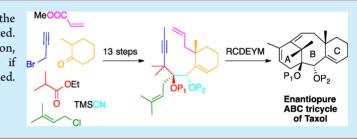
Highly Efficient Synthesis of the Tricyclic Core of Taxol by Cascade **Metathesis**

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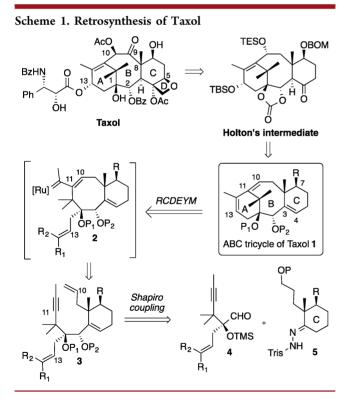
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S Supporting Information

ABSTRACT: An efficient enantioselective synthesis of the ABC tricyclic core of the anticancer drug Taxol is reported. The key step of this synthesis is a cascade metathesis reaction, which leads in one operation to the required tricycle if appropriate fine-tuning of the dienvne precursor is performed.



• axol (Scheme 1) and its derivatives are the largest selling anticancer drugs of all time,¹ and they are employed to



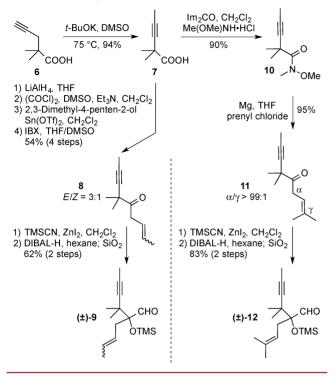
treat a wide range of malignancies.² There have been six total syntheses of Taxol by the groups of Holton,³ Nicolaou,⁴ Danishefsky,⁵ Wender,⁶ Mukaiyama,⁷ and Kuwajima,⁸ as well as one formal synthesis by Takahashi et al.9 Other elegant approaches leading to the ABC tricyclic core of taxane derivatives include work by the groups of Swindell,¹⁰ Pattenden,¹¹ Granja,¹² Winkler,¹³ and Baran.¹⁴ In our approach toward Taxol,¹⁵ we aimed for a compound described by Holton et al.^{3b} (Scheme 1). This intermediate could be synthesized from compound 1 (R = OBOM), using Danishefsky's route to convert the C3-C4 alkene into the ketone⁵ and Granja's method to install the 11-ene-10,13 diol from the C10-C13 diene.¹⁶ Herein, we report a synthetic strategy to the ABC tricyclic core of Taxol 1, featuring a challenging cascade eneyne-ene ring-closing metathesis (RCDEYM) reaction¹⁷ that allows a highly efficient access to this compound. Although such a cascade metathesis has already been employed by Granja's¹² and our group¹⁸ for the synthesis of model ABC tricycle of taxoids, in each case the relative stereochemistry of these analogues at C1, C2, and C8 was different from the one found in Taxol. Moreover, the gem dimethyl group bridging the A and B rings, whose steric hindrance presents a major difficulty in the synthesis of Taxol, was either not present¹² or not positioned at the proper place.¹⁸

We elected to validate our approach on the 7-deoxy ABC ring system 1 (R = H) (Scheme 1). Envne metathesis of compound 3 between the alkene at C10 and the alkyne at C11 would produce the intermediate carbene 2, which would lead to compound 1 after subsequent diene RCM. In order to direct the metathesis cascade so it will start with the olefin at C10, we chose to have a di- or trisubstituted olefin at C13 ($R_1 = Me_1 R_2$ = H, Me), which would react more slowly with the metathesis catalysts. Dienyne 3 would be constructed by a Shapiro coupling between aldehyde 4 and hydrazone 5.

