ARTICLE IN PRESS

Tetrahedron: Asymmetry xxx (2016) xxx-xxx

Contents lists available at ScienceDirect



Tetrahedron: Asymmetry

journal homepage: www.elsevier.com/locate/tetasy

Asymmetric synthesis of (6R)-4-hydroxy-6-substituted-δ-lactones

Ravindra D. Gaikwad, Monica D. Rane, Sujata V. Bhat*

Laboratory for Advanced Research in Natural and Synthetic Chemistry, V. G. Vaze College, Mumbai University, Mithagar Road, Mulund (E), Mumbai 400081, India

ARTICLE INFO

Article history: Received 27 October 2016 Accepted 8 December 2016 Available online xxxx

ABSTRACT

A novel approach for the asymmetric synthesis of (6R)-4-hydroxy-6-substituted- δ -lactones has been achieved using asymmetric reduction of a prochiral ketone in the presence of (S)-(-)-diphenyl-prolinol/borane as a key step. The enantiomeric purity of the products was determined by chiral GC. © 2016 Published by Elsevier Ltd.

1. Introduction

Lactones bearing 4-hydroxy-5,6-dihydropyran-2-one ring are ubiquitous in natural products and various biological activities are associated with them, such as sex pheromone, highly potent anti-inflammatory, hypocholesterolemic,¹ anti-HIV, anti-bacterial, anti-fungal, anti-tumor anticonvulsant, muscle-relaxing, sedative and analgesic activities. The statin class of drugs contain (4*R*,6*R*)-4-hydroxy-6-substituted- δ -lactone moiety **1** or the corresponding open chain structure **2** (Fig. 1), which reduce cholesterol levels in blood by inhibiting the rate limiting step of cholesterol biosynthesis in humans by binding to the enzyme HMG Co-A reductase. Simvastatin (Zocor) **1**, atorvastatin (Lipitor) **2**, pitavastatin (Livalo), rosuvastatin (Crestor), and fluvastatin (Lescol) are some examples of statins and several combinations of statins are available.²⁻⁸

1,7-Diarylheptanoids display pharmacological activities such as antioxidant, anticancer, inhibitory activity on nitric oxide production, anti-inflammatory and anti-leishmanial activities. (–)-Diospongins A **3** and B **4** are cyclic 1,7-diaryl-heptanoids, which have been isolated from the rhizomes of *Dioscorea spongiosa*, *via* a bioassay guided fractionation. Diospongins A **3** and B **4** possess a 2,6-*cis*- and 2,6-*trans*-substituted tetrahydro-2*H*-pyran-4-ol ring, respectively.^{9,10} The C1-C8-unit of antitumor (+)-discodermolide **5** contains the 4-hydroxy- δ -lactone with (4*S*,6*R*)-configuration.

Several syntheses of these molecules have been reported.^{11–14} The prominent key steps among these are the asymmetric dihydroxylation or epoxidation of 5,6-dihydropyranone; the selective reduction of the epoxide; lipase catalysed *trans*-esterification or asymmetric hydrolysis; D-2-deoxyribose-5-phosphate aldolase enzyme catalyzed sequential condensation of three aldehydes,⁹ stereocontrolled cyclization of 1,3-diols; asymmetric catalytic

* Corresponding author. *E-mail address:* sujata8b@gmail.com (S.V. Bhat).

http://dx.doi.org/10.1016/j.tetasy.2016.12.006 0957-4166/© 2016 Published by Elsevier Ltd.



Tetrahedron:

Figure 1. Bioactive 4-hydroxy-δ-lactones and analogues.

reduction of 3,5-diketo-esters, asymmetric addition of diketene to aldehydes; asymmetric addition of carbon monoxide to epoxides etc.^{12a,b}

Herein we report an efficient synthesis of (4S,6R)-4-hydroxy-6-phenyl- δ -lactone **6a** and (4R,6R)-4-hydroxy-6-(phenyl-ethyl)- δ -lactone **6b** using an inexpensive asymmetric reduction¹⁵ catalyzed by (S)-(-)-1,1-diphenyl-pyrrolidin-2-yl-methanol / borane (oxazaboro-lidine) as the key step.

2. Result and discussion

The retrosynthetic analysis for the desired lactones is shown in Fig. 2.

