Intermolecular Gold-Catalyzed Diastereo- and Enantioselective [2+2+3] Cycloadditions of 1,6-Enynes with Nitrones**

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Metal-mediated cycloaddition reactions are a powerful tool for accessing carbo- and heterocyclic frameworks.^[1] Nitrones serve as three-atom building units in various [3+2] and [3+3] cycloadditions with suitable dipolarophiles to access five- and six-membered nitrogen heterocycles.^[2,3] The utility of such reactions is manifested not only in the easy access to natural products,^[2a] but also in the development of many enantioselective cycloadditions. We are aware of few reports of nitrone cycloadditions for the synthesis of seven-membered heterocvcles.^[4] The reported examples are exclusively focused on [4+3] nitrone cycloadditions,^[4-6] as shown by the work of Hayashi and co-workers (Scheme 1).^[4c] We report herein the gold-catalyzed cyclization/[2+2+3] cycloaddition cascade^[5,6] between nitrones and the readily available 1,6-enynes 1 to give the highly substituted 1,2-oxazepane derivatives 3 and 5 with satisfactory diastereo- and enantioselectivity. The importance of this reaction is reflected by the occurrence of 1,2oxazepane moieties in several bioactive molecules.^[7]

Previous work (Hayashi et al.)[4c]



Scheme 1. Cycloaddition reactions for nitrones.

For 1,6-enynes, the interception of the 1,4-dipole equivalent **A** is challenging because of its intrinsic reactivity toward cycloisomerization.^[8,9] Echavarren and co-workers attempted to intercept **A** with carbonyl compounds in an intermolecular

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fashion,^[10a,c,11] but those reactions were inevitably accompanied by a significant amount of side products.^[10a] Helmchen^[10b] studied the same cycloaddition using unsubstituted 1,6-enynes, but carbonyl compounds other than 2-nitrobenzaldehyde were used in large excess (20 equiv). The use of 1,6enynes as efficient 1,4-dipole equivalents is only feasible in intramolecular system.^[10c]

We sought to realize the reaction using the 1,6-enyne **1a** (Table 1). The primary task was to reduce the production of 3-

 Table 1:
 Nitrone cycloaddition with various catalysts.

 Ph

la la	+ Ph N Ph O 2a (1.5 equiv)	5 mol% catalyst 25 °C, 1 h	Ph N-Ph H	4
Entry	Catalyst ^[a]	Solvent	3 a^[b] Yield [%]	4 ^[b] Yield [%]
1	AgSbF ₆	DCE	_	46
2	AgNTf ₂	DCE	-	43
3	[PPh ₃ AuCl]/AgSbF ₆	DCE	39	15
4	[PPh ₃ AuCl]/AgNTf ₂	DCE	35	20
5	[LAuCl]/AgSbF ₆	DCE	95	_
6	[LAuCl]/AgNTf ₂	DCE	86	-
7	[IPrAuCl]/AgSbF ₆	DCE	75	15
8	[IPrAuCl]/AgNTf ₂	DCE	72	18
9	[LAuCl]/ÅgSbF ₆	DCM	87	-
10	[LAuCl]/AgSbF ₆	1,4-dioxane	82	-
11	[LAuCl]/AgSbF ₆	toluene	71	-

[a] $L = P(tBu)_2(o-biphenyl)$, [substrate] = 0.19 M. [b] Product yields are reported for compounds isolated after purification on silica gel. DCE = 1,2-dichloroethane, IPr = 1,3-bis(diisopropylphenyl)imidazol-2-ylidene, Tf = trifluoromethanesulfonyl.

