LETTERS XXXX Vol. XX, No. XX 000–000

ORGANIC

Palladium-Catalyzed Medium-Ring Formation for Construction of the Core Structure of *Laurencia* Oxacycles: Synthetic Study of Laurendecumallene B

Yuji Yoshimitsu, Shinsuke Inuki, Shinya Oishi, Nobutaka Fujii,* and Hiroaki Ohno*

Graduate School of Pharmaceutical Sciences, Kyoto University, Sakyo-ku, Kyoto 606-8501, Japan

hohno@pharm.kyoto-u.ac.jp; nfujii@pharm.kyoto-u.ac.jp

Received May 2, 2013



Palladium-catalyzed medium-ring formation from a cyclic propargyl carbonate via a ring-opening and -closing cascade proceeded at the central carbon atom of the propargyl unit to provide a tetrahydro-2*H*-oxocine derivative bearing the core structure of laurencia oxacycle. The synthetic application of this reaction to a possible laurendecumallene B precursor is also presented.

The marine genus *Laurencia* specifically produces a significant subset of medium-ring haloethers as secondary metabolites: these typically consist of a C15 carbon skeleton with an envne or bromoallene side chain.¹ Since the first report of the isolation of (+)-laurencin (1) from Laurencia glandulifera by Irie et al. in 1965,² numerous medium-ring haloethers have been isolated from Laurencia red algae. Laurendecumallene B (2), a C15 acetogenin isolated from the marine red alga Laurencia decumbens, was first reported by Wang et al. in 2007 (Figure 1).³ Another structurally related laurendecumallene, known as laurendecumallene A (3), was also isolated from the same species. To date, the total synthesis of laurendecumallene B has not been reported in the literature. The relative configuration of laurendecumallene B has been partly determined, but the configurations of the axial chirality of bromoallene and the substituent at the C-13 position, and the absolute configuration of the molecule, have not yet been elucidated.

The development of synthetic strategies for the construction of these *Laurencia* oxacycles have been extensively investigated in recent years.^{4–6} However, for the syntheses of diverse collections of medium-ring natural products and

^{(1) (}a) Gribble, G. W. Acc. Chem. Res. **1998**, 31, 141–152. (b) Faulkner, D. J. Nat. Prod. Rep. **2002**, 19, 1–48. (c) Suzuki, M.; Vairappan, C. S. Curr. Top. Phytochem. **2005**, 7, 1–34.

⁽²⁾ Irie, T.; Suzuki, M.; Masamune, T. Tetrahedron Lett. 1965, 6, 1091–1099.

⁽³⁾ Ji, N. Y.; Li, X. M.; Li, K.; Wang, B. G. J. Nat. Prod. 2007, 70, 1499–1502.

⁽⁴⁾ For reviews, see: (a) Yeung, K. S.; Paterson, I. Chem. Rev. 2005, 105, 4237–4313. (b) Fujiwara, K. In Topics in Heterocyclic Chemistry, Vol. 5; Kiyota, H., Ed.; Springer-Verlag: Berlin, 2006; pp 97–148. (c) Blunt, J. W.; Copp, B. R.; Hu, W. P.; Munro, M. H. G.; Northcote, P. T.; Prinsep, M. R. Nat. Prod. Rep. 2007, 24, 31–86. (d) Morris, J. C.; Nicholas, G. M.; Phillips, A. J. Nat. Prod. Rep. 2007, 24, 87–108.

⁽⁵⁾ For selected recent syntheses based on ring-closing metathesis, see: (a) Park, J.; Kim, B.; Kim, H.; Kim, S.; Kim, D. Angew. Chem., Int. Ed. 2007, 46, 4726-4728. (b) Adsool, V. A.; Pansare, S. V. Org. Biomol. Chem. 2008, 6, 2011–2015. (c) Sasaki, M.; Hashimoto, A.; Tanaka, K.; Kawahata, M.; Yamaguchi, K.; Takeda, K. Org. Lett. 2008, 10, 1803-1806. (d) Ortega, N.; Martín, V. S.; Martín, T. J. Org. Chem. 2010, 75, 6660–6672. (e) Kim, M. J.; Sohn, T.; Kim, D.; Paton, R. S. J. Am. Chem. Soc. 2012, 134, 20178–20188.

