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Synthesis of Some Novel 4-Imino-3,5,7-Trisubstituted Pyrido[2,3-d]Pyrimidine-2(1H)-Thiones and Their Nucleosides as Potential **Therapeutic Agents**

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SYNTHESIS OF SOME NOVEL 4-IMINO-3,5,7-TRISUBSTITUTED PYRIDO[2,3-d]PYRIMIDINE-2(1*H*)-THIONES AND THEIR NUCLEOSIDES AS POTENTIAL THERAPEUTIC AGENTS

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Some newer 4-imino-3,5,7-trisubstituted pyrido[2,3-*d*]pyrimidine-2(1*H*)-thiones were synthesized by the condensation of 2-amino-3-cyano-4,6-disubstituted pyridines with phenylisothiocyanate. The nucleosides viz., 4-imino-3,5,7-trisubstituted-1-(2,3,5-tri-*O*-benzoy]- β -*D*-ribofuranosyl)pyrido[2,3,-*d*]pyrimidine-2(1*H*)-thiones were synthesized by the condensation of trimethylsilyl derivatives of pyrido[2,3-*d*]pyrimidine with sugar namely β -*D*-ribofuranose 1-acetate-2,3,5-tribenzoate while 4-imino-3,5,7-trisubstituted-1-(2,3,5-tri-*O*-benzoy]- α -*D*-ribofuranosyl)pyrido[2,3-*d*]pyrimidine-2(1*H*)-thiones were prepared by condensing trimethylsilyl derivatives of pyrido[2,3-*d*]pyrimidine with sugar in presence of SnCl₄. All the synthesized nucleosides and their precursors were characterized by spectral and elemental analysis data and have been screened for their antimicrobial activities.

Keywords: Pyrido[2,3-d]pyrimidines; nucleosides ($\alpha & \beta$ anomers); spectral data and antimicrobial activity

INTRODUCTION

Pyrido[2,3-*d*]pyrimidines were originally synthesized as compounds bearing structural kinship to many potent chemotherapeutic agents¹⁻² like pteridine, aminopterin and methotrexate. Pyrido[2,3-*d*]pyrimidines have been reported to possess wide spectrum of biological activities such as diuretic³, antimalerial⁴, antiallergic⁵, anticancer⁶ antifungal⁷, CNS depressant⁸ and antiulcer⁹ etc. Significant attention has been paid recently

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for the synthesis of nucleosides of various heterocyclic bases as potential antiviral agents against Human Immunodeficiency Virus $(HIV)^{10-12}$. Our interest in the area of pyrido[2,3-d]pyrimidines and in continuation of our earlier work¹³ we wish to report here, the synthesis of some new 4-imino-3,5,7-trisubstituted pyrido[2,3-d]pyrimidine-2(1H)-thiones and their nucleosides ($\alpha \& \beta$ anomers).

