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# Synthesis and antibacterial activity of hydroxylated 2-arylbenzothiazole derivatives

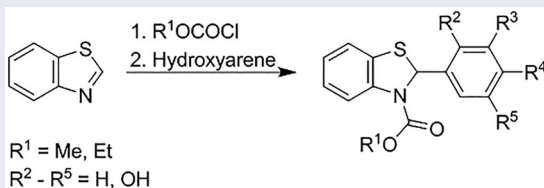
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## ABSTRACT

N-acyliminium reagents formed *in situ* from benzothiazole and alkyl chloroformates react with hydroxyarenes in a Friedel-Crafts manner, providing access to 2-(hydroxyaryl)-benzothiazolines with antibacterial properties.

## GRAPHICAL ABSTRACT



## ARTICLE HISTORY



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## KEYWORDS

2-arylbenzothiazoles;  
phenols; naphthols;  
amidoalkylation;  
antibacterial activity

## Introduction

The 2-arylbenzothiazole moiety is present in many biologically active compounds and medicines.<sup>[1]</sup> Hydroxylated 2-phenylbenzothiazoles in particular are interesting as cytotoxic agents<sup>[2]</sup> and calcium antagonists.<sup>[3]</sup> In recent years their successful application as CN-sensors,<sup>[4]</sup> photosensitizers,<sup>[5]</sup> immunomodulators,<sup>[6]</sup>  $\beta$ -glucuronidase inhibitors<sup>[7]</sup> and DNA binders<sup>[8]</sup> has also been reported. The most common use of 2-hydroxyphenyl (or naphthyl) benzothiazoles is as fluorescent probes<sup>[9–19]</sup> due to their excited-state intramolecular proton transfer (ESIPT) based emission possibility and large Stokes shifts. Because of this wide spectrum of possible applications, 2-substituted benzothiazoles have attracted considerable attention from the synthetic community.<sup>[20]</sup> The mainly employed method for the preparation of 2-arylbenzothiazoles is based on the cyclocondensation of *o*-aminothiophenol with aromatic aldehydes<sup>[21–23]</sup> or carboxylic acids.<sup>[24]</sup> Other approaches involve cyclization reactions of bis-(2-benzalaminophenyl)disulfides,<sup>[25,26]</sup> thiophenolic schiff bases<sup>[27]</sup> or thiobenzanilides.<sup>[28]</sup> This heterocyclic

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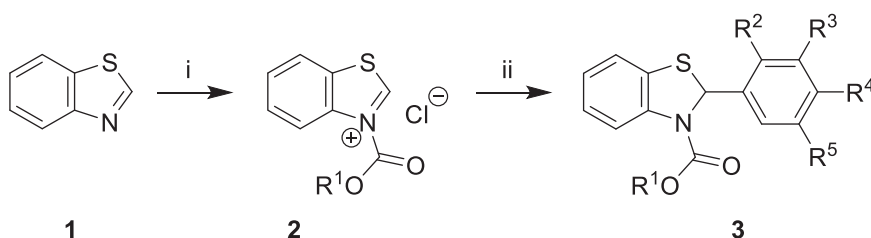
system has also been accessed via reactions of *o*-chloronitrobenzenes or *o*-iodoanilines with various cyclization agents<sup>[29–31]</sup> and arylmethyl thiols with *o*-substituted amines.<sup>[32]</sup> Less popular synthetic routes include annulation of anilines with ethers,<sup>[33]</sup> azolylolation of heteroarenes<sup>[34]</sup> and hydroxylation of heteroaryl halides.<sup>[35]</sup> Methods of direct arylation of benzothiazole<sup>[36,37]</sup> and cross-coupling of heteroaryl thiols with arylzinc reagents<sup>[38]</sup> have also been investigated.

