



Reactions of 2-(trifluoromethyl)-2-hydroxy-2H-chromenes with silyl enol ethers promoted by AlCl_3

Liang Zhang^{a,b}, Xinjin Li^b, Zhongwei Wang^{a,*}, Jingwei Zhao^{b,*}, Jinjin Wang^a, Jianwei Han^b, Shizheng Zhu^{b,*}

^a College of Material Science and Technology, Shandong University of Science and Technology, Shandong, Qingdao 266590, China

^b Key Laboratory of Organofluorine Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 345 Lingling Road, Shanghai 200032, China

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ABSTRACT

An efficient and straightforward nucleophilic substitution reaction between 2-(trifluoromethyl)-2-hydroxy-2H-chromenes and various silyl enol ethers promoted by Lewis acid has been studied. A series of novel 4-functionalized-2-trifluoromethyl-3-ethoxycarbonyl-4H-chromenes were prepared in good yields under mild conditions.

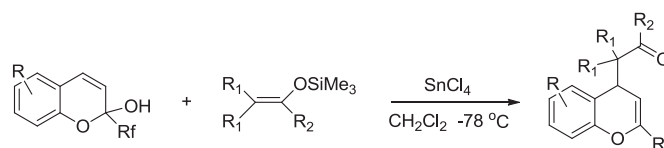
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1. Introduction

Chromenes are considered to be one of the most imperative heterocycle in nature because of their usefulness as biologically active agents.¹ In the last few years, much attention has been focused on the synthesis of chromene ring system,² for their potential application as antioxidant,³ antitubercular,⁴ anti-HIV,⁵ antitumor,⁶ cytotoxic agent,⁷ and antileishmanial.⁸ On the other hand, trifluoromethylated compounds have been recognized as one of the most potentially important and efficient probes in finding new biologically compounds⁹ because of the unique role of the CF_3 group in enhancing the bioactivity of organic molecules.¹⁰ Owing to this, 2-(trifluoromethyl)-2-hydroxy-2H-chromenes have attracted our attention to develop a new general and efficient method for the preparation of a variety of 2H-chromene heterocycles with CF_3 group.

Laurent has reported the synthesis of substituted 2-perfluoroalkyl-4H-chromenes with good yields in the presence of SnCl_4 (Scheme 1).¹¹ Hiemstra synthesized 2-substituted chromenes via ring-closing metathesis and stable 1-benzopyrylium ions.¹²

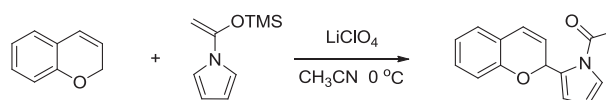
More recently, many substituted 2H-chromenes and isochromenes have been synthesized through oxidative carbon–hydrogen cleavage by Floreancig group (Scheme 2).¹³



$\text{R}=\text{H}$, p-OMe, m-OMe, o-OMe, o-Cl, p-Cl, p- NO_2

$\text{R}_1=\text{H}$, Me; $\text{R}_2=\text{H}$, Me, OMe; $\text{Rf}=\text{CF}_3$, C_3F_7 , C_6F_{11}

Scheme 1.



Scheme 2.

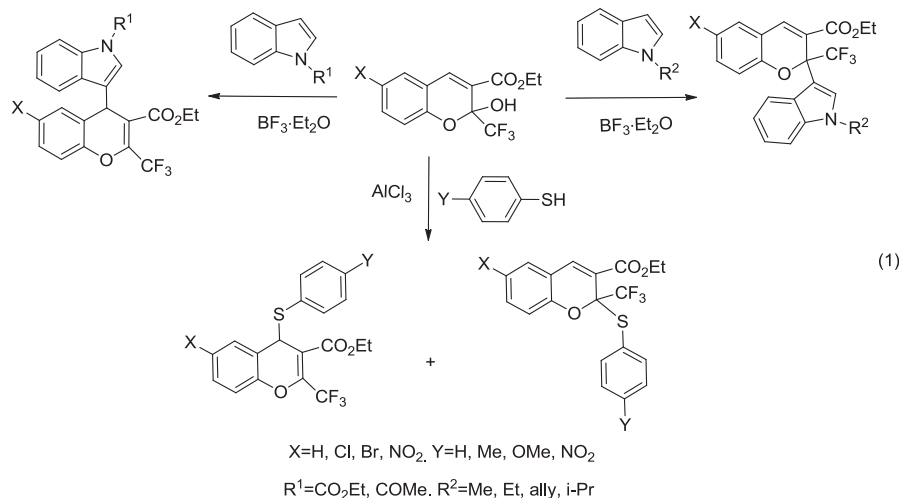
Our group has recently reported reactions of 2-(trifluoromethyl)-2-hydroxy-2H-chromenes with indoles and thiophenols promoted by Lewis acid (Scheme 3-1).^{14,15} When indoles were used as the nucleophile, only 2-substituted or 4-substituted

* Corresponding authors. E-mail address: zhuzs@mail.sioc.ac.cn (S. Zhu).

