: vcm^{-1} : NH: 3418. ¹H NMR δ ppm:CH₃-1' (s: 1.71); CH₃-C_(1') (s:1.83); CH₂-2 (3.28-3.37 J_{AB}= 12.6); CH₂-2'(m :4.97-5.28) ; Ar (m:6.64-7.43).¹³C NMR δ ppm:CH₃-1'(17.2 ¹J=126.0); CH₂-2 (31.9 ¹J=144.1); C₃ (60.74); C₂. (114.9 ¹J=157.2);C₅ (115.3 ¹J=155); C_{8a} (114.7); C₇ (117.9 ¹J=161.0); C₆ (125.5 J¹=158.7);C₃... (125.7); C₂... (126.4); C₄... (127.6); C₈ (128.6); C₁... (140.8); C₁. (148); C_{4a} (144.6). Anal. Calcd. for C₁₇H₁₇NS: C, 76.4; H,6.38; N,5.24. Found: C, 76.30; H,6.20; N,5.40.

(3-vinyl-3-phenyl)-2H-3,4-dihydro-1,4-benzothiazine: 2c

IR: vcm^{-1} : NH: 3396. ¹H NMR δ ppm: CH₂-2 (2.90-3.20 J_{AB}= 12.8); CH=CH₂-2'(m: 5.14-5.29); Ar (m: 6.57-7.41). ¹³C NMR δ ppm:CH₂-2 (36.1 ¹J= 144.0); C₃ (58.8); C_{2'} (114.5 ¹J=157.2); C₅ (115.2 ¹J=155.0); C_{8a} (116.0); C₇ (117.8 ¹J=161.0); C₆ (125.5 ¹J=158.8);C_{3''} (125.7);C_{2''} (126.5);C_{4''} (127.5); C₈ (128.5 ¹J=159.7); C_{1''} (140.9); C_{1'} (141.4 1J=155); C_{4a} (144.6).HR-SM for C₁₆H₁₅NS, cal. 253.0932, found 253.09252. Anal. Calcd. for C₁₆H₁₅SN:C, 75.89; H,5.93 N,5.53. Found: C, 76.10; H,5.90; N,5.50.

(3-vinyl-3-methyl)-2H-3,4-dihydro-1,4-benzothiazine: 2d

IR: υ cm⁻¹:NH: 3376. ¹H NMR δ ppm:CH₂-2 (2.69-2.79 J_{AB}= 12.5);CH₃-3 (s:1.36); CH=CH₂-2'(m: 5.20- 5.39); Ar (m:6.59-6.99). ¹³C NMR δ ppm: CH₃-3 (27.92 ¹J=127.6); CH₂-2 (35.76 ¹J=143.0); C₃ (52.39); C₂ (112.0 ¹J=156.1); C_{8a} (114.0); C₅ (115.0 ¹J=155.0); C₇ (117.0 ¹J=161.0); C₆ (125. 6 ¹J=159.7); C₈ (127.3 ¹J=158.0); C_{4a} (140.9); C₁ (142.0). Anal. Calcd. for C₁₁H₁₃SN : C, 69.10; H,7.02; N,7.32. Found: C, 70.10; H,6.90; N,7.30.

(2-methyl-3-propenyl-3-phenyl)-2H-3,4-dihydro-1,4-benzothiazine: 2e

(60/40)

IR: υ cm⁻¹:NH: 3458. ¹H NMR δ ppm:CH₃-2 (d:1.23; 1.39); CH₃-1' (s:1.60-1.63); CH-2 (q: 3.70; 3.80);CH2-2'(m: 4.97-5.28; 5.03-5.23); Ar (m: 6.57-7.48). ¹³C NMR δ ppm:CH₃-2 (19.01;19.27 ¹J=128.37); CH₃-1' (19.73 ¹J=125.95); C₂ (37.36;38.15 ¹J=139.27); C₃ (63.92;65.02); C₂ (111.94;113.32

