

Stereoselective tris-glycosylation to introduce β -(1 \rightarrow 3)-branches into gentiotetraose for the concise synthesis of phytoalexin-elicitor heptagluco-side†

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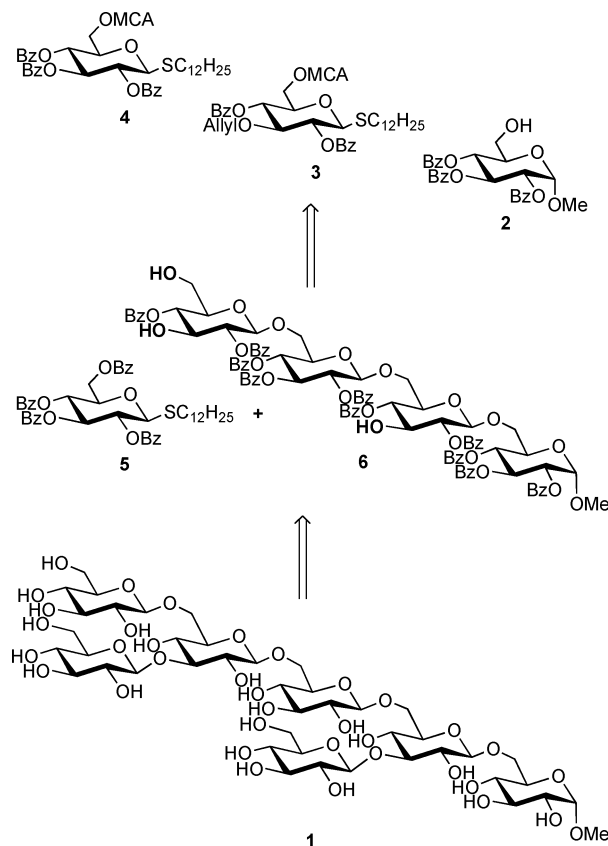
Dodecyl thioglycosides (**3**, **4**, **5**) were prepared by conventional transformation of D-glucose and used as new glycosyl donors for a short-step synthesis of phytoalexin elicitor heptagluco-side. A gentio-tetraoside derivative (**6**) having three hydroxyl groups was synthesized by NIS–TfOH promoted glycosylation in more than 90% yield followed by selective removal of temporary protective groups. Undesired formation of α -glycosides at the introduction of β -(1 \rightarrow 3)-branches into gentio-oligosaccharides was found to be suppressed by use of a thiophilic reagent system, BSP (1-benzenesulfinyl piperidine)–Tf₂O, giving the heptagluco-side in only four glycosylation steps.

Introduction

Since Albersheim *et al.* identified a heptagluco-side analogous to **1** that elicits production of phytoalexin in soybeans,¹ its unique structure has often been used as a model compound to demonstrate the feasibility of new methodologies in chemical synthesis of oligosaccharides, *e.g.*, convergent block synthesis, one-pot sequential glycosylation, and chemoselective glycosylation.² Compound **1** consists of a β -(1 \rightarrow 6)-linked D-glucopyranose backbone (gentiopentaose unit) and two β -(1 \rightarrow 3)-linked branches (laminaribiose units). Because of the difficulty of glycosylating the sterically hindered 3-hydroxyl group, the most successful syntheses of **1** reported to date involved construction of the laminaribiose units at an earlier stage of the synthetic scheme. In contrast to this laminaribiose route, van Boom *et al.* proposed a seemingly straightforward strategy, in which the β -(1 \rightarrow 6)-linked backbone of **1** would be synthesized first and two β -(1 \rightarrow 3)-linked branches would be introduced subsequently.³ However, no one has reported the total synthesis of **1** *via* this gentio-oligosaccharide route. Recently, while investigating the regioselective introduction of a β -(1 \rightarrow 3)-linked branch into gentio-oligosaccharides using our newly-developed dodecyl thioglycoside donors,⁴ we found that BSP–Tf₂O⁵ was a suitable promoter for our purpose. In this paper, we report the first successful results using this approach, which involves tris-glycosylation of a gentiotetraoside intermediate.

Results and discussion

Following the retrosynthetic analysis of the target heptagluco-side **1** proposed by van Boom, elimination of three glucose moieties at the non-reducing end and branches leaves a β -(1 \rightarrow 6)-linked gentiotetraoside **6** as shown in Scheme 1.

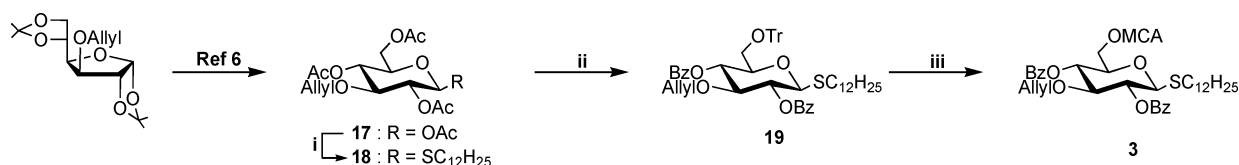


Scheme 1 Retrosynthetic analysis of the heptagluco-side **1** by tris-glycosylation.

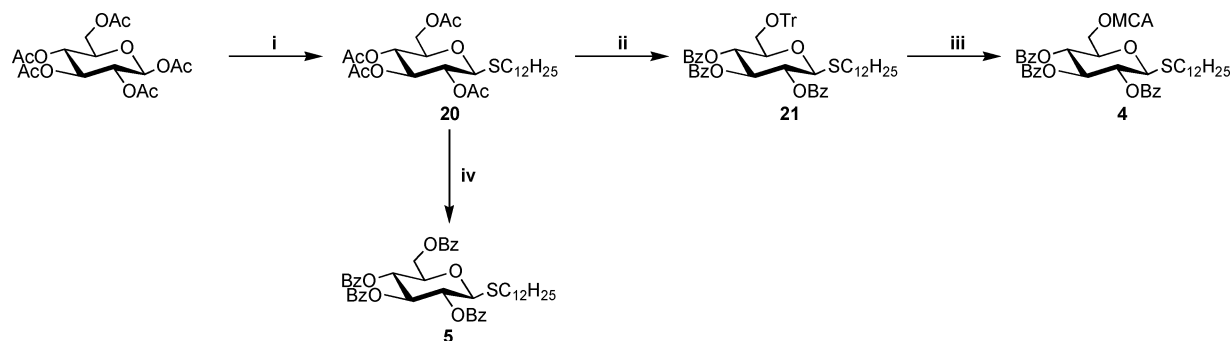
Provision for further tris-glycosylation sites in **6** is anticipated through the use of temporary protecting groups, *O*-chloroacetyl (MCA) and *O*-allyl groups, allowing for exposure of the free hydroxylic acceptor sites. Furthermore, participation of the neighboring 2-*O*-benzoyl protection groups in the dodecyl thioglycosyl donors would afford the desired β -glucosidic linkages. As a result, the total synthesis can be initiated from a known acceptor **2** and three dodecyl thioglycosyl donors (**3**, **4**, **5**). Starting from readily

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† Electronic supplementary information (ESI) available: ¹H, ¹H-¹H COSY and ¹³C NMR spectra for compounds **3**, **4**, **5**, **6**, **7**, **8**, **13**, **14**, and ¹H NMR spectrum for compound **1**. See DOI: 10.1039/b800809d



Scheme 2 Synthesis of glycosyl donor **3**. *Reagents and conditions:* (i) $\text{C}_{12}\text{H}_{25}\text{SH}$, $\text{BF}_3 \cdot \text{Et}_2\text{O}$, $(\text{CH}_2)_2\text{Cl}_2$, 0°C , 87%; (ii) (a) NaOMe , MeOH , 3 h; (b) TrCl , pyridine, DMAP, 60°C , 5 h; (c) BzCl , pyridine, 60°C , overnight, 91% over 3 steps; (iii) (a) $\text{AcOH-H}_2\text{O} = 5:1$, 70°C , 1 h; (b) $(\text{MCA})_2\text{O}$, pyridine- $\text{CH}_2\text{Cl}_2 = 1:10$, 3 h, 92%, over 2 steps. DMAP = *N,N*-dimethylaminopyridine, MCA = chloroacetyl.

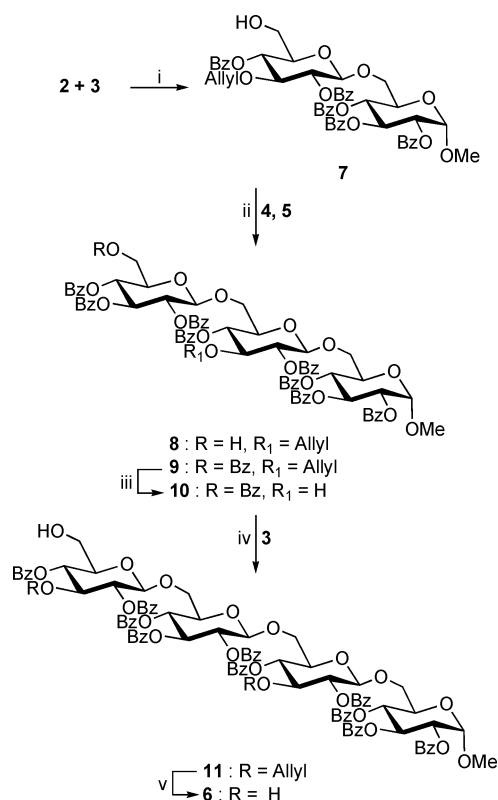


Scheme 3 Synthesis of glycosyl donor **4** and **5**. *Reagents and conditions:* (i) $\text{C}_{12}\text{H}_{25}\text{SH}$, $\text{BF}_3 \cdot \text{Et}_2\text{O}$, $(\text{CH}_2)_2\text{Cl}_2$, 94%; (ii) (a) NaOMe , MeOH , 3 h; (b) TrCl , pyridine, 60°C , overnight; (c) BzCl , pyridine, 0°C to rt, overnight, 86% over 3 steps; (iii) (a) $\text{AcOH-H}_2\text{O} = 5:1$, 70°C , 1 h; (b) MCA, pyridine- $\text{CH}_2\text{Cl}_2 = 1:10$, 87% over 2 steps; (iv) (a) NaOMe , MeOH , 5 h; (b) BzCl , pyridine, rt, overnight, 86% over 2 steps.

available D-glucose derivatives and non-volatile l-dodecanethiol, the glycosyl donors (**3**, **4**, **5**) were prepared by means of conventional transformations as shown in Schemes 2 and 3. Using the thioglycosides **3** and **4** as glycosyl donors, and *N*-iodosuccinimide (NIS)-triflic acid (TfOH),⁷ as the promoter, we proceeded with sequential glycosylation to synthesize the gentiotetraoside **6**, as outlined in Scheme 4.

