

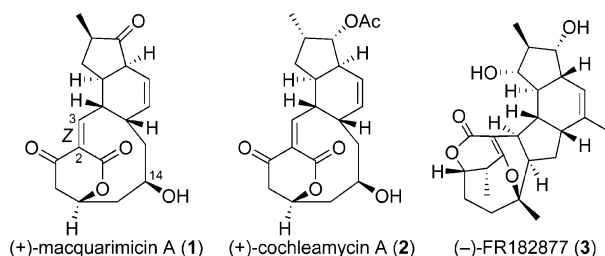
Total Synthesis of (+)-Macquarimicin A

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(+)-Macquarimicin A (**1**) was isolated from *Micromonospora chalicea* by researchers at Abbott in 1995.¹ Later, researchers at Sankyo found that **1** is a selective inhibitor of membrane-bound neutral sphingomyelinase (N-SMase) and exhibits antiinflammatory activity in vivo.² The structure of **1** is characterized by a unique tetracyclic framework, which comprises a *cis*-tetrahydroindanone ring, a β -keto- δ -lactone ring, and a 10-membered carbocycle.^{1b}

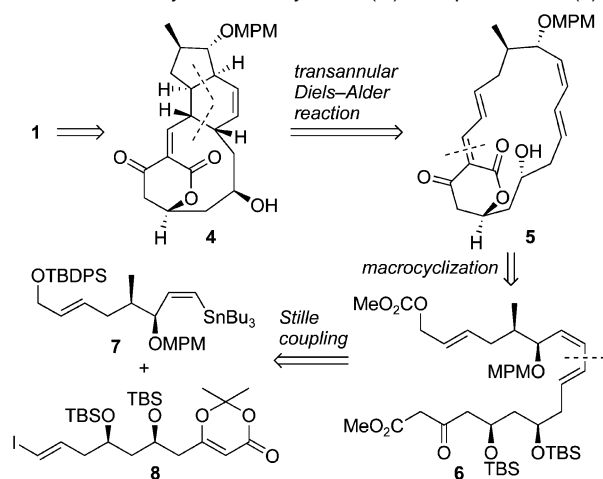
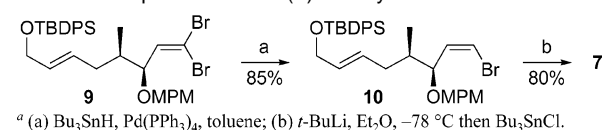
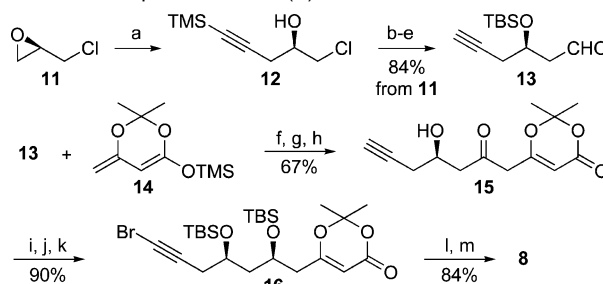


As closely related natural products, an antitumor antibiotic cochleamycin A (**2**)³ and a microtubule-stabilizing agent FR182877 (**3**)⁴ have been isolated. This class of natural products shares a biogenetic hypothesis that involves the intramolecular Diels–Alder (IMDA) reaction of polyketide intermediates.⁵ This intriguing feature, combined with biological activities and a formidable molecular architecture, makes them highly attractive synthetic targets.⁶ In 2002, Sorensen et al.⁷ and Evans and Starr⁸ achieved enantioselective total syntheses of (+)- and (-)-**3**, respectively. Very recently, Tatsuta et al.⁹ disclosed the total synthesis of (+)-**2**. Herein, we describe the first total synthesis of (+)-**1**, determination of its absolute configuration, and revision of the proposed structure concerning the C(2)–C(3) geometry.¹⁰

The retrosynthetic analysis is outlined in Scheme 1.¹¹ The tetracyclic framework of **1** was projected to arise from the transannular Diels–Alder (TADA) reaction¹² of **5**. The macrocycle **5** could be elaborated through the intramolecular Trost–Tsuji reaction of **6**, which in turn would be available via the Stille coupling of (*Z*)-stannylalkene **7** and (*E*)-iodoalkene **8**.

(*Z*)-Stannylalkene **7** was synthesized in two steps from dibromoalkene **9**¹¹ (Scheme 2). The application of Uenishi's method¹³ to **9** generated (*Z*)-bromoalkene **10** exclusively. The halogen–lithium exchange of **10** followed by treatment with Bu₃SnCl produced **7**.

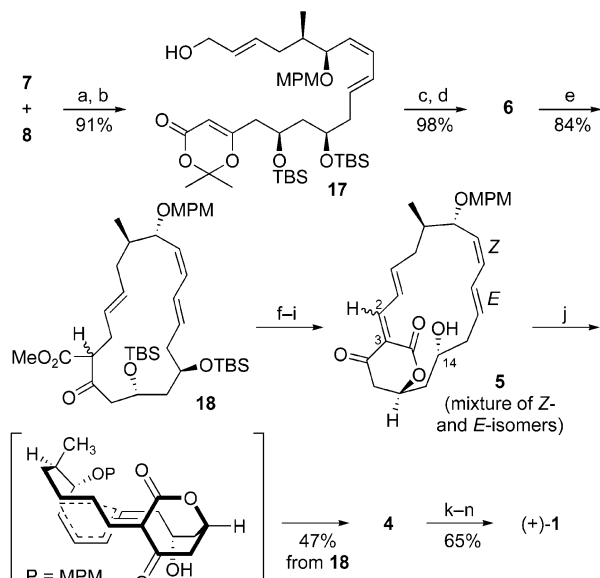
The synthesis of the other coupling substrate **8** started from (*R*)-epichlorohydrin (**11**) via known acetylenic compound **12**¹⁴ (Scheme 3). The conversion of **12** to aldehyde **13** was conducted in a straightforward manner and proceeded in 84% yield from **11**. The vinylogous Mukaiyama aldol reaction between **13** and **14**¹⁵ gave a 1:1 diastereomeric mixture of the adducts, which was converted to β -hydroxyketone **15** in two steps. The diastereoselective reduction¹⁶ of **15** gave the desired *syn*-1,3-diol exclusively. The protection of

