

A Trifluoroacetic Acid Catalyzed Domino Reaction as an Approach to Amino Acid Derived 2,3-Dihydro-1*H*-1,5-benzodiazepines

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This paper is dedicated to Prof. M. Periasamy on his 60th birthday.

Abstract: Trifluoroacetic acid catalyzed condensation of aromatic amines with substituted benzaldehydes followed by intramolecular cyclization furnishes a highly effective synthesis of amino acid derived 2,3-dihydro-1*H*-1,5-benzodiazepines. The strategy provides an efficient access to a library of benzodiazepines that can be explored for potential pharmaceutical or biological activities.

Key words: amino acids, heterocycles, benzodiazepine, domino reaction, intramolecular cyclization

Benzannulated nitrogen-containing heterocycles are core structures of numerous biologically active compounds that display a wide range of pharmacological activities.¹ Among them, benzodiazepines (Figure 1) are widely used in medicinal chemistry.² They belong to the class of psychotropics³ that have notable central nervous system depressant activity,⁴ along with various other biologically important activities⁵ such as anticancer,⁶ antiviral (HIV),⁷ and cardiovascular activity.⁸ Many members of this family are known as anti-anxiety, analgesic, antidepressive, sedative, and hypnotic agents⁹ and other applications include dyes for acrylic fibers¹⁰ and as anti-inflammatory agents.¹¹ In addition, 1,5-benzodiazepines are key intermediates for fused benzodiazepines, such as triazol¹² and oxadiazol.¹³

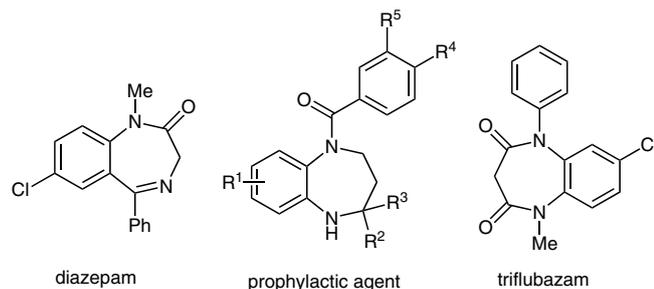


Figure 1 Examples of medicinally important benzodiazepine derivatives

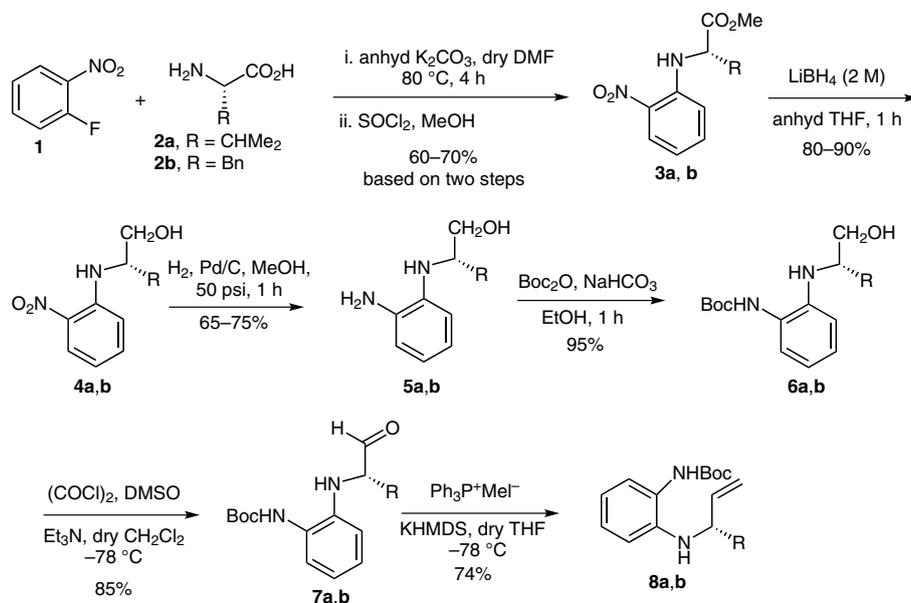
One of the most practical strategies for the synthesis of such heterocyclic compounds could be a cascade reaction in which multiple-bond formation and/or bond cleavage oc-

