

Nickel-Catalyzed Ring-Opening C–O Functionalization of *peri*-Xanthenoxanthenes for 8-Substituted Binaphthol Synthesis

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Cite This: https://doi.org/10.1021/acs.orglett.1c01053





alized binaphthols. 1,2-Bis(dicyclohexylphosphino)ethane was the best ligand for alkylations and ICy for arylation. After mechanistic investigations, we assumed that the reaction proceeds via C–O reduction and subsequent C–O functionalization. To verify the mechanism, the intermediate after reduction was isolated. Moreover, the asymmetric addition, using 8-octylbinaphthol after optical resolution, was studied.

B inaphthyl derivatives are widely applied as chiral ligands and catalysts in many asymmetric transformations.¹ Preferred synthetic strategies for these reagents are modifications of binaphthol (1,1'-bi-2-naphthol) using electrophilic substitutions at 6,6'-positions² and directed metalations at 3,3'positions.³ Although homocoupling reactions of 2-naphthol derivatives are also available for the synthesis of binaphthol with substituents in other positions,⁴ the synthesis of 8,8'-functionalized binaphthols has not been fully developed because of the steric hindrance of the 8-position of binaphthol.⁵

In our laboratory, we applied Cu-catalyzed ring-closing reactions of binaphthol to produce *peri*-xanthenoxanthene (PXX) derivatives (Scheme 1a), which are used as p-type transistors for rollable displays by Kobayashi (Sony Corporation).^{6,7} We think the reductive ring-opening reaction of PXX to introduce substituents could result in a novel synthetic strategy for binaphthol derivatives (Scheme 1b).

Scheme 1. Ring-Closing and Ring-Opening Strategy to Functionalize Binaphthols

Our previous work - Ring-closing reaction of binaphthol for PXX -



Working Hypothesis - Ring-opening reaction of PXX for binaphthol -



To prove this hypothesis, we applied the Ni-catalyzed C–O functionalization, which experienced considerable attention in this past decade.^{8,9} There have been a few reports on C–O bond activation of diaryl ethers.¹⁰ Martin reported the C–O silylation of dibenzofuran with silylborane,^{10a} and Yorimitsu and Osuka disclosed the C–O arylation of dibenzofuran with ArMgBr.^{10b} Tobisu and Chatani developed the C–O bond alkylation of diaryl ethers and dibenzofuran with Grignard reagents.^{10c}

We started our investigation to achieve the Ni-catalyzed ringopening reaction of PXX 1a employing Tobisu's reaction conditions (Table 1). Surprisingly, the reaction of 1a with Ni(cod)₂ (15 mol %), 1,2-bis(dicyclohexylphosphino)ethane (dcype) (15 mol %), and *n*-octylmagnesium iodide (2a, 6 equiv) in toluene for 40 h gave 8-*n*-octylbinaphthol (**3aa**) in 71% yield, accompanied by small amounts of binaphthol 4 (\sim 9%, Table 1, entry 1), not 8,8'-di(*n*-octyl)binaphthol.¹¹ No reaction occurred with the addition of other bidentate cyclohexylphosphine ligands, such as 1,3-bis(dicyclohexylphosphino)propane (dcypp), 1,1'-bis(dicyclohexylphosphino)ferrocene (dcypf), bipyridine, monodentate, and carbene ligands (entries 2-8). The reaction of 1a with PhMgBr 2A, adding $Ni(cod)_2$ and dcype, was unsuccessful, with no recovery of starting materials (Table 1, entry 1). After optimization of the ligands, addition of carbene ligand ICy effectively produced 8-phenylbinaphthol 3aA in 58% yield (entry 6).¹¹ No reactions occurred with 2A using other ligands listed in Table 1 (entries 2-5, 7, and 8). Alternative ligands, nickel precursors, Grignard reagents, and temperature variations were also studied (see Tables S1-S4).

