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ABSTRACT

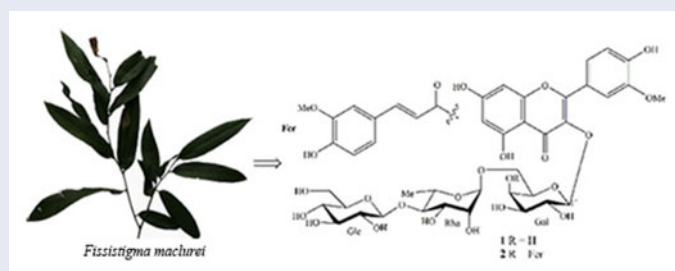
Two new flavonol glycosides, fissimacosides A (1) and B (2) along with two known flavonol glycosides, kaempferol 3-*O*- α -L-rhamnopyranosyl-(1 \rightarrow 6)- β -D-galactopyranoside (3) and kaempferol 3-*O*- β -D-glucopyranosyl-(1 \rightarrow 4)- α -L-rhamnopyranosyl-(1 \rightarrow 6)-[4-(*E*-feruloyl)- β -D-galactopyranoside (4) were isolated from the methanol extract of the leaves of *Fissistigma maclurei* Merr. Their structures were determined on the basis of extensive spectroscopic methods, including 1D-, 2D-NMR, and MS data.

ARTICLE HISTORY

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KEYWORDS

Flavonol; *Fissistigma maclurei*; fissimacoside A; fissimacoside B



1. Introduction

Fissistigma is one of the important genera of Annonaceae family. There are about 80 *Fissistigma* species distributed in Australia and Asia, especially in South-East Asia such as Thailand, Malaysia, Cambodia, and Vietnam. Chemical studies of *Fissistigma* genus indicated the main components to be flavonoids [1], alkaloids [2], and sesquiterpenes [3]. These compounds exhibited antioxidant and anticancer activities. *Fissistigma maclurei* Merr. are climbers that grow up to 6 m tall. There are few reports of pharmacological potential and chemistry of this plant [4,5]. This article reports the isolation and structure elucidation of two new and two known flavonol glycosides from the leaves of *F. maclurei* Merr (Figure 1).

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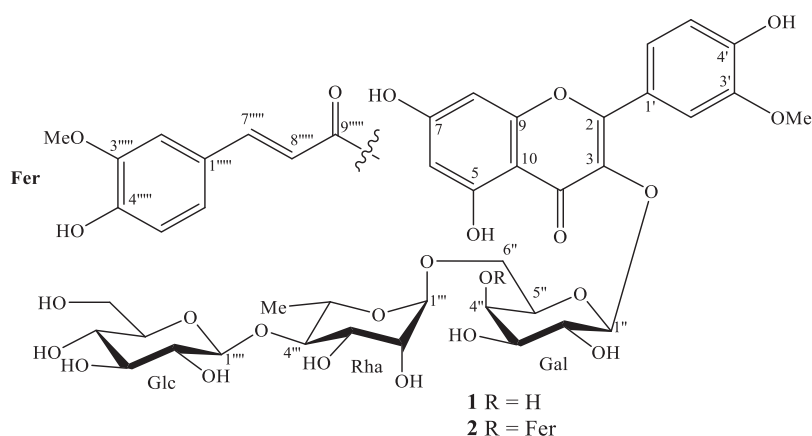


Figure 1. Chemical structures of compounds **1** and **2**.