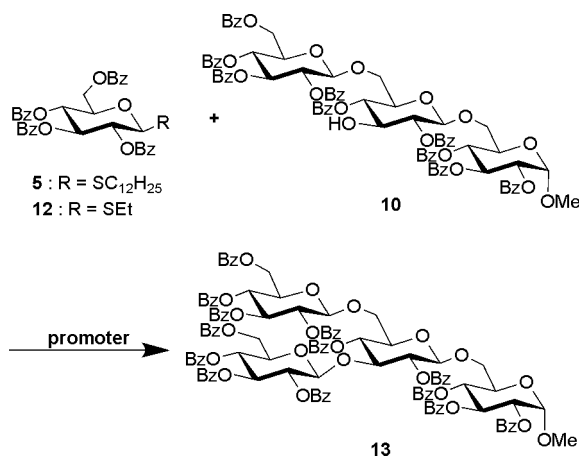
Coupling **2** and the dodecylthio donor **3**, which has a MCA and an *O*-allyl protecting group, was first performed in dry CH_2Cl_2 using NIS- TfOH as a promoter, and the temporary *O*-MCA protecting group was subsequently removed by treatment with aqueous pyridine, giving the desired disaccharide acceptor **7** in 92% overall yield. Moreover, two gentiotrioses **8** and **9** were synthesized under similar conditions with the *O*-MCA protected **4**, and the per-benzoylated donors **5** in high yields, respectively. The acceptor **8** was further glycosylated with **3** to give the tetraose **11** which has two *O*-allyl groups. Finally, selective removal of the *O*-allyl groups in **9** and **11** was successfully performed by ultrasonication with PdCl_2 ,⁸ giving the acceptors **10**³ and **6** in 90% yield. The triol **6**, obtained in an excellent overall yield, was a potential intermediate in our synthetic route to **1**. Compound **10** was used as a model acceptor for an investigation into introducing a β -(1 \rightarrow 3)-linked branch.

Our attention was focused next on tris-glycosylation to introduce β -D-glucose residues into **6** concurrently. As a preliminary experiment, the triol **6** was treated with the donor **5** (3.6 equiv.) in the presence of NIS- TfOH for 2 days. As predicted by van Boom,³ this was less successful and led to a complex mixture of products, from which crude fully benzoylated heptaglycoside, hexaoside, and gentiopentaoside were obtained in 25, 39, and 18% yields, respectively. However, ^1H NMR spectra of the crude oligosaccharide thus obtained revealed contamination of several undesired α -glucosidic linkages.



Scheme 4 Synthesis of gentio-oligosaccharide intermediate. *Reagents and conditions:* (i) NIS- TfOH , CH_2Cl_2 , -20°C to rt; then H_2O -pyridine, 50°C , overnight, **7**: 92%; (ii) NIS- TfOH , CH_2Cl_2 , -20°C to rt; then H_2O -pyridine, 50°C , overnight, **4** \rightarrow **8**: 92% and **5** \rightarrow **9**: 88%; (iii) PdCl_2 , NaOAc -95% AcOH_{aq} , ultrasonication, rt, **10**: 90%; (iv) NIS- TfOH , CH_2Cl_2 , -20°C to rt; then H_2O -pyridine, 50°C , overnight, **11**: 92%; (v) PdCl_2 , NaOAc -95% AcOH_{aq} , ultrasonication, rt, **6**: 90%.

These disappointing results forced us to seek better reaction conditions for the introduction of the β -(1 \rightarrow 3)-linked branches. Using the model acceptor **10** and the disarmed donor **5**, we evaluated the recently developed promoters of thioglycosides as shown in Scheme 5; the results are summarized in Table 1. To our surprise, the reaction using NIS-TfOH at -40°C to room temperature in dry CH_2Cl_2 (entry 1) afforded **13** which was isolated as an anomeric mixture ($\alpha : \beta = 1 : 1.4$) in 41% yield, even though a participating benzoyl ester is present at the C-2 position in **5**. Similarly, the well-known thioethyl counterpart **12** was also able to provide **13** with a slightly poorer $\alpha : \beta$ ratio (entry 3). Furthermore, we also examined milder thiophilic reagents, NIS-TfOH-AgOTf⁹ and ICl-AgOTf,¹⁰ for which a substoichiometric amount of promoter has been reported to activate the thioglycoside, producing a sulfonium byproduct that can further promote glycosylations. However, **13** was obtained in 28% yield and with disappointing stereoselectivity ($\alpha : \beta = 1 : 1.4$); representative reactions are highlighted in entry 4. Although the unusual formation of 1,2-*cis*-glycosides has been explained by unfavorable and mismatched structure between a glycosyl donor and an acceptor,¹¹ little has been reported on β -(1 \rightarrow 3)-linked systems. We hypothesized that severe steric hindrance between acceptor **10** and an orthoester intermediate derived from the donor **5** occurred in oxocarbenium ion leading to the β -glycosidic bond, but not in the transition state leading to the α -bond. We next focused on a BSP-Tf₂O promoter



Scheme 5 Model glycosylation to introduce β -(1 \rightarrow 3)-branch.

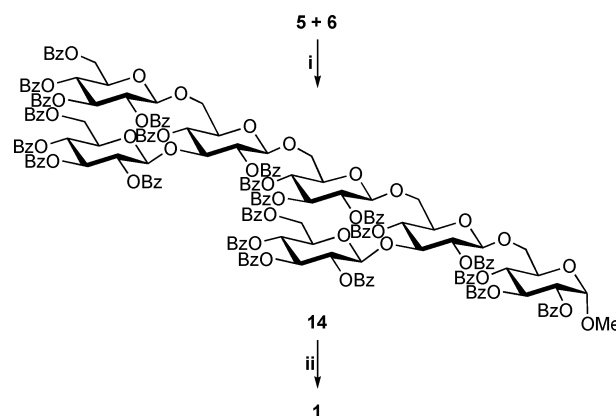
Table 1 Evaluation of promoter system of thioglycosides in β -(1 \rightarrow 3)-branch formation^a

Entry	Promoter	Temperature	Yield ($\alpha : \beta$)
1	NIS-TfOH	-40°C to rt	41% (1 : 1.4)
2 ^b	NIS-TfOH	-40°C to rt	39% (1 : 1.2)
3 ^c	NIS-TfOH	-40°C to rt	33% (1 : 1.3)
4	NIS-TfOH-AgOTf	-40°C to rt	28% (1 : 1.4)
5	ICl-AgOTf	-40°C to rt	No trace
6	BSP-Tf ₂ O	-78°C to rt	44% (β only)
7 ^c	BSP-Tf ₂ O	-78°C to rt	60% (β only)

^a All reactions were performed with 1.2 eq. of donor in CH_2Cl_2 . Anomeric ratio determined by integration of the ^1H NMR spectrum of the crude reaction mixture. ^b Glycosyl donor **12** was used. ^c Donor **5** (1.5 eq.) was used.

system developed by Crich *et al.*, which is known to facilitate formation of a highly reactive glycosyl triflate intermediate. As shown in entries 6 and 7, BSP-Tf₂O mediated condensation between **10** and **5** proceeded with complete stereoselectivity to yield the β -linked tetraglucoside **13** in 44% yield. Use of slightly excess donor was more effective to yield **13** (60%). This β -selectivity can be explained by the rapid S_N2-like displacement of the intermediate α -triflate, which has also been observed by the groups of Crich,^{5,12a} van der Marel,^{12b,c} and Yoshida.^{12d} The satisfactory stereoselectivity and product yield of the BSP-Tf₂O mediated glycosylation made it applicable to our purpose.

Based on these fruitful results, we finally undertook the culminating step of our synthetic scheme, the tris-glycosylation of **6** to construct the heptaglucoside. As shown in Scheme 6, the dodecylthio donor **5** was pre-activated with BSP-Tf₂O at -78°C for 15 min, and the tetraose acceptor **6**, which has three hydroxyl groups, was subsequently added and the reaction was completed while gradually increasing the reaction temperature from -78 to -40°C overnight. As we expected, the activation and condensation proceeded smoothly, as monitored by two-dimensional TLC analysis (hexane-EtOAc, 1 : 1 and toluene-EtOAc 5 : 1, v/v). Repeated chromatography on a silica gel column with these solvent systems furnished the fully protected heptaglucoside **14** in 47% yield together with a mixture of hexaglucosides in 42% yield. In a 600 MHz ^1H NMR spectrum of **14**, six anomeric protons, except for one at the reducing end, appeared as doublets with large coupling constants, suggesting all glucosidic linkages were β in configuration. Final removal of all benzoyl groups was performed with NaOMe in MeOH-H₂O to give known target compound **1**, the ^1H NMR spectrum of which was consistent with the reported data.^{2c,3}



Scheme 6 Synthesis of the heptaglucoside **1**. Reagents and conditions: (i) BSP-Tf₂O, CH_2Cl_2 , -78°C to -40°C , overnight, **14**: 47%; (ii) NaOMe, MeOH-H₂O = 1 : 1, 24 h, rt, **1**: 74%.

Conclusions

It has been shown for the first time that a concise synthesis of the branched structure of the phytoalexin elicitor heptaglucoside **1** is enabled by employing sulfonyl triflate mediated tris-glycosylation of the gentiotetraose intermediate **6**. Since this approach allowed the introduction of two β -(1 \rightarrow 3)-glucosyl branches at the final stage of the synthesis, this strategy would be applicable to the

synthesis of various analogues of **1** that have different mono- and/or oligosaccharide branches.

Experimental

General methods

All chemicals were purchased as reagent grade and used without further purification whereas NIS was recrystallized by 1,4-dioxane and diethyl ether (1 : 1, v/v) before use. Dichloromethane (CH_2Cl_2) and 1,2-dichloroethane were distilled over calcium hydride. Molecular sieves used for glycosylation were MS4Å, which were activated at 200 °C under reduced pressure prior to use. Reaction monitoring was done with analytical thin-layer chromatography (TLC) on silica gel 60F₂₅₄ plates (layer-thickness, 0.25 mm; E. Merck, Darmstadt, Germany), which were visualized under UV (254 nm) and/or by spraying with *p*-methoxybenzaldehyde– H_2SO_4 –MeOH (1 : 2 : 17, v/v). Medium pressure column chromatography was performed on silica gel (LiChroprep Si 60; 40–63 µm, Merck, Darmstadt, Germany). Column chromatography was performed on silica gel (Silica gel 60; 70–230 mesh ASTM, Merck, Darmstadt, Germany).

