# LETTERS

# Synthesis of (–)-Oxycodone

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**S** Supporting Information



**ABSTRACT:** Our novel synthetic route to (-)-oxycodone, a semisynthetic opioid analgesic, features a palladium-catalyzed direct intramolecular arylation of an aryl bromide, oxidative dearomatization of a dihydrophenanthrenol, formation of a benzylic quaternary carbon by an intramolecular Michael addition of a malonate moiety, and construction of the morphinan skeleton via a Hofmann rearrangement/lactamization cascade.

O xycodone (1) is a semisynthetic analgesic<sup>1</sup> that is clinically prescribed as a primary opioid for cancer pain management (Figure 1). Although the structure of oxycodone is similar to morphine (2), oxycodone has a much better oral bioavailability, making it superior for pain management in some clinical studies.<sup>2</sup> Because of these advantages, it is hoped that more efficient analogues of oxycodone will be developed. While a variety of its analogues have been prepared to date, extensive modifications of oxycodone are hampered by the fact that it is derived from thebaine (3),<sup>3</sup> a minor constituent of opium. In order to synthesize a range of analogues that are not accessible from thebaine, we initiated our studies on the first synthesis of (-)-oxycodone starting from readily available, simple substrates.



Figure 1. Structures of oxycodone, morphine, and thebaine.

Scheme 1 outlines our retrosynthesis of (-)-oxycodone. Cleavage of the two heterocycles in 1 would lead to lactone 4. The amine moiety in 4 could be derived from the carboxylic acid of 5 via either a Curtius or a Hofmann rearrangement. Stereoselective construction of the benzylic quaternary carbon in 4 could be performed via an intramolecular Michael addition of the malonate moiety of 5 by taking advantage of the stereochemistry at C-14. Requisite 5 could be synthesized via stereoselective oxidative dearomatization of dihydrophe-

#### Scheme 1. Retrosynthesis



nanthrenol 6, which in turn would be prepared by a direct intramolecular arylation of aryl halide 7.

Our synthesis commenced with the preparation of aryl bromide 13, a substrate for the direct intramolecular arylation (Scheme 2). Protection of 2-bromoisovanillin  $(8)^4$  with a MOM group,<sup>5</sup> followed by reduction of the aldehyde moiety, afforded alcohol 9, which was converted into benzyl bromide 10. Alkylation of the enolate derived from imide  $11^6$  with 10 proceeded diastereoselectively to furnish 12.<sup>7</sup> Cleavage of the oxazolidinone auxiliary with hydrogen peroxide and lithium hydroxide, and subsequent methylation of the resulting carboxylic acid with (trimethylsilyl)diazomethane provided

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# Scheme 2. Preparation of the Aryl Bromide



the desired aryl bromide **13** in good yield and in high enantiomeric purity.

The crucial direct arylation was conducted according to Fagnou's conditions<sup>8</sup> with some modifications (Scheme 3).<sup>9</sup>





Thus, 13 was treated with palladium acetate (15 mol %) and triphenylphosphine (45 mol %) in the presence of potassium carbonate (3.0 equiv) and pivalic acid (30 mol %) in 1,4-dioxane at 100 °C to provide 14 in good yield. A small amount of the undesired debrominated product 15 was formed as an inseparable mixture. However, a three-step conversion involving reduction of the ester moiety, acetylation of the resulting primary alcohol, and reductive cleavage of the benzyl ether afforded pure dihydrophenanthrenol 16 in good yield with preservation of the optical purity.

With the requisite substrate in hand, we next attempted to introduce the C-14 hydroxy group via an oxidative dearomatization (Scheme 4). An attempted reaction of 16 with singlet oxygen<sup>10</sup> produced a 2:5 diastereomeric mixture of





17, the desired isomer, and its epimer at C-14. Oxidation of 16 with (diacetoxyiodo)benzene in aqueous 1,1,1,3,3,3-hexafluoro-2-propanol (HFIP),<sup>11</sup> on the other hand, afforded the desired isomer 17 in 69% yield as the sole isomer. The drastic change in the selectivity might be attributed to the participation of the neighboring acetoxy group to form 18 as an intermediate.<sup>12</sup>

