

Domino C–N Bond Formation via a Palladacycle with Diaziridinone. An Approach to Indolo[3,2-*b*]indolets

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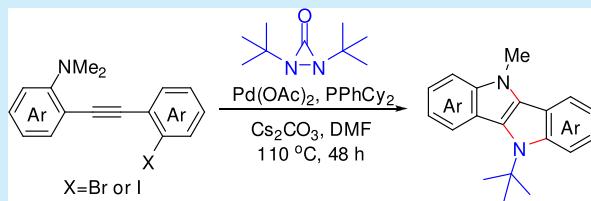
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ABSTRACT: Palladium-catalyzed C–N bond formation is one of the widely used transformations for the synthesis of structurally diverse N-heterocycles. This work describes an efficient palladium-catalyzed multiple-C–N bond formation reaction for the synthesis of highly π -conjugated N-heterocycles, indolo[3,2-*b*]indolets with di-*tert*-butyl diaziridinone. The reaction likely proceeds through the initial formation of an indole-fused palladacycle by nucleophilic aminopalladation and subsequent bisamination to give indolo[3,2-*b*]indolets.



C–N bond formation is an important transformation in organic synthesis.¹ The palladium-catalyzed reaction process plays a crucial role in this field and has constantly received attention. In our own studies, we have found that di-*tert*-butyl diaziridinone (**1**) is a class of versatile reagent for amination. In addition to diamination of olefins (**Scheme 1**, eq a),² it can also effectively react with palladacycles **4** to generate azacycles **6** likely via a Pd(IV) intermediate **5** (**Scheme 1**, eq b).^{3–5} In this bisamination process, two C–N bonds are simultaneously formed.

In general, most reported palladacycles are formed through C–H activation.⁶ To expand the synthetic utility of our bisamination process, there has been search for different ways to generate palladacycles, which could be intercepted in situ by the diaziridinone to form structurally diverse azacycles. Along this line, we have been investigating whether indole-fused palladacycle **9** could be formed via intramolecular nucleophilic aminopalladation of palladium species **8** and could be further intercepted by di-*tert*-butyl diaziridinone (**1**) to form the second indole moiety (**Scheme 1**, eq c). The indole synthesis from 2-alkynylanilines has been extensively studied.^{7,8} For example, when 2-alkynylanilines are treated with ArPdX, an intermolecular nucleophilic aminopalladation leads to the formation of the 3-palladaindolets, which give 3-aryl-substituted indoles upon reductive elimination (**Scheme 2**).⁷ However, an intramolecular version of this process (**8** to **9**) has been unexplored. The envisioned process as described in **Scheme 1** (eq c) could lead to the overall formation of indolo[3,2-*b*]indolets **10**, an important class of molecules that may possess unique electronic properties and could be used for functional materials.^{9,10} Herein, we report our preliminary studies of this subject.

In our initial studies, various nitrogen nucleophiles, including NH₂, NHCOCF₃, NHMs, NHTs, NMeMs, and NMeTs, were examined (**Scheme 3**). The results were somewhat disappointing, and no desired products were

detected. To our delight, when substrate **7a** with NMe₂ as the nucleophile was subjected to the reaction conditions [di-*tert*-butyl diaziridinone (**1**) (1.5 equiv), Pd(PPh₃)₄ (5 mol %), and Cs₂CO₃ (1.5 equiv) in DMF at 110 °C for 48 h], the envisioned indolo[3,2-*b*]indole **10a** was formed in 62% NMR yield (**Table 1**, entry 1). Encouraged by this result, we subsequently screened various ligands with Pd(OAc)₂ as the catalyst (**Table 1**, entries 2–16). PPhCy₂ was found to be the best ligand, giving product **10a** in 95% isolated yield (**Table 1**, entry 11). With PPhCy₂ as the ligand, Pd(OAc)₂ appeared to be the choice of the catalyst (**Table 1**, entries 11 and 17–19). Other bases (**Table 1**, entries 20–22) and solvents (**Table 1**, entries 23–25) were found to be less effective for the reaction. The reaction temperature appeared to be important. A lower yield was obtained when the reaction was carried out at a lower (100 °C) or higher (130 °C) temperature (**Table 1**, entry 26 or 27, respectively).

