Palladium-Catalyzed Stereoselective Synthesis of (E,E)-1-Arylselenobutadienes

Xian Huang,* Liu-Sheng Zhu

Department of Chemistry, Hangzhou University, Hangzhou, 310028, People's Republic of China Fax +86(0571)8070107

Received 11 December 1995; revised 4 April 1996

(E)-Vinylic zirconium(IV) complexes, readily obtainable via hydro-zirconation of alk-1-ynes, react with (E)-1-halo-2-arylselenoethylenes in the presence of catalytic amounts of tetrakis(triphenylphosphane)palladium to produce (E,E)-1-arylselenobutadienes in high yields.

Stereodefined conjugated dienes represent a class of important intermediates and a variety of natural products of biological interest, such as Achillea amide¹ and Pellitorine.² The synthesis of dienes for use in the Diels–Alder reaction³ is still an important challenge in synthetic organic chemistry.⁴ One particularly attractive approach has been the incorporation of heteroatom substituents that can activate the diene,⁵ thereby extending the range of workable dienophiles, and also provide a focal point for later synthetic elaboration.

To take advantage of the *endo* stereoselectivity, the *ortho* and *para*-directing regioselectivity and the *cis* stereochemistry of the Diels-Alder cycloaddition, ⁶ it is implicit that the diene should be both stereochemically defined and configurationally stable under the reaction conditions. Thus, a 1,4-disubstituted acyclic diene of the type 1, having the *E,E* stereochemistry and an *S-cis* planar conformation, is most able to overlap in the preferred *endo* transition state; other stereoisomers of 1, such as the *E,Z*, the *Z,E*, and the *Z,Z* would encounter sterior repulsion either in attaining the necessary *S-cis* conformation or in the formation of the *endo* transition state.

In the literature, we found the stereoselective synthesis of (E,E)-1-trimethylsilylbutadienes^{7,8} and (E,E)-1-phenylthiobutadienes.⁹ However, there are few reports on the synthesis of (E,E)-1-arylselenobutadienes.¹⁰ There was a highly stereo- and regioselective procedure for the synthesis of conjugated dienes by a palladium-catalyzed

reaction of alkenyl zirconium complexes with alkenyl halides. ¹¹ We tried to employ the method to develop an efficient route to (E,E)-1-arylselenobutadienes 5, and have indeed found that the selenium-containing conjugated diene synthesis shown in Scheme 1 can be successfully achieved with (E)-1-halo-2-arylselenoethylenes 4 in place of the corresponding alkenyl halides.

Table. Synthesis of Products 5a-d

2	4	Product (5)	Yield (%)
2 a	4a	5a	71
2a	4 b	5b	65
2a	4c	5a	82
2a	4d	5b	75
2b	4a	5c	73
2b	4 b	5d	68
2 b	4c	5c	78
2b	4 d	5d	83

^a Yield of isolated product

The yields of the (E)-alkenylzirconium derivatives obtained by reaction of $\mathrm{Cp_2Zr(H)Cl}$ with alkynes in a 1:1 molar ratio have been typically in the range 75–90% by $^1\mathrm{H~NMR.^{12}}$ Thus, it was necessary to use modest excesses of $\mathrm{Cp_2Zr(H)Cl}$ and alkynes for the complete conversion of alkenyl halides into the desired cross-coupled products (see Table).

The starting materials, (E)-1-halo-2-arylselenoethylenes **4**, were prepared by hydrozirconation of arylselenoethynes, followed by trapping with iodine or N-bromosuccinimide (NBS), which is shown in Scheme $2.^{13}$

All products were fully characterized. The $^1\mathrm{H}$ NMR of (E)-vinylic selenides give rise to a doublet at $\delta=6-7$ with a coupling constant 14–16 Hz. 14 The E,E configuration of the products 5 is supported by $J_{\mathrm{H,H}}$ of 14.6–15.7 Hz which could be measured. 15

$$RC \equiv CH \xrightarrow{C_{P_1}Z_r(H)Cl} \xrightarrow{R} \xrightarrow{H} \xrightarrow{X} \xrightarrow{H} \xrightarrow{(4)} \xrightarrow{R} \xrightarrow{H} \xrightarrow{SeAr} \xrightarrow{(4)} \xrightarrow{R} \xrightarrow{H} \xrightarrow{H} \xrightarrow{SeAr} \xrightarrow{SeAr} \xrightarrow{H} \xrightarrow{SeAr} \xrightarrow{SeAr}$$

$$ArSeC = CH \xrightarrow{Cp_2Zr(H)Cl} \xrightarrow{H} ArSe \xrightarrow{H} \xrightarrow{l_2 \text{ or NBS}} ArSe \xrightarrow{H} X$$

$$Ar = Ph$$
, $4-MeC_6H_4$; $X = I$, Br

Scheme 2

1192

All reactions were carried out under N₂. THF was distilled from sodium benzophenone ketyl. Mps were uncorrected. ¹H NMR spectra were recorded on a AC-P 200 (200 MHz) spectrometer with TMS as internal standard in CDCl₃. IR spectra were taken on a Shimadzu IR-435 spectrometer. MS spectra were obtained on a HP 5890 A mass spectrometer and microanalyses were measured using a Yanaco MT-3CHN microelemental analyser.