The synthesis of the required aldehydes started from the known acid 6^{19} (Scheme 2), which is easily prepared from ethyl isobutyrate in two steps (81% overall yield). Isomerization of the terminal alkyne into the more stable internal alkyne was performed by heating 6 at 75 °C in DMSO in the presence of potassium tert-butoxide²⁰ in 93% yield, and the acid 7 was then

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Scheme 2. Synthesis of Aldehydes (\pm) -9 and (\pm) -12 for Shapiro Coupling



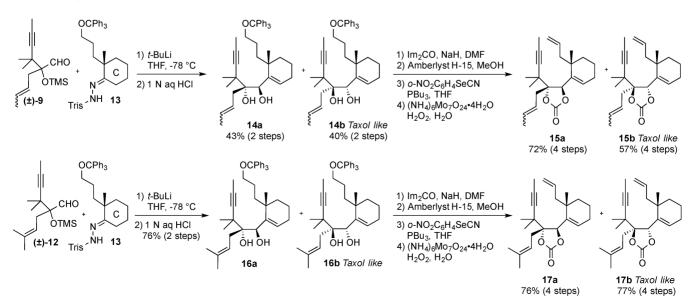
converted into ketone 8 in 4 steps as a 3:1 mixture of *E* and *Z* isomers.²¹ This ketone was then submitted to trimethylsilyl cyanide in the presence of a catalytic amount of zinc diiodide. The resulting cyanohydrin was reduced with DIBAL-H to the intermediate imine, which was hydrolyzed to the racemic aldehyde (\pm)-9 by exposure to silica gel. Acid 7 was also converted into the corresponding Weinreb amide 10, and addition of prenylmagnesium chloride to this amide furnished ketone 11 in 95% yield as the α prenylation isomer only. Cyanosilylation of ketone 11 followed by reduction of the resulting cyanohydrin with DIBAL-H gave aldehyde (\pm)-12 in excellent overall yield. Enantiopure aldehydes 9 and 12 could

Scheme 3. Synthesis of Metathesis Precursors 15a,b and 17a,b

be prepared using an enantioselective cyanation reaction,¹⁸ but we chose to pursue the synthesis of the metathesis precursors with the racemic aldehyde to study the influence of the stereochemistry of the precursors on the RCDEYM reaction outcome.

Hydrazone 13^{15h} (Scheme 3) was reacted with aldehyde (\pm) -9 using the conditions we developed previously.^{15h} As had been observed for model aldehydes,^{15d} this reaction was highly diastereoselective, giving compounds 14a and 14b after hydrolysis of the trimethylsilyl ether in excellent combined yield. These diols 14a and 14b were then submitted separately to the protection of the C1-C2 diol as the cyclic carbonate, triphenylmethyl ether hydrolysis, and dehydration of the resulting primary alcohol using the Grieco protocol²² to furnish the metathesis precursors 15a and 15b in 72% and 57% overall yield for the four steps, respectively. When the aldehyde (\pm) -12 was engaged in the Shapiro coupling, the *trans* diols 16a and 16b were obtained in 76% combined yield (Scheme 3). These diols were separately converted into the corresponding carbonates 17a and 17b in 76% and 77% overall yield, respectively, using the same protocols as that for the synthesis of 15a and 15b.

The key metathesis step was then studied. Carbonates 15a and **15b** both failed to produce tricyclic products, but led to the bicycles 18a and 18b resulting from a diene metathesis reaction between the olefins at C10 and C13 (Scheme 4). These carbonates were converted to the corresponding benzoates 19a and 19b by treatment with phenyllithium, as we have shown that the nature of the diol protecting group plays a crucial role in the outcome of metathesis reactions leading to BC ring systems of taxol.^{15h} These benzoates 19a,b also underwent diene metathesis, and the bicyclic benzoates 20a and 20b were obtained in very good yields. E and Z isomers of 15a,b and 19a,b exhibited the same behavior. We concluded that the steric hindrance around the alkyne was disfavoring the initial envne metathesis for all these substrates. Since the gemdimethyl group is an inherent part of the Taxol skeleton, it was impossible to decrease this unfavorable steric hindrance. However, we reasoned that the increased steric hindrance of the alkene at C13 of compounds 17a,b and 21a,b (Scheme 4)



