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Enantioselective Intramolecular C-H Amination of Aliphatic Azides by Dual Ruthenium and Phosphine Catalysis

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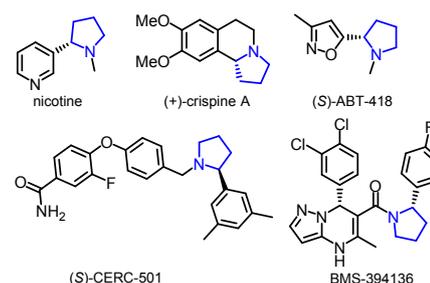
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The catalytic enantioselective intramolecular C(sp³)-H amination of aliphatic azide represents an efficient method for constructing chiral saturated cyclic amines which constitute a prominent structural motif in bioactive compounds. We report a dual catalytic system involving a chiral-at-metal bis(pyridyl-NHC) ruthenium complex and tris(4-fluorophenyl)phosphine (both 1 mol%), which in combination achieve the cyclization of aliphatic azides to chiral α -aryl pyrrolidines with enantioselectivities of up to 99% ee, including a pyrrolidine which can be converted to the anti-tumor alkaloid (*R*)-(+)-crispine. Mechanistically, the phosphine activates the organic azide to form an intermediate iminophosphorane and transfers the nitrene unit to the ruthenium providing an imido ruthenium intermediate which engages in the highly stereocontrolled C-H amination. This dual-catalysis combines ruthenium catalysis with the Staudinger reaction and provides a novel strategy for catalyzing enantioselective C-H aminations of unactivated aliphatic azides.

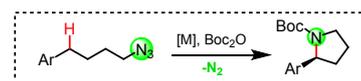
Introduction

Pyrrolidines constitute a prominent structural motif in bioactive compounds such as natural products and pharmaceuticals (Figure 1a).^{1,2} Their synthesis through a direct intramolecular C(sp³)-H amination is particularly appealing due to the lacking necessity for preinstalled functional groups which provides the prospect of an efficient synthesis with high atom economy.³ In this respect, organic azides are attractive functionalities for metal-mediated nitrene C(sp³)-H insertion reactions^{4,5} because no additional oxidant is required and molecular nitrogen is the only by-product. Whereas the amination of saturated C(sp³)-H bonds with aryl, sulfonyl, acyl and phosphoryl azides has been well established,⁶ the use of non-activated, aliphatic azides is only a recent accomplishment.^{7,8} In addition to their lower reactivity, a major pitfall for C(sp³)-H aminations of primary aliphatic azides constitutes a competing unproductive 1,2-hydride shift of the intermediate alkyl nitrenoid intermediate leading to the irreversible formation of undesirable imines.⁹ Betley introduced an elegant dipyrinato-iron(II)-catalyzed ring-closing C(sp³)-H amination of aliphatic azides but the reported turnover numbers were modest with TON < 10.^{8b,c} Subsequently reported MOF-functionalized Fe(II)- β -diketiminato,^{8d} iron(III)-coordinated redox-active pyridine-aminophenol,^{8e} and cobalt(II) porphyrin^{8g} catalysts by Lin, van der Vlugt, and de Bruin, respectively, provided improved catalytic performances for this challenging transformation. Finally, Che very recently reported an

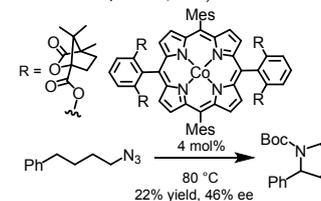
a) Chiral α -aryl pyrrolidine-containing bioactive compounds



b) Catalytic enantioselective cyclization with aliphatic azides



State of the art (de Bruin, 2017):



This study:

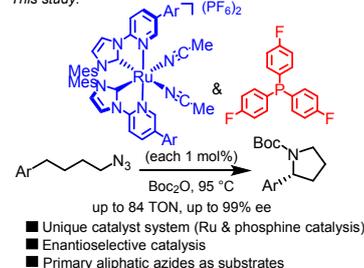


Figure 1. Previous and this work on asymmetric ring-closing C(sp³)-H aminations of aliphatic azides to provide non-racemic pyrrolidines.

