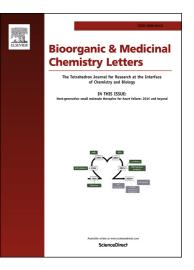
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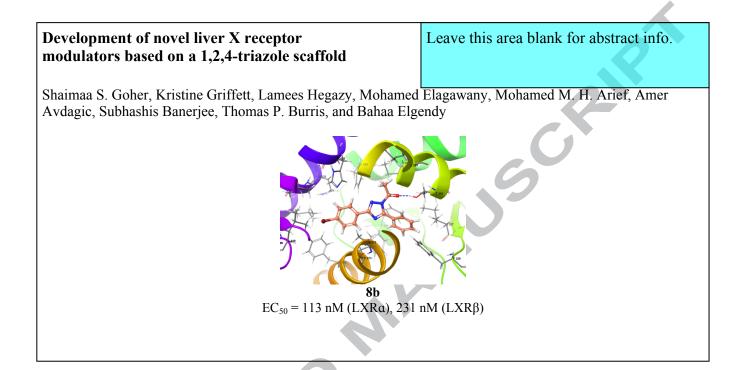
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Development of novel liver X receptor modulators based on a 1,2,4-triazole scaffold

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ABSTRACT

Liver X Receptor (LXR) agonists have been reported as a potential treatment for atherosclerosis, Alzheimer's disease and hepatitis C virus (HCV) infection. We have designed and synthesized a series of potent 1,2,4-triazole scaffolds as novel LXR modulators. In cell-based cotransfection assays these compounds generally functioned as LXR agonists and we observed compounds with selectivity towards LXR α (7-fold) and LXR β (7-fold) in terms of potency. Assessment of the effects of selected compounds on LXR target gene expression in HepG2 cells revealed that compounds **6a-b** and **8a-b** behaved as inverse agonists on FASN expression even though they were agonists in the LXR α and LXR β cotransfection assays. Interestingly, these compounds had no effect on the expression of SREBP-1c confirming a unique LXR modulator pharmacology. Molecular docking studies and evaluation of ADME properties *in-silico* show that active compounds possess favorable binding modes and ADME profiles. Thus, these compounds may be useful for in vivo characterization of LXR modulators with unique profiles and determination of their potential clinical utility.

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Liver X receptors (LXRs) have emerged as a promising drug targets because they effectively regulate lipid metabolism and inflammation. LXRs are known as oxysterol receptors and have two isoforms, LXR α (NR1H3) and LXR β (NR1H2). In humans, LXR α consists of 447 amino acids and is expressed mainly in liver, kidney, intestine, adipose tissue, and macrophages; LXR β consists of 460 amino acids and is expressed ubiquitously.^{1,2}

A growing body of literature indicates that LXR is a potential drug target for a range of diseases. LXR agonists have been shown to increase reverse cholesterol transport and reduce atherosclerosis by targeting induction of ATP-binding Cassette Transporter A1 (ABCA1) gene expression.²⁻⁷ Agonists have also been shown to have anti-inflammatory activity.^{2,8,9} Efficacy in Alzheimer's disease models and cancer models has also been demonstrated for LXR agonists.^{2,8,10-14} Stimulation of hepatic lipogenesis by agonists (via stimulation of transcription of SREBP-1c, FASN, and SCD1) has limited development of agonists for treatment of these diseases.

This has led to efforts to develop LXR modulators that would display tissue selective gene modulation effects (e.g. ABCA1) so as to retain the ability to activate key beneficial pathways while avoiding those that are detrimental (e.g. SREBP-1c). The potential utility of LXR ligands has continued to expand. LXR agonists have been shown to reduce the replication of hepatitis C virus via targeting key genes that are required for HCV replication.¹⁵ Based on the ability of LXR to regulate de novo lipogenesis, we have evaluated LXR inverse agonists for their ability to suppress tumor growth and to treat non-alcoholic steatohepatitis with success.^{14,16}

