

# Development of a Highly Selective Fluorescence Probe for Hydrogen Sulfide

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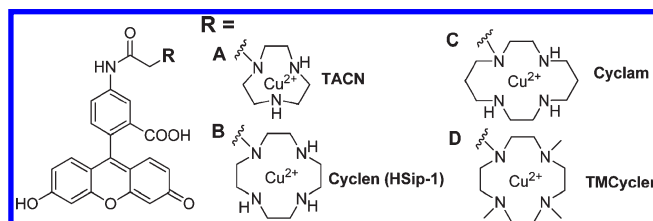
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 Supporting Information

**ABSTRACT:** Hydrogen sulfide (H<sub>2</sub>S) has recently been identified as a biological response modifier. Here, we report the design and synthesis of a novel fluorescence probe for H<sub>2</sub>S, HSip-1, utilizing azamacrocyclic copper(II) ion complex chemistry to control the fluorescence. HSip-1 showed high selectivity and high sensitivity for H<sub>2</sub>S, and its potential for biological applications was confirmed by employing it for fluorescence imaging of H<sub>2</sub>S in live cells.

Although hydrogen sulfide (H<sub>2</sub>S) is a toxic gas with the characteristic smell of rotten eggs, recent studies have regarded H<sub>2</sub>S as the third gaseous transmitter, in addition to nitric oxide (NO) and carbon monoxide (CO). H<sub>2</sub>S appears to be involved in various physiological processes, including relaxation of vascular smooth muscles,<sup>1</sup> mediation of neurotransmission,<sup>2</sup> inhibition of insulin signaling,<sup>3</sup> and regulation of inflammation and O<sub>2</sub> sensing.<sup>4,5</sup> There are several methods for selective detection of H<sub>2</sub>S, among which the most commonly used are the methylene blue method and the sulfide ion-selective electrode method.<sup>6,7</sup> However, these methods are destructive, requiring homogenization of samples. Therefore, for detailed studies of the physiological functions of H<sub>2</sub>S, a new method to measure H<sub>2</sub>S concentration in cells is required. We focused on fluorescence imaging, because it is suitable for nondestructive detection of targeted biomolecules in live cells or tissues with readily available instruments.<sup>8</sup> Thus, we set out to develop a sensitive and selective fluorescence probe for H<sub>2</sub>S. This involves two substantial challenges. One is to attain sufficient selectivity over other biothiols; i.e., the probe has to be able to measure endogenous H<sub>2</sub>S without interference from other biothiols, including reduced glutathione (GSH, present at levels of about 1–10 mM), L-cysteine (L-Cys, about 100 μM), and thiol-containing proteins.<sup>9</sup> The other challenge is to achieve sufficient sensitivity, because the concentration of H<sub>2</sub>S required to elicit physiological responses has been reported to be 10 μM–1 mM, although there remains some controversy as to the intracellular H<sub>2</sub>S levels generated in response to physiological stimuli.<sup>1–3,10</sup>

A few fluorescence probes for H<sub>2</sub>S have been reported, utilizing sulfide anion-triggered removal of a 2,4-dinitrobenzenesulfonyl group or reduction of azide to amine.<sup>11–13</sup> However, the reactions are relatively slow and show poor selectivity for H<sub>2</sub>S over reactive oxygen species (ROS), such as superoxide (O<sub>2</sub><sup>•−</sup>) or Na<sub>2</sub>SO<sub>3</sub>.<sup>12,14</sup> Recently, Chang et al. have reported that a dipicolylamine



**Figure 1.** Structures of four macrocyclic fluorescein–Cu<sup>2+</sup> conjugates: (A) TACN-AF+Cu<sup>2+</sup>, (B) Cyclen-AF+Cu<sup>2+</sup> (HSip-1), (C) Cyclam-AF+Cu<sup>2+</sup>, and (D) TMCyclen-AF+Cu<sup>2+</sup>.