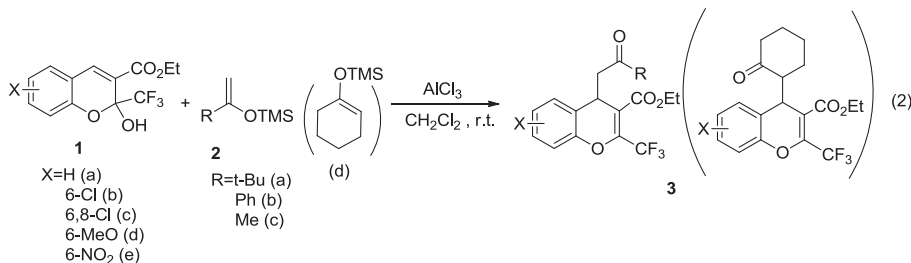
products were obtained, while both 2-substituted and 4-substituted products were isolated when the nucleophile was thiophenols. Herein, we report the reaction of 2-(trifluoromethyl)-2-hydroxy-2*H*-chromenes **1** with silyl enol ethers **2** affording a series of substituted 2-trifluoromethyl-4*H*-chromenes (Scheme 3-2).

According to Table 2, it was indicated that all these reactions occurred exclusively at 4-position of the 2*H*-chromenes. It seems that the C2 trifluoromethyl group due to its steric effect on the chromenol ring hinders the nucleophile to attack silyl enol ethers **2**. The reaction occurred between **1d** and **2a**, which both contained

Previous work:



This work:



Scheme 3.

2. Results and discussion

In the previous studies, we have reported reactions of 2-(trifluoromethyl)-2-hydroxy-2*H*-chromenes **1** with indoles and thiophenols, and found the nucleophilic substitution took place at 2-position or 4-position of **1**. In this regard, a model reaction between **1a** (1 equiv) and silyl enol ether **2a** (1.5 equiv) was examined to optimize the reaction conditions (Table 1).

It was found that no reaction took place without Lewis acids (Table 1, entries 1 and 2). When the reaction was promoted by SnCl₄, only little product was detected by ¹⁹F NMR (Table 1, entries 3 and 4). After different Lewis acids were examined, AlCl₃ was found to act as a good promoter to give the product **3aa** in 87% yield (Table 1, entry 7). When the dosage of AlCl₃ was changed, the yields decreased (Table 1, entries 6 and 8). However, when the amount of **2a** was decreased, a little lower yield of **3aa** was found under the same conditions (Table 1, entry 12). Other aprotic solvents, such as CH₃CN and THF were also examined (Table 1, entries 10 and 11). Thus, the optimal reaction conditions for this reaction were established: 1.5 equiv of **2a**, 1.0 equiv of AlCl₃, and performing the reaction at room temperature in CH₂Cl₂.

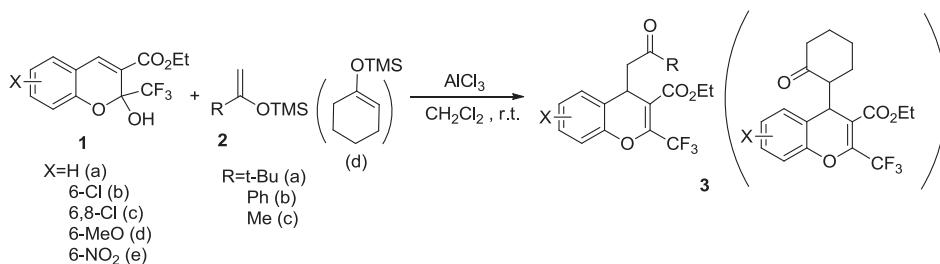
To our surprise, the nucleophilic substitution only take place at 4-position of **1a**. With the optimized reaction conditions on hand, reactions of 2-(trifluoromethyl)-2-hydroxy-2*H*-chromenes with other silyl enol ethers were further studied. The results were summarized in Table 2.

Table 1
Screening of the reaction conditions

Entry	Lewis acid (equiv)	Solvent	Time (h)	Temp	Product yield ^a
1	—	CH ₂ Cl ₂	48	rt	—
2	—	CH ₃ CN	48	rt	—
3	SnCl ₄ (0.5)	CH ₂ Cl ₂	10	−78 °C to rt	Trace
4	SnCl ₄ (1.0)	CH ₂ Cl ₂	30	−78 °C to rt	10
5	BF ₃ ·OEt ₂ (1.0)	CH ₂ Cl ₂	5	0 °C to rt	Trace
6	AlCl ₃ (0.5)	CH ₂ Cl ₂	10	rt	47
7	AlCl ₃ (1.0)	CH ₂ Cl ₂	3	rt	87
8	AlCl ₃ (1.5)	CH ₂ Cl ₂	3	rt	80
9	ZnCl ₂ (1.0)	CH ₂ Cl ₂	5	rt	23
10	AlCl ₃ (1.0)	CH ₃ CN	18	rt	54
11	AlCl ₃ (1.0)	THF	18	rt	7
12 ^b	AlCl ₃ (1.0)	CH ₂ Cl ₂	3	rt	78

^a Isolated yields.

^b The mole ratio of **1a** to **2a** was 1:1.