Compound **3** is a new LXR modulator that was recently identified using pharmacophore modeling and shape-based virtual screening.¹⁷ When this compound was evaluated in an LXR luciferase reporter gene assay, it showed moderate partial agonist activity compared to LXR full agonist GW3965 (**2**). Development of a partial agonist may be advantageous since LXR full agonists have been demonstrated to cause severe adverse side effects due to elevation of hepatic lipogenesis. There is no additional structure-activity-relationship (SAR) data in the literature on compound **3**. The 1,2,4-triazole heterocycle is an important building block for the synthesis of biologically active molecules and found wide variety of applications in medicinal chemistry.^{18,19}

As part of our ongoing research to discover novel LXR modulators of clinical importance,^{14,16} we developed new LXR agonists based on the 1,2,4-triazole scaffold **3** to evaluate them as potential drugs.

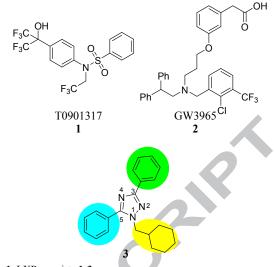
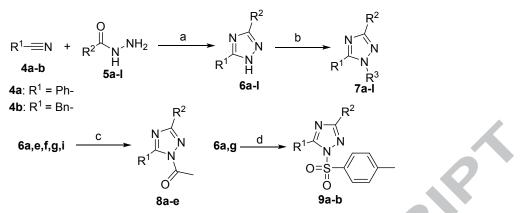


Figure 1. LXR agonists 1-3.

We have synthesized a library of new LXR agonists based on 1,2,4-triazole scaffold **3**, which we divided into three parts as highlighted in Figure 1. We studied the structure activity relationships (SAR) by modifying these three parts, namely substituents at N-1, C-3, and C-5 through different chemical transformations. The structures of synthesized compounds were elucidated using different spectroscopic techniques.

3,5-disubstituted-1*H*-1,2,4-triazoles (**6a-I**) were synthesized by reacting nitriles (**4a,b**) with hydrazides (**5a-I**) in *n*-butanol in the presence of anhydrous potassium carbonate at 150 °C.²⁰ The synthesized triazoles (**6a,d,e,g**) reacted with different alkyl halides in DMF to give 1-alkyl-3,5-disubstituted-1*H*-1,2,4-triazoles (**7a-I**).²¹ Acetyl triazoles (**8a-e**) were obtained by refluxing 3,5-disubstituted-1*H*-1,2,4-triazoles (**6a,e,f,g,i**) with acetic anhydride. Sulphonyl derivatives (**9a,b**) were obtained via reacting **6a,g** with *p*-toluene sulfonyl chloride in presence of suitable base (Scheme 1).

We evaluated the activity of synthesized compounds in a cell based (HEK293 cells) cotransfection assay using either full-length human LXR α or LXR β and a reporter containing 3 copies of an LXR response element (LXRE) within the promoter of a luciferase reporter gene. We initially decided to evaluate the activity of 3,5disubstituted-1H-1,2,4-triazoles (6a-l) as the core scaffold of ligand 3. Testing these compounds against both LXR isoforms showed potent agonistic activity. The parent triazole 6a showed potent agonist activity against LXR α (EC₅₀ for LXR α = 142 nM) and moderate activity against LXR β (EC₅₀ for LXR β = 1008 nM). Compound 6a was 7-fold selective for LXRa over LXRB. Then, we explored the effect of substituents on the phenyl ring at C-3 (R²). Electron withdrawing substituents at 2-position of the phenyl ring (6b and 6c) showed good agonistic activity against both LXR isoforms, higher efficacy toward LXRB, and slight selectivity toward LXR α (Table 1). Switching the position of Cl and Br from 2-position to 4-position (6d and 6e) enhanced the agonistic activity against both LXR isoforms and reversed the selectivity toward LXRβ (Table 1).