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N-heterocyclic carbene iron(III) porphyrin complex exhibiting high activity for this transformation under microwave conditions.^{8h} However, pyrrolidines as part of bioactive compounds are typically chiral² but only a single example of a catalytic enantioselective reaction has been reported using a chiral cobalt(III) porphyrin achieving low yields and very low enantiomeric excess (Figure 1b).^{8g,10}

Ruthenium complexes are well-established for catalyzing C(sp³)-H activation¹¹ and C(sp³)-H aminations of organic azides such as aryl, acyl, and sulfonyl azides,^{6a,d,f,l-n,12} but applying simple primary aliphatic azides has remained elusive and this has been attributed at least in parts to a very efficient 1,2-hydrogen shift of the intermediate Ru-imido complexes.^{9c-e} Our group recently reported a new class of “chiral-at-metal” ruthenium catalysts in which two bidentate *N*-(2-pyridyl)-substituted *N*-heterocyclic carbenes and two acetonitrile ligands are coordinated to a central ruthenium in a C₂-symmetric fashion.¹³ Despite all ligands being achiral, overall helical chirality originates from a stereogenic ruthenium center.^{14,15} We recently demonstrated that such complexes can indeed serve as catalysts for activating aliphatic azides towards enantioselective C-H amination, however, unfortunately, only in a very restricted structural context of converting 2-azidoacetamides into chiral imidazolidin-4-ones.¹⁶ By discovering a novel ruthenium and phosphine dual catalysis scheme to activate aliphatic azides towards C-H amination, we here report the highly enantioselective ring-closing C(sp³)-H amination of simple 4-azidobutylarenes to provide chiral α -aryl pyrrolidines with enantioselectivities of up to 99% ee (Figure 1b).

Results and discussion

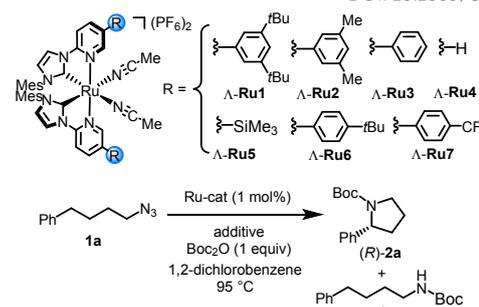
Initial optimization

We initiated our study by investigating the intramolecular C-H amination of 4-azidobutylbenzene (**1a**) to Boc-protected 2-phenylpyrrolidine (**2a**). Unfortunately, despite extensive screening of the reaction conditions, the results were unsatisfactory (see details in Tables S1-4). For example, heating **1a** under optimized conditions in 1,2-dichlorobenzene at 95 °C for 40 hours using Λ -**Ru1** (1 mol%) in the presence of Boc₂O (1 equiv) provided the desired pyrrolidine **2a** only in 26% NMR yield at a conversion of 57% but at least with an encouraging enantioselectivity of 81% ee (Table 1, entry 1). At the same time, the Boc-protected amine **3** was detected as a side-product in 14% yield. To our surprise, we finally discovered that the reaction was vastly improved when performed in the presence of catalytic amounts of PPh₃, providing **2a** in 44% yield with 79% conversion at an enantioselectivity of 82% ee (entry 2). Even slightly better yields were obtained with tris(4-fluorophenyl)phosphine (entry 3), while other phosphines provided inferior results (Table S5). The amount of the phosphine was also investigated, and it was found that 1 mol% provided the optimal results (Table S6). Next, we optimized the ruthenium catalyst for this transformation to improve the enantioselectivity. Our initial

