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Drug Annotation

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The Discovery of a Potent, Selective and Peripherally Restricted Pan-Trk Inhibitor (PF-06273340) for the Treatment of Pain

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ABSTRACT

The neurotrophin family of growth factors, comprised of nerve growth factor (NGF), brain derived neurotrophic factor (BDNF), neurotrophin 3 (NT3) and neurotrophin 4 (NT4), is implicated in the physiology of chronic pain. Given the clinical efficacy of anti-NGF monoclonal antibody (mAb) therapies, there is significant interest in the development of small molecule modulators of neurotrophin activity. Neurotrophins signal through the tropomyosin related kinase (Trk) family of tyrosine kinase receptors, hence Trk kinase inhibition represents a potentially "druggable" point of intervention. In order to deliver the safety profile required for chronic, non-life threatening pain indications, highly kinase-selective Trk inhibitors with minimal brain availability are sought. Herein we describe how the use of SBDD, 2D QSAR

models and Matched Molecular Pair data in compound design enabled the delivery of the highly potent, kinase-selective and peripherally restricted clinical candidate **PF-06273340**.

INTRODUCTION

The neurotrophin family of growth factors, comprised of nerve growth factor (NGF), brain derived neurotrophic factor (BDNF), neurotrophin 3 (NT3) and neurotrophin 4 (NT4), is implicated in the physiology of chronic pain. Preclinical and clinical studies have identified a crucial role for NGF in the pathogenesis of inflammatory pain, with neutralizing antibodies demonstrating efficacy in preclinical pain models and in clinical trials for osteoarthritis, chronic lower back pain and interstitial cystitis. BDNF has also been implicated in the pathogenesis of chronic pain, and has been shown to be upregulated in clinical pain states. Given the role of NGF and BDNF in modulating pain, and the clinical efficacy of anti-NGF monoclonal antibody (mAb) therapies, there is significant interest in the development of small molecule oral therapeutics to complement existing treatment options.

Neurotrophins signal through the tropomyosin related kinase (Trk) family of tyrosine kinase receptors, whereby NGF signals preferentially through TrkA, BDNF and NT-4 through TrkB, and NT-3 through TrkC. All neurotrophins interact with equal affinity at the p75 receptor, a member of the TNFR superfamily. An alternative point of intervention, high in the neurotrophin signaling cascade and amenable to small molecule therapy, could therefore be the inhibition of intracellular Trk kinases (Figure 1).⁷⁻⁸

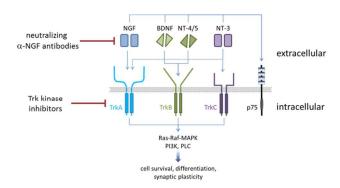


Figure 1. Neurotrophins NGF, BDNF, NT-3, NT-4/5 and their receptors TrkA, TrkB and TrkC.

A key concern in the development of Trk kinase inhibitors for the treatment of chronic pain is the potential risk of central nervous system (CNS) adverse events. TrkB is expressed throughout the CNS with the BDNF/TrkB axis involved in excitatory signaling, long-term potentiation and feeding behavior. 9-12 TrkA is highly expressed in the cholinergic neurons of the basal forebrain. with TrkA gene ablation leading to dysfunction in cholinergic circuitry in preclinical species. 13-15 In addition, clinical CNS side-effects have been noted with 1-(5-(4-amino-7-isopropyl-7Hpyrrolo[2,3-d]pyrimidine-5-carbonyl)-2-methoxyphenyl)-3-(2,4-dichlorophenyl) (CE-245677), an oral pan-Trk/Tie2 kinase inhibitor identified by Pfizer, previously in development for the treatment of certain cancers. 16 Phase I multiple dose trials were stopped due to the development of significant CNS adverse events which included cognitive deficits, personality These effects fully resolved upon cessation of dosing. changes and sleep disturbances. Preclinical studies with this compound have shown it to be efficacious in multiple pain models. However, at doses associated with significant (> 50%) CNS Trk receptor occupancy, changes in cortical electroencephalography (EEG), cognitive function, body weight and hypothalamic mRNA were observed.¹⁷ These findings were recapitulated with pan-Trk inhibitors devoid of Tie-2 kinase activity (data not shown). These data confirm the potential of small molecule Trk

kinase inhibitors as pain therapeutics, but highlight the risk of on-target adverse events resulting from pharmacologically relevant exposure in the CNS.

Given the therapeutic efficacy of brain restricted anti-NGF mAb therapies, and the known safety risks associated with CNS Trk receptor occupancy, a rational approach to delivering a welltolerated Trk kinase therapeutic is to restrict compounds to the peripheral compartment. It is known from projects within Pfizer and the wider pharmaceutical community that it is possible to deliver small molecules that are orally bioavailable yet peripherally restricted. 18-22 The design of compounds in physicochemical space appropriate for absorption across the gastrointestinal (GI) epithelium (e.g. molecular weight < 500, polar surface area < 140, < 10 rotatable bonds)²³ but that are substrates for blood-brain barrier (BBB) efflux transporters such as P-glycoprotein (Pgp) and Breast Cancer Resistance Protein (BCRP) is an emerging strategy to deliver this type of profile. According to the free drug hypothesis, unbound concentration determines a drug's pharmacological effect. If a drug is a substrate for BBB efflux transporters, the unbound brain concentration $(C_{\rm b,u})$ may be less than the unbound plasma concentration $(C_{\rm p,u})$. An example of a compound that exhibits significant peripheral restriction (via targeting BBB efflux transporters P-gp and BCRP) whilst retaining good oral bioavailability, cell penetration and pharmacological activity is the tyrosine kinase inhibitor imatinib.²⁴ Imatinib was not specifically designed to be peripherally restricted, yet discoveries such as this engender knowledge of how to prospectively design orally bioavailable, peripherally restricted molecules in a rational manner.

Targeting transporters in order to achieve peripheral restriction can introduce risk factors.²⁰ For example, the apical membrane of intestinal epithelial cells contains the same efflux transporter

proteins P-gp and BCRP found at the BBB. Drugs that are efflux substrates could therefore be restricted in their passage across the intestinal epithelium resulting in poor absorption from the gut lumen (Figure 2). Set against this risk (at commonly prescribed oral drug doses of adequately soluble compounds (10 - 500 mg)) the intraluminal compound concentration is likely to be in the range of hundreds of micromolar providing a high concentration gradient which, for a compound with sufficient permeability, will mean a high driving force for flux across the epithelium. In addition such gut luminal concentrations are likely to be high enough to saturate efflux transporters, given that substrates of P-gp typically possess Km values in the range 1 – 100 μM. These considerations suggest that in most cases, the risk of poor absorption and non-linear pharmacokinetics of orally delivered efflux substrates will be low.²⁵

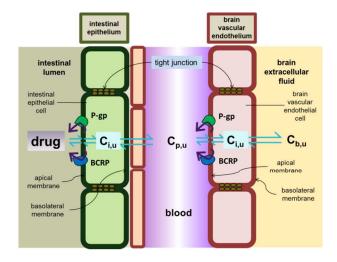


Figure 2. Schematic illustration of orally administered drug absorption across the intestinal epithelium and distribution across the brain vascular endothelium. $C_{p,u}$ = unbound plasma concentration, $C_{b,u}$ = unbound brain concentration and $C_{i,u}$ = unbound intracellular concentration.