Scheme 1. Reagents and conditions: (a) K₂CO₃, n-BuOH, 150 °C; (b) R³Cl, K₂CO₃, DMF, reflux; (c) Ac₂O, reflux 2-3 h; (d) TsCl, pyridine, reflux 6 h

 Table 1. LXR activity of 1,2,4-triazole derivatives 6a-l

			I	LXRα	I	LXRβ		
Compound	\mathbf{R}^{1}	R ²	EC ₅₀ (nM)	$E_{max}(\%)$	EC ₅₀ (nM)	$E_{max}(\%)$	EC ₅₀ α/β	ClogP
6a	Ph-	Ph-	142	56	1008	63	0.14	3.40
6b	Ph-	2-Cl-Ph-	526	60	640	74	0.82	3.57
6c	Ph-	2-Br-Ph-	390	52	493	67	0.79	3.63
6d	Ph-	4-Cl-Ph-	309	51	245	60	1.26	3.64
6e	Ph-	4-Br-Ph-	164	55	68	62	2.41	3.73
6f	Ph-	4-CH ₃ -Ph-	77	62	85	73	0.91	3.50
6g	Ph-	3-CH ₃ O-Ph-	108	60	68	72	1.59	3.24
6h	Ph-	3-Pyridyl	ia	-	ia	-	-	2.32
6i	Ph-	PhCH ₂ -	2202	45	6152	59	0.36	3.54
6j	PhCH ₂ -	Ph-	ia	-	ia	-	-	3.52
6k	PhCH ₂ -	2-Cl-Ph-	570	46	966	61	0.59	3.87
61	PhCH ₂ -	4-Cl-Ph-	ia	-	2573	73	-	3.99

ia = inactive at 10 μ M

 ${}^{a}E_{max}(\%)$ is the percentage ratio between maximum fold induction of the tested compound and fold induction for 1 at 10 μ M.

Electron donating groups on the phenyl ring of R² enhanced the agonistic activity against both LXR isoforms significantly. Compound **6f** with CH₃ group at 4-position was potent dual agonist for LXRα (EC₅₀ for LXRα = 77 nM) and LXRβ (EC₅₀ for LXRβ = 85 nM). Similarly, compound **6g** was potent dual agonist (EC₅₀ = 108 and 68 nM, for LXRα and LXRβ, respectively), but more selective toward LXRβ by 1.6-fold. Compound **6g** was docked in the LBP of LXRβ (Figure 2A). The ligand makes mainly hydrophobic contacts with Ile309, Met312, Leu313, Ile327, Phe329 and Leu274. The tautomer of compound **6g** makes hydrogen bonding interaction through the hydrogen atom on N-1 to Thr271. Methoxy group made another hydrogen bond with Thr316, which is likely responsible for higher potency of compound **6g** over compound **6a** that lacks hydrogen-bond donor at this position.

When we replaced the phenyl group in **6a** with 3-pyridyl group in **6h**, the compound lost the agonistic activity against both isoforms. Interestingly, the clogP of this compound is the lowest in this series and this might be the reason for the loss of activity. Replacing the phenyl group in **6a** with benzyl group in **6i** reduced the agonistic activity for both LXR isoforms but showed ≈ 2.8 -fold selectivity toward LXR α (Table 1).

Changing substituents at C-5 (R¹) from phenyl group (**6**a) to benzyl group (**6**j), diminished the agonistic activity toward both LXR α and LXR β . Substituting the phenyl group at C-3 of **6**j with chlorine in 2-position (**6**k) regained the agonistic activity toward LXR α (EC₅₀ = 570 nM) and LXR β (EC₅₀ = 966 nM). Transposition of chlorine atom to 4-position (**6**l) rendered the compound inactive against LXR α and lowered the activity toward LXR β by \approx 4.9-fold compared to **6**j and by \approx 2.7-fold compared to **6**k. Addition of CH₂ as a linker between the triazole ring and the phenyl ring at C-5 in **6**j, **6**k, and **6**l caused the geometry to be tilted in all three compounds if compared to their corresponding derivatives, **6a**, **6b**, and **6d**. This change in the geometry might affect the complementarity of **6**j, **6**k, and **6**l within the binding site and disrupt their π - π stacking with the surrounding aromatic residues. For example, compound **6b** was predicted to make π - π stacking with Phe271, Phe340, Phe329 and a halogen bond with Thr316 but the change in geometry in **6k** due to the extra CH₂, led to flipping of the binding pose and loss of the halogen bond interaction and one of the π - π stacking interactions (Figure S1). Similarly, if we compare **6d** to **6l**, the addition of CH₂ linker led to less complementarity of the ligand within the binding pocket where CH₂ group was displaced out of the hydrophobic region and located instead in a hydrophilic region (Figure S2). Therefore, we suggest that aromatic substitution at C-5 (R¹) is important to maintain good agonistic activity.