2. Results and discussion

Compound **1** was obtained as a yellow amorphous powder and its molecular formula was determined to be $C_{34}H_{42}O_{21}$ from HR-ESI-MS ion peak at m/z 809.2110 $[M + Na]^+$. The 1H -NMR spectrum of **1** (in methanol- d_4) showed the specific signals: two aromatic protons of A ring at δ_H 6.19 (1 H, br s) and 6.41 (1 H, br s); three aromatic protons of ABX coupling system of B ring at δ_H 6.89 (1 H, d, $J=8.4$ Hz), 7.59 (1 H, d, $J=8.4$ Hz), and 8.02 (1 H, s); one methoxy group at δ_H 3.94 (3 H, s); three anomeric protons at δ_H 4.48 (1 H, d, $J=7.6$ Hz), 4.50 (1 H, br s), and 5.20 (d, $J=7.8$ Hz) and one secondary methyl group at δ_H 1.20 (3 H, d, $J=6.0$ Hz), suggesting the presence of three sugar moieties. The ^{13}C -NMR and HSQC spectra of **1** revealed the signals of 34 carbons, of which 15 carbons were assigned to a flavonol and 18 carbons to three monosaccharide moieties (Table 1). The HMBC correlations between H-2' (δ_H 8.02)/H-6' (δ_H 7.59) and C-2 (δ_C 158.9)/C-1' (δ_C 122.9)/C-4' (δ_C 150.9); between H-5' (δ_H 6.89) and C-3' (δ_C 148.4); and between the methoxy (δ_H 3.94) and C-3' (δ_C 148.4) suggested that the methoxy and the hydroxy were at C-3' and C-4', respectively (Figure 2). Thus, the flavonol aglycone was identified as isorhamnetin [6]. Acid hydrolysis of **1** revealed D-glucose, D-galactose, and L-rhamnose (identified as trimethylsilyl (TMS) derivatives by a gas chromatography method). In addition, the multiplicity of gal H-1'' [5.20 (d, $J=7.8$ Hz)], rha H-1''' [4.50 (br s)], and glc H-1'''' [4.48 (d, $J=7.6$ Hz)] in the 1H -NMR spectrum of **1** proved configurations of sugar moieties as β -D-galactopyranosyl, β -D-glucopyranosyl, and α -L-rhamnopyranosyl. The HMBC correlations between rha H-1''' (δ_H 4.50) and gal C-6'' (δ_C 68.0); glc H-1'''' (δ_H 4.48) and rha C-4''' (δ_C 83.4); and between gal H-1'' (δ_H 5.20) and C-3 (δ_C 135.5) suggested the sugar linkages as O- β -D-glucopyranosyl-(1 \rightarrow 4)- α -L-rhamnopyranosyl-(1 \rightarrow 6)- β -D-galactopyranosyl and at C-3 of flavonol [7]. Consequently, the new compound **1** was determined to be isorhamnetin 3-O- β -D-glucopyranosyl-(1 \rightarrow 4)- α -L-rhamnopyranosyl-(1 \rightarrow 6)- β -D-galactopyranoside and named as fissmacoside A.

The HR-ESI-MS experiment of **2** resulted to the molecular formula of $C_{44}H_{48}O_{23}$. The 1H -NMR spectrum suggested that compound **2** was a flavonol glycoside with the following signals: two *meta*-protons of A ring at δ_H 6.20 (1 H, s) and 6.42 (1 H, s); an

Table 1. ^1H - and ^{13}C -NMR spectroscopic data for compounds **1** and **2** in methanol- d_4 .