Received: March 27, 2021



O Ia	* or PhMgBr (2A) (6 equiv)	Ni(cod) ₂ (15 mol %) ligand (y mol %) toluene 100 °C, 40 h	R = "Oct (3aa) Ph (3aA)
entry	ligand (y mol %)	3aa (%) ^a	3aA (%) ^a
1	dcype (15)	71 ^{b,c}	0
2	dcypp (15)	0	0
3	$dcypf^d$ (15)	trace	0
4	dppe (15)	0	0
5	PCy ₃ (30)	0	0
6	Icy·HCl (30)	0	58
7	$I(1-Ad) \cdot HBF_4$ (30)) 0	0
8	hipyridine (15)	0	0

^{*a*1}H NMR yield. ^{*b*}Isolated yield. ^{*c*}Binaphthol (R = H) 4 was obtained (9%). ^{*d*}1,1'-Bis(dicyclohexylphosphino)ferrocene.

We applied this reaction further using other Grignard reagents (Scheme 2). Butylmagnesium iodide produced **3ab** in high

Scheme 2. Ni-Catalyzed Alkylative Ring Opening of PXX



 $^{a}\mathrm{EtMgBr}$ (6 equiv) and MgI_2 (2.2 equiv) were added. $^{b_{i}}\mathrm{PrMgI}$ was used.

yields. Ethylmagnesium bromide was, together with MgI₂, the most suitable reagent to obtain **3ac** in low yields (38%).^{10c} Addition of *n*-dodecylmagnesium iodide **2d** and isobutylmagnesium iodide **2e** formed **3ad** and **3ae** also in low yields. Using isopropylmagnesium iodide **2f**, isomerization to *n*-propylnickel occurred to produce 8-*n*-propylbinaphthol in 48% yield. Probing the β -H elimination of alkylnickel was included in the later reaction mechanism study.

The results using aryl Grignard reagents are shown in Scheme 3. The reaction with *o*-tolylmagnesium bromide produced the corresponding product **3aB** in low yields (36%), related to the steric hindrance of the *o*-methyl group. Addition of *m*- and *p*-tolylmagnesium bromide successfully formed **3aC** and **3aD** in high yields. The reaction of *p*-tert-butylphenylmagnesium

Scheme 3. Ni-Catalyzed Arylative Ring Opening of PXX



^{*a*}120 °C.

bromide yielded 84% **3aE**. The presence of electron-donating groups in the Grignard reagent enhanced the reactivity, and **3aF** was also obtained in 84% yield. Electron-withdrawing fluoro groups reduced the reactivity to give **3aG** in 49% yield. When the substituents on PXX were examined, the reaction with 2,8-di(*t*-butyl)PXX and 2,8-di(*p*-*t*-butylphenyl)PXX produced the corresponding products **3bD** and **3cD**, but the phenyl groups at the 5- and 11-positions prevented the ring-opening reaction.

To gain further insight on the mechanism of the transformation, we analyzed the products of the reaction prior to full conversion (Scheme 4). The alkylation with **2a** formed the half-

Scheme 4. Half-Time Reactions to Capture Intermediates



ring-opening product **5** as the main product (56% yield), with **3aa** in 24% yield. Regarding the arylation, trace amounts of **5** were detected with ¹H NMR in the crude products, and the main product **3aA** was isolated in 48% yield. These results proved that the reaction proceeds via reductive ring-opening and alkylative/ arylative binaphthol formation.

Next, we investigated the hydrogen source of the reduction. The β -H elimination of alkylnickel intermediates was observed

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during the reaction with isopropylmagnesium iodide (3f). Therefore, we subjected Grignard reagents with no hydrogen in the β -position, such as methylmagnesium iodide and trime-thylsilylmethylmagnesium iodide, to ring-opening conditions to recover the starting materials (Scheme 5).¹²

Scheme 5. Reaction with Grignard Reagent without β -Hydrides



To identify the hydrogen source, the reaction was quenched with D_2O (Scheme 6). The addition of D_2O to **1a** and **2a** did not



result in incorporation of deuterium (D) into 3aa. Adding D₂O to the mixture of 1a and 2A also did not produce 3aA with D atoms included. Furthermore, the reaction of 1a and 2A performed in toluene- d_8 afforded 3aA without D atoms. The D-incorporated Grignard reagent (2A- d_5) was subjected to the reaction to result in 0% D at the 8-position. Mechanism of the reaction with aryl magnesium bromide did not include β -hydride elimination.