Next, we investigated the effect of substituting N-1 with different alkyl groups while maintaining R¹ as a phenyl group and changing the substituent pattern at R² by synthesizing compounds 7a-l (Table 2). Alkylating compound 6a with different alkyl groups provided compounds 7a-e. Substituting N-1 with methyl group in 7a reduced the agonistic activity dramatically toward LXRα (≈7fold) and slightly toward LXR β (\approx 2.6-fold) if compared to the parent triazole 6a. Compounds 7b-e with an ethyl, butyl, ally, and benzyl groups at N-1 were very weak agonists against LXRa and completely inactive against LXRB (Table 2). Compound 7f where R² is 3-methoxy phenyl and R³ is benzyl group showed moderate agonistic activity against LXR α (EC₅₀ for LXR α = 903 nM), and weak activity against LXR β (EC₅₀ for LXR β = 2365 nM). This compound was \approx 2.6-fold more selective toward LXR α over LXR β . On the contrary, the corresponding analogue, **6g**, was more selective toward LXR β over LXR α by \approx 1.6-fold. Comparison of the docking poses of compounds 6g and 7f in the LBP of LXR β predicts that adding alkyl groups at the N-1 position disrupted the ligand's complementarity in the ligand binding pocket leading to flipped binding pose for compound 7f and loss of both hydrogen bonds formed by compound 6g tautomer with amino acid residues Thr316 and Phe271 (Figure 2B).

Table 2. LXR activity of 1,2,4-triazole derivatives 7a-l

				L	XRa	L	XRβ		
Comp.	\mathbb{R}^1	R ²	R ³	EC ₅₀ (nM)	$E_{max}(\%)$	$EC_{50}(nM)$	$E_{max}(\%)$	EC ₅₀ α/β	ClogP
7a	Ph-	Ph-	CH ₃ -	2468	64	2637	71	0.94	3.69
7b	Ph-	Ph-	C ₂ H ₅ -	ia	-	ia	-	-	4.92
7c	Ph-	Ph-	C ₄ H ₉ -	2470	65	ia	-	-	5.08
7d	Ph-	Ph-	C ₃ H ₅ -	ia	-	ia	-	-	4.56
7e	Ph-	Ph-	PhCH ₂ -	2178	58	ia	-	-	5.40
7f	Ph-	3-CH ₃ O-Ph-	PhCH ₂ -	903	62	2365	69	0.38	3.24
7g	Ph-	4-Cl-Ph-	CH ₃ -	1189	77	1985	83	0.60	4.22
7 h	Ph-	4-Cl-Ph-	C ₂ H ₅ -	2285	60	ia	-	-	4.64
7i	Ph-	4-Cl-Ph-	C ₄ H ₉ -	ia	-	ia	-	-	5.58
7j	Ph-	4-Cl-Ph-	PhCH ₂ -	2130	53	6706	61	0.32	6.17
7k	Ph-	4-Br-Ph-	C ₂ H ₅ -	2237	45	5647	71	0.40	4.79
71	Ph-	4-Br-Ph-	PhCH ₂ -	5025	78	712	69	7.06	6.24

B.

ia = inactive at 10 µM

 ${}^{a}E_{max}$ (%) is the percentage ratio between maximum fold induction of the tested compound and fold induction for 1 at 10 μ M



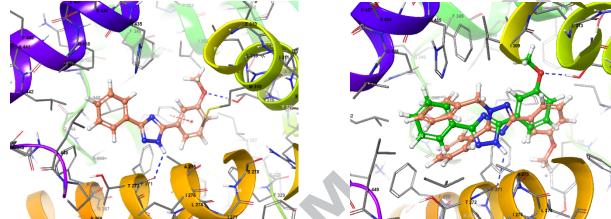


Figure 2. A. Molecular interactions of compound 6g in the LBP of LXRB. B. Overlay of compounds 6g tautomer (green) and 7f (pink). Hydrogen bonds are shown in blue dotted lines.