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¹J=157.43); C_{8a} (115.24;115.77); C_5 (115.90;116.04 ¹J=161.07); C_7 (118.57;118.64 ¹J=162.27); C_6 (124.99;125.13 ¹J=157.43); $C_{2...}$ (126.48;126.24 ¹J=157.43); $C_{4...}$ (126.75;127.75 ¹J=161.07); $C_{3...}$ (127.51;127.69 ¹J=158.65);C (128.04;128.27 ¹J=159.85); $C_{1...}$ (139.92;140.09); C_{4a} (144.82); $C_{1..}$ (145.94; 147.46). Anal. Calcd. for $C_{18}H_{19}SN$ (mixture **2e,2f**): C, 76.87; H,6.76; N,4.98. Found: C, 76.90; H,6.63; N,5.03.

(2-methyl-3-propenyl-3-phenyl)-2H-3,4-dihydro-1,4-benzothiazine: 2f

IR: υcm^{-1} : NH: 3458. ¹H NMR δ ppm:CH₃-2 (d:1.39); CH₃-1' (s:1.63); CH-2 (q: 3.80); CH₂-2'(m: 5.03-5.23); Ar (m: 6.58-7.48). ¹³C NMR δ ppm:CH₃-2 (19.27¹J=128.37); CH₃-1' (19.73 ¹J=125.95); C₂ (38.15 ¹J=139.27); C₃ (63.92); C_{2'} (111.94 ¹J=157.43); C_{8a} (115.24); C₅ (115.90 ¹J=155.01); C₇ (118.57 ¹J=162.27); C₆ (124.99 ¹J=157.43); C_{2''} (126.48 ¹J=157.43); C_{4''} (126.75 ¹J=161.07); C_{3''} (127.51 ¹J=158.65); C₈ (128.04 ¹J=159.85); C_{1''} (139.92); C_{4a} (144.82); C_{1''} (145.94).

(2-methyl-3-propenyl-3-methyl)-2H-3,4-dihydro-1,4-benzothiazine: 2g

IR: υ cm⁻¹: NH: 3380. ¹H NMR δ ppm:CH₃-2 (d:1.26, J=6.70); CH₃-3 (s:1.41); CH₃-C₁. (s: 1.82);CH-2 (q:3.23 J=6.84); CH₂-2'(m: 5.0-5.02); Ar (m: 7.14-7.46). ¹³C NMR δ ppm:CH₃-2 (19.66 ¹J=128.0); CH₃-1' (27.8 ¹J=126.0); CH₃-3 (19.4 ¹J=127.15); C₂ (40.1 ¹J=138.06); C₃ (56.7); C₂. (112.5 ¹J=162.2); C_{8a} (113.9); C₅ (114.75 ¹J=154.0); C₇ (118.0); C₆ (125.18 ¹J=158.7); C₈ (127.48); C_{4a} (140.24); C₁. (147.95). Anal. Calcd. for C₁₃H₁₇SN (mixture of 2g, 2h)C, 71.23; H,7.76; N,6.39. Found: C, 72.10; H,7.75; N,6.45.

(2-methyl-3-propenyl-3-methyl)-2H-3,4-dihydro-1,4-benzothiazine:2h

(55/45)

IR: ccm^{-1} :NH:3390. ¹H NMR δ ppm:CH₃-2 (d:1.20;1.26 J=6.90); CH₃-3 (s: 1.34;1.41); CH₃-C_(1') (s:1.82) CH-2 (q: 2.97;3.23 J=6.87) CH₂-2'(m: 4.88-4.97; 5.0-5.02); Ar (m:7.14-7.46; 7.17- 7.48). ¹³C NMR δ ppm:CH₃-1' (18.0; 27.8 ¹J= 126.5); CH₃-2 (17.05;19.66 ¹J=127.16); CH₃-3 (19.4;23.6); C₂ (38.2;40.1¹J=141.6); C₃ (56.7;58.1); C₂ (111.0;112.5 ¹J=156.2); C_{8a} (114.7;113.9); C₅ (113.92; 114.75 ¹J=155.0); C₇ (117.58;118.0); C₆ (125.18;125.30 ¹J=160.0); C₈ (127.48;127.55); C_{4a} (140.24;140.44); C₁. (147.95;148.78).