An additional requirement for a small molecule Trk kinase inhibitor, suitable for chronic administration in a non-life threatening indication, is exquisite kinase selectivity. Kinases modulate numerous physiological functions, and hence inhibiting off-target kinases can lead to

unwanted safety side effects. 26-27 The majority of kinase inhibitors bind to the highly conserved ATP binding site (Type I inhibitors), and hence can be non-selective. A strategy that has emerged to design kinase inhibitors with enhanced kinase selectivity is to target protein-ligand interactions in less conserved areas of the ligand binding domain, outside of the ATP binding pocket. 28-31 Imatinib, a Type II kinase inhibitor, is a well-established example of this whereby it binds to the inactive form of the Abl protein in which the N-terminal region of the activation loop adopts a Asp-Phe-Gly-out (DFG-out) conformation (see supplementary information S2). Since the deep pocket region accessible in a DFG-out conformation has a relatively low sequence conservation compared to other kinases, imatinib has been shown to be a relatively selective kinase inhibitor. 30-32 Indeed, when Karaman et al. generated KINOMEscan selectivity scores for a selection of marketed kinase inhibitors, they demonstrated the potential for Type II inhibitors such as imatinib, sorafenib and lapatinib to display higher selectivity than Type I inhibitors such as sunitinib and dasatinib, especially when only higher affinity off-target interactions were considered. 33-34 Whilst kinase inhibitors adopting a DFG-out conformation are certainly not guaranteed to be highly kinome-selective, 35-36 this binding mode offers the opportunity to access an allosteric binding pocket adjacent to the ATP-binding site and target unique inactive conformations.³⁷

Herein we describe the discovery of a highly potent, kinase-selective, peripherally restricted, efficacious and well-tolerated series of "DFG-out" pan-Trk inhibitors. The use of SBDD, 2D QSAR models and Matched Molecular Pair data in prospective design enabled rapid optimization of the lead chemical series and delivery of clinical candidate **14** (**PF-06273340**). 38

RESULTS AND DISCUSSION

Hit identification and characterization

A high throughput screen of the Pfizer compound collection was conducted using TrkA and TrkB cell-based assays.³⁸ Of the numerous chemical series of pan-Trk inhibitors identified, prioritization for lead optimization follow-up was based on criteria that included LipE,²³ predicted levels of kinase promiscuity,³⁹ and predicted ATP-site binding mode. Chemical series with an anticipated DFG-out binding mode were prioritized as it is known that kinase inhibitors that bind to less conserved areas of the ATP binding site can be more selective within the kinome. Indeed, when a systematic analysis of kinase structural and kinase panel screening data was undertaken within Pfizer, it was shown that ligands that contact ATP-site residues approaching the "back pocket" of the ATP site have a higher chance of being kinase-selective.⁴⁰ Assessment of compound binding mode was made *via* docking into a TrkA homology model built on a set of in-house cFMS crystal structures.

Several chemical series were identified that demonstrated moderate to good LipE profiles, were calculated to have high levels of kinase selectivity and predicted to adopt a DFG-out binding mode. One of the series selected for lead optimization was the pyrrolopyrimidine urea series.⁴¹ An example from this series, 1,⁴¹ is highlighted in Figure 3 and Table 1. Compound 1 is a potent inhibitor of TrkA, TrkB and TrkC in recombinant cellular assays.³⁸ Trk isoform selectivity is not anticipated within this series due to the highly homologous nature of the TrkA, B and C DFG-out ligand binding sites.⁴² The overall kinase selectivity profile of 1 is encouraging (Figure 12), with only 5 kinases (including TrkA) in a panel of 39 biochemical kinase assays from Invitrogen showing > 50 percent inhibition (% I) at a compound concentration of 1 μM. The kinase

selectivity profile of **1** is represented as a Gini coefficient which describes the selectivity of a compound by a single number between 0 and 1. Nonselective inhibitors are characterized by Gini values close to zero (e.g. Staurosporine, Gini 0.15), whereas compounds with complete selectivity exhibit Gini values of 1.⁴³ Compound **1** has a Gini value of 0.59, indicative of its moderate kinase selectivity, representing an encouraging start-point for further optimization.

Compound 1 was predicted to adopt a DFG-out binding pose in the TrkA homology model, which was subsequently confirmed by X-ray crystallography studies. Figure 3 shows the crystal structure of 1 bound to the TrkA kinase, and highlights the key protein-ligand interactions. The 4-amino pyrrolopyrimidine motif of 1 makes a two-point hydrogen bonding interaction with backbone polar atoms of hinge residues Glu590 and Met592, whilst the central pyridyl group of 1 makes a π-stacking interaction with the gatekeeper (Phe589), and Phe669 of the DFG triad. The carbonyl oxygen of the urea group of 1 makes a hydrogen bond interaction with the backbone N-H of Asp 668 from the DFG triad. The N-H groups of the urea group of 1 make hydrogen bond interactions with a conserved water molecule, which further interacts with catalytic lysine (Lys544) and Glu560 from the alpha-C helix. The lipophilic "DFG-out" back pocket accommodates the 1,3-difluorophenyl group.

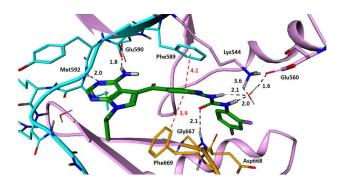


Figure 3: Crystal structure of 1 (green) bound to TrkA (pink). The hinge region of the TrkA ATP site is highlighted in blue. The "DFG" triad of the activation loop of TrkA is highlighted in orange. Key protein-ligand binding interactions are highlighted.

Although potent, the lipophilic efficiency of **1** is moderate (LipE 3.4) due to the highly lipophilic nature of the molecule (logD 4.6). The metabolic stability of **1** in human liver microsomes (HLM) and human hepatocytes (hHep) is encouraging, and may be high enough to deliver low metabolic clearance (CL) in humans. Also encouraging is the lack of blockade of the human ether-a-go-go related gene (hERG) encoded potassium channel, as demonstrated by the lack of signal in the dofetilide binding assay. Compound **1** is a weak substrate for the efflux transporter P-gp (efflux ratio (ER) in the MDCK-MDR1 assay = 2.1), is poorly soluble (< 0.3 μ M at pH 6.5), and has low passive permeability (RRCK A to B apparent permeability (P_{app}) = 0.6 x 10⁻⁶ cms⁻¹), such that limitations on oral bioavailability may be expected in the absence of formulation enhancement.

Hit to lead: Improving solubility

After identification of the pyrrolopyrimidine series (e.g. 1), an initial round of synthesis aimed at improving the aqueous solubility profile was undertaken. It is known that urea-containing compounds often have a poor solubility profile, and a commonly used strategy to try and improve solubility is to replace the urea group with an amide unit. Indeed, a Matched Molecular Pair (MMP) analysis of the Pfizer Pairwise Database⁴⁶ showed that conversion of an urea to an amide group (R₁-NHCONH-R₂ \rightarrow R₁-NHCOCH₂-R₂) increased solubility (> 2-fold) in 52% of cases, maintained solubility within 2-fold in 39% of cases and decreased solubility (> 2-fold decrease) in only 9% of cases, with a mean increase in solubility of 137 μ M (no. of pairs = 56).⁴⁷

Replacing the urea group for an amide unit in the pyrrolopyrimidine series maintained TrkA potency and on average facilitated a > 30 fold increase in solubility. In addition, the passive permeability of the amide-containing analogues was superior. An example, 2 (the direct matched pair of 1) is highlighted in Table 1. Due to the superior solubility and permeability profile, all further ligand optimization was conducted using the amide linker group.

Hit to lead: Improving kinase selectivity

Whilst the initial kinase selectivity profile of the DFG-out pyrrolopyrimidine series was promising, an early design objective was to further improve the kinase selectivity to optimize the overall safety profile. Kinase inhibitors generally form one to three hydrogen bonds with backbone polar atoms of the kinase hinge region that connects the N- and C-terminal lobes of the catalytic domain, with an increasing number of hydrogen bonds thought to negatively affect overall kinase selectivity. 48 Indeed, when a systematic analysis of kinase scaffolds extracted from the Pfizer crystal structure database (CSDb) was performed and combined with kinase biological screening data, the authors demonstrated that more hydrogen bonding interactions at the hinge region do not necessarily afford increased potency, and moreover may have deleterious effects on kinome selectivity. ⁴⁹ A strategy to optimise kinase selectivity could therefore be to minimise the number of hydrogen bonding interactions at the hinge region. Compound 3 is a representative amino pyrrolopyrimidine from the initial round of synthesis described above in which the urea to amide linker transformation was investigated (Table 1). Compound 3 is a potent inhibitor of TrkA, TrkB and TrkC, with a moderate lipophilic efficiency profile (LipE 4.0). The metabolic stability of 3 in HLM and hHep is moderate and the hERG activity modest. Compound 3 is a relatively good substrate for the efflux transporter P-gp (MDCK-MDR1 ER =

6.2), is a weak substrate for the efflux transporter BCRP, and has moderate passive permeability (RRCK A to B $P_{app} = 15.3$). The aqueous solubility is 13.2 μ M. The kinase selectivity profile of 3 is promising, with only 7 kinases (including TrkA) in a panel of 39 biochemical kinase assays from Invitrogen showing > 50 % inhibition at a compound concentration of 1 μ M, generating a Gini value of 0.59 (Figure 12).

