

Synthesis of 1,2,3-Trisubstituted Pyrrolidines and 2,3-Disubstituted Tetrahydrofurans via Diastereoselective Reductive Cyclization of γ -Chloroimines and γ -Chloroketones

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Abstract: A new diastereoselective synthetic approach towards 1,2,3-trisubstituted pyrrolidines and 2,3-disubstituted tetrahydrofurans is described. The synthesis of the pyrrolidines involves reductive cyclization of γ -chloroketimines, which were generated in situ from the reaction of 3-substituted 5-chloro-2-pentanones and a primary amine. Various reduction conditions were explored to induce a diastereoselective reductive cyclization. 2,3-Disubstituted tetrahydrofurans were obtained by the direct reduction of 3-substituted 5-chloro-2-pentanones with sodium borohydride.

Key words: pyrrolidines, tetrahydrofurans, cyclization, haloimines, haloketones

Substituted pyrrolidines and tetrahydrofurans are common structural subunits found in a broad array of natural and unnatural products with interesting and diverse biological activities.² The 1,2,3-trifunctionalized pyrrolidine moiety is present in numerous natural products such as naturally occurring proline analogues.³ Besides this, 1,2,3-trisubstituted pyrrolidines are useful building blocks for the synthesis of various natural products, as exemplified by the successful synthesis of the natural 1-amidopyrrolizidine alkaloid, (\pm)-laburnamine, starting from a 1,2,3-trisubstituted pyrrolidine precursor.⁴ Furthermore, several 1,2,3-trisubstituted pyrrolidine derivatives exhibit bioactivities such as antitumor, antiosteoarthritis,⁵ and antipsychotic activities.⁶ On the other hand, the substituted tetrahydrofuran motif is a common core structure of a large family of natural products, namely the annonaceous acetogenins, which possess a range of biological properties such as anthelmintic, antimalarial, antimicrobial, antiprotozoal, cytotoxic, and antitumor activities.⁷ In addition, substituted tetrahydrofurans are also found in many other classes of natural products including lignans,⁸ polyether ionophores,⁹ and macrodiolides.¹⁰ The significant biological interests associated with these compounds has stimulated the scientific community to develop a number of methods for the synthesis of substituted pyrrolidines and tetrahydrofurans. Common ring-formation methods involve radical, electrophilic,¹¹ and, more recently, palladium-catalyzed cyclizations^{2,12} of amino- and hydroxyalkenes,¹³ 1,3-dipolar cycloadditions,¹⁴ reductive

cyclizations of γ -amino- and γ -hydroxyketones, olefin metathesis reactions, and intramolecular cyclizations by S_N2 nucleophilic substitution.^{15,16} The latter method mainly involves the stereospecific intramolecular S_N2 cyclization of an acyclic substrate in which all stereocenters are already in place.^{15,16} In the present paper, an alternative, unexplored synthetic route to 1,2,3-trisubstituted pyrrolidines via diastereoselective reductive S_N2 cyclization of γ -chloroketimines and a similar approach for the synthesis of 2,3-disubstituted tetrahydrofurans using γ -chloroketones is disclosed,^{17,18} in extension of our research on the application of ω -functionalized imines, and more specifically ω -chloroimines,¹⁹ for the synthesis of heterocyclic compounds.²⁰ Whereas α -, β -, and δ -haloimines have been used successfully in the reductive cyclization to the corresponding aziridines,^{19d-19f} azetidines^{19a} and piperidines,^{19c,d} respectively, a similar approach towards the five-membered azaheterocycles via reduction of γ -halogenated imines has proven elusive up to now.^{19h,20} This important gap in the application of haloimines towards the synthesis of azaheterocycles is partly filled by the results presented herein.

