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fibers of *Trichosanthes cucumerina* L.†

New triterpenoid saponin glycosides from the fruit

Five new triterpenoid saponin glycosides, trichocucumerisides A–E (1–5), together with eleven known compounds (6–16) were isolated from *Trichosanthes cucumerina* fruit fibers. The structures of the new compounds were elucidated by detailed analysis of NMR and mass spectroscopic data as well as chemical reactions. The anti-inflammatory study against nitric oxide (NO) production in lipopolysaccharide (LPS)-stimulated RAW264.7 cells shows that compounds 7 and 9 exhibited stronger NO inhibitory activity, with IC50 values of 3.0 and 2.7 μ M, respectively, with comparison to positive references Celecoxib and aminoguanidine (IC50 values 75.7 and 75.0 μ M, respectively). Compounds 7 and 9 also possessed a greater selectivity index (SI) of approximately 3–4-fold activity than that of the positive references.

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Introduction

Trichosanthes cucumerina L. is a Thai medicinal plant in the Cucurbitaceae family and the fruits are known for their strong bitter taste. Trichosanthes is mainly distributed in the tropical zone of Southeast Asia and Australia, and it is the largest genus of the family with over 100 species, 17 species of which are found in Thailand.1 Previous reports revealed its pharmacological effects including cytotoxic,2 anti-inflammatory,3 antidiabetic, 4,5 hypoglycemic, 6 hepatoprotective, 7 gastroprotective, 8 antifertility9 and larvicidal10 activities. The main chemical constituents present in T. cucumerina are triterpenoids, especially cucurbitacin groups,2 which make the plant pharmacologically and therapeutically important. Studies on the cytotoxic activities of cucurbitacins from T. cucumerina have been reported. 11-13 However, the phytochemical investigation of the constituents of this plant has not much been undertaken. In the present study, we wish to report five new triterpenoid saponin glycosides 1-5 together with eleven known compounds 6-16 14-24 from the fruit fibers of T. cucumerina. These compounds were elucidated by analysis of varying spectroscopic evidence, as well as comparison with the literature data of known compounds. All these compounds were evaluated against nitric oxide (NO) production in lipopolysaccharide (LPS)-stimulated RAW264.7 cells.

Results and discussion

The crushed fruit fiber of T. cucumerina was extracted successively with n-hexane, EtOAc and MeOH. The extracts were repeatedly subjected to silica gel, Sephadex LH-20 and reversed phase RP C-18 column chromatography to afford five new triterpenoid saponin glycosides (1-5), two known steroids (6 and 13) and nine known cucurbitacins (7-12 and 14-16). The structures of the new compounds were elucidated by spectroscopic techniques and chemical means to be hederagenin-3-O- β -D-glucopyranosyl-28-O- β -D-xylopyranosyl-(1 → 6)-[β -D-xylopyranosyl- $(1 \rightarrow 4)$ - α -1-rhamnopyranosyl- $(1 \rightarrow 2)$]- β -D-glucopyranoside (1), bayogenin-3-O- β -D-glucopyranosyl-28-O- β -D-xylopyranosyl-(1 \rightarrow 6)- $[\beta-D-xylopyranosyl-(1 \rightarrow 4)-\alpha-L-rhamnopyranosyl-(1 \rightarrow 2)]-\beta-D-glu$ copyranoside (2), oleanolic acid-3-O-β-D-glucopyranosyl-28-O-β-D-xylopyranosyl- $(1 \rightarrow 6)$ - $[\beta$ -D-xylopyranosyl- $(1 \rightarrow 2)$ - β -D-xylopyrano-syl- $(1 \rightarrow 4)$ - α -L-rhamnopyranosyl- $(1 \rightarrow 2)$]- β -D-glucopyranoside (3), oleanolic acid-3-O-β-D-glucopyranosyl-28-O-β-D-xylopyranosyl- $(1 \rightarrow 4)$ - β -D-xylopyranosyl- $(1 \rightarrow 4)$ - α -L-rhamnopyranosyl- $(1 \rightarrow 2)$ - β -D-glucopyranosyl-(1 \rightarrow 3)- β -D-xylopyranoside (4), and gypsogenin-3-O- β -D-glucopyranosyl-28-O- β -D-xylopyranosyl- $(1 \rightarrow 4)$ - β -Dxylopyranosyl- $(1 \rightarrow 4)$ - α -L-rhamnopyranosyl- $(1 \rightarrow 2)$ - β -D-glucopyranosyl- $(1 \rightarrow 3)$ - β -D-xylopyranoside (5) (Fig. 1). The NMR and mass spectra of the new compounds are included in the ESI.† The known compounds were identified as α-spinasterol (6),14 cucurbitacin B (7),15-17 cucurbitacin E (8),17-19 isocucurbitacin B (9),20 isocucurbitacin D (10),21 cucurbitacin D (11),17,18,22 25-acetyl cucurbitacin F (12),²³ α-spinasterol-3-*O*-β-D-

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glucopyranoside (13),¹⁴ 2-*O*- β -D-glucopyranosyl-cucurbitacin B (14),²⁴ 2-*O*- β -D-glucopyranosyl-cucurbitacin D (15),²⁴ and 2-*epi*-*O*- β -D-glucopyranosyl-cucurbitacin B (16)²⁴ by spectroscopic data comparisons with those of the literature values (Fig. S1†).

Compound 1 was isolated as a white powder, $[\alpha]_D^{26}$ -17.2° (c 0.39 in MeOH). Its ESI-TOF-MS data showed a sodium adduct ion $[M + Na]^+$ at m/z 1229.5924 (calcd for $C_{58}H_{94}NaO_{26}$, 1229.5926), corresponding to the molecular formula of C₅₈H₉₄O₂₆. The IR spectrum indicated the presence of hydroxyl (3337 cm⁻¹), ester (1734 cm⁻¹) and olefinic (1639 cm⁻¹) groups. The ¹H NMR data (Table 1, the NMR spectra are presented in Fig. S19-S42†) showed six tertiary methyl groups at δ 0.91 (3H, s, H-24), 0.94 (3H, s, H-25), 1.11 (3H, s, H-26), 1.18 (3H, s, H-27), 0.80 (3H, s, H-29) and 0.90 (3H, s, H-30) and characteristic signals for olefinic group at δ 5.40 (1H, brt, I = 3.2 Hz, H-12). The primary alcoholic function placement at C-23 was consistent with the down-field shift of CH₂-23 at δ 3.74 and 4.29 (each 1H, d, J = 10.9 Hz). The HMBC correlations between these two protons with C-3 at $\delta_{\rm C}$ 82.8 confirmed the location of the alcoholic function at C-23. The secondary alcoholic function in the aglycone unit was at C-3 (δ 4.26, 1H, m, H-3). HMBC correlation between CH_3 -24 (δ_H 0.91) and C-3 ($\delta_{\rm C}$ 82.8), and between CH₂-23 and C-3 allowed the location of this alcoholic function at C-3. The presence of the ester group ($\nu_{\rm max}$ CO at 1727 cm⁻¹ and $\delta_{\rm C}$ CO at 177.1) was proven to be at C-28 by HMBC correlation between H-16 ($\delta_{\rm H}$ 2.07) and C-28 (Fig. S34-S38†). The aglycone moiety of 1 was therefore identified to be hederagenin (1a, Fig. S2†).25

The anomeric protons at δ 5.10 (1H, d, J = 7.2 Hz, H-1′), 6.14 (1H, d, J = 8.0 Hz, H-1″), 6.46 (1H, brs, H-1‴), 5.05 (1H, d, J = 7.5 Hz, H-1″″) and 4.87 (1H, d, J = 7.4 Hz, H-1″″′) suggested that 1 had five sugar units. The five anomeric proton signals at δ 5.10, 6.14, 6.46, 5.05 and 4.87 showed correlations with five anomeric carbon signals at δ 106.3 (C-1′), 95.1 (C-1″), 101.9 (C-1″″), 108.2 (C-1″″′) and 106.1 (C-1″″″), respectively, in the HMQC spectra (Fig. S30–S33†). The sugar moieties were initially deduced from the coupling constants of anomeric protons, followed by acid hydrolysis of 1 with TFA and comparison with standard sugars (see