Alkylation of 6d and 6e with different alkyl groups afforded compound 7g-l. These compounds showed similar pattern to their corresponding analogues 7a-f, where they showed weak activity against both LXR isoforms. Compound 71 showed good activity toward LXR β (EC₅₀ for LXR β = 712 nM) and was 7-fold more selective to LXR β over LXR α (Table 2). This compound was the most lipophilic among synthesized compounds (clogP = 6.24) and it was shown by Ishikawa that higher lipophilicity is important for LXRß selectivity.²² It is noteworthy that the ligand binding pockets for both LXR α and LXR β are very similar. Therefore, rationalizing ligand selectivity to both receptors based on ligands' interactions only is not possible and other aspects of molecular recognition such as receptor-ligand dynamics and entropy should be taken into consideration.¹¹ Molecular dynamics simulations and free energy calculations are currently underway to determine the molecular basis of differential selectivity between LXRa and LXR_β.

We next explored the effect of acylating parent triazoles 6a, 6e, 6f, 6g, and 6i using acetic anhydride to form acetyl derivatives 8ae. Substituting N-1 with acetyl group in 8a lowered the agonistic activity toward LXRa by 2.8-fold, but increased the agonistic activity and selectivity toward LXRB significantly (EC50 for LXRB = 232 nM, EC₅₀ α/β = 1.71) if compared to the parent triazole **6a**. Compound 8b showed a higher agonistic activity toward LXRa and slight increase in E_{max} (58% for **8b** vs 55% for **6e**) if compared to 6e (Table 1 and Table 3). The agonistic activity of 8b toward LXR β was significantly lower than **6e** but E_{max} increased from 62% to 78% (Table 1 and Table 3). Compound 8b was 2-fold more selective for LXRa. Docking of compound 8b in the LBP of LXRB predicts that this compound is stabilized by forming a hydrogen bond with Thr316 and with van der Waals forces and hydrophobic

contacts with surrounding residues such as Phe271, Ile327, Phe329, Leu313, Met312, Ile309, and Phe349 (Figure 3).

Acetylation of 6f led to 8c, which exhibited 3- and 4-fold lower activity against LXRa and LXRB, respectively. Compound 8d showed good agonistic activity against both LXR isoforms. There was no real difference in agonistic activity between 6g and 8d toward LXR α , while there was a change in EC₅₀ from 68 nM in 6g to 109 nM in 8d. This difference was compensated slightly by increase in E_{max} from 72% in 6g to 78% in 8d. Compound 8e showed more than 3- and 4-fold enhancement in agonistic activity over the parent **6i** toward LXR α and LXR β , respectively. Similar to the parent 6i, compound 8e is 2-fold more selective toward LXRa over LXRB.

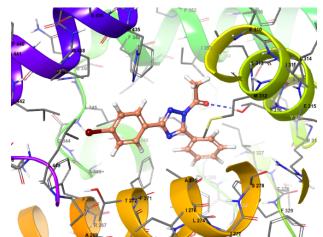


Figure 3. Molecular interactions of compound 8b in the LBP of LXRB. Hydrogen bonds are shown as blue dotted lines and π - π stacking interactions are shown as dotted red lines

Finally, we explored the possibility of substituting N-1 with tosyl group by synthesizing **9a** and **9b**. Compound **9a** was > 4- and 12-fold less active against LXR α than the corresponding **6a** and **8a**, respectively. On the other hand, **9a** possess similar activity to **8a** against LXR β and 3.9-fold higher activity than **6a** against the same isoform. Compound **9b** was completely inactive against both LXR α and LXR β . We did not explore scaffold **9** further because it is not stable for long time at room temperature and tend to decompose to the parent triazole.

