

CARBOHYDRATE RESEARCH

Carbohydrate Research 300 (1997) 183-189

## Note

# Synthesis of new anhydro and branched-chain cyclitols <sup>1,2</sup>

Zoltán G. Tóth, István F. Pelyvás \*, Csaba Szegedi, Péter Benke, Erika Magyar, Tünde Miklovicz, Gyula Batta, Ferenc Sztaricskai \*

Research Group of Antibiotics of the Hungarian Academy of Sciences, P.O. Box 70, Debrecen H-4010, Hungary

Received 4 October 1996; accepted 3 January 1997

### Abstract

Starting from D-glucose and D-(-)-quinic acid (5) (1S,5R,6S)-5-azido-6-benzyloxycyclohex-2-en-1-ol (3), and the structurally related  $\alpha,\beta$ -unsaturated alcohols 7 and 8, respectively, were prepared. They have been transformed, by treatment with 3-chloroperoxybenzoic acid, into (1R,2S,3R,5S,6R)-3-azido-2-benzyloxy-5,6-epoxycyclohexane-1-ol (4) and the two diastereoisomeric 4,5-isopropylidenedioxycyclohexane-1-ols 9 and 10. Thermal Claisen rearrangement of the allylic alkcohols 3, 7 and 8 resulted in the functionalized branched-chain cyclohexenyl acetamides 12, 13 and 14, respectively. The prepared new cyclitols are useful starting materials for further derivatization to obtain novel enzyme-inhibitors, including phosphorylated cyclitols with "second-messenger" properties. © 1997 Elsevier Science Ltd.

Keywords: Cyclitols; Enzyme inhibitors

Since the discovery of the role of D-myo-inositol 1,4,5-triphosphate (Ins  $P_3$ ) as an intracellular second messenger for calcium mobilization [1], as well as certain anhydro and aminocyclitol derivatives (such as cyclophellitol [2] and conduramines [3], respectively) possessing glycosidase enzyme inhibitory activity [4], interest in the chemical synthesis of cyclitol derivatives [5] has greatly increased. In continuation of our research aimed at the preparation of chiral

cyclitols from carbohydrates [6] and related substances, we now report on the synthesis of novel azido, anhydro and branched-chain cyclitols for enzyme-inhibitory investigations and further chemical derivatization, including phosphorylation.

The carbohydrate-based synthesis of anhydrocyclitols started from (2S, 3R, 5R)-3-azido-2-benzyloxy-5-hydroxycyclohexanone (1), prepared [7] in this laboratory from D-glucose in eleven steps with a ca. 6% overall yield.  $\beta$ -Elimination of the C-5 hydroxyl group of 1 was effected by treatment with methanesulfonyl chloride in pyridine to give the unstable enone 2 as shown by the appearance of two olefinic protons at  $\delta$  7.0 and 6.02 ppm in the <sup>1</sup>H NMR spectrum. For the stereoselective reduction of the keto function in 2, the Luche procedure [8] employ-

<sup>\*</sup> Corresponding authors.

<sup>&</sup>lt;sup>1</sup> A part of this work was presented at the 5th International Conference on Chemical Synthesis of Antibiotics and Related Microbial Products, Debrecen (Hungary), 1–6 September, 1996, Abstr. OL-8.

<sup>&</sup>lt;sup>2</sup> Dedicated to Professor Dr. Hans Paulsen on the occasion of his 75th birthday.



ing sodium borohydride and cerium(III) chloride in methanol was found the best, affording a ca. 9:1 mixture of the (1S,5R,6S)-allyl alcohol **3** and the corresponding C-1 diastereoisomer (the latter of which could not be isolated in pure form). The values of the <sup>1</sup>H NMR coupling constants observed for **3** ( $J_{1.6}$  7,  $J_{5.6} = J_{4a.5} = 10.5$ , and  $J_{4e.5}$  6 Hz) revealed that the C-1 hydroxyl group is *trans-pseudoequatorially* oriented to the C-6 benzyloxy function in the favoured  ${}^{6}H_{5}$  half-chair conformer.

