

Tandem Palladium-Catalyzed Urea Arylation–Intramolecular Ester Amidation: Regioselective Synthesis of 3-Alkylated 2,4-Quinazolinones

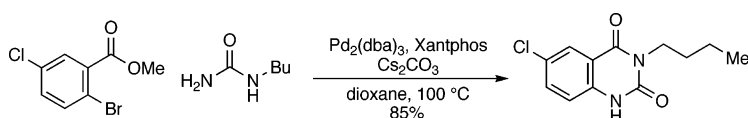
Michael C. Willis,* Robert H. Snell, Anthony J. Fletcher, and Robert L. Woodward

Department of Chemistry, University of Bath, Bath, BA2 7AY, United Kingdom

m.c.willis@bath.ac.uk

Received August 14, 2006

ABSTRACT



o-Halo benzoates can be combined with monoalkyl ureas in a tandem palladium-catalyzed arylation–ester amidation sequence to deliver quinazolinone products. The reactions are regioselective for formation of the 3-*N*-alkyl isomers. Significant variation of both coupling partners is possible, allowing the synthesis of a diverse array of substituted quinazolinones, exemplified by the preparation of a simple unsymmetric-dialkylated natural product.

The quinazolinone motif is an important heterocycle, as not only are these units embedded in a variety of natural products and designed molecules, but they also serve as precursors to a range of related structures.¹ Quinazolinones are responsible for a variety of biological responses, including applications to hypertension,² diabetes,³ and immunosuppression.⁴ A large number of these biologically interesting examples feature unsymmetrical alkylation patterns across the two *N*-atoms; the natural product **1**,⁵ and medicinal agents **2**⁶ and **3**³ are illustrative examples (Figure 1). A significant complication in the synthesis of these molecules is that alkylation of the parent heterocycle often results in the

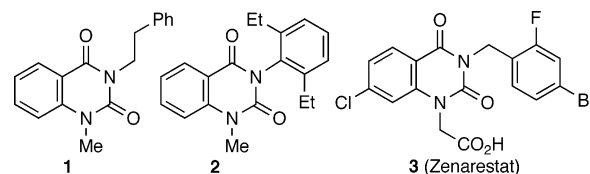


Figure 1. Representative alkylated quinazolinones.

formation of mixtures of regioisomers,⁷ and this has led to the development of synthetic routes involving the stepwise introduction and functionalization of the individual *N*-atoms.⁸ In this Letter we report a new quinazolinone synthesis that involves the regioselective introduction of both *N*-atoms in a single operation.

(1) Undheim, K.; Benneche, T. In *Comprehensive Heterocyclic Chemistry II*; Katritzky, A. R., Rees, C. W., Scriven, E. F. V., Eds.; Elsevier: Oxford, UK, 1996; Vol. 6, p 93.

(2) Ismail, M. A. H.; Barker, S.; El Ella, D. A. A.; Abouzid, K. A. M.; Toubar, R. A.; Todd, M. H. *J. Med. Chem.* **2006**, *49*, 1526.

(3) Hashimoto, M.; Oku, T.; Ito, Y.; Namiki, T.; Sawada, K.; Kasahara, C.; Baba, Y. European Patent Application EP 218999 A2, 1987. Goto, S.; Tsuboi, H.; Kagara, K. *Chem. Exp.* **1993**, *8*, 761.

(4) Buckley, G. M.; Davies, N.; Dyke, H. J.; Gilbert, P. J.; Hannah, D. R.; Haughan, A. F.; Hunt, C. A.; Pitt, W. R.; Profit, R. H.; Ray, N. C.; Richard, M. D.; Sharpe, A.; Taylor, A. J.; Whitworth, J. M.; Williams, S. C. *Bioorg. Med. Chem. Lett.* **2005**, *15*, 751.

(5) Dreyer, D. L.; Brenner, R. C. *Phytochemistry* **1980**, *19*, 935.

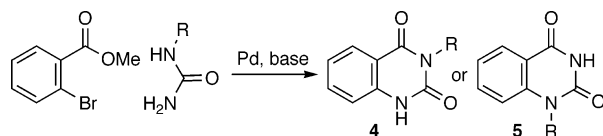
(6) Kakuta, H.; Tanatani, A.; Nagasawa, K.; Hashimoto, Y. *Chem. Pharm. Bull.* **2003**, *51*, 1273.