A possible reaction mechanism is shown in Scheme 7. The oxidative addition of a Ni catalyst into the C–O bond, using the Grignard reaction, is followed by the transmetalation with Grignard reagents to produce complex II. Oxidative addition of Ni(0) species was well examined on Ni-catalyzed reduction of diphenyl ether. Oxidative addition proceeded via the η^6 -benzene complex.¹³ Ni(0) complex might be slightly sterically accessible to the benzene structure, including a C_a–O atom to form the η^6 -complex. The inhibition of free rotation of the axial bond caused steric congestion and was strict with the reductive elimination, and β -H elimination is preferred, forming a nickel hydride

Scheme 7. Possible Mechanisms of Ni-Catalyzed Ring Opening with Alkyl Grignard Reagent



complex. Reductive elimination yields half-ring-opening product **5** (cycle I).¹⁴ The oxidative addition of Ni to another C–O bond, followed by transmetalation and reductive elimination, results in the products shown in cycle II. In the case of aryl Grignard reagents, **5** is produced similarly to that with alkyl Grignard reagents, following a cross-coupling sequence to obtain the corresponding products; however, the hydrogen source could not be determined.

Finally, to ensure the suitability of the 8-alkyl substituent for asymmetric reactions, the enantioselective alkylation of carbonyl compounds with a zinc reagent was used as a model reaction (Scheme 8).¹⁵ The enantiomerically pure binaphthol **3aa** was

Scheme 8. Enantioselective Alkylation of 1-Naphthaldehydes



resolved using camphor sulfonyl ester column chromatography (Scheme S2). The reaction of α -naphthaldehyde and diethyl zinc in the presence of Ti(OⁱPr)₄ and (*R*)-**3aa** produced 1-(α -naphthyl)-1-propanol **6** in 74% yield and 98% ee. The control experiment using (*R*)-binaphthol instead of (*R*)-**3aa** yielded the product in 81% yield and 78% ee (ref 15 reports products obtained in 94% yield and 85% ee). These results show that the 8-substituted binaphthol can control asymmetric reactions.

In conclusion, we achieved the Ni-catalyzed C–O bond activation of PXX to afford 8-substituted binaphthol derivatives in high yields. The reaction proceeds through the reductive halfring-opening of PXX. Hydride is derived from β -H elimination of alkyl nickel intermediates. Cross-coupling with alkylation or arylation provides the 8-substituted binaphthol derivatives. The suitability of the products as ligands in asymmetric reactions was examined using (R)-8-*n*-octylbinaphthol. Studies to determine the hydrogen source in the case of arylations and apply the 8substituted binaphthol in other asymmetric reactions are ongoing.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.orglett.1c01053.

Experimental procedures and compound characterization (PDF)

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Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

We are grateful to Prof. Mamoru Tobisu (Osaka University) for his helpful advice. This work was supported by Grant-in-Aid for Young Scientists (B) JP15K17859. We would like to thank Editage (www.editage.com) for English language editing.

REFERENCES

(1) (a) Chen, Y.; Yekta, S.; Yudin, A. Modified BINOL Ligands in Assymmetric Catalysis. *Chem. Rev.* **2003**, *103*, 3155–3211. (b) Kočovský, P.; Vyskočil, Š.; Smrčina, M. Non-Symmetrically Substituted 1,1'-Binaphthyls in Enantioselective Catalysis. *Chem. Rev.* **2003**, *103*, 3213– 3245. (c) Berthod, M.; Mignani, G.; Woodward, G.; Lemaire, M. Modified BINAP: The How and the Why. *Chem. Rev.* **2005**, *105*, 1801–1836 and references therein.

(2) (a) Vondenhof, M.; Mattay, J. Sulfonic Acid Esters Derived from 1,1'-Binaphtharene as New Axially Chiral Photosensitizers. *Tetrahedron Lett.* **1990**, *31*, 985–988. (b) Kamei, T.; Shibaguchi, H.; Sako, M.; Toribatake, K.; Shimada, T. Scandium Triflate-Catalyzed 6,6'-Diiodination of 2,2'-Dimethoxy-1,1'-binaphthyl with 1,3-Diiodo-5,5-dimethylhydantoin. *Tetrahedron Lett.* **2012**, *53*, 3894–3896.

