

Malaria-Infected Mice Live until at Least Day 30 after a New Monomeric Trioxane Combined with Mefloquine Are Administered Together in a Single Low Oral Dose[†]

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In only five simple steps and 48% overall yield from the natural trioxane artemisinin, the thermally and hydrolytically stable trioxane fluoroanilide **4b** has been prepared. Upon one oral dose of only 6.8 mg/kg of monomeric trioxane **4b** combined with 20 mg/kg of mefloquine hydrochloride, all of the malaria-infected mice lived until at least day 30 post infection. Of the five mice in this surviving group, four (80%) were completely cured (no parasites in their blood) and one mouse had 4% blood parasitemia. Importantly, the efficacy of this ACT chemotherapy using monomeric trioxane **4b** plus mefloquine hydrochloride is considerably better than the efficacy under the same conditions using the popular trioxane drug artemether plus mefloquine hydrochloride.

Introduction

Resistance of malaria parasites, especially to chloroquine, is now widespread, seriously compromising the efficacy of this antimalarial drug that has been so widely and successfully used during the past 60 years.¹ Today, as recommended by the World Health Organization (WHO)² and as adopted by most countries where malaria is endemic, artemisinin (**1**) combination therapy (ACT^a) is popular.^{3–7} Typically, natural trioxane **1** derivatives like artemether (**2b**) or sodium artesunate (**2c**) are used. One leading ACT drug for chemotherapy of people sick with malaria is a fixed 1:6 combination of the trioxane **2b** with the amino-alcohol lumefantrine.⁸ Typically, a six-dose adult regimen requires a total of approximately 320 mg of trioxane **2b** and 1920 mg of lumefantrine. A second ACT drug requires a three-dose adult regimen totaling 600 mg of trioxane **2c** and 750 mg of the quinoline antimalarial mefloquine. Patient compliance with adhering to a repeated dose regimen, however, is often a serious problem. Therefore, a single dose oral cure is highly desirable. Toward this goal, we have recently reported a “proof of principle” advance in malaria chemotherapy: a single 144 mg/kg oral dose cure of malaria-infected mice by a new trioxane dimer sulfone carbamate.⁹ We have now prepared and tested the *in vivo* antimalarial efficacies of the fluorobenzyl amide trioxane dimer **3**, of the corresponding trioxane monomer fluorobenzyl amide **4a**, and of the trioxane monomer fluoroanilide **4b** (artefanilide). Monomeric trioxane

fluoroanilide **4b** is the most efficacious; using only one single-digit mg/kg oral dose of this new trioxane monomer combined with a 3-fold higher amount of mefloquine hydrochloride prolonged survival of malaria-infected mice until at least day 30. The total amount of fluoroanilide **4b** and mefloquine hydrochloride needed to achieve this single oral dose high efficacy compares favorably with the amounts of the antimalarial trioxane drugs **2b** and **2c** plus amine currently used clinically in repeated oral dose human ACT chemotherapy (Scheme 1).⁸

