

## N-Alkylation of N-trimethylsilyl derivatives of lactams, amides, and imides with alkyl sulfonates\*

A. D. Shagina,<sup>\*</sup> E. P. Kramarova, A. G. Shipov, D. V. Tarasenko, Vad. V. Negrebetsky, and Yu. I. Baukov

Pirogov Russian National Research Medical University,  
1 ul. Ostrovityanova, 117997 Moscow, Russian Federation.  
Fax: (495) 434 0329. E-mail: nasya.shagina@gmail.com

The reaction of *N*-trimethylsilyl derivatives of amides and imides with alkyl sulfonates on heating affords the corresponding *N*-alkyl derivatives and trimethylsilyl sulfonates.

**Key words:** *N*-alkyllactams, *N*-alkylamides, *N*-alkylimides; synthesis, "silyl method" of alkylation.

Methods of alkylation, based on the reaction of amides with various alkylation agents (dialkyl sulfates,<sup>1–5</sup> diazo-alkanes,<sup>6,7</sup> diacetals,<sup>8</sup> alkyl halogenides<sup>9</sup>, etc.) and with variations of conditions, that lead to *N*- or O-alkylation products, are widely used nowadays.

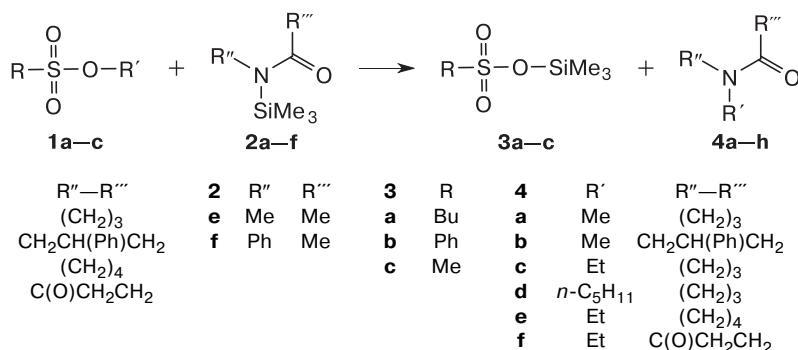
"Silyl method" is one of common and convenient methods for the *N*-alkylation of amides and imides. For example, the reactions of *N*-trimethylsilyl derivatives of lactams with active halogen derivatives such as benzyl bromide, chloro- and bromoacetyl acid esters,<sup>10</sup> chloro-acetamide derivatives (*N*-chloroacetyl-L-proline, *N*-chloroacetyl-L-valine,<sup>11</sup> and 3-chloroacetyl-5-oxazolidone<sup>12</sup>),  $\alpha$ -chloro ethers,<sup>13</sup> acetylated aldonic acid halides,<sup>14</sup> leads to *N*-substituted lactams. *N*-TMS-derivatives of amides, imines, and imides also participate in the reactions of this type, and *N*-trimethylsilyloxymethyl and similar derivatives of these compounds<sup>15</sup> or oxazolidin-5-ones<sup>16</sup> in the presence of electrophilic catalysts can serve as alkylation agents in this case. Known nootropic drugs such as nootropil and phenotropil as well as their topologic counterypes con-

taining peptide bonds<sup>17</sup> were synthesized by amination of the products mentioned above that comprise ester bonds. Thus, "silyl version" of *N*-alkylation means the introduction of methyl group with non-alkylic substitutes to a nitrogen atom of amide or imide.

The current work contains data of use of sulfonic esters based on primary alcohols  $\text{RSO}_2\text{OR}'$  in *N*-alkylation of lactams, amides and imides. This helps to avoid alkaline medium in the course of alkylation and to obtain simple *N*-alkyl derivatives. For instance, heating of a mixture of alkyl sulfonates **1a–c** with trimethylsilyl derivatives of lactams **2a–f**, *N*-TMS-succinimide **2d**, and amides **2e,f** at 140–170 °C for several hours leads to TMS-sulfonates **3a–c** and the corresponding alkylation products **4a–h** (Scheme 1).

The starting silylated lactams **2a–c** are shown as *N*-silyl tautomers for simplicity. Tautomerism of silyllactams, *i.e.* equilibrium of their *N*- and O-derivatives ("silyltautomerism") was described in Ref. 18. TMS-derivatives of amides **2e,f** and succinimide **2d** are also shown as *N*-silyltautomers.