In order to design compounds with the minimal number of hydrogen bond contacts at the hinge region, an analysis of the crystal structure of **3** bound to TrkA and the non-liganded *apo* structure was undertaken. The crystal structure of **3** bound to TrkA is shown in Figure 4. Compound **3** adopts a DFG-out binding mode, with the amino pyrrolopyrimidine motif making a two-point hydrogen bonding interaction with hinge residues Glu590 and Met592. When **3** is superpositioned onto the TrkA *apo* crystal structure, the -NH₂ motif of the amino pyrrolopyrimidine group overlays directly with a conserved water molecule observed in the *apo* structure (Figure 5). It was hoped that removal of the –NH₂ motif as highlighted in Figure 5 would deliver a ligand that makes a single hydrogen bonding interaction with the backbone N-H of hinge residue Met592, with a water molecule replacing the –NH₂ group, satisfying the backbone carbonyl oxygen at Glu590, and with the overall effect of improving kinase selectivity.

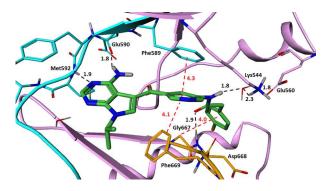


Figure 4: Crystal structure of **3** (green) bound to TrkA (pink). The hinge region of the TrkA ATP site is highlighted in blue. The "DFG" triad of the activation loop of TrkA is highlighted in orange. Key binding interactions are highlighted.

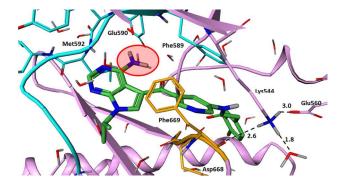


Figure 5: Superposition of **3** (green) onto the TrkA *apo* crystal structure (pink), all water molecules included. The hinge region of the TrkA ATP *apo* site is highlighted in blue. The "DFG" triad of the activation loop of TrkA is highlighted in orange. The red circle highlights the -NH₂ motif of **3** that overlays directly onto a conserved water molecule observed in the *apo* structure

Deleting $-NH_2$ at the 4-position affords ligands with comparable TrkA activity, and a much improved kinase selectivity profile. An example, 4, 38 is highlighted in Table 1. Compound 4 is a potent pan-Trk inhibitor with moderate LipE and moderate turnover in HLM and hHep. Removal of the amino group from the pyrrolopyrimidine unit does not affect the propensity of the chemotype to be a substrate for P-gp, with 4 having an ER in the MDCK-MDR1 assay of 8.2. As with previous pyrrolopyrimidine amide analogues, the passive permeability of 4 is moderate (RRCK $P_{app} = 9.9$) and the aqueous solubility relatively low (17.5 μ M). A significant difference between 4 and its amino pyrrolopyrimidine analogue 3 is in the kinase selectivity profile. Whereas 3 strongly inhibits (> 50% I) 7 kinases (including TrkA) in an Invitrogen panel of 39 biochemical kinase assays (Gini score 0.59), 4 is much more selective (Gini score of 0.82), Figure 12.

The TrkA co-crystal structure of *des*-amino pyrrolopyrimidine **5**,³⁸ a close-in analogue of **4** in which the *p*-Cl phenyl group is replaced with a *p*-CN phenyl, is shown in Figure 6. Compound **5** adopts a DFG-out binding mode, and in general makes similar protein-ligand binding interactions as amino pyrrolopyrimidine **3** (Figure 4). A notable difference is the interaction network around the hinge region. The amino pyrrolopyrimidine motif of **3** makes a two-point hydrogen bonding interaction with backbone polar atoms of hinge residues Glu590 and Met592. As Figure 6 highlights, the hydrogen bond interaction between the pyrimidyl nitrogen of **5** and the backbone N-H of Met592 is maintained, but a conserved water molecule replaces the –NH₂ group to make a hydrogen bond interaction that bridges the main chain carbonyl Glu590 and the ketone oxygen atom of **5**.

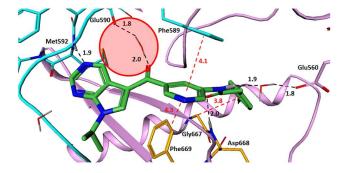


Figure 6: Crystal structure of **5** (green) bound to TrkA (pink). The hinge region of the TrkA ATP site is highlighted in blue. The "DFG" triad of the activation loop of TrkA is highlighted in orange. Key protein-ligand binding interactions are outlined. The red circle highlights the hydrogen bond interactions made by a conserved water molecule bridging the main chain carbonyl Glu590 and the ketone oxygen atom of *des*-amino pyrrolopyrimidine **5**.

The presence of a bridging water between the main chain carbonyl oxygen of TrkA Glu590 (or the equivalent across other kinases) and a polar atom of a bound ligand is a relatively rare occurrence. The available kinase crystal structures from the Protein Data Bank (PDB) and the

Pfizer crystal structure database (CSDb) were searched to identify similar water-bridging examples using Interaction Miner¹; an in-house tool to search structural databases for specific interactions and structural motifs.⁵⁰ This search retrieved < 10 examples from other kinases within the RMSD threshold of < 1.0 Angstrom to the matching atoms. Hence, as replacing the direct H-bond with a "through water" H-bond maintained Trk activity and improved kinase selectivity, follow-up Trk analogues within pyrrolopyrimidine series were devoid of an –NH₂ group at the *4*-position.

Compound	1	2	3	4
	HN NH ₂ NH ₂ N	F HN O	NH ₂ N	HN O CI
Trk A/B/C IC ₅₀	2/5/4	3/ND/ND	6/2/1	12/4/3
(nM)				
MW/logD	451/4.6	450/4.3	449/4.0	434/3.7
LipE	4.1	4.2	4.3	4.2
HLM	18	23	27	22
$(\mu L/min/mg)$				
hHeps	12	ND	12	37
$(\mu L/min/mill)$				
Dof Ki/hERG IC ₅₀ (μM)	> 40/ND	8.3/ND	10.8/7.8	5.2/ND

Interaction Miner is a Pfizer tool to search structural databases for specific interactions and structural motifs. Briefly, Interaction Miner identifies matching interactions or structural features based on user-defined atom and bond match criteria as well as geometric constraints. The matching database structures are superposed to the query using RMSD overlay of the matching substructure, enabling users to visually analyze database hits in the context of the query. The RMSD threshold is exposed as an additional filtering criterion to restrict the database hits based on their geometric similarity to the query substructure.