Experimental section) and D-glucose, L-rhamnose and D-xylose, together with the aglycone hederagenin (1a, Fig. S2†).25 The magnitude of the coupling constants of the anomeric protons of the sugar units mentioned above were therefore concluded to exist as two β-D-glucose, one α-L-rhamnose and two β-D-xylose moieties. 26-28 In the HMBC spectra (Fig. S34-S38†) of 1, the correlations between signal at $\delta_{\rm H}$ 5.10 (Glc-1') and $\delta_{\rm C}$ 82.8 (C-3) was observed, which indicated that the glucose unit was attached at the 3-position of the aglycone. The NMR data did not show any connection of this glucose unit with other sugar moiety, suggesting that the sugar residue at the 3-position was a monoglucoside. The correlations of signals between $\delta_{\rm H}$ 6.16 (Glc-1") and $\delta_{\rm C}$ 177.1 (C-28), $\delta_{\rm H}$ 6.46 (Rha-1"') and $\delta_{\rm C}$ 76.7 (Glc-2"), $\delta_{\rm H}$ 5.05 (Xyl-1"") and $\delta_{\rm C}$ 86.0 (Rha-4"'), and $\delta_{\rm H}$ 4.87 (Xyl-1"''') and $\delta_{\rm C}$ 69.4 (Glc-6") characterized that the sequence of the sugar chain was β -D-Xyl- $(1 \rightarrow 6)$ - $[\beta$ -D-Xyl- $(1 \rightarrow 4)$ - α -L-Rha- $(1 \rightarrow 2)$]- β -D-Glc and that the glycosidic site was at C-28. The sugar moieties at C-3 and C-28 were further confirmed by alkaline hydrolysis of 1 with NaOH (see Experimental section) to give hederagenin-3-O-β-D-glucopyranoside (1b, Fig. S2†),29 the structure of which was confirmed by NMR data (Tables S1 and S2†). The β-orientation of the C-3 hydroxyl group in 1b was confirmed by the large coupling constant (11.9 Hz) of H-3 at δ 4.32 with H-2ax (Table S1†). The presence of a mono glucose unit at the 3-position of 1 was therefore confirmed. In addition, ¹H-¹H COSY correlations H-1'/H-2', H-2'/H-3', H-5'/H-4', H-5'/H-6', H-1"/H-2", H-2"/H-3", H-5"/H-4", H-5"/H-6", H-1""/H-2"", H-2""/H-3"", H-3""/H-4"', H-4"'/H-5"', H-5"'/H-6"', H-1""/H-2"", H-2""/H-3"", H-3""/H-4"", H-4""/H-5"", H-1"""/H-2""", H-2"""/H-3""", H-4"""/H-5""" were observed (Fig. 2). Moreover, the NOESY correlations of H-1'/ H-3, H-1"'/H-2", H-1""/H-4"", and H-1""'/H-6" (Fig. 2) were also key interactions to support this gross structure. Accordingly, compound 1 was identified as hederagenin-3-O-β-D-glucopyranosyl-28-O- β -D-xylopyranosyl- $(1 \rightarrow 6)$ - $[\beta$ -D-xylopyranosyl- $(1 \rightarrow 4)$ - α -Lrhamnopyranosyl- $(1\rightarrow 2)$]-β-D-glucopyranoside and was named trichocucumeriside A.

Compound 2 was isolated as a white powder, $[\alpha]_D^{26}$ –22.7° (*c* 0.24 in MeOH). Its ESI-TOF-MS data showed a sodium adduct ion

Fig. 1 Structures of compounds 1-5.

Table 1 1 H NMR data for aglycone moieties of compounds 1–5 in pyridine- d_5 at 400 MHz (δ in ppm, J in Hz)

Position	1	2	3	4	5
1	0.94^{a}	1.30 m	0.91^{a}	0.85 m	0.88 m
	1.45 m	2.32 dd (12.7, 4.1)	1.44 m	1.41 ^a	1.40^{a}
2	1.89 ^a	4.17 m	1.84 m	1.83 ^a	1.84 m
	2.23 m		2.24 m	2.43 dd (11.8, 3.2)	2.17 m
3	4.26 m	4.25 m	3.39 dd (11.3, 4.6)	3.31 dd (11.8, 3.9)	4.04 m
5	1.66 m	1.84^{a}	0.84 m	0.74 brd (11.4)	1.66 dd (9.2, 8.1
6	1.45 m	1.49 ^a	1.44 m	1.31 ^a	1.06^{a}
	1.89 ^a	1.92^{a}	1.68 m	1.51 m	1.40^{a}
7	1.76 m	1.77 ^a	1.82 ^a	1.41 m	1.25^{a}
	2.02 m	2.06 m	2.08 m	1.47 m	1.43^{a}
9	1.75 ^a	1.84 ^a	1.67 m	1.60 t (8.1)	1.36 m
11	1.89 ^a	1.94^{a}	1.97 m	1.87 m	1.84 m
	1.94 m	2.08 m			
12	5.40 brt (3.2)	5.40 brt (3.2)	5.46 brs	5.46 brt (3.2)	5.44 brs
15	1.46 m	1.49 m	1.57 m	1.27^{a}	1.22^{a}
	2.05 m	2.08^{a}	2.12 m	2.11 m	2.07 m
16	1.92^{a}	1.96^{a}	2.00 m	2.03 m	2.02 m
	2.05 m	2.04^{a}	2.16 m	2.11 m	2.10^{a}
18	3.13 dd (14.0, 4.3)	3.14 dd (13.9, 3.6)	3.18 brd (12.1)	3.31 dd (13.7, 3.9)	3.28 brd (11.6)
19	1.14 m	1.19 ^a	1.25 m	1.27^{a}	1.22^{a}
	1.68 m	1.68 brd (13.6)	1.80 ^a	1.81 ^a	1.78 m
21	1.11 ^a	1.14^{a}	1.19 m	1.18 brd (12.7)	1.75^{a}
	1.26 brd (12.0)	1.30 m	1.35 ^a	1.41 ^a	2.01 m
22	1.75 ^a	1.49 m	1.44^{a}	1.75 m	1.17 m
		1.77 ^a	1.75 m	2.11 m	1.40^{a}
23	3.74 d (10.9)	3.76 d (10.6)	0.99 s	1.12 s	9.95 s
	4.29 d (10.9)	4.51 d (10.6)			
24	0.91 s	0.98 s	1.32 s	1.34 s	1.43 s
25	0.94 s	1.06 s	0.91 s	0.82 s	0.79 s
26	1.11 s	1.14 s	1.20 s	1.07 s	1.02 s
27	1.18 s	1.16 s	1.32 s	1.29 s	1.25 s
29	0.80 s	0.83 s	0.89 s	0.94 s	0.94 s
30	0.90 s	0.93 s	0.96 s	1.01 s	1.00 s

^a Signals overlapped.

 $[M + Na]^+$ at m/z 1245.5869 (calcd for $C_{58}H_{94}NaO_{27}$, 1245.5875), corresponding to the molecular formula of C₅₈H₉₄O₂₇. The IR spectrum indicated the presence of hydroxyl (3442 cm⁻¹), ester (1727 cm⁻¹) and olefinic (1643 cm⁻¹) groups. The ¹H NMR data (Table 1 and Fig. S44-S66†) showed very similar spectral feature to that of compound 1. Comparison of the molecular formula of compound 2 with that of compound 1 indicated that 2 constituted one oxygen atom more than 1, suggesting the presence of one additional hydroxyl group in the aglycone unit. The large downfield shifts at C-2 signals of 2 ($\delta_{\rm H}$ 4.17 and $\delta_{\rm C}$ 67.6) comparing with those of 1 ($\delta_{\rm H}$ 1.89, 2.23 and $\delta_{\rm C}$ 26.4) were observed. Considerable down-field shifts of proton and carbon resonances at C-1 and C-3 of 2 (Tables 1 and 3) were also noted. These observations indicated that the hydroxyl group should be located at the 2-position. This was confirmed by the ¹H-¹H COSY correlation of H-2 and H-3, and HMBC correlation between H-2 and C-3.

The orientation of H-2 was difficult to determine owing to the obscured nature of its 1 H signal. However, bayogenin-3-O- β -D-glucopyranoside (**2b**, Fig. S2 \dagger), 30 the alkaline hydrolysis product of compound **2** (see Experimental section), exhibited well-resolved 1 H-NMR spectrum (Table S1 and Fig. S13 \dagger) and H-2 signal (δ 4.24, J = 11.6, 3.9 Hz) showed

relatively large coupling constant with the neighbouring H-1ax. It was therefore concluded that H-2 was in the β -orientation. From the existing NMR spectra (Fig. S5 and S6†) and comparison with those of glycosides with bayogenin as the aglycone, ^{31,32} it could therefore be concluded that the aglycone moiety was 2β , 3β , 23-trihydroxyolean-12-en-28-oic acid (bayogenin) (2a).

