

CYCLIC β -ENAMINOESTERS DECARBOXYLATION WITH BORIC ACID :
 A CONVENIENT SYNTHESIS OF CYCLIC IMINES

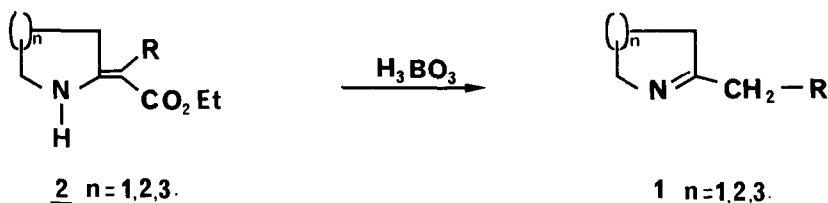
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Summary : Δ^1 -pyrrolines, Δ^1 -piperideines and 1-aza 1-cycloheptenes are formed in good yields from the decarboxylation of alkylated cyclic β -enaminoesters with boric acid.

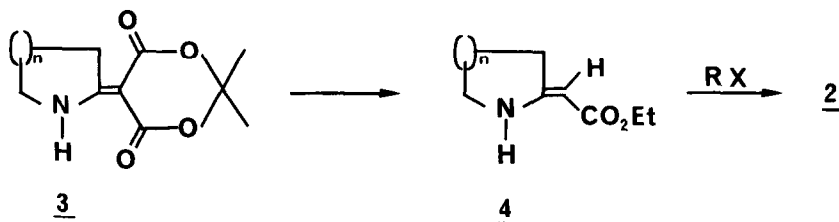
Cyclic 2-alkyl imines (**1**) are useful synthons for indolizidine and quinolizidine alkaloids synthesis¹. Numerous methods are available for the preparation of 2-alkyl pyrrolines (**1**, $n = 1$)²⁻⁷ but these do not give 2-alkyl piperideines or 1-aza 1-cycloheptenes (**1**, $n = 2$ or 3) in useful amounts^{8,9}. The literature also shows that cyclic imines containing a long chain are not easily prepared.

In this letter, we report that 2-alkyl cyclic imines (**1**, $n = 1, 2, 3$) can be readily obtained from cyclic β -enaminoesters precursors (**2**, $n = 1, 2, 3$) which are heated at 180°C during 45 minutes with two equivalents of boric acid (scheme 1). Direct hydrolysis or saponification did not give good yields.



scheme 1

In previous papers, we described a monodecarboxylating transesterification reaction^{11,12} of β -enaminoesters (**3**, $n = 1, 2, 3$)¹³ in basic or neutral medium and the alkylation¹⁴ of β -enaminoesters **4** which need a basic (**4**, $n = 1, 2$) or a neutral medium (**4**, $n = 3$) (scheme 2).



scheme 2

Table. Cyclic imines 1.

n	R	Yield %	b.p. (°C)/Torr	n	R	Yield %	b.p. (°C)/Torr
1	H ^a	80	94/760 ^{2,3}	2	CH ₂ -CH ₃	65	60/15 ^{3,8}
1	CH ₃	95	128/760 ²	2	CH ₂ -Ph	70	85/0.01
1	CH ₂ -CH ₃	70	135/760 ^{2,3}	2	(CH ₂) ₈ 	70	110/0.01
1	CH ₂ -Ph	80	110/0.01	2	(CH ₂) ₉ -CH ₃	80	115/0.01
1	(CH ₂) ₈ 	50	105/0.01	3	H	40	110/760 ^{3,9}
1	(CH ₂) ₉ -CH ₃	90	100/0.01	3	CH ₃	35	60/20 ⁹
2	H ^a	70	120/769 ^{3,8}	3	CH ₂ -CH ₃	30	78/20 ^{3,9}
2	CH ₃	65	70/60	3	CH ₂ -Ph	40	105/0.05

a. In same conditions the cyclic imines 1 (n = 1), 1 (n = 2) are directly prepared from 3 (n = 1,2) with respectively 80, 47 % yields.

All the ¹H-NMR spectra of the cyclic 2-alkyl imines present a multiplet at $\delta = 3.40\text{--}4.00$ ppm, corresponding to the -CH₂- bearing by the nitrogen atom.

The synthetic pathway in this work provides a useful method for preparation for cyclic 2-alkyl imines from a very available starting material through easy transformations of fonctionnal groups.

References and notes.

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