Table 1. Initial C-H amination experiments.^a

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Entry	Cat.	Additive	Conv. (%) ^b	NMR yield (%) ^b		ee (%) ^f
				2a	3	
1 ^d	Λ - Ru1	no	57	26	14	81
2 ^d	Λ - Ru1	P(Ph) ₃	79	44	17	82
3 ^d	Λ - Ru1	P(4-F-Ph) ₃	81	46	18	82
4	Λ - Ru2	P(4-F-Ph) ₃	77	46	18	89
5	Λ - Ru3	P(4-F-Ph) ₃	79	44	16	90
6	Λ - Ru4	P(4-F-Ph) ₃	75	44	20	87
7	Λ - Ru5	P(4-F-Ph) ₃	70	43	17	89
8	Λ - Ru6	P(4-F-Ph) ₃	65	23	18	80
9	Λ - Ru7	P(4-F-Ph) ₃	77	54 (51) ^e	18	95
10	Λ - Ru7 ^f	P(4-F-Ph) ₃	65	42	16	94
11	Λ - Ru7	no	45	12	14	95
12 ^g	Λ - Ru7	P(4-F-Ph) ₃	<3	0 ^h	0 ^h	n.a. ⁱ

^aStandard conditions: **1a** (0.2 mmol), Boc₂O (0.2 mmol, 1 equiv), catalyst (0.002 mmol, 1 mol%), and additive (0.002 mmol, 1 mol%) in 1,2-dichlorobenzene (0.5 mL) at 95 °C for 60 h under N₂ unless otherwise noted. ^bDetermined by ¹H NMR of crude products using Cl₂CHCHCl₂ as internal standard. ^cDetermined by HPLC of crude main product on a chiral stationary phase. ^d40 h was used. ^eIsolated yield in parentheses. ^f0.5 mol% Λ -**Ru7** was used. ^gWithout Boc₂O. ^hRefers to cpds without Boc-protection. ⁱn.a. = not applicable.

experiments were performed with the chiral-at-ruthenium catalyst Λ -**Ru1** which bears two very bulky 3,5-di(*tert*-butyl)phenyl substituents at the coordinating pyridine ligands. Interestingly, Λ -**Ru2** with less bulky 3,5-(dimethyl)phenyl substituents at the pyridyl moieties provided an even higher enantioselectivity of 89% ee (entry 4). The phenyl-modified catalyst Λ -**Ru3** afforded a further slightly increased enantioselectivity of 90% ee (entry 5), whereas the plain catalyst devoid of additional substituents (Λ -**Ru4**) yielded the Boc-protected pyrrolidine with reduced 87% ee (entry 6). Furthermore, a trimethylsilyl (TMS)-functionalized ruthenium complex Λ -**Ru5**, which is the optimal catalyst in our previous work on enantioselective C-H aminations of 2-azidoacetamides,¹⁶ did not



provide better results here (entry 7). Adding a *t*Bu-moiety at the 4-position of the phenyl groups (Λ -Ru6) decreased the enantioselectivity to 80% ee (entry 8). However, the best result was obtained with a 4-(CF₃)Ph modification (Λ -Ru7) which afforded (*R*)-2a in 54% NMR yield at 77% conversion and with 95% ee (entry 9). Interestingly, even at a catalyst loading of just 0.5 mol%, the Boc-protected pyrrolidine (*R*)-2a was still formed with 42% yield (65% conversion) and 94% ee, reflecting a turnover number of 84 (entry 10). Thus, a careful optimization of steric and electronic effects provided a ruthenium catalyst (Λ -Ru7) which, in the presence of tris(4-fluorophenyl)phosphine, effectively discriminates between the two benzylic C-H bonds of 4-azidobutylbenzene to provide the corresponding chiral pyrrolidine with modest yield but outstanding enantioselectivity.^{17,18}

Control experiments revealed that tris(4-fluorophenyl)phosphine is crucial for obtaining a satisfactory yield. In its absence, the yield diminished to merely 12% even after an extended reaction time, while the enantioselectivity was not affected, thus implying that the phosphine is not involved in the stereocontrolling step (entry 11). Finally, Boc₂O is also required for this reaction to proceed (entry 12).¹⁶