isobutenyl-1H-indene (4), which results from a competitive cycloisomerization.^[12] In a standard operation, the nitrone 2a (ca. 0.38 M) was first stirred with the catalyst (5 mol%) in DCE (25°C) for five minutes, and then 1a in DCE (equal volume) was slowly added to this solution, using a syringe pump, over a 1 hour period.^[13] The conversion was complete at the end of the addition. As shown in entries 1 and 2, we obtained only 4 in 46 and 43% yield using $AgSbF_6$ and AgNTf₂, respectively. Cationic gold complexes, [PPh₃AuCl]/ AgX (X = SbF₆, NTf₂), gave the nitrone cycloadduct 3a as a single diastereomer in moderate yield (35-39%) together with 4 in 15-20% yield (entries 3 and 4). Pleasingly, the use of $[P(o-biphenyl)(tBu)_2AuCl]/AgX (X = SbF_6, NTf_2)$ gave the desired 3a in 95 and 86% yield (entries 5 and 6). [IPrAuCl]/ AgX ($X = SbF_6$, NTf₂) gave **3a** in 75–72 % yield in addition to a small amount of 4 (15-18%; entries 7 and 8). For

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[(*o*-biphenyl)P(tBu)₂AuSbF₆], the yields of **3a** depended on solvents: CH₂Cl₂ (87%), 1,4-dioxane (82%), and toluene (71%). Structural characterization of **3a** relied on an X-ray diffraction study^[14] (see the Supporting Information).

We examined the reactions of **2a** with various 1,6-enynes (**1b–m**) to assess the substrate scope, including benzenoid and nonbenzenoid bridges (Scheme 2). The cycloaddition reac-



Scheme 2. [2+2+3] Nitrone cycloadditions on various 1,6-enynes. [a] [substrate] = 0.19 M. [b] All product yields are reported for compounds isolated after purification on silica gel. L = P(tBu)₂(*o*-biphenyl), DCE = 1,2-dichloroethane.

tions were performed in DCE (25 °C) using [P(*o*-biphenyl)-(*t*Bu)₂AuCl]/AgSbF₆ (5 mol %). This cycloaddition is applicable to substrates bearing various alkenes, such as 3pentylidene, cyclopentylidene, and cyclohexylidene, thus giving the desired cycloadducts **3b–d** in 62–84% yield. For the benzenoid substrates **1e–g** bearing a C4 substituent (X = OMe, F, Cl), the reactions proceeded smoothly to give the desired compounds **3e–g** with yields exceeding 91%. Excellent yields (85–92%) were obtained also for products **3h–j** bearing various C5 substituents (X = OMe, Me, Cl). The scope of this cycloaddition is substantially expanded with its applicability to nonbenzenoid substrates containing a cyclopentenyl, cyclohexenyl, and cycloheptenyl bridge, thus giving the corresponding products **3k–m** in 78–90% yield.

Scheme 3 shows the applicability of the cycloaddition of the 1,6-enyne **1a** with various nitrones (**2b–k**). We obtained only the desired cycloadducts **5b–k** as single diastereomers with yields exceeding 76%. We tested the cycloadditions on nitrones **2b–e**, bearing both electron-deficient and electronrich aniline moieties ($Ar^1 = 4-XC_6H_4$; X = OMe, Me, Cl, CO_2Et), and the corresponding cycloadducts **5b–e** were obtained in excellent yields (92–97%). These cycloadditions can also be extended to the nitrones **2 f–i** bearing electrondeficient imines ($Ar^2 = 4-XC_6H_4$; X = Cl, Br, CF₃, NO₂), thus giving the corresponding cycloadducts **5 f–i** with yields of



Scheme 3. Cycloadditions of enyne **1 a** with various nitrones. [a] [substrate] = 0.19 m. [b] All product yields are reported for compounds isolated after purification on silica gel. $L = P(tBu)_2(o\text{-biphenyl})$.

greater than 91%. For the nitrones **2j** and **2k** bearing a 2- and 3-thienyl group, respectively, the corresponding cycloadducts **5j** and **5k** were obtained in 76–78% yield.

Scheme 4 shows a plausible mechanism to rationalize the stereochemistry of the resulting cycloadducts **3a** and **3l** using the 1,6-enynes **1a** and **1l**, respectively, and nitrone **2a**. The substrates **1a** and **1l** are less flexible in conformation and facilitate the initial cyclization. In the presence of a gold catalyst, we postulate that an initially formed cyclopropylgold carbenoid^[4-5] has a strong alkenylgold carbocation character (**B**'),^[15] which is attacked by nitrone **2a** in a concerted pathway, thus giving either the cycloadduct **3a** or **3l** with excellent diastereoselectivity.