Scheme 1



<b>1</b>	<b>R</b>	<b>R'</b>	<b>2</b>	<b>R''–R'''</b>	<b>2</b>	<b>R''</b>	<b>R'''</b>	<b>3</b>	<b>R</b>	<b>4</b>	<b>R'</b>	<b>R''–R'''</b>	<b>4</b>	<b>R'</b>	<b>R''</b>	<b>R'''</b>
<b>a</b>	Bu	Me	<b>a</b>	$(\text{CH}_2)_3$	<b>e</b>	Me	Me	<b>a</b>	Bu	<b>a</b>	Me	$(\text{CH}_2)_3$	<b>g</b>	Et	Me	Me
<b>b</b>	Ph	Et	<b>b</b>	$\text{CH}_2\text{CH}(\text{Ph})\text{CH}_2$	<b>f</b>	Ph	Me	<b>b</b>	Ph	<b>b</b>	Me	$\text{CH}_2\text{CH}(\text{Ph})\text{CH}_2$	<b>h</b>	Et	Ph	Me
<b>c</b>	Me	$n\text{-C}_5\text{H}_{11}$	<b>c</b>	$(\text{CH}_2)_4$				<b>c</b>	Me	<b>c</b>	Et	$(\text{CH}_2)_3$				
			<b>d</b>	$\text{C}(\text{O})\text{CH}_2\text{CH}_2$						<b>d</b>	$n\text{-C}_5\text{H}_{11}$	$(\text{CH}_2)_3$				
										<b>e</b>	Et	$(\text{CH}_2)_4$				
										<b>f</b>	Et	$\text{C}(\text{O})\text{CH}_2\text{CH}_2$				

\* Based on the materials of the 5th EuChemS Inorganic Chemistry Conference (EICC-5) (June 24–28, 2019, Moscow, Russia).

Note, TMS-sulfonates **3a–c** that form in these reactions probably serve as catalysts for alkylation similar to what was found for trimethylsilyl triflate and trimethylsilyl halogenides,<sup>10</sup> *i.e.* these reactions are autocatalytic.

The herein synthesized *N*-alkyl derivatives of lactams, amides, and imides were described in the literature repeatedly. Their spectral characteristics, for instance their IR spectra and <sup>1</sup>H NMR spectra, are similar to those reported.

In conclusion, we developed the new method for *N*-alkylation of lactams, aliphatic amides, and imides, comprising reaction of their *N*-trimethylsilyl derivatives with alkyl sulfonates at 140–170 °C for several hours in absence of solvents. The advantage of this method is obtaining of *N*-alkylation products without thermal or catalytic initiation.

## Experimental

IR spectra in solid phase were recorded at Bruker Tensor-2 spectrometer with use of incomplete internal reflection module (IIRM). <sup>1</sup>H NMR spectra were registered at 20 °C with Bruker Avance II 300 instrument (<sup>1</sup>H, 300 MHz) in CDCl<sub>3</sub>; in pulse mode with consequent Fourier transform and with <sup>2</sup>H-stabilization of the resonance conditions. Me<sub>4</sub>Si was used as an internal standard.

Alkyl sulfonates **1a–c** were commercial reagents (Aldrich). NMR spectroscopy was used as purity control method. Initial *N*-silylated lactams, *N*-TMS-2-pyrrolidone (**2a**),<sup>19</sup> *N*-TMS-4-phenyl-2-pyrrolidone (**2b**),<sup>17</sup> *N*-TMS-azepan-2-one (**2c**),<sup>20</sup> *N*-TMS-succinimide (**2d**),<sup>21</sup> *N*-TMS-*N*-methylacetamide (**2e**),<sup>22</sup> *N*-TMS-acetanilide (**2f**)<sup>23</sup> were obtained as described; their constants and yields correspond with literature data.

**Synthesis of *N*-alkyl-substituted derivatives **4a–h** (general procedure).** A mixture of alkyl sulfonate **1** (0.05–0.1 mol) with equimolar amount of *N*-silyl derivative of **2** was heated for 3–5 h at 140–170 °C, then fractioned.

***N*-Methyl-2-pyrrolidone (4a).** Yield 72%, b.p. 94–97 °C (18 Torr), *n*<sub>D</sub><sup>20</sup> 1.4705 (see Ref. 24: b.p. 94–96 °C (20 Torr), *n*<sub>D</sub><sup>20</sup> 1.4700). Trimethylsilyl sulfonate **3a** was also obtained (yield 75%), b.p. 115–120 °C (18 Torr), *n*<sub>D</sub><sup>20</sup> 1.4298 (see Ref. 25: b.p. 72–73 °C (3 Torr), *n*<sub>D</sub><sup>20</sup> 1.4300).