P-gp/ BCRP ER	2.1	10.9	6.2/1.2	8.2/1.5
RRCK A-B P _{app}	0.6	24.1	15.3	9.9
(x 10 ⁻⁶ cms ⁻¹)				
Solubility (μM)	< 0.3	13.6	13.2	17.5
Gini score	0.59	ND	0.59	0.82

Table 1: Structure and properties of compounds 1 to 4. ND = not determined

Peripheral restriction: Building an in vitro to in vivo correlation

The initial rounds of compound design (as discussed in the previous section) generated a number of potent and selective pan-Trk inhibitors that displayed a range of ER in the MDCK-MDR1 assay, demonstrating a spectrum of ability to act as substrates for the efflux transporter P-gp. In order to build an understanding of the *in vitro* P-gp activity required to deliver significant peripheral restriction in vivo, brain distribution experiments were conducted in rats. Compounds with a P-gp ER in the range of 1 - 50 were administered orally at a dose of 100 mg/kg and terminal plasma and brain samples were taken 3 hours after dosing. Compound concentrations in plasma and homogenized brain samples were quantified in parallel with assessment of binding of compounds to plasma and brain tissue, such that unbound concentration ratios $(C_{b,u}/C_{p,u})$ could be determined.⁵¹ The data from the brain penetration experiments are shown in Figure 7A. Increasing P-gp ER results in decreased brain penetration, such that compounds with a P-gp ER > 10 are significantly excluded from the CNS ($C_{\rm b,u}$ / $C_{\rm p,u}$ < 0.05). In addition, the occupancies of TrkB and TrkC receptors in brain samples were determined using the chemical proteomic technology developed by ActivX Biosciences. This approach utilizes biotinylated acyl phosphates of ADP or ATP that transfer biotin to conserved lysine residues within the ATP binding site of kinases. Binding of a test compound reduces binding of the probes to individual kinases, which when combined with kinase proteomic analysis, allows receptor occupancy and

broad kinase selectivity in complex tissue lysates to be determined.^{52,53} Figure 7B shows the kinase receptor occupancy profile of **3** and **6** in rat brain lysates following oral administration of compound as determined using the ActivX biotinylated acyl phosphate ADP probe. The x-axis of Figure 7B represents the individual kinases detected in the rat brain lysate panel using the ActivX ADP probe, the y-axis represents the fold change in kinase receptor occupancy relative to vehicle-treated animals. Compound **6** (shown as a red circle in Figure 7B) has a high P-gp ER, is significantly restricted from the CNS (P-gp ER = 27.0, $C_{b,u}$ / $C_{p,u}$ = 0.013) and has no detectable TrkB or TrkC occupancy in brain lysates. Compound **3** (shown as a blue circle in Figure 7B) has a lower P-gp ER which translates into higher brain availability (P-gp ER = 6.2, $C_{b,u}$ / $C_{p,u}$ = 0.22). Compound **3** causes marked blockade of binding of the chemical proteomic probe to TrkB and TrkC within the CNS (TrkA not detected due to low expression), but does not display significant occupancy of any of the other kinases in the brain lysate panel, demonstrating the high selectivity of this ligand for TrkB and TrkC over other kinases.

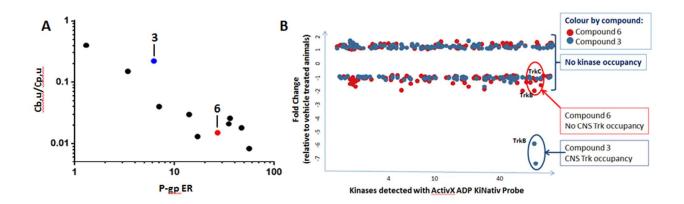


Figure 7A: Relationship of P-gp ER in the MDCK-MDR1 assay vs. rat $C_{b,u}$ / $C_{p,u}$ for a range of pan Trk compounds from the pyrrolopyrimidine series. Figure 7B: Kinase receptor occupancy profile of **3** and **6** in rat brain lysates determined using the ActivX biotinylated acyl phosphate ADP probe.

Peripheral restriction: Use of 2D QSAR models to guide design

As highlighted by the rat brain availability data in Figures 7A and 7B, a compound with a high P-gp ER should be significantly restricted from the CNS. In order to guide the prospective design of compounds with a high P-gp ER on the Trk project, a 2D QSAR regression model built on the Pfizer P-gp dataset was utilised.⁵⁴ The model predicts a P-gp ER value for proposed compounds, along with a confidence value that incorporates both the chemical similarity and activity landscape of a test compound's neighbourhood, calibrating the value to a probability that the prediction will be within 2-fold of the actual experimental result. To assess the performance of the QSAR model and confidence metric, prospective predictions of P-gp ER over time can be evaluated by comparing the model predictions made prior to testing of the compound vs. the actual experimental P-gp data generated once tested. The result of this analysis is shown in Figure 8. The y-axis of the trellised pie-chart in Figure 8 shows the binned P-gp ER predictions (low predicted efflux; P-gp ER < 5, high predicted efflux; P-gp ER > 5) and the x-axis shows the confidence values (0.0 to 1.0) of the P-gp ER model predictions. A cut-off of P-gp ER > 5 (cf. the stricter cut-off of P-gp ER > 10) was chosen so as not to omit potentially interesting molecules from consideration given that the model does not have perfect predictivity. The pies are coloured by real experimental P-gp ER data (cerise: P-gp ER < 5, blue: P-gp ER > 5). As highlighted in Figure 8, as the confidence in the model prediction increases, the probability of achieving the predicted P-gp ER outcome increases. In the case of pan-Trk, the design strategy was to prioritise compounds predicted to have high efflux (with high model confidence) for synthesis, disregard compounds predicted to have low efflux (with high model confidence), and more sparingly synthesise compounds in the space for which the P-gp ER QSAR model confidence was not high. Utilising TrkA co-crystal structure information alongside the 2D QSAR P-gp ER models to prioritise design ideas enabled an efficient triage process in which 69% of compounds in the Hit to Lead phase met TrkA potency (< 100 nM) and P-gp ER (ER > 5) criteria.

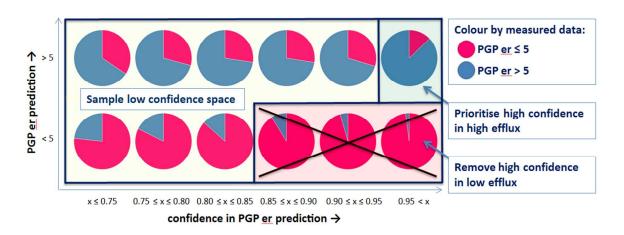


Figure 8: Binned 2D QSAR P-gp ER prediction data vs. Binned confidence in 2D QSAR P-gp ER prediction. Pies are coloured by real experimental P-gp ER data (cerise: P-gp ER < 5, blue: P-gp ER > 5).

Lead optimization

In addition to utilizing 2D P-gp QSAR model data (*vide supra*) to guide design of compounds with a high P-gp ER, design efforts in the lead optimization phase focused on increasing compound metabolic stability and lipophilic efficiency. In order to increase metabolic stability, *in vitro* metabolic identification (MetID) data were used to identify metabolic "soft spots". Metabolite profiling over a range of pyrrolopyrimidine analogues in human liver S9 fractions (in the absence of the CYP cofactor NADP) identified Aldehyde Oxidase (AO) as a likely source of metabolic oxidation. The metabolic liabilities of representative *des*-amino pyrrolopyrimidines and amino pyrrolopyrimidines **5** (HLM < 9.8 μ l/min/mg protein, hHep 13.0 μ l/min/10⁶ cells) and **6** (HLM 21.6 μ l/min/mg protein, hHep 6.6 μ l/min/10⁶ cells) are

highlighted in Figure 9 (full HPLC-UV chromatograms are available in supplementary information S3 and S4). Amino pyrrolopyrimidine 6 has a small amount of oxidation on the amino pyrrolopyrimidine hinge binding group and the central pyridine substituent. Conversely, significant metabolism is observed at multiple sites of des-amino pyrrolopyrimidine 5, the major sites being the pyrrolopyrimidine hinge binding group and the central pyridine substituent. A very minor product of amide hydrolysis was also observed. Incubation of structurally similar analogue (R)-2-(4-cyanophenyl)-N-(4-(7-(1hydroxypropan-2-yl)-7H-pyrrolo[2,3-d]pyrimidine-5-carbonyl)pyridin-2-yl)acetamide 05302191)⁵⁵ in human liver S9 (no additional NADP) showed formation of a mono-oxidised metabolite (see supplementary information S5). In the presence of 50 μM menadione (an AO inhibitor), a dramatic reduction in the formation of this metabolite was observed, whereas preincubation with 50 µM allopurinol (XO inhibitor) showed no change in metabolite profile. These observations would suggest that metabolism in human liver S9 fractions in this series is most likely due to AO.