¹H and ¹³C NMR spectra were recorded with a Bruker ASX 300 (300 and 75.1 MHz, respectively), JEOL A 500 (500 and 125 MHz) or JEOL ECA 600 (600 and 150 MHz). Splitting patterns are indicated as s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet; brs, broad singlet for ¹H NMR data. Signals were assigned on the basis of ¹H–¹H COSY and ¹H–¹H TOCSY NMR experiments. ESI-HR mass spectra were recorded on a JEOL JMS-100TJ spectrometer and ESI-TOF HR mass spectra were measured on a Bruker micro TOF focus spectrometer. MALDI-TOF mass spectrometry was carried out using a Bruker-Daltonik Ultraflex TOF mass spectrometer equipped with a pulsed ion extraction system.

Dodecyl 2,4,6-tri-*O*-acetyl-3-allyl-*O*-1-thio-β-D-glucopyranoside (**18**)

Compound **17**⁶ (500 mg, 1.3 mmol) and 1-dodecanethiol (0.5 mL, 2.0 mmol) was dissolved in 1,2-dichloroethane (10 mL) and the solution was cooled to 0 °C. To the solution was added slowly $\text{BF}_3 \cdot \text{Et}_2\text{O}$ (0.45 mL, 1.9 mmol), then the mixture was stirred for 30 min at 0 °C. TLC (toluene–EtOAc, 5 : 1, v/v) confirmed complete disappearance of the starting material. The reaction mixture was poured into ice-water, and extracted with CHCl_3 . The resulting mixture was successively washed with saturated aqueous (sat. aq.) NaHCO_3 , brine, dried (MgSO_4), and concentrated. The residue was recrystallized in EtOH to yield **18** (600 mg, 87%). $[\alpha]_D^{27} = -30.8$ (*c* 1.24, CHCl_3); ¹H NMR (300 MHz, CDCl_3): δ 5.81 (ddd, 1H, *J* = 5.5 Hz, *J* = 10.6 Hz, *J* = 22.6 Hz, $\text{CH}_2\text{--CH=CH}_2$), 5.30 (d, 1H, *J* = 17.0 Hz, $\text{CH}_2\text{--CH=CH}_2$), 5.18 (d, 1H, *J* = 7.7 Hz, $\text{CH}_2\text{--CH=CH}_2$), 5.10 (t, 1H, *J*_{4,5} = 10.2 Hz, H-4), 5.04 (t, 1H, *J*_{2,3} = 9.8 Hz, H-2), 4.40 (d, 1H, *J*_{1,2} = 10.0 Hz, H-1), 4.26 (dd, 1H, *J*_{5,6b} = 4.1 Hz, H-6b), 4.13 (dd, 1H, *J*_{5,6a} = 2.2 Hz, *J*_{6a,6b} = 14.7 Hz, H-6a), 4.30–4.00 (m, 2H, $\text{CH}_2\text{--CH=CH}_2$), 3.61 (t, 1H, *J*_{3,4} = 9.2 Hz, H-3), 3.70–3.50 (m, 1H, H-5), 2.90–2.60 (m, 2H, SCH_2), 2.10, 2.10, 2.09 (3 s, each 3H, CH_3Ac), 1.87–1.65 (m, 20H, $\text{SCH}_2(\text{CH}_2)_{10}\text{CH}_3$), 0.92 (t, 3H, *J* = 4.1 Hz, CH_2CH_3); ¹³C NMR (75 MHz, CDCl_3): δ 171.0, 169.5, 169.4 (C=O Ac), 134.4,

131.1, 117.2, 84.0, 81.5, 76.8, 76.4, 73.3, 71.4, 69.8, 62.8, 32.2, 30.6, 30.1, 29.9, 29.8, 29.6, 29.4, 29.1, 22.9, 21.2, 21.1, 14.4, 11.2; Anal. Calcd for $\text{C}_{27}\text{H}_{46}\text{O}_8\text{S}$: C, 61.10; H, 8.74; S, 6.04%; Found: C, 60.96; H, 8.75; S, 6.07%.

Dodecyl 3-*O*-allyl-2,4-di-*O*-benzoyl-6-*O*-trityl-1-thio-β-D-glucopyranoside (**19**)

To a solution of **18** (500 mg, 0.9 mmol) in MeOH (10 mL) was added a 25% solution of NaOMe in MeOH to adjust the solution to pH >12, and the mixture was stirred at room temperature for 3 h, when TLC (CHCl_3 –MeOH, 10 : 1, v/v) indicated that the reaction was complete. The reaction mixture was neutralized with Amberlite IR-120 (H^+ form), filtered, and concentrated to give the 2,4,6-triol. To a dry pyridine (10 mL) solution of the resulting triol, DMAP (165 mg, 1.35 mmol), and chlorotriphenylmethane (362 mg, 1.3 mmol) were added, and the solution was stirred at 60 °C for 5 h, when TLC (toluene–EtOAc, 20 : 1, v/v) indicated that the reaction was completed. After addition of benzoyl chloride (0.6 mL, 5.6 mmol), the reaction mixture was stirred at 60 °C, overnight. The mixture was partitioned between CHCl_3 and ice-water. The organic phase was washed with 1 M HCl, sat. aq. NaHCO_3 , brine, dried (MgSO_4), and concentrated. Purification on silica gel column chromatography with hexane–EtOAc, 20 : 1→5 : 1, v/v as the eluant afforded compound **19** (700 mg, 91%). $[\alpha]_D^{26} = -19.8$ (*c* 1.22, CHCl_3); ¹H NMR (300 MHz, CDCl_3): δ 8.20–7.10 (m, 25H, CH_{arom}), 5.62 (ddd, 1H, *J* = 5.8 Hz, *J* = 8.0 Hz, *J* = 16.9 Hz, $\text{CH}_2\text{--CH=CH}_2$), 5.52 (t, 1H, *J*_{2,3} = 9.6 Hz, H-2), 5.51 (t, 1H, *J*_{4,5} = 9.6 Hz, H-4), 5.05 (dd, 1H, *J* = 1.4 Hz, *J* = 17.2 Hz, $\text{CH}_2\text{--CH=CH}_2$), 4.93 (d, 1H, *J* = 10.4 Hz, $\text{CH}_2\text{--CH=CH}_2$), 4.77 (d, 1H, *J*_{1,2} = 10.0 Hz, H-1), 4.09 (d, 2H, $\text{CH}_2\text{--CH=CH}_2$), 4.00 (t, 1H, *J*_{3,4} = 9.2 Hz, H-3), 3.84–3.81 (m, 1H, H-5), 3.50–3.30 (m, 2H, H-6a, 6b), 3.04–2.80 (m, 2H, SCH_2), 1.85–1.68, 1.47–1.28 (m, 20H, $\text{SCH}_2(\text{CH}_2)_{10}\text{CH}_3$), 0.98 (t, 3H, *J* = 6.9 Hz, CH_2CH_3); ¹³C NMR (75 MHz, CDCl_3): δ 171.3, 165.2, 164.9, 147.1, 143.9, 143.5, 134.5, 133.4, 133.3, 130.0, 129.9, 129.2, 128.8, 128.7, 128.5, 128.2, 128.1, 127.9, 127.4, 127.3, 127.1, 117.6, 86.8, 83.9, 81.6, 78.5, 73.5, 72.6, 71.0, 63.0, 60.6, 32.1, 30.2, 30.1, 29.9, 29.8, 29.6, 29.3, 29.3, 22.9, 14.4; ESI-HRMS (*m/z*) calcd for $\text{C}_{54}\text{H}_{62}\text{O}_7\text{SNa}^+$: 877.4108; Found: 877.4113.

Dodecyl 3-*O*-allyl-2,4-di-*O*-benzoyl-6-*O*-chloroacetyl-1-thio-β-D-glucopyranoside (**3**)

To a solution of the tritylated compound **19** (500 mg, 0.58 mmol) in acetic acid (40 mL) was added H_2O (8 mL) at 70 °C. The mixture was stirred at this temperature for 1 h, after which time the reaction was completed as indicated by TLC (hexane–EtOAc, 3 : 1, v/v). The reaction mixture was evaporated, then subjected to co-evaporation with toluene three times. Purification on silica gel column chromatography (hexane–EtOAc, 7 : 1→5 : 1, v/v) gave a colorless oil (330 mg, 93%).

A solution of the resulting de-*O*-tritylated thioglycoside (330 mg, 0.54 mmol) in pyridine– CH_2Cl_2 (1 : 10, v/v, 10 mL) was treated with chloroacetic anhydride (103 mg, 0.6 mmol); the reaction was allowed to process for 3 h. The mixture was successively washed with ice-water, sat. aq. NaHCO_3 , and brine, dried (MgSO_4), and concentrated. Purification with silica gel column chromatography (hexane–EtOAc, 7 : 1→5 : 1, v/v) gave

compound **5** (380 mg, 92% in two steps). $[\alpha]_D^{25} = -13.7$ (*c* 0.96, CHCl₃); ¹H NMR (300 MHz, CDCl₃): δ 8.18–7.27 (m, 10H, CH_{arom}), 5.54 (ddd, 1H, *J* = 5.8 Hz, *J* = 10.4 Hz, *J* = 22.6 Hz, CH₂–CH=CH₂), 5.39 (t, 1H, *J*_{4,5} = 9.7 Hz, H-4), 5.35 (t, 1H, *J*_{2,3} = 9.7 Hz, H-2), 5.03 (dd, 1H, *J* = 1.4 Hz, *J* = 17.2 Hz, CH₂–CH=CH₂), 4.90 (d, 1H, *J* = 10.3 Hz, CH₂–CH=CH₂), 4.66 (d, 1H, *J*_{1,2} = 10.0 Hz, H-1), 4.38–4.31 (m, 2H, H-6a, 6b), 4.10–4.00 (m, 4H, COCH₂Cl, CH₂–CH=CH₂), 3.98 (t, 1H, *J*_{3,4} = 9.1 Hz, H-3), 3.95–3.78 (m, 1H, H-5), 2.78–2.63 (m, 2H, SCH₂), 1.30–1.08 (m, 20H, SCH₂(CH₂)₁₀CH₃), 0.88 (t, 3H, *J* = 7.0 Hz, CH₂CH₃); ¹³C NMR (75 MHz, CDCl₃): δ 167.4, 165.4, 165.3, 140.0, 133.7, 134.4, 130.2, 130.1, 130.0, 129.4, 129.0, 128.8, 117.9, 84.3, 81.1, 73.8, 72.3, 70.8, 64.7, 63.3, 41.1, 30.3, 30.5, 30.0, 29.9, 29.8, 29.7, 29.5, 29.1, 23.0, 14.5; ESI-HRMS (*m/z*) calcd for C₃₇H₄₉ClO₈SN⁺: 711.2729; Found: 711.2748.