C	1		2	
	δ_{C}	δ_{H} (mult., $J = \text{Hz}$)	δ_{C}	δ_{H} (mult., $J = \text{Hz}$)
Flavonol				
2	158.9	–	159.0	–
3	135.5	–	135.4	–
4	179.5	–	179.3	–
5	163.0	–	163.0	–
6	100.0	6.19 (s)	100.0	6.20 (s)
7	166.0	–	166.0	–
8	94.9	6.41 (s)	95.0	6.42 (s)
9	158.5	–	158.5	–
10	105.7	–	105.7	–
1'	122.9	–	123.0	–
2'	114.6	8.02 (s)	114.9	8.05 (s)
3'	148.4	–	148.3	–
4'	150.9	–	151.1	–
5'	116.0	6.89 (d, 8.4)	116.0	6.91 (d, 8.4)
6'	123.8	7.59 (d, 8.4)	123.9	7.63 (d, 8.4)
3'-OMe	57.0	3.94 (s)	57.0	3.96 (s)
3-OGal				
1''	104.9	5.20 (d, 7.8)	105.0	5.29 (d, 6.0)
2''	73.0	3.78–3.82 (m)	73.5	3.81–3.85 (m)
3''	75.0	3.53–3.55 (m)	73.5	3.83–3.85 (m)
4''	70.1	3.75 (br d, 2.8)	71.4	5.38 (br s)
5''	75.5	3.63–3.66 (m)	73.9	3.88–3.90 (m)
6''	68.0	3.47–3.49 (m)	67.1	3.21–3.25 (m)
		3.69–3.71 (m)		3.47–3.50 (m)
6''-ORha				
1'''	102.0	4.50 (br s)	102.3	4.46 (br s)
2'''	72.0	3.58 (br s)	72.0	3.62 (br s)
3'''	72.3	3.67 (dd, 2.0, 8.0)	72.0	3.70 (dd, 2.0, 8.0)
4'''	83.4	3.45–3.49 (m)	82.8	3.42–3.46 (m)
5'''	68.4	3.55–3.59 (m)	68.2	3.41–3.45 (m)
6'''	18.2	1.20 (d, 6.0)	18.2	1.09 (d, 5.0)
4'''-OGlc				
1''''	105.7	4.48 (d, 7.6)	105.2	4.54 (d, 7.6)
2''''	76.0	3.14–3.18 (m)	75.9	3.19–3.23 (m)
3''''	78.3	3.30–3.32 (m)	78.2	3.33–3.37 (m)
4''''	71.6	3.24–3.28 (m)	71.7	3.22–3.26 (m)
5''''	78.0	3.19–3.23 (m)	77.9	3.20–3.24 (m)
6''''	62.8	3.62 (dd, 4.8, 12.0)	62.9	3.59 (dd, 4.0, 12.0)
		3.81 (br d, 12.0)		3.79 (br d, 12.0)
4''-OFer				
1'''''			127.7	–
2'''''			111.6	7.20 (s)
3'''''			149.4	–
4'''''			150.8	–
5'''''			116.5	6.83 (d, 8.0)
6'''''			124.5	7.10 (d, 8.0)
7'''''			147.8	7.64 (d, 16.0)
8'''''			115.1	6.38 (d, 16.0)
9'''''			168.8	–
OMe			56.3	3.93 (s)

Assignments were done by HSQC, HMBC, and COSY experiments; Glc, glucopyranosyl; Rha, rhamnopyranosyl; Gal, galactopyranosyl; Fer, feruloyl.

ABX coupling system of B ring at δ_{H} 6.91 (1 H, d, $J=8.4$ Hz), 7.63 (1 H, d, $J=8.4$ Hz), and 8.05 (1 H, s); one methoxy group at δ_{H} 3.96 (3 H, s); three anomeric protons at δ_{H} 4.46 (1 H, br s), 4.54 (1 H, d, $J=7.6$ Hz), and 5.29 (1 H, d, $J=6.0$ Hz); the proton signals of (*E*)-feruloyl moiety at δ_{H} 7.20 (1 H, s), 6.83 (1 H, d, $J=8.0$ Hz),

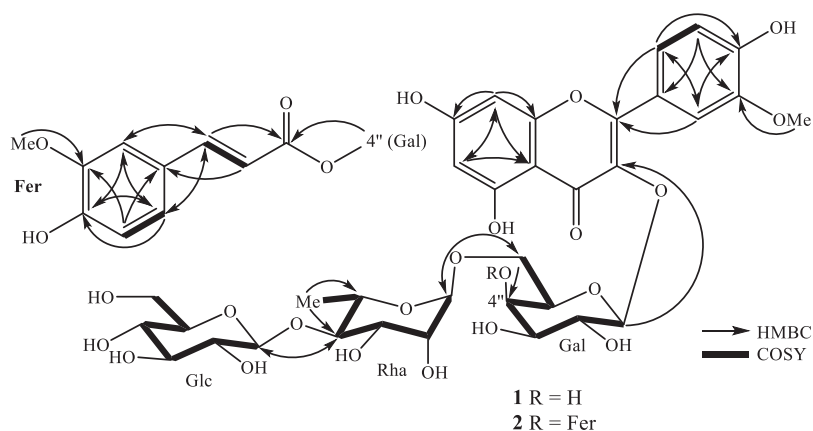


Figure 2. The key HMBC and COSY correlations of compounds **1** and **2**.