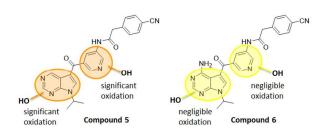


Figure 9: Sites of AO-mediated oxidation of des-amino pyrrolopyrimidines 5 and amino pyrrolopyrimidine 6.

Over the last few years, AO has been increasingly recognized as playing an important role in drug metabolism, with reported examples in which metabolism by this enzyme has had significant clinical impact or led to the termination of a drug development program. In drug discovery, a method of predicting CL parameters in humans with high confidence is highly desirable to make accurate dose predictions and, whilst methods of scaling *in vitro* intrinsic CL data to predict the *in vivo* metabolic CL of compounds are well-established for CYP

P450-dominated CL, similar methods have not yet been established for drugs metabolized by AO. As a result, the impact of having AO-mediated metabolism cannot be fully established until the compound has reached the clinic. Numerous medicinal chemistry design strategies aimed at avoiding AO metabolism have been reported.⁵⁷ As highlighted in Figure 9, the AOmediated oxidation of the des-amino pyrrolopyrimidine group is more significant than for the 4-amino pyrolopyrimidine motif. However, as the des-amino pyrrolopyrimidine confers an enhanced kinase selectivity profile, strategies to minimise AO oxidation on the more kinaseselective des-amino pyrrolopyrimidine template were investigated. The first step in an AOmediated oxidation involves a nucleophilic attack on a heterocyclic ring system by the molybdenum pyranopterin cofactor MoCo,⁵⁸ hence an initial strategy to minimize AO liability on the des-amino pyrrolopyrimidine template was to increase the electron density of that group by replacing the pyrimidine with a pyridyl motif. Gratifyingly, the pyrimidyl to pyridyl transformation generated compounds that maintained potent Trk activity, high kinase selectivity and were devoid of AO-mediated oxidation on the pyridyl-containing hinge binding moiety. An example of from the pyrrolopyridine series, 7,⁵⁹ is highlighted in Table 2.

Compound 7 is a potent pan-Trk inhibitor, with an improved lipophilic efficiency of LipE 4.9 due to a reduced logD relative to previous analogues. Compound 7 has low metabolic turnover in HLM and hHep, is a good substrate for both the efflux transporters P-gp (ER = 29) and BCRP (ER = 13), and has moderate passive permeability (RRCK $P_{app} = 11.3 \times 10^{-6} \text{ cms}^{-1}$). The aqueous solubility of 7 is 7.5 μ M. Compound 7 has an exquisitely selective kinase selectivity profile, with a Gini score of 0.96 (Figure 12). However, the introduction of the weakly basic pyridyl atom (pKa 6.5) as part of the pyrrolopyridyl ring engenders increased hERG activity (functional hERG IC₅₀ = 1.6 μ M). Incubation of 7 in human S9 (no

added NADP) showed no evidence of AO-mediated oxidation on the pyrrolopyridine hinge binding group, and a low level of metabolism of the central pyridine substituent (HPLC-UV chromatograms are available in supplementary information S6). Interestingly, AO-mediated metabolism on the central pyridyl ring is ablated if the N-atom is located at the 2'- or 4'-position (Table 2), in these cases however compounds are subject to chemical instability (if the nitrogen atom is located at the 2'- position) or significant GSK3 β kinase activity (if the nitrogen atom is located at the 4'- position). Hence, the 3'-pyridyl ring was established as the best isomer with which to continue the lead optimization process.

Compound	5	6	7
	HN O CN	HN O CN	HN O 2 2 1 N N CF3
Trk A/B/C IC ₅₀	62/42/ND	73/8/ND	13/7/5
(nM)			
MW/logD	424/2.6	439/2.5	456/3.0
LipE	4.6	4.6	4.9
HLM	< 9	22	< 8
(μL/min/mg)			
hHeps	13	7	12
(μL/min/mill)			
Dof Ki/hERG	7.1/ND	9.7/ND	1.4/1.6
IC ₅₀ (μM)			
P-gp/ BCRP ER	31/ND	27/ND	29/13
RRCK A-B P _{app}	19.1	12.1	11.3
(x 10 ⁻⁶ cms ⁻¹)			
Solubility (µM)	5.2	4.4	7.5
Gini score	ND	ND	0.96

Table 2: Structure and properties of compounds 5 to 7. ND = not determined

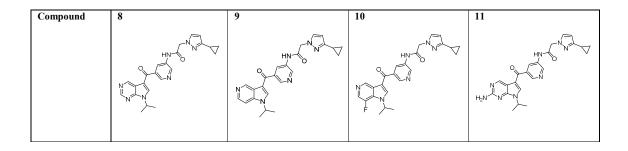
The pharmacokinetic and brain penetration profile of 7 was assessed *in vivo*. Intravenous (i.v.) and oral (p.o.) dosing in rats was conducted at 1 mg/kg and 3 mg/kg respectively. Compound 7 has a high systemic CL (blood CL 50 mL/min/kg, \sim 70% hepatic blood flow), moderate volume of distribution (V_{ss} 1.5 L/kg), short half-life (2.3 h), and a relatively low bioavailability (F% 16). The calculated fraction absorbed (0.55) was however encouraging, suggesting that active transport may not significantly impede oral absorption in this series. To determine the extent of peripheral restriction, 7 was administered orally at 100 mg/kg, with terminal plasma and brain samples taken 3 hours post dose. This study demonstrated a very low unbound brain/plasma concentration ratio ($C_{b,u}$ / $C_{p,u}$) of 0.0083. In addition, Trk receptor occupancies in brain lysates were assessed using the ActivX biotinylated acyl phosphate ADP probe. No measurable TrkB or TrkC occupancy was detected.

The ability of 7 to reverse pain responses *in vivo* was assessed in an ultraviolet (UV) irradiation induced hyperalgesia (UVIH) model of inflammatory pain (see supplementary information, S7). Analysis of the associated plasma exposure of 7 indicated that statistically significant efficacy was observed at unbound plasma concentrations of ≥ 1 x TrkA IC₅₀ with a maximal efficacy response at unbound plasma concentration of ~ 10 x TrkA IC₅₀. Compound 7 was well tolerated in a 7-day rodent toxicology study up to the highest dose tested of 400 mg/kg/day. Thus we demonstrate that 7 is a potent and selective inhibitor of Trk tyrosine kinases, is orally absorbed and highly peripherally restricted, exhibits robust antinociceptive effects in the UVIH model of inflammatory pain, and is well tolerated *in vivo*. Whilst the hERG liability of 7 was not optimal, the promising data generated on 7 enabled the project team to enter candidate-seeking mode.

Candidate-seeking

As outlined above, the AO-liability of the pyrrolopyrimidine series can be mitigated by replacing the pyrimidyl "head group" with a pyridyl motif, albeit with the introduction of hERG activity. In order to attenuate the hERG liability associated with the pyrrolopyridine series, the basicity of the pyridyl group was modulated *via* the addition of a fluorine atom at the *meta* position. This pKa-lowering strategy delivered compounds with comparable Trk potency (albeit with lower LipE), metabolic stability and kinase selectivity, and with an improved hERG liability profile relative to pyrrolopyridine analogues. Table 3 highlights example fluoro-pyrrolopyridine 10⁵⁹ and comparator pyrimidine and pyridine analogues 8³⁸ and 9.⁵⁹ MetID analysis of the fluoropyridyl ring of 2-(4-chlorophenyl)-N-(2-(7-fluoro-1-isopropyl-1H-pyrrolo[3,2-c]pyridine-3-carbonyl)pyridin-4-yl)acetamide (PF-06287755,⁵⁹ a structurally similar analogue of fluoropyridyl 10), in human S9 in the absence of NADP, confirmed no AO oxidation at the fluoro-pyrrolopyridine motif (see supplementary information S8).

