



Cite this: *Chem. Commun.*, 2014, 50, 13706

Received 14th August 2014,
Accepted 2nd September 2014

DOI: 10.1039/c4cc06395c

www.rsc.org/chemcomm

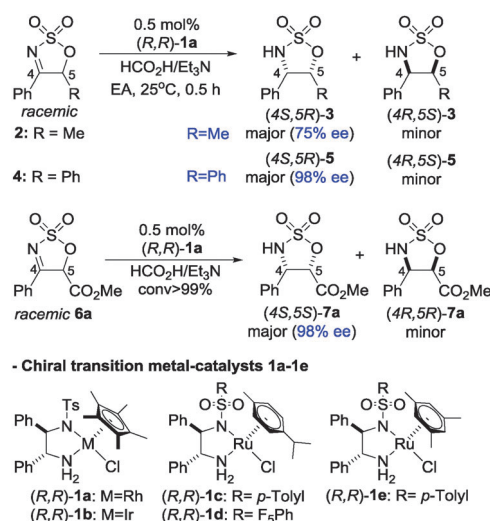
Stereoselective synthesis of 4-substituted-cyclic sulfamidate-5-carboxylates by asymmetric transfer hydrogenation accompanied by dynamic kinetic resolution and applications to concise stereoselective syntheses of (–)-*epi*-cytoxazone and the taxotere side-chain†

Jin-ah Kim,‡^a Yeon Ji Seo,‡^{ab} Soyeong Kang,^{ab} Juae Han^{ab} and Hyeon-Kyu Lee^{*ab}

Dynamic kinetic resolution driven, asymmetric transfer hydrogenation reactions of cyclic sulfamidate imine-5-carboxylate esters were developed. Applications of the new methodology to stereoselective syntheses of the taxotere side-chain and (–)-*epi*-cytoxazone are described.

1,2-Amino alcohol motifs, including those found in β -amino- α -hydroxy acids, are present in a vast range of natural products and pharmaceutically related compounds.¹ In addition, the relative and absolute stereochemistry of the 1,2-amino alcohol moiety generally governs the biological activities of these substances. Therefore, the development of methods for stereoselective synthesis of members of this family has received considerable attention.^{1a,2}

Transition metal catalyzed-asymmetric transfer hydrogenation reactions (ATH)³ of carbonyl compounds containing configurationally labile stereogenic C–H centers, accompanied by dynamic kinetic resolution (DKR), have become efficient and powerful techniques for controlling the stereochemistry at two contiguous stereogenic centers. Examples of processes of this type include ATH of α -substituted- β -ketoesters,⁴ β -ketoamides,⁵ α -alkoxy- β -keto phosphonates,⁶ 1,2-diketones,⁷ α -ketoesters,⁸ and α -ketophosphonates.⁹ However, only a few reports exist describing ATH reactions of imines that are accompanied by DKR.¹⁰ In this context, we recently described a highly efficient procedure for ATH–DKR of prochiral cyclic sulfamidate imines, using $\text{HCO}_2\text{H}/\text{Et}_3\text{N}$ as the hydrogen source and chiral Rh-catalysts (Scheme 1).^{10a,b} In this early effort,



Scheme 1

we showed that ATH of 4,5-disubstituted cyclic sulfamidate imines **2**, possessing configurationally labile stereogenic centers (C5), is accompanied by DKR. It was also observed that DKR is caused by rapid racemization at the acidic stereogenic C5 position adjacent to the imine carbon under the reaction conditions. In fact, introduction of an aryl in place of a methyl group at C-5 of **2** leads to drastic improvement in the stereoselectivity of the ATH reaction (e.g., from 75% ee for **3** to 98% ee for **5**), an obvious consequence of the enhanced acidity of H-5 (Scheme 1).^{10a}

While considering other strategies to improve the stereoselectivity of ATH–DKR reactions of cyclic imine **2**, we envisioned that introduction of a carboxylate group at C-5 would also enhance the acidity of H-5 and, as a result, would promote high levels of stereoselectivity in the ATH–DKR reaction.