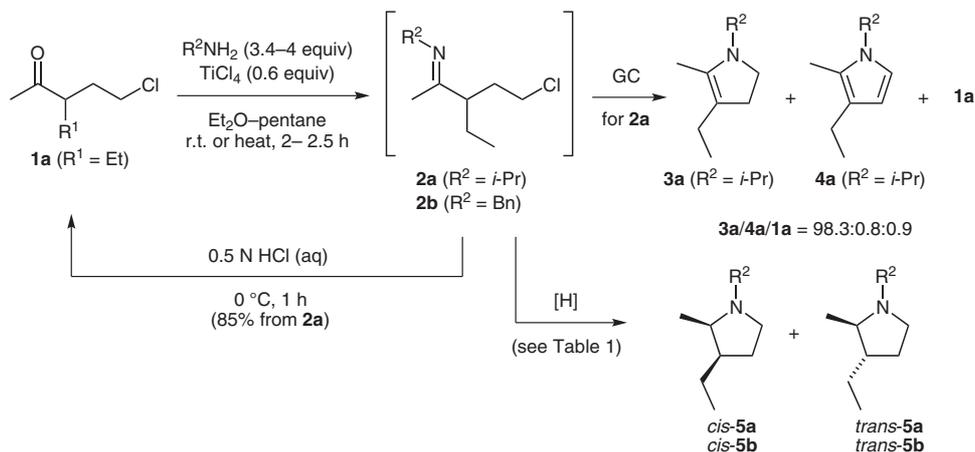
The required starting compounds, that is, 3-alkyl-5-chloro-2-pentanones **1**, were prepared in analogy with alkylation procedures on 2-acetyl- γ -butyrolactone reported in the literature.²¹ The imination of γ -chloroketone **1a** with isopropylamine (4 equiv, r.t.) or benzylamine (3.4 equiv, reflux), in the presence of 0.6 equivalents titanium(IV) chloride,²² resulted in reaction mixtures which were difficult to characterize (Scheme 1). The obtained intermediate imines **2** were not stable enough for characterization by ¹H NMR or ¹³C NMR spectroscopy, and all attempts to isolate them in pure form by distillation failed. The imines **2** tend to cyclize and decompose upon storage even at low temperature. Some analytical and chemical proof for the formation and reactivity of ketimine **2a** was obtained as follows. GC-MS analysis (HP5-MS capillary column) of the reaction mixture containing imine **2a** resulted in thermolysis during analysis with formation of one large peak, together with two minor peaks in a ratio of 98.3:0.8:0.9. The large peak was assigned to pyrroline **3a** (MW = 153), resulting from intramolecular N-alkylation with elimination of HCl. The first eluting minor peak corresponded with the γ -chloroketone **1a** (m/z = 148/150). The mass spectrum of the second minor peak showed a signal at m/z = 151 and corresponded with the molecular weight of

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Scheme 1

pyrrole **4a**. All attempts to repeat the thermolysis of imine **2a** under more standard conditions by heating in EtOH or DMF under reflux or by microwave heating in DMF or neat resulted in complex reaction mixtures.

The hydrolysis of imine **2a** under mild reaction conditions upon treatment with aqueous 0.5 N HCl at 0 °C for one hour resulted in the regeneration of γ -chloroketone **1a** in 85% yield.

The following reductive cyclizations, with the aim to induce diastereoselectivity during the reduction, were performed on the crude reaction mixtures of imines **2** (Scheme 1, Table 1 and Table S1 in the Supporting Information). Reduction of imine **2a** with two equivalents of sodium borohydride in methanol under reflux resulted in a clean conversion into pyrrolidines *cis*-**5a** and *trans*-**5a**, which were isolated by vacuum distillation as an inseparable 1:1 mixture in 67% overall yield for the two-step process from ketone **1a** (Table 1, entry 1). The *cis/trans* ratios and degree of conversion into pyrrolidines **5** were

determined by GC. Analogously, reduction of *N*-benzyl imine **2b** with sodium borohydride resulted in the isolation of pyrrolidines *cis*-**5b** and *trans*-**5b** as a 1:1 mixture in 78% overall yield (Table 1, entry 2). The pyrrole **4b** was present as a small impurity in the reaction mixture (ca. 6% determined by GC) and the distillation fraction, and was identified by NMR spectroscopy of an analytical sample obtained via column chromatography on silica gel (Figure 1).