(7) For example; Goto, S.; Tsuboi, H.; Kanoda, M.; Mukai, K.; Kagara, K. *Org. Proc. Res. Dev.* **2003**, *7*, 700. Also see, ref 2.

(8) For some recent examples, see; (a) Nikpour, F.; Paibast, T. *Chem. Lett.* **2005**, *34*, 1438. (b) Tran, T. P.; Ellsworth, E. L.; Watson, B. M.; Sanchez, J. P.; Showalter, H. D. H.; Rubin, J. R.; Stier, M. A.; Yip, J.; Nguyen, D. Q.; Bird, P.; Singh, R. J. *Heterocycl. Chem.* **2005**, *42*, 669. (c) Rivero, I. A.; Espinoza, K.; Somanathan, R. *Molecules* **2004**, *9*, 609. (d) Makino, S.; Suzuki, N.; Nakanishi, E.; Tsuji, T. *Synlett* **2001**, 333. See also, ref 4.

Continuing our interest in the application of palladium-catalyzed C–N and C–O bond forming reactions in tandem processes for heterocycle synthesis,⁹ we speculated that the combination of an *N*-alkyl urea with an *o*-halo benzoate would provide a direct route to monoalkylated quinazolinodiones (Scheme 1).^{10,11} The heterocycle would be con-

Scheme 1. Tandem Urea Arylation–Ester Amidation Route to Quinazolinodiones



structed by using a tandem reaction sequence involving Pd-catalyzed urea arylation and base-promoted ester amidation. The utility of the process would be dependent on achieving selectivity for either the 3-*N*-alkyl (**4**) or 1-*N*-alkyl (**5**) regioisomer. We anticipated that regiocontrol would be possible by dictating the order of the two bond-forming steps, and by the regioselectivity of the arylation reaction.

Although palladium-catalyzed *N*-arylation is now a routine operation in synthetic chemistry,¹² there are only a handful of reports of the use of urea nucleophiles.^{13,14} To establish that the proposed route was feasible we chose to ignore the issue of regioselectivity and to focus on the coupling of *o*-bromo benzoate **6** and urea (Table 1). On the basis of literature precedent we selected Xantphos¹⁵ as the initial ligand for study;¹³ reaction between benzoate **6** and urea employing NaO^tBu as base delivered only amide **7** in 21% yield (entry 1). However, the use of the weaker base Cs₂CO₃ provided the parent quinazolinodione **8** in 43% yield (entry 2). Simply extending the reaction time to 48 h increased the yield of **8** to 89% (entry 3). In an effort to reduce the reaction time we briefly explored the use of the

(9) (a) Willis, M. C.; Taylor, D.; Gillmore, A. D. *Org. Lett.* **2004**, 6, 4755. (b) Willis, M. C.; Brace, G. N.; Holmes, I. P. *Angew. Chem., Int. Ed.* **2005**, 44, 403. (c) Willis, M. C.; Brace, G. N.; Findlay, T. J. K.; Holmes, I. P. *Adv. Synth. Catal.* **2006**, 348, 851.

(10) For the use of *o*-halobenzaldehydes as substrates for a Pd-catalyzed amination based route to naphthyridinones, see: Manley, P. J.; Bilodeau, M. T. *Org. Lett.* **2004**, 6, 2433.

(11) For a Pd-catalyzed cyclocarbonylation route to quinazolinones, see: Larksarp, C.; Alper, H. *J. Org. Chem.* **2000**, 65, 2773.

(12) General Pd-catalyzed C–N arylation reviews: (a) Hartwig, J. F. In *Handbook of Organopalladium Chemistry for Organic Synthesis*; Negishi, E., Ed.; Wiley-Interscience: New York, 2002; Vol. 1, p 1051. (b) Muci, A. R.; Buchwald, S. L. *Top. Curr. Chem.* **2002**, 219, 131. (c) Prim, D.; Campagne, J.-M.; Joseph, D.; Andrioletti, B. *Tetrahedron* **2002**, 58, 2041.