Although the variety of known reactions leading to 2-substituted benzothiazoles is significant, many of them are limited with regard to functional group tolerance and/or availability of reagents and catalysts. Such limitations are particularly relevant to the synthesis of hydroxylated 2-arylbenzothiazoles, and for this reason we chose to investigate an approach for alkylative coupling of the benzothiazole ring to the aromatic rings of various phenols and naphthols. This approach relies on the initial conversion of the thiazole ring to a N-acyliminium species, which is sufficiently electrophilic to further react in a Friedel-Crafts manner with activated aromatic nucleophiles. Combined with a subsequent oxidative rearomatization, this approach has previously been very useful to us in the synthesis 2-heteroaryl substituted thiazoles, such as the natural product *Camalexin* and related analogues.<sup>[39,40]</sup> Additional interest in the hydroxyarene-coupled benzothiazoles is based on analogy with some antibacterial and cytotoxic compounds.<sup>[41,42]</sup>

## Results and discussion

The N-acyliminium reagents **2** were prepared simply by adding methyl or ethyl chloroformate to a solution of benzothiazole in acetonitrile and were used immediately in reactions with activated aromatic substrates. We studied the reactivity of **2** toward five different phenols – phenol, pyrocatechol, pyrogallol, resorcinol and hydroquinone (Scheme 1, Table 1). In the initial experiments all hydroxyarenes were successfully amidoalkylated in the aromatic ring for 1 to 6 h at reflux temperature, but in nearly all cases except for phenol these conditions led to formation of positional isomers and overalkylation. The best yields of mono amidoalkylated products **3** were obtained when the N-acyliminium agent was taken in twofold excess and the reaction mixtures were allowed to stand at 20 °C for two to three days (**3a–d**, **3i–j**) or at 0 °C for 3 to 5 h (**3e–h**). Under these conditions only pyrogallol gave minor amount (10%) of dialkylated products.

The amidoalkylation was also tried with a set of three naphthols (Table 2). Here the reactions proceeded better with no excess of the N-acyliminium reagent. The

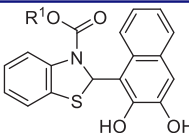
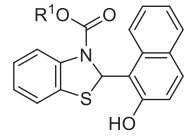
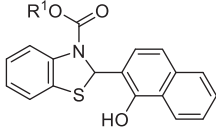


**Scheme 1.** Amidoalkylative coupling of benzothiazole and hydroxyarenes. Reagents and conditions: (i) 1–1.5 equiv.  $R^1OCOCI$  in  $CH_3CN$ ; (ii) Hydroxyarene. See Table 1 for temperature and duration.

**Table 1.** Products obtained from the corresponding phenols according to Scheme 1.

3	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>	R <sup>4</sup>	R <sup>5</sup>	T (°C)	Time (h)	Yield (%)
<b>a</b>	Et	H	H	OH	H	20	80	83
<b>b</b>	Me	H	H	OH	H	20	80	75
<b>c</b>	Et	H	OH	OH	H	20	80	94
<b>d</b>	Me	H	OH	OH	H	20	80	91
<b>e</b>	Et	OH	OH	OH	H	0	5	60
<b>f</b>	Me	OH	OH	OH	H	0	5	68
<b>g</b>	Et	OH	H	OH	H	0	3	52
<b>h</b>	Me	OH	H	OH	H	0	3	65
<b>i</b>	Et	OH	H	H	OH	20	50	75
<b>j</b>	Me	OH	H	H	OH	20	50	55

**Table 2.** Products obtained from the corresponding naphthols and acyliminium reagents **2** in acetonitrile at 20 °C.

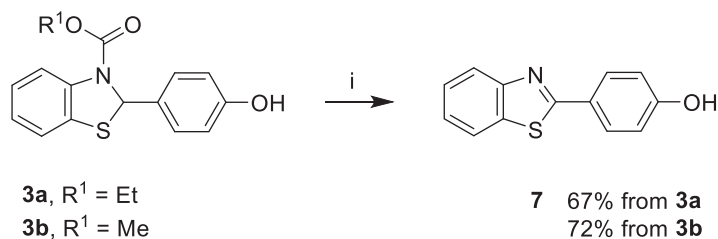
Product		R <sup>1</sup>	Time (h)	Yield (%)
	<b>4a</b>	Me	50	60
	<b>4b</b>	Et	50	40
	<b>5a</b>	Me	80	97
	<b>5b</b>	Et	80	80
	<b>6a</b>	Me	24	72
	<b>6b</b>	Et	24	53

amidoalkylated products **4a**, **b–6a**, **b** were obtained in 40–97% yield at 20 °C, with reaction durations indicated in Table 2.