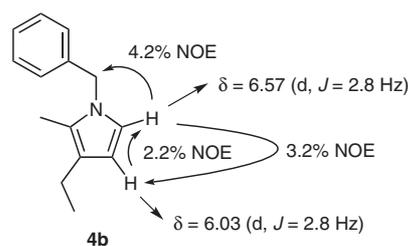


Figure 1 Structural identification of pyrrole **4b** via ^1H NMR and DIFNOE experiments

Table 1 Reductive Cyclization of γ -Chloroketimine **2a** and **2b**

Entry	Substrate	Reduction conditions	dr <i>cis/trans</i> ^a	Conv. to 5 (%) ^{a,b}
1	2a	NaBH ₄ (2 equiv), MeOH, heat, 2 h	50:50	98 (67)
2	2b	NaBH ₄ (2 equiv), MeOH, heat, 2 h	50:50	96 (78) ^c
3	2a	NaBH ₄ (1.2 equiv), MeOH, 0 °C, 2 h	60:40	98
4	2a	NaBH ₄ (1.3 equiv), MeOH, -40 °C to -20 °C, 2 h	52:48	92
5	2a	NaBH ₄ (1.3 equiv), MeOH, -78 °C to -40 °C, 2 h	55:45	86
6	2a	H ₂ (5 atm)/Pd(0) (10 mol%), MeOH, r.t., 3 h	89:11	90 (52)
7	2a	DIBAL (1.3 equiv), THF, -78 °C, 3 h	77:23	64
8	2a	9-BBN (1.3 equiv), THF, 0 °C, 3 h	76:24	76
9	2b	9-BBN (1.3 equiv), THF, 0 °C, 4 h	87:13	90 (48)

^a Determined via GC.

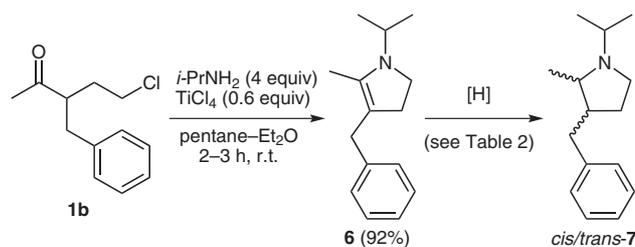
^b Parentheses indicate overall isolated yields for the two-step process from ketone **1a**.

^c Contained pyrrole **4b** (< 4% as determined by GC).