With an optimized protocol in hand, the substrate scope was subsequently investigated. Bromide **7a'** was also shown to be an effective substrate, while a slightly lower yield was obtained compared to that of iodide **7a** (**Table 2**, entry 1). The bromides were used as substrates in cases in which the corresponding iodides were less accessible. As shown in **Table 2**, the reaction process can be extended to various substituted iodides and bromides, giving a variety of substituted indolo[3,2-*b*]indolets in 52–93% yields. Substrates bearing substituents such as Me, Cl, and CF₃ groups on the phenyl rings containing the halogen (Br or I) were effective under the

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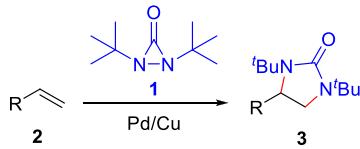
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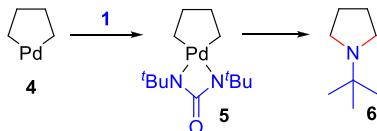
Scheme 1. Diamination and Bisamination with Di-*tert*-butyldiaziridinone

Previous work

(a) Diamination of olefin with di-*t*-butyldiaziridinone

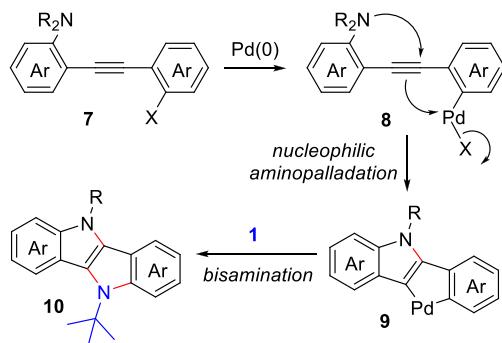


(b) Bisamination of palladacycle with di-*t*-butyldiaziridinone



Present work

(c) Sequential nucleophilic aminopalladation and bisamination



reaction conditions, affording the corresponding products **10b–e** in 70–91% yields (Table 2, entries 2–5, respectively). Substituents like Me, Ph, F, Cl, CN, CO₂Me, and NO₂ groups on the phenyl rings containing the NMe₂ were found to be compatible with the reaction, generating the corresponding indolo[3,2-*b*]indoles **10f–l** in 52–85% yields (Table 2, entries 6–12, respectively). Substrate **7m** with *N,N*-dimethylnaphthalen-1-amine also underwent the reaction smoothly to give product **10m** in 77% yield (Table 2, entry 13). The reaction can also effectively apply to substrates bearing different substituents in both phenyl rings, providing various substituted indolo[3,2-*b*]indoles **10n–u** in 69–93% yields (Table 2, entries 14–21, respectively). A substrate with a diethylamine group (**7v**) was also effective for the reaction, giving product **10v** in 78% yield (Table 2, entry 22).

As exemplified with **7a**, the reaction can be carried out on a gram scale (Scheme 4). The removal of a *tert*-butyl group for **10a** was achieved with trifluoroacetic acid (TFA) and *n*-hexane, affording indolo[3,2-*b*]indole **10a'** in 86% yield (Scheme 5). Gratifyingly, highly conjugated bisindolo[3,2-

Scheme 3. Initial Studies of Nitrogen Nucleophiles

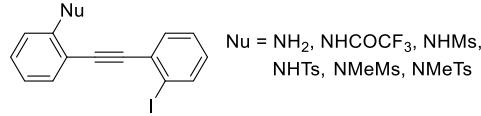
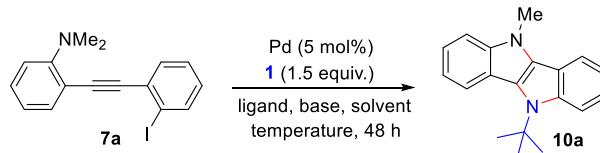