For the sugar moiety of compound **2**, the ¹H NMR data were closely resembled those of compound **1** (Table 2) and the ¹³C NMR data of the sugar moiety of **2** (Table 4) were almost identical to those of compound **1** (Table 4). Acid hydrolysis of **2** also produced the same sugars, p-glucose, L-rhamnose and p-xylose, together with the aglycone which was identified as bayogenin (**2a**, Fig. S2†). ^{31,32} HMBC correlations of the proton of one sugar unit to the carbon of the next sugar unit were in the same manner to those occurred in the sugar unit of compound **1**. The sugar linkages of compounds **1** and **2** were therefore identical. Alkaline hydrolysis of **2** (see Experimental section) gave bayogenin-3-*O*-β-p-glucopyranoside (**2b**, Fig. S2†), ³⁰ the structure of which was confirmed by NMR data (Tables S1 and S2†). In addition, the ¹H-¹H COSY correlation of the sugar units (Fig. 2 and S51–S53†) and the NOESY correlations of the sugar

Fig. 2 Key correlations of compounds 1, 3 and 4.

and the aglycone units (Fig. 2 and S64–S66†) confirmed the structure of **2**. Compound **2** was therefore identified as bayogenin-3-O- β -D-glucopyranosyl-28-O- β -D-xylopyranosyl- $(1 \rightarrow 4)$ - α -L-rhamnopyranosyl- $(1 \rightarrow 2)$]- β -D-glucopyranoside and was named trichocucumeriside B.

Compound 3 was isolated as a white powder with $[\alpha]_D^{26}$ –39.6° (c 0.20 in MeOH). The ESI-TOF-MS gave a sodiated molecular ion at m/z 1345.6331 [M + Na]⁺ compatible with the molecular formula C₆₃H₁₀₂O₂₉. The ¹H and ¹³C NMR data (Tables 1-4 and Fig. S68-S90†) revealed that the aglycone moiety of 3 was similar to that of 1. The significant difference was the absence of an oxymethylene group at C-23 and this was replaced by a methyl group at $\delta_{\rm H}$ 0.99 and $\delta_{\rm C}$ 17.6. The aglycone moiety of 3 was suggested to be 3 β hydroxyolean-12-en-28-oic acid (oleanolic acid, 3a) from its NMR spectra (Fig. S7 and S8†) and by comparison with the NMR data of the glycosides of oleanolic acid.33 Acid hydrolysis of 3 (see Experimental section) gave D-glucose, L-rhamnose and D-xylose, together with the aglycone oleanolic acid (3a). The NMR data (Tables 2 and 4) of 3 suggested that the sugar chain at C-28 consisted of three β -Dxylose, one β -D-glucose and one α -L-rhamnose and the sugar unit at C-3 was determined as one β-p-glucose in the molecule. The assignments of the sugar moieties at both positions were further confirmed by alkaline hydrolysis (see Experimental section) which yielded oleanolic acid-3-O-β-D-glucopyranoside (3b, Fig. S2 \dagger),³⁴ the NMR spectral data of which are presented in Fig. S15, S16, Tables S1 and S2.† The anomeric protons appeared at $\delta_{\rm H}$ 4.94 (1H, d, J=7.8 Hz, H-1'), 6.16 (1H, d, J = 8.0 Hz, H-1"), 6.47 (1H, brs, H-1""), 5.08 (1H, d, J = 7.0 Hz, H-1''''), 4.90 (1H, d, J = 7.4 Hz, H-1''''') and 5.23 (1H, d, J = 7.5 Hz, H-1'''''). These six anomeric proton signals

showed correlations with six anomeric carbon signals at $\delta_{\rm C}$ 107.4 (C-1'), 95.2 (C-1"), 101.9 (C-1"), 108.2 (C-1""), 106.1 (C-1"") and 106.4 (C-1""), respectively, in the HMQC spectrum (Fig. S78-S81†). These observations suggested that 3 contained six sugar units, with one more sugar unit than that of compound 1. Comparison of the molecular formula of 3 with that of 1 also suggested that the extra sugar moiety was compatible with a xylose unit deduced from the NMR data. The HMBC experiments (Fig. 2) showed correlations between $\delta_{\rm H}$ 4.94 (Glc-1') and $\delta_{\rm C}$ 89.6 (C-3), $\delta_{\rm H}$ 6.16 (Glc-1") and $\delta_{\rm C}$ 177.3 (C-28), $\delta_{\rm H}$ 6.47 (Rha-1"") and $\delta_{\rm C}$ 76.9 (Glc-2"), $\delta_{\rm H}$ 5.08 (Xyl-1"") and $\delta_{\rm C}$ 85.9 (Rha-4""), and between $\delta_{\rm H}$ 4.90 (Xyl-1'''') and $\delta_{\rm C}$ 69.5 (Glc-6"). The sequence of connection of the sugar units was therefore the same as those of compounds 1 and 2. The additional sixth sugar moiety, the xylose unit, was determined to reside at the lower xylose unit from the HMBC correlation between $\delta_{\rm H}$ 5.23 (Xyl-1''''') and $\delta_{\rm C}$ 87.8 (Xyl-2''''). These data have been used to establish the positions of the sugar units in the chain. In addition, ¹H-¹H COSY correlations of the sugar protons were observed (Fig. 2). Furthermore, key NOESY correlations of the sugar moieties (Fig. 2) were also observed. These correlations were key interactions to support the gross structure of 3. Compound 3 was therefore identified as oleanolic acid-3-O-β-D-glucopyranosyl-28-O- β -D-xylopyranosyl-(1 → 6)-[β -D-xylopyranosyl-(1 → 2)- β -D-xylopyranosyl- $(1 \rightarrow 4)$ - α -L-rhamnopyranosyl- $(1 \rightarrow 2)$]- β -D-glucopyranoside and was named trichocucumeriside C.

Compound 4 was isolated as a white powder with $[\alpha]_D^{26} - 31.1^\circ$ (c 0.38 in MeOH). The ESI-TOF-MS gave a sodiated molecular ion at m/z 1345.6340 [M + Na]⁺ which was compatible with the molecular formula $C_{63}H_{102}O_{29}$. The ¹H and ¹³C NMR data (Tables 1–4 and

Fig. S92-S115†) showed that the aglycone of 4 was almost identical to 3. Acid hydrolysis of 4 (see Experimental section) yielded pxylose, p-glucose, L-rhamnose and the triterpenoid oleanolic acid

(3a).33 Alkaline hydrolysis of 4 (see Experimental section) gave oleanolic acid-3-O-β-D-glucopyranoside (3b, Fig. S2†)34 which was identical to that obtained from alkaline hydrolysis of compound 3.