All attempts to improve the diastereoselectivity of the reduction of imine **2a** by lowering the reaction temperature during the reaction with sodium borohydride (Table 1, entries 3–5) or by using alternative reducing reagents, that is, $\text{NaBH}_4 \cdot \text{CeCl}_3 \cdot 7\text{H}_2\text{O}$, NaCNBH_3 , $\text{Zn}(\text{BH}_4)_2$, $\text{Li}(\text{Et})_3\text{BH}$, LiBH_4 , LiAlH_4 , L-Selectride, and $\text{BH}_3 \cdot \text{SMe}_2$, failed, leading to poor diastereomeric ratios (see Supporting Information: Table S1, entries 1–3, 5, 6, 8–10) and/or resulting in lower degrees of conversion into pyrrolidines **5a** (see Supporting Information: Table S1, entries 1, 2, 4, 7, 9–11). However, palladium-catalyzed hydrogenation of γ -chloroketimine **2a** resulted in an improved diastereoselectivity with good conversion affording pyrrolidines *cis*-**5a** and *trans*-**5a** in an 89:11 diastereomeric ratio in an overall yield of 52% (Table 1, entry 6). Alternatively, the use of DIBAL-H or 9-BBN in tetrahydrofuran also afforded good *cis* selectivity with moderate conversion into pyrrolidines **5a** (Table 1, entries 7 and 8). Following the latter method, *N*-benzyl pyrrolidine **5b** as an 87:13 *cis/trans* mixture was isolated in 48% yield (Table 1, entry 9). The identification of the *cis*- and *trans*-diastereomers **5a,b** was made by NMR analysis of diastereomerically pure samples of *cis*-**5a** and *trans*-**5a** obtained via preparative GC and samples of *cis*-**5b** and *trans*-**5b** obtained via column chromatography on silica gel. In the literature, the *cis/trans* configuration of a related 1,2,3-trisubstituted pyrrolidine was assigned based on the ^{13}C NMR data of the 2-methyl group.²³ Based on the γ -gauche effect, the *cis* configuration was assigned to the diastereomer with the resonance of the 2-methyl group shifted upfield by 4 ppm (CDCl_3) relative to the other diastereomer, that is, the *trans*-pyrrolidine. In analogy, for 3-ethyl-1-isopropyl-2-methylpyrrolidine (**5a**), as well as for 1-benzyl-3-ethyl-2-methylpyrrolidine (**5b**), the resonance of the 2-methyl group of the *cis* diastereomer was shifted upfield relative to the *trans* diastereomer and allowed the assignment of *cis*- and *trans*-**5a,b**. The assignment of the *cis* stereochemistry of the *N*-benzylpyrrolidine *cis*-**5b** was further supported by the fact that in this diastereomer C-2 and C-3 were shielded in the ^{13}C NMR spectrum when compared with the corresponding resonances of the *trans*-isomer **5b**.²⁴

Having found the most stereoselective reduction conditions, that is, palladium-catalyzed hydrogenation at room temperature and the most efficient reaction conditions, that is, reduction with NaBH_4 in methanol, the generality of these reductive cyclizations was extended towards the

synthesis of 3-benzylated pyrrolidines. Imination of 3-benzyl-5-chloro-2-pentanone (**1b**) with isopropylamine directly afforded, somewhat surprisingly, the corresponding 2-pyrroline **6** in 92% crude yield in >90% purity as determined by GC (Scheme 2). Reduction of the reaction mixture with sodium borohydride or sodium cyanoborohydride in methanol afforded the expected pyrrolidines *cis*- and *trans*-**7** with poor diastereoselectivity (Table 2, entries 1 and 2). The diastereoselectivity could be improved again by palladium-catalyzed hydrogenation leading to the isolation of pyrrolidines **7** as a 78:22 *cis/trans* mixture in 42% yield after vacuum distillation (Table 2, entry 3).



Scheme 2

With the positive results on the synthesis of 1,2,3-trisubstituted pyrrolidines in hand, the synthesis of valuable 2,3-disubstituted tetrahydrofurans via direct reductive cyclization of γ -chloroketones **1** was investigated. Attempted palladium-catalyzed hydrogenation of γ -chloroketone **1a** failed to give reasonable conversion into the corresponding tetrahydrofurans. When γ -chloroketone **1a** was reduced with sodium borohydride in methanol at 0 °C for three hours no cyclization occurred and the δ -chloroalcohol **8** (dr 67:33) was obtained in 82% yield.

However, when the reduction of ketone **1a** with NaBH_4 was performed under reflux conditions in MeOH, the corresponding cyclized 2,3-disubstituted tetrahydrofurans **9** were obtained in 63% yield as a 70:30 *cis/trans* mixture in >90% purity (Scheme 3). In analogy with the latter reduction conditions, the 3-benzyl-, 3-methyl-, and 3-allyl-5-chloro-2-pentanones **1b–d** were reductively cyclized to diastereomeric mixtures of *cis*- and *trans*-tetrahydrofurans **10–12** in good to excellent yields and purity. The synthesized tetrahydrofurans could be further purified via fractionated distillation. 2,3-Dimethyltetrahydrofuran (**11**) has been identified as a significant volatile compo-