An alternative strategy to remove AO-liability is to add a "blocking group" at the metabolically labile position. In the pyrrolopyrimidine series, an -NH₂ group was assessed as a blocking group at the labile 2-position. This strategy delivered compounds with comparable or improved potency, LipE, and hERG profiles. MetID studies with human S9 in the absence of NADP confirmed no AO-mediated oxidation on the 2-aminopyrimidyl ring. Table 1 highlights example 2-aminopyrimidine, compound 11.³⁸



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AO liability on	Yes	No	No	No
head group				
Trk A/B/C IC ₅₀	18/4/ND	17/5/ND	33/7/ND	8/5/4
(nM)				
MW/logD	429/2.5	428/2.7	446/3.5	444/2.5
LipE	5.2	5.0	4.0	5.6
HLM	25	35	37	8
$(\mu L/min/mg)$				
hHeps	12	23	ND	24
$(\mu L/min/mill)$				
Head Group	4.4	6.6	4.8	4.4
pKa				
Dof Ki (μM)	11.9	1.4	11.3	38.9
P-gp/ BCRP ER	5.8/38.0	12.5/ND	3.8/ND	8.7/13.6
RRCK A-B P _{app}	18.4	19.8	15.4	15.8
(x 10 ⁻⁶ cms ⁻¹)				
Solubility (µM)	ND	ND	ND	30
Gini score	0.85	0.82	ND	0.91

Table 3: Structure and property comparison of pyrimidine, pyridine, fluoro-pyrrolopyridine and 2-aminopyrimidine containing Trk ligands. ND = not determined

Interestingly, the kinase selectivity selective profile of 2-aminopyrrolopyrimidine 11 is comparable to *des*-amino pyrrolopyrimidine 8 (Table 2), even though there is the potential for an additional H-bond with the conserved hinge region. However, as exemplified in Figure 10, the -NH₂ group of the 2-aminopyrrolopyrimidine hinge binder makes an imperfect hydrogen bonding interaction with the backbone carbonyl of hinge residue Met592. Due to the sub-optimal geometry of the donor–acceptor-donor arrangement (O···H-N bond angle is 123° cf. an ideal O···H-N bond angle of 180°) the additional hydrogen bond may not significantly contribute to the free energy of binding, and hence not erode kinase selectivity. 60

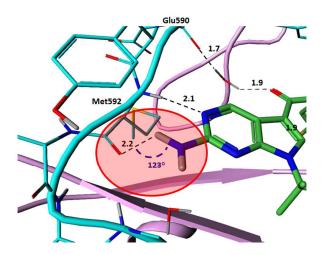


Figure 10: Hydrogen bond interactions of 2-aminopyrrolopyrimidine group with the TrkA hinge binding region (blue). The red circle highlights the non-optimal bond angle between backbone carbonyl A592 and the ligand – NH₂.

Overall, the 2-aminopyrrolopyrimidine head group conferred the best balance of properties of the hinge binding groups assessed although the aqueous solubility was still relatively low. Compound 12, ³⁸ an exemplar 2-aminopyrrolopyrimidine (Table 4), is a potent and selective Trk kinase inhibitor with a high P-gp ER and minimal hERG liability. As with other 2-aminopyrrolopyrimidines, there is no AO oxidation on the aminopyrrolopyrimidine ring. The logD is high (4.2), likely contributing to the relatively poor solubility of 12 μM. In order to design 2-aminopyrrolopyrimidine analogues with enhanced solubility and an improved LipE profile, compounds of reduced logD were required. An analysis of available Trk co-crystal structures suggested polar groups would be tolerated at the *N*-ⁱPr motif of 2-aminopyrrolopyrimidine compounds such as 12, as this region of the binding site is highly solvent accessible. Addition of a solubilising hydroxymethylene group to *N*-ⁱPr motif generated compounds such as 13³⁸ that retained Trk potency, increased LipE due to a lower logD, and effected an improved solubility profile. In addition, replacement of the chlorophenyl ring of with a lower logD pyridyl motif delivered ligands such as 14 which had further optimised LipE and solubility profiles.

Compound	12	13	14
	HN O CI	H ₂ N N N N OH	H _N OH
Trk A/B/C IC	6/2/1	3/2/2	6/4/3
(nM)			
MW/logD	448/4.2	478/3.7	479/2.7
LipE	4.0	4.8	5.5
HLM	< 8.0	< 8.0	< 8.0
(μL/min/mg)			
hHeps	14.0	11.2	4.0
(µL/min/mill)			
Dof Ki/hERG	16.9/ ND	> 4.3/ND	> 40.3/> 30.0
IC ₅₀ (μM)			
P-gp/ BCRP ER	10.9/ND	> 34.0/ND	35.7/4.0
RRCK A-B P _{app}	5.2	6.4	5.4
(x 10 ⁻⁶ cms ⁻¹)			
Solubility (µM)	12	29	131
Gini score	0.93	0.70	0.92

Table 4: Trk pharmacology and *in vitro* ADME profiles of 2-aminopyrimidine compounds **12**, **13** and **14**.

As highlighted in Table 4, **14** is a highly potent pan-Trk inhibitor, with an excellent LipE profile. Compound **14** has low metabolic turnover in HLM and hHep, is a good substrate for efflux transporters P-gp (ER = 35.7) and BCRP (ER = 4.0), and has moderate passive permeability (RRCK $P_{app} = 5.4 \times 10^{-6} \text{ cms}^{-1}$). Its oxidation by AO on the amino pyrrolopyrimidine substituent in human S9 is minimal (see supplementary information S9), and only a low level of metabolism by AO is detected on the central pyridyl ring. The aqueous solubility of **14** is 131 μ M, much improved over previous analogues, it is highly kinase-selective (Gini score of 0.92), and has no measurable activity at the hERG channel.

Compound 14 was profiled in a series of *in vitro* safety assays, showing little cytotoxicity in THLE or HepG2 cell lines (IC₅₀ > 42 μ M and > 300 μ M, respectively), and was evaluated for broader pharmacological activity in a panel of receptors, ion channels and enzymes. In this broad panel, all IC₅₀/K_i values were > 10 μ M except for COX-1 (IC₅₀ = 2.7 μ M) and dopamine transporter assays (K_i = 5.2 μ M) and PDEs 4D, 5A, 7B, 8B and 11 (54 - 89% inhibition at 10 μ M). Compound 14 was screened in the Invitrogen wide kinase panel of 309 kinases, all were inhibited by < 40% when tested at 1 μ M except the following: MUSK (IC₅₀ 53nM), FLT-3 (IC₅₀ 395 nM), IRAK1 (IC₅₀ 2.5 μ M), MKK (90% @ 1 μ M) and DDR1 (60% @ 1 μ M).

Figure 11 shows the crystal structure of **14** bound to TrkA kinase, and highlights key protein-ligand interactions. Compound **14** adopts a DFG-out binding mode, with the 2-amino pyrrolopyrimidine hinge binding motif making a hydrogen bonding interaction to the backbone N-H of Met592 via a pyrimidyl nitrogen and with a "through water" hydrogen bond network between main chain carbonyl Glu590 and the ketone oxygen atom of **14**. An additional hydrogen bond exists between the -NH₂ group of the 2-aminopyrrolopyrimidine hinge binder and the backbone carbonyl of hinge residue A592, but with sub-optimal geometry (O"H-N bond angle, 122°). The central pyridyl group of **14** makes a π -stacking interaction with the gatekeeper (Phe589), and Phe669 of the DFG triad. The carbonyl oxygen of the amide group makes a hydrogen bond interaction with the backbone N-H of Asp 668 from the DFG triad, and the amide N-H group makes a hydrogen bond to a conserved water molecule which further interacts with catalytic lysine (Lys544) and Glu560 from the alpha-C helix. The lipophilic back pocket accommodates the chloropyridyl group.

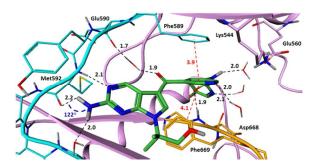


Figure 11: Crystal structure of **14** (green) bound to TrkA (pink). The hinge region of the TrkA ATP site is highlighted in blue. The "DFG" triad of the activation loop of TrkA is highlighted in orange. Key protein-ligand binding interactions are outlined.