Synthesis of 5a-d; General Procedure:

Short Papers

A mixture of $Cp_2Zr(H)Cl$ (0.8 mmol) and 2 (0.8 mmol) in THF (4 mL) was stirred at r.t. for 20 min. To the resulting solution was added 4 (0.6 mmol) and $Pd(PPh_3)_4$ (0.6 × 5% mmol) and this was stirred at r.t. for 3 h. It was then diluted with Et_2O and after 5 min of additional stirring, the supernatant liquid was filtered through a short plug of silica gel. After removal of solvent, the residue was purified by preparative TLC on silica gel [petroleum ether (bp 30–60°C) as eluent for 5a-b, $Et_2O/petroleum$ ether (1:20) for 5c-d].

(E,E)-4-Phenyl-1-phenylselenobuta-1,3-diene (5a): mp 40-41 °C. IR (KBr): $v = 3041, 1591, 980, 734, 686 \text{ cm}^{-1}$.

¹H NMR: $\delta = 7.26-7.34$ (m, 10 H), 6.42–7.12 (m, 4 H).

MS: m/z = 286.05 (M⁺, 21.00), 128.20 (100.00).

Anal. Calcd. for $C_{16}H_{14}Se$: C, 67.38; H, 4.95. Found C, 67.05; H, 5.08.

(E,E)-4-Phenyl-1-(4-methylphenylseleno)buta-1,3-diene (5b):

IR (film): $v = 3102, 1593, 982, 800, 687 \text{ cm}^{-1}$.

¹H NMR: $\delta = 7.05 - 7.49$ (m, 9 H), 6.42 - 7.85 (m, 4 H); 2.39 (s, 3 H).

MS: m/z = 300.05 (M⁺, 25.93), 128.05 (100.00).

Anal. Calcd. for C₁₇H₁₆Se: C, 68.24; H, 5.39.

Found C, 68.53; H, 5.30.

(E,E)-5-Methoxy-1-phenylselenopenta-1,3-diene (5c):

IR (film): v = 3035, 2905, 1574, 993, 733, 686 cm⁻¹.

¹H NMR: δ = 7.29–7.50 (m, 5 H), 6.25–6.74 (m, 3 H); 5.64–5.72 (m, 1 H), 3.95 (d, 2 H), 3.30 (s, 3 H).

MS: $m/z = 254.00 \text{ (M}^+, 11.33), 97.15 (100.00).$

Anal. Calcd. for C₁₂H₁₄OSe: C, 56.94; H, 5.57. Found C, 56.82; H, 5.71.

SYNTHESIS

(E,E)-5-Methoxy-1-(4-methylphenylseleno)penta-1,3-diene (5d):

IR (film): v = 3010, 2905, 1569, 978, 801 cm⁻¹.

¹H NMR: δ = 7.11–7.43 (q, 4 H), 6.19–6.71 (m, 3 H); 5.62–5.70 (m, 1 H), 3.95 (d, 2 H), 3.34 (s, 3 H), 2.26 (s, 3 H).

MS: m/z = 268.00 (M⁺, 20.02), 97.05 (100.00).

Anal. Calcd. for $C_{13}H_{16}OSe:$ C, 58.44; H, 6.04. Found C, 58.80; H, 6.38.

We thank the National Natural Science Foundation of China and the Natural Science Foundation of Zhejiang Province for financial support.

- Huang, Y.Z.; Shi, L.; Yang, J.; Zhang, J. Tetrahedron Lett. 1987, 28, 2159.
- Shi, L.; Yang, J.; Wen, X.; Huang, Y. Z. Tetrahedron Lett. 1988, 29, 3949.
- (3) Oppolzer, W. Angew. Chem., Int. Ed. Engl. 1984, 23, 876.
- (4) March, J. Advances Organic Chemistry; 2nd ed.; McGraw-Hill: Toronto, 1977.
- (5) Petrzika, M.; Grayson, J.I. Synthesis 1981, 753.
- (6) Danishefsky, S.; Kitahara, T.; Yan, C.F.; Morris, J. J. Am. Chem. Soc. 1979, 101, 6996.
- (7) Fiandanese, V.; Marchese, G.; Mascolo, G.; Naso, F.; Ronzini, L. Tetrahedron Lett. 1988, 29, 3705.
- (8) Chan, T. H.; Li, J. S. J. Chem. Soc., Chem. Commun. 1982, 969.
- (9) Naso, F. Pure Appl. Chem. 1988, 60, 79.
- (10) Comasseto, J. V.; Brandt, C. A. Synthesis 1987, 146.
- (11) Okukado, N.; Van Horn, D.E.; Klima, W.L.; Negishi, E. Tetrahedron Lett. 1978, 1027.
- (12) Negishi, E.; Van Horn, D. E. J. Am. Chem. Soc. 1977, 99, 3168.
- (13) This method for preparation of (E)-1-halo-2-arylselenoethylenes was invented by us.
- (14) Braga, A.L.; Reckziegel, A.; Silverira, C.C. Synth. Commun. 1994, 24, 1165.
- (15) Lee, C.W.; Koh, Y.J.; Oh, D.Y. J. Chem. Soc., Perkin Trans. 1 1994, 717.