(3) (a) Cox, P. J.; Wang, W.; Snieckus, V. Expedient route to 3- and 3,3'-substituted 1,1'-bi-2-naphthols by directed ortho metalation and Suzuki cross coupling methods. *Tetrahedron Lett.* **1992**, *33*, 2253–2256. (b) Ahmed, I.; Clark, D. A. Rapid Synthesis of 3,3' Bis-Arylated BINOL Derivatives Using a C-H borylation *in Situ* Suzuki-Miyaura Coupling Sequence. *Org. Lett.* **2014**, *16*, 4332–4335. (c) Yang, J.-F.; Wang, R.-H.; Wang, Y.-X.; Yao, W.-W.; Liu, Q.-S.; Ye, M. Ligand-Accelerated Direct C-H Arylation of BINOL: A Rapid One-Step

Synthesis of Racemic 3,3'-Diaryl BINOLs. Angew. Chem., Int. Ed. 2016, 55, 14116–14120.

(4) For example, see: (a) Noji, M.; Nakajima, M.; Koga, K. A New Catalytic System for Aerobic Oxidative Coupling of 2-Naphthol Derivatives by the Use of CuCl-amine Complex: A Practical Synthesis of Binaphthol Derivatives. *Tetrahedron Lett.* **1994**, 35, 7983–7984.
(b) Egami, H.; Katsuki, T. Iron-Catalyzed Asymmetric Aerobic Oxidation: Oxidative Coupling of 2-Naphthols. *J. Am. Chem. Soc.* **2009**, *131*, 6082–6083. (c) Liu, Q.-Z.; Xie, N.-S.; Luo, Z.-B.; Cui, X.; Cun, L.-F.; Gong, L.-Z.; Mi, A.-Q.; Jiang, Y.-Z. Facile Preparation of Optically Pure 7,7'-Disubstituted BINOLs and Their Application in Asymmetric Catalysis. *J. Org. Chem.* **2003**, *68*, 7921–7924. (d) Dewar, M. J.; Nakaya, T. Oxidative Coupling of Phenols. *J. Am. Chem. Soc.* **1968**, *90*, 7134–7135.

(5) 8,8'-Disubstituted binaphthols: (a) Katsuki, N.; Isshiki, S.; Fukatsu, D.; Okamura, J.; Kuramochi, K.; Kawabata, T.; Tsubaki, K. Total Synthesis of Dendrochrysanene through a Frame Rearrangement. *J. Org. Chem.* **2017**, *82*, 11573–11584. (b) Podlesny, E. E.; Carroll, P. J.; Kozlowski, M. C. Selective Oxidation of 8,8'-Hydroxylated Binaphthols to Bis-spironaphthalenones or Binaphtho-*para-* and Binaphtho-*ortho*quinones. *Org. Lett.* **2012**, *14*, 4862–4865. (c) Terrasson, V.; Roy, M.; Moutard, S.; Lafontaine, M.-P.; Pèpe; Fèlix, G.; Gingras, M. Benzylictype couplings provide an important asymmetric entry to functionalized, non-racemic helicenes. *RSC Adv.* **2014**, *4*, 32412–32414. (d) Vesely, V.; Stursa, F. 1-Methyl-7-naphthol. *Collect. Czech. Chem. Commun.* **1933**, *5*, 170–178.

(6) (a) Kamei, T.; Uryu, M.; Shimada, T. Cu-Catalyzed Aerobic Oxidative C-H/C-O Cyclization of 2,2'-Binaphthols: Practical Synthesis of PXX Derivatives. Org. Lett. **2017**, 19, 2714–2717. (b) Kamei, T.; Nishino, S.; Yagi, A.; Segawa, Y.; Shimada, T. Ni-Catalyzed α -Selective C-H Borylation of Naphthalene-Baseed Aromatic Compounds. J. Org. Chem. **2019**, 84, 14354–14359.