Table 2 Reduction of 2-Pyrroline **6**

Entry	Reduction conditions	dr <i>cis/trans</i> ^a	Conv. to 7 (%) ^{a,b}
1	NaBH_4 (2 equiv), MeOH, heat, 2 h	55:45	96 (66)
2	NaCNBH_3 (2 equiv), MeOH AcOH (1 equiv), heat, 2 h	40:60	94
3	H_2 (5 atm)/Pd(0) (10 mol%), MeOH, r.t., 3 h	78:22	93 (42)

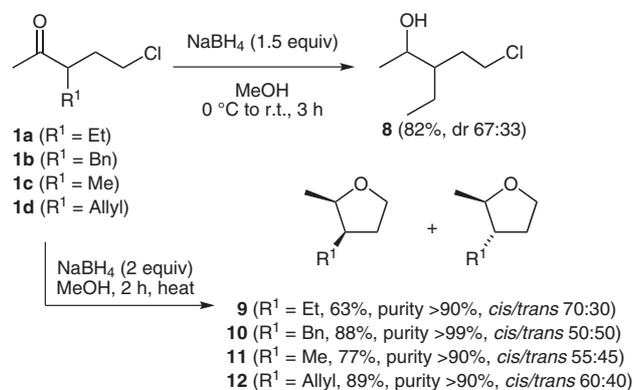
^a Determined via GC.

^b Parentheses indicate overall isolated yields for the two-step process from ketone **1b**.

Table 3 Reduction of γ -Chloroketone **1c**

Entry	Reaction conditions	Result
1	NaBH ₄ (2 equiv), MeOH, 2 h, heat	11 (77%, purity >90%, <i>cis/trans</i> = 55:45)
2	NaCNBH ₃ (2 equiv) AcOH (1 equiv), MeOH, 2 h, heat	11 (66%, purity >90%, <i>cis/trans</i> = 40:60)
3	LiAlH ₄ (2 equiv), Et ₂ O, 2 h, heat	11 (55%, purity >80%, <i>cis/trans</i> = 70:30)

ment of the odor spectrum produced by streptomycetes, which are cultivated for the production of important biologically active compounds, for example, antibiotics.²⁵ The use of alternative reducing reagents had only a moderate influence on the diastereoselectivity of the reductive cyclization of γ -chloroketone **1c** (Table 3), with NaCNBH₃ affording a 40:60 *cis/trans* mixture of tetrahydrofurans **11** (Table 3, entry 2) and LiAlH₄ (Table 3, entry 3) providing a better diastereoselectivity (70:30 *cis/trans*) but lower yield and purity.

**Scheme 3**

The assignment of the *cis* and *trans* configuration of the 2,3-disubstituted tetrahydrofurans **9–12** was based on comparison of spectroscopic data with literature data for the known 2,3-dimethyltetrahydrofurans (*cis*- and *trans*-**11**),²⁶ ¹H NMR NOE experiments, and analysis of ¹³C NMR data of the *cis* and *trans* diastereomers **9–12**, which could be separated via preparative GC.

It is stated in the literature that the carbon atoms of the ring and the α -carbon atoms of the substituents on C-2 and C-3 in *cis*-tetrahydrofurans are shielded in the ¹³C NMR spectrum when compared with the corresponding resonances of the *trans* isomers.²⁷

In summary, a novel unexplored diastereoselective synthetic approach towards the synthesis of 1,2,3-trisubstituted pyrrolidines and 2,3-disubstituted tetrahydrofurans is described. The synthesis of the pyrrolidines involves reductive cyclization of in situ prepared γ -chloroketimines or reduction of the 2-pyrroline formed directly by imination of 3-benzyl-5-chloro-2-pentanone. An extensive study was performed to obtain good diastereoselectivity during the synthetic protocol. In terms of yields, sodium borohydride came out as the best reductant, whereas for

diastereoselectivity, palladium-catalyzed hydrogenation or, alternatively, reduction with 9-BBN proved to be the best choice. Similarly, 2,3-disubstituted tetrahydrofurans were obtained by the direct reduction of γ -chloroketones.