Table 2Reactions of **1** with **2** promoted by AlCl₃ in CH₂Cl₂ at room temperature

Entry	1	2	Time (h)	Product	Yield ^a (%)
1	1a	2a	3	3aa	87
2	1a	2b	3	3ab	85
3	1a	2c	5	3ac	63
4	1a	2d	3	3ad	83
5	1b	2a	3	3ba	83
6	1b	2b	3	3bb	57
7	1b	2c	5	3bc	34
8	1b	2d	3	3bd	62
9	1c	2a	3	3ca	38
10	1c	2b	3	3cb	29
11	1c	2d	3	3cd	33
12	1d	2a	3	3da	89
13	1d	2b	3	3db	82
14	1d	2c	5	3dc	77
15	1d	2d	3	3dd	81
16	1e	2a	10	3ea	n.r.
17	1e	2b	10	3eb	n.r.

n.r.=no reaction.

^a Isolated yields.

electron-donating functional groups ($\text{X}=\text{OCH}_3$, $\text{R}=\text{t-Bu}$), corresponding 4-substituted chromene was obtained in a good yield (Table 2, entry 12). However, when the electron-donating ability of the group X was decreased, yield of the affording product was reduced accordingly (Table 2, entries 3, 7, and 14). What is more, once a nitro group was introduced into the 2H-chromene, the reaction could not proceed to produce the desired products (Table 2, entries 16 and 17). This is probably due to the strong electron-withdrawing effect of the nitro group.

All new compounds were characterized by ¹H, ¹³C, ¹⁹F NMR spectroscopy and mass spectrometry. All of the data in the spectra were in good accordance with the structures. The structure of compound **3ca** was also confirmed by X-ray crystallography (Fig. 1).

According to the related literature¹¹ and previous research in our group, the possible reaction mechanism is proposed in Scheme 4. Initially, the hydroxy group of **1** is activated by AlCl₃ to form the intermediate **A**. Subsequently, an elimination of AlCl₂O[−] produces the oxonium intermediate **B**, which is attacked by silyl enol ethers to form product **3**. Because of the electron delocalization of oxonium intermediate, the intermediate **C** is formed

and followed by the attack of **2** to produce substituted 4H-chromenes **3**. It is noted that the attack prefer to take place at 4-position rather than at 2-position probably due to the combined action of the steric hindrance of CF₃ group and the effect of ester carbonyl group in 2H-chromenes, which can explain why 4H-chromenes **3** are the exclusive products.

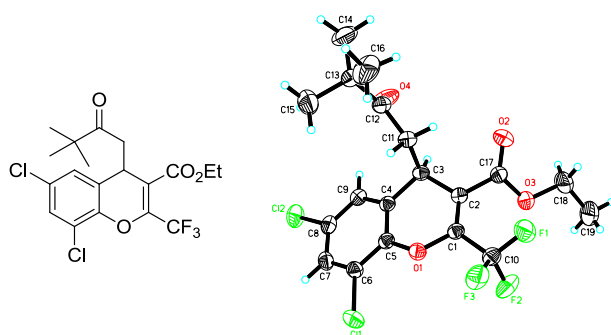
3. Conclusion

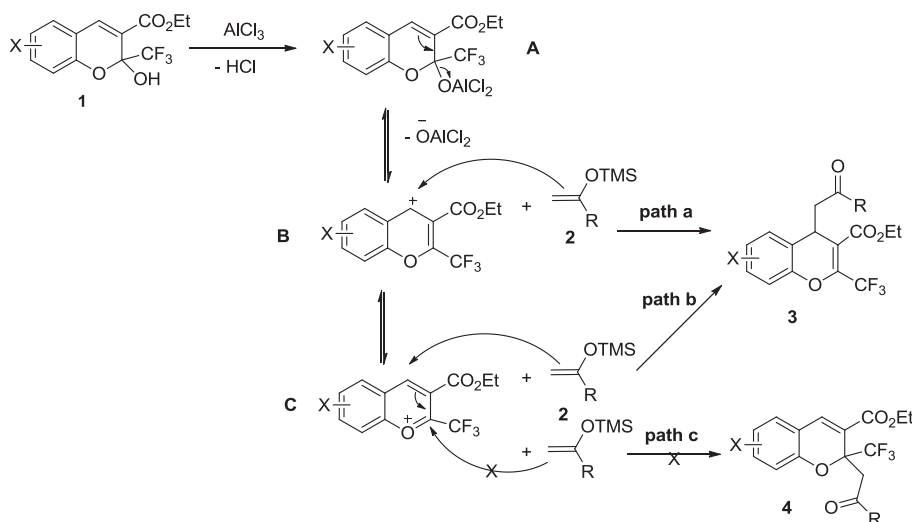
In conclusion, we have developed an efficient and new nucleophilic substitution reaction of 2-(trifluoromethyl)-2-hydroxy-2H-chromenes **1** and the silyl enol ether **2** in heterocyclic systems under mild conditions. It was found that the regioselective nucleophilic attack occurred at 4-position of the 2H-chromenes promoted by AlCl₃. The possible mechanism of the reaction was also discussed.

4. Experimental

4.1. General information

Commercial reagents were used without further purification. All solvents used were dried and purified by distillation. Chromenes **1a–d** and silyl enol ethers **2a–d** were prepared according to the literature of **2e** and **16**, accordingly. Melting points were measured on a Temp-Melt apparatus and uncorrected. ¹H NMR and ¹⁹F NMR spectra were recorded on an Agilent AM-400 instrument with Me₄Si and CFCl₃ as the internal standard, respectively. ¹³C NMR spectra were recorded on an Agilent AM-400 instrument with Me₄Si and as the internal standard. FT-IR spectra were obtained with a Nicolet AV-360 spectrophotometer. Mass spectra were obtained on a Finnigan GC-MS 4021 or a Finnigan MAT8430 instrument. Single crystal X-ray structure analysis was performed on a Bruker P4 instrument. All reactions were monitored by TLC or ¹⁹F NMR.