Substrate scope investigation

With the optimized catalyst Λ -Ru7 and optimized reaction conditions in hand, we investigated the substrate scope of this transformation. As shown in Figure 2, methyl groups in *para* or *meta* position of the benzene moiety are well tolerated (pyrrolidines 2b,c), but a sterically demanding *ortho*-methyl group leads to vastly diminished yields of 15%. Both electron-donating (leading to pyrrolidines 2e-k) and electron-accepting substituents (leading to pyrrolidines 2l, m) are tolerated, although electron-accepting substituents led to decreased yields. The phenyl moiety can also be replaced by a naphthyl (pyrrolidine 2n) and by heteroaromatic moieties (pyrrolidines 2o-s). For example, a carbazole moiety provides the Boc-protected pyrrolidine 2s in 52% yield and 93% ee. However, replacing the aryl group with alkyl, alkenyl, or alkynyl groups suppresses the C-H amination reaction (Table S8, entries 1-3). Further substrates including bridging aryl groups could also be transformed to their pyrrolidine products (2t,u). Finally, a racemic substrate with a tertiary C-H group only provided a very modest kinetic resolution (Table S8, entry 4).

Overall, enantioselectivities of 76-99% ee were observed and isolated yields of 15-57%. However it has to be noted that none of the reactions proceed to full conversion and thus allow to reisolated unreacted starting materials (Figure 2). Despite the modest yields it is remarkable that chiral α -aryl pyrrolidines with outstanding enantioselectivities of up to 99% ee can be obtained through this challenging ring-closing C(sp³)-H amination of aliphatic azides.¹⁹ Importantly, chiral α -aryl pyrrolidines are prominent structural motifs in bioactive compounds.² For example, pyrrolidine 2g can be converted to the anti-tumor alkaloid (*R*)-(+)-crispine in 4 steps using a Pummerer cyclization (Figure S1).^{2c}

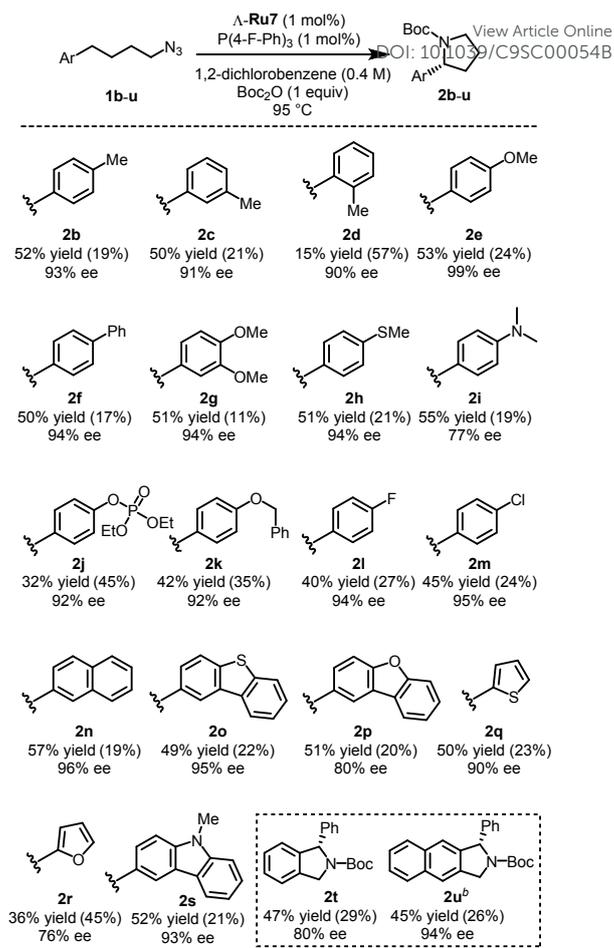


Figure 2. Substrate scope with isolated yield.^a

^aRecovered starting materials are shown in parentheses. ^bLower substrate concentration (0.1 M) and 85 °C was used instead.