	1	2	3	4	5
Position	Glc'	Glc'	Glc'	Glc'	Glc'
1'	5.10 d (7.2)	5.20^{a}	4.94 d (7.8)	4.93 d (7.6)	4.83 d (7.6)
2'	4.04 t (8.3)	4.11 m	4.06 m	4.20 d (8.2)	4.12 m
- 3'	4.18 d (8.8)	4.13 t (8.4)	4.31 m	4.34 t (8.8)	4.27 t (8.8)
4'	4.26 m	4.27 t (10.1)	4.27 m	4.07 m	4.06 m
5'	3.90 ddd (9.0, 4.6, 2.5)	3.93 ddd (9.4, 5.0, 2.2)	4.02 t (7.7)	3.95 ddd (9.5, 5.0, 1.9)	3.87 m
6'	4.40 brd (10.2)	4.32 dd (10.2, 4.9)	4.43 dd (11.8, 5.0)	4.37 dd (8.8, 4.5)	4.34 dd (8.8, 3.8)
Ü	4.51 m	4.52 dd (9.5, 6.5)	4.62 brd (11.0)	4.57 m	4.55 m
^a Signals ove	rlapped.				
	1	2	3	4	5
Position	Glc"	Glc"	Glc"	Xyl"	Xyl"
1"	6.14 d (8.0)	6.16 d (7.8)	6.16 d (8.0)	6.54 d (7.7)	5.24 d (7.1)
2"	4.29 d (8.3)	4.32 m	4.32 m	4.60 t (9.6)	4.59 m
3"	4.21 m	4.24 m	4.26 t (8.9)	4.23 m	4.15 m
4''	4.29 m	4.11 m	4.36 t (8.7)	4.08 m	4.07 t (8.1)
5"	4.04 t (8.3)	4.05 m	4.07 m	4.00 dd (10.4, 4.0) 4.39 m	3.99 dd (7.0, 3.5) 4.45 m
6"	4.28 m 4.64 dd (9.6, 1.6)	4.30 m 4.67 dd (11.3, 1.6)	4.31 m 4.66 brd (10.9)		
	1	2	3	4	5
Position	Rha‴	Rha‴	Rha‴	Glc'''	Glc‴
1‴	6.46 brs	6.47 brs	6.47 brs	5.24 d (7.6)	5.18 d (7.2)
2""	4.82 brs	4.84 brs	4.84 m	4.64 t (9.2)	4.59 m
3‴	4.70 brd (8.8)	4.72 brd (9.4)	4.72 m	4.74 brs	4.61 m
4'''	4.37 t (9.4)	4.39 t (9.3)	4.49 dd (9.2, 3.4)	4.19 d (8.6)	4.17 m
5‴	4.50 t (6.4)	4.52 t (9.5)	4.53 m	4.07 t (6.6)	4.06 m
6′′′	1.82 d (6.1)	1.84 d (6.1)	1.81 d (5.9)	4.64 m	4.17 m
	1102 & (011)	1101 ti (011)	1.01 & (0.5)	4.42 m	4.56 m
	1	2	3	4	5
Position	Xyl''''	Xyl''''	Xyl''''	Rha''''	Rha''''
1''''	5.05 d (7.5)	5.08 d (7.2)	5.08 d (7.0)	5.78 brs	5.82 brs
2''''	4.09 m	4.08 m	4.09 m	4.60 m	4.61 m
3''''	4.09 m	4.24 m	4.09 m	4.45 m	4.45 m
4''''	4.46 m	4.32 m	4.36 t (8.7)	4.41 m	4.41 m
5''''	3.52 t (10.3) 4.26 m	3.55 t (10.2)	3.57 t (10.8)	4.41 m	4.40 m
6''''	4.20 111	4.27 t (10.1)	4.31 t (9.7)	1.79 d (5.0)	1.76 d (4.7)
	1	2	3	4	5
Position	Xyl''''	Xyl''''	Xyl''''	Xyl'''''	Xyl''''
1''''	4.87 d (7.4)	4.90 d (7.4)	4.90 d (7.4)	5.19^{a}	5.15 d (7.5)
2''''	3.97 t (7.9)	3.99 t (7.8)	4.00 t (8.1)	4.07 m	4.07 m
3''''	4.10 m	4.11 m	4.17 t (8.7)	4.59 m	4.59 m
4''''	4.17 t (8.8)	4.19 m	4.21 t (8.2)	4.08 m	4.07 m
5'''''	3.63 t (10.6)	3.67 t (10.8)	3.66 t (10.6)	3.48 t (9.8)	3.51 t (10.1)
	4.29 t (10.8)	4.32 t (8.2)	4.31 m	4.23 t (9.4)	4.24 t (9.4)

Table 2 (Contd.)

			3	4	5
Position	1	2	Xyl'''''	Xyl'''''	Xyl'''''
1'''''			5.23 d (7.5)	5.28 d (7.6)	5.23^{a}
2'''''			4.13 t (8.4)	4.07 m	4.06 m
3'''''			4.22 m	4.19 m	4.17 m
4'''''			4.09 m	4.20 m	4.19 m
5'''''			3.53 t (10.0)	3.70 t (10.0)	3.70 t (10.0)
			4.31 t (9.7)	4.31 m	4.29 t (10.4)

The sugar unit at C-3 was therefore determined as one β-D-glucose and the sugar chain was located at C-28. Six anomeric proton signals at δ 4.93 (1H, d, J = 7.6 Hz, H-1′), 6.54 (1H, d, J = 7.7 Hz, H-1″), 5.24 (1H, d, J = 7.6 Hz, H-1‴), 5.78 (1H, brs, H-1′′′′), 5.19 (1H, overlapping signals, H-1′′′′′) and 5.28 (1H, d, J = 7.6 Hz, H-1′′′′′) showed correlations with six anomeric carbon signals at δ 105.5 (C-1′′), 93.8 (C-1″), 107.7 (C-1‴), 101.4 (C-1′′′′′), 106.9 (C-1′′′′′′) and 106.5 (C-1′′′′′′′′), respectively, in the HMQC spectrum (Fig. S102–S105†). In the HMBC spectrum (Fig. 2 and S106–S112†), the correlation between $\delta_{\rm H}$ 4.93 (Glc-1′) and $\delta_{\rm C}$ 89.4 (C-3), which was the mono sugar unit of β -D-glucose, was observed. The correlations between $\delta_{\rm H}$ 6.54 (Xyl-1″) and $\delta_{\rm C}$ 176.8 (C-28), $\delta_{\rm H}$ 5.24 (Glc-1″′) and $\delta_{\rm C}$ 84.9

Table 3 13 C NMR data for aglycone moieties of compounds 1–5 in pyridine- d_5 at 100 MHz (δ in ppm)

Position	1	2	3	4	5
1	39.3	47.9	39.4	39.2	38.6
2	26.4	67.6	27.2	27.1	25.4
3	82.8	89.1	89.6	89.4	83.7
4	43.9	45.2	40.1	40.0	55.6
5	48.3	47.9	56.5	56.3	48.7
6	18.9	18.9	19.2	18.9	21.0
7	32.9	32.9	33.0	33.7	33.0
8	40.5	40.5	40.5	40.3	40.6
9	48.7	48.6	48.6	48.4	48.2
10	37.5	38.6	37.6	37.4	36.7
11	24.4	23.8	24.4	24.2	24.2
12	123.3	123.2	123.5	123.5	123.1
13	144.5	144.5	144.6	144.7	144.8
14	42.7	42.8	42.9	42.6	42.6
15	29.1	29.1	29.0	28.7	28.7
16	23.8	24.4	24.0	23.6	23.6
17	47.7	47.7	47.8	47.8	47.8
18	42.4	42.3	42.5	42.2	42.2
19	46.8	46.8	47.0	46.7	46.7
20	31.2	31.2	31.3	31.4	31.4
21	34.4	34.4	34.5	34.6	33.2
22	33.6	33.4	33.7	33.2	34.6
23	65.2	64.4	17.6	17.3	210.2
24	14.2	15.3	28.8	28.7	11.4
25	16.8	18.0	16.2	16.0	16.2
26	18.1	18.1	18.1	18.0	17.9
27	26.5	26.4	26.6	26.5	26.5
28	177.1	177.1	177.3	176.8	176.8
29	33.6	33.6	33.7	33.7	33.7
30	24.4	24.4	24.4	24.2	24.2

(Xyl-3"), $\delta_{\rm H}$ 5.78 (Rha-1"") and $\delta_{\rm C}$ 75.3 (Glc-2""), $\delta_{\rm H}$ 5.19 (Xyl-1""") and $\delta_{\rm C}$ 84.3 (Rha-4""), $\delta_{\rm H}$ 5.28 (Xyl-1""") and $\delta_{\rm C}$ 87.5 (Xly-4""") characterized that the sequence of the sugar chain was β -D-Xyl-(1 \rightarrow 4)- β -D-Xyl-(1 \rightarrow 4)- α -L-Rha-(1 \rightarrow 2)- β -D-Glc-(1 \rightarrow 3)- β -D-Xyl and that the glycosidic site was at C-28. In addition, 1 H- 1 H COSY correlations of the sugar protons were observed (Fig. 2) and the NOESY correlations (Fig. 2) were also key interactions to support this gross structure. Compound 3 was therefore assigned as oleanolic acid-3-O- β -D-glucopyranosyl-28-O- β -D-xylopyranosyl-(1 \rightarrow 4)- α -L-rhamnopyranosyl-(1 \rightarrow 2)- β -D-glucopyranosyl-(1 \rightarrow 3)- β -D-xylopyranoside and was named trichocucumeriside D.