**Fig. 1.** The structure of compound **3ca**.



Scheme 4. The possible mechanism for the formation of 4-substituted products 3.

4.2. Typical procedure for the reaction of ethyl 2-hydroxy-2-(trifluoromethyl)-2H-chromene-3-carboxylate (1a) with silyl enol ether (2a) in the presence of AlCl_3 in CH_2Cl_2

To a stirred solution of chromene (1a, 288 mg, 1 mmol) and anhydrous aluminum chloride (133 mg, 1 mmol) in anhydrous dichloromethane (12 mL) under nitrogen was added silyl enol ether (2a, 258 mg, 1.5 mmol) at room temperature. At the end of the reaction as judged by ^{19}F NMR spectroscopy or TLC, the mixture was diluted with water (10 mL), quenched carefully with a solution of sodium bicarbonate (15 mL), and then extracted with dichloromethane (3×20 mL). The organic layers were washed with brine (2×20 mL). Then the organic layers were dried over sodium sulfate, filtered, and concentrated in vacuo. The crude products were chromatographed over silica gel using an eluent of ethyl acetate/petroleum ether (1:20) to afford the desired chromene.

4.2.1. 2-(Trifluoromethyl)-3-ethoxycarbonyl-4-(3,3-dimethyl-2-oxobutyl)-4H-chromene (3aa). Yield 87%, yellow oil. IR (KBr): 2973, 1734, 1710, 1489, 1479, 1460, 1377, 1300, 1232, 1197, 1150, 1111, 1087, 1034, 759 cm^{-1} . ^1H NMR (400 MHz, CDCl_3): δ 0.99 (s, 9H), 1.29 (t, $J=7.2$ Hz, 3H), 2.85–3.00 (m, 2H), 4.23 (q, $J=7.2$ Hz, 2H), 4.36–4.39 (m, 1H), 7.05–7.10 (m, 2H), 7.17–7.26 (m, 2H). ^{13}C NMR (100 MHz, CDCl_3): δ 13.8, 25.8, 32.2, 44.2, 45.2, 61.9, 114.0, 115.0 (q, $^1J_{\text{C-F}}=271$ Hz, CF_3), 116.4, 123.4, 125.3, 128.2, 128.5, 142.0 (q, $^2J_{\text{C-F}}=37$ Hz, CF_3), 150.0, 165.2, 212.4. ^{19}F NMR (376 MHz, CDCl_3): δ –67.3 (s, 3F, CF_3). MS (ESI): $m/z=393.2$ ($\text{M}+\text{Na}^+$). HRMS (ESI): m/z calcd for $(\text{C}_{19}\text{H}_{21}\text{F}_3\text{O}_4+\text{Na})^+$: 393.1290; found: 393.1284.

4.2.2. 2-(Trifluoromethyl)-3-ethoxycarbonyl-4-(2-oxo-2-phenylethyl)-4H-chromene (3ab). Yield 85%, yellow oil. IR (KBr): 1732, 1686, 1586, 1489, 1449, 1377, 1292, 1226, 1197, 1163, 1111, 1085, 1002, 757, 689 cm^{-1} . ^1H NMR (400 MHz, CDCl_3): δ 1.27 (t, $J=7.2$ Hz, 3H), 3.30–3.39 (m, 1H), 3.54–3.61 (m, 1H), 4.22 (q, $J=7.2$ Hz, 2H), 4.58–4.61 (m, 1H), 7.06–7.11 (m, 2H), 7.20–7.30 (m, 2H), 7.44 (t, $J=10.0$ Hz, 2H), 7.53 (t, $J=10.0$ Hz, 1H), 7.91 (d, $J=10.0$ Hz, 2H). ^{13}C NMR (100 MHz, CDCl_3): δ 13.7, 32.4, 47.2, 62.0, 113.8, 116.5, 119.1 (q, $^1J_{\text{C-F}}=271$ Hz, CF_3), 123.2, 125.5, 128.1, 128.4, 128.6, 128.7, 133.4, 136.6, 142.8 (q, $^2J_{\text{C-F}}=37$ Hz, CF_3), 150.0, 165.3, 196.8. ^{19}F NMR (376 MHz, CDCl_3): δ –67.6 (s, 3F, CF_3). MS (ESI): $m/z=413.1$ ($\text{M}+\text{Na}^+$). HRMS (ESI): m/z calcd for $(\text{C}_{21}\text{H}_{17}\text{F}_3\text{O}_4+\text{Na})^+$: 413.0977; found: 413.0971.