Results and Discussion

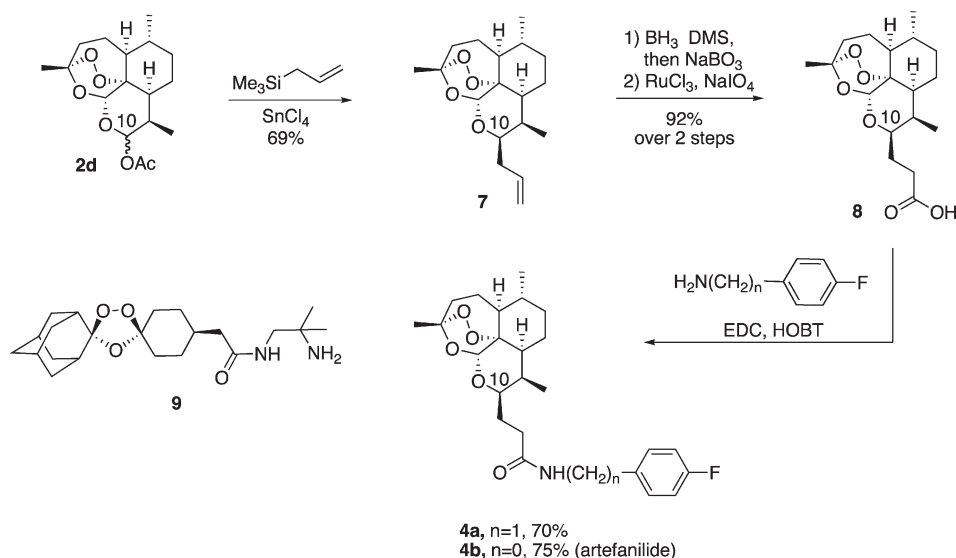
Chemistry. Monomeric trioxane dihydroartemisin C-10 acetate (**2d**) has been converted into dimeric trioxane **5** and then into dimer carboxylic acid **6** (Scheme 2).¹⁰ Facile amidation produced trioxane dimer fluorobenzyl amide **3**. Also, C-10 acetate **2d** has been converted into C-10-allyl derivative **7** and then, via hydroboration followed by oxidation, into monomeric trioxane carboxylic acid **8** (Scheme 3).¹¹ Monomeric trioxane carboxylic acid **8** can be transformed in one step directly into a diverse library of monomeric trioxane amides.¹¹ For example, one-step amidation of carboxylic acid **8** produced trioxane fluorobenzyl amide **4a**, a monomeric version of dimer fluorobenzyl amide **3** (Scheme 3). In the same way, monomer trioxane fluoroanilide **4b** was prepared directly and in good yield from carboxylic acid **8** (Scheme 3). The overall yield of **4b** is approximately 48% from natural trioxane **1**, and scale up to multigram or even kilogram amounts is expected to be straightforward. Fluoroanilide **4b** is stable as a solid in the absence of solvent for at least 7 days at 60 °C and for at least 1 day at 70 °C. Because amides **3**, **4a**, and **4b** are also C-10 nonacetal trioxanes, they are all more hydrolytically stable than the C-10 acetal trioxane drugs **2b** and **2c**.

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^a Abbreviations: ACT, artemisinin combination therapy; DMSO, dimethyl sulfoxide; DMS, dimethyl sulfide; EDC, *N*-(3-dimethylaminopropyl)-*N'*-ethylcarbodiimide hydrochloride; HOBT, 1-hydroxybenzotriazole.

Scheme 3

**Table 1.** Antimalarial Efficacy Using a Single Oral Dose of Trioxane Combined with Mefloquine Hydrochloride in *Plasmodium berghei*-Infected Mice

Trioxane	Oral Dose		Average survival (days) after infection	% Suppression of parasitemia (on day 3 post infection)
	Trioxane (mg/kg)	Mefloquine Hydrochloride (mg/kg)		
2b	6.8	20	19.8 (16,17,21,22,23) ^a	99.8
2b	13	13	19.6 (16,17,21,22,22) ^a	99.7
3	6.8	20	27.4 (22,25,30,30,30) ^a	99.5
3	13	13	24.6 (17,21,25,30,30) ^a	99.9
4a	6.8	20	27.8 (22,27,30,30,30) ^a	99.9
4a	13	13	26.4 (21,23,26,30,30) ^a	99.6
4b	6.8	20	30	99.0
4b	13	13	30	99.6
4b	72	0	24.4 (21,23,24,27,27) ^a	99.0
Controls				
vehicle (no drug)	0	0	6.4 (6,6,6,7,7) ^a	0
2c	30	0	7.6 ^b	- ^c
9	30	0	10.7 ^b	99.95 ^b
Mefloquine	0	20	17.6 (16,16,17,19,20) ^a	99.6

^a Actual mouse survival until day. ^b Data from Supporting Information in ref 13 using a single oral dose of 30 mg/kg. ^c Dash indicates "not measured".

6.8 mg/kg of the popular trioxane drug **2b** combined with 20 mg/kg of mefloquine hydrochloride prolonged average survival to only day 19.8 (Table 1). Neither overt toxicity nor behavioral change attributable to trioxane drug administration was observed in any of the malaria-infected animals cured by trioxane fluoroanilide **4b** plus mefloquine hydrochloride combination. The water-soluble monomeric trioxane antimalarial drug **2c** and the trioxolane antimalarial drug candidate **9**, although able to lower parasitemia levels considerably by day 3 post infection, were not efficacious in prolonging the mouse average survival time beyond day 11 when used as monotherapy at a dose of 30 mg/kg.