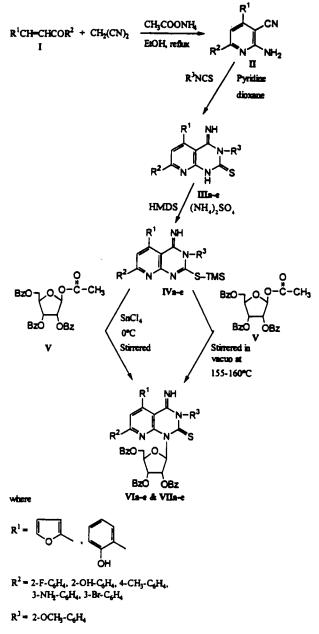
RESULTS AND DISCUSSION

4-Imino-3,5,7-trisubstituted pyrido[2,3-d]pyrimidine-2(1H)-thiones III were synthesized by the condensation of 2-amino-3-cyano-4,6-disubstituted pyridines II with arylisothiocyanate in dioxane and pyridine. Compounds II were synthesized by condensing chalcones I with malononitrile in the presence of ammonium acetate through Michael type reaction. 4-Imino-3,5,7-trisubstituted-1-(2,3,5-tri-O-benzoyl- β -D-ribofuranosyl) pyrido[2,3-d]pyrimidine-2(1H)-thiones VI were prepared by the condensation of trimethylsilyl derivatives of pyrido[2,3-d]pyrimidine IV with sugar namely β -D-ribofuranose 1-acetate-2,3,5-tribenzoate V in vacuum at 155–160°C whereas, 4-imino-3,5,7-trisubstituted-1-(2,3,5-tri-O-benzoyl- α -D-ribofuranosyl)pyrido[2,3-d]-pyrimidine-2(1H)-thiones VII were synthesized by condensing compounds IV with sugar V in presence of SnCl₄ at 0°C in 1,2-dichloroethane.

The trimethylsilyl derivatives of pyrido[2,3-d]pyrimidines IV were synthesized by the reaction of pyrido[2,3-d]pyrimidines III with hexamethyldisilazane in presence of few crystals of ammonium sulphate. Formation of VIa-e (β -anomers) may be attributed due to SN²-mechanism via neighbouring group participation and is in consonance with the earlier report. However, the formation of compounds VIIa-e (α -anomers) is perhaps due to coordination of SnCl₄ with -OCOCH₃ group of the sugar moiety and thus permitting condensation selectively. The reaction progress was monitored by TLC the products obtained were characterized by IR & ¹H NMR spectral data (Scheme 1).

SPECTRAL DATA

The proposed structure of the synthesized compounds are well supported by the spectral and elemental analysis data (Table. Ia, b).



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TABLE la Characterization data of synthesized 4-imino-3,5,7-trisubstituted pyrido[2,3-d]pyrimidine-2(1H)-thiones III a-e

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	Ta	20	3	Malandar Formula	N.al.d of	J. 9 M	Element	al analysis	Elemental analysis calc. (Found) %	% (pu
comp. No.	<	4	c			м.г. с	J	Н	2	s
IIIa	-C4H3O	-C ₄ H ₃ O 4-CH ₃ -C ₆ H ₄ 2-OCH ₃ -C ₆ H ₄	2-OCH ₃ -C ₆ H ₄	C ₂₅ H ₂₀ N ₄ O ₂ S	72	61	68.18	4.54	12.73	7.27
							(68.22)	(4.59)	(12.68)	(7.24)
qIII	-C4H3O	2-OH-C ₆ H ₄	2-0CH ₃ -C ₆ H ₄	C ₂₄ H ₁₈ N ₄ O ₃ S	68	82	65.16	4.07	12.67	7.24
							(65.21)	(4.12)	(12.62)	(1.21)
IIIc	-C4H3O	2-F-C ₆ H ₄	2-0CH ₃ -C ₆ H ₄	C ₂₄ H ₁₇ N ₄ O ₂ FS	81	85	64.86	3.83	12.61	7.21,
							(16.19)	(3 89)	(12 57)	(7.17)
PHI	-C ₄ H ₃ O	3-BrC ₆ H ₄	2-0CH ₃ -C ₆ H ₄	C ₂₄ H ₁₇ N ₄ O ₂ BrS	68	93	57.03	3.36	60.11	6.34
							(56.98)	(3 39)	(11.04)	(6.29)
lile	-C ₆ H ₅ O		3-NH ₂ -C ₆ H ₄ 2-OCH ₃ -C ₆ H ₄	C ₂₆ H ₂₀ N ₅ O ₂ S	75	90-92	66.95	4.29	15.02	6.87
							(16:99)	(4 25)	(14.97)	(181)