For obtaining the desired anhydrocyclitols the olefinic bond of the allyl alcohol 3 was epoxidized with 3-chloroperoxybenzoic acid to furnish a separable 8:1 mixture of two compounds. The major epox-

ide **4**, isolated in 62% yield, had an (1R,2S,3R,5S,6R) absolute configuration as proved by NOE difference experiments. Thus, irradiation of the <sup>1</sup>H NMR spectrum of **4** with the frequencies of H-4a and H-4e caused a 1.5% and 5% NOE enhancement, respectively, of the H-5 signal ( $\delta$  3.18). This observation, as well as that irradiation with the H-1 frequency, resulted in a 6% NOE enhancement of the signal of H-6 at  $\delta$  = 3.25 clearly proved that the protons H-5 and H-6 are closer to H-4e and H-1, and thus the epoxy function, the C-1 hydroxyl group, and the H-4a proton are located on the same side of the cyclohexane ring.

A noncarbohydrate-based route to anhydrocyclitols structurally related to **4** involved conversion of D-(-)-quinic acid (**5**) into (4S,5R)-4,5-O-isopropylidenedioxycyclohex-5-en-4-one (**6**) by means of modifications of known procedures [9,10]. Then the Luche reduction of **6**, as described for **3**, afforded a 2:1 separable mixture of the two  $\alpha$ , $\beta$ -unsaturated alcohols (1*S*)-**7** and (1*R*)-**8**. Based on the values of the <sup>1</sup>H NMR coupling constants measured for the major product **7** ( $J_{1,6a} = J_{5,6a} = 9$ ,  $J_{1,6e}$  4.5,  $J_{5,6e}$  4.5,  $J_{3,4}$  2, and  $J_{4,5}$  5.5 Hz) the C-1 hydroxyl group is *pseudoequatorial* and *cis*-oriented to the dioxolane ring (<sup>5</sup>H<sub>6</sub> conformation). In the minor isomer **8** the hydroxyl group at C-1 is also *pseudoequatorial* ( $J_{1,6a}$ 10 and  $J_{1,6e}$  4.5 Hz), but *trans* to the acetal function in the favoured <sup>6</sup>H<sub>5</sub> conformer ( $J_{5,6a}$  2.5 and  $J_{5,6e}$  2 Hz).

According to <sup>13</sup>C NMR data, epoxidation of the allyl alcohols 7 and 8 with 3-chloroperoxybenzoic acid in dichloromethane produced the anhydrocyclitols 9 and 10 as the major products, with 92 and 86%diastereomeric excess, respectively. The conformation of 9 and 10 and the steric arrangement of the anhydro ring was studied by means of <sup>1</sup>H NMR. COSY, and NOE experiments. Thus, the small values (2.0-4.5 Hz) of the coupling constants observed for 9. as well as irradiation with the frequency of H-1 (causing a 5% NOE enhancement of the H-6a signal) and with that of H-2 (resulting in a small NOE effect in the signal of H-1), indicated that the anhydro ring and the OH-1 substituent are in trans relation in 9. On the other hand, the <sup>1</sup>H NMR data of 10 ( $J_{1,2}$  =  $J_{1,6a} = J_{1,6e} = 3$ ,  $J_{5,6a}$  11,  $J_{5,6e}$  1.5, and  $J_{3,4} = J_{4,5} = 5.2$  Hz) suggested a *cis* steric arrangement of the epoxide and OH-1 functions.

Opening of the oxirane ring of compounds 4, 9, and 10 with nucleophiles permits the synthesis of functionalized (azido and amino) cyclitol derivatives that are suitable for phosphorylation to obtain various analogues of Ins  $P_3$  with potential 'second messenger' properties. For such purposes **4** was treated with allyl alcohol in the presence of boron trifluoride etherate to afford, in 80% yield, (1S,2S,3S,4R,6R)-6-allyloxy-4-azido-3-benzyloxycyclohexane-1,2-diol (**11**). A *diaxial*-opening of the oxirane ring was concluded from the values ( $J_{1,2}$  3.0,  $J_{2,3} = J_{3,4} = 9.0$ ,  $J_{4,5a}$  11.5,  $J_{4,5e}$  4,  $J_{5a,6}$  3, and  $J_{5e,6}$  4 Hz) of the <sup>1</sup>H NMR coupling constants, demonstrating that compound **11** exists in a  ${}^{3}C_{6}$  chair conformation. This high-yielding oxirane ring cleavage with allyl alcohol is of practical importance for distinguishing between the C-3 and C-6 hydroxyl functions upon deprotection (and subsequent phosphorylation) of a 1,2-acetal derivative of the azidocyclitol **11**.