cur.¹⁴ The salient features of cascade or domino reactions are simplicity, efficiency, and shortened timing for diversity-oriented synthesis in one pot.¹⁵ Several approaches to benzodiazepines have been reported.¹⁶ The most general and simplest method for accessing 1,5-benzodiazepines involves the acid-catalyzed reaction of *o*-phenylenediamine with α,β -unsaturated carbonyl compounds, β -haloketones or ketones. Many reagents including $\text{BF}_3 \cdot \text{OEt}_2$,¹⁷ polyphosphoric acid– SiO_2 ,¹⁸ NaBH_4 ,¹⁹ MgO/POCl_3 ,²⁰ $\text{Yb}(\text{OTf})_3$,²¹ Al_2O_3 – P_2O_5 ,²² and acetic acid under microwave conditions²³ have been reported for this reaction. Recently, these condensations have been reported even in monosaccharide and ionic liquid media.^{24,25}

The synthesis and biology of (*S*)-amino acid based chiral heterocycles and natural product like molecules²⁶ have been reported by our group. Recently, we synthesized a novel series of amino acid derived benzoxazepines as potential antitumor agents^{26e} and diversity-oriented benzannulated heterocyclic scaffolds employing inter- and intramolecular Mitsunobu reactions.^{26g} Herein we disclose an innovative route towards the synthesis of chiral 2,3-dihydro-1*H*-1,5-benzodiazepines through an acid-catalyzed domino reaction.

(*S*)-Amino acids **2a,b** were reacted with 1-fluoro-2-nitrobenzene (**1**) in the presence of K_2CO_3 and dry DMF at 80 °C to furnish the corresponding 2-nitrophenyl *N*-protected amino acid derivatives which were converted into methyl esters **3a,b** in the presence of SOCl_2 and MeOH (Scheme 1). Nucleophilic aromatic substitution of 2-nitrofluorobenzene with amino acids occurs without any racemization^{26h} (ee > 99%). LiBH_4 reduction of esters **3a,b** gave carbinols **4a,b** in 80–90% yield. The nitro group was reduced to an amine by hydrogenation to provide carbinols **5a,b** in 65–75% yield. Selective protection of the primary amine using Boc anhydride and sodium bicarbonate in ethanol afforded **6a,b** both in 95% yield. Swern oxidation at –78 °C gave aldehydes **7a,b** which underwent one-carbon Wittig olefination to afford **8a,b**.

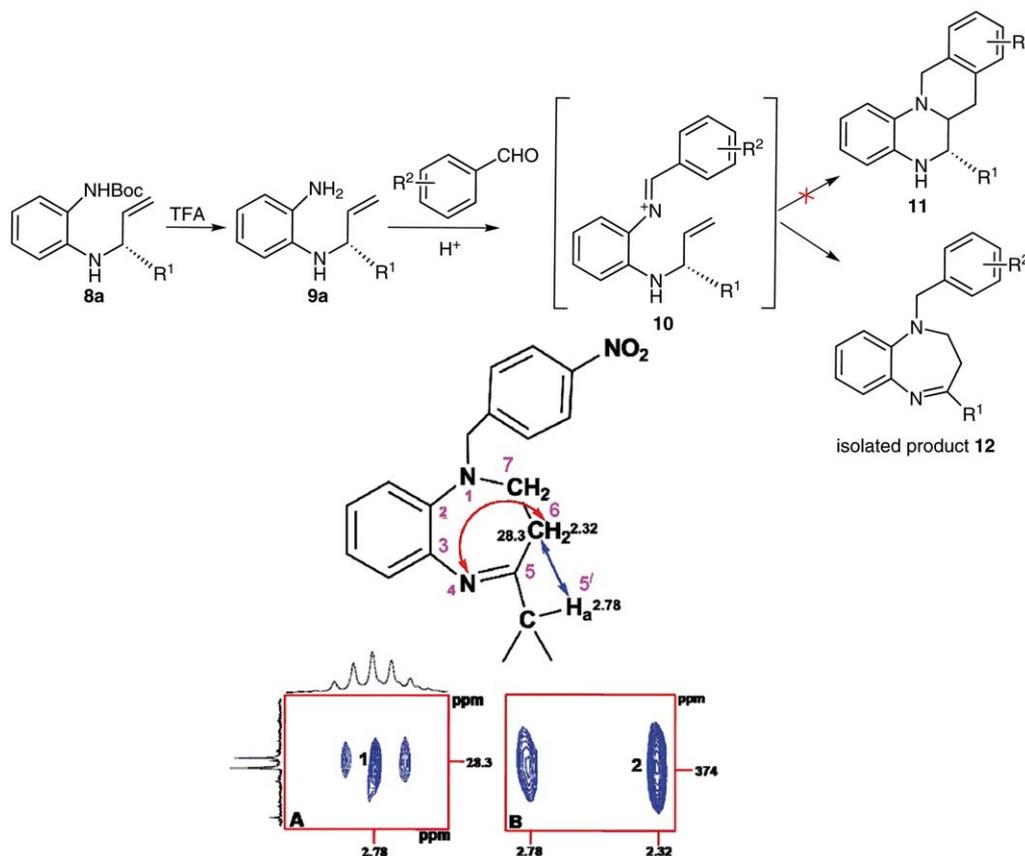
With intermediate **8a** in hand, we targeted enantiomerically pure tetrahydro-5*H*-isoquinolino[2,3-*a*]quinoxaline **11** in one pot via an inverse electron-demand Diels–Alder reaction (Scheme 2). However, when **8a** was treated with trifluoroacetic acid (TFA) followed by addition of substituted benzaldehydes at room temperature, dihydrobenzodiazepines **12** were unexpectedly isolated instead of **11**.



