

Note

Synthesis, Characterization and Antifungal Evaluation of Novel Thiochromanone Derivatives Containing Indole Skeleton

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Received May 1, 2016; accepted June 20, 2016; advance publication released online July 2, 2016

Invasive fungal disease constitutes a growing health problem and development of novel antifungal drugs with high potency and selectivity against new fungal molecular targets are urgently needed. In order to develop potent antifungal agents, a novel series of 6-alkyl-indolo[3,2-*c*]-2*H*-thiochroman derivatives were synthesized. Microdilution broth method was used to investigate antifungal activity of these compounds. Most of them showed good antifungal activity *in vitro*. Compound 4o showed the best antifungal activity, which (inhibition of *Candida albicans* and *Cryptococcus neoformans*) can be achieved at the concentration of 4 μg/mL. Compounds 4b (inhibition of *Cryptococcus neoformans*), 4j (inhibition of *Cryptococcus neoformans*), 4d (inhibition of *Candida albicans*) and 4h (inhibition of *Candida albicans*) also showed the best antifungal activity at the concentrations of 4 μg/mL. The molecular interactions between 4o and the *N*-myristoyltransferase of *Candida albicans* (PDB ID: 1HYL) were finally investigated through molecular docking. The results indicated that these thiochromanone derivatives containing indole skeleton could serve as promising leads for further optimization as novel antifungal agents.

Key words antifungal activity; antifungal lead compound; thiochromanone; indole; molecular docking

Over the past two decades, the incidence of invasive fungal infections and associated mortality has been increasing dramatically, due to a huge increase in the number of patients undergoing organ transplants, anticancer chemotherapy, or other procedures that cause potential hosts to become immunocompromised.¹⁾ However, effective and low toxic antifungal agents are limited. Clinically, *Aspergillus fumigatus* (mortality rate: 50–90%), *Cryptococcus neoformans* (mortality rate: 20–70%), and *Candida albicans* (mortality rate: 20–40%) have been identified as the most common causes of fungal infections.^{2–4)} For the treatment of these infections, orally active polyenes (e.g., amphotericin B),⁵⁾ fluorinated pyrimidines (e.g., 5-fluorocytosine), azoles (e.g., fluconazole and voriconazole),⁶⁾ and echinocandins (e.g., caspofungin and micafungin),⁷⁾ have been widely used in clinic use⁸⁾ (Fig. 1). However, several factors have limited their practical applications. For example, amphotericin B has significant nephrotoxicity and many other side effects. Severe resistance to azoles is increasingly being reported. Echinocandins cannot be orally administrated and are not active against *Cryptococcus neoformans*.^{9–11)} To sum up, these antifungal agents have achieved limited success in terms of severe resistance, limited efficacy and spectrum, drug related toxicity, non-optimal pharmacokinetics, and other problems.¹²⁾ Hence, it is urgent to elaborate new, highly potent antibiotics with alternative modes of actions. In this regard, the myristoyl coenzyme A (MCoA)/protein *N*-myristoyltransferase (NMT) is an attractive target.¹³⁾

NMT is a ubiquitous enzyme present in eukaryotes such as fungi, protozoa and mammals. It catalyzes the transfer of the 14-carbon saturated fatty acid myristate from myristoyl-CoA to the N-terminal glycine residue of a variety of eukaryotic cellular and viral proteins.¹³⁾ NMT participates in membrane

targeting and the function of many proteins in a variety of signal transduction cascades and other critical cellular functions.¹⁴⁾ Moreover, NMTs are striking in their remarkable diversity of peptide substrates and diversity in the peptide-binding groove has offered potential of developing species-specific inhibitors.¹⁴⁾ As a result, NMT has been identified as a potential chemotherapeutic target enzyme for antifungal agents.¹⁾

Previously, Sheng and colleagues¹²⁾ had prepared tetrahydrocarbazole derivatives (Fig. 1) and tested their antimicrobial activity (minimal inhibitory concentration (MIC) range: 0.0156 to 64 μg/mL). It had broad-spectrum inhibitory activity against a wide range of fungal pathogens. Because the chemical scaffold of this compound differs from that of all reported antifungal agents, it is interesting to investigate its structure–activity relationships (SARs) and discover the novel antifungal lead compound which has a similar skeleton structure to tetrahydrocarbazole derivatives.