Compound 5 was isolated as a white powder, $\left[\alpha\right]_{D}^{26}$ -11.6° (c 0.30 in MeOH), with the molecular formula of C₆₃H₁₀₀O₃₀ obtained by the ESI-TOF-MS from the $[M + Na]^+$ ion at m/z 1359.6145 (calcd for $C_{63}H_{100}NaO_{30}$, 1359.6192). The ¹³C and ¹H-NMR spectra (Tables 1-4) of the aglycone moiety of 5 was similar to that of 4. The major difference was that the chemical shifts of 23-CH₃ at $\delta_{\rm H}$ 1.12 and $\delta_{\rm C}$ 17.3 of 4 were replaced by the CHO group at $\delta_{\rm H}$ 9.95 (1H, s, H-23) and $\delta_{\rm C}$ 210.2 in the ¹H and ¹³C-NMR spectrum, respectively, indicated an aldehydic group in the aglycone moiety of 5 at C-23. The aglycone moiety of 5 was therefore suggested to be gypsogenin (4a, Fig. S2†)35 by comparison of its spectroscopic data with those reported in the literature. Acid hydrolysis of 5 yielded Dxylose, D-glucose, L-rhamnose and the triterpenoid 4a³⁵ (Fig. S2†). Alkaline hydrolysis gave the glycoside gypsogenin-3-O-β-D-glucopyranoside (4b)³⁶ (Fig. S2, Tables S1 and S2†). The ¹H-¹H COSY (Fig. 2 and S126-S129†), HMBC (Fig. 2 and S134-S141†) and HMQC data (Fig. S130-S133†) confirmed the same sugar moiety of β-D-Glc at C-3. Analysis of the HMBC correlations (Fig. 2 and S134-S141†) supported by the COSY (Fig. 2 and S126-S129†) and NOESY (Fig. 2 and S142–S144†) correlations revealed the sugar linkage at C-28 as β -D-Xyl- $(1 \rightarrow 4)$ - β -D-Xyl- $(1 \rightarrow 4)$ - α -L-Rha- $(1 \rightarrow 2)$ - β -D-Glc- $(1 \rightarrow$ 3)-β-D-Xyl, which is the same as that of compound 4. All of the above data led to the formulation of 5 as gypsogenin-3-O-β-D-glucopyranosyl-28-O- β -D-xylopyranosyl-(1 \rightarrow 4)- β -D-xylopyranosyl-(1 \rightarrow 4)-α-L-rhamnopyranosyl- $(1 \rightarrow 2)$ -β-D-glucopyranosyl- $(1 \rightarrow 3)$ -β-D-xylopyranoside and was named trichocucumeriside E.

The anti-inflammatory activity of all the isolated compounds was evaluated against nitric oxide (NO) production in LPS-stimulated RAW264.7 cells using the Griess assay³⁷ (Table 5). Compared to the positive references, Celecoxib and aminoguanidine (IC₅₀ 75.7 and 75.0 μ M, respectively), cucurbitacin B (7) and isocucurbitacin B (9) showed more potent inhibitory

Table 4 13 C NMR data for sugar moieties of compounds 1–5 in pyridine- d_5 at 100 MHz (δ in ppm)

Position	1	2	3	4	5
	Glc'	Glc'	Glc'	Glc'	Glc'
1′	106.3	106.3	107.4	105.5	103.4
2'	76.4	76.0	76.4	72.0	71.7
3′	79.3	79.1	79.3	78.9	78.8
4'	72.1	71.8	72.3	75.8	75.5
5′	78.6	78.9	78.9	78.5	78.8
6′	63.2	62.8	63.5	63.3	63.1
	1	2	3	4	5
Position	Glc"	Glc"	Glc"	Xyl"	Xyl"
1"	95.1	95.1	95.2	93.8	93.9
2"	76.7	76.7	76.9	70.0	76.6
3"	79.9	79.9	79.9	84.9	83.3
4"	71.4	76.7	71.2	69.5	69.5
5"	78.1	78.1	78.1	63.3	63.4
6"	69.4	69.5	69.5	00.0	00.1
	1	2	3	4	5
Position	Rha‴	Rha‴	Rha‴	Glc'''	Glc'''
1′′′	101.9	101.9	101.9	107.7	106.8
2′′′	72.3	72.3	72.1	75.3	75.4
3‴	73.1	73.1	73.1	70.0	75.0
4'''	86.0	86.0	85.9	75.5	75.4
5‴	68.8	68.8	68.8	75.5	75.6
6′′′	19.2	19.2	19.2	61.7	62.6
	1	2	3	4	5
Position	Xyl''''	Xyl''''	Xyl''''	Rha''''	Rha''''
1''''	108.2	108.2	108.2	101.4	101.5
2''''	79.2	79.3	87.8	73.2	73.2
3''''	76.7	76.7	75.8	66.4	66.7
4''''	71.6	71.4	71.4	84.3	84.4
5''''	68.0	68.0	67.9	69.0	69.0
6''''				18.9	18.9
	1	2	3	4	5
Position	Xyl''''	Xyl''''	Xyl''''	Xyl''''	Xyl''''
		106.1	106.1	106.9	107.6
	106.1	106.1		100.9	107.0
2''''	106.1 75.3	75.3	75.4	77.4	76.6
2''''	75.3	75.3	75.4	77.4	76.6
2''''' 3'''''	75.3 78.9	75.3 78.6	75.4 78.6	77.4 72.3	76.6 72.3
2'''' 3'''' 4'''''	75.3 78.9 71.4	75.3 78.6 71.6	75.4 78.6 71.6	77.4 72.3 87.5	76.6 72.3 87.6
3''''' 4'''''	75.3 78.9 71.4	75.3 78.6 71.6	75.4 78.6 71.6 67.6	77.4 72.3 87.5 67.4	76.6 72.3 87.6 67.4
2""" 3""" 4""" 5""" Position	75.3 78.9 71.4 67.6	75.3 78.6 71.6 67.6	75.4 78.6 71.6 67.6 3 Xyl'''''	77.4 72.3 87.5 67.4 4 Xyl'''''	76.6 72.3 87.6 67.4 5 Xyl'''''
2""" 3""" 4""" 5""" Position	75.3 78.9 71.4 67.6	75.3 78.6 71.6 67.6	75.4 78.6 71.6 67.6 3 Xyl'''''	77.4 72.3 87.5 67.4 4 Xyl''''' 106.5	76.6 72.3 87.6 67.4 5 Xyl'''''
2""" 3""" 4""" 5""" Position 1""" 2"""	75.3 78.9 71.4 67.6	75.3 78.6 71.6 67.6	75.4 78.6 71.6 67.6 3 Xyl''''' 106.4 76.6	77.4 72.3 87.5 67.4 4 Xyl''''' 106.5 75.6	76.6 72.3 87.6 67.4 5 Xyl''''' 106.5 75.8
2'''' 3''''' 4''''' 5'''''	75.3 78.9 71.4 67.6	75.3 78.6 71.6 67.6	75.4 78.6 71.6 67.6 3 Xyl'''''	77.4 72.3 87.5 67.4 4 Xyl''''' 106.5	76.6 72.3 87.6 67.4 5 Xyl'''''

Table 5 Anti-inflammatory activity of compounds 1–16

	$(IC_{50}, \mu M)$		
Compound	NO^a	MTT^b	SI^c
1	97.7 ± 12.4	>160	>1.6
2	99.0 ± 15.6	>160	>1.6
3	115.7 ± 16.9	>160	>1.4
4	29.2 ± 5.3	49.0 ± 6.4	1.7
5	22.5 ± 3.8	52.0 ± 3.8	2.3
7	3.0 ± 1.1	19.4 ± 1.6	6.5
9	2.7 ± 0.9	17.2 ± 4.5	6.4
10	13.3 ± 2.3	41.7 ± 9.4	3.1
11	21.7 ± 8.1	67.0 ± 16.4	3.1
12	11.3 ± 3.3	19.0 ± 2.3	1.7
13	$\textbf{71.7} \pm \textbf{3.0}$	>160	>2.2
14	148.3 ± 6.4	141.7 ± 9.3	1.0
15	153.0 ± 6.5	>160	>1.1
16	110.0 ± 9.0	>160	>1.5
Celecoxib ^d	$\textbf{75.7} \pm \textbf{12.8}$	$\textbf{102.7} \pm \textbf{8.8}$	1.4
Aminoguanidine ^d	$\textbf{75.0} \pm \textbf{15.5}$	>160	>2.1

^a Nitric oxide (NO) production in LPS-stimulated RAW264.7 cells. ^b Cell viability in LPS-stimulated RAW264.7 cells. ^c Selectivity index. ^d Positive references. Mean \pm SD, n=3.

activity against NO production in LPS-stimulated RAW264.7 cells with IC_{50} values of 3.0 and 2.7 μ M, respectively. More importantly, both natural compounds showed approximately 3–4-fold higher selectivity index (SI) than that of positive references. Compounds 4, 5 and 10–12 exhibited moderate inhibitory activity towards NO production with IC_{50} range from 11.3 to 29.2 μ M, while compounds 1–3 and 13–16 exhibited weaker activity with the IC_{50} values of 71.7–153.0 μ M.

