

One-Pot Pictet-Spengler Reaction and Esterification for the Preparation of a Key Tadalafil Synthetic Intermediate

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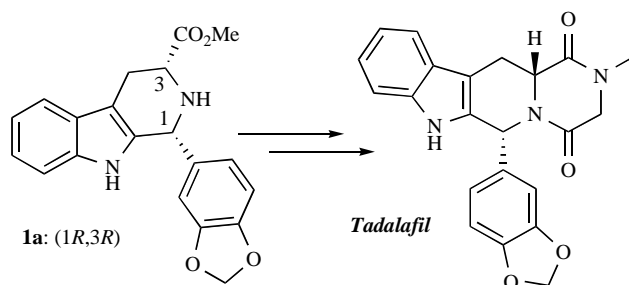
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Abstract: A new method is reported for the preparation of 2,3,4,9-tetrahydro-1H- β -carboline-3-carboxylic acid methyl ester, a key intermediate in the synthesis of Tadalafil which is used for the treatment of male erectile dysfunction, via one-pot esterification and Pictet-Spengler reaction.

Keywords: Pictet-Spengler, esterification, isomerization, Tadalafil.

The synthesis of compound **1** (Scheme 1) and its analogs have been recently studied extensively because of their importance as precursors to obtain pharmaceutical compounds such as alkaloids and drugs, the best and well-known being the Tadalafil (Scheme 1) [1]. This compound, commercialized as CialisTM (Icos/Lilly), was approved in 2003 by the FDA for the treatment of male erectile dysfunction (MED) and received much attention because of its decreased side effects and increased and effective duration when compared to Sildenafil (ViagraTM, Pfizer). Only **1a**, i.e. the (1*R*,3*R*)- or *cis*-diastereoisomer, can be used in the synthesis of Tadalafil and in the literature, countless methods have been reported to obtain *cis*-**1a**, many of them are patented. Despite the fact that numerous methods have been published or patented, the interest in **1a** and Tadalafil is still high as observed by the continuous emergence appearance of new procedures for its preparation [2]. The reaction of **1a** with chloroacetyl chloride, followed by treatment with methylamine of the so formed *N*-chloroacetylcarboline, afforded the target compound Tadalafil [2, 3].



Scheme 1.

All of the methods published so far always employed D-tryptophan methyl ester as a precursor together with 3,4-(methylenedioxy)benzaldehyde (**3**, Scheme 2), compound **1**

being usually obtained as a 1:1 mixture of diastereoisomers by means of an acid-catalyzed Pictet-Spengler reaction [4]. The reaction conditions varied for solvent (chlorinated solvents [3], isopropyl alcohol [5], methanol [6], ethyl acetate [7], and *N,N*-dimethyl acetamide [8]), temperature (from 25 [3] to 80 °C [5]), reaction time (from 16 h [5] to 7 days [7]) and isolation of the desired *cis*-**1a** (preparative chromatography [3], fractioned crystallization [6]). The 1:1 diastereomeric mixture could be isomerized in good yield to *cis*-**1a** by heating in aqueous HCl for 2-3 days [3].

The method that we report here allows to obtain compound **1** starting from the amino acid D-tryptophan **2** (and aldehyde **3**) in methanol so that Pictet-Spengler reaction occurs concomitantly with the esterification of the carboxylic moiety, with further advantage in using the less expensive D-Trp-OH instead of its methyl ester.

We studied two different strategies. The first one employed SOCl₂ for the esterification, with the HCl that developed during the reaction acting as acid catalyst for the Pictet-Spengler reaction [9]. Although the yield was very good (92% in the final product as hydrochloride) with this procedure, the use of SOCl₂ as a reagent is not appealing for possible industrial purposes, and therefore, we decided to test some protic acids that could act as catalysts for both reactions.

Whereas H₂SO₄ proved to be efficient but afforded the final product with low purity grade, HCl gave the best results, both for yield (89%) and purity of the final product [10].

To drive the esterification to completion, the water formed had to be removed. The use of dehydrating agents, such as trimethyl orthoformate failed, mainly because the Pictet-Spengler reaction did not occur; moreover, both molecular sieves [6] and anhydrous sodium sulfate [8] were used in recent patented procedures and their use was not interesting for industrial scale-up. We therefore removed the water by a continuous stripping with methanol (i.e. removal of the solvent and its continuous replacement with fresh

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- ¹³C NMR (50.33 MHz, DMSO-d₆) δ (ppm): 168.5 (s), 148.5 (s), 147.1 (s), 136.7 (s), 128.9 (s), 127.0 (s), 125.4 (s), 125.0 (d), 122.0 (d), 119.2 (d), 118.2 (d), 111.6 (d), 110.4 (d), 108.3 (d), 106.3 (s), 101.5 (t), 57.6 (d), 55.2 (d), 53.0 (q), 22.2 (t).
- [11] Borzatta, V.; Scarpi, D.; Guarna, A.; Occhiato, E.G. Process for the preparation of 2,3,4,9-tetrahydro-1H-beta-carboline-3-carboxylic esters. PCT WO2009/103787 A1, August 27, 2009. The procedure differs from the one reported in ref. 10, because compound **1** was not isolated but directly isomerized to **1a** by treatment with

aqueous 1.0 N HCl (see Ref. 3 for experimental conditions of the isomerization). The product was recovered by filtration (I cycle yield: 67% in **1a**). The residue, resulting from the mother liquor and containing a mixture of **1a**, **1b** and their corresponding carboxylic acids, was added to fresh reagents (i.e. D-Trp-OH and piperonal) in the next process, allowing the recovery of **1a** in 84% yield (II cycle only). The experiments stopped with the second cycle, but we believe that following cycles could further improve the overall yield.