Thiochromanone is a versatile reagent that has been extensively utilized in heterocyclic synthesis.¹⁵⁾ It had been reported to possess important biological activities. Hoettecke *et al.*¹⁶⁾ had prepared 2-alkenylthiochroman-4-ones and tested their antimicrobial activity. Moreover, in our previous work,¹⁷⁾ 1-(4-phenylthiazol-2-yl)-1,4-dihydrothiochroman[4,3-*c*]pyrazole was synthesized and showed good inhibition for *Cryptococcus neoformans*. Therefore, the biological activity of thiochromanone is well documented and thiochromanone derivatives with antifungal^{18,19)} and antitumor²⁰⁾ activity have been synthesized. These results promoted us to synthesize novel derivatives substituted cyclohexane of tetrahydrocarbazole with thiochroman.

Continuing our efforts on the discovery of thiochroma-

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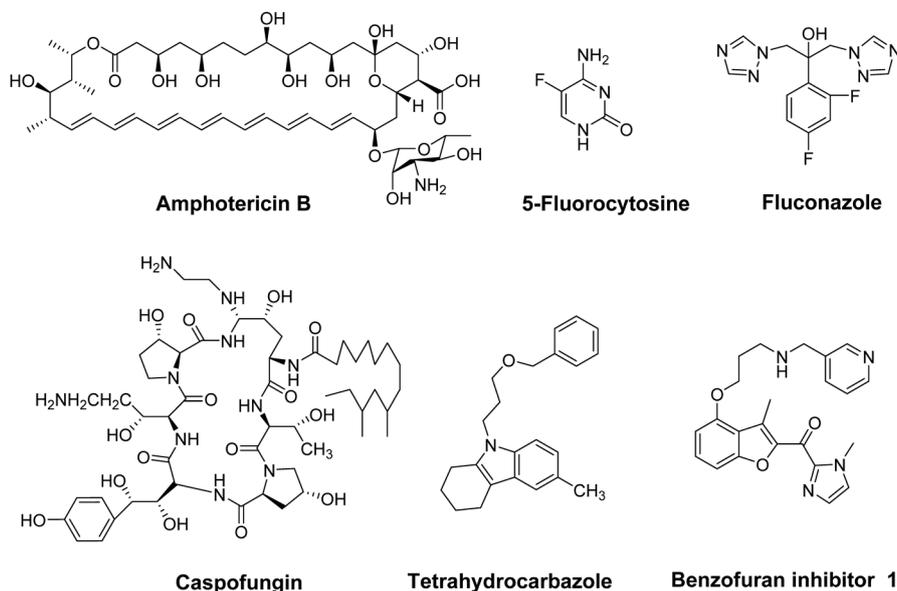
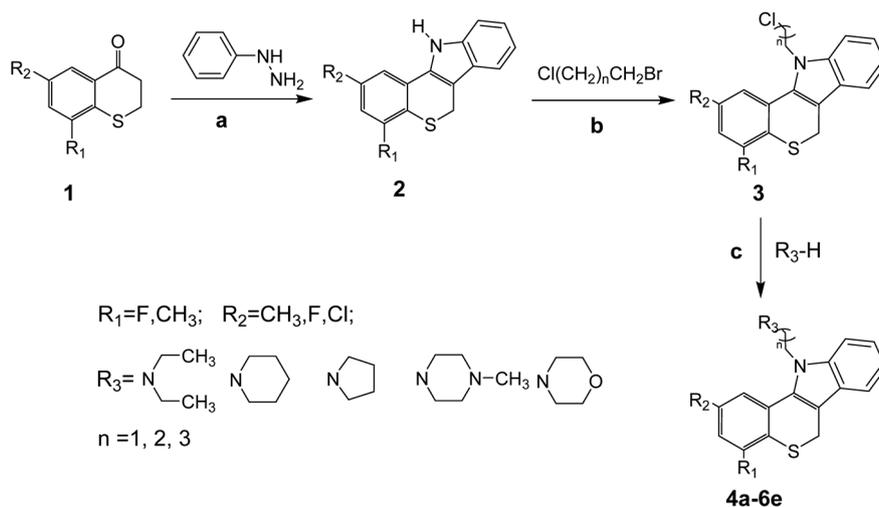


Fig. 1. Chemical Structures of Representative Antifungal Agents



Reagents and conditions: (a) EtOH/reflux/6h/66–75% (b) CH₃CN/NaOH/reflux/6min/70–78% (c) CH₃CN/K₂CO₃/KI/reflux/2h/70–91%.