Conclusions

In summary, five new triterpenoid saponin glycosides (1–5) have been isolated from *T. cucumerina* fruit fibers, together with eleven known compounds (6–16) in this work. The anti-inflammatory assays showed that the natural compounds 7 and 9 exhibited more potent NO inhibitory activities than those of positive references Celecoxib and aminoguanidine. In addition, these two compounds possessed approximately 3–4-fold higher selectivity index (SI) than the positive references.

Experimental

General

Melting points were determined with an Electrothermal melting point apparatus and are uncorrected. Optical rotations were acquired using a JASCO P-1020 digital polarimeter. IR spectra were recorded on a PerkinElmer FT-IR 400 spectrophotometer. Spectra of solid sample were recorded in the ATR mode. NMR spectra were recorded on a Bruker AVANCE 400 FT-NMR spectrometer, operating at 400 (1 H) and 100 (13 C) MHz. For the spectra taken in CDCl₃ ($\delta_{\rm H}$ 7.24 and $\delta_{\rm C}$ 77.0) and pyridine- $d_{\rm 5}$ ($\delta_{\rm H}$ 8.74 and $\delta_{\rm C}$ 150.3), the residual nondeuterated solvent were used as references for 1 H and 13 C NMR spectra. The electrospray ionization mass spectra (ESIMS) were measured with a Finnigan

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LC-Q mass spectrometer. High resolution electrospray ionization time-of-flight mass spectra (ESI-TOF-MS) were measured with a Bruker micrOTOF-II mass spectrometer. Column chromatography was carried out using Merck silica gel 60 (<0.063 mm) and reversed phase RP C-18 (40–63 μ m, E. Merck). For TLC, Merck precoated silica gel 60 F₂₅₄ plates and RP-TLC, RP-18 F₂₅₄ precoated on aluminium plate (E. Merck) were used. Spots on TLC were detected under UV light and by spraying with anisaldehyde–H₂SO₄ reagent followed by heating. The standard sugars, D(+)-glucose, L(+)-rhamnose and D(+)-xylose, were purchased from Sigma, St. Louis, USA.

Plant material

The dried fruit fibers of *T. cucumerina* used in this study were collected from Takli district, Nakhon Sawan province, Thailand in 2009 and were identified by Assoc. Prof. Nopporn Damrongsiri, Department of Biology, Faculty of Science, Ramkhamhaeng University. A voucher specimen (Apichart Suksamrarn, No. 058) was deposited at the Faculty of Science, Ramkhamhaeng University, Bangkok, Thailand.

Extraction and isolation

The dried fruit fibers of T. cucumerina (18.0 kg), whose fruit peels and seeds were separated, were chilled in liquid N2 and crushed into small pieces and extracted successively with nhexane (3 \times 50 L), EtOAc (3 \times 50 L) and MeOH (3 \times 50 L). The extracts were evaporated to dryness under reduced pressure at temperature 40-45 °C to afford the hexane (168.0 g), EtOAc (350.0 g) and MeOH (830.0 g) extracts, respectively. The hexane extract (25.0 g) was subjected to column chromatography (CC) (Merck silica gel 60, 0.063-0.200 mm, 300 g) using gradient elution of *n*-hexane: EtOAc (100:0 to 0:100, v/v) and EtOAc: MeOH (100: 0 to 0: 100, v/v) to give 3 fractions (H1 to H3). H1 (8.5 g) was repeatedly CC over silica gel eluted with nhexane: EtOAc (100:10, v/v) to afford 4 subfractions (H1-1 to H1-4), then H1-3 (1.2 g) was purified by CC eluted with CH₂-Cl₂: MeOH (100: 0.4, v/v) to give compound 6 (148.9 mg). H2 (10.8 g) was applied to silica gel using isocratic elution of nhexane: EtOAc (50: 50, v/v) to obtain 4 subfractions (H2-1 to H2-4). H2-3 was identified to be compound 7 (6.5 g). Subfraction H2-1 (0.3 g) was subjected to silica gel CC eluted with n-hexane: EtOAc (50: 50, v/v) to yield compound 8 (20.0 mg). H2-2 (3.5 g) was purified by CC eluted with n-hexane: EtOAc (50:50, v/v) to afford compound 9 (2.8 g). The EtOAc extract (140.0 g) was fractionated by CC on silica gel (400 g), eluting with n-hexane: EtOAc (100: 0 to 0: 100, v/v) and EtOAc: MeOH (100:0 to 0:100, v/v) to give 4 fractions (E1 to E4). E3 (21.4 g) was applied to silica gel using isocratic elution of CH₂Cl₂-: MeOH (98: 2, v/v) to yield 4 subfractions (E3-1 to E3-4). E3-2 (1.8 g) was repeated CC with EtOAc: n-hexane (50: 35, v/v) to give compound 10 (137.7 mg). E3-3 (3.2 g) was subjected to silica gel CC eluted with EtOAc: n-hexane (50:35, v/v) to yield 3 subfractions (E3-3-1 to E3-3-3). E3-3-3 (0.8 g) was repeatedly CC over silica gel eluting with CH₂Cl₂: MeOH (98: 2, v/v) to afford compounds 11 (30.4 mg) and 12 (6.4 mg). E4 (67.0 g) was subjected to silica gel CC eluted with CH₂Cl₂: MeOH (95: 5, v/v) to

yield 4 subfractions (E4-1 to E4-4). E4-2 (5.4 g) was chromatographed on silica column using gradient elution of CH2Cl2-: MeOH (98: 2 to 95: 5, v/v) to give 4 subfractions (E4-2-1 to E4-2-4). E4-2-3 (1.5 g) yielded compound 13 (49.0 mg) by recrystallization with MeOH. The filtrate upon evaporation gave a residue (849.0 mg) which was CC on silica gel eluting under gradient condition of CH₂Cl₂: MeOH (98:2 to 96:4, v/v) to give compound 14 (217.8 mg). E4-3 (6.9 g) was subjected to silica gel CC eluted with EtOAc: MeOH (97:3, v/v) to yield 4 subfractions (E4-3-1 to E4-3-4). E4-3-3 (3.5 g) was purified by CC eluted with CH₂Cl₂: MeOH (95: 5, v/v) to afford 2 subfractions (E4-3-3-1 and E4-3-3-2). E4-3-3-2 was identified to be compound 15 (34.9 mg). E4-3-3-1 (0.4 g) was purified by CC eluted with CH_2Cl_2 : MeOH (98: 2, v/v) to give compound 16 (35.4 mg). The MeOH part (60.0 g) was fractionated by CC on silica gel (300 g), and it was eluted with CH₂Cl₂: MeOH (100: 0 to 0: 100, v/v) to give 6 fractions (M1 to M6). M5 (22.9 g) was subjected to reversed phase RP-18 CC eluted with MeOH: H_2O (50: 25, v/v) to afford 6 subfractions (M5-1 to M5-6). M5-2 (3.2 g) was repeatedly chromatographed on Sephadex LH-20 eluting with 100% MeOH, followed by reversed phase RP-18 CC using MeOH: $H_2O(50:25, v/v)$ to afford compound 2 (105.0 mg). M5-3 (1.6 g) was purified by reversed phase RP-18 CC eluted with MeOH: H₂O (50: 25, v/v) to give compound 1 (46.4 mg). Similarly, M5-4 (5.6 g) was subjected to reversed phase RP-18 gel CC to afford 5 subfractions (M5-4-1 to M5-4-5). M5-4-5 (2.3 g) was separated by reversed phase RP-18 CC eluted with MeOH: H2O (50:20, v/v) to give 3 subfractions (M5-4-5-1 to M5-4-5-3). Then, M5-4-5-3 (0.5 g) was purified by reversed phase RP-18 CC using H_2O : ACN (50: 25, v/v) to afford compound 4 (26.4 mg). M5-4-3 (2.5 g) was subjected to reversed phase RP-18 CC eluted with MeOH: H_2O (50: 20, v/v) to obtain 3 subfractions (M5-4-3-1 to M5-4-3-3). Then, M5-4-3-2 (0.4 g) was repeatedly chromatographed on RP-18 column using H2O: ACN (50:25, v/v) to afford compound 5 (42.8 mg) and M5-4-3-3 (0.9 g) was CC using RP-18 with MeOH: H₂O (50: 25, v/v) as eluting solvent to yield compound 3 (158.8 mg).