⁽⁶⁾ For selected recent syntheses without using ring-closing metathesis, see: (a) Kim, H.; Lee, H.; Lee, D.; Kim, S.; Kim, D. J. Am. Chem. Soc. 2007, 129, 2269–2274. (b) Li, J.; Suh, J. M.; Chin, E. Org. Lett. 2010, 12, 4712–4715. (c) Snyder, S. A.; Treitler, D. S.; Brucks, A. P.; Sattler, W. J. Am. Chem. Soc. 2011, 133, 15898–15901. (d) Snyder, S. A.; Brucks, A. P.; Treitler, D. S.; Moga, I. J. Am. Chem. Soc. 2012, 134, 17714–17721.



Figure 1. Structures of naturally occurring haloethers from *Laurencia* species.

their derivatives, the development of efficient approaches for the construction of medium-ring ethers, not only those based on ring-closing metathesis,⁵ remains important.⁶ Because of the interesting activities of eight-membered *Laurencia* oxacycles,⁷ attempts were made in this study to complete the first total synthesis of laurendecumallene B.

Despite the presence of several stereogenic centers, the configuration of the axial chirality of the bromoallene moiety in *Laurencia* oxacycles can be predicted from their strong optical rotation values, which are in good agreement with Lowe's rule.⁸ Based on the positive value of the optical rotation of laurendecumallene B ($[\alpha]^{18}_{D} = +60.6$ in CHCl₃), its axial chirality was assumed to be (*S*). The target structures were therefore identified as compounds **2a–d** (Figure 2). The decision was taken to develop a synthetic route to compounds **2a/b** based on the absolute configuration of the core structures of the related bromoallenes from *Laurencia* species, such as (+)-laurallene (**4**), (+)-pannosallene (**5**), and (+)-itomanallene A (**6**) (Figure 1). If necessary, the other possible isomers **2c/d** could be formed



Figure 2. Possible structures of laurendecumallene B.

(7) Watanabe, K.; Umeda, K.; Miyakado, M. Agric. Biol. Chem. 1989, 53, 2513–2515. as their antipodes (ent-2c/d) from the common intermediate for the synthesis of 2a/b at a later stage of the synthesis.

In 2003, the author's group reported that bromoallenes such as 7, which are synthetic equivalents of propargylic compounds, were extremely useful for the synthesis of medium-sized rings 8 (Scheme 1, eq 1).9 We have also shown that cyclization through ring opening and closing is a convenient strategy for the construction of bicvclic structures.¹⁰ In 2001, Yoshida, Ihara, and co-workers developed a methodology for the synthesis of cyclic carbonates, based on a Pd-catalyzed cascade reaction involving a CO₂ elimination-fixation process.¹¹ In the current work, we tried to apply these chemistries to the synthesis of Laurencia oxacycles (eq 2), in which a cyclic propargyl carbonate 9 possessing a strategically positioned hydroxy functionality could undergo Pd-mediated medium-ring formation through ring-opening and -closing reactions via an n^3 -allylpalladium complex **B**. These compounds could then be trapped by the pendant carbonate, providing medium-sized ethers 10.

Scheme 1. Cyclization of Propargylic/Allenic Compounds



The retrosynthetic analysis of the possible stereoisomers 2a/b is shown in Scheme 2. It was envisaged that 2a/b could be derived from 12 by deoxygenation at the C-5 position, followed by the introduction of a bromoallene side chain and a Br-atom at the C-13 position. The fused tetrahydrofuran 12 could be constructed from the cyclic carbonate 13 via a sequence of hydroboration—oxidation and cyclization by intramolecular $S_N 2$ displacement reactions, which would use the resulting hydroxy group as a leaving group. The eight-membered ring in 13 could in turn be constructed via the Pd-catalyzed cyclization of the cyclic propargyl carbonate 14, which could be synthesized via a

^{(8) (}a) Lowe, G. *Chem. Commun.* **1965**, 411–413. The absolute configurations of the bromoallene moiety can be sometimes predicted by their optical rotation values, even if the molecules have several chiral centers. For example, see: (b) Ohno, H.; Ando, K.; Hamaguchi, H.; Takeoka, Y.; Tanaka, T. J. Am. Chem. Soc. **2002**, *124*, 15255–15266.

^{(9) (}a) Ohno, H.; Hamaguchi, H.; Ohata, M.; Tanaka, T. Angew. Chem., Int. Ed. 2003, 42, 1749–1753. (b) Ohno, H.; Hamaguchi, H.; Ohata, M.; Kosaka, S.; Tanaka, T. J. Am. Chem. Soc. 2004, 126, 8744–8754.

⁽¹⁰⁾ Okano, A.; Oishi, S.; Tanaka, T.; Fujii, N.; Ohno, H. J. Org. Chem. 2010, 75, 3396–3400.