Dodecyl 2,3,4,6-tetra-*O*-acetyl-1-thio- β -D-glucopyranoside (**20**)

To a solution of 1,2,3,4,6-penta-*O*-acetyl- β -D-glucopyranose (780 mg, 2.0 mmol) and 1-dodecanethiol (0.53 mL, 2.2 mmol) in 1,2-dichloroethane (10 mL) was added BF₃·Et₂O (0.3 mL, 2.4 mmol) and the solution was stirred for 40 min at room temperature. The reaction was quenched with Et₃N and evaporated. The residue was recrystallized in EtOH to yield **20** (1.0 g, 94%). $[\alpha]_D^{25} = -29.2$ (*c* 0.12, CHCl₃); ¹H NMR (300 MHz, CDCl₃): δ 5.24 (t, 1H, *J*_{3,4} = 9.3 Hz, H-3), 5.10 (t, 1H, *J*_{4,5} = 9.5 Hz, H-4), 5.05 (t, 1H, *J*_{2,3} = 9.4 Hz, H-2), 4.49 (d, 1H, *J*_{1,2} = 10.0 Hz, H-1), 4.26 (dd, 1H, *J*_{5,6b} = 4.9 Hz, H-6b), 4.05 (dd, 1H, *J*_{5,6a} = 2.2 Hz, *J*_{6a,6b} = 12.3 Hz, H-6a), 3.75–3.65 (m, 1H, H-5), 2.76–2.60 (m, 2H, SCH₂), 2.10, 2.08, 2.04, 2.03 (4 s, each 3H, CH₃Ac), 1.66–1.26 (m, 20H, SCH₂(CH₂)₁₀CH₃), 0.88 (t, 3H, *J* = 4.1 Hz, CH₂CH₃); ¹³C NMR (75 MHz, CDCl₃): δ 171.1, 170.7, 169.9, 169.9 (C=O), 84.2, 76.4, 74.5, 70.5, 68.9, 62.8, 32.4, 30.6, 30.2, 30.1, 30.1, 29.9, 29.7, 29.3, 23.2, 21.2, 21.1, 14.6; Anal. Calcd for C₂₆H₄₄O₉S: C, 58.62; H, 8.33; S, 6.02%; Found: C, 58.55; H, 8.46; S, 6.17%.

Dodecyl 2,3,4-tri-*O*-benzoyl-6-*O*-trityl-1-thio- β -D-glucopyranoside (**21**)

To a solution of **20** (1.0 g, 1.9 mmol) in MeOH (50 mL) was added a 25% solution of NaOMe in MeOH to adjust the pH to >12 at room temperature. After 3 h, TLC (CHCl₃–MeOH, 5 : 1, v/v) showed that the reaction was completed. The reaction mixture was neutralized with Amberlite IR-120 (H⁺ form), filtered, and concentrated *in vacuo*. The product and chlorotriphenylmethane (1.2 g, 4.0 mmol) were dissolved in dry pyridine (10 mL) and the solution was heated to 60 °C, overnight. TLC (toluene–EtOAc, 20 : 1, v/v) indicated that the reaction was completed, and that one product was formed. To the reaction mixture was added benzoyl chloride (1.3 mL, 11.1 mmol) at 0 °C, the reaction mixture was stirred at the same temperature for 30 min and at room temperature overnight. TLC (hexane–EtOAc, 5 : 1, v/v) then showed that the reaction was completed. The mixture was diluted with CHCl₃, then poured into ice-water. The organic phase was washed with 1 M HCl, sat. aq. NaHCO₃, brine, dried (MgSO₄), and concentrated. Purification on silica gel column chromatography (hexane–EtOAc, 15 : 1→5 : 1, v/v) as the eluant afforded compound **21** (1.5 g, 86%) as a colorless oil. $[\alpha]_D^{25} = +0.26$ (*c* 0.99, CHCl₃); ¹H NMR (300 MHz, CDCl₃): δ 8.29–7.07 (m, 30H, CH_{arom}), 5.86 (t, 1H,

*J*_{3,4} = 9.5 Hz, H-3), 5.68 (t, 1H, *J*_{4,5} = 9.8 Hz, H-4), 5.64 (t, 1H, *J*_{2,3} = 9.5 Hz, H-2), 4.85 (d, 1H, *J*_{1,2} = 9.8 Hz, H-1), 3.95–3.89 (m, 1H, H-5), 3.40 (dd, 1H, *J*_{5,6a} = 2.2 Hz, *J*_{6a,6b} = 10.6 Hz, H-6a), 3.30 (dd, 1H, *J*_{5,6b} = 4.9 Hz, H-6b), 2.97–2.82 (m, 2H, SCH₂), 1.87–1.25 (m, 20H, SCH₂(CH₂)₁₀CH₃), 0.92 (t, 3H, *J* = 7.0 Hz, CH₂CH₃); ¹³C NMR (75 MHz, CDCl₃): δ 166.2, 165.6, 165.1, 143.9, 133.5, 133.4, 133.4, 130.5, 130.2, 130.1, 130.0, 129.6, 129.3, 128.9, 128.8, 128.7, 128.6, 128.5, 128.0, 127.2, 87.0, 84.0, 74.8, 71.2, 69.7, 62.9, 60.7, 32.2, 30.3, 30.2, 30.0, 29.8, 29.7, 29.5, 29.3, 23.0, 14.5; ESI-HRMS (*m/z*) calcd for C₅₈H₆₂O₈SN⁺: 941.4058; Found: 941.4061.

Dodecyl 2,3,4-tri-*O*-benzoyl-6-*O*-chloroacetyl-1-thio- β -D-glucopyranoside (**4**)

To a solution of the tritylated compound **21** (500 mg, 0.5 mmol) in acetic acid (40 mL) was added H₂O (8 mL) at 70 °C. The mixture was stirred at the same temperature for 1 h, when the reaction was complete as indicated by TLC (hexane–EtOAc, 3 : 1, v/v). The reaction mixture was evaporated, then subjected to co-evaporation with toluene three times and purification by silica gel column chromatography (hexane–EtOAc, 7 : 1→5 : 1, v/v) gave a colorless oil (318 mg, 94%).

A solution of the resulting de-*O*-tritylated thioglycoside (318 mg, 0.47 mmol) in pyridine–CH₂Cl₂ (1 : 10, v/v, 10 mL) was treated with chloroacetic anhydride (100 mg, 0.6 mmol), and the reaction was stirred for 2 h. The mixture was successively washed with ice water, sat. aq. NaHCO₃, and brine, dried (MgSO₄), and concentrated. Purification by silica gel column chromatography (hexane–EtOAc, 7 : 1→5 : 1, v/v) gave compound **4** (312 mg, 87% in two steps). $[\alpha]_D^{25} = -6.2$ (*c* 1.24, CHCl₃); ¹H NMR (300 MHz, CDCl₃): δ 8.09–7.10 (m, 15H, CH_{arom}), 5.93 (t, 1H, *J*_{3,4} = 9.5 Hz, H-3), 5.59 (t, 1H, *J*_{4,5} = 10.0 Hz, H-4), 5.64 (t, 1H, *J*_{2,3} = 9.8 Hz, H-2), 4.83 (d, 1H, *J*_{1,2} = 10.0 Hz, H-1), 4.50–4.30 (m, 2H, H-6a, 6b), 4.20–4.00 (m, 3H, H-5, COCH₂Cl), 2.81–2.70 (m, 2H, SCH₂), 1.34–1.24 (m, 20H, SCH₂(CH₂)₁₀CH₃), 0.92 (t, 3H, *J* = 6.9 Hz, CH₂CH₃); ¹³C NMR (75 MHz, CDCl₃): δ 167.3, 165.5, 165.1, 165.3, 133.9, 133.6, 130.1, 130.1, 129.9, 129.3, 128.9, 128.8, 128.7, 128.6, 128.6, 84.3, 74.2, 70.7, 69.4, 64.3, 40.9, 32.2, 30.4, 29.9, 29.8, 29.8, 29.6, 29.4, 29.0, 22.9, 14.4; ESI-HRMS (*m/z*) calcd for C₄₁H₄₉ClO₉SN⁺: 775.2678; Found: 775.2683.

Dodecyl 2,3,4,6-tetra-*O*-benzoyl-1-thio- β -D-glucopyranoside (**5**)

To a solution of the acetate **20** (600 mg, 1.13 mmol) in MeOH (20 mL) was added NaOMe (20 mg), and the solution was stirred at room temperature for 5 h. After neutralization with Amberlite IR-120 (H⁺ form), the solvent was evaporated and the residue was dissolved in pyridine (8 mL). Benzoyl chloride (0.8 mL, 6.8 mmol) was added to the solution. The mixture was stirred at room temperature overnight, poured into crushed ice-water and extracted with CHCl₃. The extract was washed successively with 1 M HCl, sat. aq. NaHCO₃ and brine, dried with MgSO₄, and evaporated. The residue was chromatographed on silica gel (toluene–EtOAc 20 : 1→10 : 1, v/v) to give the benzoate **5** (757 mg, 86%) as a colorless oil. $[\alpha]_D^{25} = +12.4$ (*c* 1.70, CHCl₃); ¹H NMR (300 MHz, CDCl₃): δ 8.03–7.17 (m, 20H, CH_{arom}), 5.95 (t, 1H, *J*_{3,4} = 9.5 Hz, H-3), 5.69 (t, 1H, *J*_{4,5} = 9.7 Hz, H-4), 5.56 (t, 1H, *J*_{2,3} = 9.6 Hz, H-2), 4.87 (d, 1H, *J*_{1,2} = 10.0 Hz, H-1), 4.65 (dd,

1H, $J_{5,6b} = 3.1$ Hz, H-6b), 4.51 (dd, 1H, $J_{5,6a} = 5.4$ Hz, $J_{6a,6b} = 12.2$ Hz, H-6a), 4.26–4.16 (m, 1H, H-5), 2.76–2.60 (m, 2H, SCH₂), 1.66–1.26 (m, 20H, SCH₂(CH₂)₁₀CH₃), 0.88 (t, 3H, $J = 4.1$ Hz, CH₂CH₃); ¹³C NMR (75 MHz, CDCl₃): δ 166.4, 166.2, 165.6, 165.5, 133.8, 133.6, 133.6, 133.4, 130.2, 130.2, 130.1, 129.6, 129.2, 129.2, 128.7, 128.7, 128.6, 84.4, 76.7, 74.6, 71.1, 70.1, 63.7, 32.3, 30.5, 30.1, 30.0, 30.0, 29.9, 29.8, 29.7, 29.4, 29.1, 23.0, 14.4; Anal. Calcd for C₄₆H₅₂O₉S: C, 70.74; H, 6.71; S, 4.11%; Found: C, 70.92; H, 6.65; S, 3.97%.