Scheme 1. Retrosynthetic Analysis for (+)-Macquarimicin A (**1**)Scheme 2. Preparation of the (*Z*)-Stannylalkene **7**^aScheme 3. Preparation of the (*E*)-Iodoalkene **8**^a

^a (a) trimethylsilylacetylene, *n*-BuLi, BF₃·OEt₂, THF, –78 to –30 °C; (b) KCN, NaI, DMSO/H₂O (10:1); (c) K₂CO₃, MeOH; (d) TBSCl, imidazole, DMF; (e) Dibal-H, toluene, –78 °C; (f) BF₃·OEt₂, CH₂Cl₂, –78 °C; (g) Dess–Martin reagent, CH₂Cl₂; (h) 48% aq. HF/MeCN (5:95); (i) Et₃BOMe, NaBH₄, THF/MeOH (4:1), –78 °C; (j) TBSCl, imidazole, DMF; (k) NBS, AgNO₃, acetone; (l) Bu₃SnH, Pd(dba)₃, PPh₃, THF; (m) I₂, CH₂Cl₂, 0 °C.

the resulting diol and conversion to bromoalkyne **16** followed by one-pot hydrostannylation–iodination¹⁷ produced (*E*)-iodoalkene **8**.

With stannane **7** and iodide **8** in hand, assembly was undertaken (Scheme 4). The cuprous chloride-promoted Stille coupling^{18,19} (97%), followed by selective deprotection²⁰ of the TBDPS group (94%), afforded **17**. Conversion of **17** to the methyl carbonate followed by thermolysis in toluene/MeOH furnished the β -keto ester **6**. Macroallylation²¹ was successfully carried out to form a 17-membered macrocycle **18** (ca. 3:2 diastereomeric mixture) in 84% yield using Pd(PPh₃)₄/dppe (1:1) as a catalyst. After removal of

Scheme 4. Total Synthesis of (+)-**1** via TADA of **5**^a

^a (a) Pd(PPh₃)₄, CuCl, DMSO–THF; (b) NH₄F, MeOH; (c) ClCO₂Me, pyr, CH₂Cl₂; (d) MeOH, toluene, 110 °C, in a sealed tube; (e) Pd(PPh₃)₄, dppe, THF; (f) HF·pyr., pyr.; (g) MeOH–*i*-Pr₃NEt (10:1); (h) PhSeCl, Et₃N, CH₂Cl₂, –78 °C; (i) mCPBA, CH₂Cl₂, –50 °C; (j) BHT, toluene, 130 °C, in a sealed tube; (k) TESOTf, lutidine, CH₂Cl₂, –78 °C; (l) DDQ, CH₂Cl₂/pH 7 buffer (10:1); (m) Dess–Martin reagent, NaHCO₃, CH₂Cl₂; (n) PPTS, MeOH.

the TBS groups in **18**, the formation of the β -keto- δ -lactone ring under basic conditions followed by a double-bond introduction was carried out to afford **5** as a mixture of C(2)–C(3) geometrical isomers.²²

The stage was set for the key TADA reaction. Under thermal conditions (130 °C), the cycloaddition of **5** furnished the desired diastereomer **4** as a sole cycloadduct. In this reaction, the (*Z,E*)-geometry of the reacting diene is the origin of *endo* selectivity,¹¹ while the lactone ring restricts conformation to control the diastereofacial selectivity completely.²³

The cycloadduct **4** was converted to (+)-**1** as follows. Silylation and removal of the MPM group, followed by Dess–Martin oxidation, gave 14-*O*-TES-**1**. Finally, PPTS-catalyzed cleavage of the TES ether afforded (+)-**1**. Spectral properties (¹H and ¹³C NMR and IR) of synthetic (+)-**1** were completely identical to those of a natural sample, and optical rotation of synthetic (+)-**1** ([α]_D²³ = +270; *c* 0.20, MeOH) established the absolute configuration of natural (+)-**1** ([α]_D²⁵ = +285.6; *c* 0.2, MeOH). Furthermore, extensive NOE experiments on synthetic (+)-**1** revealed that the C(2)–C(3) geometry must be *Z*, not *E* as reported^{1b} (see Supporting Information for details).

In conclusion, the first total synthesis of (+)-macquarimicin A (**1**) has been accomplished with 27 linear steps from **11** in 9.9% overall yield (92% average yield per step). The synthesis features the transannular Diels–Alder reaction, which constructed the tetracyclic framework of **1** stereoselectively. Also, the present work established the absolute configuration of (+)-**1** and revised its C(2)–

C(3) geometry. Further study for the syntheses of the other members of this class of natural products is currently underway.

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Supporting Information Available: Experimental procedures and characterization data for all new compounds and details on the determination of the C(2)–C(3) geometry (PDF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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- (22) Compound **5** gave a complicated ¹H NMR spectrum, making elucidation of the *E/Z* ratio difficult. We attribute the complication to the tautomerization of **5** (such as ketalization), in addition to the geometry of the C(2)–C(3) double bond.
- (23) A TADA substrate without the lactone ring did not afford the desired cycloadduct.

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