7.10 (1 H, d, $J = 8.0$ Hz), 6.38 (1 H, d, $J = 16.0$ Hz), and 7.64 (1 H, d, $J = 16.0$ Hz), and one methoxy group at δ_{H} 3.93 (s). The ^{13}C -NMR and HSQC spectra of **2** showed the signals of 44 carbons, including 2 carbonyl, 12 non-protonated carbons, 25 methines, 2 methylenes, and 3 methyl carbons (Table 1). The ^1H - and ^{13}C -NMR data of **2** showed almost the same to those of fismacoside A (**1**) except for the addition of feruloyl moiety at gal C-4. Addition of feruloyl(2) proved the presence of D-galactose, L-rhamnose, and D-glucose (identified as TMS derivatives by a gas chromatography method). The HMBC correlations between rha H-1''' (δ_{H} 4.46) and gal C-6'' (δ_{C} 67.1); glc H-1'''' (δ_{H} 4.54) and rha C-4''' (δ_{C} 82.8); and between gal H-4'' (δ_{H} 5.38) and fer C-98 (δ_{C} 168.8) confirmed the sugar linkages as *O*- β -D-glucopyranosyl-(1 \rightarrow 4)- α -L-rhamnopyranosyl-(1 \rightarrow 6)- β -D-galactopyranosyl and feruloyl group at gal C-4'. Finally, the HMBC correlation from gal H-1'' (δ_{H} 5.29) to C-3 (δ_{C} 135.4) indicated the trisaccharide position at C-3 of flavonol. Based on the above evidence, compound **2** was determined to be isorhamnetin 3-*O*- β -D-glucopyranosyl-(1 \rightarrow 4)- α -L-rhamnopyranosyl-(1 \rightarrow 6)-[4-(*E*)-feruloyl]- β -D-galactopyranoside, a new compound and named as fismacoside B.

The known compounds were elucidated as kaempferol 3-*O*- α -L-rhamnopyranosyl-(1 \rightarrow 6)- β -D-galactopyranoside (**3**) [8] and kaempferol 3-*O*- β -D-glucopyranosyl-(1 \rightarrow 4)- α -L-rhamnopyranosyl-(1 \rightarrow 6)-[4-(*E*)-feruloyl]- β -D-galactopyranoside (**4**) [7] by comparing their observed and reported physical data.

All isolated compounds were evaluated for cytotoxic activity against three human cancer cell lines, A-549 (human lung cancer), MCF-7 (human breast cancer), and MKN-45 (human gastric cancer) at 10 μM . However, none of them showed cytotoxic activity (cell viability >50%).

3. Experimental

3.1. General experimental procedures

Optical rotations were determined on a JASCO P-2000 digital polarimeter (JASCO Ltd., Tokyo, Japan). UV spectra were obtained from Labtron LUS-B14 spectrophotometer (Labtron Equipment Ltd., Camberley, UK). The IR spectra were obtained from a Bruker TENSOR 37 FT-IR spectrometer (Bruker, Ettlingen, Germany). NMR spectra were

recorded on an Agilent 400 MHz (Agilent Technologies, Santa Clara, CA, USA). Data processing was carried out with the MestReNova ver. 9.0.1. HR-ESI-MS were recorded on an AGILENT 6550 iFunnel Q-TOF LC/MS system (Agilent Technologies). HPLC was carried out using an AGILENT 1200 HPLC system (Agilent Technologies). Column chromatography was performed on silica gel (Kieselgel 60, 230–400 mesh, Merck, Darmstadt, Germany) or RP-18 resins (30–50 μm , Fuji Silysia Chemical Ltd., Tokyo, Japan). For thin layer chromatography (TLC), pre-coated silica-gel 60 F₂₅₄ (0.25 mm, Merck) and RP-18 F₂₅₄S (0.25 mm, Merck) plates were used.

3.2. Plant material

The leaves of *Fissistigma maclurei* Merr. were collected at Curut, Daknong, Vietnam in April 2018, and identified by Dr. Nguyen, The Cuong, Institute of Ecology and Biological Resources, VAST. A voucher specimen (HVQY1804) was deposited at the Military Medical University.

3.3. Extraction and isolation

The dried powdered leaves of *F. maclurei* (2.0 kg) were sonicated with hot methanol two times (each 5 L, 2 h) and removed the solvent *in vacuo* to yield methanol (MeOH) extract (120.0 g). The MeOH extract was suspended in H₂O (1.0 L) and successively partitioned with chloroform, and ethyl acetate (EtOAc) to yield chloroform (FM1, 18.0 g), EtOAc (FM2, 8.0 g) residues, and water layer (FM3).