Table 1. Studies of the Reaction Conditions^a



entry	Pd catalyst	ligand	yield (%) ^b
1	Pd(PPh ₃) ₄	—	62
2	Pd(OAc) ₂	PPh ₃	36
3	Pd(OAc) ₂	P(<i>p</i> -tolyl) ₃	33
4	Pd(OAc) ₂	P(<i>o</i> -tolyl) ₃	6
5	Pd(OAc) ₂	P(<i>p</i> -FPh) ₃	46
6	Pd(OAc) ₂	P(<i>p</i> -MeOPh) ₃	41
7	Pd(OAc) ₂	P(C ₆ F ₅) ₃	—
8	Pd(OAc) ₂	P(<i>p</i> -CF ₃ Ph) ₃	41
9	Pd(OAc) ₂	P(<i>o</i> -furyl) ₃	12
10	Pd(OAc) ₂	PCy ₃	31
11	Pd(OAc) ₂	PPhCy ₂	99 (95) ^c
12	Pd(OAc) ₂	PPh ₂ Cy	60
13	Pd(OAc) ₂	dpppe	18
14	Pd(OAc) ₂	dppb	43
15	Pd(OAc) ₂	dppf	25
16	Pd(OAc) ₂	BINAP	11
17	Pd(TFA) ₂	PPhCy ₂	44
18	PdCl ₂	PPhCy ₂	53
19	Pd ₂ (dba) ₃	PPhCy ₂	47
20 ^d	Pd(OAc) ₂	PPhCy ₂	57
21 ^e	Pd(OAc) ₂	PPhCy ₂	37
22 ^f	Pd(OAc) ₂	PPhCy ₂	—
23 ^g	Pd(OAc) ₂	PPhCy ₂	44
24 ^h	Pd(OAc) ₂	PPhCy ₂	23
25 ⁱ	Pd(OAc) ₂	PPhCy ₂	32
26 ^j	Pd(OAc) ₂	PPhCy ₂	50
27 ^k	Pd(OAc) ₂	PPhCy ₂	65

^aAll reactions were carried out with **7a** (0.15 mmol), Pd (0.0075 mmol), ligand (0.015–0.030 mmol; 1:4 Pd:P), di-*tert*-butyldiaziridinone **1** (0.225 mmol), and Cs₂CO₃ (0.225 mmol) in DMF (2 mL) at 110 °C under N₂ for 48 h unless otherwise stated. ^bThe yields were determined by ¹H NMR analysis of the crude reaction mixture using PhCH₂OCH₃ as the internal standard. ^cIsolated yield. ^dWith K₂CO₃. ^eWith Na₂CO₃. ^fWith KOAc. ^gIn MeCN. ^hIn 1,4-dioxane. ⁱIn PhCH₃. ^jAt 100 °C. ^kAt 130 °C.

b]indole **10w** was achieved from **7w** via double-reaction processes (Scheme 6).

Scheme 2. Palladium-Catalyzed Indole Formation from 2-Alkynylanilines

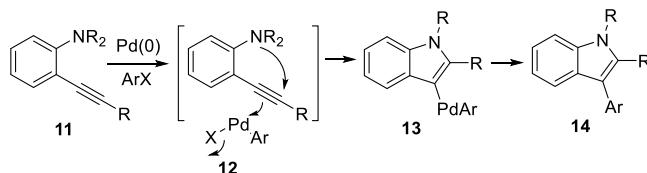
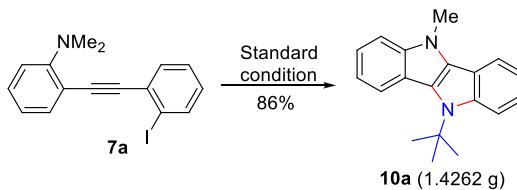
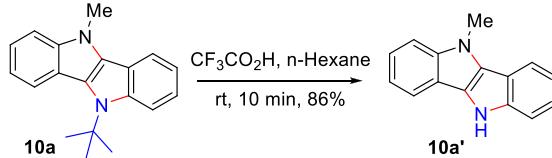
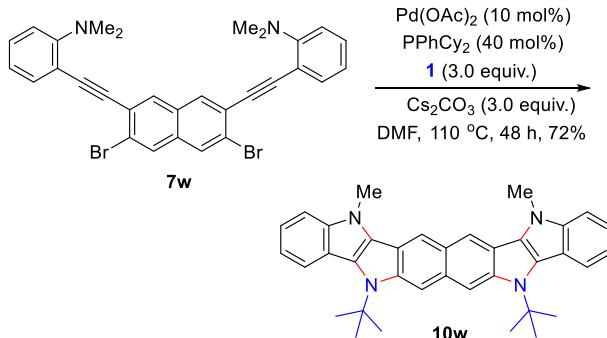