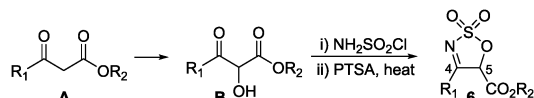
Below, we describe the results of an investigation exploring this proposal, which led to the first examples of highly efficient ATH reactions of cyclic imines **6**, which are accompanied by

^a Korea Chemical Bank, Korea Research Institute of Chemical Technology, PO Box 107, Yuseong, Daejeon 305-600, Korea. E-mail: leehk@kriict.re.kr; Fax: +82 42 860-7096; Tel: +82 42 860-7016

^b Department of Medicinal and Pharmaceutical Chemistry, University of Science and Technology, Daejeon 305-333, Korea

† Electronic supplementary information (ESI) available: Experimental procedures and characterization data with the copies of ¹H-, ¹³C-NMR spectra, chiral HPLC chromatograms of all chiral compounds and X-ray crystallography data of (*S,S*)-**7j**. CCDC 1007235. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c4cc06395c

‡ These authors contributed equally.

Scheme 2 Synthesis of 4-substituted sulfamate imine-5-carboxylates **6**.

DKR and can be applied in the synthesis of stereochemically enriched, chiral cyclic sulfamate-5-carboxylate esters **7**.

The racemic cyclic imine-5-carboxylate esters **6**, used in this work, were prepared from α -hydroxy- β -keto ester (**B**) and sulfamoyl chloride by modification of a previously described procedure (Scheme 2).¹¹

Racemic 4-phenyl-5-methoxycarbonyl cyclic imine **6a** was selected as the model substrate in initial efforts aimed at the identification of the most suitable catalyst systems for the ATH reactions. Reactions of **6a** were performed using the known chiral transition metal catalysts **1a–e** (0.5 mol%) and employing $\text{HCO}_2\text{H}/\text{Et}_3\text{N}$ as the hydrogen source in EtOAc at rt (Table 1).

As the results given in Table 1 show, the efficiencies and stereochemical outcomes of ATH reactions of **6a** are strongly affected by both the nature of the transition metal and the diamine ligands. For example, reaction of this substrate using Ru-catalysts with diamine ligands bearing electron rich or electron deficient arylsulfonyl groups (**1c**,^{3c} **1d**,¹² **1e**^{4a}) proceed to very low conversions (Table 1, entries 3–5). However, ATH of **6a** using Ir-catalyst **1b**¹³ reaches completion in 0.5 h (conversion > 99%) but takes place with a low level enantioselectivity (30% ee). Finally, the results reveal that ATH-DKR of **6a** with Rh-catalyst (*R,R*)-**1a**,¹⁴ which possesses TsDPEN and pentamethylcyclopentadienyl ligands, for 0.5 h at rt produces (*S,S*)-**7a** in the highest conversion (> 99%) and level of stereoselectivity (> 25 : 1 dr, 98% ee).

The influence of solvent on the ATH reaction of **6a** catalyzed by (*R,R*)-**1a** was investigated. In most of the solvents tested (EtOAc, CH_2Cl_2 , $\text{Cl}(\text{CH}_2)_2\text{Cl}$, CHCl_3 , toluene, DMF, MeOH, THF, and 2-propanol), ATH of **6a** takes place completely to form (*S,S*)-**7a** with high levels of stereoselectivity (> 25 : 1 dr, 84–99% ee) (see, Table S2 in ESI†). For the purpose of experimental convenience and based on optimization of stereoselectivity, further ATH reactions were carried out in EtOAc as solvent.

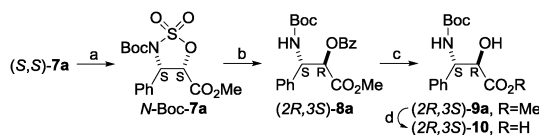
Table 1 Optimization of chiral catalysts **1a–e** for ATH-DKR of **6a**^a

Entry	Cat. 1	Conv. ^b (%)	dr (<i>syn</i> : <i>anti</i>)	ee ^d (%)	Config. ^e
1	(<i>R,R</i>)- 1a	> 99	> 25 : 1 ^c	98	<i>S,S</i>
2	(<i>R,R</i>)- 1b	> 99	> 25 : 1 ^c	30	<i>S,S</i>
3	(<i>R,R</i>)- 1c	13	—	95	<i>S,S</i>
4	(<i>R,R</i>)- 1d	6	—	—	—
5	(<i>R,R</i>)- 1e	17	—	83	<i>S,S</i>