The water-soluble fraction (FM3) was chromatographed on a Diaion HP-20P column eluting with water then increasing the concentration of MeOH in water (25, 50, 75, and 100%) to obtain four sub-fractions, FM3A (5.0 g), FM3B (8.0 g), FM3C (16.0 g), and FM3D (7.0 g). The FM3C fraction was chromatographed on a silica gel column eluting with gradient solvents of CHCl₃–MeOH (10: 1 M3C fraction was chromatographed on a silica, FM3C1–FM3C3. The FM3C2 fraction was chromatographed on a Sephadex LH-20 column eluting with MeOH to give two fractions, FM3C2A–FM3C2B. The fraction FM3C2A was chromatographed on HPLC using J'sphere ODS-H80 (150 mm \times 20 mm ID) with solvent condition of 35% acetonitrile in water, the flow rate of 3 ml/min, and the wavelength of 254 nm to yield compound **2** (5.0 mg, t_R 38.8 min). The FM3C3 fraction was chromatographed on a RP-18 column using MeOH–water (1: 1, v/v) as eluent solvent to give three fractions, FM3C3A–FM3C3C. The FM3C3B was chromatographed on HPLC using J'sphere ODS-H80 (150 mm \times 20 mm ID) with solvent condition of 25% acetonitrile in water, the flow rate of 3 ml/min, and the wavelength of 254 nm to yield compounds **1** (5.0 mg, t_R 44.2 min) and **3** (11.0 mg, t_R 51.7 min). Compound **4** (5.0 mg, t_R 32.3 min) was obtained from FM3C3C using J'sphere ODS-H80 column (150 mm \times 20 mm ID) on Agilent HPLC 1290 with solvent condition of 30% acetonitrile in water, the flow rate of 3 ml/min, and the wavelength of 254 nm

3.3.1. Fissinoside A (1)

Yellow amorphous powder. $[\alpha]_D^{25}$ -67.0 (c 0.1 MeOH). UV (MeOH) λ_{max} : 264, 370 nm. IR (KBr) ν_{max} : 3390, 1658 cm^{-1} . ¹H- and ¹³C-NMR spectral data (methanol-

d_4): see Table 1. HR-ESI-MS m/z : 809.2110 $[M + Na]^+$ (calcd for $C_{34}H_{42}NaO_{21}$, 809.2111).

3.3.2. Fissmacoside B (2)

Yellow amorphous powder. $[\alpha]_D^{25}$ -86.0 (c 0.1 MeOH). UV (MeOH) λ_{max} : 265, 370 nm. IR (KBr) ν_{max} : 3410, 1675, 1658 cm^{-1} . 1H - and ^{13}C -NMR spectral data (methanol- d_4): see Table 1. HR-ESI-MS m/z : 985.2606 $[M + Na]^+$ (calcd for $C_{44}H_{50}NaO_{24}$, 985.2584).

3.4. Acid hydrolysis

Each compound (1 and 2, 2.0 mg) was separately dissolved in 1.0 N HCl (dioxane- H_2O , 1:1, v/v, 1.0 ml) and heated to 80 °C in a water bath for 3 h. The acidic solution was neutralized with silver carbonate; the solvent was thoroughly removed under a nitrogen stream overnight. After extraction with $CHCl_3$, the water layer was concentrated to dryness using N_2 . The residue was dissolved in pyridine (0.1 ml) and then added L-cysteine methyl ester hydrochloride in pyridine (0.06 M, 0.1 ml). The reaction mixture was heated at 60 °C for 2 h and then added trimethylsilylimidazole solution (0.1 ml), heating at 60 °C for 1.5 h. The dried product was partitioned with *n*-hexane and H_2O (0.1 ml each), and the organic layer was analyzed by gas chromatography: column DB-5 (0.32 mm ID \times 30 m length), detector FID, column temperature 210 °C, injector temperature 270 °C, detector temperature 300 °C, carrier gas He (2 ml/min). Under these conditions, the standard sugars gave peaks at t_R (min) 14.11 and 14.26 for D- and L-glucose, 18.71 and 22.92 for D- and L-galactose, and 4.50 for L-rhamnose, respectively. Peaks at t_R (min) 14.11, 18.71, and 4.50 of D-glucose, D-galactose, and L-rhamnose for all compounds were observed.

3.5. Cytotoxic assays

See Reference [9].

Disclosure statement

No potential conflict of interest was reported by the authors.

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