Synthetic analogues of the natural branched-chain cyclitols and anhydrocyclitols may be of interest for biological investigation including structure-activity relationship studies. For the synthesis of related compounds, unsaturated cyclitols 3, 7, and 8 were subjected to a thermal Claisen rearrangement [11] with N.N-dimethylacetamide dimethyl acetal. The functionalized cyclohexenyl acetamides 12, 13 and 14, respectively, were isolated in over 70% yield. The NMR data clearly showed that the amide-acetal rearrangement expectedly proceeded with a complete chirality transfer from carbon C-1 of the enol to the  $\beta$ -position. As an example, the values of the  $J_{16}$ coupling constants ( $J_{1.6}$  10 and  $J_{1.6'}$  2.5 Hz) indicate the (1R, 4R, 5R) absolute configuration of 12, and thus the cis arrangement of the C-1 and C-5 substituents.

Further chemical derivatization and biological investigation of the prepared new anhydro and branched-chain cyclitol derivatives are in progress

#### 1. Experimental

General.—Melting points were determined on a Kofler hot-stage apparatus and are uncorrected. Optical rotations were measured with a Perkin–Elmer 283 B instrument. <sup>1</sup>H (200 MHz) and <sup>13</sup>C NMR spectra (50.3 MHz) were recorded with a Bruker 200 SY spectrometer (internal TMS). See Tables 1 and 2. Mass spectra were recorded with AEI-MS 902 and VG-7035 instruments. TLC and column chromatography were performed on Kieselgel 60  $F_{254}$  (E. Merck) and Silica Gel 60 (E. Merck), using (A) 6:4 hexane–ethyl acetate, (B) 96:4 toluene–methanol, (C) 9:1

Compound	Solvent	Chemic	al shifts (8) i	in ppm							
		C-1	C-2	C-3	C-4	C-5	C-6	PhCH <sub>2</sub> -	$C(CH_3)_2$	$C(CH_3)_2$	Others
3	Acetone- $d_6$	85.52	131.56 <sup>h</sup>	124.61 <sup>b</sup>	31.99	61.53	73.55	75.02			
4	Acetone- $d_{\kappa}$	82.20	72.81	60.80	30.19	51.73	58.32	75.44	1	I	1
7	Acetone- $d_6$	65.16	137.73 <sup>b</sup>	123.92 <sup>b</sup>	73.02	71.17	35.77	Ι	28.30	109.89	I
	, ,								25.96		
8	Acetone- $d_6$	62.51	135.30 <sup>b</sup>	126.73 <sup>b</sup>	73.36	71.90	35.13	Ι	27.76	108.64	1
	2								26.24		
6	Acetone- $d_6$	64.60	54.56	52.49	72.51	71.22	25.92	Ι	28.19	109.36	1
	•								25.39		
10	Acetone- $d_6$	63.21	56.55	55.05	74.19	70.54	26.94	1	27.53	109.41	1
	2								25.24		
11	CDCI,	74.76	81.99	71.79	60.12	29.73	70.27	75.04	ł	I	-OCH, CH=CH, ; 117.09
	ì										-OCH, CH = CH, 134.31
											$-OCH_{2}CH = CH_{2}; 138.04$
12	CDCI	44.91	132.73 <sup>b</sup>	122.04 <sup>b</sup>	78.21	57.83	38.70	72.13	ł	I	-CH <sub>2</sub> CO 35.85
	)										$-N(CH_3)_2$ 35.39; 37.20
13	Acetone- $d_6$	35.08	132.22	126.13	34.96	74.90	77.23	ł	26.60	107.90	- <i>C</i> H <sub>2</sub> CÕ 28.94
	;								24.81		$-N(CH_3)$ , 35.36; 37.32
14	Acetone- $d_6$	37.89	132.39 <sup>b</sup>	125.65 <sup>b</sup>	36.53	73.93	79.15	Ι	28.48	108.69	$-CH_2CO$ 29.99
									25.88		-N(CH <sub>3</sub> ) <sub>2</sub> 35.64; 37.88
	-										

Table 1  $^{13}$ C NMR data for compounds 3, 4 and 7–14  $^{a}$ 

<sup>a</sup> Obtained at 50.3 MHz. <sup>b</sup> Signals are interchangeable.