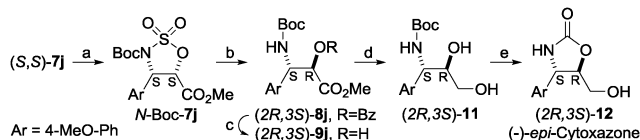
^a Reaction conditions: **6a** (0.5 mmol), cat-1 (0.5 mol%), $\text{HCO}_2\text{H}/\text{Et}_3\text{N}$ (5 : 2, 0.5 mL), EtOAc (5 mL), rt, 0.5 h. ^b Determined by ^1H NMR analysis of crude products. ^c Only 4,5-*cis* products were detected by using ^1H NMR analysis of crude product mixtures. ^d Determined by chiral HPLC. ^e See Scheme S1 in ESI.

Table 2 ATH-DKR of cyclic sulfamate imine-5-carboxylates **6**^a

<div> 6 $\xrightarrow[\text{EtOAc (0.1M), 25}^\circ\text{C}]{(R,R)\text{-1a (0.5 mol\%)} \text{ HCO}_2\text{H/Et}_3\text{N (5:2)}}$ (S,S)-7 + (R,R)-7</div>							
Entry	Substrate			Time (h)	Conv. ^b (%)	ee ^c (%)	Conf. ^d
	6, 7	R ₁	R ₂				
1	a		Me	0.5	> 99(92)	98	<i>S,S</i> ^e
2	b		i-Pr	0.5	> 99(85)	98	<i>S,S</i>
3	c		Bn	0.5	> 99(87)	98	<i>S,S</i>
4	d		<i>t</i> -Bu	0.5	> 99(87)	> 99	<i>S,S</i>
5 ^f	a		Me	0.5	> 99(94)	98	<i>R,R</i> ^f
6	e		Me	12	20	—	—
7 ^g	e		Me	12 ^g	> 99(95) ^g	92 ^g	—
8	f		Me	0.5	> 99(99)	99	<i>S,S</i>
9	g		Me	0.5	> 99(99)	99	<i>S,S</i>
10	h		Me	0.5	> 99(99)	97	<i>S,S</i>
11	i		Me	0.75	> 99(92)	97	<i>S,S</i>
12	j		Me	0.5	> 99(99)	> 99	<i>S,S</i> ^h
13 ⁱ	j		Me	3.5 ⁱ	> 99 ⁱ	99 ⁱ	<i>S,S</i> ^h
14	k		Me	0.5	> 99(96)	97	<i>S,S</i>
15	l		Me	0.5	> 99(88)	99	<i>S,S</i>
16	m		Me	0.5	> 99(95)	96	<i>S,S</i>
17	n		Me	0.5	> 99(92)	97	<i>S,S</i>
18	o		Me	0.5	> 99(91)	97	<i>S,S</i>
19	p		Me	1.0	> 99(92)	95	<i>S,S</i>
20	q		Me	2.0	> 99(94)	99	<i>S,S</i>
21	r	<i>n</i> -Pr-	Me	1.5	> 99(69)	91 ^j	ND ^k
22	s	Ph(CH ₂) ₂ -	Me	12	> 99(54)	76	ND ^k



Scheme 4 Reaction conditions: (a) (Boc)₂O, Et₃N, cat. DMAP, CH₂Cl₂, 100%; (b) (i) PhCO₂NH₄, DMF, 55 °C, 12 h; (ii) 1N HCl, CH₂Cl₂, 6 h, rt, 82%; (c) KCN, MeOH, 65 °C, 85%; (d) 1N NaOH, MeOH–THF, rt, 88%.



Scheme 5 Reaction conditions: (a) (Boc)₂O, Et₃N, cat. DMAP, CH₂Cl₂, 94%; (b) (i) PhCO₂NH₄, DMF, 55 °C, 12 h; (ii) 1N HCl, CH₂Cl₂, 6 h, rt, 100%; (c) KCN, MeOH, 65 °C, 86%; (d) NaBH₄, MeOH, rt, 92%; (e) NaH, THF, rt, 95%.