General procedure for the preparation of gentio-oligosaccharides using NIS–TfOH

A solution of the glycosyl donor (1.2 equiv. to acceptor), glycosyl acceptor (1.0 equiv.) and freshly recrystallized NIS (1.2 equiv. to donor) in CH₂Cl₂ was stirred for 30 min over activated molecular sieves under a nitrogen atmosphere. TfOH (5 μ L) was added at –20 °C with a micro syringe. The reaction mixture was stirred and the temperature was warmed slowly to room temperature. After TLC (toluene–EtOAc, 5 : 1, v/v) analysis to check that the glycosyl donor was consumed, the reaction was quenched with Et₃N (about 2 mL) and then diluted with CHCl₃. The mixture was filtered through a Celite pad, and the filtrate was washed successively with sat. aq. Na₂S₂O₃, sat. aq. NaHCO₃, and brine, dried (MgSO₄), and concentrated to give a crude oil, which mainly consisted of a chloroacetated product.

The crude mixture of the above mentioned glycosylation was dissolved in pyridine (20 mL) and H₂O (4 mL). After stirring overnight at 50 °C, the reaction was completed as indicated by TLC (toluene–EtOAc, 2.5 : 1, v/v). The reaction mixture was evaporated, then subjected to co-evaporation with toluene three times. Purification was carried out by silica gel column chromatography (toluene–EtOAc).

Methyl 6-*O*-(3-*O*-allyl-2,4-di-*O*-benzoyl- β -D-glucopyranosyl)-2,3,4-tri-*O*-benzoyl- α -D-glucopyranoside (7)

The glycosyl donor **3** (689 mg, 1.0 mmol) was condensed with glycosyl acceptor **2** (761 mg, 1.2 mmol) using NIS (270 mg, 1.2 mmol), and TfOH (0.02 mmol, 5 μ L) in dry CH₂Cl₂ (20 mL) according to the general procedure described above. Compound **7** was obtained as a colorless oil (845 mg, 92% in two steps). $[a]_D^{26} = +2.4$ (*c* 1.25, CHCl₃); ¹H NMR (600 MHz, CDCl₃): δ 8.13–7.22 (m, 25H, CH_{arom}), 6.07 (t, 1H, $J_{3,4} = 9.9$ Hz, H-3), 5.60–5.50 (ddd, 1H, $J = 6.0$ Hz, $J = 11.0$ Hz, $J = 22.5$ Hz, CH₂–CH=CH₂), 5.42 (t, 1H, $J_{4,5} = 9.3$ Hz, H-4), 5.30 (t, 1H, $J_{2',3'} = 8.2$ Hz, H-2'), 5.19 (t, 1H, $J_{4',5'} = 9.3$ Hz, H-4'), 5.11 (dd, 1H, $J = 3.8$ Hz, $J_{2,3} = 10.2$ Hz, H-2), 5.01 (d, 1H, $J = 10.4$ Hz, CH₂–CH=CH₂), 4.95 (d, 1H, $J_{1,2} = 3.8$ Hz, H-1), 4.90 (d, 1H, $J = 10.4$ Hz, CH₂–CH=CH₂), 4.79 (d, 1H, $J_{1',2'} = 8.2$ Hz, H-1'), 4.23–4.17 (m, 1H, H-5), 4.09 (dd, 1H, $J_{5,6b} = 2.2$ Hz, H-6b), 4.03 (d, 2H, $J = 5.5$ Hz, CH₂–CH=CH₂), 3.98 (t, 1H, $J_{3',4'} = 9.3$ Hz, H-3'), 3.75 (dd, 1H, $J_{5,6a} = 6.6$ Hz, $J_{6a,6b} = 11.0$ Hz, H-6a), 3.73–3.67 (m, 1H, H-6'b), 3.65–3.60 (1 m, 1H, H-5'), 3.58–3.52 (m, 1H, H-6'b), 3.15 (s, 3H, OMe), 2.80–2.72 (m, 1H, OH); ¹³C NMR (75 MHz, CDCl₃): δ 169.7, 166.1, 166.0, 134.6, 134.0, 133.7, 133.4, 130.3, 130.2, 130.1, 130.0, 129.6, 129.4, 129.2, 128.9, 128.8, 128.6, 117.5, 101.0, 97.2, 80.0, 75.3, 73.4, 72.4, 70.9, 69.9, 68.7, 67.2, 61.8, 55.9; ESI-HRMS (*m/z*) calcd for C₅₁H₄₈O₁₆Na⁺: 939.2835; Found: 939.2833.

Methyl 6-*O*-(6-*O*-2,3,4-tri-*O*-benzoyl- β -D-glucopyranosyl)-3-*O*-allyl-2,4-di-*O*-benzoyl- β -D-glucopyranosyl)-2,3,4-tri-*O*-benzoyl- α -D-glucopyranoside (8)

The glycosyl donor **4** (459 mg, 0.5 mmol) was condensed with glycosyl acceptor **7** (452 mg, 0.42 mmol) using NIS (270 mg, 1.2 mmol), and TfOH (0.02 mmol, 5 μ L) in dry CH₂Cl₂ (10 mL) according to the general procedure described above. Compound **8** was obtained as colorless oil (643 mg, 92% in two steps). $[a]_D^{26} = +0.18$ (*c* 0.95, CHCl₃); ¹H NMR (600 MHz, CDCl₃): δ 8.10–7.10 (m, 40H, CH_{arom}), 6.06 (t, 1H, $J_{3,4} = 9.7$ Hz, H-3), 5.93 (t, 1H, $J_{3'',4''} = 9.7$ Hz, H-3''), 5.57–5.49 (ddd, 1H, $J = 4.4$ Hz, $J = 5.5$ Hz, $J = 16.4$ Hz, CH₂–CH=CH₂), 5.41 (t, 1H, $J_{4,5} = 9.7$ Hz, H-4), 5.27 (t, 1H, $J_{2',3'} = 9.8$ Hz, H-2'), 5.26 (t, 1H, $J_{4',5'} = 9.1$ Hz, H-4'), 5.23 (t, 1H, $J_{4'',5''} = 9.8$ Hz, H-4''), 5.19 (t, 1H, $J_{2'',3''} = 9.3$ Hz, H-2''), 5.11 (dd, 1H, $J = 3.8$ Hz, $J_{2,3} = 10.2$ Hz, H-2), 5.02 (d, 1H, $J_{1,2} = 3.8$ Hz, H-1), 4.98 (dd, 1H, $J = 1.6$ Hz, $J = 17.3$ Hz, CH₂–CH=CH₂), 4.93 (d, 1H, $J_{1',2'} = 7.7$ Hz, H-1'), 4.87 (dd, 1H, $J = 1.6$ Hz, $J = 13.2$ Hz, CH₂–CH=CH₂), 4.59 (d, 1H, $J_{1'',2''} = 8.2$ Hz, H-1''), 4.07–4.02 (m, 1H, H-5), 4.15–3.93 (m, 4H, H-6b, 6'b, CH₂–CH=CH₂), 3.85 (t, 1H, $J_{3',4'} = 9.3$ Hz, H-3'), 3.84–3.72 (m, 4H, H-5', 5'', 6'a, 6'b), 3.63–3.57 (m, 1H, H-6'a), 3.52 (dd, 1H, $J_{5,6a} = 6.0$ Hz, $J_{6a,6b} = 11.5$ Hz, H-6a), 3.13 (s, 3H, OMe), 2.93–2.87 (m, 1H, OH); ¹³C NMR (75 MHz, CDCl₃): δ 166.1, 166.0, 165.7, 165.4, 165.2, 134.5, 133.8, 133.7, 133.6, 133.6, 133.6, 133.5, 133.4, 133.3, 130.2, 130.0, 130.0, 130.0, 130.0, 130.0, 129.4, 129.4, 129.3, 129.2, 129.0, 125.6, 117.7, 101.7, 100.9, 96.8, 79.6, 74.8, 74.0, 73.5, 73.3, 73.1, 72.4, 72.2, 72.0, 70.6, 70.1, 69.5, 68.9, 68.4, 61.5, 55.5, 21.8; ESI-HRMS (*m/z*) calcd for C₇₈H₇₀O₂₄Na⁺: 1413.4149; Found: 1413.4143.