Table 2 <sup>1</sup> H NMR data	tor compound	ls 2-4 and 7-	- <b>14</b> <sup>a</sup>								
Compound	Solvent	Chemical shi	ifts (8) i	u ppm							
		H-1	H-2	H-3	H-4	H-5	H-6	$PhCH_2^-$	$C(CH_3)_2$	HO	Others
2	Acetone-d <sub>6</sub>	I	6.02	7.00	2.44 (a)	4.50	4.49	4.72 5.00	١	I	1
3	Acetone-d <sub>6</sub>	4.33	5.59	5.59	2.88 (e) 2.02 (a) 2.42 (e)	3.73	3.45	5.08 4.95	l	4.50	1
4	Acetone- $d_6$	4.05	3.38	3.57	1.68 (a) 2.43 (e)	3.18	3.25	4.85	ł	4.57	1
7	Acetone- $d_6$	4.07	5.71	5.95	4.41	4.28	1.62 (a)	I	1.28 1 37	3.80	I
œ	Acetone- $d_6$	4.27	5.54	5.84	4.42	4.43	2.00 (c) 1.62 (a)	ŀ	1.26	3.46	1
6	Acetone- $d_6$	3.92	3.26	2.89	4.22	4.21	1.67 (a)	I	1.19	3.24	I
10	Acetone- $d_6$	4.27	3.00	3.23	4.17	4.19	1.55 (a) 1.55 (a) 1.91 (e)	I	1.22	4.10	I
11	CDCI <sub>3</sub>	3.94-4.05	3.81	3.52	3.94-4.05	1.81 (a) 2.06 (e)	3.94-4.05	4.80	I	2.82 2.82	$-OCH_2CH=CH_2 3.67$ $-OCH_2CH=CH_2 5.86$ $OCH_2CH=CH_2 5.17.5 35$
12	CDCI 3	3.08	5.74	5.81	4.81	4.53	2.03 (a) 2.19 (e)	4.20 4.38	I	I	-OCH3/CH - CH2 - CH2 - CH3/CH - CH3/CH - CH, CO 2.34
13	Acetone-d <sub>6</sub>	2.56	5.54	5.72	1.98 (a) 2.25 (e)	4.48	4.48	I	1.22 1.25	I	-N(CH <sub>3</sub> ) <sub>2</sub> 2.89; 3.04 -CH,CO 2.48
14	Acetone-d <sub>6</sub>	2.49–2.60	5.68	5.68	2.10 (a) 2.38 (e)	4.30	3.89	ŧ	1.28 1.38	1	-N(ČH <sub>3</sub> ) <sub>2</sub> 2.83; 3.04 -CH <sub>2</sub> CO 2.65

<sup>a</sup> Obtained at 200 MHz.

toluene–MeOH, (D) 7:3 hexane–ethyl acetate, (E) 98:2  $CH_2Cl_2$ –MeOH, (F) 1:1 hexane–ethyl acetate, and (G) 95:5 ether–MeOH as the developing system/eluent. Evaporations were carried out under diminished pressure at 35–40 °C.

(5R,6S)-5-Azido-6-benzyloxycyclohex-2-enone (2). —To a stirred, cold (0 °C) solution of 1 (1.02 g, 3.9 mmol) in dry pyridine (20 mL) was added dropwise methanesulfonyl chloride (0.37 mL, 4.7 mmol), and then the reaction mixture was allowed to warm up to room temperature. After 4 h it was diluted with CH<sub>2</sub>Cl<sub>2</sub> (40 mL) and washed with satd aq NaHCO<sub>3</sub> (2 × 20 mL) and water (2 × 20 mL), dried (Na<sub>2</sub>SO<sub>4</sub>) and concentrated. Co-evaporation with dry toluene (2 × 15 mL) resulted in unstable syrupy **2**, which was used without further purification,  $R_f$  0.5 (A): IR  $\nu_{max}$ (KBr) 2106 (N<sub>3</sub>), 1694 (C=O) cm<sup>-1</sup>.