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;	14	70	2		V: .1.1 (4.	J. a M	Element	al analysis	Elemental analysis calc. (Found) %	% (pu
Comp. No.	×	×	×	Molecular Formula	or, man	י ה. איני ר	C	Н	N	S
Vla	-C ₄ H ₃ O	4-CH ₃ .C ₆ H ₄	2-OCH ₃ -C ₆ H ₄	C51H4009N4S	82	161	69.23	4 52	6 33	3 62
							(69.29)	(4.56)	(6.27)	(3.58)
VIb	-C4H3O	2-OH-C ₆ H ₄	2-OCH ₃ -C ₆ H ₄	C ₅₀ H ₃₈ O ₁₀ N ₄ S	61	145	67.72	4.29	6 32	3.61
							(67.77)	(4 35)	(6.26)	(3.56)
VIc	-C ₄ H ₃ O	2-F-C ₆ H ₄	2-0CH ₃ -C ₆ H ₄	C ₅₀ H ₃₇ O ₉ N ₄ FS	18	148	67.57	4.17	6.31	3.60
							(67.62)	(4.23)	(6.28)	(3.55)
ΡIΛ	-C ₄ H ₃ O	3-Br-C ₆ H ₄	2-0CH ₃ -C ₆ H ₄	C ₅₀ H ₃₇ O ₉ N ₄ BrS	76	156	63.22	3.90	5.90	3.37
							(63.27)	(3.97)	(5.85)	(3.32)
Vie	-C ₆ H ₅ O	3-NH ₂ -C ₆ H ₄	2-0CH ₃ -C ₆ H ₄	C ₅₂ H ₄₀ O ₉ N ₅ S	78	215	68.57	4.39	7.69	3.52
							(68.63)	(4.44)	(1.62)	(3.45)
VIIa	-C4H3O	4-CH ₃₋ C ₆ H ₄	2-0CH ₃ -C ₆ H ₄	C ₅₁ H ₄₀ O ₉ N ₄ S	72	230	69.23	4.52	6.33	3.62
							(69.28)	(4.55)	(6.26)	(3.57)
VIIb	-C4H3O	2-OH-C ₆ H ₄	2-OCH ₃ -C ₆ H ₄	C ₅₀ H ₃₈ O ₁₀ N ₄ S	68	245	67.72	4.29	6.32	3.61
							(67.76)	(4.34)	(6.25)	(3.55)
VIIc	-C4H3O	2-F-C ₆ H ₄	2-OCH ₃ -C ₆ H ₄	C ₅₀ H ₃₇ O ₉ N ₄ FS	11	250	67.57	4.17	6.31	3.60
							(19.79)	(4.22)	(6.27)	(3.55)
NII	-C ₄ H ₃ O	3-Br-C ₆ H ₄	2-OCH ₃ -C ₆ H ₄	C ₅₀ H ₃₇ O ₉ N ₄ BrS	69	235	63.22	3.90	5.90	3.37
							(63.26)	(3.96)	(5.84)	(18.8)
VIIe	-C ₆ H ₅ O	3-NH ₂ -C ₆ H ₄	2-OCH ₃ -C ₆ H ₄	C ₅₂ H ₄₀ O ₉ N ₅ S	73	280	68.57	4.39	7.69	3.52
							(68.62)	(4.45)	(7.63))	(3,44)

PYRIDO PYRIMIDINES

IR spectra of compounds II showed a sharp band in the region 2230–2210 cm⁻¹, indicating the presence of -C=N group. The stretching and bending vibrations due to - NH₂ group was appeared in the region 3450–3310 cm⁻¹ and 1525–1500 cm⁻¹ respectively. Disappearance of band due to -C=N and appearance of new bands due to >C=S, >C=NH, and >NH in the region 1220–1185, 3190–3160, and 3140–3110 cm⁻¹ respectively, indicated the formation of compounds III from compounds II. The absence of band due to >NH stretching vibrations in the spectra of compounds VI & VII indicated the site of ribosylation by substitution of hydrogen atom of >NH group. In compounds VI & VII, bands due to carbonyl group and C-O-C linkage were appeared at 1750–1725 cm⁻¹ and 1170–1050 cm⁻¹, respectively.