The SAR of synthesized compounds revealed that 3,5disubstituted 1,2,4-triazoles are generally good modulators of LXR when the two substituents are phenyl rings or substituted phenyl rings (e.g. 6a-6g). Substituting the phenyl ring at C-3 of the triazole ring with electron withdrawing groups make the compounds behave as dual agonists toward both LXR isoforms with slight selectivity toward LXRB (e.g. 6b-6e). Electron donating substituents at the same ring increased the potency toward both LXR isoforms significantly (e.g. 6f and 6g). Substituting C-3 of the triazole ring with pyridine renders the compound inactive (e.g. 6h). Similarly, the compounds lost activity when C-5 was substituted with benzyl group (e.g. 6j-6l). Substituting N-1 of the triazole ring with different alkyl groups led to mostly inactive compounds (e.g. 7a-7l). On the other hand, substituting N-1 with acetyl group gave compounds of comparable potency to the parent triazole with higher efficacy (e.g. 8a-8d).

We measured the effect of T0901317 (1), **6a-b**, and **8a-b** on gene expression of direct target genes of LXR: Fatty acid synthase (FAS), Sterol Regulatory Element Binding Protein 1c (SREBP-1c), 3-Hydroxy-3-Methylglutaryl-CoA Reductase (HMGCR), and LXRa. LXR agonists increase the expression of LXR target genes *FASN* and *SREBP-1c* in the liver causing increased *de novo* lipogenesis. Long term effects of increased expression of these genes can lead to hepatosteatosis, severe inflammation, and liver damage. We utilized the liver HepG2 immortalized cell line as a model to examine the effects of LXR ligands on these two genes. While we observed very efficacious agonist activity for T0901317 (1) in terms of induction of *FASN*, compounds **6a-b** and **8a-b**

Table 3. LXR activity of 1,2,4-triazole derivatives 8a-e and 9a-b

unexpectedly decreased FASN expression even though they behaved as agonists in both LXR α and LXR β cotransfection assays. SREBP-1c is another gene that is involved in lipogenesis and T0901317 (1) increased expression of this gene almost 9-fold (Table 4). Compounds **6a-b** and **8a-b** that acted as agonists in the cotransfection assays and inverse agonists in terms of *FASN* expression had no effect on expression of the well characterized LXR target gene *SREBP-1c*. These data suggest that the 1,2,4triazole compounds are a novel class of LXR modulators that display context dependent agonism/inverse agonism.

We calculated a set of physical descriptors and pharmaceutical properties of the selected compounds using Qikprop²³ to predict their absorption, distribution, metabolism, and excretion (ADME) properties (Table 5). All compounds comply with Lipniski's rule of five²⁴ and Jorgensen's rule of three²⁵ and are predicted to have good oral bioavailability and excellent absorption through cell membranes.

In conclusion, we have developed novel and potent LXR modulators based on 1,2,4-triazole scaffolds. Several of our new agonists exhibit good isoform selectivity in a cell based cotransfection assay where they generally behave as agonists. Interestingly, these compounds display "modulator" pharmacology since they function as inverse agonists in terms of FASN expression and are inactive on SREBP-1c expression in contrast to the activity observed in cotransfection assays. These new modulators possess excellent ADME properties and can be developed further to examine their unique modulator properties in vivo.