Scheme 1 Synthesis of substrate **8**

This was confirmed by ^1H - ^1H homonuclear, ^{13}C - ^1H and ^{15}N - ^1H heteronuclear long-range correlations for the representative compound **12g**. Long-range correlation between $\text{H}_{5'}$ ($\delta = 2.78$ ppm) and C_6 ($\delta = 28.3$ ppm), supports

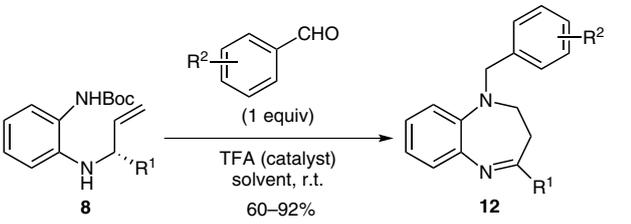
the seven-membered ring formation. This is further confirmed by the long-range correlation between N_4 and C_6H . During optimization of this TFA-catalyzed domino reaction, varying solvents and stoichiometries of reagents, noticeable results were examined (Table 1). The maximum



Scheme 2 Strategy for preparation of benzodiazepines. (A) ^{13}C - ^1H HMBC correlation of **12g** is shown between $\text{H}_{5'}$ \leftrightarrow C_6 with blue arrow and labelled as **1**. (B) ^{15}N - ^1H HMBC correlation of **12g** is shown between N_4 \leftrightarrow C_6H with red arrow and labelled as **2**.

yield (92%) was obtained with two equivalents TFA at room temperature for 30 minutes. With one equivalent TFA and **8** yields were lower.

Table 1 Screening of the Substrates and Optimization of the Reaction



Entry	Substrate	TFA (equiv)	Solvent	Time (min)	Yield (%) ^a
1	8a	0.0	CH ₂ Cl ₂	50	0
2	8a	0.1	CH ₂ Cl ₂	45	51
3	8a	1.0	CH ₂ Cl ₂	48	65
4	8a	1.5	CH ₂ Cl ₂	48	72
8	8a	2.0	CH ₂ Cl ₂	30	92
6	8b	0.1	CH ₂ Cl ₂	38	62
7	8b	1.0	CH ₂ Cl ₂	44	53
8	8b	1.5	CH ₂ Cl ₂	50	69
9	8b	2.0	CH ₂ Cl ₂	35	85
10	8a	0.1	MeCN	50	0
11	8a	1.0	MeCN	45	15
12	8b	2.0	MeCN–H ₂ O (1:1)	55	42

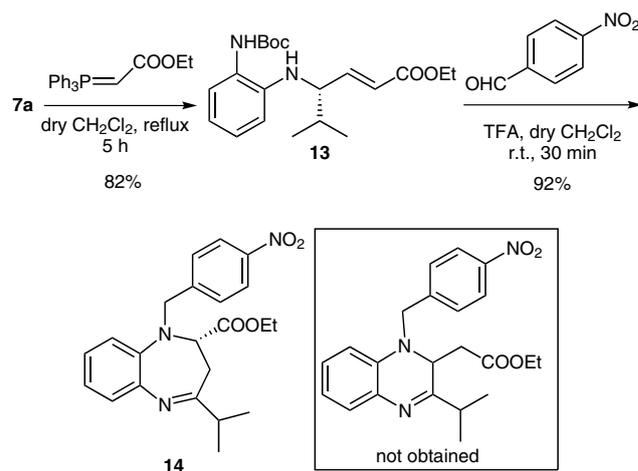
^a Isolated yield after column chromatography.