(13) (a) Yin, J.; Buchwald, S. L. *J. Am. Chem. Soc.* **2002**, 124, 6043. (b) Artamkina, G. A.; Sergeev, A. G.; Beletskaya, I. P. *Tetrahedron Lett.* **2001**, 42, 4381. (c) Sergeev, A. G.; Artamkina, G. A.; Beletskaya, I. P. *Tetrahedron Lett.* **2003**, 44, 4719. (d) Abad, A.; Agulló, C.; Cuñat, A. C.; Vilanova, C. *Synthesis* **2005**, 915. (e) Artamkina, G. A.; Sergeev, A. G.; Stern, M. M.; Beletskaya, I. P. *Synlett* **2006**, 235.

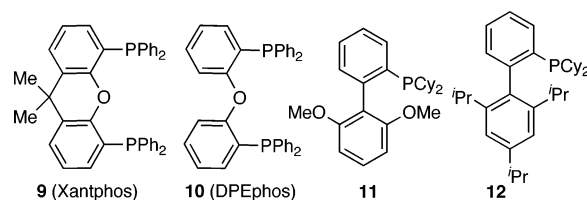
(14) For the use of Pd-catalyzed intramolecular urea arylation in the synthesis of dihydroquinazolinones, see: (a) Ferraccioli, R.; Carenzi, D. *Synthesis* **2003**, 1383. (b) Schlappbach, A.; Heng, R.; Di Padova, F. *Bioorg. Med. Chem. Lett.* **2004**, 14, 357. Also see: (c) Benedí, C.; Bravo, F.; Uriz, P.; Fernández, E.; Claver, C.; Castillón, S. *Tetrahedron Lett.* **2003**, 44, 6073.

(15) Kranenburg, M.; van der Burgt, Y. E. M.; Kamer, P. C. J.; van Leeuwen, P. W. N. M.; Goubitz, K.; Fraanje, J. *Organometallics* **1995**, 14, 3081.

Table 1. Reaction Optimization^a

entry	ligand	base	time (h)	product	yield ^b (%)
1	9	NaO ^t Bu	24	7	21
2	9	Cs ₂ CO ₃	24	8	43
3	9	Cs ₂ CO ₃	48	8	89
4	10	Cs ₂ CO ₃	24		0
5	11	Cs ₂ CO ₃	24		0
6	12	Cs ₂ CO ₃	24		0
7 ^c	9	Cs ₂ CO ₃	24	8	79

^a Reaction conditions; substrate (1.0 equiv), urea (1.2 equiv), Pd₂(dba)₃ (0.5 mol %), ligand (1 mol %), base (2.0 equiv). ^b Isolated yields. ^c 5.0 mol % Pd₂(dba)₃, 10.0 mol % ligand.



structurally similar ligand DPEphos, and the two biphenyl ligands **11**¹⁶ and **12**¹⁷ however, even when Cs₂CO₃ was used as base, no reaction was observed (entries 4–6). Finally, the simplest method to reduce the reaction times was to increase the catalyst loading from 1 mol % palladium to 10 mol % (entry 7).

With conditions for the synthesis of the parent heterocycle available, we next explored the scope of the urea coupling partner and the regioselectivity of the process (Table 2). The

Table 2. Scope of the Urea Component^a

entry	R ¹ , R ²	time (h)	yield (%) ^b
1	Me, Me	24	90
2	H, Me	48	99
3	H, Bu	48	95
4	H, Oct	48	84
5	H, Cy	48	97
6	H, allyl	48	62
7	H, Bn	48	95
8	H, Ph	48	63

^a Reaction conditions; substrate (1.0 equiv), urea (1.2 equiv), Pd₂(dba)₃ (2.5 mol %), ligand (5.0 mol %), base (2.0 equiv). ^b Isolated yields of pure regioisomer.

use of *N,N'*-dimethylurea led to a faster reaction with the desired product obtained in 90% yield after 24 h (entry 1).

(16) Walker, S. D.; Barder, T. E.; Martinelli, J. R.; Buchwald, S. L. *Angew. Chem., Int. Ed.* **2004**, 43, 1871.

(17) Huang, X.; Anderson, K. W.; Zim, D.; Jiang, L.; Klapars, A.; Buchwald, S. L. *J. Am. Chem. Soc.* **2003**, 125, 6653.