Conclusions

In conclusion, a single-digit oral dose of any one of the three new trioxane fluorinated amides **3**, **4a**, and **4b** combined with mefloquine hydrochloride is considerably more antimalarially efficacious than the popular ACT trioxane drug **2b** combined with mefloquine hydrochloride.^{14,15} Monomer trioxane fluoroanilide **4b** stands out as being the most powerful antimalarial in this series of semisynthetic trioxane fluorophenyl amides. Preclinical drug development is continuing on trioxane lead optimization and on use of other amine combination partners.

Experimental Section

High-pressure liquid chromatography (HPLC) was performed on a Rainin HPLX system equipped with two 25 mL pump heads and a Rainin Dynamax UV-C dual-beam variable wavelength detector set at 254 using a Phenomenex Luna 5 μ C18 250 mm \times 10 mm column. The purity of analogs **3**, **4a**, and **4b** was $\geq 98\%$ based on HPLC analysis.

Synthesis of Trioxane Dimer Fluorobenzyl Amide (3). To a solution of acid **6**¹⁰ (600 mg, 0.967 mmol) in CH_2Cl_2 (10 mL) were added *N*-(3-dimethylaminopropyl)-*N'*-ethylcarbodiimide hydrochloride (222 mg, 1.16 mmol) and 1-hydroxybenzotriazole (157 mg, 1.16 mmol) and it was stirred for 1 h at rt. To the reaction were added 4-fluorobenzylamine (0.33 mL, 2.9 mmol), and the solution was stirred for 5 h. It was quenched with water (3 mL). Layers were separated and the aqueous layer was extracted with EtOAc (2 \times 4 mL). The combined organic solution was dried (MgSO_4) and concentrated. The residue was purified by flash column chromatography (elution with EtOAc:hexanes = 1:3) to provide **3** (651 mg, 93%) as a white solid: $[\alpha]_{\text{D}}^{24} = +82.1$ ($c = 1.55$, CHCl_3); mp = 110 $^\circ\text{C}$. IR (thin film) 3312, 2939, 1669, 1510, 1377, 1221, 1052, 1012, 735 cm^{-1} . ^1H NMR (400 MHz, CDCl_3) δ 7.32 (m, 2H), 6.97 (m, 2H), 6.23 (t, $J = 5.6$ Hz, 1H), 5.27 (s, 1H), 5.20 (s, 1H), 4.41 (s, 1H), 4.39 (s, 1H), 4.09 (m, 2H), 2.76 (dq, $J = 13.2$, 7.2 Hz, 1H), 2.66 (dq, $J = 13.6$, 6.4 Hz, 1H), 2.54 (octet, $J = 4.0$ Hz, 1H), 2.31 (m, 2H), 2.18 (m, 1H), 2.01–1.95 (m, 3H), 1.92–1.18 (m, 24H including s at 1.35 and 1.26), 0.98–0.79 (m, 14H including d at 0.95 with $J = 5.6$ Hz, 0.93 with $J = 6.0$ Hz, 0.85 with $J = 7.6$ Hz, and 0.82 with $J = 7.2$ Hz). ^{13}C NMR (100 MHz, CDCl_3) δ 175.8, 160.8, 134.3, 129.8, 129.7, 115.3, 115.1, 103.4, 102.9, 100.8, 88.6, 88.4, 81.2, 81.1, 76.4, 73.7, 52.5, 52.4, 44.7, 44.5, 44.3, 43.3, 37.4, 37.2, 36.5, 34.5, 33.3, 32.9, 30.2, 29.9, 26.2, 26.0, 24.9, 24.8, 24.6, 24.5, 20.2, 13.5, 13.0. ^{19}F NMR (282 MHz, CDCl_3) δ –115.7. HRMS (FAB) calculated for $\text{C}_{41}\text{H}_{59}\text{FNO}_9$ $[(\text{M} + \text{H})^+]$ 728.4174, found 728.4177.