To explore this synthetic strategy, substituted benzaldehydes with a variety of electron-releasing groups; that is, Me, OMe, and electron-withdrawing groups, such as NO₂ and CN, were reacted with substrates **8** under similar conditions (Table 2). The three-step reaction proceeded very efficiently under mild conditions at room temperature to produce the corresponding 2,3-dihydro-1*H*-1,5-benzodiazepines **12a–n** in high yields in each case.

To examine the mechanism of the reaction, further investigations were undertaken under the optimized reaction conditions. With intermediate aldehyde **7a** in hand, Horner–Wadsworth–Emmons olefination was performed in the presence of Ph₃P=CHCO₂Et in dry CH₂Cl₂ to give **13** (Scheme 3). Upon treatment of **13** with 4-nitrobenzaldehyde and TFA in dry CH₂Cl₂, (*S*)-ethyl-4-isopropyl-1-(4-nitrobenzyl)-2,3-dihydro-1*H*-1,5-benzodiazepine-2-carboxylate (**14**) was isolated. NMR analysis of compound **14** also displayed similar, long-range correlations as detected in **12g**, supporting the same mechanistic route like compound **12**.

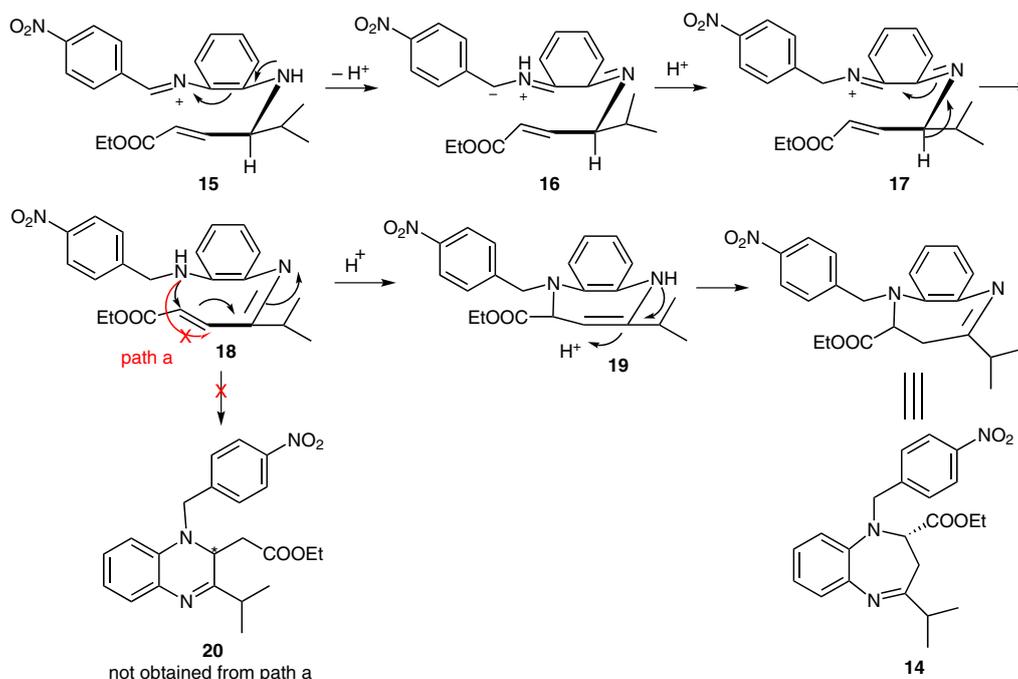
Table 2 Synthesis of 2,3-Dihydro-1*H*-1,5-benzodiazepine

