

1,4- and 1,7-Addition Reactions of 4-(Substituted benzylidene)-3,5-dimethylisopyrazoles¹⁾

TAKUSHI KURIHARA, YASUHIKO SAKAMOTO, TOSHIKO SAKAGUCHI,
and HIROSHI HIRANO

Osaka College of Pharmacy²⁾

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Reactions of 4-(substituted benzylidene)-3,5-dimethylisopyrazoles (**1a—1c**) with acetic anhydride, hydrochloric acid in methanol, bromine in acetic acid, dimethyl sulfate, and acyl chlorides (acetyl chloride, benzoyl chloride, ethyl chloroformate, and *p*-toluenesulfonyl chloride) in pyridine gave 1,4-addition products (**2—10**) in fairly good yield. On the other hand, 4-(*o*-nitrobenzylidene)isopyrazole (**1a**) was converted to 5-chloro-3-(1-substituted 3,5-dimethylpyrazolyl)anthranils (**13**, **14**, and **15**) by treatment with acetyl chloride, benzoyl chloride, or ethyl chloroformate without pyridine.

Keywords—isopyrazole; pyrazole; 1,4-addition; 1,7-addition; betaine; anthranil

In contrast to pyrazole chemistry, the reactivity of isopyrazoles has not been investigated extensively. Recently we have reported the 1,4-addition reaction of 4-(*m*-nitrobenzylidene)-3,5-dimethylisopyrazole (**1b**)³⁾ and the reaction of 4-(*o*-nitrobenzylidene)-3,5-dimethylisopyrazole (**1a**) with acyl chlorides⁴⁾ in brief communications. The present paper describes further reactions of isopyrazoles (**1a—1c**), which exist in a betaine form,⁵⁾ with a full account of the previous communications.^{3,4)}

Treatment of **1b** with excess acetic anhydride at 50° for 10 hr gave 1-acetyl-4-(α -acetoxy-*m*-nitrobenzyl)-3,5-dimethylpyrazole (**2b**) in 92% yield, whose structure was supported by its analytical and spectral data [IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 1735, 1730; NMR (CDCl₃) δ : 2.10 and 2.21 (each 3H, each s, C₃- and C₅-CH₃), 2.60 and 2.63 (each 3H, each s, 2 $\times$ COCH₃); mass spectrum (MS) m/e : 331 (M⁺)]. Similarly, reactions of **1a** and **1c** with acetic anhydride under the same condition gave the diacetates (**2a** and **2c**). Their analytical and spectral data are summarized in Table I.

Reaction of **1a** to **1c** with a catalytic amount of hydrochloric acid in methanol gave 4-(α -methoxybenzyl)-3,5-dimethylpyrazoles (**3a—3c**) in 85—90% yield. These are presumably obtained by 1,4-addition of hydrochloric acid to isopyrazoles, followed by substitution of the chloro group with methanol. Their analytical and spectral data are summarized in Table II.

Brominations of **1a** to **1c** in acetic acid with an equivalent amount of bromine at 35° gave a mixture of benzaldehyde derivatives (**4a—4c**) in 40—45% yield and 4-bromo-3,5-dimethylpyrazole⁶⁾ (**5**) in 50—55% yield. The mechanism of the formation of aldehydes and bromopyrazole seemed to start with 1,4-addition of acetic acid to isopyrazoles, accompanied by bromination as shown in Chart 1.

It is interesting to note that treatment of **1b** in absolute toluene with dimethyl sulfate under reflux in less than 0.5 hr gave 4-(α -methoxy-*m*-nitrobenzyl)-1,3,5-trimethylpyrazole (**6**)

1) A part of this work was presented at the 25th Meeting of Kinki Local Meeting of the Pharmaceutical Society of Japan, Kobe, November 1975.

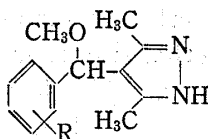
2) Location: Kawai, 2-10-65, Matsubara, Osaka 580, Japan.

3) T. Kurihara, E. Araya, and T. Sakaguchi, *Heterocycles*, **3**, 543 (1975).