Chart 1. Reagents and Conditions of the Synthetic Compounds **4a-6e**

none,¹⁹ novel 6-alkyl-indolo[3,2-*c*]-2*H*-thiochroman structures were designed and synthesized. We studied their antifungal activities against *Cryptococcus neoformans* (*C. neo.*), *Candida albicans* (*C. alb.*), *Epidermophyton floccosum* (*E. flo.*), *Mucor racemosus* (*M. rac.*) and *Candida krusei* (*C. kru.*). Crystal structures of *C. alb.* NMT complexed with two classes of inhibitors competitive for peptide substrates have been presented.¹³ One is a nonpeptidic inhibitor having a benzofuran core (Fig. 1). Molecular docking was performed using the X-ray crystallographic structure of *C. alb.* NMT in complex with the benzofuran inhibitor **1** (Fig. 1) to explore the binding mode of the compound at the active site.

Results and Discussion

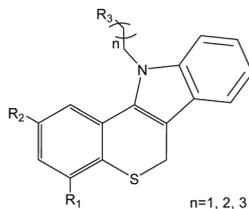
Chemistry By means of Fischer indole synthesis, phenylhydrazine hydrochloride and thiochromanone^{21,22} in an ethanol solution were heated at reflux for 6h to yield compound **2** in 66–75% yield. Compound **2** was then alkylated in acetonitrile solutions under reflux for 6min to obtain compound

3 in 70–78% yield. Compounds **4-6** was then obtained by treating **3** with secondary amine ($R_3\text{-H}$), potassium iodide and anhydrous potassium carbonate in acetonitrile under reflux for about 2h. After completion of the reaction as indicated by TLC, the reaction mixture was purified by column chromatography (Chart 1).

The structures of all compounds are shown in Table 1. All the compounds were elucidated by high resolution mass spectroscopy (HR-MS), ¹H-NMR, and ¹³C-NMR.

Table 2 is shown the reaction conditions exploration of synthetic compound **3**. It is clear that the solvent of acetonitrile and reagent of sodium hydroxide are the best reaction conditions of synthetic compound **3**.

Biological Activity *In vitro* antifungal activity of each compound was expressed as the MIC. As shown in Table 3, it is clear that target compounds showed weak antifungal activity against *E. flo.*, *M. rac.* and *C. kru.*, while there is a good antifungal activity against *C. neo.* and *C. alb.* which are invasive fungus.

Table 1. Structure, Yield, Reaction Time and Melting Point of 6-Alkyl-indolo[3,2-*c*]-2*H*-thiochroman **4a–6e**

No.	R ₁	R ₂	R ₃	<i>n</i>	Time (h)	Yield (%)	mp (°C)
4a	H	F		2	1.5	70	102–104
4b	H	F		2	1.5	90	70–72
4c	H	F		2	1.5	85	104–107
4d	H	F		2	2	85	84–86
4e	H	CH ₃		2	2	88	112–115
4f	H	CH ₃		2	1.5	78	87–88
4g	H	CH ₃		2	2.5	80	133–136
4h	H	CH ₃		2	1	85	115–118
4i	CH ₃	CH ₃		2	1.5	70	132–134
4j	CH ₃	CH ₃		2	1.5	90	110–114
4k	CH ₃	CH ₃		2	2	75	120–124
4l	CH ₃	CH ₃		2	2.5	75	138–142
4m	CH ₃	CH ₃		2	2	70	100–103
4n	H	CH ₃		2	1.5	70	98–103
4o	H	Cl		2	1.5	91	94–96
5a	H	CH ₃		3	2	85	134–135
5b	H	CH ₃		3	1.5	80	122–123
5c	CH ₃	CH ₃		3	2	80	122–124
5d	CH ₃	CH ₃		3	1.5	85	124–126
5e	H	F		3	2	87	82–84
5f	H	F		3	1.5	83	111–113
5g	H	Cl		3	2	77	124–126
5h	H	Cl		3	1.5	81	97–99
6a	H	F		1	2	70	102–104
6b	H	Cl		1	2	75	98–100
6c	H	CH ₃		1	1.5	75	100–102
6d	CH ₃	CH ₃		1	1.5	70	102–104
6e	H	F		1	1.5	70	99–102