(7) Kobayashi, N.; Sasaki, M.; Nomoto, K. Stable *peri*-Xanthenoxanthene Thin-Film Transistors with Efficient Carrier Injection. *Chem. Mater.* **2009**, *21*, 552–556.

(8) For selected reviews, see: (a) Tobisu, M.; Chatani, N. Cross-Couplings Using Aryl Ethers via C–O Bond Activation Enabled by Nickel Catalysts. Acc. Chem. Res. 2015, 48, 1717–1726. (b) Li, B.-J.; Yu, D.-G.; Sun, C.-L.; Shi, Z.-J. Activation of "Inert" Alkenyl/Aryl C–O Bond and Its Application in Cross-Coupling Reactions. Chem. - Eur. J. 2011, 17, 1728–1759. (c) Rosen, B. M.; Quasdorf, K. W.; Wilson, D. A.; Zhang, N.; Resmerita, A.-M.; Garg, N. K.; Percec, V. Nickel-Catalyzed Cross- Couplings Involving Carbon–Oxygen Bonds. Chem. Rev. 2011, 111, 1346–1416. (d) Cornella, J.; Zarate, C.; Martin, R. Metal-catalyzed activation of ethers via C–O bond cleavage: a new strategy for molecular diversity. Chem. Soc. Rev. 2014, 43, 8081–8097. (e) Qiu, Z.; Li, C.-J. Transformations of Less-Activated Phenols and Phenol Derivatives via C–O bond Cleavage. Chem. Rev. 2020, 120, 10454–10515.

(9) Wenkert, E.; Michelotti, E. L.; Swindell, C. S. Nickel-induced conversion of carbon-oxygen into carbon-carbon bonds. One-step transformations of enol ethers into olefins and aryl ethers into biaryls. *J. Am. Chem. Soc.* **1979**, *101*, 2246–2247.

(10) (a) Zarate, C.; Nakajima, M.; Martin, R. A Mild and Ligand Free Ni-Catalyzed Silylation via C–OMe Cleavage. J. Am. Chem. Soc. 2017, 139, 1191–1197. (b) Kurata, Y.; Otsuka, S.; Fukui, N.; Nogi, K.; Yorimitsu, H.; Osuka, A. Aromatic Metamorphosis of Dibenzofurans into Triphenylenes Starting with Nickel-Catalyzed Ring-Opening C–O Arylation. Org. Lett. 2017, 19, 1274–1277. (c) Tobisu, M.; Takahira, T.; Morioka, T.; Chatani, N. Nickel-Catalyzed Alkylative Cross-Coupling of Anisoles with Grignard Reagents via C–O Bond Activation. J. Am. Chem. Soc. 2016, 138, 6711–6714.

(11) The ring-opening product at another C–O bond, 8-hydroxy-1,1'-binaphthyl, was not observed.

(12) The reactions of 1a with MeMgI or TMSCH₂MgI using $Ni(cod)_2/ICy$ ·HCl also resulted in no reaction: Tobisu, M.; Takahira, T.; Chatani, N. Nickel-Catalyzed Cross-Coupling of Anisoles with Alkyl Grignard Reagents via C–O Bond Cleavage. *Org. Lett.* 2015, 17, 4352–4355.

Nickel Complexes of N-Heterocyclic Carbene Ligands. J. Am. Chem. Soc. 2017, 139, 17667–17676. (b) Xu, L.; Chung, L. W.; Wu, Y.-D. Mechanism of Ni-NHC Catalyzed Hydrogenolysis of Aryl Ethers: Roles of the Excess Base. ACS Catal. 2016, 6, 483–493.

(14) Reduction via β -H elimination: Tobisu, M.; Morioka, T.; Ohtsuki, A.; Chatani, N. Nickel-catalyzed reductive cleavage of aryl alkyl ethers to arenes in absence of external reductant. *Chem. Sci.* **2015**, *6*, 3410–3414.

(15) Harada, T.; Kanda, K. Enantioselective Alkylation of Aldehydes Catalyzed by a Highly Active Titanium Complex of 3-Substituted Unsymmetric BINOL. *Org. Lett.* **2006**, *8*, 3817–3819.