4) T. Kurihara, T. Sakaguchi, and H. Hirano, *Chem. Pharm. Bull.* (Tokyo), **24**, 1106 (1976).

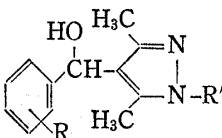
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TABLE II. 4-(α -Methoxybenzyl)-3,5-dimethylpyrazoles

Compd. No.	Yield (%)	mp (°C) (Recryst. solvent)	Formula	Analysis (%)			IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm ⁻¹ (NH)	NMR (CDCl ₃) δ	
				Calcd. (Found)	C	H	N	-OCH ₃	-CH
3a	90	54—55 (L)	C ₁₃ H ₁₅ N ₃ O ₃	59.75 (59.77)	5.79 5.54	16.08 16.15	3460	3.33	5.95
3b	86	139—140 (L)	C ₁₃ H ₁₅ N ₃ O ₃	59.75 (59.66)	5.79 5.80	16.08 16.10	3460	3.37	4.05
3c	90	112—113 (PE)	C ₁₃ H ₁₅ ClN ₂ O	62.27 (62.14)	6.03 6.14	11.17 11.30	3460	3.37	5.50

Solvent: L=ligroin, PE=petroleum ether.

TABLE III. 1-Substituted 4-(α -Hydroxy)-3,5-dimethylpyrazoles

Compd. No.	R'	Yield (%)	mp (°C) (Recryst. solvent)	Formula	Analysis (%)			IR $\nu_{\text{max}}^{\text{KBr}}$ cm ⁻¹		NMR (CDCl ₃) δ -CH
					Calcd. (Found)	C	H	N	(OH) (C=O)	
7a	COCH ₃	72	152—153 (B-L)	C ₁₄ H ₁₅ N ₃ O ₄	58.12 (57.91)	5.23 5.22	14.53 14.26	3400	1740	6.45
7b	COCH ₃	78	144—145 (M)	C ₁₄ H ₁₅ N ₃ O ₄	58.12 (58.09)	5.23 5.25	14.53 14.79	3400	1740	5.95
7c	COCH ₃	81	Oil ^c					3600 ^a	1735 ^a	6.05
8a	COC ₆ H ₅	65	Oil ^d					3650 ^a	1710 ^a	6.45
8b	COC ₆ H ₅	80	147—148 (B-PE)	C ₁₉ H ₁₇ N ₃ O ₄	64.95 (65.22)	4.88 4.87	11.96 11.98	3400	1715	5.90
8c	COC ₆ H ₅	75	Oil ^c					3650 ^a	1710 ^a	6.05
9a	CO ₂ C ₂ H ₅	88	123—124 (B)	C ₁₅ H ₁₇ N ₃ O ₅	56.42 (56.68)	5.37 5.35	13.16 12.93	3400	1760	6.45
9b	CO ₂ C ₂ H ₅	72	114—115 (B-L)	C ₁₅ H ₁₇ N ₃ O ₅	56.42 (56.67)	5.37 5.17	13.16 13.33	3350	1760	5.95
9c	CO ₂ C ₂ H ₅	75	Oil ^d					3600	1755	6.15
10a	SO ₂ C ₆ H ₅ CH ₃	65	171—172 (M)	C ₁₉ H ₁₉ N ₃ O ₅ S	56.86 (56.96)	4.77 4.61	10.46 10.23	3350	1380, 1190(SO ₂)	6.15 ^b
10b	SO ₂ C ₆ H ₅ CH ₃	60	188—190 (M)	C ₁₉ H ₁₉ N ₃ O ₅ S	56.86 (56.64)	4.77 4.49	10.46 10.49	3350	1390, 1190(SO ₂)	5.90 ^b
10c	SO ₂ C ₆ H ₅ CH ₃	58	153—154 (M)	C ₁₉ H ₁₉ ClN ₂ O ₅ S	58.38 (58.59)	4.89 4.99	7.16 7.45	3350	1395, 1190(SO ₂)	5.75 ^b

Solvent B=benzene, L=ligroin, M=methanol, PE=petroleum ether.