ARTICLE IN PRESS

R. D. Gaikwad et al./Tetrahedron: Asymmetry xxx (2016) xxx-xxx



Figure 2. Retrosynthetic analysis of lactones 6a and 6b.

The ethyl (*R*)-3-hydroxy-phenyl-propionate **8a** ($R = -C_6H_5$) was obtained by the asymmetric reduction of β -keto-ester **7a** in the presence of (S)-(-)- α , α -diphenyl-prolinol, trimethylborate and sodium borohydride (Scheme 1). The reaction was carried out under dry conditions and in an N₂ atmosphere at 0 °C for 4 h. The (*R*)-stereochemistry of the newly formed 3-hydroxy-group was assigned based on the positive specific rotation, and the ¹H and ¹³C NMR values of compound **8a**, which were in agreement with the literature values.¹⁶ Subsequent condensation of ethyl (R)-3hydroxy-3-phenyl-propanoate **8a** with *t*-butyl acetate in the presence of LDA at -78 °C in dry THF afforded *t*-butyl (*R*)-5-hydroxy-3-oxo-5-phenyl-pentanoate **9a**. Product **9a** was confirmed by ¹H NMR absorption at δ 0.92 (s, 9H), which corresponded to the *t*-butyl group and at δ 5.14 for proton under –OH. ¹³C NMR of **9a** showed two peaks at δ 199.8 and 172.6 for the ketone and ester, respectively. The IR spectrum of 9a showed bands at 1728, 1714, 3392 cm⁻¹ for the ester, ketone and hydroxyl groups, respectively. The stereospecific reduction^{15c} of **9a** was effected by using NaBH₄ and triethylborane at -30 °C to give t-butyl (3S,5R)-3,5dihydroxy-5-phenyl-pentanoate 10a. The IR spectrum of 10a had bands at 3414 and 1735 cm⁻¹ due to the hydroxyl and ester groups, respectively. Diol 10a was cyclized in the presence of ptoluenesulfonic acid in dry CH₂Cl₂ to yield the lactone (4S,6R)tetrahydro-4-hydroxy-6-phenyl-pyran-2-one **6a**. The specific rotation, and the ¹H NMR and ¹³C NMR data of lactone **6a** were in agreement with the previously reported values.^{12b}



Scheme 1. Synthesis of (4R,6R)-tetrahydro-4-hydroxy-6-phenylpyran-2-one.

Similarly, the ethyl (*R*)-3-hydroxy-5-phenylpentanoate **8b** ($R = -C_6H_5$) was obtained by the asymmetric reduction of 3-keto-ester **7b** in the presence of (*S*)-(–)-1,1-diphenyl-prolinol, triethylborane and sodium borohydride (Scheme 2). The (*R*)-stereochemistry of the newly formed 3-hydroxy-group was assigned based on positive

specific rotation, and the ¹H and ¹³C NMR values of compound **8b**, which were in agreement with the literature values.¹⁷ Subsequent condensation of ethyl ester (*R*)-**8b** with *t*-butyl acetate in the presence of LDA at -78 °C in dry THF afforded *t*-butyl (*R*)-5-hydroxy-3-oxo-5-phenylheptanoate **9b**. Product **9b** was supported by ¹H NMR peak at δ 1.48 (s, 9H) corresponding to the *t*-butyl group. ¹³C NMR of **9b** showed two peaks at δ 192.2 and 173.2 for ketone and ester respectively. IR spectrum of **9b** showed bands at 1730, 1714, 3428 cm⁻¹ for ester, ketone and hydroxyl group respectively. The stereospecific reduction of **9b** was carried out by using NaBH₄ and triethylborane at -30 °C to give *t*-butyl (3*R*,5*R*)-3,5-dihydroxy-7-phenylheptanoate **10b**.

Diol **10b** was cyclised in the presence of *p*-toluene-sulfonic acid in dry CH_2CI_2 to yield the product (4*R*,6*R*)-tetrahydro-4-hydroxy-6phenethylpyran-2-one **6b**. The specific rotation, and ¹H NMR and ¹³C NMR data of lactone **6b** were in agreement with the previously reported values.¹⁴

3. Conclusion

In conclusion, a facile asymmetric synthesis of (6*R*)-4-hydroxy-6-substituted- δ -lactones has been achieved using an asymmetric reduction in the presence of (*S*)-(-)-1,1-diphenyl-prolinol/ trimethylborate (oxazaborolidine)/sodium borohydride as the key step. The resulting (*R*)-(+)- β -hydroxy esters **8a** and **8b** were condensed with the enolate of *t*-butyl acetate at low temperature to yield the corresponding β -keto- δ -hydroxy esters, which were further subjected to stereoselective reduction to yield β , δ -dihydroxy esters. Further acid catalysed cyclization gave the title molecules in good yield and with good enantiomeric excess. The enantiomeric purity of the products was checked by chiral GC.