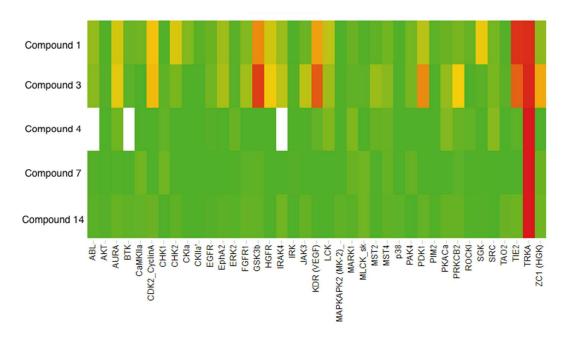


Figure 12: Heat map representation of the Invitrogen selectivity panel data of 39 kinases for 1, 3, 4, 7 and 14, %I values generated at a compound concentration of 1 μ M. 0% I = green, 50% I = yellow, 100% I = red

Brain availability experiments with **14** in rat confirmed significant peripheral restriction, with $C_{b,u}/C_{p,u} = 0.026$. In addition, no measurable TrkB or TrkC occupancy was detected from brain lysates using the ActivX biotinylated acyl phosphate ADP probe. Compound **14** was assessed in the UVIH model of inflammatory pain in rodents. In this study, statistically significant efficacy was observed at unbound plasma concentrations of ≥ 1 x TrkA IC₅₀, with maximal efficacy response at an unbound plasma concentrations of ~ 10 x TrkA IC₅₀ (see

supplementary information, S10). The safety profile of **14** was assessed *in vivo*. In rats, decreases in white blood cell count were observed from 150 mg/kg/day. At doses \geq 250 mg/kg, increases in body weight gain and food consumption were observed, effects that could be rationalized as being mediated by central inhibition of TrkB, agonists of which are known to be anorexigenic in rodents. Adaptive changes in the liver were observed microscopically and accompanied by increased liver weight at \geq 250 mg/kg and increased cholesterol at 1000 mg/kg. Overall **14** was well tolerated up to 1000 mg/kg/day where plasma exposure (unbound C_{avg}) was approximately 400 x TrkA IC₅₀.

In summary, **14** is a highly potent, kinase-selective, peripherally restricted, efficacious and well-tolerated pan-Trk inhibitor and was selected as a candidate for clinical development.

Compound Synthesis

The synthetic route towards **14** is described in Scheme 1.³⁸ Iodination of **15** with NIS gave mono-iodinated **16** in good yield. Compound **16** was subsequently alkylated with methyl 2-bromo-2-methylpropanoate using cesium carbonate as base to afford **17**. Ester hydrolysis of **17** generated **18** in good yield, which was then reduced to give **19**. The reduction step to afford **19** proved challenging and a range of conditions were examined with DIBAL-H and super-hydride proving most effective. DIBAL-H was utilised as the reducing agent upon scale-up. A subsequent protection of the resulting primary alcohol with *t*-butyldimethylsilyl chloride and imidazole in DMF (0°C to rt, 16 hr) delivered **20**, the iodide coupling partner required for the subsequent Weinreb ketone formation, in 96% yield. The additional coupling partner for the ketone formation step, Weinreb amide **22**, was synthesized *via* the palladium mediated cross-coupling of **21** and benzophenone imine. To elicit the Weinreb ketone formation, isopropyl magnesium chloride was added to **20** in THF at 0°C. After stirring at 0°C for 1 hr a solution of Weinreb amide **22** in THF was added, and the resulting

mixture stirred at rt for 16 hr. Following ammonium chloride quench and purification *via* column chromatography, **23** was delivered in 66% yield. Chloro displacement of **23** with 2,4-dimethoxybenzylamine in DMAP and 1,4-dioxane, followed by deprotection of the benzophenoneimine group with citric acid in THF, afforded **24** in 78% yield. Amidation of **24** was achieved *via* treatment with (5-chloropyridin-2-yl)acetic acid, 1-propylphosphonic acid cyclic anhydride (T₃P), and triethylamine in THF. After work-up and *in vacuo* concentration of the organic extracts, the resultant crude oil was treated with trifluoroacetic acid and stirred at 50°C for 3 hours. After this time, the solution was concentrated, methanol and potassium carbonate were added, and the resulting mixture stirred at rt for 16 hrs. The mixture was poured into water and the resultant solid was filtered and triturated with EtOH to afford **14** in 48% yield.

Scheme 1: Synthesis of **14**. Reagents and conditions: i) NIS, MeCN, 12°C to rt, 1 hr, 82%. ii) BrMe₂CO₂Me, KI, Cs₂CO₃, DMF, 60°C, 19 hr, 92%. iii) LiOH, THF/H₂O, 60°C, 3 hr, 90%. iv) DIBAL-H, THF, 0°C, 1.5 hr, 56%. v) TBMS-Cl, imidazole, DMF, 0°C to rt, 16 hr, 96%. vi) Benzophenone imine, Pd₂(dba)₃, K₃PO₄, DME, 50 °C, 17 hr, 51%. vii) **20**, i-PrMgCl, THF, 0°C, then **22**, THF, 0°C to rt, 16 hr, 66%. viii) 2,4-dimethoxybenzylamine, DMAP, 1,4-dioxane, reflux, 2 d, then citric acid, THF, rt, 5 hr, 78%. viiii) 2-(5-chloropyridin-2-yl)acetic acid, T₃P, Et₃N, THF, rt, 2 hr, then TFA, 50°C, 3 hr, then K₂CO₃, MeOH, rt, 16 hr, 48%.

CONCLUSION

The tropomyosin related kinase (Trk) family of tyrosine kinase receptors represents an important target class in the pain therapy field. In addition to delivering the requisite analgesic effects, a key requirement is to deliver a therapeutic agent with a very benign safety side effect profile. Kinases modulate numerous physiological functions, ²⁶⁻²⁷ so designing highly kinome-selective Trk kinase inhibitors is a key objective. Within the Pfizer Trk project, chemical series with an anticipated DFG-out binding mode were prioritized for follow-up, as kinase inhibitors that bind to less conserved areas of the ligand binding domain, outside of the ATP binding pocket, can sometimes be more kinase-selective. ^{32-33,35-36,40} An additional requirement is to deliver Trk ligands with minimal brain availability at therapeutically relevant doses, due to the known safety (cognitive) side effects associated with inhibiting Trk receptors in the CNS. The strategy employed to deliver peripherally restricted, orally bioavailable compounds was to design compounds in physicochemical space appropriate for absorption across the gastrointestinal epithelium that act as substrates for blood-brain barrier efflux transporters such as P-glycoprotein (P-gp). ²²

A high throughput screening campaign identified a "DFG-out" pyrrolopyrimidine series; potent pan-Trk inhibitors with moderate LipE, encouraging kinase selectivity and P-gp affinity, but very poorly soluble. Following a Matched Molecular Pair analysis, the urea motif was replaced with an amide group to improve solubility whilst maintaining Trk activity. The extent of *in vitro* P-gp activity required to deliver significant peripheral restriction *in vivo* was established by assessing the $C_{b,u}$ / $C_{p,u}$ and brain Trk RO (using a chemical proteomic technology developed by ActivX Biosciences Inc.) of compounds with P-gp ER in the range 1 – 50. These data, alongside an analysis of the performance of an inhouse 2D QSAR P-gp ER model, were utilized to guide the prospective design of compounds with a high *in vitro* P-gp ER, and hence significant peripheral restriction *in vivo*.

Optimization of kinase selectivity was achieved by reducing the extent of polar contact made between the ligand and the backbone polar atoms of the conserved kinase hinge residues *via* analysis of the *apo* versus *holo* crystal structures, and replacing a protein-ligand hydrogen bond with a relatively rare "bridging" water molecule. Further optimization of the hinge binding group to minimize AO liability and the use of SBDD to add additional polar group functionality to modulate logD delivered highly potent, kinase-selective, peripherally restricted, efficacious and well-tolerated pan-Trk inhibitors. Compound **14** was selected as a candidate for clinical development and clinical data will be published in due course.