^a) In CHCl₃ solution.^b) In DMSO-*d*₆.^c) 7c and 8c were analysed by leading to crystalline derivatives.^d) 8a and 9c failed to form crystalline derivatives, and these were determined by mass spectrum [8a: *m/e* 351 (M⁺), 9c: *m/e* 308 (M⁺)].

in 25% yield, which was identified with an authentic sample⁷⁾ obtained by the reaction of *m*-nitrobenzylideneacetylacetone with methylhydrazine hydrochloride in methanol. However similar reactions of **1a** and **1c** with dimethyl sulfate failed.

Reaction of **1a** with a slight excess of acetyl chloride in pyridine at 50° for 10 hr, followed by treatment with ice-water gave 1-acetyl-4-(α -hydroxy-*o*-nitrobenzyl)-3,5-dimethylpyrazole (**7a**) in 72% yield, whose structure was determined by derivation to its diacetate (**2a**) by treatment with acetic anhydride. Similarly, reactions of **1a** to **1c** with acetyl chloride, benzoyl chloride, ethyl chloroformate, and *p*-toluenesulfonyl chloride (tosyl chloride) in pyridine gave the corresponding 1-acetyl (**7b**, **7c**), 1-benzoyl (**8a**—**8c**), 1-ethoxycarbonyl (**9a**—**9c**), and 1-tosyl (**10a**—**10c**) derivatives, respectively, in a fairly good yield. Their analytical and spectral data are summarized in Table III.

Introduction of a hydroxyl group can be reasonably explained by postulating the pyridinium chloride (**11**) as an intermediate because 1-acetyl-4-(α -chlorobenzyl)-3,5-dimethylpyrazole (**12**) was isolated by treatment of **1b** with a large excess of acetyl chloride without pyridine.

On the other hand, reaction of **1a** with acetyl chloride without pyridine gave an entirely different result. Treatment of **1a** with a large excess of acetyl chloride at 50° for 10 hr resulted in isolation of crystalline **13** of mp 130—131° as a sole product, whose structure was assigned as 5-chloro-3-(1'-acetyl-3,5-dimethylpyrazolyl)anthranil from analytical and spectral data [IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 1740 (CO), no NO₂ group; UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 261 (3.86), 342 (4.06); NMR (DMSO-*d*₆) δ : 2.42 and 2.72 (each 3H, each s, CH₃), 2.80 (3H, s, COCH₃), 7.20—7.75 (3H, m, aromatic H)]. Hydrolysis of **13** with ethanolic sodium hydroxide gave 5-chloro-3-(3,5-dimethylpyrazolyl)anthranil (**16**) in a good yield.