4.2.3. 2-(Trifluoromethyl)-3-ethoxycarbonyl-4-(2-oxopropyl)-4H-chromene (3ac). Yield 63%, yellow oil. IR (KBr): 2983, 2926, 1725,

1587, 1490, 1459, 1377, 1299, 1227, 1198, 1161, 1112, 1088, 1039, 759 cm^{-1} . ^1H NMR (400 MHz, CDCl_3): δ 1.29 (t, $J=7.2$ Hz, 3H), 2.08 (s, 3H), 2.75 (q, $J=7.6$ Hz, 1H), 2.80 (q, $J=4.4$ Hz, 1H), 4.23 (q, $J=7.2$ Hz, 2H), 4.35–4.36 (m, 1H), 7.04–7.12 (m, 2H), 7.22 (d, $J=6.4$ Hz, 2H). ^{13}C NMR (100 MHz, CDCl_3): δ 13.7, 30.4, 31.9, 51.6, 61.9, 113.4, 114.9 (q, $^1J_{\text{C-F}}=273$ Hz, CF_3), 116.5, 123.0, 125.5, 128.4, 128.5, 141.8 (q, $^2J_{\text{C-F}}=37$ Hz, CF_3), 149.8, 165.2, 205.2. ^{19}F NMR (376 MHz, CDCl_3): δ –67.4 (s, 3F, CF_3). MS (ESI): $m/z=351.1$ ($\text{M}+\text{Na}^+$). HRMS (ESI): m/z calcd for $(\text{C}_{16}\text{H}_{15}\text{F}_3\text{O}_4+\text{Na})^+$: 351.0820; found: 351.0814.

4.2.4. 2-(Trifluoromethyl)-3-ethoxycarbonyl-4-(2-oxocyclohexyl)-4H-chromene (3ad). Yield 83%, yellow oil. IR (KBr): 2938, 2866, 1731, 1712, 1585, 1487, 1459, 1379, 1299, 1235, 1197, 1149, 1112, 1085, 994, 760 cm^{-1} . ^1H NMR (400 MHz, CDCl_3): δ 1.31 (t, $J=7.2$ Hz, 3H), 1.33–1.45 (m, 2H), 1.62–1.84 (m, 2H), 1.95–2.02 (m, 2H), 2.23–2.35 (m, 1H), 2.41–2.47 (m, 1H), 2.82–2.88 (m, 1H), 4.21 (q, $J=7.2$ Hz, 2H), 4.72 (s, 1H), 6.34 (d, $J=10.8$ Hz, 1H), 7.26 (q, $J=10.4$ Hz, 1H), 7.23 (t, $J=10.0$ Hz, 1H), 7.41 (d, $J=11.2$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3): δ 13.7, 24.7, 26.3, 26.9, 34.9, 41.8, 56.0, 61.9, 113.1, 116.0, 119.1 (q, $^1J_{\text{C-F}}=273$ Hz, CF_3), 120.8, 125.4, 128.3, 130.1, 143.5 (q, $^2J_{\text{C-F}}=37$ Hz, CF_3), 151.3, 165.4, 209.5. ^{19}F NMR (376 MHz, CDCl_3): δ –67.5 (s, 3F, CF_3). MS (ESI): $m/z=391.0$ ($\text{M}+\text{Na}^+$). HRMS (ESI): m/z calcd for $(\text{C}_{19}\text{H}_{19}\text{F}_3\text{O}_4+\text{Na})^+$: 391.1133; found: 391.1128.

4.2.5. 2-(Trifluoromethyl)-3-ethoxycarbonyl-4-(3,3-dimethyl-2-oxobutyl)-6-chloro-4H-chromene (3ba). Yield 83%, yellow oil. IR (KBr): 2971, 1735, 1710, 1480, 1373, 1307, 1236, 1198, 1152, 1119, 1080, 998, 819 cm^{-1} . ^1H NMR (400 MHz, CDCl_3): δ 1.23 (s, 9H), 1.29 (t, $J=7.2$ Hz, 3H), 2.85–3.02 (m, 2H), 4.23 (q, $J=7.2$ Hz, 2H), 4.32–4.25 (m, 1H), 6.99–7.02 (m, 1H), 7.17–7.19 (m, 2H). ^{13}C NMR (100 MHz, CDCl_3): δ 13.8, 25.9, 32.0, 44.1, 45.0, 62.0, 113.6, 114.8 (q, $^1J_{\text{C-F}}=271$ Hz, CF_3), 117.8, 125.0, 128.2, 128.4, 130.3, 141.8 (q, $^2J_{\text{C-F}}=37$ Hz, CF_3), 148.4, 164.9, 212.2. ^{19}F NMR (376 MHz, CDCl_3): δ –67.3 (s, 3F, CF_3). MS (ESI): $m/z=427.2$ ($\text{M}+\text{Na}^+$). HRMS (ESI): m/z calcd for $(\text{C}_{19}\text{H}_{20}\text{ClF}_3\text{O}_4+\text{Na})^+$: 427.0900; found: 427.0894.

4.2.6. 2-(Trifluoromethyl)-3-ethoxycarbonyl-4-(2-oxo-2-phenylethyl)-6-chloro-4H-chromene (3bb). Yield 57%, yellow solid, mp 108–110 °C. IR (KBr): 1740, 1690, 1487, 1377, 1277, 1244, 1229, 1198, 1141, 1079, 1003, 830, 760 cm^{-1} . ^1H NMR (400 MHz, CDCl_3): δ 1.25 (t, $J=7.2$ Hz, 3H), 3.32 (q, $J=7.2$ Hz, 1H), 3.55 (q, $J=4.0$ Hz, 1H), 4.22 (q, $J=7.2$ Hz, 2H), 4.55 (m, 1H), 7.00 (d, $J=8.8$ Hz, 1H), 7.16 (q, $J=2.4$ Hz, 1H), 7.30 (d, $J=2.4$ Hz, 1H), 7.43 (t, $J=8.0$ Hz, 2H), 7.54 (t, $J=7.6$ Hz, 1H), 7.90 (d, $J=7.6$ Hz, 2H). ^{13}C NMR (100 MHz, CDCl_3): δ 13.7, 32.0,