Compound 1. White powder; $[a]_D^{26} - 17.2^\circ$ (c 0.39 in MeOH); Mp 219–222 °C; IR (ATR) ν_{max} : 3337, 2932, 1734, 1639, 1455, 1387, 1255, 1073, 894, 815 cm⁻¹; ¹H-NMR (pyridine- d_5 , 400 MHz) data and ¹³C-NMR (pyridine- d_5 , 100 MHz) data: see Tables 1–4; ESI-TOF-MS m/z 1229.5924 [M + Na]⁺ (calcd for $C_{58}H_{94}O_{26}Na$, 1229.5926).

Compound 2. White powder; $[\alpha]_D^{26} - 22.7^\circ$ (c 0.24 in MeOH); Mp 215–217 °C; IR (ATR) ν_{max} : 3442, 2928, 1727, 1643, 1455, 1374, 1258, 1073, 894, 837 cm⁻¹; ¹H-NMR (pyridine- d_5 , 400 MHz) data and ¹³C-NMR (pyridine- d_5 , 100 MHz) data: see Tables 1–4; ESI-TOF-MS m/z 1245.5869 [M + Na]⁺ (calcd for $C_{58}H_{94}O_{27}Na$, 1245.5875).

Compound 3. White powder; $[a]_{\rm D}^{26} - 39.6^{\circ}$ (c 0.20 in MeOH); Mp 204–206 °C; IR (ATR) $\nu_{\rm max}$: 3332, 2933, 1751, 1639, 1455, 1361, 1258, 1030, 894, 820, 809 cm⁻¹; ¹H-NMR (pyridine- d_5 , 400 MHz) data and ¹³C-NMR (pyridine- d_5 , 100 MHz) data: see Tables 1–4; ESI-TOF-MS m/z 1345.6331 [M + Na]⁺ (calcd for $C_{63}H_{102}O_{29}Na$, 1345.6399).

Compound 4. White powder; $[\alpha]_D^{26} - 31.1^\circ$ (*c* 0.38 in MeOH); Mp 213–215 °C; IR (ATR) ν_{max} : 3338, 2934, 1734, 1639, 1458,

1390, 1258, 1034, 892, 818, 779 cm $^{-1}$; 1 H-NMR (pyridine- d_{5} , 400 MHz) data and 13 C-NMR (pyridine- d_{5} , 100 MHz) data: see Tables 1–4; ESI-TOF-MS m/z 1345.6340 [M + Na] $^{+}$ (calcd for $C_{63}H_{102}O_{29}Na$, 1345.6399).

Compound 5. White powder; $[\alpha]_{\rm D}^{26}$ –11.6° (c 0.30 in MeOH); Mp 180–182 °C; IR (ATR) $\nu_{\rm max}$: 3344, 2933, 1719, 1640, 1455, 1387, 1261, 1042, 894, 818, 782 cm⁻¹; ¹H-NMR (pyridine- d_5 , 400 MHz) data and ¹³C-NMR (pyridine- d_5 , 100 MHz) data: see Tables 1–4; ESI-TOF-MS m/z 1359.6145 [M + Na]⁺ (calcd for $C_{63}H_{100}O_{30}$ Na, 1359.6192).

Acid hydrolysis

Paper

The new compounds 1-5 (5.0 mg each) were individually dissolved in MeOH (0.2 mL) and TFA (2 N, 1 mL each) was added. The mixture was heated at 95 °C for 24 h and the progress of the reaction was monitored by TLC. The residues were evaporated to dryness by repeatedly adding MeOH to remove the acid. The residue was analyzed by TLC over silica gel together with authentic sugar standards using CHCl₃: MeOH: H₂O (3:1.5:0.2) and the sugars were then separated by silica gel using the same eluting solvent system as the mobile phase. 38,39 The monosaccharides of compounds 1-5 were identified as D-glucose, L-rhamnose and Dxylose by TLC and NMR comparisons with those of standard sugars. It was detected at 589 nm on a polarimeter that the optical rotation of 1.02% (w/v) of p-glucose ($[\alpha]_D^{26} + 18.7^{\circ}$), 40-43 1.12% (w/v) of L-rhamnose $(\alpha)_{D}^{26} + 5.0^{\circ}$ and 0.35% (w/v) of D-xylose $(\alpha)_{D}^{26} + 11.6^{\circ}$ consisting with a homologous standard distilled water solution. Moreover, by comparison of the spectroscopic data (Fig. S3-S10†) with those reported in literature, the aglycones obtained from acid hydrolysis were identified as hederagenin²⁵ (1a, Fig. S2†) from compound 1, bayogenin31,32 (2a, Fig. S2†) from compound 2, oleanolic acid33 (3a, Fig. S2†) from compounds 3 and 4, and gypsogenin³⁵ (4a, Fig. S2†) from compound 5.

Alkaline hydrolysis

The new compounds 1–5 (5 mg each) were dissolved in MeOH (0.2 mL each), then saturated solution of NaOH (1 mL each) were added. The mixture was heated at 60 °C for 24 h, acidified with 1 N HCl and the mixture was then evaporated to dryness *in vacuo*. The triterpenoid glycosides were separated by silica column chromatography using CHCl₃: MeOH: H_2O (3:1.5:0.2) as the mobile phase. The glycosides hederagenin-3-O- β -D-glucopyranoside glycones, the glycosides hederagenin-3-O- β -D-glucopyranoside (2b, Fig. S2, Tables S1 and S2†), bayogenin-3-O- β -D-glucopyranoside (3b, Fig. S2, Tables S1 and S2†) and gypsogenin-3-O- β -D-glucopyranoside (3b, Fig. S2, Tables S1 and S2†) and gypsogenin-3-O- β -D-glucopyranoside (4b, Fig. S2, Tables S1 and S2†), which was derived from compounds 1–5, respectively.

Anti-inflammatory assay

Cell culture. RAW264.7 murine macrophages (ATCC, Rockville, MD, USA) were cultured in RPMI-1640 medium (Corning, NY, USA) supplemented with 10% fetal bovine serum, and 1% penicillin/streptomycin (Gibco, MA, USA). Cells were incubated at 37 $^{\circ}$ C with 5% CO₂ in a humidified incubator.

Cell viability test. RAW264.7 cells were grown in a 96-well plate at a density of 1.0×10^5 cells per well for 24 h. Cells were treated with the compounds at various concentrations ranged from 5–160 μ M in combination with 10 ng mL⁻¹ of LPS. After 24 h, cells were then incubated with 0.5 mg mL⁻¹ of 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) solution for another 3 h. After removal of MTT solution, the formazan crystal products were dissolved in DMSO (200 μ L per well). The absorbance was measured using a microplate reader at a wavelength of 560 nm. Three independent experiments were performed in triplicates. IC₅₀ of cell viability by MTT was calculated from a dose–response curve using GraphPad Prism 6 (GraphPad Software Inc., CA, USA).

Nitric oxide assay. RAW264.7 cells of 1.0×10^5 cells per well were seeded into 96-well plates and grown for 24 h. Cells were pretreated with various concentrations of compounds (5–160 μ M) followed by 10 ng mL⁻¹ LPS stimulation for 24 h. Then, culture supernatants were collected and Griess reagent (1% sulfanilamide and 0.1% naphthylethylene in 2.5% phosphoric acid solution) was added in a 96-well plate according to the described method.³⁷ The absorbance was measured at 540 nm using a microplate reader. Nitrite levels were determined using a NaNO₂ standard curve with $r^2 > 0.999$. Data were obtained from three independent experiments in triplicates. IC₅₀ by nitric oxide assay was determined from a dose–response curve using GraphPad Prism 6 (GraphPad Software Inc., CA, USA). The selectivity index (SI) was calculated as the ratio of IC₅₀ (MTT) to IC₅₀ (NO).