(1S,5R,6S)-5-Azido-6-benzyloxycyclohex-2-en-1-ol (3).—To a solution of crude 2 (0.12 g, 0.46 mmol) in dry methanol (3 mL) was added  $CeCl_3 \cdot 6H_2O$  (0.16 g, 0.46 mmol), the mixture was cooled to 0 °C and then NaBH<sub>4</sub> (17.4 mg, 0.46 mmol) was added in three portions with vigorous stirring. After 3 h the reaction mixture was evaporated, the residue was diluted with  $CH_2Cl_2$  (3 × 15 mL), and the combined organic layers was concentrated. The residue was dissolved in water (20 mL) and extracted with CH<sub>2</sub>Cl<sub>2</sub>  $(2 \times 10 \text{ mL})$ , the organic layer was dried (MgSO<sub>4</sub>) and concentrated, and the residue was subjected to column chromatography  $(B \rightarrow C)$  to obtain 80 mg (70%) of **3** as a pale yellow syrup:  $[\alpha]_{\rm D}$  +106.2° (c 1.16 CHCl<sub>3</sub>), IR:  $\nu_{max}$  (KBr) 2106 (N<sub>3</sub>), 3418 (OH) cm<sup>-1</sup>. MS: m/z 245 [M]<sup>+</sup>, 244 [M – H]<sup>+</sup>, 216 [M –  $H - N_2$ ]<sup>+</sup>. Anal. Calcd for C<sub>13</sub>H<sub>15</sub>N<sub>3</sub>O<sub>2</sub> (245.28): C, 63.66; H, 6.16; N, 17.13. Found: C, 63.92; H: 6.29; N, 17.25.

(*IR*,2S,3R,5S,6R)-3-Azido-2-benzyloxy-5,6-epoxycyclohexan-1-ol (4).—A mixture of 3 (0.45 g, 1.84 mmol), 85% 3-chloroperoxybenzoic acid (0.56 g, 2.8 mmol) and  $CH_2Cl_2$  (8.0 mL) was stirred at room temperature for 22 h, when TLC (D) showed that the conversion of 3 to 4 was complete. The reaction mixture was diluted with  $CH_2Cl_2$  (10 mL) and then washed with 5% aq Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> (2×5 mL), 1% aq NaOH (2  $\times$  5 mL), and water (2  $\times$  7 mL), dried  $(MgSO_4)$ , concentrated, and purified by means of column chromatography  $(10:0 \rightarrow 3:7 \text{ hexane-ethyl})$ acetate) to obtain pure 4 (297 mg, 62%): mp 119-120 °C;  $[\alpha]_D + 76.9^\circ$  (c 0.9 in CHCl<sub>3</sub>),  $R_f$  0.17 (D). Anal. Calcd for  $C_{13}H_{15}N_3O_3$  (261.28): C, 59,76; H, 5.78; N, 16.08. Found: C, 59.55; H, 6.00; N, 15.98. The (1S)-diastereoisomer of 4 was also isolated as a

byproduct: 39.8 mg (8%):  $[\alpha]_D + 50.9^\circ$  (c 0.91 in CHCl<sub>3</sub>),  $R_f$  0.31 (D).