46.8, 62.1, 113.4, 114.8 (q, $^1J_{C-F}=273$ Hz, CF₃), 117.9, 124.8, 128.1, 128.5, 128.7, 130.4, 133.5, 136.3, 142.1 (q, $^2J_{C-F}=37$ Hz, CF₃), 148.5, 164.9, 196.3. ¹⁹F NMR (376 MHz, CDCl₃): δ -67.2 (s, 3F, CF₃). MS (ESI): $m/z=447.0$ (M+Na)⁺. HRMS (ESI): m/z calcd for (C₂₁H₁₆ClF₃O₄+Na)⁺: 447.0587; found: 447.0581.

4.2.7. 2-(Trifluoromethyl)-3-ethoxycarbonyl-4-(2-oxopropyl)-6-chloro-4H-chromene (3bc). Yield 34%, yellow oil. IR (KBr): 2983, 2920, 1726, 1483, 1374, 1307, 1244, 1199, 1162, 1082, 1032, 814, 732 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ 1.29 (t, $J=7.2$ Hz, 3H), 2.11 (s, 3H), 2.77 (q, $J=7.2$ Hz, 1H), 3.03 (q, $J=4.0$ Hz, 1H), 4.23 (q, $J=7.2$ Hz, 2H), 4.30–4.32 (m, 1H), 6.99 (d, $J=8.4$ Hz, 1H), 7.17–7.24 (m, 2H). ¹³C NMR (100 MHz, CDCl₃): δ 13.7, 30.3, 31.6, 51.3, 62.1, 113.1, 114.8 (q, $^1J_{C-F}=273$ Hz, CF₃), 117.9, 124.7, 128.3, 128.5, 130.4, 142.1 (q, $^2J_{C-F}=37$ Hz, CF₃), 148.4, 164.9, 204.8. ¹⁹F NMR (376 MHz, CDCl₃): δ -67.4 (s, 3F, CF₃). MS (ESI): $m/z=385.2$ (M+Na)⁺. HRMS (ESI): m/z calcd for (C₁₆H₁₃ClF₃O₄+Na)⁺: 385.0430; found: 385.0425.

4.2.8. 2-(Trifluoromethyl)-3-ethoxycarbonyl-4-(2-oxocyclohexyl)-6-chloro-4H-chromene (3bd). Yield 62%, yellow oil. IR (KBr): 2938, 2858, 1712, 1480, 1450, 1374, 1351, 1307, 1238, 1199, 1151, 1083, 997, 818, 740 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ 0.91 (m, $J=3.6$ Hz, 1H), 1.28 (t, $J=7.2$ Hz, 3H), 1.46–1.53 (m, 1H), 1.60–1.66 (m, 1H), 1.82–1.85 (m, 1H), 1.92–2.00 (m, 2H), 2.24–2.33 (m, 1H), 2.44–2.47 (m, 1H), 2.82 (m, 1H), 4.24 (q, $J=7.2$ Hz, 2H), 4.67 (s, 1H), 6.96 (d, $J=8.8$ Hz, 1H), 7.17 (q, $J=2.4$ Hz, 1H), 7.43 (d, $J=2.4$ Hz, 1H). ¹³C NMR (100 MHz, CDCl₃): δ 13.7, 24.7, 26.4, 26.8, 34.9, 41.8, 55.8, 62.0, 112.7, 114.8 (q, $^1J_{C-F}=273$ Hz, CF₃), 117.4, 122.5, 128.4, 129.8, 130.4, 142.9 (q, $^2J_{C-F}=37$ Hz, CF₃), 149.8, 165.1, 209.3. ¹⁹F NMR (376 MHz, CDCl₃): δ -67.2 (s, 3F, CF₃). MS (ESI): $m/z=425.1$ (M+Na)⁺. HRMS (ESI): m/z calcd for (C₁₉H₁₈ClF₃O₄+Na)⁺: 425.0743; found: 425.0738.

4.2.9. 2-(Trifluoromethyl)-3-ethoxycarbonyl-4-(3,3-dimethyl-2-oxobutyl)-6,8-dichloro-4H-chromene (3ca). Yield 38%, yellow solid, mp 93–94 °C. IR (KBr): 2978, 1723, 1703, 1574, 1458, 1375, 1309, 1244, 1197, 1165, 1090, 1016, 872, 775 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ 1.04 (s, 9H), 1.29 (t, $J=7.2$ Hz, 3H), 2.84–3.03 (m, 2H), 4.24 (q, $J=7.2$ Hz, 2H), 4.35–4.37 (m, 1H), 7.10 (d, $J=3.2$ Hz, 1H), 7.28 (d, $J=2.4$ Hz, 1H). ¹³C NMR (100 MHz, CDCl₃): δ 13.7, 25.9, 32.5, 44.1, 44.9, 62.2, 114.5, 114.5 (q, $^1J_{C-F}=271$ Hz, CF₃), 122.9, 126.4, 126.7, 128.8, 130.2, 141.8 (q, $^2J_{C-F}=37$ Hz, CF₃), 144.8, 164.4, 211.8. ¹⁹F NMR (376 MHz, CDCl₃): δ -67.1 (s, 3F, CF₃). MS (ESI): $m/z=461.1$ (M+Na)⁺. HRMS (ESI): m/z calcd for (C₁₉H₁₉Cl₂F₃O₄+Na)⁺: 461.0510; found: 461.0505.