***N*-Methyl-4-phenyl-2-pyrrolidone (4b).** Yield 66%, b.p. 182–184 °C (15 Torr), *n*<sub>D</sub><sup>20</sup> 1.5510 (see Ref. 26: b.p. 127–130 °C (1 Torr), *n*<sub>D</sub><sup>20</sup> 1.5538). Trimethylsilyl sulfonate **3a** was also obtained (yield 72%), b.p. 117–120 °C (15 Torr), *n*<sub>D</sub><sup>20</sup> 1.4320.

***N*-Ethyl-2-pyrrolidone (4c).** Yield 59%, b.p. 102–105 °C (20 Torr), *n*<sub>D</sub><sup>20</sup> 1.4660 (see Ref. 27: b.p. 93–97 °C (16 Torr), *n*<sub>D</sub><sup>20</sup> 1.4650). Trimethylsilyl sulfonate **3b** was also obtained (yield 72%), b.p. 160–163 °C (20 Torr), *n*<sub>D</sub><sup>20</sup> 1.4935.

***N*-Pentyl-2-pyrrolidone (4d).** Yield 58%, b.p. 117–120 °C (18 Torr), *n*<sub>D</sub><sup>20</sup> 1.4620 (see Ref. 28: b.p. 87–88 °C (1 Torr), *n*<sub>D</sub><sup>20</sup> 1.4619). Trimethylsilyl sulfonate **3c** was also obtained (yield 51%), b.p. 100–102 °C (18 Torr), *n*<sub>D</sub><sup>20</sup> 1.4240 (comp. Ref. 26: b.p. 80–82 °C (8 Torr), *n*<sub>D</sub><sup>20</sup> 1.4235).

***N*-Ethylazepan-2-one (4e).** Yield 78%, b.p. 125–128 °C (16 Torr), *n*<sub>D</sub><sup>20</sup> 1.4790 (see Ref. 29: b.p. 97 °C (5.5 Torr), *n*<sub>D</sub><sup>20</sup> 1.4772). Trimethylsilyl sulfonate **3b** was also obtained (yield 86%), b.p. 155–157 °C (16 Torr), *n*<sub>D</sub><sup>20</sup> 1.4940.

***N*-Ethylsuccinimide (4f).** Yield 86%, the product crystallized when standing, m.p. 24 °C. (see Ref. 30: b.p. 119–121 °C (15 Torr), m.p. 26 °C).

***N*-Ethyl-*N*-methylacetamide (4g).** Yield 87%, b.p. 185–189 °C (750 Torr), *n*<sub>D</sub><sup>20</sup> 1.4280 (see Ref. 31: b.p. 180 °C (750 Torr)). Trimethylsilylsulfonate **3b** was also obtained (yield 45%), b.p. 168–172 °C (25 Torr), *n*<sub>D</sub><sup>20</sup> 1.4950.

***N*-Ethylacetanilide (4h).** Yield 78%, b.p. 126–130 °C (12 Torr), m.p. 53–54 °C (see Ref. 32: b.p. 138 °C (20 Torr), m.p. 53–54 °C, *n*<sub>D</sub><sup>20</sup> 1.4772). Trimethylsilylsulfonate **3b** was also obtained (yield 71%), b.p. 145–146 °C (12 Torr), *n*<sub>D</sub><sup>20</sup> 1.4950.

This work is financially supported by Ministry of Health of the Russian Federation as part of a state assignment for Pirogov Russian National Research Medical University for 2018–2020 (State registration number AAAA-A18-118051590108-1).

