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α,β -Dihydroxy Ketones

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SYNTHESIS OF α-DIKETONES FROM α,β-DIHYDROXY KETONES

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ABSTRACT: Treatment of α , β -dihydroxy ketones with carbonyldiimidazole resulted in the formation of α -diketones based on the elimination of the cyclic carbonates formed *in situ*.

 α -Diketones 2 have attracted considerable interest because of their utility as synthetic intermediates¹ to undergo a variety of reactions, for example, the benzilbenzilic acid rearrangement² and dioxaphospholene formation,³ and etc., Several methods for preparation of α -diketones have been reported which include oxidation of acetylenes,⁴ selenium dioxide oxidation of various monoketones,⁵ acyloin condensation of esters followed by subsequent oxidation,⁶ acylation of acyl anions or acyl anion equivalents,⁷ coupling of acyl chlorides mediated by SmI₂ and SmCp₂,⁸ and others.⁹ Here we wish to report a convenient one-pot procedure for the synthesis of α -diketones 2 from α , β -dihydroxy ketones 1 based on the elimination of the carbonates 3 formed *in situ* by treating with carbonyldiimidazole (Scheme 1).



Scheme 1

The results are summarized in Table 1. Treatment of 1a with carbonyldiimidazole (1.5 equiv) in CH_2Cl_2 at room temperature for 0.5 h gave a mixture of the diketone 2a and the cyclic carbonate 3a by checking tlc, which was subjected to prolonged stirring for 2 h to afford the diketone 2a as the only isolated product in 89% yield (entry 1). As indirect evidence for the formation of the cyclic carbonate 3a in the above reaction, reaction of 1a with triphosgene (0.5 equiv)¹⁰ in the presence of 6 equiv of pyridine at -70°C rt gave pure cyclic carbonate 3a in 80% yield, which was treated with 2 equiv of imidazole in CH_2Cl_2 to afford the diketone 2a (Scheme 2).



This methodology was also applied to prepare α -diketones 2b-c, which is shown in Table 1 (entries 2 and 3). For the *threo*-dihydroxy ketone 1d, reaction with carbonyldiimidazole at room temperature for 10 min gave the cyclic carbonate 3d (checked by tlc) and a mixture of the carbonate and the diketone 2d after stirring for 2 h, which without separation was refluxed for 24 h to afford 2d (entry 4). For the cyclic dihydroxy ketone 1e, stirring for 0.5 h provided the cyclic carbonate 3e and stirring for 5 h afforded the diketone 2e¹¹ (entry 5), which was an equilibrium mixture of keto and enol form.

In summary, α -diketones were synthesized by mild and one-pot dehydration of α , β -dihydroxy ketones.



Table 1. Systhesis of α -Diketones from α , β -Dihydroxy Ketones

EXPERIMENTAL

5-Benzyloxy-2,3-pentanedione (2a): General procedures:

To a stirred solution of CO(Im)₂ (290 mg, 1.8 mmol) in dry CH₂Cl₂ (3 ml) was added the diol 1a (270 mg, 1.2 mmol) in dry CH₂Cl₂ (1.5 ml) at room temperature. The reaction mixture was stirred for 2 h. The mixture was quenched with 5% HCl (10 ml) and then extracted with ether (15 ml) for three times. The organic layer was washed with water three times and dried over anhydrous MgSO₄ and then concentrated *in vacuo*. The crude product was purified by column chromatography on silica gel using EtOAc/hexanes 1 : 3 as eluent to afford 2a (219 mg, 89%). TLC ; SiO₂, EtOAc/hexanes 1 : 3 R_r = 0.70. IR (neat) : 1711 cm⁻¹. ¹H NMR (60 MHz, CDCl₃) δ 2.30 (s, 3H), 3.01 (t, 2H, J = 6Hz), 3.78 (t, 2H, J = 6Hz), 4.52 (s, 2H), 7.15 (s, 5H). GC-MS (HP-1, 0.2mm X 25m, oven temp 200 - 280°C, carrier gas : N₂), retention time: 3.20 min. MS (m/e) 206 (M⁺), 163 (6), 100 (22), 91 (base peak), 77 (11), 65 (16), 55 (12), 43 (39). Anal. Calcd for C₁₂H₁₄O₃ : C, 69.88 ; H, 6.84. Found: C, 70.06 ; H, 6.97.