Mechanism study

We performed experiments to gain insight into the reaction mechanism and started with the unusual function of the phosphine. The role of the phosphine as a co-catalyst to activate the organic azide is advocated by the well-established reactivity of azides towards phosphines.²⁰ Indeed, P(4-F-Ph)₃ starts to react with (4-azidobutyl)benzene already at room temperature and full conversion is obtained at 95 °C for 2 hours to form the corresponding iminophosphorane 4 (Figure 3a). We also confirmed that iminophosphorane 4 is catalytically competent itself (Figure 3b). Interestingly, although such a role of phosphines in the activation of organic azides towards C-H amination has not been reported to our knowledge, the opposite reaction namely the phosphine-induced extraction of a nitrene from a metal imido



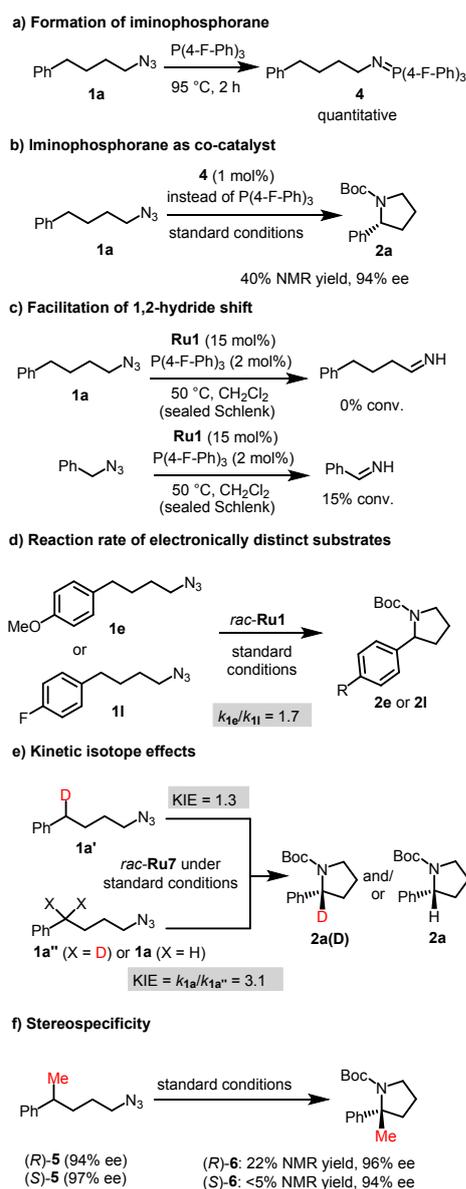


Figure 3. Mechanistic experiments.

complex was disclosed independently by McElwee-White^{21a} and Sundermeyer.^{21b}

Next, we attempted to gain insight into the competing 1,2-hydride shift by comparing (4-azidobutyl)benzene and benzyl azide as substrates at a temperature where C-H amination does not yet occur. As a result, only benzyl azide provided significant amounts of the imine product which can be traced back to the higher activity of the benzylic C-H group in the ruthenium imido intermediate towards 1,2-H shift (Figure 3c). This is consistent with a recent report by Park who showed that the degree of 1,2-H shift correlates with the nature of the α -C-H bond.^{9c}

To understand the electronic nature of the proposed ruthenium nitrenoid intermediate, initial C-H amination rates of electronically distinct substrates **1e** and **1l** were determined. As a result, the cyclization rate of electron-rich **1e** was 1.7 times faster than electron-deficient **1l** suggesting the electrophilic nature of the ruthenium nitrene (Figure 3d). This is consistent with the substrate scope in which electron-rich substrates provided better yields