4.2.10. 2-(Trifluoromethyl)-3-ethoxycarbonyl-4-(2-oxo-2-phenylethyl)-6,8-dichloro-4H-chromene (3cb). Yield 29%, yellow solid, mp 74–75 °C. IR (KBr): 1729, 1682, 1598, 1456, 1360, 1307, 1266, 1242, 1202, 1168, 1086, 1011, 978, 862, 780, 756, 692 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ 1.26 (t, $J=7.2$ Hz, 3H), 3.32 (q, $J=7.2$ Hz, 1H), 3.54 (q, $J=4.0$ Hz, 1H), 4.23 (q, $J=7.2$ Hz, 2H), 4.57–4.58 (m, 1H), 7.25 (q, $J=2.0$ Hz, 2H), 7.44 (t, $J=6.8$ Hz, 2H), 7.56 (t, $J=6.8$ Hz, 1H), 7.89 (d, $J=8.4$ Hz, 2H). ¹³C NMR (100 MHz, CDCl₃): δ 13.7, 32.6, 46.6, 62.3, 114.4, 114.7 (q, $^1J_{C-F}=273$ Hz, CF₃), 116.5, 123.0, 125.9, 126.3, 127.0, 128.1, 128.8, 129.0, 130.3, 133.6, 136.3, 142.2 (q, $^2J_{C-F}=37$ Hz, CF₃), 145.0, 164.5, 196.0. ¹⁹F NMR (376 MHz, CDCl₃): δ -67.0 (s, 3F, CF₃). MS (ESI): $m/z=457.0$ (M-H)⁻. HRMS (ESI): m/z calcd for (C₂₁H₁₄Cl₂F₃O₄+Na)⁺: 457.0221; found: 457.0223.

4.2.11. 2-(Trifluoromethyl)-3-ethoxycarbonyl-4-(2-oxocyclohexyl)-6,8-dichloro-4H-chromene (3cd). Yield 33%, yellow solid, mp 82–83 °C. IR (KBr): 2932, 1724, 1705, 1575, 1461, 1373, 1309, 1256, 1240, 1198, 1166, 1091, 1004, 870, 787 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ 0.95 (m, $J=3.6$ Hz, 1H), 1.29 (t, $J=7.2$ Hz, 3H), 1.47–1.54 (m, 1H), 1.60–1.67 (m, 1H), 1.85–1.88 (m, 1H), 1.93–2.03 (m, 2H), 2.25–2.34 (m, 1H), 2.45–2.48 (m, 1H), 2.82–2.87 (m, 1H), 4.25 (q,

$J=7.2$ Hz, 2H), 4.68 (s, 1H), 7.30 (d, $J=2.4$ Hz, 1H), 7.37 (d, 2.4 Hz, 1H). ¹³C NMR (100 MHz, CDCl₃): δ 13.7, 24.7, 26.4, 27.0, 35.4, 41.8, 55.7, 62.2, 113.6, 114.7 (q, $^1J_{C-F}=273$ Hz, CF₃), 122.7, 123.9, 128.4, 129.0, 130.3, 142.9 (q, $^2J_{C-F}=37$ Hz, CF₃), 146.0, 164.6, 209.0. ¹⁹F NMR (376 MHz, CDCl₃): δ -66.9 (s, 3F, CF₃). MS (ESI): $m/z=459.1$ (M+Na)⁺. HRMS (ESI): m/z calcd for (C₁₉H₁₇Cl₂F₃O₄+Na)⁺: 459.0354; found: 459.0348.

4.2.12. 2-(Trifluoromethyl)-3-ethoxycarbonyl-4-(3,3-dimethyl-2-oxobutyl)-6-methoxy-4H-chromene (3da). Yield 89%, yellow oil. IR (KBr): 2967, 1710, 1499, 1466, 1397, 1375, 1307, 1265, 1213, 1151, 1088, 1036, 738 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ 1.00 (s, 9H), 1.28 (t, $J=7.2$ Hz, 3H), 2.85–2.99 (m, 2H), 3.72 (s, 3H), 4.21 (q, $J=7.2$ Hz, 2H), 4.34–4.36 (m, 1H), 6.66 (d, $J=2.8$ Hz, 1H), 6.72 (d, $J=2.8$ Hz, 1H), 6.96 (d, $J=9.2$ Hz, 1H). ¹³C NMR (100 MHz, CDCl₃): δ 13.8, 25.9, 32.5, 44.2, 55.7, 61.8, 112.1, 112.8, 114.5, 114.9 (q, $^1J_{C-F}=271$ Hz, CF₃), 117.3, 124.2, 142.4 (q, $^2J_{C-F}=37$ Hz, CF₃), 143.8, 156.8, 165.3, 212.6. ¹⁹F NMR (376 MHz, CDCl₃): δ -67.2 (s, 3F, CF₃). MS (ESI): $m/z=423.1$ (M+Na)⁺. HRMS (ESI): m/z calcd for (C₂₀H₂₃F₃O₅+Na)⁺: 423.1395; found: 423.1390.