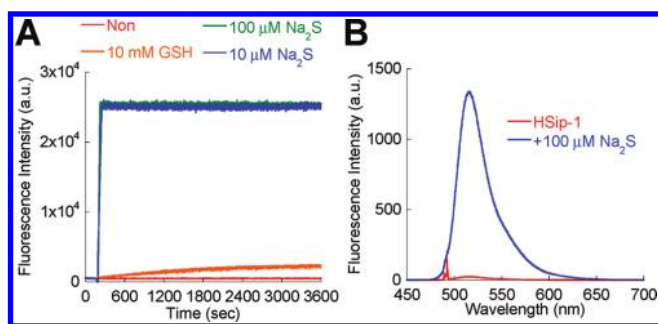
(DPA)—fluorescein complex with Cu<sup>2+</sup> (DPA-AF+Cu<sup>2+</sup>) showed a selective turn-on fluorescence response to sulfide anion.<sup>15</sup> We confirmed that DPA-AF+Cu<sup>2+</sup> could detect 10 μM H<sub>2</sub>S in an aqueous solution at pH 7.4 (Supporting Information, Figure S1), but this probe also showed fluorescence enhancement upon addition of 10 mM GSH. We hypothesized that improvement of the chelator–Cu<sup>2+</sup> complex stability in the presence of biothiols would result in a good selectivity for H<sub>2</sub>S at low concentrations in the presence of high concentrations of GSH. It is well known that azamacrocyclic rings form stable metal complexes with Cu<sup>2+</sup>, and the paramagnetic Cu<sup>2+</sup> center has a pronounced quenching effect on fluorophores.<sup>16</sup> On the basis of these facts, we expected that Cu<sup>2+</sup> would be released from the azamacrocyclic ring when H<sub>2</sub>S binds to the Cu<sup>2+</sup> center, resulting in fluorescence enhancement, whereas the azamacrocyclic Cu<sup>2+</sup> complex would retain its structure in the presence of high concentrations of GSH, showing no fluorescence enhancement.

On the basis of this hypothesis, we designed and synthesized four sensor probes based on a fluorescein scaffold conjugated with an azamacrocyclic Cu<sup>2+</sup> complex, employing 1,4,7-triazacyclononane (TACN), 1,4,7,10-tetraazacyclododecane (Cyclen), 1,4,8,11-tetraazacyclotetradecane (Cyclam), and *N,N',N''*-trimethylcyclen (TMCyclen) as chelators for Cu<sup>2+</sup> instead of DPA (Figure 1 and Supporting Information, Schemes S1–S4).

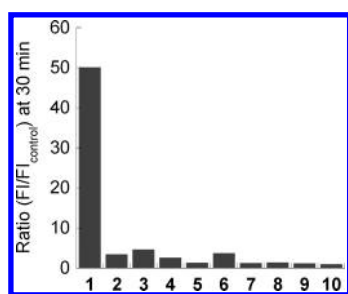
Next, we examined the sensitivity and selectivity of the synthesized compounds for H<sub>2</sub>S over GSH by measuring the fluorescence increment upon addition of 10 μM H<sub>2</sub>S or 10 mM GSH. TACN-AF+Cu<sup>2+</sup> showed high sensitivity and low selectivity (Figure S2), while Cyclam-AF+Cu<sup>2+</sup> (Figure S3) and TMCyclen-AF+Cu<sup>2+</sup> (Figure S4) showed low sensitivity and high

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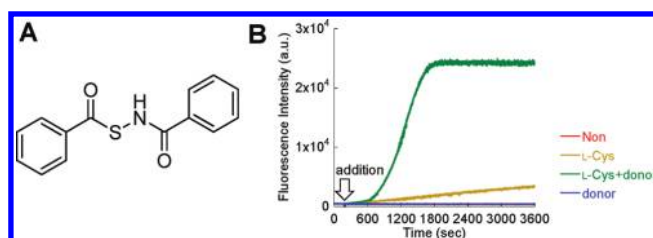


**Figure 2.** (A) Time course of reactions of HSip-1 with no addition (red) or addition of 10 mM GSH (orange), 100  $\mu$ M Na<sub>2</sub>S (green), or 10  $\mu$ M Na<sub>2</sub>S (blue) in 30 mM HEPES buffer (pH 7.4) at 37 °C. Na<sub>2</sub>S and GSH were added at 180 s. Ex/Em = 491/516 nm. (B) Fluorescence spectra of 1  $\mu$ M HSip-1 before (red) and after (blue) reaction with 100  $\mu$ M Na<sub>2</sub>S in 30 mM HEPES buffer (pH 7.4).