Methyl 6-*O*-(6-*O*-(6-*O*-(3-*O*-allyl-2,4-di-*O*-benzoyl- β -D-glucopyranosyl)-2,3,4-tri-*O*-benzoyl- β -D-glucopyranosyl)-3-*O*-allyl-2,4-di-*O*-benzoyl- β -D-glucopyranosyl)-2,3,4-tri-*O*-benzoyl- α -D-glucopyranoside (11)

The glycosyl donor **3** (83 mg, 0.12 mmol) was condensed with glycosyl acceptor **8** (143 mg, 0.1 mmol) using NIS (32.4 mg, 0.144 mmol), and TfOH (0.02 mmol, 5 μ L) in dry CH₂Cl₂ (10 mL) according to the general procedure described above. Compound **11** was obtained as a colorless oil (198 mg, 92%). $[a]_D^{23} = -7.6$ (*c* 1.00, CHCl₃); ¹H NMR (600 MHz, CDCl₃): δ 8.20–7.13 (m, 50H), 6.07 (t, 1H, $J_{3,4} = 9.9$ Hz, H-3), 5.73 (t, 1H, $J_{3'',4''} = 9.9$ Hz, H-3''), 5.61–5.51 (m, 2H, CH₂–CH=CH₂), 5.38 (t, 1H, $J_{4,5} = 9.9$ Hz, H-4), 5.29 (t, 1H, $J_{2',3'} = 8.8$ Hz, H-2'), 5.21 (t, 1H, $J_{4',5'} = 9.9$ Hz, H-4'), 5.17 (t, 1H, $J_{2'',3''} = 8.9$ Hz, H-2''), 5.11 (dd, 1H, $J = 7.8$ Hz, $J_{2',3'} = 9.8$ Hz, H-2'), 5.10 (dd, 1H, $J = 3.8$ Hz, $J_{2,3} = 10.2$ Hz, H-2), 5.03 (d, 2H, $J = 17.0$ Hz, CH₂–CH=CH₂), 5.00 (t, 1H, $J_{4'',5''} = 9.9$ Hz, H-4''), 5.00 (t, 1H, $J_{4''',5'''} = 9.9$ Hz, H-4'''), 4.99 (d, 1H, $J_{1,2} = 3.8$ Hz, H-1), 4.91 (d, 2H, $J = 15.9$ Hz, CH₂–CH=CH₂), 4.90 (d, 1H, $J_{1'',2''} = 8.2$ Hz, H-1''), 4.60 (d, 1H, $J_{1',2'} = 8.2$ Hz, H-1'), 4.57 (d, 1H, $J_{1',2'} = 8.2$ Hz, H-1'), 4.12–3.95 (m, 6H, H-5, 6b, CH₂–CH=CH₂), 3.89 (t, 1H, $J_{3'',4''} = 9.9$ Hz, H-3''), 3.88 (t, 1H, $J_{3',4'} = 9.9$ Hz, H-3'), 3.86–3.81 (m, 3H, H-6'b, 5'', 6''a), 3.79–3.69 (m, 3H, H-6'a, 5'', 6''b), 3.66–3.61 (m, 1H, H-5'), 3.59 (dd, 1H, $J_{5'',6''b} = 4.9$ Hz, H-6''b), 3.53 (dd, 1H, $J_{5',6'a} = 1.6$ Hz, $J_{6'a,6'b} = 7.2$ Hz, H-6'a), 3.51 (dd, 1H, $J_{5,6a} = 6.1$ Hz, $J_{6a,6b} = 11.3$ Hz, H-6a), 3.09 (s, 3H, OMe), 2.79–2.89 (brs, 1H, OH); ¹³C NMR (75 MHz, CDCl₃): δ 166.5, 166.3, 166.3, 166.1, 166.0, 165.9,

165.8, 165.6, 165.4, 134.9, 134.8, 134.1, 134.0, 134.0, 133.9, 133.8, 133.7, 133.6, 133.5, 133.4, 131.5, 130.6, 130.5, 130.3, 130.2, 130.2, 130.1, 130.0, 129.8, 129.7, 129.5, 129.4, 129.3, 129.2, 129.1, 129.0, 129.0, 128.8, 128.7, 118.0, 117.9, 102.1, 101.3, 100.9, 97.0, 80.1, 79.9, 75.0, 74.5, 74.2, 74.0, 73.8, 73.1, 73.0, 72.8, 72.6, 71.6, 70.8, 70.2, 69.3, 68.7, 68.0, 62.1, 55.6, 39.3, 30.9, 30.3, 29.5, 24.6, 23.6, 14.6, 11.5; ESI-HRMS (m/z) calcd for $C_{101}H_{92}O_{31}Na^+$: 1823.5515; Found: 1823.5517.

Methyl 6-*O*-(6-*O*-(6-*O*-(2,4-di-*O*-benzoyl- β -D-glucopyranosyl)-2,3,4-tri-*O*-benzoyl- β -D-glucopyranosyl)-2,4-di-*O*-benzoyl- β -D-glucopyranosyl)-2,3,4-tri-*O*-benzoyl- α -D-glucopyranoside (6)

To a solution of compound **11** (158 mg, 0.09 mmol) in acetic acid (3.8 mL) and H_2O (0.2 mL) were added sodium acetate (74 mg, 0.9 mmol) and palladium(II) chloride (64 mg, 0.36 mmol), and the reaction mixture was sonicated at room temperature for 2 h. The solution was then filtered through a celite bed and the filtrate was extracted with $CHCl_3$. The organic layer was washed with H_2O , sat. aq. $NaHCO_3$, and brine, dried ($MgSO_4$), and then purification by silica gel column chromatography (toluene–EtOAc, 5 : 1 \rightarrow 2 : 1, v/v) gave compound **6** (139 mg, 90%) as a syrup foam. $[a]_D^{25} = -8.7$ (c 1.04, $CHCl_3$); 1H NMR (500 MHz, $CDCl_3$): δ 8.20–7.13 (m, 50H), 6.08 (t, 1H, $J_{3,4} = 9.8$ Hz, H-3), 5.79 (t, 1H, $J_{3'',4''} = 9.8$ Hz, H-3''), 5.45 (t, 1H, $J_{4,5} = 9.8$ Hz, H-4), 5.23 (dd, 1H, $J = 7.9$ Hz, $J_{2'',3''} = 9.8$ Hz, H-2''), 5.18 (t, 1H, $J_{4'',5''} = 9.6$ Hz, H-4''), 5.14 (t, 1H, $J_{4',5'} = 10.4$ Hz, H-4'), 5.14 (t, 1H, $J_{2',3'} = 9.2$ Hz, H-2'), 5.12 (t, 1H, $J_{2,3} = 9.2$ Hz, H-2), 5.06 (t, 1H, $J_{2''',3'''} = 8.5$ Hz, H-2'''), 5.06 (t, 1H, $J_{4''',5'''} = 8.5$ Hz, H-4'''), 5.02 (d, 1H, $J_{1,2} = 3.7$ Hz, H-1), 4.88 (d, 1H, $J_{1''',2'''} = 7.9$ Hz, H-1'''), 4.76 (d, 1H, $J_{1'',2''} = 7.9$ Hz, H-1''), 4.62 (d, 1H, $J_{1',2'} = 7.9$ Hz, H-1'), 4.17 (t, 1H, $J_{3''',4'''} = 9.2$ Hz, H-3'''), 4.15–4.90 (m, 1H, H-5), 4.01 (dd, 1H, $J_{5,6b} = 1.8$ Hz, H-6b), 3.96 (t, 1H, $J_{3',4'} = 9.2$ Hz, H-3'), 3.92–3.98 (m, 1H, H-5''), 3.93 (dd, 1H, $J_{5'',6''b} = 3.7$ Hz, H-6''b), 3.89 (dd, 1H, $J_{5',6'b} = 6.1$ Hz, H-6'b), 3.81 (dd, 1H, $J_{5'',6''a} = 6.1$ Hz, $J_{6'a,6''b} = 11.0$ Hz, H-6''a), 3.78 (dd, 1H, $J_{5''',6'''} = 3.0$ Hz, H-6'''b), 3.75 (dd, 1H, $J_{5',6'a} = 1.6$ Hz, $J_{6'a,6'b} = 7.2$ Hz, H-6'a), 3.72–3.60 (m, 3H, H-5', 5'', 6''a), 3.59 (dd, 1H, $J_{5,6a} = 6.1$ Hz, $J_{6a,6b} = 11.3$ Hz, H-6a), 3.16 (s, 3H, OMe), 3.00–2.88 (brs, 3H, OH); ^{13}C NMR (75 MHz, $CDCl_3$): δ 166.9, 166.7, 166.5, 166.4, 166.1, 166.0, 165.9, 165.8, 165.4, 133.9, 133.8, 130.7, 130.6, 130.5, 130.4, 130.3, 130.2, 130.1, 129.9, 129.7, 129.4, 129.2, 128.9, 128.8, 128.7, 128.6, 128.5, 101.4, 101.1, 101.0, 97.0, 75.1, 74.8, 73.9, 73.4, 73.0, 72.3, 70.7, 70.3, 70.1, 69.2, 68.7, 68.4, 68.3, 61.9, 55.6; ESI-HRMS (m/z) calcd for $C_{95}H_{84}O_{31}Na^+$: 1743.5889; Found: 1743.5891.

Methyl 6-*O*-(3,6-di-*O*-(2,3,4,6-tetra-*O*-benzoyl- β -D-glucopyranosyl)-2,4-di-*O*-benzoyl- β -D-glucopyranosyl)-2,3,4-tri-*O*-benzoyl- α -D-glucopyranoside (13)

(a) NIS–TfOH promoted glycosylation. A solution of the glycosyl donor **5** (141 mg, 0.18 mmol), glycosyl acceptor **10** (170 mg, 0.12 mmol), freshly recrystallized NIS (49 mg, 0.22 mmol), and $MS4\text{\AA}$ (300 mg) in dry CH_2Cl_2 (5 mL) was stirred at room temperature for 30 min under a nitrogen atmosphere. To the mixture was added TfOH (5 μ L, 0.02 mmol) at $-40^\circ C$. The reaction mixture was allowed to slowly warm to room temperature overnight (two-dimensional TLC, hexane–EtOAc, (1 : 1, v/v) and toluene–EtOAc (5 : 1, v/v)), then quenched with Et_3N and

diluted with $CHCl_3$, filtered, successively washed with sat. aq. $Na_2S_2O_3$, sat. aq. $NaHCO_3$, and brine. After drying ($MgSO_4$) and concentration, the residue was purified twice by column chromatography (hexane–EtOAc 1 : 1, followed by toluene–EtOAc 10 : 1) to provide **13** as a colorless oil (100.1 mg, 41%, α : β ; 1 : 1.4).

(b) BSP–Tf₂O promoted glycosylation. A solution of the glycosyl donor **5** (141 mg, 0.18 mmol), BSP (56 mg, 0.27 mmol) and $MS4\text{\AA}$ (300 mg) was stirred in dry CH_2Cl_2 (3 mL) at room temperature for 30 min under a nitrogen atmosphere. The reaction mixture was cooled to $-78^\circ C$, followed by addition of Tf_2O (45 μ L, 0.27 mmol) and stirred at this temperature for 15 min. Then a solution of glycosyl acceptor **10** (179 mg, 0.12 mmol) in dry CH_2Cl_2 (2 mL) was added and the reaction mixture was allowed to slowly warm to room temperature overnight (two-dimensional TLC, hexane–EtOAc, (1 : 1, v/v) and toluene–EtOAc (5 : 1, v/v)). Purification as described above gave **13** (107.8 mg, 44%, β only).