 ^{(11) (}a) Yoshida, M.; Ihara, M. Angew. Chem., Int. Ed. 2001, 40, 616–619. (b) Yoshida, M.; Fujita, M.; Ishii, T.; Ihara, M. J. Am. Chem. Soc. 2003, 125, 4874–4881.

^{(12) (}a) Okude, Y.; Hirano, S.; Hiyama, T.; Nozaki, H. J. Am. Chem. Soc. 1977, 99, 3179–3181. (b) Jin, H.; Uenishi, J.; Christ, W. J.; Kishi, Y. J. Am. Chem. Soc. 1986, 108, 5644–5646. (c) Takai, K.; Tagashira, M.; Kuroda, T.; Oshima, K.; Utimoto, K.; Nozaki, H. J. Am. Chem. Soc. 1986, 108, 6048–6050.

Scheme 2. Retrosynthetic Analysis of Possible Isomers 2a/b



Nozaki–Hiyama–Kishi (NHK) coupling reaction¹² between aldehyde **15** and alkyne **16**, and several subsequent functional group manipulations. It was envisaged that the known L-arabinose-derived hemiacetal 17^{13} could be used as a common intermediate for the syntheses of **15** and **16**.

We started our synthesis by preparing **19**, the precursor of aldehyde **15** (Scheme 3). The addition of lithium acetylide (prepared from but-1-yne) to the known hemiacetal **17** was quenched with ClCO₂Me to furnish the corresponding methyl carbonate, which was subsequently treated with Pd(OAc)₂/dppb/ammonium formate to give the deoxygenated product **18**.¹⁴ Subsequent Birch reduction of the alkyne moiety occurred, with concomitant carbonate cleavage, to afford **19**.

Scheme 3. Synthesis of Alkene 19 (Precursor of 15)^a



We then proceeded toward the synthesis of alkyne **16** (Scheme 4). Using the method reported by Myers,¹⁵ **17**

was converted to alkyne **20** in 69% yield by treatment with lithiated TMS-diazomethane. The hydroxy group was then protected as the corresponding *p*-methoxyphenyl ether, using Mitsunobu conditions, to give **21** in 66% yield.¹⁶ The conversion of the acetonide to the corresponding carbonate group was conducted by exposure of **21** to TsOH·H₂O in MeOH, followed by treatment with triphosgene in the presence of pyridine to give cyclic carbonate **22** in 67% yield. The iodoalkyne **16** was obtained by iodination of the terminal alkyne in **22** using NIS/AgNO₃ in 79% yield.¹⁷

Scheme 4. Synthesis of Alkyne 16



The synthesis and cyclization of carbonate 14 is illustrated in Scheme 5. Dess–Martin oxidation of the primary alcohol 19 and subsequent Nozaki–Hiyama–Kishi coupling¹² with alkyne 16 gave a propargyl alcohol, which was transformed into thiocarbonate 23. This material was then immediately subjected to a Sharpless asymmetric dihydroxylation, followed by sequential selective monosilylation and deoxygenation reactions because the thiocarbonates of this series are relatively labile. We then





 a DMP = Dess-Martin periodinane; (DHQ)₂PHAL = hydroquinine 1,4-phthalazinediyl diether.

(16) Fukuyama, T.; Laird, A. A.; Hotchkiss, L. M. Tetrahedron Lett. 1985, 26, 6291–6292.

⁽¹³⁾ Chapman, T. M.; Courtney, S.; Hay, P.; Davis, B. G. Chem.-Eur. J. 2003, 9, 3397-3414.

^{(14) (}a) Mandai, T.; Matsumoto, T.; Tsujiguchi, Y.; Matsuoka, S.; Tsuji, J. J. Organomet. Chem. **1994**, 473, 343–352. (b) Sawada, D.; Kanai, M.; Shibasaki, M. J. Am. Chem. Soc. **2000**, 122, 10521–10532. (c) Ohmiya, H.; Yang, M.; Yamauchi, Y.; Ohtsuka, Y.; Sawamura, M. Org. Lett. **2010**, 12, 1796–1799.

⁽¹⁵⁾ Myers, A. G.; Goldberg, S. D. Angew. Chem., Int. Ed. 2000, 39, 2732–2735.