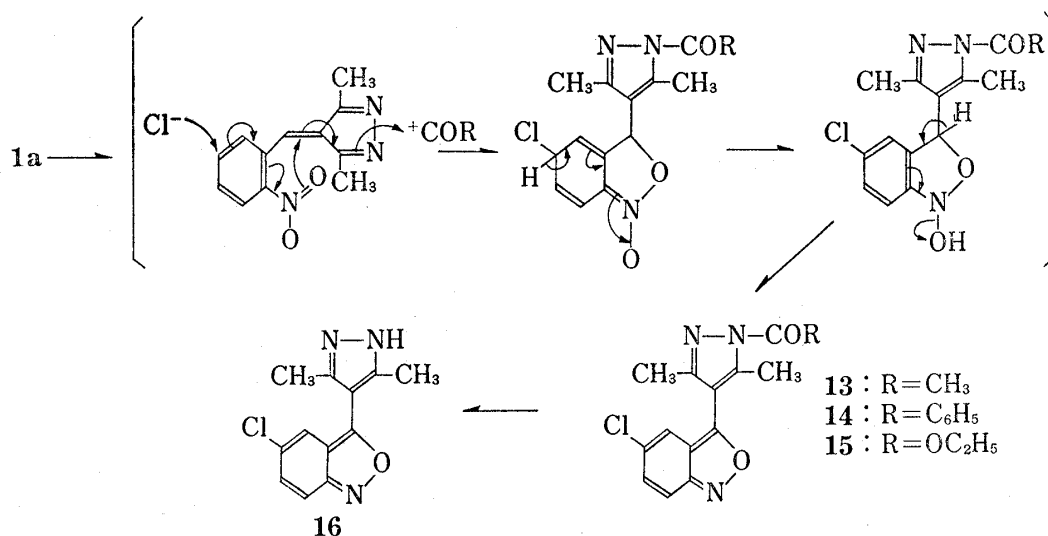


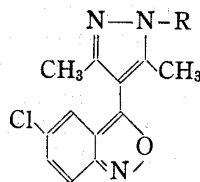
Chart 2

On the basis of the Dickenson's report⁸⁾ that *o*-nitrobenzhydrol cyclizes with introduction of a chlorine atom upon heating with thionyl chloride in chloroform to give 5-chloro-3-phenylanthranil, the mechanism of the formation of **13** was considered as the attack of nitro-oxygen on the benzyl-carbon of isopyrazole with the 1,7-addition of acetyl chloride, followed by prototropy and dehydration, resulting in the formation of the anthranil. Similarly, reaction of **1a** with benzoyl chloride and ethyl chloroformate gave 5-chloro-3-(1-benzoyl-3,5-dimethylpyrazolyl)anthranil (**14**) and 5-chloro-3-(1-ethoxycarbonyl-3,5-dimethylpyrazolyl)anthranil (**15**), respectively. Analytical and spectral data are summarized in Table IV.

7) T. Kurihara, T. Sakaguchi, and H. Hirano, *Heterocycles*, **3**, 633 (1975).

8) W.B. Dickenson, *J. Am. Chem. Soc.*, **86**, 3580 (1964).

TABLE IV. 5-Chloro-3-(1-substituted 3,5-dimethylpyrazolyl)anthranils



Compd. No.	R	Yield (%)	mp (°C) (Recryst. Solvent)	Formula	Analysis (%)			IR $\nu_{\text{max}}^{\text{KBr}}$ cm ⁻¹ (C=O)	UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ)
					Calcd. (Found)	C	H	N	
13	COCH ₃	92	130—131 (MeOH)	C ₁₄ H ₁₂ ClN ₃ O ₂	58.03 (58.33)	4.17 4.36	14.50 14.39	1740	261(3.86), 342(4.06)
14	COC ₆ H ₅	58	155—156 (EtOH)	C ₁₉ H ₁₄ ClN ₃ O ₂	64.86 (65.07)	4.01 4.13	11.94 11.91	1700	248(4.07), 342(4.07)
15	CO ₂ C ₂ H ₅	87	103—104 (Petr. ether)	C ₁₅ H ₁₄ ClN ₃ O ₃	56.36 (56.61)	4.41 4.45	13.14 13.06	1760	253(3.92), 337(4.00)

Experimental

Melting and sublimating points are uncorrected. Infrared (IR) spectra were determined on a JASCO Model IRA-1 and ultraviolet (UV) spectra on a Shimadzu UV-200 spectrophotometer. Nuclear magnetic resonance (NMR) spectra were determined, using tetramethyl silane as the internal standard, with a Hitachi R-24A spectrometer, and mass spectra on a Hitachi RMU-7L.

1-Acetyl-4-(α -acetoxybenzyl)-3,5-dimethylpyrazoles (2a—2c)—A suspension of isopyrazoles (**1**) (0.01 mol) in Ac₂O (20 ml) was stirred at 50° for 10 hr. After decomposition of Ac₂O with saturated NaHCO₃ solution, the whole mixture was extracted with CHCl₃. The extract was washed with H₂O, dried over anhyd. Na₂SO₄, and the solvent was evaporated to afford a crystalline residue which was recrystallized from a suitable solvent to **2a**, **2b**, and **2c**.

4-(α -Methoxybenzyl)-3,5-dimethylpyrazoles (3a—3c)—A solution of **1** (0.01 mol) and 35% hydrochloric acid (0.5 ml) in MeOH (100 ml) was heated for 2 hr. After evaporation of the solvent, H₂O was added and the whole mixture was extracted with CHCl₃. The extract was washed with saturated NaHCO₃ solution and H₂O, and dried over anhyd. Na₂SO₄. The solvent was evaporated and the crystalline residue was recrystallized from a suitable solvent to **3a**, **3b**, and **3c**.