Scheme 4. A proposed mechanism for the diastereoselective [2+2+3] cycloaddition.

Our results for the enantioselective cycloaddition of the 1,6-enyne **1a** with nitrone **2e** are shown in Table 2. Among the commonly used C2-bisphosphine ligands, (*R*)-DTBM-MeO-Biphep gave the best performance for enantioselectivity (see the Supporting Information). The use of $[LAu_2Cl_2]/2AgSbF_6$ (L = (*R*)-DTBM-MeO-Biphep, 5 mol%) gave the desired compound (-)-**5e** in 71% yield with 70% *ee*

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(entry 1). [LAu₂Cl₂]/2 AgOTf at a 5 mol% loading improved the enantioselectivity of compound (–)-**5e** to 75% *ee* (entry 2). The reaction in a cold DCE solution (0°C) had little influence on its *ee* value (entry 3). However, a change of the Ag/Au ratio to 1:1 for [LAu₂Cl₂]/AgOTf increased the *ee* value to 84% at 0°C; here, one Au–Cl bond in gold catalyst remains intact (entry 4). We managed eventually to obtain (–)-**5e** with 92% *ee* and 74% yield at a low loading catalyst loading (Au/Ag = 1:1, 3.8 mol%, entry 5). We assigned the absolute configuration of (–)-**5e** according to an X-ray diffraction study on its analogue (–)-**5m** (see the Supporting Information).

With $[LAu_2Cl_2]/AgOTf$ (3.8 mol%, L = (R)-DTBM-MeO-Biphep), we conducted enantioselective cycloadditions on additional 1,6-envnes and nitrones (Scheme 5). The initial results show the enantioselectivity of the prototype product (-)-3a for which the *ee* value was up to 95%. We tested this chiral gold catalyst on the benzenoid substrates 1e, 1g, 1h, and 1j, and their corresponding products (-)-3e, (-)-3g, (-)-**3h**, and (-)-**3j** were obtained with high *ee* values (89–94%). We studied the reaction of nitrones bearing an electron-rich aniline moiety and the resulting product (-)-5b was obtained in 73% yield and 89% ee. The same reactions on several nitrones bearing an electron-deficient imine produced the cycloadducts (-)-5i, (-)-5l, and (-)-5m with satisfactory ee values (87-93%). For the nonbenzenoid substrate 11, the resulting cycloadduct (-)-31 was obtained in 84% ee. X-xay diffraction of compound (-)-5m was performed to clarify its absolute configuration.

Before this work, there was no report on the [2+2+3] cycloadditions of nitrones. The use of a 1,6-enyne as a 1,4-dipole equivalent remains a challenging task. We have developed a new diastereoselective cyclization/[2+2+3] cycloaddition cascade between 1,6-enynes and nitrones. The reactions are applicable not only to a broad range of nitrones,



Scheme 5. Gold-catalyzed enantioselective cycloadditions. [a] [substrate] = 0.19 M. [b] All product yields are reported for compounds isolated after purification on silica gel. [c] The *ee* values were determined by HPLC using a Chiralpak OD-H or AD-H column. L = (*R*)-DTBM-OMe-Biphep.

but also 1,6-enynes of benzenoid and nonbenzenoid systems. We also report the success on the enantioselective synthesis of cycloadducts using $[LAu_2Cl_2]/AgOTf$ (L = (*R*)-DTBM-MeO-Biphep). Most nitrone cycloadducts were obtained with high enantiopurity.

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Communications



Intermolecular Gold-Catalyzed Diastereo and Enantioselective [2+2+3] Cycloadditions of 1,6-Enynes with Nitrones

Going for gold: The title reaction has been developed and demonstrates a wide substrate scope with respect to the 1,6enynes and nitrones (see scheme; DCE = 1,2-dichloroethane, Tf = trifluoromethanesulfonyl). The results for the enantio-selective versions are also presented.