⁽¹⁷⁾ Hofmeister, H.; Annen, K.; Laurent, H.; Wiechert, R. Angew. Chem., Int. Ed. Engl. 1984, 23, 727–729.

investigated the cyclization of 14 using Pd(0). Treatment of 14 with Pd₂(dba)₃·CHCl₃ and dppf in MeOH at 50 °C only afforded the solvolysis product. However, the reaction in DMF produced the desired oxocine derivative 13, albeit in 14% yield. The best result was obtained using 30 mol % of Pd₂(dba)₃ and 80 mol % of dppf in the presence of 3 equiv of H₂O in DMF (58% yield) (for details, see Supporting Information).

The investigation proceeded toward the construction of the tetrahydrofuran ring (Scheme 6). Unfortunately, initial attempts to introduce the required hydroxy group into acetonide 13 via a hydroboration-oxidation sequence proved unsuccessful. In contrast, application of hydroboration-oxidation to the deprotected enol ether 24 proceeded smoothly with high levels of regio- and stereoselectivity to give triol 25 as a single diastereomer. Reinstallation of the acetonide group, followed by sequential mesylation and methanolysis of the cyclic carbonate, promoted the facile formation of the tetrahydrofuran ring to give 12 bearing the requisite bicyclic core structure. Xanthate formation was followed by CAN-mediated deprotection¹⁸ and Barton-McCombie deoxygenation¹⁹ to give the primary alcohol 27. The construction of the bromoallene moiety was accomplished using the well-established strategy developed by Overman and Kim.²⁰ Dess-Martin oxidation of 27 gave the corresponding aldehyde, which was treated with ethynyltitanium triisopropoxide and then desilylated to give the propargyl alcohol 28 as the sole stereoisomer.^{20,21} The trysilate 29 was converted to bromoallene 30 via the established anti S_N2' substitution using the bromocuprate reagent.²² The optical rotation value of the bromoallene 30 ($[\alpha]_{D}^{25} = +68.4$) supports the (S)-axial chirality. Exposure of 30 to the dppe-mediated bromination conditions²³ followed by cleavage of the acetonide with AcOH led to detection of the bromination product, which corresponded well with natural laurendecumallene B (2) by ¹H NMR and optical rotation values (see Supporting Information). However, further investigations are necessary to achieve the total synthesis elucidating the C-13 stereochemistry.

Kim, D.; Shin, K. J. Angew. Chem., Int. Ed. 2010, 49, 752–756.

(21) For determination of the relative configuration at the propargylic position, see Supporting Information.

(22) (a) Montury, M.; Goré, J. Synth. Commun. 1980, 10, 873–879.
(b) Elsevier, C. J.; Meijer, J.; Tadema, G.; Stehouwer, P. M.; Bos, H. J. T.; Vermeer, P.; Runge, W. J. Org. Chem. 1982, 47, 2194–2196.

(23) (a) Fujiwara, K.; Kobayashi, M.; Awakura, D.; Murai, A. Synlett **2000**, 1187–1189. See also: (b) Bratz, M.; Bullock, W. H.; Overman, L. E.; Takemoto, T. J. Ann. Chem. Soc. **1995**, 117, 5958– 5966. (c) Fujiwara, K.; Tsunashima, M.; Awakura, D.; Murai, A. Tetrahedron Lett. **1995**, 36, 8263–8266. Scheme 6. Construction of THF Ring and Bromoallene Moiety^a



In conclusion, we demonstrated that Pd-catalyzed medium-ring formation is useful for construction of the core structure of *Laurencia* oxacycles. Further studies to achieve the total synthesis of laurendecumallene B, including optimization of bromination conditions and the appropriate protecting group for the vicinal diol moiety, are now underway.

Acknowledgment. This work was supported by a Grantin-Aid for the Encouragement of Young Scientists (A) (H.O.) and Platform for Drug Design, Discovery, and Development from the MEXT, Japan. Y.Y. and S.I. are grateful for Research Fellowships from the Japan Society for the Promotion of Science (JSPS) for Young Scientists.

Supporting Information Available. Our attempts toward total synthesis, detailed results of the Pd-catalyzed cyclization, experimental procedures, and characterization data for all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

⁽¹⁸⁾ For removal of a PMP group under buffered conditions, see: Floreancig, P. E.; Swalley, S. E.; Trauger, J. W.; Dervan, P. B. J. Am. Chem. Soc. 2000, 122, 6342–6350.

⁽¹⁹⁾ Barton, D. H. R.; McCombie, S. W. J. Chem. Soc., Perkin Trans. 1 1975, 1574–1585.

^{(20) (}a) Grese, T. A.; Hutchinson, K. D.; Overman, L. E. J. Org. Chem. **1993**, 58, 2468–2477. (b) Jeong, W.; Kim, M. J.; Kim, H.; Kim, S.;

The authors declare no competing financial interest.