(3R,4S)-5-Benzyloxy-3,4-dihydroxy-2-pentanone 3,4-cyclic carbonate (3a):

To a stirred solution of triphosgene (200 mg, 0.68 mmol) in dry CH₂Cl₂ (3 ml) was added dropwise to a solution of pyridine (641 mg, 8.10 mmol) and the diol 2a (300 mg, 1.35 mmol) in dry CH₂Cl₂ (5 ml) cooled to -70°C. Once addition was complete the reaction was then allowed to warm room temperature. The reaction mixture was gunched with saturated aqueous ammonium chloride and the aqueous portion was separated and extracted with CH₂Cl₂ (15 ml). The organic layer was washed with 1N HCl, saturated aqueous NaHCO₃, brine, dried over anhydrous MgSO₄ and then concentrated in vacuo. The crude product was purified by column chromatography on silica gel using EtOAc/hexanes 1 : 1 as eluent to afford 3a (270 mg, 80%). TLC; SiO₂, EtOAc/hexanes 1:1 R, = 0.49. IR (neat): 1800, 1717 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ 2.35 (s, 3H), 3.65 (dd, 1H, J = 9, 8 Hz), 3.85 (dd, 1H, J = 9, 3 Hz), 4.6 (m, 2H), 4.79 (m, 1H), 4.86 (m, 1H), 7.3 (m, 5H).GC-MS (HP-1, 0.2mm X 25m, oven temp 220 - 280 °C, carrier gas : N₂), retention time: 8.22 min. MS (m/e) 250 (M⁺), 162 (13), 105 (base peak), 91 (54), 79 (12), 77 (64), 51 (23). Anal. calcd for $C_{13}H_{14}O_5$: C, 61.65; H, 5.73. Found : C, 63.39; H, 5.64.

5-Benzyloxy-2,3-pentanedioneb (2a) :

To a stirried solution of the carbonate **3a** (225 mg. 0.9 mmol) in CH₂Cl₂ (3 ml) was added imidazole (123 mg, 1.8 mmol). and stirred at room temperature for 2 h. The mixture was quenched with 5% HCl (9 ml) and then extracted with ether (15 ml) for three times. The organic layer was washed with water and dried over anhydrous MgSO₄ and then concentrated *in vacuo*. The crude product was purified by column chromatography on silica gel using 1 : 3 as eluent to afford **2a** (152 mg, 82%). TLC; SiO₂, EtOAc/hexanes 1 : 3 R_f = 0.70, IR (neat) : 1711 cm⁻¹. ¹H NMR (60 MHz, CDCl₃) δ 2.31 (s, 3H), 3.01 (t, 2H, J = 6 Hz), 3.78 (t, 2H, J = 6 Hz), 4.52 (s, 2H), 7.15 (s, 5H).

1,3-Diphenyl-1,2-propanedione (2b):

TLC; SiO₂, EtOAc/hexanes 1 : 3 R_f = 0.56. IR (neat) : 1714, 1672, 1667 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ 4.20 (s, 2H), 7.20 - 7.89 (m, 10H).

4-Benzyloxy-1-phenyl-1,2-butanedione (2c) :

TLC; SiO₂, EtOAc/hexanes 1; 1 R_f = 0.79. IR (neat): 3400, 1720, 1707 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ 3.16 (t, 2H, J = 6 Hz), 3.88 (t, 2H, J = 6 Hz) 4.46 (s, 2H), 7.29 - 7.85 (m, 10H) MS (m/e) 268 (M⁺), 91 (base peak).

2,3-Heptanedione (2d):

TLC; SiO₂, EtOAc/hexanes $R_f = 0.71$. IR (neat) : 1714 cm⁻¹. ¹H NMR (60 MHz, CDCl₃) δ 0.95 (t, 3H), 1.65 (m, 4H), 2.32 (s, 3H), 2.7 (t, 2H).

1,2-Cyclohexanedione(2e):

TLC; SiO₂, EtOAc $R_r = 0.71$. IR (neat) : 3380, 1720, 1660 cm⁻¹. ¹H NMR (60 MHz, CDCl₃) δ 2.1 (m, 3H), 2.45 (m, 4H), 6.18 (t, 1H).

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