4.2.13. 2-(Trifluoromethyl)-3-ethoxycarbonyl-4-(2-oxo-2-phenylethyl)-6-methoxy-4H-chromene (3db). Yield 82%, yellow solid, mp 53–55 °C. IR (KBr): 2932, 1732, 1688, 1500, 1444, 1375, 1308, 1242, 1217, 1151, 1087, 1032, 689 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ 1.25 (t, $J=7.2$ Hz, 3H), 3.31 (q, $J=8.0$ Hz, 1H), 3.54 (q, $J=4.0$ Hz, 1H), 3.72 (s, 3H), 4.22 (q, $J=7.2$ Hz, 2H), 4.56–4.57 (m, 1H), 6.73 (m, 2H), 6.99 (d, $J=8.8$ Hz, 1H), 7.42 (t, $J=7.2$ Hz, 2H), 7.53 (t, $J=7.2$ Hz, 1H), 7.90 (d, $J=7.6$ Hz, 2H). ¹³C NMR (100 MHz, CDCl₃): δ 13.7, 32.6, 47.2, 55.6, 61.9, 112.5, 114.5, 115.0 (q, $^1J_{C-F}=273$ Hz, CF₃), 117.4, 123.9, 128.1, 128.6, 133.4, 136.5, 142.7 (q, $^2J_{C-F}=37$ Hz, CF₃), 156.9, 165.4, 196.9. ¹⁹F NMR (376 MHz, CDCl₃): δ -67.1 (s, 3F, CF₃). MS (ESI): $m/z=443.1$ (M+Na)⁺. HRMS (ESI): m/z calcd for (C₂₂H₁₉F₃O₅+Na)⁺: 443.1082; found: 443.1077.

4.2.14. 2-(Trifluoromethyl)-3-ethoxycarbonyl-4-(2-oxopropyl)-6-methoxy-4H-chromene (3dc). Yield 77%, yellow oil. IR (KBr): 2937, 1727, 1500, 1467, 1436, 1375, 1308, 1215, 1151, 1088, 1035, 817 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ 1.29 (t, $J=7.2$ Hz, 3H), 2.09 (s, 3H), 2.76 (q, $J=7.6$ Hz, 1H), 3.02 (q, $J=4.0$ Hz, 1H), 3.76 (s, 3H), 4.22 (q, $J=7.2$ Hz, 2H), 4.33 (m, 1H), 6.75 (d, $J=7.6$ Hz, 2H), 6.98 (d, $J=8.4$ Hz, 1H). ¹³C NMR (100 MHz, CDCl₃): δ 13.7, 30.4, 32.2, 51.6, 55.7, 61.9, 112.2, 112.3, 114.5, 115.0 (q, $^1J_{C-F}=273$ Hz, CF₃), 117.4, 123.9, 142.3 (q, $^2J_{C-F}=37$ Hz, CF₃), 143.7, 157.0, 165.3, 205.4. ¹⁹F NMR (376 MHz, CDCl₃): δ -67.2 (s, 3F, CF₃). MS (ESI): $m/z=381.1$ (M+Na)⁺. HRMS (ESI): m/z calcd for (C₁₇H₁₇F₃O₅+Na)⁺: 381.0926; found: 381.0920.

4.2.15. 2-(Trifluoromethyl)-3-ethoxycarbonyl-4-(2-oxocyclohexyl)-6-methoxy-4H-chromene (3dd). Yield 81%, yellow solid, mp 79–81 °C. IR (KBr): 2940, 1719, 1705, 1500, 1317, 1294, 1244, 1212, 1191, 1153, 1136, 1088, 885, 820 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ 0.90 (q, $J=10.4$ Hz, 1H), 1.28 (t, $J=7.2$ Hz, 3H), 1.43–1.46 (m, 1H), 1.63–1.67 (m, 2H), 1.79–1.82 (m, 1H), 1.91–1.99 (m, 2H), 2.24–2.32 (m, 1H), 2.41–2.45 (m, 1H), 2.81–2.86 (m, 1H), 3.76 (s, 3H), 4.22 (q, $J=7.2$ Hz, 2H), 4.68 (s, 1H), 6.74 (d, $J=8.8$ Hz, 1H), 6.94 (d, $J=6.4$ Hz, 2H). ¹³C NMR (100 MHz, CDCl₃): δ 13.7, 24.7, 26.4, 26.7, 35.2, 41.8, 55.7, 56.0, 61.8, 111.7, 113.6, 114.6, 115.0 (q, $^1J_{C-F}=273$ Hz, CF₃), 116.9, 121.5, 143.6 (q, $^2J_{C-F}=37$ Hz, CF₃), 145.3, 156.9, 165.5, 209.8. ¹⁹F NMR (376 MHz, CDCl₃): δ -67.0 (s, 3F, CF₃). MS (ESI): $m/z=421.1$ (M+Na)⁺. HRMS (ESI): m/z calcd for (C₂₀H₂₁F₃O₅+Na)⁺: 421.1240; found: 421.1233.

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Supplementary data

Supplementary data associated with this article can be found in the online version, at <http://dx.doi.org/10.1016/j.tet.2013.07.004>. These data include MOL files and InChIKeys of the most important compounds described in this article.

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