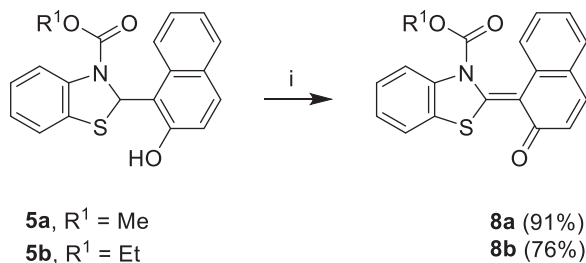
After the successful amidoalkylative coupling of benzothiazole with hydroxyaryl substrates, we experimented with oxidative rearomatization of the benzothiazoline ring in the obtained products **3–6**. In the 2-(hydroxyphenyl) series **3** these experiments led to complex mixtures of oxidized products and failed to provide access to the desired benzothiazole derivatives. An exception was the pair **3a**, **b** – both compounds were cleanly converted to the known<sup>[23,35]</sup> product **7** upon oxidation with *o*-chloranil at 20 °C in acetonitrile (Scheme 2).

In the 2-(hydroxynaphthyl) series **4–6**, under the same conditions, only the oxidation of beta-naphthol derivatives **5a**, **b** proceeded cleanly and with good yields. However, the spectral characteristics of the obtained products **8** indicated that their structure is of quinone methide type rather than aromatic (Scheme 3).

The newly synthesized benzothiazoline derivatives **3–7** were screened for antibacterial activity against two Gram-positive strains (*Staphylococcus aureus* ATCC 6538 P and *Bacillus licheniformis* ATCC 14580) and one Gram-negative strain (*Escherichia coli* ATCC 8739), using the hole plate method in Mueller-Hinton agar with 100 µL loading of 1 mg/mL solutions in DMSO-Water (1:1 vol.). Inhibition of bacterial growth was



**Scheme 2.** Oxidative rearomatization of **3a**, **b** to **7**. Reagents and conditions: (i) o-chloranil, acetonitrile, 20 °C, 2 h.



**Scheme 3.** Oxidation of beta-naphthol derivatives **5a**, **b**. Reagents and conditions: (i) o-chloranil, acetonitrile, 20 °C, 2 h.

**Table 3.** Antibacterial activity of the obtained benzothiazole derivatives **3–7**.

Compound	<i>S. aureus</i> ATCC 6538P		<i>B. licheniformis</i> ATCC 14580	
	Inhibition zone (mm) <sup>a</sup>	MIC (mg/mL) <sup>b</sup>	Inhibition zone (mm) <sup>a</sup>	MIC (mg/mL) <sup>b</sup>
<b>3a</b>	19	0.0063	–	–
<b>3b</b>	13	0.0125	–	–
<b>3c</b>	19	0.0063	17	0.0125
<b>3d</b>	20	0.0063	19	0.0032
<b>3e</b>	17	0.0063	15	0.0063
<b>3f</b>	16	0.0032	14	0.0500
<b>3g</b>	19	0.0063	15	0.0016
<b>3h</b>	16	0.0016	15	0.0250
<b>3i</b>	16	0.0016	16	0.0008
<b>3j</b>	14	0.0063	15	0.0016
<b>4b</b>	15	0.0063	12	0.0125
<b>5a</b>	14	0.0125	–	–
<b>5b</b>	13	0.0032	–	–
<b>6b</b>	13	0.0016	–	–
<b>7</b>	15	0.0063	–	–

<sup>a</sup>Assessed by the hole-plate method with 100 µL loading of 1 mg/mL solutions in 1:1 DMSO-Water.

<sup>b</sup>Ampicillin was used as a positive control with MIC = 0.0004 mg/mL against both strains.

observed only for the Gram-positive strains. Minimal inhibitory concentrations (MIC) were determined for all active compounds with Ampicillin as positive control (Table 3). The selectivity of the tested compounds toward Gram-positive bacteria along with the low MICs observed for some of them (notably the hydroquinone derivative **3i**) provide a good lead for further structure optimization.