Table 2. The Influence of Solvent and Reagent on the Yield of Compound **3**^{a)}

Solvent	Reagent	Yield (%)
C ₂ H ₅ OH	K ₂ CO ₃	3
C ₂ H ₅ OH	(Et) ₃ N	—
C ₂ H ₅ OH	NaOH	11
CH ₃ CN	K ₂ CO ₃	5
CH ₃ CN	(Et) ₃ N	—
CH ₃ CN	NaOH	76
None	NaOH	34

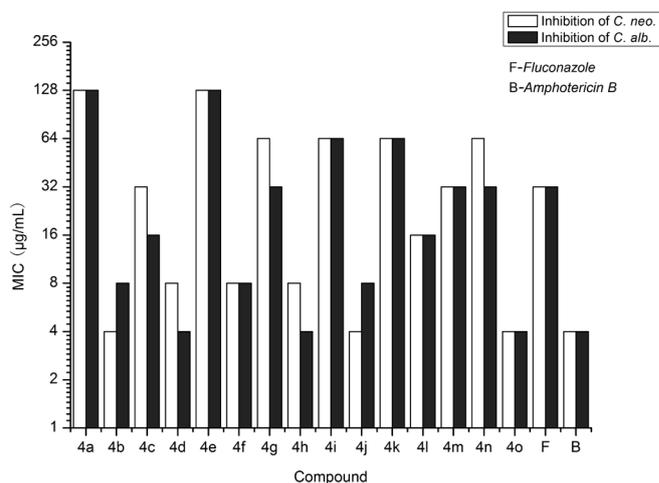
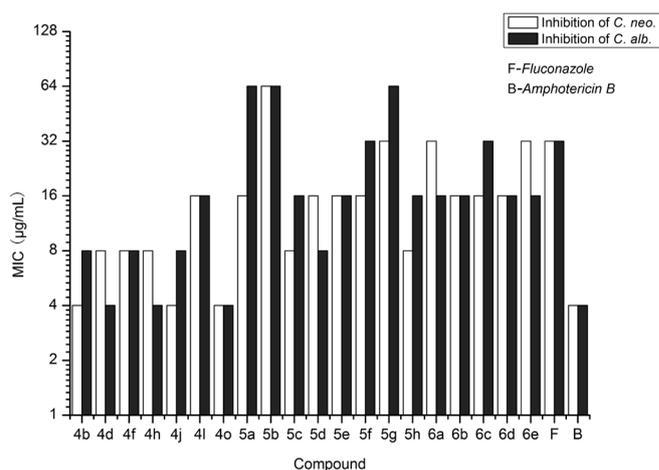
a) All conditions are reflux. "—" represents no reaction.

Table 3. Minimum Inhibitory Concentrations (MIC $\mu\text{g/mL}$) of Target Compounds^{a)}

Compound	MIC ($\mu\text{g/mL}$)				
	<i>C. neo.</i>	<i>C. alb.</i>	<i>E. flo.</i>	<i>M. rac.</i>	<i>C. kru.</i>
4a	128	128	128	>128	>128
4b	4	8	16	64	128
4c	32	16	32	128	>128
4d	8	4	8	64	64
4e	128	128	64	64	64
4f	8	8	16	64	64
4g	64	32	32	>128	128
4h	8	4	8	32	32
4i	64	64	128	128	128
4j	4	8	16	64	64
4k	64	64	64	>128	128
4l	16	16	32	64	>128
4m	32	32	32	64	32
4n	64	32	64	32	64
4o	4	4	16	64	32
5a	16	64	32	128	128
5b	64	64	64	>128	>128
5c	8	16	16	64	64
5d	16	8	16	64	64
5e	16	16	16	128	64
5f	16	32	32	64	128
5g	32	64	64	128	64
5h	8	16	16	32	64
6a	32	16	64	64	64
6b	16	16	64	128	64
6c	16	32	64	128	>128
6d	16	16	32	64	64
6e	32	16	16	64	64
F	32	32	8	64	16
B	4	4	2	4	2

a) F, Fluconazole. B, Amphotericin B. *C. neo.*, *Cryptococcus neoformans*. *C. alb.*, *Candida albicans*. *E. flo.*, *Epidermophyton floccosum*. *M. rac.*, *Mucor racemosus*. *C. kru.*, *Candida krusei*.