compared with electron-deficient substrates. The C-H amination with mono-deuterated substrate **1a'** using a **racemic catalyst** provided an intramolecular kinetic isotope effect (KIE) of 1.3 (Figure 3e), which is much lower than the value reported for iron^{8b,e} and cobalt^{8g} catalytic systems. This might suggest that the C-H amination appears to occur by a concerted-insertion mechanism or a hydrogen abstraction mechanism with fast radical recombination.^{8h} However, the interpretation of the intramolecular KIE is complicated by the fact that the ruthenium catalyst is intrinsically chiral, although used as a racemic mixture for this experiment, and the monodeuterated substrate **1a'** as well. Indeed the cyclization of the chiral substrate (*R*)-**5** to (*R*)-**6** but not (*S*)-**5** to (*S*)-**6** demonstrates the high stereospecificity of the C-H amination (Figure 3f). Finally, we determined a pronounced (noncompetitive) intermolecular KIE value of 3.1 by measuring initial C-H amination rates of non-deuterated (**1a**) and bis-deuterated (**1a''**) substrates (Figure 3e). Using alternatively a 1:1 mixture of non-deuterated (**1a**) and bis-deuterated substrate (**1a''**) provides a (competitive) intermolecular KIE of 3.9 (see Figure S6). An observation of a significant intermolecular KIE reveals that the C-H amination is the rate limiting step in the overall process.

Based on previous work^{8b-h} on the ring-closing C-H amination of (4-azidobutyl)arenes and our mechanism study, the following mechanism is proposed (Figure 4). P(4-F-Ph)₃ activates the organic azide to form an intermediate iminophosphorane (I) through the well-known Staudinger reaction which then transfers a nitrene to the ruthenium center to afford a ruthenium imido complex (intermediate II), followed by a stereo-controlled insertion of the nitrene moiety into the δ -C-H bond (transition state III) to provide a ruthenium-coordinated pyrrolidine (intermediate IV). Alternatively, a stepwise process through H-atom transfer cannot be excluded at this point. Finally, the product is released after Boc-protection.

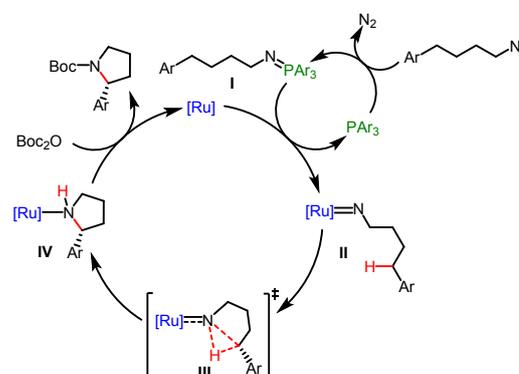


Figure 4. Proposed mechanism.

Conclusions

We here presented a highly enantioselective catalytic ring-closing benzylic C(sp³)-H amination of primary aliphatic azides to provide chiral 2-aryl pyrrolidines by combining of chiral-at-metal transition metal catalysis with nucleophilic phosphine catalysis. In this unique dual catalysis system, the phosphine activates the organic azide and transfers a nitrene to the ruthenium complex, which then executes the enantioselective C-H amination. This combination of ruthenium catalysis and Staudinger reaction introduces a novel direction for C-H



amination of unactivated aliphatic azides which are very desirable but challenging substrates for this transformation.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

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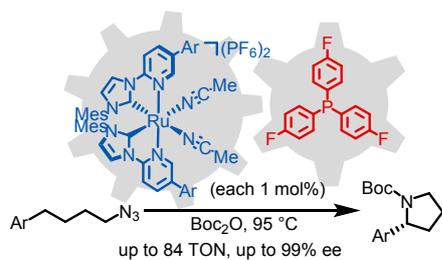
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Enantioselective Intramolecular C-H Amination of Aliphatic Azides by Dual Ruthenium and Phosphine Catalysis

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By combining a chiral-at-metal ruthenium catalyst with catalytic amounts of tris(p-fluorophenyl)phosphine (both 1 mol%) the challenging catalytic enantioselective ring-closing $\text{C}(\text{sp}^3)\text{-H}$ amination of unactivated aliphatic azides has been achieved with high enantioselectivities.