Supporting Information for this article is available online at <http://www.thieme-connect.com/ejournals/toc/synlett>.

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References and Notes

- Postdoctoral Fellow of the Research Foundation-Flanders (FWO).
- Wolfe, J. P. *Eur. J. Org. Chem.* **2007**, 571; and references cited therein.
- Mauger, A. B. *J. Nat. Prod.* **1996**, *59*, 1205.
- Norton Matos, M.; Afonso, C. A. M.; Batey, R. A. *Tetrahedron* **2005**, *61*, 1221.
- Kluender, H. C. E.; Benz, G. H. H.; Britelli, D. R.; Bullock, W. H.; Combs, K. J.; Dixon, B. R.; Schneider, S.; Wood, J. E.; Van Zandt, M. C.; Wolanin, D. J.; Wilhelm, S. M. US 5789434, **1998**; *Chem. Abstr.* **1998**, *129*, 161412h.
- Huang, P. Q.; Wang, S. L.; Ye, J. L.; Ruan, Y. P.; Huang, Y. Q.; Zheng, H.; Gao, J. X. *Tetrahedron* **1998**, *54*, 12547.
- Bermejo, A.; Figadère, B.; Zafra-Polo, M.-C.; Barrachina, I.; Estornell, E.; Cortes, D. *Nat. Prod. Rep.* **2005**, *22*, 269.
- Saleem, M.; Kim, H. J.; Ali, M. S.; Lee, Y. S. *Nat. Prod. Rep.* **2005**, *22*, 696.
- Faul, M. M.; Huff, B. E. *Chem. Rev.* **2000**, *100*, 2407.
- Kang, E. J.; Lee, E. *Chem. Rev.* **2005**, *105*, 4348.
- Jones, A. D.; Knight, D. W.; Hibbs, D. E. *J. Chem. Soc., Perkin Trans. 1* **2001**, 1182.
- (a) Minatti, A.; Muñoz, K. *Chem. Soc. Rev.* **2007**, *36*, 1142. (b) Wolfe, J. P. *Synlett* **2008**, 2913.
- Miura, K.; Hosomi, A. *Synlett* **2003**, 143.
- (a) Husinec, S.; Savic, V. *Tetrahedron: Asymmetry* **2005**, *16*, 2047. (b) Pandey, G.; Banerjee, P.; Gadre, S. R. *Chem. Rev.* **2006**, *106*, 4484.
- (a) Pichon, M.; Figadère, B. *Tetrahedron: Asymmetry* **1996**, *7*, 927. (b) Mitchinson, A.; Nadin, A. *J. Chem. Soc., Perkin Trans. 1* **2000**, 2862.
- Wolfe, J. P.; Hay, M. B. *Tetrahedron* **2007**, *63*, 261.
- For isolated examples of reductive cyclizations of aromatic γ -chloroketones to 2-phenyltetrahydrofurans, see: (a) Dahlenburg, L.; Götz, R. *Eur. J. Inorg. Chem.* **2004**, 888. (b) Rang, H.; Goetz, N.; Harreus, A.; Kast, J. DE 4314656, **1993**; *Chem. Abstr.* **1994**, *120*, 269823.
- For a related example on the stereospecific cyclization of diastereomeric quaternized amino alcohols, prepared by reduction of the corresponding ketones, to 2,3-disubstituted