Entry	R	ArCHO	Product	Time (min)	Yield (%)
1	CHMe ₂	4-FC ₆ H ₄	12a	35	89
2	CHMe ₂	4-NCC ₆ H ₄	12b	30	92
3	CHMe ₂	2-FC ₆ H ₄	12c	32	90
4	CHMe ₂	3-MeC ₆ H ₄	12d	38	65
5	CHMe ₂	2-MeC ₆ H ₄	12e	40	68
6	CHMe ₂	3-O ₂ NC ₆ H ₄	12f	32	91
7	CHMe ₂	4-O ₂ NC ₆ H ₄	12g	30	92
8	CHMe ₂	3-FC ₆ H ₄	12h	35	88
9	CHMe ₂	4-ClC ₆ H ₄	12i	45	82
10	CHMe ₂	3-ClC ₆ H ₄	12j	40	79
11	CHMe ₂	3-thiophene	12k	45	78
12	Bn	4-NCC ₆ H ₄	12l	42	80
13	CHMe ₂	4-MeC ₆ H ₄	12m	34	61
14	CHMe ₂	4-MeOC ₆ H ₄	12n	38	60



Scheme 3 Synthesis of compound **14**

The probable mechanism for the TFA-catalyzed transformation of compounds **8a,b** into benzodiazepines **12a–n**²⁷ is shown in Scheme 4. Initially, the iminium ion **15**, formed from condensation between the benzaldehyde and amine, undergoes sigmatropic 1,5-H shift (NH → CH)²⁸ to furnish the intermediate *o*-quinonediimine (type **16** without anion). Subsequently, intermediate **16** through another 1,5-H shift (CH → NH) generates the intermediate **18**. Of the two reaction pathways possible from **18**, intramolecular 7-*endo-trig* aza-Michael addition onto the most activated α,β-unsaturated iminium ion followed by 1,7-electrocyclization and final rearomatization affords benzodiazepine **14**. Intermediate **18** does not furnish **20**.



Scheme 4 Possible mechanism for the formation of product **14**

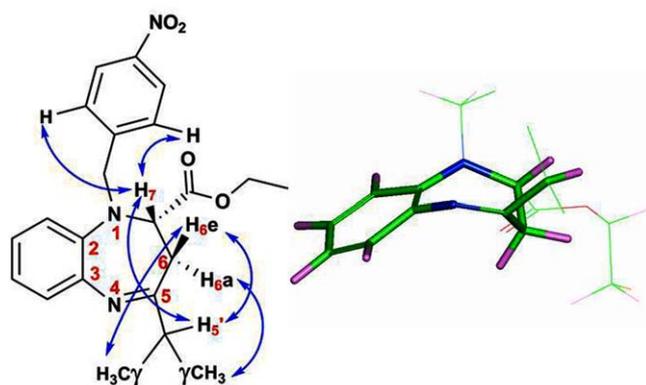


Figure 2 Confirmation of stereochemistry of **14** by NOESY (400 MHz, CDCl_3). Characteristic NOE values and average structure from 1 ns MD run. (A) NOE enhancements are shown in blue arrows and support the proposed twist-chair conformation of the seven-membered ring. (B) Average of the 15 lowest-energy structures resulted from the restrained molecular dynamics (MD) run. For clarity, the nitrophenyl group is replaced with a methyl group after the MD run.

For compound **14**, a vicinal coupling constant of 11.8 Hz between H_{6a} and H_7 suggests their diaxial relative disposition. Additionally, NOE enhancements between $H_7 \rightarrow \text{Ph}$ (*ortho*), $H_{6a} \rightarrow \text{Ph}$ (*ortho*), $\text{CH}_3(\gamma) \rightarrow H_{6e}$ and $\text{CH}_3(\gamma) \rightarrow H_{6a}$ (weak) further support the proposed conformation for the seven-membered ring. Restrained molecular dynamics (MD) calculations were performed with a Discovery Studio 3.0 client program using CHARMM force field²⁹ with default parameters throughout the simulation. The average structure, given in Figure 2, suggests that the cycloheptadiene ring adopts the proposed twisted-chair conformation, which is flattened at the Val residue due to the double bond.

In conclusion, we have established a new strategy for the synthesis of 1,5-benzodiazepines from amino acid derived substrates and substituted benzaldehydes through TFA-catalyzed domino transformation. This synthesis proceeds via Boc deprotection, imine formation, and intramolecular cyclization to form the seven-membered heterocycle. This simple, convenient, easily reproducible, and high-yielding procedure for accessing functionalized 1,5-benzodiazepine scaffolds may be useful in both synthetic and medicinal chemistry.

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Supporting Information for this article is available online at <http://www.thieme-connect.com/ejournals/toc/synlett>.

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