## References

- R. E. Benson, Th. L. Cairns, *J. Am. Chem. Soc.*, 1948, **70**, 2115.
- H. Bredereck, F. Effenberger, G. Simehen, *Ber.*, 1963, **96**, 1350.
- H. Bredereck, R. Gompper, H. Rempfer, K. Klem, H. Kuk, *Ber.*, 1959, **92**, 329.
- M. Matsui, *Mem. Coll. Sci. Eng. Kyoto*, 1909, **2**, 37.
- A. Buhner, *Ann. Chem.*, 1904, **333**, 289.
- R. Gompper, *Chem. Ber.*, 1960, **93**, 187.
- J. W. Ralls, *J. Org. Chem.*, 1961, **21**, 66.
- H. Bohme, G. Berg, *Chem. Ber.*, 1966, **99**, 2127.
- R. Roger, D.G. Neilson, *Chem. Rev.*, 1961, **61**, 179.
- E. P. Kamarova, A. G. Shipov, N. A. Anisimova, N. A. Orlova, O. B. Artamkina, I. Yu. Belavin, Yu. I. Baukov, *J. Gen. Chem. USSR (Engl. Transl.)*, 1988, **58**, 970.
- A. G. Shipov, E. P. Kamarova, N. A. Kalashnikova, E. A. Besova, Yu. I. Baukov, *Russ. J. Gen. Chem. (Engl. Transl.)*, 1997, **67**, 1736.
- E. A. Zheltonogova, N. A. Orlova, V. P. Kobzareva, A. G. Shipov, Yu. I. Baukov, *J. Gen. Chem. USSR (Engl. Transl.)*, 1991, **61**, 2092.
- N. A. Anisimova, E. P. Kamarova, I. Yu. Belavin, Yu. I. Baukov, *J. Gen. Chem. USSR (Engl. Transl.)*, 1986, **56**, 1631.
- N. A. Anisimova, Abstract of Ph.D. Theses, MGPI, Moscow, 1983, 16 p. (in Russian).
- N. A. Orlova, I. Yu. Belavin, V. N. Sergeev, A. G. Shipov, Yu. I. Baukov, *J. Gen. Chem. USSR (Engl. Transl.)*, 1984, **54**, 635.
- A. G. Shipov, N. A. Orlova, I. A. Savost'yanov, O. B. Artamkina, Yu. I. Baukov, *J. Gen. Chem. USSR (Engl. Transl.)*, 1989, **59**, 959.
- A. G. Shipov, E. P. Kamarova, V. V. Negrebetsky, V. I. Akhapkina, S. A. Pogozhykh, U. I. Baukov, *Vestnik RGMU*, 2006, No 1, 56.
- A. I. Albanov, M. F. Larin, V. A. Pestunovich, M. G. Voronkov, E. P. Kamarova, Yu. I. Baukov, *Zh. Obshch. Khim.*, 1981, **51**, 488 (in Russian).
- K. A. Andrianov, A. I. Nogaydeli, D. Sh. Akhobadze, L. M. Khananashvili, L. Sh. Tkeshelashvili, *Bull. Acad. Sci. USSR, Div. Chem. Sci. (Engl. Transl.)*, 1972, **41**, 1100.

20. K. Ruhlman, B. Ruppert, *Lieb. Ann.*, 1965, **686**, 226.
21. S. A. Pogozhykh, Yu. E. Ovchinnikov, E. P. Kamarova, Vad. V. Negrebetsky, A. G. Shipov, A. I. Albanov, M. G. Voronkov, V. A. Pestunovich, I. Yu. Baukov, *Russ. J. Gen. Chem.*, 2004, **74**, 1501.
22. F. Lutsenko, Yu. I. Baukov, A. S. Kostyuk, N. I. Savelyeva, V. K. Krysina, *J. Organomet. Chem.*, 1969, **17**, 241.
23. V. Bazant, V. Chvalovsky, J. Rathousky, *Organosilicon Compounds*, Czechoslovakia, Prague, 1973, **4**, 31.
24. L. Craig, *J. Am. Chem. Soc.*, 1933, **55**, 295.
25. A. G. Shipov, Yu. I. Baukov, *J. Gen. Chem. USSR (Engl. Transl.)*, 1984, **54**, 1642.
26. E. P. Kamarova, N. A. Anisimova, Yu. I. Baukov, *J. Gen. Chem. USSR (Engl. Transl.)*, 1991, **61**, 1284.
27. N. J. Leonard, A. B. Simon, *J. Org. Chem.*, 1952, **17**, 1262.
28. J. H. Padon, H. Adkins, *J. Am. Chem. Soc.*, 1936, **58**, 2487.
29. C. S. Marvel, W. W. Moyer, *J. Org. Chem.*, 1957, **22**, 1065.
30. G. Tsolomiti, K. Tsolomiti, A. Tsolomitis, *Heterocycl. Commun.*, 2006, **12**, 179.
31. W. Titherley, *J. Chem. Soc., Trans.*, 1901, **79**, 391.
32. J. Barluenga, F. Arnar, R. Lis, R. Rodes, *Chem. Soc., Perkin Trans. 1*, 1980, 2732.

*Received April 7, 2019;  
in revised form November 5, 2019;  
accepted November 6, 2019*