¹HNMR

All the synthesized compounds gave a complex multiplet for aromatic protons in the region of δ 7.0–7.9 ppm. The imino protons in compounds III, VI & VIIappeared as singlet at δ 8.9–9.1 ppm while the >NH proton showed their presence in compounds III at δ 8.1–8.3 ppm. Peaks due to -CH₃, -OCH₃, & -NH₂ protons were appeared in the region δ 2.1–2.3, δ 3.76–3.99 & δ 5.20–5.41 ppm, respectively while peak due to -OH proton was found to be merged with Ar-H wherever it present. Absence of peak due to >NH proton in compounds VI & VII confirm the ribosylation at this position & formation nucleosides. In compounds VI C₁-H i.e. anomeric proton appeared as singlet at δ 6.40 ppm (It should appear as doublet with J = 2–3 Hz but at 90 MHz it appeared as singlet) confirm the β configuration while it appeared as doublet at δ 6.52 ppm with J = 8Hz in compounds VII confirm the α configuration.

ANTIMICROBIAL ACTIVITY

Synthesized pyrido [2,3-d] pyrimidines and their nucleosides were evaluated for their antibacterial and antifungal activity following the method Bauer *et.* al^{14} , using streptomycin in antibacterial and mycostatin in antifungal activity, as the reference compounds. All the synthesized compounds showed moderate to good activity against the organisms viz. *Escherichia coli* (gram negative bacteria), *Staphylococcus aureus* (gram positive bacteria) *Aspergillus niger, Aspergillus flavus*, and *Fusarium oxysporium* (fungi). The results have been tabulated in the form of activity indices. A close look on activity indices indicated that nucleosides showed better activity than their precursor. Further, it was also observed thus α -anomer showed better activity than β -anomer in most of the cases (table II).

EXPERIMENTAL

Melting point of all the synthesized compounds were determined in open capillary tube and are uncorrected. The IR spectra were determined in KBr disc on a NICOLET MEGNA FT-IR 550 spectrometer and ¹HNMR spectra in CDCl₃/DMSO-d₆ on a FX 90Q JEOL type spectrophotometer using TMS as internal standard (chemical shift in δ ppm). The purity of compounds were checked by TLC using silica gel "G" as adsorbent and visualization was accomplished by UV light or iodine.

Chalcones were synthesized by reported methods¹⁵.

Synthesis of 2-amino-3-cyano-4,6-disubstituted pyridines II

A mixture of appropriate chalcones I (0.05 mole), malononitrile (0.05 mole) and ammonium acetate (0.4 mole) in ethanol (50 ml) was refluxed on a water-bath for 20-22 hrs, cooled and poured onto crushed ice with constant striring. The solid thus obtained was washed with water and cold ethanol and recrystallized from ethanol.

Synthesis of 4-imino-3,5,7-trisubstituted pyrido [2,3-*d*]pyrimidine -2(1*H*)-thiones III

Compounds II (0.01 mole), appropriate arylisothiocyanate (0.01 mole), dioxane (15ml) and pyridine (2.0 ml) was refluxed at 150°C at 20–22 hrs. After cooling, the contents of flask were poured onto crushed ice with constant stirring. The solid mass thus obtained was washed with water, aqueous sodium bicarbonate 5% (W/V) and finally with water. The dried crude mass was recrystallized from glacial acetic acid.

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TABLE II Antimicrobial activity of pyrido[2,3-d]pyrimidines and their nucleosides