In order to demonstrate the utility of the methodology developed in this effort, we employed it in the synthesis of the taxotere side-chain 10¹⁷ (Scheme 4).

Accordingly, (S,S)-7a formed by ATH–DKR reaction of 6a is converted to its N-Boc derivative, which upon treatment with PhCO₂NH₄ undergoes ring opening^{10a,19} to form (2R,3S)-8a. Selective removal of the O-benzoyl group in 8a using KCN¹¹ in MeOH and subsequent hydrolysis of methyl ester 9a produce the taxotere side-chain 10¹⁷ (ca. 61% overall yield over 4 steps from (S,S)-7a).

An additional example demonstrating the usefulness of the methodology is found in the synthesis of (–)-epi-cytoxazone (12) starting with (S,S)-7j (Scheme 5)^{2b,18} (ca. 70% overall yield over 5 steps from (S,S)-7j).

In summary, a convenient and highly stereoselective method for the preparation of 4-substituted cyclic sulfamidate-5-carboxylate esters 7 was developed in this investigation. The process, involving asymmetric transfer hydrogenation accompanied by dynamic kinetic resolution (ATH–DKR), uses HCO₂H/Et₃N as the hydrogen source and chiral Rh catalysts (S,S)- or (R,R)-Cp*⁺RhCl(TsDPEN). Most of the ATH–DKR reactions probed in this study occur rapidly (30 min) and highly stereoselectively under mild and experimentally convenient conditions (rt, without the need for solvent degassing or an inert atmosphere). The utility of this methodology was demonstrated by its application to stereoselective syntheses of the taxotere side-chain and (–)-epi-cytoxazone.

This research was financially supported by grants from the National Research Foundation of Korea (2008-2004732) and Korea Research Institute of Chemical Technology (SI-1405).