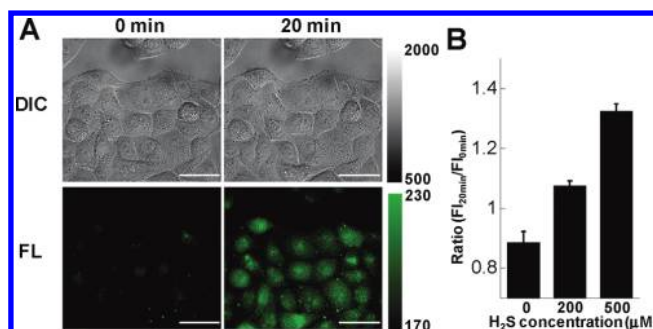


**Figure 3.** Fluorescence enhancement of 1  $\mu$ M HSip-1 in the presence of thiols, inorganic sulfur compounds, and sodium ascorbate. Bars represent relative fluorescence intensity 30 min after addition of thiols, inorganic sulfur compounds, or sodium ascorbate. The reactions of 1  $\mu$ M HSip-1 with (1) 10  $\mu$ M Na<sub>2</sub>S, (2) 10 mM GSH, (3) 1 mM L-Cys, (4) 1 mM DL-Hcy, (5) 1 mM 2-ME, (6) 100  $\mu$ M DTT, (7) 1 mM NaSCN, (8) 1 mM Na<sub>2</sub>SO<sub>3</sub>, (9) 1 mM Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>, and (10) 10 mM sodium ascorbate were performed in 30 mM HEPES buffer (pH 7.4) at 37 °C.

selectivity. Fortunately, Cyclen-AF+Cu<sup>2+</sup> (HSip-1 = hydrogen sulfide imaging probe-1) showed excellent properties as a fluorescence probe for H<sub>2</sub>S (Abs<sub>max</sub>/Em<sub>max</sub> = 491/516 nm,  $\Phi_f$  = 0.019 in 30 mM HEPES buffer at pH 7.4); i.e., it showed a large and immediate increment of fluorescence intensity by 50-fold upon addition of 10  $\mu$ M H<sub>2</sub>S, whereas almost no fluorescence increment was observed upon addition of 10 mM GSH (Figure 2). In addition, the fluorescence enhancement induced by H<sub>2</sub>S was retained even in the presence of 10 mM GSH (Figure S5). HSip-1 also showed high selectivity over other thiols (1 mM L-Cys, DL-Hcy, 2-mercaptoethanol (2-ME), and 100  $\mu$ M dithiothreitol (DTT)), inorganic sulfur compounds (1 mM NaSCN, Na<sub>2</sub>SO<sub>3</sub>, and Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>), and a reducing condition (10 mM sodium ascorbate) (Figures 3 and S6). Moreover, HSip-1 did not show any fluorescence enhancement in response to ROS or reactive nitrogen species (RNS), such as hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>), hydroxyl radical ( $\cdot$ OH), peroxynitrite (ONOO<sup>-</sup>), hypochlorite (<sup>-</sup>OCl), superoxide (O<sub>2</sub><sup>-</sup>), singlet oxygen (<sup>1</sup>O<sub>2</sub>), and nitric oxide ( $\cdot$ NO). Upon addition of Angeli's salt (NO<sup>-</sup> donor), a small fluorescence enhancement was observed (Figures S7 and S8). Thus, HSip-1 offers high selectivity for H<sub>2</sub>S compared with previously reported fluorescence probes utilizing a 2,4-dinitrosulfonyl group or an azide group.



**Figure 4.** (A) Structure of cysteine-activated H<sub>2</sub>S donor. (B) Time course of reactions of 1  $\mu$ M HSip-1 with no addition (red) or addition of 50  $\mu$ M H<sub>2</sub>S donor (blue), 1 mM L-Cys (yellow), or 50  $\mu$ M H<sub>2</sub>S donor + 1 mM L-Cys (green) in 30 mM HEPES buffer (pH 7.4) with 0.1% tetrahydrofuran as a cosolvent at 37 °C. H<sub>2</sub>S donor and L-Cys were added at 180 s, as indicated by an arrow. Ex/Em = 491/516 nm.