(1S, 4S, 5R)- and (1R, 4S, 5R)-4,5-O-isopropylidenedioxycyclohex-2-en-1-ol (7 and 8).-Reduction of 6 [9,10] (0.2 g, 1.19 mmol) with sodium borohydride (45 mg, 1.19 mmol) in methanol (5 mL) in the presence of CeCl<sub>3</sub>  $\cdot$  6H<sub>2</sub>O (0.422 g, 1.19 mmol) was carried out as described above for the preparation of 3. The reduction was complete in ca. 30 min, and workup that included chromatographic purification  $(100:0 \rightarrow 98:2 \text{ CH}_2\text{Cl}_2\text{-MeOH})$  gave as the first product to elute (1*S*)-7 (118 mg, 58%):  $[\alpha]_{D} - 68.7^{\circ}$ (c 1.16 in CHCl<sub>3</sub>),  $R_f$  0.42 (E). Anal. Calcd for C<sub>9</sub>H<sub>14</sub>O<sub>3</sub> (170.2): C, 63.51; H, 8.29. Found: C, 63.63; H, 8.22. Eluted second was (1*R*)-8 (59 mg, 29%):  $[\alpha]_{\rm D} + 40.2^{\circ} (c \ 2.56 \text{ in CHCl}_3), R_f \ 0.21 \text{ (E). Anal.}$ Calcd for C<sub>9</sub>H<sub>14</sub>O<sub>3</sub> (170.2): C, 63.51; H, 8.29. Found: C, 63.77; H, 8.34.

(1S,2R,3R,4S,5R)-2,3-Epoxy-4,5-isopropylidenedioxycyclohexane-1-ol (9).—Epoxidation of 7 (0.2 g, 1.16 mmol) with 85% 3-chloroperoxybenzoic acid (0.47 g, 2.3 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (6.0 ml) for 48 h was carried out as described for the preparation of 4, to give syrupy 9 (187 mg, 87%):  $[\alpha]_D - 58.1^\circ$  (*c* 1.15 in CHCl<sub>3</sub>),  $R_f$  0.5 (*A*). Anal. Calcd for C<sub>9</sub>H<sub>14</sub>O<sub>4</sub> (186.20): C, 58.05; H, 7.57. Found: C, 58.21; H, 7.62.

(1R, 2R, 3R, 4S, 5R)-2, 3-Epoxy-4, 5-isopropylidenedioxycyclohexane-1-ol (10). —Conversion of 8 (0.2 g, 1.16 mmol) into the epoxy compound 10 was performed as described for 4 and 9 to give 153 mg (72%) of crystalline 10: mp 98–99 °C,  $[\alpha]_D + 9.5^\circ$  (c 1.09 in CHCl<sub>3</sub>),  $R_f$  0.29 (E). Anal. Calcd for C<sub>9</sub>H<sub>14</sub>O<sub>4</sub> (186.20): C, 58.05; H, 7.57. Found: C, 57.86; H, 7.73.

(1S, 2S, 3S, 4R, 6R) - 6 - Allyloxy - 4 - azido - 3 benzyloxycyclohexane - 1, 2 - diol (11).—To a stirred solution of 4 (76 mg, 0.29 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (10 mL), allyl alcohol (128  $\mu$ L, 1.88 mmol) and BF<sub>3</sub> · Et<sub>2</sub>O (75  $\mu$ L, 0.61 mmol) were added. After stirring at room temperature for 3.5 h, the mixture was neutralized with Et<sub>3</sub>N, diluted with CH<sub>2</sub>Cl<sub>2</sub> (10 mL), washed with water (3 × 5 mL), dried (Na<sub>2</sub>SO<sub>4</sub>) and concentrated. Column chromatography of the residue (with eluent F) gave syrupy **11** (74 mg, 80%): [ $\alpha$ ]<sub>D</sub> + 27.9° (*c* 0.8 in CHCl<sub>3</sub>), *R<sub>f</sub>* 0.43 (*A*). Anal. Calcd for C<sub>16</sub>H<sub>21</sub>N<sub>3</sub>O<sub>4</sub> (319.02): C, 60.18; H, 6.58; N, 13.17. Found: C, 60.33; H, 6.71; N, 13.35.

N,N-Dimethyl [(IR,4R,5R)-5-azido-4-benzyloxycyclohex - 2 - enyl]acetamide (12).—A mixture of 3 (0.505 g, 2.06 mmol) and N,N-dimethylacetamide dimethyl acetal (8.7 mL, 59.9 mmol) was slowly heated from 25 °C to 140 °C (bath temperature) in the distillation flask of a micro-distillation apparatus fitted with a short Vigreux column. The distillate collected at 80–110 °C was recycled into the distillation flask, and the mixture was kept at 140 °C for 30 min. It was then cooled and co-evaporated with toluene  $(3 \times 10 \text{ mL})$ , and the residue was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (20 mL). The solution was washed with water (2 × 5 mL), dried (MgSO<sub>4</sub>), and concentrated. Flash column chromatography of the residue (*G*) resulted in syrupy **12**: 479 mg (74%): [ $\alpha$ ]<sub>D</sub> – 117° (*c* 1.0 in CHCl<sub>3</sub>), *R*<sub>f</sub> 0.28 (*G*). Anal. Calcd for C<sub>17</sub>H<sub>22</sub>N<sub>4</sub>O<sub>2</sub> (314.39): C, 64.89; H, 7.00; N, 17.81. Found: C, 65.11; H, 7.13; N, 18.01.