4. Experimental

4.1. General

Analytical TLC was performed on silica gel 60 F254 plates and visualized using anisaldehyde sulfuric acid reagent. ¹H NMR and ¹³C NMR spectra were recorded at ambient temperature on spectrometers at 500 MHz and 125 MHz, respectively. Chemical shifts are reported in parts per million relative to residual solvent CDCl₃ (¹H, 7.27 ppm; ¹³C, 77.00 ppm). Multiplicities are reported as follows: s = singlet, d = doublet, dd = doublet of doublets, t = triplet, td = triplet of doublets, m = multiplet, q = quartet, bs = broad singlet. Electrospray ionization and a TOF mass analyzer were used for HRMS measurements. GC-MS analysis was carried on instrument, where GC-6890 with coupled mass spectrometer MS-5973N with quadrupole mass detector, using phenyl-methyl-siloxane (5%) column. The enantiomeric excess (*ee*%) of products was obtained by GC analysis using chiral Beta-Dex column (30 m × 0.25 µm × 0.25 mm.

4.2. Synthesis of ethyl 3-oxo-3-phenylpropanoate 7a

A mixture of acetophenone (10 g, 0.083 mol), diethyl carbonate (29.49 g, 0.249 mol) and dry toluene (30 mL) was stirred for 5 min.

Please cite this article in press as: Gaikwad, R. D.; et al. Tetrahedron: Asymmetry (2016), http://dx.doi.org/10.1016/j.tetasy.2016.12.006

ARTICLE IN PRESS

R. D. Gaikwad et al./Tetrahedron: Asymmetry xxx (2016) xxx-xxx



Scheme 2. Synthesis of (4R,6R)-tetrahydro-4-hydroxy-6-phenethylpyran-2-one.

Next, NaH (3.1 g, 0.129 mol) (55%) was added in small portions to the resulting solution at ambient temperature. After the addition, exotherm was observed. The reaction was carried out at rt for 30 min. The reaction mixture was then poured into ice water (100 mL) containing a small amount of acetic acid (1 mL). The resulting mixture was extracted with EtOAc (50×2 mL). The extract was washed with water and dried over anhydrous Na₂SO₄. The solvent was removed in *vacuum* and the residue was subjected to column chromatography to yield **7a**, Yield: 11.90 g (74.4%), colorless liquid; IR ν_{max} : 3063, 2983, 1742, 1683, 1598, 1449, 1360, 1265, 1198 cm⁻¹; ¹H NMR (CDCl₃, 60 MHz): δ 7.95–7.10 (m, 5H), 3.96 (q, 2H), 2.40 (s, 2H), 1.10 (t, *J* = 7 Hz, 3H); GCMS (*m*/*z*): 192 (M⁺), 192, 146, 122, 105, 91, 77, 51; HRMS (*m*/*z*) calcd for C₁₁H₁₂O₃; 192.2140.; found: 192.2132.