[2-methyl-3-(phenylamino) thiomethylene]-1H-indene: 3b

IR: vcm^{-1} : NH₂:3458-3360. ¹H NMR δ ppm:CH₃-2 (s: 1.62);CH₂-3 (s: 3.20); CH₂-1 (s: 3.81); NH₂ (s: 4.26); Ar (m: 6.57-7.39). ¹³C NMR δ ppm:CH₃-2 (13.28 ¹J=125.94); CH₂-S (29.54, ¹J=132); CH₂-1 (42.59 1J=127); C₁. (117.3); C₃ (132.23)); C_{3a} (142.12); C_{7a} (142.52); C₂ (145.13); C₂. (149.18); (114.47); (118.11); (118.54); .18);(123.88); (126.12); (130.18); (137.45).HR-SM for C₁₇ H₁₇NS, cal. 267.0332, found, 267.0335. Anal. Calcd. for C₁₇H₁₇SN: C, 76.40; H,6.38; N,11.98. Found: C, 76.80; H,6.10; N,12.10.

[3-(phenylamino) thiomethylene]-1H-indene: 3c

IR: vcm^{-1} : NH₂:3455-3360. ¹H NMR δ ppm:CH₂-3 (s: 3.10); CH₂-1 (s: 3.90); NH₂ (s: 4.0); CH-2(d: 5.2); Ar (m: 6.57-7.50). ¹³C NMR δ ppm: CH₂-S (28.0, ¹J=135); CH₂-1 (42.50 ¹J=127); C₂ (111,15); C₁. (117.2); C₃ (132.50); C_{3a} (142.20); C_{7a} (142.52); C₂. (149.18); Ar: (114.54); (118.10); (118.54); (123.20); (123.80); (126.24); (131.18); (138.45). Anal. Calcd. for C₁₆H₁₅SN: C, 75.89; H,5.93; N,5.53. Found: C, 76.15; H,5.85; N,5.56.

[2-methyl-3-(phenylamino) thiomethyl]-1H-indene: 3e or 3f

IR: vcm^{-1} : NH₂:3460-3360. ¹H NMR δ ppm:CH₃-2 (s: 1.69); CH₃(CH-S) (d: 1.70); CH₂-1 (3.09-3.22 J_{AB}=22.57); NH₂ (s: 4.27); CH-S (q: 4.36-4.40); Ar (m: 6.51-7.54). ¹³C NMR δ ppm:CH₃-2 (13.73 ¹J=125.95); CH₃(CH-S) (19.39, ¹J=128.37); CH-S (40.64 ¹J=141.69); CH₂-1 (42.71 ¹J=128.37); C₁. (117.70); C₃ (136.84); C_{3a} (140.62); C_{7a} (142.53); C₂ (144.13); C₂. (149.14); Ar: (114.42); (117.92); (120.48); (123.29);(123.55); (125.75); (130.08); (137.56). Anal. Calcd. for C₁₈H₁₉SN :C, 76.87; H,6.76; N,4.98. Found: C, 76.90; H,6.70; N,5.10.

4-(amino phenylthio)-2-methylbutane-1,3-diene: 4a

IR: vcm^{-1} : C=CH₂: 1650,1600,890. ¹H NMR δ ppm:CH₃-2 (s: 1.84); CH₂-4 (s: 3.50); NH₂ (s: 4.20); CH₂-1, CH₂-3(dxd: 4.70-5.20); Ar (m: 6.51-7.30). ¹³C NMR δ ppm:CH₃-2 (21.06 ¹J=126.80); CH₂-4 (38.56 ¹J=141.50);CH₂-1 (113.39 ¹J=157); C₅ (114.72); CH₂- C₃ (txt: 115.53 ¹J=158); C₁ (118.09); C₅ (118.35 ${}^{1}J=161$); C₄. (136.54 ${}^{1}J=161$); C₂. (140.87); C₃ (143.31); C₂ (148.63). Anal. Calcd. for C₁₂H₁₅SN: C, 70.24; H,7.32; N,15.61. Found: C, 71.10; H,7.14; N,15.60.

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