**Figure 5.** Visualization of H<sub>2</sub>S in live cells using HSip-1 DA. HeLa cells were incubated with 30  $\mu$ M HSip-1 DA in DMEM containing 0.3% DMSO for 30 min. (A) Differential interference contrast (DIC) and fluorescence (FL) images were captured after addition of 500  $\mu$ M Na<sub>2</sub>S in HBSS solution. (B) Average FI<sub>20 min</sub>/FI<sub>0 min</sub> intensity ratios in fluorescence images after addition of 0, 200, or 500  $\mu$ M Na<sub>2</sub>S in HBSS buffer. The excitation and emission wavelengths were 470–490 and 515–550 nm, respectively. Scale bar, 10  $\mu$ m. Representative fluorescence images from replicate experiments (*n* = 3) are shown. Error bars are  $\pm$ SD.

Xian et al. have reported cysteine-activated H<sub>2</sub>S donors which mimic the slow and continuous endogenous biosynthesis of H<sub>2</sub>S from L-Cys mediated by enzymes such as cystathionine  $\beta$ -synthase (CBS) and cystathionine  $\gamma$ -lyase (CSE).<sup>17</sup> We synthesized this H<sub>2</sub>S donor (Figure 4A) to examine the feasibility of using HSip-1 to monitor pseudo-enzymatic H<sub>2</sub>S production. We could detect time-dependent H<sub>2</sub>S release associated with decomposition of the H<sub>2</sub>S donor by using HSip-1 in the presence of 1 mM L-Cys (Figure 4B).

The spectroscopic properties of HSip-1, as well as its selectivity for H<sub>2</sub>S, seemed appropriate for cellular application, so we next considered the suitability of HSip-1 for fluorescence imaging of cellular H<sub>2</sub>S. However, this probe is membrane-impermeable because of its hydrophilicity derived from the carboxyl group and the cyclen-Cu<sup>2+</sup> complex moiety. Therefore, we employed a standard diacetylation approach to improve the membrane permeability and synthesized diacetylated HSip-1 (HSip-1 DA); this was confirmed to be a membrane-permeable precursor that is hydrolyzed to HSip-1 by intracellular esterases (Scheme S2). Next, HeLa cells were incubated with 30  $\mu$ M HSip-1 DA containing 0.3% DMSO as a cosolvent in Dulbecco's modified Eagle's medium (DMEM) for 30 min and then washed with Hanks' Balanced Salt Solutions (HBSS), and various concentrations of Na<sub>2</sub>S (0, 200, or 500  $\mu$ M) were added to the medium. Upon addition of 500  $\mu$ M Na<sub>2</sub>S, a large intracellular fluorescence enhancement was observed, while no significant fluorescence

increment was seen in the absence of Na<sub>2</sub>S (Figure S). The results of CCK-8 assay showed that HSip-1 DA exhibits no cytotoxicity at concentrations up to 100  $\mu$ M (Figure S9).

In summary, we have developed a novel fluorescence probe for H<sub>2</sub>S, HSip-1, based on azamacrocyclic Cu<sup>2+</sup> complex chemistry. HSip-1 can sensitively detect H<sub>2</sub>S in aqueous solution with high selectivity over biothiols, inorganic sulfur compounds, ROS, and RNS and has excellent photophysical properties for biological applications, arising from its fluorescein scaffold. We confirmed that HSip-1 could be used both for detection of pseudo-enzymatic H<sub>2</sub>S release in a cuvette and for real-time fluorescence imaging of intracellular H<sub>2</sub>S in live cells. In addition, we also confirmed that HSip-1 could detect H<sub>2</sub>S produced by 3-mercaptopyruvate sulfurtransferase (3MST), or lysate of 3MST-expressing cells, and 3-mercaptopyruvate, i.e., H<sub>2</sub>S-producing enzyme and its substrate (Figure S10).<sup>18</sup> We anticipate that HSip-1 will be useful for high-throughput screening of CBS, CSE, and 3MST agonists and antagonists, as well as for detailed investigation of a wide range of biological functions of H<sub>2</sub>S.

## ■ ASSOCIATED CONTENT

**S Supporting Information.** Synthesis; experimental details; characterization of developed compounds; and experiments using living cells. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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