4.3. Synthesis of ethyl (R)-3-hydroxy-3-phenylpropionoate 8a

To a stirred solution of NaBH₄ (2.03 g, 0.052 M) in dry THF (20 mL), cooled at 0 °C under an N₂ atmosphere was added catalyst (S)-(-)-1,1-diphenyl-prolinol (0.554 g, 2.2 mM). The mixture was stirred for 5 min. To this solution was added trimethylborate (0.245 mL, 2.17 mmol) followed by the dropwise addition of 3oxo-3-phenylpropanoate 7a (10 g, 0.052 M), while maintaining the temperature at 0 °C and the stirring was continued for 4 h. The reaction was quenched with 2 N HCl solution, after which THF was removed in vacuum. To the residue, EtOAc was added. Aqueous layer was extracted with EtOAc. The combined organic layer was washed with brine and dried over anhydrous Na₂SO₄. The solvent was removed in vacuum and the residue was subjected to column chromatography on silica gel to yield hydroxyl compound **8a**, Yield: 7.40 g (73.20%), colorless liquid; $[\alpha]_D^{20}$ = +28.6 (*c* 0.50, CHCl₃); (lit¹⁶ $[\alpha]_D^{20}$ = +34.7 (*c* 1.4, CHCl₃); Chiral GC analysis: $t_R(\text{minor}) = 15.67 \text{ min}, t_R(\text{major}) = 15.90 \text{ min}; \text{ IR } v_{\text{max}}$: 3392, 3029, 2927, 1729, 1493, 1452, 1371, 1297 $\rm cm^{-1};\ ^1H\ NMR$ $(500 \text{ MHz}, \text{CDCl}_3)$: δ 7.40–7.26 (m, 5H), 5.13 (dd, J = 3.65, 8.95 Hz, 1H), 4.18 (q, J = 7.5 Hz, 2H), 3.34 (s, 1H), 2.75 (dd, J = 9.0, 16.35 Hz, 1H), 2.70 (dd, *J* = 3.85, 16.35, 1H), 1.27 (t, *J* = 7.5, 3H); ¹³C NMR (125 MHz, CDCl₃): δ 172.6, 142.7, 128.7, 127.9, 127.7,

127.1, 125.8, 70.5, 61.0, 43.5, 14.3; GCMS (m/z): 194 (M^+) , 194, 165, 147, 120, 107, 91, 79, 60, 43; HRMS (m/z) calcd for C₁₁H₁₄O₃; 194.2298; found: 194.2292.

4.4. Synthesis of *t*-butyl (*R*)-5-hydroxy-3-oxo-5-phenylpentanoate 9a

A mixture of diisopropyl amine (4.97 g, 22.4 mmol) and dry THF (40 mL), cooled to 0 °C under an N₂ atmosphere was added *n*-BuLi (48 mL, 73.4 mmol) over 10 min, after which stirring was continued for 20 min. After 20 min, the solution was cooled to -78 °C and t-butyl acetate (6.38 g, 47.0 mmol) was added over 5 min. After 20 min, the resulting solution at -78 °C was transferred via syringe over 5 min to a solution of the reactant (2.3 g, 11.0 mmol) and THF at 0 °C. The reaction flask was placed in a water bath and stirred for 1.5 h at room temperature. The reaction was quenched with saturated NH₄Cl solution (1.0 mL). THF was removed in vacuum and the residue was extracted in EtOAc. The aqueous layer was washed by EtOAc. The solvent was removed in vacuum and the residue was subjected to column chromatography on silica gel to yield **9a** 2.26 g (71.50%), colourless liquid; $[\alpha]_D^{20} = +36.95$ (c 0.4385, CHCl₃); lit^{12d} $[\alpha]_D^{20} = +32.3$ (c 0.1, CHCl₃); Chiral GC analysis: *t*_R(major) = 24.13 min; IR *v*_{max}: 3392, 3030, 2930, 1728, 1714, 1493, 1453, 1369, 1154, 1078, 1029, 899, 767 cm⁻¹; ¹H NMR (500 MHz, CDCl₃): δ 7.42–7.26 (m, 5H), 5.14 (dd, J = 3.75, 8.85 Hz, 2H), 4.13 (d, J = 6.70 Hz, 1H), 3.40 (s, 1H), 2.78 (dd, J = 5.45, 16.25 Hz, 1H), 2.71 (dd, J = 3.80, 16.25 Hz, 1H), 0.92 (s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 199.8, 172.6, 142.7, 133.2, 128.7, 128.5, 128.3, 127.8, 70.4, 64.8, 45.5, 43.4, 30.7, 19.2, 13.8; GCMS (*m*/*z*): 264 (M⁺), 265, 241, 207, 161, 130, 115, 88, 69, 43; HRMS (m/z) calcd for C₁₅H₂₀O₄; 264.3206; found: 264.3202.