EXPERIMENTAL SECTION

Starting materials and other reagents were purchased from commercial suppliers and were used without further purification unless otherwise indicated. All reactions were performed under a positive pressure of nitrogen, argon or with a drying tube at ambient temperature (unless otherwise stated) in anhydrous solvents, unless otherwise indicated. Analytical thin-layer chromatography was performed on glass-backed silica gel 60_F 254 plates (Analtech (0.25 mm)) and eluted with the appropriate solvent ratios (v/v). The reactions were assayed by high performance liquid chromatography (HPLC) or thin-layer chromatography (TLC) and terminated as judged by the consumption of starting material. The TLC plates were visualized by phosphomolybdic acid stain or iodine stain. Microwave assisted reactions were run in a Biotage initiator. ¹H NMR spectra were recorded on a Bruker instrument operating at 400 MHz unless otherwise indicated. ¹H NMR spectra are obtained as DMSO-d₆ or CDCl₃ solutions as indicated (reported in ppm), using chloroform as the reference standard (7.25 ppm) or DMSO-d₆ (2.50 ppm). Other NMR solvents were used as needed. When peak multiplicities are reported, the following abbreviations are used: s = singlet, d = doublet, t = triplet, m = multiplet, br = broadened, dd = doublet of doublets, dt = doublet of triplets.

Coupling constants, when given, are reported in hertz. Mass spectral date were obtained using Waters ZQ ESCI or Applied Biosystem's API-2000. Chemical purities were > 95% for all final compounds as assessed by LCMS, with UV detection at 220 and/or 254 nm. Further details on the analytical conditions used for individual compounds may be found in the Supporting Information. The chemical yields reported below are unoptimized specific examples of one preparation. The chemical yields reported are unoptimized specific examples of one preparation. Compound 14 is commercially available *via* Sigma-Aldrich (catalogue no. PZ0254).

N-(5-{[2-Amino-7-(2-hydroxy-1,1-dimethylethyl)-7H-pyrrolo[2,3-d]pyrimidin-5-

yl|carbonyl} pyridin-3-yl)-2-(5-chloropyridin-2-yl)acetamide (14). (5-Chloropyridin-2-yl)acetic acid (26.1 g, 152 mmol) was added to 24 (75.0 g, 130 mmol), 1-propylphosphonic acid cyclic anhydride (187 mL, 317 mmol, 50% solution in EtOAc) and triethylamine (61.9 mL, 444 mmol) in THF (423 mL). The mixture was stirred at 25°C for 2 hours then saturated aqueous sodium bicarbonate (400 mL) was added and the organic layer was separated. The aqueous phase was extracted with EtOAc (400 mL) and all organic phases were combined and dried over sodium sulfate then evaporated *in vacuo*. The residual brown solid was dissolved in trifluoroacetic acid (300 mL) and the solution was stirred at 50°C for 3 hours then evaporated *in vacuo*. Methanol (1800 mL) was added to the residue and the mixture was filtered. The filtrate was evaporated *in vacuo* and azeotroped with ethanol (3 x 200 mL). Potassium carbonate (87.7 g, mmol) was added to the crude trifluoroacetamide in methanol (300 mL) and the mixture was stirred at room temperature for 16 hours. The mixture was poured into water (2000 mL) and filtered. The solid was washed with water (200 mL) then triturated with ethanol (2 x 200 mL at room temperature then 380 mL at 50°C) to afford the title compound as a yellow solid in 48% yield, 29.9 g. ¹H NMR (400 MHz, DMSO-d6) δ:

1.64 (s, 6H), 3.90 (d, J = 5.5, 2H), 3.95 (s, 2H), 5.05 (dd, J = 5.7, 5.5, 1H), 6.54 (br s, 2H), 7.49 (d, J = 8.4, 1H), 7.69 (s, 1H), 7.92 (dd, J = 8.3, 2.4, 1H), 8.40 (m, 1H), 8.56 (d, J = 2.5, 1H), 8.64 (d, J = 1.8, 1H), 8.94 (d, J = 2.2, 1H), 8.96 (s, 1H), 10.71 (s, 1H); HPLC (6 min, acid) R_t 1.26 min, UV 220 nM 100% purity; LC-MS (ES⁻) m/z 478 (M - H⁺); HRMS (ES⁺) m/z 480.15468 (M + H⁺).

ASSOCIATED CONTENT

Supporting information (SI) is available at http://pubs.acs.org: SI contains crystal structure of imatinib bound to Abl, HPLC-UV chromatograms of metabolites of 5, 6, 7, PF-06287755 and 14 (in HS9 in the absence of additional NADP), incubation of PF-05302191 in human liver S9 fraction in the presence and absence of Aldehyde Oxidase (AO) and Xanthene Oxidase (XO) inhibitors, ultraviolet burn induced hyperalgesia (UVIH) model data for 7 and 14. Experimental and spectral data for 16, 17, 18, 19, 20, 22, 23, 24 and 14, crystallography experimental data, human liver microsome (HLM) depletion assay information, human hepatocyte (hHEP) depletion assay information, metabolism of compounds in human liver S9 fraction in vitro assay information, incubation in human liver S9 fraction in the presence and absence of Aldehyde Oxidase (AO) and Xanthene Oxidase (XO) inhibitors assay information, permeability in RRCK cell monolayers assay information, determination of efflux by P-gp and BCRP assay information and Trk cellular primary pharmacology assay information.

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The authors declare no competing financial interest.

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ACCESSION CODES

PDB codes are 5JFS for TrkA + Compound 1, 5JFV for TrkA + Compound 3, 5JFW for TrkA + Compound 5, and 5JFX for TrkA + Compound 14

ABBREVIATIONS USED

AB, apical basolateral; ADME, absorption, distribution, metabolism and excretion; ADP, adenosine diphosphate; AO, aldehyde oxidase; ATP, adenosine triphosphate; BA, basolateral to apical; BCRP, breast cancer resistance protein; BDNF, brain derived neurotrophic factor; BBB, blood-brain barrier; $C_{b,u}/C_{p,u}$, unbound brain/plasma concentration ratio; C_{avg} , average plasma concentration; CL, clearance; CL_{int}, intrinsic clearance; CNS, central nervous system; CSDb, Pfizer crystal structure database; CYP, cytochrome P450; DDI, drug-drug interaction; DIBAL-H, diisobutylaluminium hydride; Dof., dofetilide; ER, efflux ratio; F, bioavailability; fu, fraction unbound; hERG, human ether-a-go-go related gene; hHep, human hepatocytes; HLM, human liver microsomes; IC₅₀, half-maximum inhibitory concentration; Ki, dissociation constant; Km, Michaelis-Menten constant; LBF, liver blood flow; LipE, lipophilic efficiency; log D, distribution coefficient at pH = 7.4; mAb, monoclonal antibody; MDCK, Madine Darby canine kidney; MDR1, multidrug resistance protein (p-glycoprotein); MetID, *in vitro* metabolite identification; MMP, matched molecular pair; NADP,

Nicotinamide Adenine Dinucleotide Phosphate; ND, not determined; NT3, neurotrophin 3; NT4, neurotrophin 4; NIS *N*-iodosuccinimide; NGF, nerve growth factor; P_{app}, apparent permeability; PDB, Protein Data Bank; P-gp, p-glycoprotein; PK, pharmacokinetics; QSAR, quantitative structure-activity relationship; RMSD, root mean square deviation; RO, receptor occupancy; RRCK, Ralph Russ canine kidney cell line, permeability measured with low-efflux MDCK cell; SBDD, structure based drug design; T_{1/2}, half-life; THLE, transformed human liver epithelial cells; Trk, tropomyosin related kinase; UVIH, ultraviolet burn induced hyperalgesia; V_{ss}, volume of distribution at steady state.

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