(c) NIS–TfOH–AgOTf promoted glycosylation. A solution of the glycosyl donor **5** (31 mg, 40 μ mol), glycosyl acceptor **10** (116 mg, 80 μ mol), freshly recrystallized NIS (6.3 mg, 20 μ mol), and $MS4\text{\AA}$ (300 mg) in dry CH_2Cl_2 (4 mL) was stirred at room temperature for 30 min under a nitrogen atmosphere. To the mixture was added TfOH (2 μ L, 20 μ mol) at $-40^\circ C$ and the mixture was stirred at this temperature for 30 min. Then a solution of AgOTf (7.2 mg, 28 μ mol) in Et_2O (1 mL) was added and the reaction mixture was allowed to slowly warm to room temperature overnight (two-dimensional TLC, hexane–EtOAc, (1 : 1, v/v) and toluene–EtOAc (5 : 1, v/v)). Purification as described above gave **13** (22.4 mg, 28%, α : β ; 1 : 1.4).

(d) ICl–AgOTf promoted glycosylation. A solution of the glycosyl donor **5** (31 mg, 40 μ mol), glycosyl acceptor **10** (116 mg, 80 μ mol), AgOTf (15 mg, 60 μ mol), and $MS4\text{\AA}$ (300 mg) in dry CH_2Cl_2 (4 mL) was stirred at room temperature for 30 min under a nitrogen atmosphere. The reaction mixture was cooled to $-40^\circ C$, followed by the addition of ICl (10 μ L, 0.28 mmol) in CH_2Cl_2 (1 mL), and the temperature was increased gradually from $-40^\circ C$ to room temperature overnight (two-dimensional TLC, hexane–EtOAc, (1 : 1, v/v) and toluene–EtOAc (5 : 1, v/v)) to give no trace condensation products.

Repeated column chromatography of the anomeric mixture of the tetraglucoside with toluene–EtOAc gave pure α - and β -anomers. Spectroscopic data for the β -anomer was identical with that previously prepared by the van Boom group.³

α -Anomer. $[a]_D^{25} = -0.54$ (c 1.00, $CHCl_3$); 1H NMR (600 MHz, $CDCl_3$): δ 8.14–7.03 (m, 65H, CH_{arom}), 6.08 (t, 1H, $J_{3,4} = 9.9$ Hz, H-3), 6.03 (t, 1H, $J_{3'',4''} = 9.9$ Hz, H-3''), 5.87 (t, 1H, $J_{3''',4'''} = 9.9$ Hz, H-3'''), 5.56 (t, 1H, $J_{4'',5''} = 9.9$ Hz, H-4''), 5.51 (t, 1H, $J_{4,5} = 9.9$ Hz, H-4), 5.48 (t, 1H, $J_{4''',5'''} = 9.8$ Hz, H-4'''), 5.46 (d, 1H, $J_{1''',2'''} = 3.8$ Hz, H-1'''), 5.35 (t, 1H, $J_{2',3'} = 8.8$ Hz, H-2'), 5.35 (t, 1H, $J_{2'',3''} = 8.8$ Hz, H-2''), 5.13 (dd, 1H, $J = 3.9$ Hz, $J_{2,3} = 10.1$ Hz, H-2'), 5.11 (dd, 1H, $J = 2.8$ Hz, $J_{2''',3'''} = 10.4$ Hz, H-2'''), 5.08 (t, 1H, $J_{4',5'} = 9.9$ Hz, H-4'), 5.04 (d, 1H, $J_{1,2} = 3.3$ Hz, H-1), 5.01 (d, 1H, $J_{1'',2''} = 7.7$ Hz, H-1''), 4.53 (dd, 1H, $J_{5'',6''b} = 3.3$ Hz, H-6''b), 4.49 (d, 1H, $J_{1',2'} = 7.7$ Hz, H-1'), 3.37 (dd, 1H, $J_{5',6'a} = 5.5$ Hz, $J_{6'a,6''b} = 12.1$ Hz, H-6''a), 4.29 (t, 1H, $J_{3',4'} = 9.4$ Hz, H-3'), 4.25–4.19 (m, 2H, H-6b, 5''), 4.20–4.13 (m, 1H, H-5), 4.02 (dd, 1H, $J_{5''',6'''} = 2.8$ Hz, H-6'''b), 4.01 (m, 1H, H-5'''), 3.77 (dd, 1H,

$J_{5',6'b} = 8.2$ Hz, H-6'b), 3.72 (dd, 1H, $J_{5',6'a} = 2.4$ Hz, $J_{6'a,6'b} = 9.9$ Hz, H-6'a), 3.67–3.61 (m, 1H, H-5'), 3.53 (dd, 1H, $J_{5,6a} = 2.8$ Hz, $J_{6a,6b} = 12.4$ Hz, H-6a), 3.47 (dd, 1H, $J_{5'',6''a} = 4.9$ Hz, $J_{6''a,6''b} = 11.3$ Hz, H-6''a), 2.15 (s, 3H, OMe); ^{13}C NMR (150.9 MHz, CDCl_3): δ 207.0, 166.2, 166.1, 165.8, 165.8, 165.7, 165.6, 165.4, 165.3, 165.2, 165.2, 164.9, 164.8, 133.6, 133.6, 133.5, 133.2, 133.2, 133.2, 133.1, 133.1, 133.1, 133.0, 133.0, 133.0, 133.0, 133.0, 129.4, 129.4, 129.3, 129.2, 129.1, 129.1, 129.0, 128.9, 128.7, 128.6, 128.6, 128.6, 128.5, 128.3, 128.2, 128.1, 101.4, 101.1, 96.7, 74.8, 72.8, 72.5, 72.2, 72.2, 72.1, 70.8, 70.3, 70.2, 69.3, 68.7, 68.6, 68.5, 68.1, 63.3, 61.7, 55.3, 31.0, 29.8; ESI-TOF HRMS (m/z) calcd for $\text{C}_{116}\text{H}_{96}\text{O}_{34}\text{Na}^+$: 2055.5675; Found: 2055.5669.

β -Anomer. $[\alpha]_{\text{D}}^{25} = +0.78$ (c 1.02, CHCl_3); ^1H NMR (600 MHz, CDCl_3): δ 8.04–7.09 (m, 65H, CH_{arom}), 6.05 (t, 1H, $J_{3,4} = 9.9$ Hz, H-3), 6.01 (t, 1H, $J_{3'',4''} = 9.9$ Hz, H-3''), 5.60 (t, 1H, $J_{4,5} = 9.9$ Hz, H-4), 5.55 (t, 1H, $J_{3''',4'''} = 9.3$ Hz, H-3'''), 5.43 (t, 1H, $J_{2'',3''} = 9.8$ Hz, H-2''), 5.40 (t, 1H, $J_{4'',5''} = 9.9$ Hz, H-4''), 5.31 (t, 1H, $J_{2''',3'''} = 8.2$ Hz, H-2'''), 5.28 (t, 1H, $J_{4''',5'''} = 9.9$ Hz, H-4'''), 5.08 (dd, 1H, $J = 3.3$ Hz, $J_{2,3} = 10.4$ Hz, H-2), 5.05 (d, 1H, $J_{1'',2''} = 6.6$ Hz, H-1''), 5.03 (t, 1H, $J_{2',3'} = 7.7$ Hz, H-2'), 5.01 (d, 1H, $J_{1,2} = 3.3$ Hz, H-1), 4.94 (t, 1H, $J_{4',5'} = 8.8$ Hz, H-4'), 4.90 (d, 1H, $J_{1''',2'''} = 7.7$ Hz, H-1'''), 4.58 (dd, 1H, $J_{5,6b} = 2.3$ Hz, H-6b), 4.41 (dd, 1H, $J_{5,6a} = 4.9$ Hz, $J_{6a,6b} = 11.8$ Hz, H-6a), 4.37 (d, 1H, $J_{1',2'} = 7.7$ Hz, H-1'), 4.30–4.25 (m, 1H, H-5), 4.21 (t, 1H, $J_{3',4'} = 8.8$ Hz, H-3'), 4.11 (dd, 1H, $J_{5'',6''a} = 2.1$ Hz, $J_{6''a,6''b} = 8.7$ Hz, H-6''a), 4.00 (dd, 1H, $J_{5',6'b} = 5.5$ Hz, H-6'b), 3.98–3.82 (m, 4H, H-5', 5'', 6''b, 6''b), 3.77 (dd, 1H, $J_{5',6'a} = 8.2$ Hz, $J_{6'a,6'b} = 12.1$ Hz, H-6'a), 3.72–3.69 (m, 1H, H-5'), 3.35 (dd, 1H, $J_{5'',6''a} = 6.6$ Hz, $J_{6''a,6''b} = 12.6$ Hz, H-6''a), 3.04 (s, 3H, OMe); ^{13}C NMR (75 MHz, CDCl_3): δ 166.8, 166.5, 166.3, 166.2, 166.1, 165.9, 165.8, 165.7, 165.6, 164.7, 134.0, 133.8, 133.7, 133.5, 133.4, 130.5, 130.4, 130.3, 130.0, 130.0, 130.0, 129.9, 129.8, 129.7, 129.7, 129.6, 129.5, 129.4, 129.3, 129.2, 129.1, 129.0, 128.9, 128.8, 128.7, 101.9, 101.4, 97.1, 75.4, 74.1, 73.5, 73.3, 72.8, 72.7, 72.5, 72.3, 70.9, 70.4, 70.2, 70.1, 69.2, 68.9, 68.6, 63.8, 55.7, 30.3; ESI-TOF HRMS (m/z) calcd for $\text{C}_{116}\text{H}_{96}\text{O}_{34}\text{Na}^+$: 2055.5675; Found: 2055.5672.