Having successfully installed the C-14 hydroxy group, we turned our attention to the formation of the benzylic quaternary carbon (Scheme 5). Dienone 17 was subjected to



# Scheme 5. Construction of the Benzylic Quaternary Carbon

rhodium-catalyzed site-selective hydrogenation<sup>13</sup> to give enone **19**. Acylation of the tertiary alcohol of **19** with methyl malonyl chloride afforded enone **20**. Upon treatment of **20** with cesium carbonate in acetonitrile, an intramolecular Michael addition proceeded smoothly to yield the desired lactone **21** with a benzylic quaternary carbon. Dealkoxycarbonylation of **21** and subsequent cleavage of the methoxymethyl ether gave ketone **22** in high yield. With **22** in hand, the remaining tasks involved construction of the dihydrobenzofuran and the piperidine (Scheme 6). Site-



#### Scheme 6. Completion of the Synthesis

selective  $\alpha$ -bromination of **22** was conducted by means of pyridinium tribromide<sup>14</sup> to afford  $\alpha$ -bromoketone **23** as a diastereomeric mixture. Upon treatment with lithium iodide and triethylamine, both diastereomers were converted into the desired product 24.<sup>15</sup> After methanolysis of the acetyl group in 24, the resulting primary alcohol 25 was oxidized into a carboxylic acid, which was subsequently condensed with ammonia using 4-(4,6-dimethoxy-1,3,5-triazin-2-yl)-4-methylmorpholinium chloride (DMTMM)<sup>16</sup> to give amide 26. A (diacetoxyiodo)benzene-mediated Hofmann rearrangement<sup>17</sup> followed by hydrolysis of the resulting isocyanate afforded a primary amine, which spontaneously attacked the lactone to form the desired lactam 27 in good yield. Reduction of the lactam in 27 and subsequent methylation of the resulting amine gave diol 28. Finally, oxidation of the secondary alcohol to the ketone with Dess-Martin periodinane<sup>18</sup> furnished (-)-oxycodone. The spectroscopic data of (-)-oxycodone thus prepared are consistent with those reported in the literature.<sup>19</sup>

In conclusion, we have completed the first synthesis of (-)-oxycodone. The key features of our synthesis include a palladium-catalyzed direct arylation, oxidative dearomatization, formation of the benzylic quaternary carbon by an intramolecular Michael addition, and construction of the morphinan skeleton via a Hofmann rearrangement/lactamization cascade. Further improvement in the synthesis of morphinan alkaloids is currently underway in our laboratories.

## ASSOCIATED CONTENT

#### **Supporting Information**

Experimental details and spectroscopic data. This material is available free of charge via the Internet at http://pubs.acs.org.

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# Notes

The authors declare no competing financial interest.

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## REFERENCES

(1) (a) Freund, M.; Speyer, E. DE 286431, 1914. (b) Freund, M.; Speyer, E. DE 296916, 1916. (c) Freund, M.; Speyer, E. J. Prakt. Chem. **1916**, *94*, 135.

(2) (a) Heiskanen, T.; Kalso, E. Pain 1997, 73, 37. (b) Mercadante, S.; Arcuri, E. Cancer Treat. Rev. 1998, 24, 425. (c) Watson, C. P.; Babul, N. Neurology 1998, 50, 1837. (d) Silvestri, B.; Bandieri, E.; Del Prete, S.; Ianniello, G. P.; Micheletto, G.; Dambrosio, M.; Sabbatini, G.; Endrizzi, L.; Marra, A.; Aitini, E.; Calorio, A.; Garetto, F.; Nastasi, G.; Piantedosi, F.; Sidoti, V.; Spanu, P. Clin. Drug Investig. 2008, 28, 399.

(3) (a) Sándor, B.; Csaba, C.; Attila, S. *Curr. Med. Chem.* **2009**, *16*, 3215. (b) Halvorsen, H.; Lovli, T. WO 2009004491, 2009. (c) Weber, B.; Sahli, S. WO 2011117172, 2011. (d) Keskeny, E. M.; Mencel, J. J.; Dung, J.-S. WO 2012003468, 2012.