We mainly discuss the target compounds antifungal activity against *C. neo.* and *C. alb.*, **4b**, **d**, **h**, **j** and **o** which have better antifungal activity. Among them, **4o** showed the best antifungal activity. Not only is compound **4o** antifungal activity against *C. neo.* and *C. alb.* superior to the positive control Fluconazole but also surpassed Amphotericin B activity. Compounds **4b** and **j** (inhibition of *C. neo.*) can be achieved at 4 $\mu\text{g/mL}$. Compounds **4d** and **h** (inhibition of *C. alb.*) can be achieved at 4 $\mu\text{g/mL}$. In addition, antifungal activity of other

Fig. 2. MIC ($\mu\text{g/mL}$) of Target Compounds **4a–o** against *C. neo.* and *C. alb.*Fig. 3. MIC ($\mu\text{g/mL}$) of Target Compounds That R₃ Is Pyrrolidin or *N*-Methylpiperazine against *C. neo.* and *C. alb.*

compounds also exceeded the level of fluconazole.

As shown in Fig. 2, the target compounds **4a–o**, when R₃ is pyrrolidine, **4b**, **j**, **o** (MIC=4 $\mu\text{g/mL}$) showed better inhibition activity to *C. neo.* When R₃ is *N*-methylpiperazine, the MIC of **4d** and **h** is 8 $\mu\text{g/mL}$. When R₃ is morpholine, piperidine or diethylamine, the compounds showed weak antifungal activity (MIC>16 $\mu\text{g/mL}$).

Compounds **4d** and **h** (R₃ is *N*-methylpiperazine) showed better inhibition activity to *C. alb.* (MIC=4 $\mu\text{g/mL}$). When R₃ is pyrrolidine, the MIC of **4b**, **f** and **j** is 8 $\mu\text{g/mL}$. When R₃ is morpholine, piperidine or diethylamine, the compounds showed weak antifungal activity (MIC>16 $\mu\text{g/mL}$).

Alkane chain of **4a–o** is three carbon chains, **5a–h** is four carbon chains and **6a–e** is two carbon chains. As shown in Fig. 3, while R₃ is pyrrolidin or *N*-methylpiperazine, the three-carbon chain alkane compounds **4a–o** showed the highest antifungal activity.

The inhibition activity to *C. neo.*, when R₃ is pyrrolidine or *N*-methylpiperazine, **4b**, **d**, **h**, **j**, **o** (three carbon chains) >**5a–f** (four carbon chains) or **6a**, **b** (two carbon chains). The inhibition activity to *C. alb.*, when R₃ is *N*-methylpiperazine or pyrrolidine, **4b**, **d**, **f**, **h**, **j** (three carbon chains) >**5a–h** (four

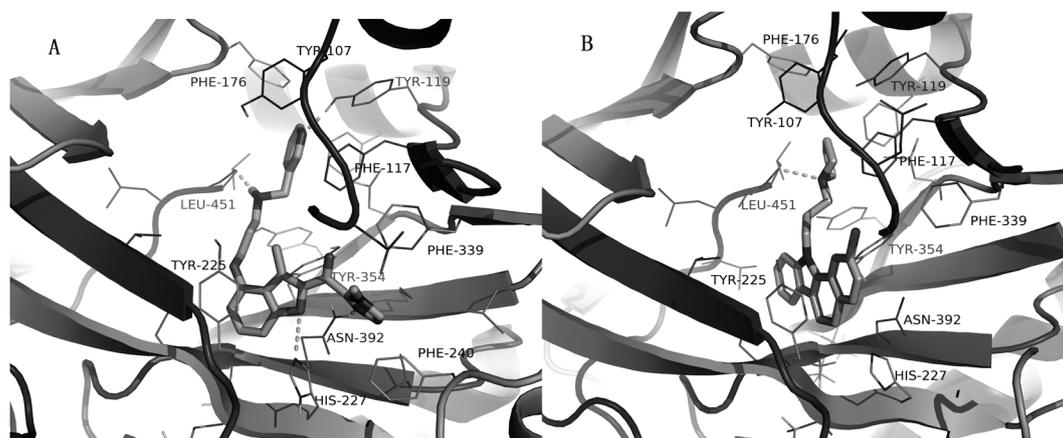


Fig. 4. The Predicted Binding Modes of the Benzofuran Inhibitor **1** /**4o**-*N*-Myristoyltransferase

A: Predicted binding mode of the benzofuran inhibitor **1** docked into the binding site of *N*-myristoyltransferase; B: Predicted binding mode of **4o** docked into the binding site of *N*-myristoyltransferase.

carbon chains) or **6a–e** (two carbon chains).