4.5. Synthesis of *t*-butyl (3S,5*R*)-3,5-dihydroxy-5-phenylpentanoate 10a

To a stirred solution of compound **9a** (0.8 g, 3.03 mmol) in dry THF (15 mL), cooled at -10 °C under an N₂ atmosphere was added triethylborane (0.278 mL, 2.7 mmol). The reaction was stirred for

15 min. To this mixture was added dropwise a solution of NaBH₄ (0.238 g, 6.1 mmol) in dry THF (10 mL). The reaction mixture was allowed to stir for 4 h. The reaction was monitored by TLC, then quenched with water, and extracted with EtOAc, washed with brine and dried over anhydrous Na₂SO₄. The solvent was removed in *vacuum* and residue was subjected to column chromatography on silica gel to give **10a**, Yield: 0.693 g (84.7%); $[\alpha]_D^{20} = -20.2$ (*c* 0.54, CHCl₃); IR ν_{max} : 3414, 2978, 1732, 1404, 1224, 1049, 758, 700 cm⁻¹, ¹H NMR (500 MHz, CDCl₃): δ 7.35–7.15 (m, 5H), 4.97 (dd, *J* = 3.0, 9.0 Hz, 1H), 4.30 (m, 1H), 3.62 (bs, 1H), 2.41 (m, 1H),1.96–1.72 (m, 2H), 1.46 (s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 172.4, 144.7, 128.5, 127.6, 125.8, 81.5, 74.4, 68.8, 44.2, 42.7, 28.3; GCMS (*m*/*z*): 266 (M⁺), 251, 236, 221, 209, 193, 177, 165, 151, 134, 121, 107, 90, 77, 64, 51, 40; HRMS (*m*/*z*) calcd for C₁₅H₂₂O₄; 266.3364; found: 266.3358.

4.6. Synthesis of (4*S*,6*R*)-tetrahydro-4-hydroxy-6-phenylpyran-(2*H*)-2-one 6a

A catalytic amount of *p*-toluenesulfonic acid (0.052 g) was added to a stirred solution of **10a** (0.5 g, 1.87 mmol) in dry CH₂Cl₂ (10 mL) under an N₂ atmosphere and the reaction mixture was stirred at rt for 3 h. After completion of the reaction, the reaction mixture was quenched by the addition of solid NaHCO₃ and the solvent was removed under vacuum. The crude residue was subjected to column chromatography to give lactone 6a, Yield: 0.216 g (60.0%); $[\alpha]_D^{20}$ = +5.6 (c 0.16, CHCl₃); lit⁸ (4R,6S isomer, $[\alpha]_{D}^{20} = -9.4$ (c 1.3, CHCl₃); Chiral GC analysis: t_{R} (major) = 34.76 min; IR v_{max}: 3430, 3015, 2927, 1728, 1495, 1455, 1348, 1257, 1044, 860, 768 cm⁻¹; ¹H NMR (500 MHz, CDCl₃): δ 7.40-7.35 (m, 5H), 5.76 (dd, J = 3.0, 11.50 Hz, 1H), 4.45 (q, J = 3.9, 1H), 2.86 (dd, 4.9, 17.5, H), 2.74 (ddd, J = 1.5, 3.5, 17.5 Hz, H), 2.25-2.05 (m, 3H); $^{13}\mathrm{C}$ NMR (125 MHz, CDCl_3): δ 170.8, 139.4, 128.9, 128.8, 128.5, 125.8, 76.9, 62.6, 38.7, 38.4; GCMS (m/z): 192 (M⁺), 176, 160, 148, 134, 117, 104, 90, 77, 65, 52, 40; HRMS (m/z) calcd for C₁₁H₁₂O₃: 192.2140, found: 192.2132.