Methyl 6-O-(6-O-(6-O-(3,6-di-O-(2,3,4,6-tetra-O-benzoyl- β -D-glucopyranosyl)-2,4-di-O-benzoyl- β -D-glucopyranosyl)-2,3,4-tri-O-benzoyl- β -D-glucopyranosyl)-3-O-(2,3,4,6-tetra-O-benzoyl- β -D-glucopyranosyl)-2,4-O-benzoyl- β -D-glucopyranosyl)-2,3,4-tri-O-benzoyl- α -D-glucopyranoside (14)

A solution of the glycosyl donor **5** (185 mg, 0.24 mmol), BSP (63 mg, 0.29 mmol) and MS4Å (500 mg) was stirred in dry CH_2Cl_2 (4 mL) at room temperature for 30 min under a nitrogen atmosphere. The reaction mixture was cooled to -78°C , and stirred at the same temperature for 15 min after addition of Tf_2O (0.29 mmol, 49 μL). Then a solution of glycosyl acceptor **6** (109 mg, 0.06 mmol) in dry CH_2Cl_2 (2 mL) was added to the reaction mixture. It was allowed to slowly warm to -40°C with stirring overnight (two-dimensional TLC, hexane–EtOAc, (1 : 1, v/v) and toluene–EtOAc (5 : 1, v/v)), then quenched with Et_3N and diluted CHCl_3 , filtered, washed with sat. aq. NaHCO_3 , and brine. After drying (MgSO_4) and concentration, the residue was purified by repeated column chromatography with hexane–EtOAc (1 : 1, v/v), followed by toluene–EtOAc (10 : 1, v/v) to provide **14** (98 mg, 47%) and hexaglucoisides (72 mg, 42%) as a colorless oil. Compound **14** had $[\alpha]_{\text{D}}^{24.6} = -14.4$ (c 0.13, CHCl_3); ^1H NMR

(600 MHz, CDCl_3): δ 8.15–7.08 (m, 110H, CH_{arom}), 6.10 (t, 1H, $J_{3A',4A'} = 9.9$ Hz, H-3A'), 6.00 (t, 1H, $J_{3B,4B} = 9.9$ Hz, H-3B), 5.79 (t, 1H, $J_{3C,4C} = 9.9$ Hz, H-3C), 5.58 (dd, 3H, $J = 9.3$ Hz, $J = 19.2$ Hz, H-3A'', 3A''', 4A''), 5.44–5.24 (m, 9H, H-2A', 2A'', 2A''', 4A'', 4A''', 2B, 4B, 2C, 4C), 5.18 (t, 1H, $J_{2D',3D'} = 8.8$ Hz, H-2D'), 5.03 (d, 1H, $J_{1A',2A'} = 8.2$ Hz, H-1A'), 4.98 (t, 1H, $J_{4D',5D'} = 9.9$ Hz, H-4D'), 4.97 (d, 1H, $J_{1B,2B} = 3.7$ Hz, H-1B), 4.95 (d, 1H, $J_{1A''',2A'''} = 9.3$ Hz, H-1A'''), 4.93 (d, 1H, $J_{1A'',2A''} = 7.7$ Hz, H-1A''), 4.85 (t, 1H, $J_{2D'',3D''} = 8.8$ Hz, H-2D''), 4.70 (t, 1H, $J_{4D'',5D''} = 8.8$ Hz, H-4D''), 4.65 (d, 1H, $J_{1C,2C} = 8.2$ Hz, H-1C), 4.58 (dd, 1H, $J_{5A',6A'b} = 2.2$ Hz, H-6A'b), 4.41 (dd, 1H, $J_{5A',6A'a} = 5.5$ Hz, $J_{6A'a,6A'b} = 14.8$ Hz, H-6A'a), 4.35 (d, 1H, $J_{1D',2D'} = 7.7$ Hz, H-1D'), 4.37–4.31 (m, 1H, H-5A'), 4.26 (d, 1H, $J_{1D'',2D''} = 7.7$ Hz, H-1D''), 4.26 (t, 1H, $J_{3D',4D'} = 8.8$ Hz, H-3D'), 4.22 (t, 1H, $J_{3D'',4D''} = 8.8$ Hz, H-3D''), 4.19–4.08 (m, 2H, H-6A''a, 6A''b), 4.05–3.93 (m, 3H, H-5A'', 5A''', 6A''a), 3.87 (d, 1H, $J_{5D',6D'b} = 11.5$ Hz, H-6D'b), 3.82–3.71 (m, 5H, H-5B, 5C, 6Ba, 6Bb, 6Da), 3.70–3.52 (m, 4H, H-5D', 5D'', 6Ca, 6D'a), 3.44 (dd, 1H, $J_{5D',6D'b} = 7.7$ Hz, H-6D'b), 3.25 (dd, 1H, $J_{5C,6Cb} = 4.4$ Hz, H-6Cb), 3.17 (dd, 1H, $J_{5B,6Ba} = 3.8$ Hz, $J_{6Ba,6Bb} = 11.3$ Hz, H-6Bb), 3.06 (s, 3H, OMe); ^{13}C NMR (75 MHz, CDCl_3): δ 175.6, 165.9, 165.6, 165.4, 165.3, 165.1, 165.0, 164.9, 164.7, 164.7, 163.9, 134.1, 133.4, 133.0, 132.9, 132.8, 129.7, 129.6, 129.5, 129.4, 129.3, 129.2, 129.1, 129.0, 128.9, 128.7, 128.6, 128.4, 128.2, 128.1, 128.0, 100.9, 100.8, 100.7, 100.4, 100.1, 96.3, 81.6, 73.3, 72.9, 72.8, 72.7, 72.6, 72.0, 71.8, 70.2, 69.9, 69.7, 68.4, 63.2, 55.1, 52.8, 52.7, 51.5, 31.9; ESI-HRMS (m/z) calcd for $\text{C}_{197}\text{H}_{162}\text{O}_{58}\text{Na}^+$: 3477.9625; Found: 3477.9663.

Methyl 6-O-(3-O-(β -D-glucopyranosyl)-6-O-(6-O-(3,6-di-O-(β -D-glucopyranosyl)- β -D-glucopyranosyl)- β -D-glucopyranosyl)- β -D-glucopyranosyl)- α -D-glucopyranoside (1)

To a solution of **14** (14 mg, 4.0 μmol) in MeOH (0.8 mL) and H_2O (0.7 mL) was added NaOMe (28 mg) at room temperature and the mixture was stirred at room temperature for 24 h. The reaction mixture was treated with Amberlite IR-120B ion exchange resin (H^+ form), to give the title compound **1** (3.5 mg, 74%) as a white solid. The spectroscopic data were consistent with the previous reports.^{2c,3} ^1H NMR (600 MHz, D_2O , 300 K): δ 4.64 (d, 1H, $J = 3.6$ Hz), 4.57 (d, 1H, $J = 7.8$ Hz), 4.56 (d, 1H, $J = 7.8$ Hz), 4.39 (d, 1H, $J = 7.2$ Hz), 4.38 (d, 1H, $J = 7.2$ Hz), 4.37 (d, 1H, $J = 7.2$ Hz), 4.35 (d, 1H, $J = 7.8$ Hz), 3.80–3.10 (m, 41H); MALDI-TOF MS (m/z) calcd for $\text{C}_{43}\text{H}_{74}\text{O}_{36}\text{Na}^+$: 1189.4; Found: 1189.9.

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References

- (a) J. K. Sharp, B. Valent and P. Albersheim, *J. Biol. Chem.*, 1984, **259**, 11312–11320; (b) J. K. Sharp, M. McNeil and P. Albersheim, *J. Biol. Chem.*, 1984, **259**, 11321–11336; (c) J. K. Sharp, P. Albersheim and B. Lindberg, *J. Biol. Chem.*, 1984, **259**, 11341–11345.
- (a) S. Yamago, T. Yamada, O. Hara, H. Ito, Y. Mino and J. I. Yoshida, *Org. Lett.*, 2001, **3**, 3867–3870; (b) H. Yamada, H. Takimoto, T. Ikeda, H. Tsukamoto, T. Harada and T. Takahashi, *Synlett*, 2001, 1751–1754; (c) H. Tanaka, M. Adachi, H. Tsukamoto, T. Ikeda, H. Yamada and T. Takahashi, *Org. Lett.*, 2002, **4**, 4213–4216; (d) S. Yamago, T. Yamada, H. Ito, O. Hara, Y. Mino and J. I. Yoshida, *Chem.–Eur. J.*, 2005, **11**, 6159–6174 and references therein.

- 3 R. Verduyn, M. Douwes, P. A. M. van der Klein, E. M. Mosinger, G. A. van der Marel and J. H. van Boom, *Tetrahedron*, 1993, **49**, 7301–7316.
- 4 H. Matsui, J. Furukawa, T. Awano, N. Nishi and N. Sakairi, *Chem. Lett.*, 2000, 326–327.
- 5 (a) D. Crich and M. Smith, *Org. Lett.*, 2000, **2**, 4067–4069; (b) D. Crich and M. Smith, *J. Am. Chem. Soc.*, 2001, **123**, 9015–9020; (c) D. Crich and M. Smith, *J. Am. Chem. Soc.*, 2002, **124**, 8867–8869; (d) D. Crich and L. B. L. Lim, *Org. React.*, 2004, **64**, 115–251; (e) D. Crich, A. Banerjee, W. Li and Q. Yao, *J. Carbohydr. Chem.*, 2005, **24**, 415–424.
- 6 J. D. C. Codée, B. Stubba, M. Schiattarella, H. S. Overkleeft, C. A. A. van Boeckel, J. H. van Boom and G. van der Marel, *J. Am. Chem. Soc.*, 2005, **127**, 3767–3773.
- 7 (a) G. H. Veeneman, S. H. van Leeuwen and J. H. van Boom, *Tetrahedron Lett.*, 1990, **31**, 1331–1334; (b) P. Konradsson, D. R. Mootoo, R. E. McDevitt and B. Fraser-Reid, *J. Chem. Soc., Chem. Commun.*, 1990, 270–272.
- 8 T. Furuike, K. Yamada, T. Ohta, K. Monde and S. I. Nishimura, *Tetrahedron*, 2003, **59**, 5105–5113.
- 9 Z. Zhang, K. Niikura, X. F. Huang and C. H. Wong, *Can. J. Chem.*, 2002, **80**, 1051–1054.
- 10 (a) T. Ercegovic, A. Meijer, U. Ellervik and G. Magnusson, *Org. Lett.*, 2001, **3**, 913–915; (b) A. Meijer and U. Ellervik, *J. Org. Chem.*, 2004, **69**, 6249–6256.
- 11 N. M. Spijker and C. A. A. van Boeckel, *Angew. Chem., Int. Ed. Engl.*, 1991, **30**, 180–183.
- 12 (a) This unprecedented β -selectivity can be explained by the rapid direct S_N2 -like displacement of the intermediate α -triflate. See: D. Crich and W. Cai, *J. Org. Chem.*, 1999, **64**, 4926–4930; (b) J. D. C. Codée, R. E. J. N. Litjens, R. Den Heeten, H. S. Overkleeft, J. H. van Boom and G. A. van der Marel, *Org. Lett.*, 2003, **5**, 1519–1522; (c) J. D. C. Codée, L. J. van den Bos, R. E. J. N. Litjens, H. S. Overkleeft, C. A. A. van Boeckel, J. H. van Boom and G. A. van der Marel, *Tetrahedron*, 2004, **60**, 1057–1064; (d) S. Yamago, T. Yamada, O. Hara, H. Ito, Y. Mino and J. I. Yoshida, *Angew. Chem., Int. Ed.*, 2004, **43**, 2145–2148.