Conflicts of interest

Authors declare no conflict of interest.

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Notes and references

- 1 B. E. E. Duyfjes and K. Pruesapan, *Thai Forest Bulletin* (Botany), 2004, 32, 76–109.
- 2 S. Kongtun, W. Jiratchariyakul, T. Kummalue, P. Tan-ariya, S. Kunnachak and A. W. Frahm, *Planta Med.*, 2009, 75, 839–842.
- 3 R. M. Kolte, V. V. Bisan, C. R. Jangde and A. A. Bhalerao, *Indian J. Indig. Med.*, 1997, **8**, 117–121.
- 4 M. Arawwawala, I. Thabrew and L. Arambewela, *Int. J. Biol. Chem. Sci.*, 2009, 3, 287–296.
- 5 H. Kirana and B. Srinivasan, *Indian J. Pharmacol.*, 2008, 40, 103–106.

- 6 A. Kar, B. K. Choudhury and N. G. Bandyopadhyay, *J. Ethnopharmacol.*, 2003, **84**, 105–108.
- 7 S. S. Kumar, B. R. Kumar and G. K. Mohan, *J. Ethnopharmacol.*, 2009, **123**, 347–350.
- 8 L. D. Arawwawala, M. I. Thabrew and L. S. Arambewela, *J. Ethnopharmacol.*, 2010, **127**, 750–754.
- N. D. Kage, V. B. Malashetty, Y. N. Seetharam, P. Suresh and S. B. Patil, *Int. J. Morphol.*, 2009, 27, 173–182.
- 10 A. A. Rahuman and P. Venkatesan, *Parasitol. Res.*, 2008, **103**, 133–139.
- 11 S. Duangmano, P. Sae-lim, A. Suksamrarn, F. E. Domann and P. Patmasiriwat, *BMC Complementary Altern. Med.*, 2012, 12(185), 1–12.
- 12 K. Dittharot, S. Dakeng, P. Suebsakwong, A. Suksamrarn, P. Patmasiriwat and M. Promkan, *Planta Med.*, 2019, 85, 370–378.
- 13 P. Suebsakwong, J. Wang, P. Khetkam, N. Weerapreeyakul, J. Wu, Y. Du, Z.-J. Yao, J.-X. Li and A. Suksamrarn, ACS Med. Chem. Lett., 2019, 10, 1400–1406.
- 14 H. Kojima, N. Sato, A. Hatano and H. Ogura, *Phytochemistry*, 1990, **29**, 2351–2355.
- 15 H. Jacobs and T. Singh, J. Nat. Prod., 1990, 53, 1600-1605.
- 16 S. Y. Ryu, S. H. Lee, S. U. Choi, C. O. Lee, Z. No and J. W. Ahn, *Arch. Pharmacal Res.*, 1994, 17, 348–353.
- 17 Y. Yamada, K. Hagiwara, K. Iguchi and Y. Takahashi, *Chem. Lett.*, 1978, 319–322.
- 18 V. V. Velde and D. Lavie, Tetrahedron, 1983, 39, 317-321.
- 19 C. Seger, S. Sturm, M. Mair, E. Ellmerer and H. Stuppner, *Magn. Reson. Chem.*, 2005, 43, 489-491.
- 20 M. Arisawa, J. M. Pezzuto, A. D. Kinghorn, G. A. Cordell and N. R. Farnsworth, J. Pharmaceut. Sci., 1984, 73, 411–413.
- 21 S. M. Kupchan, H. Meshulam and A. T. Sneden, *Phytochemistry*, 1978, **17**, 767–769.
- 22 C. Seger, S. Sturm, E. Haslinger and H. Stuppner, *Monatsh. Chem.*, 2005, **136**, 1645–1649.
- 23 H. Abd El-Fattah, Phytochemistry, 1994, 36, 159-161.
- 24 N. Kawahara, A. Kurata, T. Hakamatsuka, S. Sekita and M. Satake, *Chem. Pharm. Bull.*, 2004, **52**, 1018–1020.
- 25 X. C. Li, C. R. Yang, Y. Q. Liu, R. Kasai, K. Ohtani, K. Yamasaki, K. Miyahara and K. Shingu, *Phytochemistry*, 1995, **39**, 1175–1179.
- 26 Y. Shao, B.-N. Zhou, L.-Z. Lin and G. A. Cordell, *Phytochemistry*, 1995, **38**, 927–933.
- 27 E. Bedir, N. J. Toyang, I. A. Khan, L. A. Walker and A. M. Clark, *J. Nat. Prod.*, 2001, **64**, 95–97.

- 28 B. Yeskaliyeva, M. A. Mesaik, A. Abbaskhan, A. Kulsoom, G. S. Burasheva, Z. A. Abilov, M. I. Choudhary and Atta-ur-Rahman, *Phytochemistry*, 2006, **67**, 2392–2397.
- 29 J. M. Augustin, S. Drok, T. Shinoda, K. Sanmiya, J. K. Nielsen, B. Khakimov, C. E. Olsen, E. H. Hansen, V. Kuzina, C. T. Ekstrom, T. Hauser and S. Bak, *Plant Physiol.*, 2012, 160, 1881–1895.
- 30 Y. Shao, B. N. Zhou, K. Ma and H. M. Wu, *Planta Med.*, 1995, 61, 246–249.
- 31 R. Kasai, M. Miyakoshi, R. L. Nie, J. Zhou, K. Matsumoto, T. Morita, M. Nishi, K. Miyahara and O. Tanaka, *Phytochemistry*, 1988, 27, 1439–1446.
- 32 T. Schopke, V. Wray, A. Kunath and K. Hiller, *Phytochemistry*, 1992, **31**, 2555–2557.
- 33 H. Zhang, A. K. Samadi, K. V. Rao, M. S. Cohen and B. N. Timmermann, *J. Nat. Prod.*, 2011, 74, 477–482.
- 34 M. Miyakoshi, K. Shirasuna, Y. Hirai, K. Shingu, S. Isoda, J. Shoji, Y. Ida and T. Shimizu, *J. Nat. Prod.*, 1999, **62**, 445–448.
- 35 T. Kuljanabhagavad, P. Thongphasuk, W. Chamulitrat and M. Wink, *Phytochemistry*, 2008, **69**, 1919–1926.
- 36 B. Yeskaliyeva, M. A. Mesaik, A. Abbaskhan, A. Kulsoom, G. S. Burasheva, Z. A. Abilov, M. I. Choudhary and Atta-ur-Rahman, *Phytochemistry*, 2006, 67, 2392–2397.
- 37 J. Sun, X. Zhang, M. Broderick and H. Fein, Sensors, 2003, 3, 276–284.
- 38 A. Rezgui, A. C. Mitaine-Offer, T. Miyamoto, C. Tanaka, S. Delemasure, P. Dutartre and M. A. Lacaille-Dubois, *Phytochemistry*, 2016, 123, 40–47.
- 39 Y. Ding, X. R. Tian, H. F. Tang, J. T. Feng, W. Zhang, W. L. Hai, X. Y. Wang and Y. Wang, *Phytochem. Lett.*, 2012, 5, 668–672.
- 40 A. Bagno, F. Rastrelli and G. Saielli, *J. Org. Chem.*, 2007, 72, 7373–7381.
- 41 A. Penzkofer, J. Anal. Sci., 2013, 3, 234–239.
- 42 S. Y. Wu, Y. H. Fu, Q. Zhou, M. Bai, G.-Y. Chen, S. Y. Zhao, C. R. Han and X. P. Song, *Molecules*, 2018, 23, 1–9.
- 43 H. S. Isbell, Bur. Stand. J. Res., 1929, 3, 1041-1052.
- 44 J. D. Britto, V. S. Manickam, S. Gopalakrishnan, T. Ushioda and N. Tanaka, *Chem. Pharm. Bull.*, 1995, **43**, 338–339.
- 45 P. A. Levene and R. S. Tipson, *J. Biol. Chem.*, 1936, **115**, 731–747.
- 46 S. Kirmizigul, H. Anil and M. E. Rose, *Phytochemistry*, 1995, **39**, 1171–1174.