

CLEAVAGE REACTIONS OF THE AZIRIDINE RING

III. Reaction of N-Arylethylenimines with Three-Membered Heterocycles*

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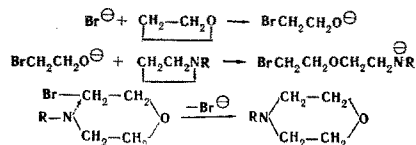
N-Arylethylenimines react with ethylene oxide in the presence of tetraethylammonium bromide with the formation of N-arylmorpholines. In a number of cases, N,N'-diarylpiperazines were isolated as by-products. Ethylene sulfide does not react with N-arylethylenimines.

It is known that three-membered heterocyclic compounds — ethylene oxide, ethylene sulfide, and N-substituted ethylenimines — are capable of dimerizing to form dioxane [2], dithiane, and N,N'-disubstituted piperazines [4, 5], respectively.

We assumed that in the presence of tetraethylammonium bromide (TEAB), which causes the opening of an oxide ring [6] and the dimerization of N-phenylethylenimine [1], the reaction of N-arylethylenimines with ethylene oxide (sulfide) would lead to the formation of N-arylmorpholines (triamorpholines).

In actual fact, when N-arylethylenimines were heated with ethylene oxide at 150°C the expected compounds were isolated in high yields. To confirm their structure, analogous substances were obtained by the dehydration of bis(β-hydroxyethyl)arylamines with concentrated sulfuric acid [7]. The constants of the compounds synthesized by the two methods and their IR spectra were identical.

The following scheme of the reaction is proposed:



In order to suppress the possible dimerization of the ethylenimines, an excess of ethylene oxide was used. However, in the case of N-o-tolyl- and N-o-methoxyphenylethylenimines the corresponding piperazines were obtained with a yield of about 30%. It might be assumed that the morpholines are obtained through a stage of the formation of piperazines with their subsequent reaction with ethylene oxide. However, N,N'-diphenylpiperazine does not react with ethylene oxide even at 200°C.

The reaction of N-phenylethylenimine with ethylene sulfide in the presence of TEAB did not lead to the expected thiamorpholine because of the polymerization of the ethylene sulfide.

EXPERIMENTAL

The N-arylethylenimines were obtained by the dehydrochlorination of N-(β-chloroethyl)arylamine hydrochlorides by analogy with the dehydrobromination of their hydrobromides [8]. The β-chloroethylamine hydrochlorides (Table 1) were synthesized in a similar manner to β-chloroethylaniline hydrochloride [9] by treating the corresponding β-hydroxyethylarylamines with phosphorus oxychloride [10].

N-o-Methoxyphenylethylenimine: Bp 77–80°C (5 mm); n_D^{20} 1.5625. Found %: C 72.20; 72.33; H 7.36, 7.37; N 9.59, 9.63; mol. wt. 150; 148. $\text{C}_9\text{H}_{11}\text{NO}$. Calculated %: C 72.48; H 7.31; N 9.39; mol. wt. 149.

N-p-Methoxyphenylethylenimine: Bp 88–90°C (4 mm); n_D^{20} 1.5540. Found %: C 72.40, 72.57; H 7.41, 7.35; N 9.37, 9.43; mol. wt. 147; 148. $\text{C}_9\text{H}_{11}\text{NO}$. Calculated %: C 72.48; H 7.37; N 9.39; mol. wt. 149.

*For part II, see [1].

Table 1. $\text{RC}_6\text{H}_4\text{NHCH}_2\text{CH}_2\text{Cl} \cdot \text{HCl}$

R	Mp, °C	Empirical formula	Equiv.	
			found:	calculated
H	154—155 ⁹	$\text{C}_8\text{H}_{10}\text{ClN} \cdot \text{HCl}$	193.5; 183	182
<i>o</i> -CH ₃	166—168	$\text{C}_9\text{H}_{12}\text{ClN} \cdot \text{HCl}$	205; 206	206
<i>p</i> -CH ₃	161—163	$\text{C}_9\text{H}_{12}\text{ClN} \cdot \text{HCl}$	204; 205	206
<i>o</i> -CH ₃ O	163—164	$\text{C}_9\text{H}_{12}\text{ClNO} \cdot \text{HCl}$	221; 220	221
<i>p</i> -CH ₃ O	150—151	$\text{C}_9\text{H}_{12}\text{ClNO} \cdot \text{HCl}$	220; 219	221

Table 2. N-arylmorpholines

Aryl	Bp, °C (pressure, mm)	Mp, °C	Empirical formula	Found				Calculated				Yield, %
				C, %	H, %	N, %	M	C, %	H, %	N, %	Mol. wt.	
C ₆ H ₅	96—98 (2)	53 [7]	C ₁₀ H ₁₃ NO	73.45 73.53	8.13 8.20	8.70 8.58	163 162	73.62	7.97	8.58	163	94
<i>o</i> -CH ₃ C ₆ H ₄	102—104 (4) ^a	—	C ₁₁ H ₁₅ NO	74.47 74.52	8.38 8.51	7.55 7.20	179 180	74.50	8.47	7.89	177	65
<i>p</i> -CH ₃ C ₆ H ₄	108—110 (4)	42	C ₁₁ H ₁₅ NO	74.25 74.37	8.33 8.46	7.70 7.67	176 180	74.50	8.47	7.89	177	85
<i>o</i> -CH ₃ OC ₆ H ₄	114—116 (2) ^b	—	C ₁₁ H ₁₅ NO ₂	68.30 68.59	7.75 7.62	7.04 6.90	196 192	68.40	7.77	7.25	193	70
<i>p</i> -CH ₃ OC ₆ H ₄	125—126 (2)	67	C ₁₁ H ₁₅ NO ₂	68.35 68.41	7.72 7.68	7.11 7.22	196 195	68.40	7.77	7.25	193	90

^a n_D^{20} 1.5435. ^b n_D^{20} 1.5560.

Reaction of N-arylethylenimines with ethylene oxide. A mixture of 0.02 mole of a N-arylethylenimine, 0.08 mole of ethylene oxide, and 0.01 g of TEAB was heated in a tube at 140–150°C for 40–50 hr. After the elimination of the excess of ethylene oxide, the reaction products were fractionated in vacuum. The constants of the N-arylmorpholines are given in Table 2.

In the cases of N-*o*-tolylethylenimine and N-*o*-methoxyphenylethylenimine, N,N'-ditolyl- and N,N'-dimethoxyphenylpiperazines were isolated as by-products with yields of 25–30%.

N,N'-Di-*o*-tolylpiperazine: mp 170–171°C (from acetone). Found %: C 81.06, 81.37; H 8.29, 8.29; N 10.20, 10.39. C₁₈H₂₂N₂. Calculated %: C 81.20; H 8.27; N 10.52.

N,N'-Di-*o*-methoxyphenylpiperazine: mp 173–174°C (from acetone). Found %: C 73.04, 73.00; H 7.55, 7.40; N 9.75, 9.83. C₁₈H₂₂N₂O₂. Calculated %: C 72.48; H 7.39; N 9.39.

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