A screening was also done against *Candida albicans* and *Saccharomyces cerevisiae*, but only three compounds (**3a**, **3b** and **4b**) showed weak activity against these yeasts.

## Experimental

### General information

All solvents and reagents, such as benzothiazole, phenols, naphthols and alkyl chloroformates were purchased from Sigma-Aldrich or Merck and were used without further purification. Melting points were measured on a Boetius PHMKO5 hot stage apparatus and are uncorrected. IR spectra were measured on VERTEX 70 FT-IR spectrometer (Bruker Optics, Germany).  $^1\text{H}$ -,  $^{13}\text{C}$ -NMR spectra were measured on Bruker Avance AV600 or DRX250 spectrometers and chemical shifts ( $\delta$ , ppm) are downfield from TMS. The NMR spectra of the amidoalkylated hydroxyaryl derivatives, both phenyl and naphthyl, indicated slow rotameric interconversions with broad coalescent signals at 20 °C, which is most likely due to the presence of tertiary carbamate group. Adequate assignment of peaks and structure determination in most cases became possible only when the spectra were measured at 80 °C in DMSO- $d_6$ , as indicated in the text below. TLC was done on precoated 0.2 mm Merck silica gel 60 plates. Silica gel or neutral alumina were used for column chromatography. Mass spectral measurements were performed on a Thermo Scientific Q Exactive hybrid quadrupole-orbitrap mass spectrometer.

### Synthesis of amidoalkylated phenols (3), general procedure

The corresponding alkyl chloroformate (2 mmol) was added dropwise to a magnetically stirred solution of benzothiazole (0.270 g, 2 mmol, 0.22 mL) in acetonitrile (10 mL), followed immediately by the corresponding phenol (1 mmol). The stirring was then continued under the conditions indicated in Table 1. After that, the solvent was evaporated under reduced pressure, the residue was dissolved in dichloromethane (40 mL) and washed twice with water. The organic layer was dried ( $\text{Na}_2\text{SO}_4$ ) and the crude mixture was dry-loaded onto silica gel. The products were isolated by column chromatography on silica gel with mixtures of petrol/diethyl ether as the eluent.

### Synthesis of amidoalkylated naphthols (4, 5 and 6), general procedure

The corresponding alkyl chloroformate (1.2 mmol) was added dropwise to a magnetically stirred solution of benzothiazole (0.135 g, 1 mmol, 0.11 mL) in acetonitrile (7 mL), followed immediately by the corresponding naphthol (1 mmol). The stirring was then continued at 20 °C for the time indicated in Table 2. The solvent was evaporated under reduced pressure and the crude mixture was dry-loaded onto silica gel. The products were isolated by column chromatography on silica gel with mixtures of petrol/diethyl ether as the eluent.

### Oxidation of the amidoalkylated phenols. Synthesis of compounds 7 and 8, general procedure

To the corresponding compound 3 or 5 (0.5 mmol) in acetonitrile (7 mL) was added o-chloranil (0.5 mmol, 0.123 g). The reaction mixture was stirred for 2 h at 20 °C and the solvent was then removed under reduced pressure dry loading the crude mixture onto silica gel. The products were isolated by silica gel column chromatography with petrol – diethyl ether as the eluent.

4-(benzo[d]thiazol-2-yl)phenol 7: The spectral and melting point data for this compound coincided with the literature.<sup>[23,35]</sup>

## Conclusions

We have demonstrated that the N-acyliminium reagents formed *in situ* from benzothiazole and alkyl chloroformates react with hydroxyarenes in a Friedel-Crafts manner, providing access to 2-(hydroxyaryl)-benzothiazoline derivatives. Although the attempts for oxidative rearomatization of the obtained benzothiazolines have been only partly successful, these compounds have shown encouraging antibacterial properties.

Full experimental details, characterization data and copies of <sup>1</sup>H, <sup>13</sup>C-NMR spectra for synthesized compounds can be found via the “Supplementary content” section.

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