Bromination of Isopyrazoles (1a—1c)—To a solution of **1** (0.005 mol) in AcOH (25 ml), Br₂ (0.8 g) (0.005 mol), dissolved in AcOH (2 ml) was added in small portions under stirring at 35°. After stirring for 1 hr, AcOH was evaporated *in vacuo* and the viscous oily residue was dissolved in CHCl₃. The CHCl₃ solution was washed with saturated NaHCO₃ solution and H₂O, and dried over anhyd. Na₂SO₄. The solvent was evaporated and the pale yellow oily residue was submitted to an Al₂O₃ column chromatography using benzene as an eluant. From the first fraction were obtained benzaldehydes (**4a**, **4b**, and **4c**) in 40—45% yield, which were identified with each authentic sample by comparison of IR spectra. From the latter fractions was obtained 4-bromo-3,5-dimethylpyrazole (**5**) of mp 121—123° (lit.⁶ 123°) recrystallized from hexane in 50—55% yield.

4-(α -Methoxy-*m*-nitrobenzyl)-1,3,5-trimethylpyrazole (6)—To a suspension of **1b** (1.15 g, 0.005 mol) in absolute toluene (30 ml) Me₂SO₄ (2.52 g, 0.01 mol) was added and the mixture was heated under reflux for 0.5 hr. The solvent was removed by concentration *in vacuo* and the residue was dissolved in CHCl₃. The CHCl₃ solution was washed with 5% NH₄OH and H₂O, and dried over anhyd. Na₂SO₄. The solvent was evaporated and the viscous oily residue was purified by column chromatography over an Al₂O₃. Elution with benzene gave **6** as colorless crystals (0.34 g) (25%). Sublimation at 130° (oil bath)/3 mmHg gave an analytical sample as colorless needles of mp 124—125°. NMR (CDCl₃) δ : 2.11 and 2.15 (each 3H, each s, C₃- and C₅-CH₃), 3.38 (3H, s, OCH₃), and 3.37 (3H, s, NCH₃). MS *m/e*: 275 (M⁺). Anal. Calcd. for C₁₄H₁₇O₃N₃: C, 61.08; H, 6.22; N, 15.26. Found: C, 61.09; H, 6.30; N, 14.92.

1-Substituted 4-(α -Hydroxybenzyl)-3,5-dimethylpyrazoles (7a—7c, 8a—8c, 9a—9c, and 10a—10c)—To a suspension of **1** (0.005 mol) in pyridine (10 ml) acetyl chloride, benzoyl chloride, ethyl chloroformate, or tosyl chloride (0.0055 mol) was added. The reaction mixture was stirred for 10 hr at 50—60°. After evaporation of pyridine *in vacuo*, the residue was triturated with saturated NaHCO₃ solution and extracted with CHCl₃. The extract was washed with cold 5% HCl and H₂O, and dried over anhyd. Na₂SO₄. After evaporation of the solvent, the crystalline residue was recrystallized from a suitable solvent. The oily products were derived to crystalline derivatives and determined as described below.

Acetylation of 7c: A solution of **7c** (278 mg, 1 mmol) in Ac_2O (5 ml) and pyridine (1 drop) was allowed to stand overnight at room temperature. The mixture was poured into ice-water, made alkaline with NaHCO_3 , and extracted with CHCl_3 . The extract was washed with H_2O and dried over anhyd. Na_2SO_4 . The solvent was evaporated and the crystalline residue was recrystallized from petr. ether to colorless needles (75 mg) of mp 112–113°. This was identified with the authentic sample of **2c**.

Benzoylation of 8c: A solution of **8c** (1.7 g, 0.005 mol) in pyridine (10 ml) and BzCl (1.4 g, 0.01 mol) was stirred at 60° for 5 hr. After pyridine was evaporated *in vacuo*, H_2O was added and the whole mixture was extracted with CHCl_3 . The solvent was evaporated and the crystalline residue was recrystallized from benzene-ligroin to 1-benzoyl-4-(α -benzoxy-*o*-chlorobenzyl)-3,5-dimethylpyrazole as colorless needles (0.85 g), mp 115–117°. IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 1717 and 1705. NMR (CDCl_3) δ : 2.20 (3H, s, $\text{C}_3\text{-CH}_3$), 2.75 (3H, s, $\text{C}_5\text{-CH}_3$). Anal. Calcd. for $\text{C}_{26}\text{H}_{21}\text{ClN}_2\text{O}_3$: C, 70.19; H, 4.76; N, 6.30. Found: C, 70.26; H, 4.77; N, 6.54.