N,N-Dimethyl [(1S,5R,6S)-5,6-O-isopropylidenedioxycyclohex-2-enyl]acetamide (13).—The Claisen rearrangement of 7 (0.406 g, 2.38 mmol) with N,Ndimethylacetamide dimethyl acetal (6.0 mL, 40.5 mmol) was carried out as described above for the preparation of 12 to furnish pure 13 (429 mg, 75.3%): mp. 65–68 °C, [ $\alpha$ ]<sub>D</sub> – 16° (c 0.9 in CHCl<sub>3</sub>),  $R_f$  0.5 (G). IR:  $\nu_{max}$  (KBr) 1646 (amide) cm<sup>-1</sup>. Anal. Calcd for C<sub>13</sub>H<sub>21</sub>NO<sub>3</sub> (239.31): C, 65.25; H, 8.85; N, 5.85. Found: C, 64.97; H, 9.05; N, 5.71.

N,N-Dimethyl [(1R,5R,6S)-5,6-O-isopropylidenedioxycyclohex-2-enyl]acetamide (14).—The Claisen rearrangement of **8**, performed in an analogous manner as described for the conversion of **7**, gave 447 mg (78%) of syrupy 14:  $[\alpha]_D - 85^\circ$  (c 1.16 in CHCl<sub>3</sub>),  $R_f$  0.46 (G). Anal. Calcd for C<sub>13</sub>H<sub>21</sub>NO<sub>3</sub> (239.31): C, 65.25; H, 8.85; N, 5.85. Found: C, 65.51; H, 8.92; N, 6.08.

#### Acknowledgements

Financial support for this work was provided by the Grants OTKA 19327 and 23778 given by the National Science Foundation (Hungary). One of us (Z.G.T.) is indebted to the Hungarian Credit Bank for a grant ('A Magyar Tudományért').

#### References

- [1] M.J. Berridge, Annu. Rev. Biochem., 56 (1987) 159-169.
- [2] S. Atsumi, K. Umezawa, H. Iinuma, H. Naganawa, H. Nakamura, Y. Iitaka, and T. Takeuchi, *J. Antibiot.*, 43 (1990) 49-53.
- [3] G. Legler and R. Bollhagen, Carbohydr. Res., 233 (1992) 113-123.
- [4] G. Legler, Adv. Carbohydr. Chem. Biochem., 48 (1990) 319-384.
- [5] D.C. Billington, The Inositol Phosphates: Chemical Synthesis and Biological Significance, VCH Verlagsgesellschaft mbH, Weinheim, 1993.
- [6] P. László, I.F. Pelyvás, F. Sztaricskai, L. Szilágyi, and Á. Somogyi, *Carbohydr. Res.*, 175 (1988) 227– 239.
- [7] I.F. Pelyvás, M. Mádi-Puskás, Z. G. Tóth, Zs. Varga, M. Hornyák, Gy. Batta, and F. Sztaricskai, J. Antibiot., 48 (1995) 683-695.
- [8] N. Chida, M. Ohtsuka, K. Nakazawa, and S. Ogawa, J. Org. Chem., 56 (1991) 2976-2983.
- [9] B.M. Trost and A.G. Romero, J. Org. Chem., 51 (1986) 2332-2342.
- [10] J.E. Audia, L. Boisvert, A.D. Patten A. Villalobos, and S.J. Danishefsky, J. Org. Chem., 54 (1989) 3738-3740.
- [11] I.F. Pelyvás, T. Lindhorst, and J. Thiem, *Liebigs Ann. Chem.*, (1990) 761–769.