Table 2. Substrate Scope^a

Entry	7	10	Yield ^b
1			93
2			83
3			79
4			91
5			70
6			85
7			84
8			72
9			58
10			81
11			52
12			71
13			77
14			81
15			81
16			83
17			77
18			87
19			89
20			93
21			69
22			78

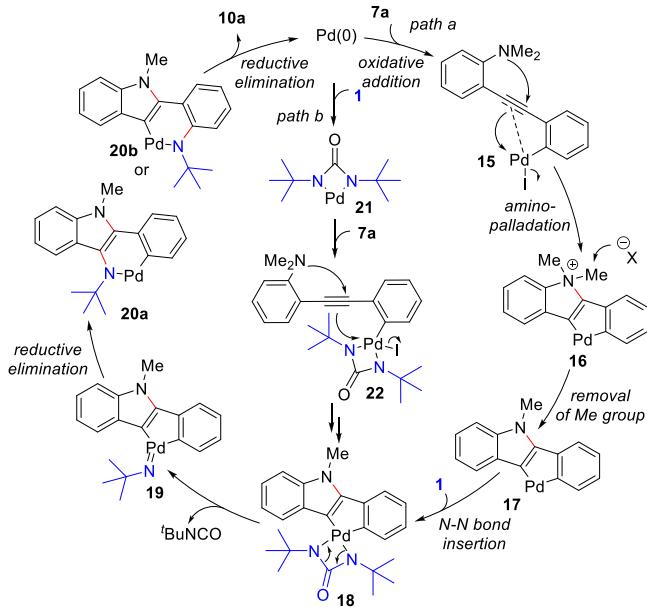
^aAll reactions were carried out with 7 (0.30 mmol), Pd (0.015 mol), PPh₂Cy₂ (0.060 mmol), 1 (0.45 mmol), and Cs₂CO₃ (0.45 mmol) in DMF (3 mL) at 110 °C under N₂ for 48 h. ^bIsolated yield (%)

Scheme 4. Gram Scale Reaction with Substrate 7a

Scheme 5. Removal of the *tert*-Butyl Group of Product 10aScheme 6. Formation of Highly Conjugated Bisindolo[3,2-*b*]indole

A precise understanding of the reaction mechanism requires further study. On the basis of earlier studies,³ a plausible catalytic cycle is proposed in Scheme 7 (with 7a as an example). The process starts with the coordination of the Pd(0) to the substrate (7a), followed by the oxidative addition of the Pd(0) to the C—I bond to give Pd(II) species 15 (path

Scheme 7. A Proposed Catalytic Cycle



a), which undergoes nucleophilic aminopalladation and demethylation to form indole-fused pallada(II)cycle **17**. The oxidative addition of **17** to di-*tert*-butyldiaziridinone (**1**) gives pallada(IV)cycle **18**, which is converted to pallada(IV)nitrene species **19** with release of *t*BuNCO. Indolo[3,2-*b*]indole **10a** is formed from **19** by two consecutive reductive eliminations via **20a** or **20b** with regeneration of the Pd catalyst. Alternatively, the Pd(0) could be oxidatively inserted into the N–N bond of di-*tert*-butyldiaziridinone (**1**) to form four-membered Pd species **21** (path b). The oxidative addition of **21** to substrate **7a** gives Pd intermediate **22**, which is converted to pallada(IV)cycle **18**, upon nucleophilic aminopalladation and subsequent demethylation.

In summary, we have developed an efficient palladium-catalyzed nucleophilic aminopalladation and bisamination sequence with 2-[(2-haloaryl)ethynyl]-*N,N*-dialkylanilines (**7**) and di-*tert*-butyl-diaziridinone (**1**), affording a wide variety of substituted indolo[3,2-*b*]indoles in 52–93% yields. To the best of our knowledge, this is the first example in which an indole-fused palladacycle is formed through intramolecular nucleophilic aminopalladation to form overall two rings with generation of three C–N bonds. Highly conjugated linear bisindolo[3,2-*b*]indole can also be achieved via the aforementioned process. Further exploration of new reaction processes with the indole-fused palladacycle is currently underway.

■ ASSOCIATED CONTENT

SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.orglett.1c00466>.

Experimental procedures, characterization data, and NMR spectra ([PDF](#))

Accession Codes

CCDC 2051396 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

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Notes

The authors declare no competing financial interest.

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