Test organism							Inhibit	Inhibition Zone (mm)	(uuu).						
	IIIa	<i>q111</i>	1110	PIII	IIIe	Vla	VIb	VIC	PIA	Vle	VIIa	qIIA	VIIc	VIId	VIIe
Gram negative bacteria	14.0	13.0	14.5	15.0	14.5	16.0	17.0	18.0	19.0	19.5	20.5	17.5	18.5	19.0	20.0
E. coli	(0.96)	(0.95)	(0.1)	(1.02)	(0.1)	(1.04)	(1.08)	(01.10)	(1.12)	(1.14)	(1.17)	(1.09)	(111)	(1.12)	(1.15)
Gram positive bacteria	20.0	0.61	21.0	19.5	22.0	23.0	25.0	27.0	22.0	26.0	28.0	22.5	25.0	27.0	27.5
S. aureus	(0.99)	(0.97)	(0.1)	(0.98)	(1.02)	(1.05)	(1.12)	(1.16)	(1.02)	(1.14)	(1.20)	(1.03)	(1.12)	(1.16)	(1.18)
Fungi															
Aspergillus niger	9.9	9.6	9.7	10.2	6.6	9.11	12.5	11.7	12.5	12.0	10.0	11.5	12.0	11.5	12.0
	(1.02)	(0.98)	(0.1)	(1.05)	(1.02)	(61.19)	(1.28)	(1.21)	(1.28)	(1.24)	(1.03)	(1.17)	(1.26)	(1.17)	(1.24)
Aspergillus flavus	9.6	9.7	9.9	10.3	9.6	10.7	11.7	11.5	12.5	0.11	12.0	11.5	11.0	12.4	11.7
	(10.01)	(0.98)	(0.1)	(1.04)	(0.97)	(60'1)	(1.19)	(1.17)	(1.26)	(1.13)	(1.22)	(1.16)	(1.13)	(1.25)	(61.1)
Fusarium oxysporium	9.2	10.3	9.6	10.5	10.0	12.5	12.6	11.2	11.4	12.0	13.0	12.5	12.4	11.5	12.0
	(0.94)	(1.05)	(0.98)	(1.07)	(1.02)		(1.27) (1.28)	(1.14)	(1.16)	(1.22) (1.31) (1.27)	(181)	(1.27)	(1.25)	(1.18)	(1.22)
Activity index = Inhibition area of the sample/Inhibition area of the standard	on area of	the sam	ple/Inhil	bition arc	a of the	standard									

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Synthesis of 4-imino-3,5,7-trisubstituted-1-(2,3,5-tri-O-benzoyl- β -D-ribofuranosyl) pyrido[2,3-d]pyrimidine-2(1H)-thiones VI

Synthesized compounds III (0.02 mole) were refluxed with HMDS (hexamethyldisilazane) (0.0124 mole) alongwith a few crystals of ammonium sulphate in toluene (30 ml) for 8 hrs under anhydrous condition. The coloured solution thus obtained was filtered and the solvent was removed under vacuum at 100°C. The sugar viz. β -D-ribofuranose 1-acetate-2,3,5-tribenzoate (0.02 mole) was added to the above pasty mixture and it was stirred at 155–160°C under vacuum for 15 minutes in absence of moisture. The reaction mixture was stirred for 10 hrs. During the reaction period, the vacuum was regularly applied for five minutes, at the end of every hour. The melt was boiled in methanol for 10 minutes, cooled and filtered. The solid mass of nucleosides VI thus obtained was recrystallized from diethyl ether.

Synthesis of 4-imino-3,5,7-trisubstituted-1-(2,3,5-tri-O-benzoyl- α -D-ribofuranosyl) pyrido[2,3-d]pyrimidine-2(1H)-thiones VII

Compounds III (0.01 mole) was refluxed with HMDS (60 ml) alongwith a few crystals of ammonium sulphate in toluene (20 ml) for 8 hrs under anhydrous condition. The coloured solution thus obtained was filtered and the solvent was removed in under vacuum at 100°C. The above pasty mass was dissolved in anhydrous 1,2-dichloroethane (40 ml) and a solution of sugar (β -D-ribofuranose 1-acetate-2,3,5-tribenzoate) (0.011 mole) in dry 1,2- dichloroethane (5ml) was added with stirring. The mixture was cooled to 0°C and a solution of SnCl₄(1.6 ml) was added dropwise with stirring and the completion of reaction was judged by TLC (2,3 hrs.) and then poured onto saturated NaHCO₃ solution. It was extracted with chloroform, dried over anhydrous MgSO₄ and filtered to get compounds VII. It was recrystallized from EtOH.

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