Notes and references

- (a) M. Villacrez and P. Somfai, *Tetrahedron Lett.*, 2013, **54**, 5266; (b) O. K. Karjalainen and A. M. P. Koskinen, *Org. Biomol. Chem.*, 2012, **10**, 4311; (c) J. A. Bodkin and M. D. McLeod, *J. Chem. Soc., Perkin Trans. 1*, 2002, 2733; (d) S. C. Bergmeier, *Tetrahedron*, 2000, **56**, 2561; (e) T. J. Dobohoe, C. K. A. Callens, A. Flores, A. R. Lacy and A. H. Rath, *Chem. – Eur. J.*, 2011, **17**, 58.
- (a) Y. Zhao, N. Jiang, S. Chen, C. Peng, X. Zhang, Y. Zou, S. Zhang and J. Wang, *Tetrahedron*, 2005, **61**, 6546; (b) Y. Qian, X. Xu, L. Jiang, D. Prajapati and W. Hu, *J. Org. Chem.*, 2010, **75**, 7483; (c) S. Matsunaga, T. Yoshida, H. Morimoto, N. Kumagai and M. Shibasaki, *J. Am. Chem. Soc.*, 2004, **126**, 8777; (d) P. Dziedzic, P. Schyman, M. Kullberg and A. Córdova, *Chem. – Eur. J.*, 2009, **15**, 4044; (e) Y. Wang, Q.-F. He, H.-W. Wang, X. Zhou, Z.-Y. Huang and Y. Qin, *J. Org. Chem.*, 2006, **71**, 1588; (f) M. Bruncko, G. Schlingloff and K. B. Sharpless, *Angew. Chem., Int. Ed. Engl.*, 1997, **36**, 1483; (g) P. O'Brien, *Angew. Chem., Int. Ed.*, 1999, **38**, 326.
- For selected reviews, (a) T. Ikariya and A. J. Blacker, *Acc. Chem. Res.*, 2007, **40**, 1300; (b) T. Ikariya, K. Murata and R. Noyori, *Org. Biomol. Chem.*, 2006, **4**, 393; (c) R. Noyori and S. Hashiguchi, *Acc. Chem. Res.*, 1997, **30**, 97.
- (a) D. Cartigny, K. Puntener, T. Ayad, M. Scalone and V. Ratovelomanana-Vidal, *Org. Lett.*, 2010, **12**, 3788; (b) B. Mohar, A. Valleix, J.-R. Desmurs, M. Felemez, A. Wagner and C. Mioskowski, *Chem. Commun.*, 2001, 2572; (c) B. Seashore-Ludlow, P. Vilho, C. Hacker and P. Somfai, *Org. Lett.*, 2010, **12**, 5274; (d) A. Ros, A. Magriz, H. g. Dietrich, J. M. Lassaletta and R. Fernandez, *Tetrahedron*, 2007, **63**, 7532.
- (a) J. Limanto, S. W. Krska, B. T. Dorner, E. Vazquez, N. Yoshikawa and L. Tan, *Org. Lett.*, 2010, **12**, 512; (b) S.-M. Son and H.-K. Lee, *J. Org. Chem.*, 2013, **78**, 8396.
- S.-M. Son and H.-K. Lee, *J. Org. Chem.*, 2014, **79**, 2666.
- (a) T. Koike, K. Murata and T. Ikariya, *Org. Lett.*, 2000, **2**, 3833; (b) K. Murata, K. Okano, M. Miyagi, H. Iwane, R. Noyori and T. Ikariya, *Org. Lett.*, 1999, **1**, 1119.
- (a) K. M. Steward, M. T. Corbett, C. G. Goodman and J. S. Johnson, *J. Am. Chem. Soc.*, 2012, **134**, 20197; (b) C. G. Goodman, D. T. Do and J. S. Johnson, *Org. Lett.*, 2013, **15**, 2446.
- M. T. Corbett and J. S. Johnson, *J. Am. Chem. Soc.*, 2013, **135**, 594.
- (a) J. Han, S. Kang and H.-K. Lee, *Chem. Commun.*, 2011, **47**, 4004; (b) S. Kang, J. Han, E. S. Lee, E. B. Choi and H.-K. Lee, *Org. Lett.*, 2010, **12**, 4184; (c) C. Schüttler, Z. Li-Böhmer, K. Harms and P. von Zezschwitz, *Org. Lett.*, 2013, **15**, 800.
- H.-K. Lee, S. Kang and E. B. Choi, *J. Org. Chem.*, 2012, **77**, 5454.
- J. Limanto, S. W. Krska, B. T. Dorner, E. Vazquez, N. Yoshikawa and L. Tan, *Org. Lett.*, 2009, **12**, 512.
- K. Mashima, T. Abe and K. Tani, *Chem. Lett.*, 1998, 1199.
- J. Mao and D. C. Baker, *Org. Lett.*, 1999, **1**, 841.
- X. Zhou, X. Wu, B. Yang and J. Xiao, *J. Mol. Catal. A: Chem.*, 2012, **357**, 133.
- (a) R. E. Meledez and W. D. Lubell, *Tetrahedron*, 2003, **59**, 2581; (b) J. F. Bower, J. Rujirawanich and T. Gallagher, *Org. Biomol. Chem.*, 2010, **8**, 1505.
- X. Shen, J. Yang, H. Zhan, H. Wang, S. Wu and Z. Chen, *Chin. J. Chem.*, 2013, **31**, 31.
- (a) S. V. Narina, T. S. Kumar, S. George and A. Sudalai, *Tetrahedron Lett.*, 2007, **48**, 65; (b) I. S. Kim, J. D. Kim, C. B. Ryu, O. P. Zee and Y. H. Jung, *Tetrahedron*, 2006, **62**, 9349; (c) R. K. Mishra, C. M. Coates, K. D. Revell and E. Tuross, *Org. Lett.*, 2007, **9**, 575; (d) R. S. Reddy, P. V. Chouthaiwale, G. Suryavanshi, V. B. Chavan and A. Sudalai, *Chem. Commun.*, 2010, **46**, 5012; (e) S.-G. Kim and T.-H. Park, *Tetrahedron: Asymmetry*, 2008, **19**, 1626.
- H. Leisch, B. Sullivan, B. Fonovic, T. Dudding and T. Hudlicky, *Eur. J. Org. Chem.*, 2009, 2806.