

REGIO- AND STEREODEFINED SYNTHESIS OF TRIMETHYLSILYL ENOL ETHERS  
RESULTED FROM THE ISOMERIZATION OF  $\alpha$ -TRIMETHYLSILYL KETONES

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**Summary:** The isomerization of an  $\alpha$ -trimethylsilyl ketone is lead to the corresponding trimethylsilyl enol ether with the enhanced regioselectivity by heating or by the assist of trimethylsilyl trifluoromethanesulfonate. The thermal reaction discloses a new regiodefined (E)-selective route to silyl enol ethers.

Silyl enol ethers have been employed as one of the most promising nucleophilic substrates resulting carbon carbon bond formation or functionalization onto  $\alpha$ -carbons of carbonyl compounds in organic synthesis.<sup>1</sup> Since the conventional method for their preparation includes the silylation of preformed enolate anion, it is difficult to differentiate two  $\alpha$ -positions of an unsymmetrically substituted ketone except the case of which the appreciable energy difference is present between two enolate forms.<sup>2</sup> Stereochemical control forming enolate anion is also pursued related to stereoselective aldol reactions.<sup>3</sup> Few methods have succeeded to control the stereochemistry of enolate anions, (Z)- or (E)-selectively in the case of freed from regiochemistry<sup>4</sup>.

Recently, the reliable regioselection of an unsymmetrical ketone have been attained by two indirect methods depending on the isomerization of  $\alpha$ -trimethylsilyl-<sup>5</sup> or  $\beta$ -trimethylsilylallyl alcohols.<sup>6</sup> The former method gave 2 (Z)-selectively as the result of the intermediacy of equilibrated homoenolate anion. On the contrary, the latter one suggested (E)-selective formation of 2 as the result of consecutive isomerization of isolable  $\alpha$ -trimethylsilyl ketone. Therefore, the isolated  $\alpha$ -trimethylsilyl ketone would promise the regio- and stereocontrol in the formation of trimethylsilyl enol ethers. We describe herein a successful method to control the regio- and stereochemistry of trimethylsilyl enol ethers in unsymmetrical substituted ketones.

Since the regiodefined synthesis of aliphatic  $\alpha$ -trimethylsilyl ketones was attained by the rhodium catalyzed methods under the nonaqueous

Table 1. Results of the isomerization of  $\alpha$ -trimethylsilyl ketones.

Entry	$\alpha$ -Silyl ketone (1)		Silyl enol ether (2)	Method <sup>i</sup>	Yield <sup>ii</sup> (%)	E/Z <sup>iv</sup>
1		a		A	56 <sup>iii</sup>	-
2				C	92(1)	-
3		b		A	100(1)	93/7
4				C	86(1)	11/89
5		c		A	96(0.5)	93/7
6				B	97(2)	75/25
7				C	95(0.5)	14/86
8				D	90(1)	25/75
9		d		A	98	96/4 <sup>v</sup>
10				B	90	78/22 <sup>v</sup>
11				C	95(2)	28/72 <sup>v</sup>
12				D	96	18/82 <sup>v</sup>
13		e		A	97(0.5)	93/7
14				C	89	19/81
15		f		A	98	95/5 <sup>v</sup>
16				C	86	21/79 <sup>v</sup>
17		g		A	97	98/2
18		h		A	94	94/6
19		i		A	96	94/6

i, A; neat/175°C/15 min. B; 5 mol% HRh(CO)(PPh<sub>3</sub>)<sub>3</sub>/C<sub>6</sub>H<sub>6</sub>/105°C/12h.

C; 10 mol% Me<sub>3</sub>SiOTf/(Me<sub>3</sub>Si)<sub>2</sub>NH/CH<sub>2</sub>Cl<sub>2</sub>/0°C/5 min.

D; 5 mol% Me<sub>3</sub>SiI/HgI<sub>2</sub>/(Me<sub>3</sub>Si)<sub>2</sub>NH/CH<sub>2</sub>Cl<sub>2</sub>/r.t./15 h.

ii, Distilled yield. The content of the regioisomers is shown in parentheses.

iii, 35% of 1a was recovered.

iv, The ratio was determined by GLC analyses (OV 101, 50m column).

v, The ratio was determined by GLC analyses (PEG-HT Bonded, 25m column).

conditions<sup>6b, 7</sup> a series of silylated nonanones (1a - 1f) was designed and isolated in order to evaluate the regio- and stereoselectivity of the resultant silyl enol ethers 2. Results are summarized in Table 1.

First of all 3-nonanone derivatives 1c and 1d were checked under the following conditions; heat (method A),  $\text{HRh}(\text{CO})(\text{PPh}_3)_3$  catalysis (method B),  $\text{Me}_3\text{SiOTf}$  catalysis (method C),<sup>8, 9</sup> and  $\text{Me}_3\text{SiI}$  catalysis (method D).<sup>8</sup> The reaction of 1c and 1d proceeded smoothly to give 2c and 2d, respectively under all conditions. The  $\pi$ -bond location of the resultant 2c and 2d is strictly (more than 98% selectivity<sup>10</sup>) directed to the original position of trimethylsilyl group in 1. Different from the regiochemistry, the stereoselectivity depends on the reaction conditions. (E)-Form of 2c and 2d was obtained selectively in method A or B (entry 5, 6, 9, and 10), whereas (Z)-form was given predominantly in method C or D (entry 7, 8, 11, and 12). Based on the above results, methods A and C were applied to the other  $\alpha$ -silyl ketones as the standard conditions.

An excellent (E)-selective formation of 2 is observed in the thermal isomerization of 1 except 1a. It is noteworthy that such a high (E)-selectivity can be attained even in the formation of 2g, 2h, and 2i where the selection of (E)-form is relatively difficult in the conventional method. The present method is sharply contrasted with Kuwajima's work<sup>5</sup> which brought about 2, (Z)-selectively and discloses a novel route to the (E)-selective formation of 2. The high (E)-selectivity can be rationalized by the concerted intramolecular migration of the trimethylsilyl group via the four membered transition state which was proposed in the thermal isomerization of silylmethyl ketones.<sup>12</sup>

On the other hand,  $\text{Me}_3\text{SiOTf}$  catalyzed isomerization of 1 gives predominantly (Z)-2. The reverse selectivity in method C is pertinent to the intermolecular migration of the trimethylsilyl group mediated by the catalyst.<sup>13</sup> Thus, the regio- and stereoselection can be attained uniformly by this approach in the formation of trimethylsilyl enol ethers derived from unsymmetrically substituted ketones. Although (Z)-selectivity in 2 is rather moderate in the present time, it is important to make possible the choice of the stereochemical course from the identical starting materials.

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10. The regio- and stereochemical purity was evaluated by GLC analyses. Each isomer of 2 was identified by the comparison of  $^1\text{H}$  and  $^{13}\text{C}$  n.m.r. spectra. The diagnosis of stereoisomers in 2 is based on the chemical shift of the allyl carbon which is located in  $\beta$ -position of the trimethylsiloxy group. The signal of (Z)-isomers was shown in 5 - 6 ppm lower field than (E)-isomers.<sup>11</sup>
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