Synthesis of Trioxane Monomer Fluorobenzyl Amide (4a). Into a flame-dried 5 mL RBF was charged acid **8**¹¹ (75 mg, 0.22 mmol), *N*-(3-dimethylaminopropyl)-*N'*-ethylcarbodiimide hydrochloride (67 mg, 0.35 mmol), and 1-hydroxybenzotriazole (35 mg, 0.26 mmol). Dichloromethane (2.5 mL) was then added, and the mixture was stirred for an hour, at which time 4-fluorobenzylamine (95 μL , 0.84 mmol) was added by syringe. The reaction was allowed to stir at room temperature for 3 h. It was then quenched with 1 N HCl, extracted with dichloromethane (3 \times 5 mL), washed with aqueous NaHCO_3 and brine, dried over magnesium sulfate, and evaporated. The crude product was purified by preparative thin layer chromatography (silica gel, 100% diethyl ether) to afford **4a** as an amorphous, white solid (69 mg, 0.15 mmol, 70%). IR (thin film) 3321, 2947, 2875, 1648, 1546, 1510, 1453, 1378, 1223, 1127, 1095, 1052, 1011, 940, 879, 822, 756 cm^{-1} . ^1H NMR (400 MHz, CDCl_3) δ 7.25 (m, 2H), 7.00 (m, 2H), 6.08 (br. s, 1H), 5.27 (s, 1H), 4.40 (m, 2H), 4.09 (m, 1H), 2.69 (m, 1H), 2.46 (m, 1H), 2.33 (m, 2H), 2.04–1.77 (m, 5H), 1.64–1.55 (m, 2H), 1.48–1.20 (m, 7H, including singlet at 1.36), 0.99–0.92 (m, 4H), 0.85 (d, 3H, $J = 6.0$ Hz). ^{13}C NMR (100 MHz, CDCl_3) δ 172.97, 163.40, 160.96, 134.27, 129.54, 115.41, 103.28, 88.91, 81.14, 75.75, 52.44, 44.45, 43.01, 37.43, 36.58, 34.64, 34.48, 30.22, 26.11, 25.15, 24.89, 24.68, 20.12, 13.00; $[\alpha]_{\text{D}}^{22} = +72$ ($c = 0.97$, CHCl_3). HRMS (FAB) m/z calcd for $\text{C}_{25}\text{H}_{34}\text{FNO}_5\text{Na}$ $(\text{M} + \text{Na})^+$ 470.2313, found 470.2300.

Synthesis of Trioxane Monomer Fluoroanilide (4b). Into a flame-dried 5 mL RBF was charged carboxylic acid **8**¹¹ (55 mg, 0.16 mmol), *N*-(3-dimethylaminopropyl)-*N'*-ethylcarbodiimide hydrochloride (50 mg, 0.26 mmol), and 1-hydroxybenzotriazole (30 mg, 0.19 mmol). Dichloromethane (2.5 mL) was then added and the mixture was stirred for an hour, at which time 4-fluoroaniline (60 μL , 0.61 mmol) was added by syringe. The reaction was allowed to stir at room temperature for 3 h. It was

then quenched with 1 N HCl, extracted with dichloromethane (3 \times 5 mL), washed with aqueous NaHCO_3 and brine, dried over magnesium sulfate, and evaporated. The crude product was purified by flash column chromatography (silica gel, 30% ethyl acetate/hexanes) to afford **4b** (artefanilide) as an amorphous, white solid (51 mg, 0.12 mmol, 75%). IR (thin film) 3313, 2939, 2874, 1663, 1614, 1543, 1509, 1451, 1406, 1377, 1212, 1124, 1091, 1055, 1012, 876, 835, 754 cm^{-1} . ^1H NMR (400 MHz, CDCl_3) δ 8.07 (br. s, 1H), 7.50 (m, 2H), 6.96 (m, 2H), 5.34 (s, 1H), 4.14 (m, 1H), 2.73 (m, 1H), 2.60 (m, 1H), 2.46 (m, 1H), 2.31 (m, 1H), 2.02–1.78 (m, 5H), 1.65–1.55 (m, 2H), 1.47–1.20 (m, 7H, including singlet at 1.35), 0.97–0.93 (m, 4H), 0.87 (d, 3H, $J = 7.6$ Hz). ^{13}C NMR (100 MHz, CDCl_3) δ 171.33, 157.89, 134.17, 121.63, 115.33, 103.42, 88.79, 81.08, 76.12, 52.30, 44.31, 37.34, 36.42, 35.62, 34.32, 30.85, 30.13, 25.98, 24.82, 24.55, 20.09, 13.04; $[\alpha]_{\text{D}}^{22} = +60$ ($c = 0.47$, CHCl_3). HRMS (FAB) m/z calcd for $\text{C}_{24}\text{H}_{33}\text{FNO}_5$ $(\text{M} + \text{H})^+$ 434.2343, found 434.2335.

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Supporting Information Available: ^1H , ^{13}C NMR spectra for all of the new trioxanes **3**, **4a**, and **4b**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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