ŌH OBz

Õ ō 17a

0

ÕH OBz

21a

PhLi, THF

-78 °C, 83%

19a

152

PhLi. THF

-78 °C, 94%

Scheme 4. Attempts at RCDEYM and Synthesis of the ABC Tricycle of Taxol

18a

ŌН ŌВz

20a

18a

ō

ŌH ŌBz

20a

10 mol %

Grubbs 2

Toluene

110 °C, 12 h

63%

10 mol %

Grubbs 2

Toluene

110 °C, 12 h

78%

10 mol %

Grubbs 2

Toluene

110 °C, 24 h

79%

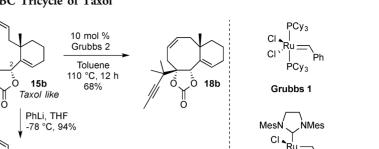
10 mol %

Grubbs 2

Toluene

110 °C, 24 h

80%



ÖH ÖBz

ō

+ 18b, 45%

ŌH ÓBz

22

45%

20b

20b

10 mol %

Grubbs 2

Toluene

110 °C. 12 h

86%

10 mol %

Grubbs 2

Toluene

110 °C, 24 h

10 mol %

Grubbs 2

Toluene 110 °C, 24 h

90%

ÔH ÔBz

19b

17b

Taxol like

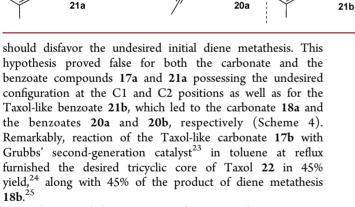
PhLi, THF

-78 °C, 93%

Ó

0

OH OBz

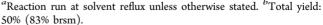


We then varied the reaction conditions in order to optimize the yield of compound 22 (Table 1). Lowering the reaction concentration did not change the ratio of 22 and 18b (entry 1 vs 2). Different catalysts were then evaluated. Unsurprisingly, the Grubbs 1 catalyst²⁶ was not active enough to perform any metathesis reaction (entry 3). The yield of 22 increased to 59% with the Hoveyda-Grubbs 2 complex²⁷ (entry 4), but the best yield (70%) was obtained with a variant of this catalyst bearing an electron-withdrawing group on the benzylidene ligand, the Zhan-1B catalyst²⁸ (entry 5), along with 20% of bicycle 18b. Changing the nature of the solvent and the reaction temperature while using the Zhan-1B catalyst did not bring any improvement (entries 6–9). The reaction did not proceed in refluxing dichloromethane (40 °C), and only degradation was observed in refluxing xylene (140 °C). Interestingly, the reaction proceeded faster in refluxing 1,2-dichloroethane (80 °C) than in toluene at the same temperature (entry 7 vs 9).

In summary, we have constructed the ABC tricyclic core of Taxol in 14 steps and 11% overall yield from ethyl isobutyrate.

Table 1. Optimization of the RCDEYM

	or 17b Taxol lin		÷	0 0 18b
entry	catalyst	solvent ^a	[C] 10 ⁻³ M	yield (%) 22:18b
1	Grubbs 2	toluene	15	45:45
2	Grubbs 2	toluene	3.5	40:40
3	Grubbs 1	toluene	5	0:0
4	HG 2	toluene	4	59:38
5	Zhan-1B	toluene	3.2	70:20
6	Zhan-1B	CH_2Cl_2	3.5	0:0
7	Zhan-1B	$Cl(CH_2)_2Cl$	3.5	65:34
8	Zhan-1B	xylene	3	degradation
9	Zhan-1B	toluene, 80 °C	3	40:10 ^b
^a Deaction run at solvent reflux unless otherwise stated				ad bratal wield.



The key step of this synthesis is an RCDEYM that leads in one operation to the required tricycle, and we have shown that we can direct the course of this metathesis reaction by adding an extra methyl substituent to the olefin at C13, which does not appear in the structure of the metathesis products. Both the nature of the protecting group and the stereochemistry of the diol at C1-C2 have a profound influence on the outcome of the metathesis reaction, and only the diastereomer with the required stereochemistry for Taxol for the C1-C2 diol,

CL,

MesN

CL Ru :

CI `

PCy₃

NMes

Grubbs 2

Hoveyda-Grubbs 2

NMes

Zhan 1B

SO₂NMe₂

MesN

CL **`**Ru

CI `

protected as a cyclic carbonate, undergoes the desired RCDEYM. Calculations are in progress to rationalize these results, and work is currently underway to achieve the formal synthesis of Taxol.

ASSOCIATED CONTENT

Supporting Information

Detailed experimental procedures, compound characterization, and crystallographic data (CIF). This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

The authors declare no competing financial interest.

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