- tetrahydrofurans, see: Grasshoff, L.; Haller, R. *Arch. Pharm.* **1986**, *319*, 493.
- (19) For selected examples, see: (a) Sulmon, P.; De Kimpe, N.; Schamp, N.; Tinant, B.; Declercq, J. P. *Tetrahedron* **1988**, *44*, 3653. (b) Verniest, G.; Claessens, S.; Bombeke, F.; Van Thienen, T.; De Kimpe, N. *Tetrahedron* **2005**, *61*, 2879. (c) Verniest, G.; Surmont, R.; Van Hende, E.; Deweweire, A.; Deroose, F.; Thuring, J. W.; De Kimpe, N. *J. Org. Chem.* **2008**, *73*, 5458. (d) D'hooghe, M.; De Meulenaer, B.; De Kimpe, N. *Synlett* **2008**, 2437. (e) Denolf, B.; Leemans, E.; De Kimpe, N. *J. Org. Chem.* **2007**, *72*, 3211. (f) Denolf, B.; Mangelinckx, S.; Törnroos, K. W.; De Kimpe, N. *Org. Lett.* **2006**, *8*, 3129. (g) De Kimpe, N.; Stevens, C. *Synthesis* **1993**, 89. (h) Leemans, E.; Mangelinckx, S.; De Kimpe, N. *Chem. Commun.* **2010**, *46*, 3122. (i) De Kimpe, N.; Verhé, R.; De Buyck, L.; Schamp, N. *J. Org. Chem.* **1980**, *45*, 5319. (j) De Kimpe, N.; Sulmon, P.; Verhé, R.; De Buyck, L.; Schamp, N. *J. Org. Chem.* **1983**, *48*, 4320. (k) Sulmon, P.; De Kimpe, N.; Schamp, N. *J. Chem. Soc., Chem. Commun.* **1985**, 715.
- (20) (a) Mangelinckx, S.; Giubellina, N.; De Kimpe, N. *Chem. Rev.* **2004**, *104*, 2353; and references cited therein. (b) Giubellina, N.; Aelterman, W.; De Kimpe, N. *Pure Appl. Chem.* **2003**, *75*, 1433; and references cited therein. (c) Aelterman, W.; De Kimpe, N.; Tyvorskii, V.; Kulinkovich, O. *J. Org. Chem.* **2001**, *66*, 53.
- (21) (a) Brimble, M. A.; Prabakaran, H. *Tetrahedron* **1998**, *54*, 2113. (b) Allen, P. R.; Brimble, M. A.; Prabakaran, H. *J. Chem. Soc., Perkin Trans. 1* **2001**, 379. (c) Johnson, W. S. DT 1948369, **1970**; *Chem. Abstr.* **1970**, *73*, 14222. (d) Roman, B. I.; De Kimpe, N.; Stevens, C. V. *Chem. Rev.* **2010**, *110*, 5914.
- (22) (a) De Kimpe, N.; Verhé, R.; De Buyck, L.; Moens, L.; Schamp, N. *Synthesis* **1982**, 43. (b) De Kimpe, N.; Verhé, R. In *The Chemistry of α -Haloketones, α -Haloaldehydes and α -Haloamines*; Patai, S.; Rappoport, Z., Eds.; Wiley: New York, **1988**. (c) De Kimpe, N.; Schamp, N. *Org. Prep. Proced. Int.* **1979**, *11*, 115.
- (23) Meyers, A. I.; Snyder, L. *J. Org. Chem.* **1992**, *57*, 3814.
- (24) Katritzky, A. R.; Feng, D.; Qi, M.; Aurrecoechea, J. M.; Suero, R.; Aurrekoetxea, N. *J. Org. Chem.* **1999**, *64*, 3335; and references cited therein.
- (25) (a) Řezanka, T.; Prell, A.; Sigler, K. *Folia Microbiol.* **2008**, *53*, 315. (b) Jáchymová, J.; Votruba, J.; Viden, I.; Řezanka, T. *Folia Microbiol.* **2002**, *47*, 37.
- (26) Frauenrath, H.; Philipps, T. *Liebigs Ann. Chem.* **1985**, 1951.
- (27) Eliel, E. L.; Rao, V. S.; Pietrusiewicz, K. M. *Org. Magn. Reson.* **1979**, *12*, 461.

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