Molecular Docking In an attempt to correlate the SARs with the binding mode of the synthesized compounds, the benzofuran inhibitor **1** which was reported as NMT inhibitor by Ohtsuka and colleagues²³ and **4o** were docked into the crystal structure of NMT from *C. alb.* (CaNMT).¹³ The docking results of the benzofuran inhibitor **1** and **4o** were shown in Fig. 4.

The overall conformation of the benzofuran inhibitor **1** in the active site of CaNMT was shown in Fig. 4A. The benzofuran ring was located at the center of the active site, surrounded by some hydrophobic residues, such as Tyr225, Tyr354 and Leu394. The phenyl part of the benzofuran ring could form π - π interaction with Tyr225. The substituted phenyl group of C-2 side chain formed hydrophobic interaction with Phe115 and Phe240. The terminal pyridinyl group C-4 side chain was surrounded by hydrophobic aromatic residues, such as Tyr107, Phe117 and Phe176. Four hydrogen bonds were observed between the benzofuran inhibitor **1** and CaNMT. The second amine group of C-4 side chain made a hydrogen bond with C-terminal carboxylate of Leu451, which is an important functional residue in the catalytic cycle of CaNMT. The oxygen atom of the benzofuran ring and the nitrogen atom of the methylimidazole ring formed two hydrogen bonds with His227 and Asn392, respectively. Another hydrogen bond was formed between the pyridine nitrogen of C-4 side chain and Tyr119 hydroxyl group.

The overall conformation of compound **4o** in the active site of CaNMT was shown in Fig. 4B. The nitrogen atom of the pyrrolidine ring formed one hydrogen bond with Leu451. The indolo[3,2-*c*]-2*H*-thiochroman ring was surrounded by some hydrophobic residues, such as Tyr225, Tyr354 and Leu394. The phenyl part of the thiochroman ring could form π - π interaction with Tyr225. The pyrrolidine ring was surrounded by hydrophobic aromatic residues, such as Tyr107, Phe117 and Phe176. However, three hydrogen bonds were lost for compound **4o**, mainly because of the structure of **4o** lacking of a chain interacting with the residues His227 and Asn392. The binding mode of compound **4o** in the active site of CaNMT was very similar to that of the benzofuran inhibitor. As a result, the similar binding mode contributed to the excellent inhibition activity of compound **4o** against *C. alb.*

Conclusion

In summary, a novel series of thiochromanone derivatives containing indole skeleton were synthesized and evaluated as antifungal inhibitors. Compound **4o** (4-chloro-6-(3-pyrrolidin-1-yl-propyl)-indolo[3,2-*c*]-2*H*-thiochroman) showed the best antifungal activity *in vitro*, which (inhibition of *C. neo.* and *C. alb.*) can be achieved at the concentration of 4 μ g/mL. Moreover, molecular docking was performed to position compound **4o** into the *C. alb.* NMT active site to determine the potential binding model. Molecular docking studies rationalized the biological properties of compounds by identifying the possible interactions with the molecular targets. In conclusion, these preliminary results are promising and some of these compounds may be potential candidates for antifungal agents.

Experimental

General chemistry methods, synthesis procedures, spectral data, biological assays, molecular docking are given in Supplementary materials.

Acknowledgments We gratefully acknowledge the Open Project Program of Key Laboratory of Pharmaceutical Quality Control of Hebei Province, for financial support; the Youth Foundation of Science and Technology Research in Higher Education Institutions of Hebei Province (QN2015039) for financial support; Hebei University innovative funding projects for graduate students. We gratefully acknowledge Dr. Carl LeBlond (Chemistry Department, College of Natural Sciences and Mathematics of IUP) for his helping for grammar and spelling of the paper.

Conflict of Interest The authors declare no conflict of interest.

Supplementary Materials The online version of this article contains supplementary materials.

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