Table 3. Scope of the Halo Benzoate Component^a

entry	substrate	product	yield (%) ^b
1 ^c			33
2			72
3			92
4			85
5			68
6			99
7			95

^a Reaction conditions; substrate (1.0 equiv), urea (1.2 equiv), Pd₂(dba)₃ (2.5 mol %), ligand (5.0 mol %), base (2.0 equiv). ^b Isolated yields. ^c 2-(Dimethylamino)-2'-dicyclohexylphosphino-1,1'-diphenyl employed as ligand (10.0 mol %).

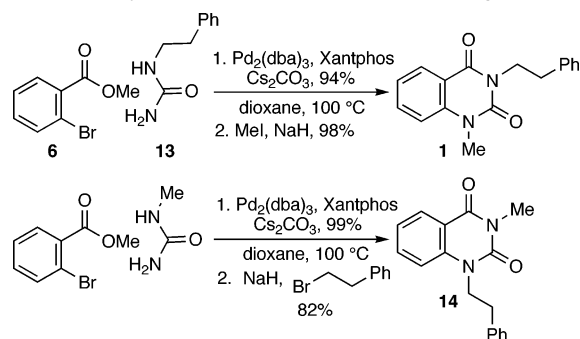
When monoalkylated ureas were employed the reactions required 48 h to reach completion; methyl-, butyl-, octyl-, cyclohexyl-, allyl-, benzyl-, and phenyl-substituted ureas all performed well, delivering the quinazolin-4-one products in good yields (entries 2–8). In all of these examples the 3-alkyl regioisomer was obtained as the exclusive product. This was established by comparison with literature data, nOe studies, and by conversion to known compounds. We are confident that this selectivity arises from an initial arylation reaction followed by a ring-closing amidation. In addition, the arylation reaction occurs on the least hindered, unsubstituted *N*-atom of the urea.¹⁸ In a control reaction we established that combination of bromobenzoate **6** with *N,N*-dimethylurea, under our standard reaction conditions, leads to formation of the *N*-arylated urea with the ester group remaining intact.

(18) For an example of similar selectivity in the arylation of monoalkyl ureas with 2-chloropyridines, see ref 13d.

(19) Old, D. W.; Wolfe, J. P.; Buchwald, S. L. *J. Am. Chem. Soc.* **1998**, *120*, 9722.

We next explored the variation possible in the benzoate component (Table 3). Although reaction of the *o*-chlorobenzoate substrate with our standard conditions delivered only trace amounts of the desired product, this could be increased to 33% by employing Buchwald's 2-(dimethylamino)-2'-dicyclohexylphosphine-substituted biphenyl ligand (entry 1).¹⁹ Returning to the Br-substituted substrates we were pleased to find that both an electron-donating methoxy substituent and an electron-withdrawing nitro group could be readily incorporated (entries 2 and 3). The remainder of the examples serve to demonstrate that functionality suitable for modification of the heterocyclic products can be incorporated into the cyclization substrates; Cl substituents could be introduced at both the quinazolin-4-one 5- and 6-positions and methyl esters at both positions 6 and 7 (entries 4–7).

Finally, to demonstrate the utility of the quinazolin-4-one synthesis we prepared the simple natural product **1**, isolated from Mexican seed husks (Scheme 2).^{5,8b} Tandem urea

Scheme 2. Synthesis of Natural Product **1** and Regioisomer **14**

arylation—ester amidation with bromobenzoate **6** and urea **13** delivered the 3-*N*-alkylquinazolin-4-one in 94% yield. Methylation with methyl iodide and sodium hydride provided the natural product. The regioisomeric compound (**14**) could also be readily prepared by simply employing *N*-methylurea for initial heterocycle formation and then alkylating with the indicated electrophile.

In conclusion, we have demonstrated an efficient one-step route to 3-alkyl quinazolin-4-ones using *o*-bromobenzoates as substrates. Heterocycle formation involves tandem, regio-selective, palladium-catalyzed arylation of a monoalkyl urea followed by an intramolecular amidation reaction. A variety of both functionalized ureas and benzoate coupling partners can be employed, allowing access to a diverse range of functionalized quinazolin-4-one products.

Acknowledgment. This work was supported by the EPSRC and the University of Bath. The EPSRC Mass Spectrometry Service at the University of Wales Swansea is also thanked for their assistance.

Supporting Information Available: Experimental procedures and full characterization for all novel compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

OL062009X