Acknowledgments

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				LX	Rα	LX	KRβ		
Comp.	\mathbb{R}^1	R ²	R ³	EC ₅₀ (nM)	E_{max} (%)	EC ₅₀ (nM)	$E_{max}(\%)$	EC ₅₀ α/β	ClogP
8a -	Ph-	Ph-	CH ₃ CO-	396	56	232	63	1.71	2.84
8b	Ph-	4-Br-Ph-	CH ₃ CO-	113	58	231	78	0.49	3.43
8c	Ph-	4-CH ₃ -Ph-	CH ₃ CO-	228	53	351	62	0.65	3.15
8d	Ph-	3-CH ₃ O-Ph-	CH ₃ CO-	115	62	109	78	1.06	2.91
Be	Ph-	PhCH ₂ -	CH ₃ CO-	701	50	1425	68	0.49	3.20
9a	Ph-	Ph-	4-CH ₃ -C ₆ H ₄ -SO ₂ -	1705	75	259	81	6.58	3.66
9b	Ph-	3-CH ₃ O-Ph-	4-CH ₃ -C ₆ H ₄ -SO ₂ -	ia	-	ia	-	-	3.73

ia = inactive at $10 \,\mu M$

 ${}^{a}E_{max}(\%)$ is the percentage ratio between maximum fold induction of the tested compound and fold induction for 1 at 10 μ M.

Gene	Compound	Normalized mRNA Expression	SEM (+/-)
	T0901317 (1)	2.6210	0.630
FASN	6a	0.7264	0.156
FASIN	6b	0.6393	0.390
	8a 0.6517 0.075 8b 0.4448 0.091	0.075	
	8b	0.4448	0.091
	T0901317 (1)	8.950	0.215
SREBP-1c	6a	0.988	0.190
SKEDP-IC	6b	0.865	0.157
	8a	0.890	0.091
	8b	0.893	0.715

Table 4. Normalized mRNA expression of *FASN* and *SREBP-1c* with **6a-b**, **8a-b**.

 Table 5. Calculated Molecular Descriptors for Prediction of ADME properties for selected compounds. Recommended values or range for 95% of known drugs is shown in parenthesis.

Comp.	^a mol_MW	^b QPlogPo/w	۴HBD	dHBA	°% Human Oral	^f PSA	^g QplogS	hQPPCaco	ⁱ #metab	^j QplogBB	^k QPPMDCK
		(-2 – 6)	(0 - 6)		Absorption				(2 - 8)	(-3 – 1.2)	

	(130 -			(2-	(<25 % Low, >	(7 –	(-6.5 -	(< 25 Poor,			(< 25 Poor, >
	725)			20)	80% High)	200)	.0.5)	> 500			500
								Excellent)			Excellent)
6a	221.26	3.4	1	2.5	100	39.2	-4.012	2905.99	0	0.059	1567.17
6g	251.29	3.24	1	3.25	100	47.5	-4.27	2897.74	1	-0.016	1562.36
8a	263.298	2.84	0	5	100	53.2	-3.81	2973.15	0	0.042	1606.355
8b	342.194	3.428	0	5	100	53.1	-4.69	2973.84	0	0.216	4258.532
8d	293.324	2.9	0	5.7	100	61.1	-3.654	3028.45	1	-0.024	1638.673

^aMolar weight in Daltons; ^bLogarithm of partitioning coefficient between *n*-octanol and water phases (range for 95% of drugs: -2 to 6); ^cEstimated number of hydrogen bonds that would be donated by the solute to water molecules in an aqueous solution. Values are averages taken over a number of configurations, so they can be non-integer; ^d Estimated number of hydrogen bonds that would be accepted by the solute from water molecules in an aqueous solution. Values are averages taken over a number of configurations, so they can be non-integer; ^ePredicted human oral absorption on a 0 -100 % scale, based on a multiple linear regression model; ^fPolar surface area; ^gPredicted aqueous solubility, log S. S in mol dm⁻³; ^hPredicted apparent Caco-2 cell permeability in nm/sec as a model for the gut-blood barrier. ⁱNumber of possible metabolic reactions; ^jPredicted brain/blood partition coefficient; ^kPredicted apparent MDCK cell permeability in nm/sec as a mimic for blood/ brain barrier. Qikrop predictions are for non-active transport.

References and notes

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Supplementary Material

Supplementary material that may be helpful in the review process should be prepared and provided as a separate electronic file. That file can then be transformed into PDF format and submitted along with the manuscript and graphic files to the appropriate editorial office.