1-Acetyl-4-(α -chloro-*m*-nitrobenzyl)-3,5-dimethylpyrazole (12)—To a suspension of **1b** (1.15 g, 0.005 mol) in MeCOCl (30 ml), a drop of pyridine was added and the mixture was heated under stirring at 50° for 10 hr. After evaporation of MeCOCl *in vacuo*, the residue was dissolved in CHCl_3 . CHCl_3 solution was washed with H_2O and saturated NaHCO_3 solution, and dried over anhyd. Na_2SO_4 . Evaporation of the solvent gave a clear viscous oil which showed one spot on thin-layer chromatography (Al_2O_3 /benzene) and a positive Beilstein test. IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm^{-1} : no OH, 1738 (CO). NMR (CDCl_3) δ : 2.05 and 2.55 (each 3H, each s, C_3 - and $\text{C}_5\text{-CH}_3$), 2.65 (3H, s, COCH_3), and 6.20 (1H, s, CH). These data are in fair agreement with the structure of **12**. The oily compound **12** (1 g) dissolved in MeOH (20 ml) was refluxed for 1 hr. After evaporation of the solvent, the residue was dissolved in CHCl_3 . CHCl_3 solution was washed with saturated NaHCO_3 solution and H_2O , and dried over anhyd. Na_2SO_4 . The solvent was evaporated and the crystalline residue was recrystallized from ligroin to colorless needles (0.42 g) of mp 139–140° which was identified with the authentic sample of **3b**.

5-Chloro-3-(1-substituted 3,5-dimethylpyrazolyl)anthranils (13, 14, and 15)—A mixture of **1a** (0.01 mol) and MeCOCl , BzCl , or ethyl chloroformate (30 ml) was heated under stirring at 50–60° for 10 hr. MeCOCl , BzCl , or ethyl chloroformate was removed by evaporation *in vacuo* and the oily residue was triturated with saturated NaHCO_3 solution. The whole mixture was extracted with CHCl_3 , and the extract was washed with H_2O and dried over anhyd. Na_2SO_4 . The solvent was evaporated and the crystalline residue was recrystallized from a suitable solvent. NMR ($\text{DMSO}-d_6$) δ : **13**; 2.42 and 2.72 (each 3H, each s, $2 \times \text{CH}_3$), 2.80 (3H, s, COCH_3), 7.20–7.75 (3H, m, aromatic H). **14**; 2.30 and 2.70 (each 3H, each s, $2 \times \text{CH}_3$), 7.30–8.00 (8H, m, aromatic H). **15**; 1.40 (3H, t, $J=6$ Hz, CH_2CH_3), 2.30 and 2.60 (each 3H, each s, $2 \times \text{CH}_3$), 4.50 (2H, q, $J=6$ Hz, CH_2CH_3), 7.30–7.90 (3H, m, aromatic H).

5-Chloro-3-(3,5-dimethylpyrazolyl)anthranil (16)—A mixture of **13** (293 mg, 1 mmol) and NaOH (44 mg, 1.1 mmol) in EtOH (10 ml) was allowed to stand overnight at room temperature. EtOH was removed by concentration *in vacuo*, and the crystalline residue was recrystallized from MeOH to **16** (205 mg) (82%) as colorless needles, mp 186–187°. IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3200 (NH). UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 248 (4.02), 348 (3.82). NMR ($\text{DMSO}-d_6$) δ : 2.33 (6H, s, $2 \times \text{CH}_3$). Anal. Calcd. for $\text{C}_{12}\text{H}_{10}\text{ClN}_3\text{O}$: C, 58.19; H, 4.07; N, 16.97. Found: C, 58.27; H, 4.05; N, 16.88.

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