(4) Toth, J. E.; Hamann, P. R.; Fuchs, P. L. J. Org. Chem. 1998, 53, 4694.

(5) Duchek, J.; Piercy, T. G.; Gilmet, J.; Hudlicky, T. *Can. J. Chem.* **2011**, *89*, 709.

(6) For preparation of imide 11, see the Supporting Information: Arvanitis, E.; Ernst, H.; Ludwig, A. A.; Robinson, A. J.; Wyatt, P. B. J. Chem. Soc., Perkin Trans. 1 1998, 521.

(7) (a) Evans, D. A.; Ennis, M. D.; Mathre, D. J. J. Am. Chem. Soc. **1982**, 104, 1737. (b) Heemstra, J. M.; Kerrigan, S. A.; Doerge, D. R.; Helferich, W. G.; Boulanger, W. A. Org. Lett. **2006**, *8*, 5441.

(8) (a) Campeau, L.-C.; Parisien, M.; Leblanc, M.; Fagnou, K. J. Am. Chem. Soc. 2004, 126, 9186. (b) Campeau, L.-C.; Parisien, M.; Jean, A.; Fagnou, K. J. Am. Chem. Soc. 2006, 128, 581. (c) Lafrance, M.; Fagnou, K. J. Am. Chem. Soc. 2006, 128, 16496.

(9) The reaction using *N*,*N*-dimethylacetamide (DMA), a common solvent for palladium-catalyzed direct arylation, gave 14 in low yield with an unacceptable amount of 15.

(10) (a) Saito, I.; Chujo, Y.; Shimazu, H.; Yamane, M.; Matsuura, T.; Cahnmann, H. J. *J. Am. Chem. Soc.* **1975**, *97*, 5272. (b) Breton, J. L.; Llera, L. D.; Navarro, E.; Trujillo, J. *Tetrahedron* **1987**, *43*, 4447.
(c) Carreño, M. C.; González-López, M.; Urbano, A. Angew. Chem., Int. Ed. **2006**, *45*, 2737.

(11) Dohi, T.; Yamaoka, N.; Kita, Y. Tetrahedron 2010, 66, 5775.

(12) Under the (diacetoxyiodo)benzene-mediated oxidative dearomatization reaction conditions, a trace amount of acetate **29** was also produced, strongly supporting the proposed reaction pathway.



(13) Fang, L.; Chen, Y.; Huang, J.; Liu, L.; Quan, J.; Li, C.-C.; Yang, Z. J. Org. Chem. **2011**, *76*, 2479.

(14) Pyridinium tribromide was recrystallized twice from ethanol prior to use. Use of commercial pyridinium tribromide without recrystallization caused overbromination at the less hindered  $\alpha$ -position.

(15) (a) Mulzer, J.; Dürner, G.; Trauner, D. Angew. Chem., Int. Ed. 1996, 35, 2830. (b) Nagata, H.; Miyazawa, N.; Ogasawara, K. Chem. Commun. 2001, 1094. (c) Wang, P. X.; Jiang, T.; Cantrell, G. L.; Berberich, D. W.; Moser, F. W.; Bao, J. US 20090221825, 2009.

(16) Kunishima, M.; Kawachi, C.; Iwasaki, F.; Terao, K.; Tani, S. Tetrahedron Lett. **1990**, 40, 5327.

(17) Moriarty, R. M.; Chany, C. J., II; Vaid, R. K.; Prakash, O.; Tuladhar, S. M. J. Org. Chem. **1993**, 58, 2478.

(18) Dess, D. B.; Martin, J. C. J. Org. Chem. 1983, 48, 4155.

(19) <sup>1</sup>H NMR: Dung, J.-S.; Mudryk, B.; Sapino, C.; Sebastian, A. EP0889045, 1998. <sup>13</sup>C NMR: Tavakol, H.; Esfandyari, M.; Taheri, S.; Heydari, A. *Spectrochim. Acta, Part A* **2011**, *79*, 574. Optical rotation: Currie, A. C.; Gillon, J.; Newbold, G. T.; Spring, F. S. J. Chem. Soc. **1960**, 773.