4.7. Synthesis of ethyl 3-oxo-5-phenylpentanone 7b

To a solution of diisopropyl amine (3.84 g, 27.0 mmol) in dry THF (20 mL), cooled to -50 °C under an N₂ atmosphere was added *n*-BuLi (20 mL, 30.6 mmol) over 10 min, after which stirring was continued for 20 min, while the reaction mixture was maintained at -50 °C. Next, ethyl acetoacetate (2 g, 15.4 mmol) was added dropwise within period of 15 min, and stirring was continued for 30 min. A solution of benzyl bromide (2.63 g, 15.4 mmol) and dry THF (6 mL) was added dropwise within 20 min and the reaction was maintained at $-50 \degree C$ for 3 h. After completion of the reaction, it was quenched with saturated NH₄Cl (1 mL). THF was removed in vacuum. The residue was extracted in EtOAc and washed with water (10 mL), and dried over anhydrous Na₂SO₄. The solvent was removed in vacuum and residue was subjected to column chromatography to give compound 7b 2.31 g (63.0%), liquid; IR vmax: 3028, 2984, 1743, 1714, 1497, 1454, 1368, 1317, 1031, 757 cm⁻¹; ¹H NMR (60 MHz, CDCl₃): δ 7.27 (m, 5H), 4.10 (q, J = 7.0 Hz, 3H), 3.40 (s, 2H), 2.90 (m, 4H), 1.22 (t, J = 7.0 Hz, 2H); GCMS (m/z): 220 (M⁺), 202, 177, 159, 147, 131, 119, 104, 91, 78, 65, 41; HRMS (*m*/*z*) calcd for C₁₃H₁₆O₃; 220.2676; found: 220.2668.

4.8. Synthesis of ethyl (R)-3-hydroxy-5-phenylpentanoate 8b

To a stirred solution of NaBH₄ (0.740 g, 14.2 mmol) in dry THF (15 mL), cooled at 0 °C under an N₂ atmosphere was added (*S*)-(–)-1,1-diphenyl-prolinol (0.290 g, 1.2 mmol). The mixture was stirred for 5 min. To this solution was added trimethylborate (0.2–0.3 mL) followed by the dropwise addition of ethyl 3-oxo-5-

phenylpentanone 7b (1.5 g, 6.8 mmol) while maintaining the temperature at 0 °C and the reaction was kept stirring for 4 h. The reaction was quenched with NH₄Cl solution. THF was removed in vacuum. The residue was extracted with EtOAc and the aqueous layer was separated and extracted with EtOAc. The combined organic layer was washed with brine and dried over anhydrous Na₂SO₄. Solvent was removed in vacuum and the residue was subjected to column chromatography on silica gel to give hydroxyl compound **8b**, Yield: 1.20 g (79.50%); $[\alpha]_{D}^{20}$ = +5.2 (*c* 0.24, CHCl₃); lit¹⁷ $[\alpha]_D^{20} = +1$ (*c* 1, CHCl₃); chiral GC analysis: $t_R(\text{minor})$ = 32.53 min, $t_{\rm R}$ (major) = 33.65 min; IR $v_{\rm max}$: 3429, 3027, 2931, 1732, 1603, 1495, 1454, 1373, 1320, 1183 cm⁻¹; ¹H NMR (500 MHz, CDCl₃): δ = 7.24 (m, 5H), 4.17 (q, J = 7.5 Hz, 2H), 4.05-3.95 (m, 1H), 3.05 (bs, -OH), 2.50 (dd, J = 3.5, 16.5 Hz, 1H), 2.43 (dd, J = 8.5, 16.5 Hz, 1H), 2.86–2.79 (m, 1H), 2.73–2.67 (m, 1H), 1.89–1.70 (m, 2H), 1.27 (t, J = 7.5 Hz, 3H); ¹³C NMR (125 MHz, $CDCl_3$): $\delta = 173.0, 141.8, 128.5, 128.5, 125.9, 67.3, 60.7, 41.4, 38.2, 128.5, 128$ 31.8, 28.0, 14.2; GCMS (m/z): 222 (M⁺), 204, 176, 159, 145, 130, 117, 104, 91, 77, 60, 43. HRMS (m/z) calcd for $C_{13}H_{18}O_3$; 222.2834; 222.2842.

4.9. Synthesis of *t*-butyl (*R*)-5-hydroxy-3-oxo-7-phenylheptanoate 9b

To a mixture of diisopropyl amine (2.26 g, 0.022 mol) and dry THF (20 mL), cooled to 0 °C under an N₂ atmosphere was added *n*-BuLi (14.12 mL, 0.021 M) over 10 min, stirring was continued for 20 min. After 20 min, the solution was cooled to -78 °C and *t*-butyl acetate (2.5 g, 5.4 mmol) was added over 5 min. The stirring was continued for 20 min, and the resulting solution at -78 °C was transferred via syringe over 5 min to a solution of compound 8b (1.2 g, 5.4 mmol) and THF at 0 °C. The reaction flask was placed in a rt water bath and stirred for 1.5 h. The reaction mixture was quenched with saturated NH₄Cl solution. THF was removed in vacuum and the residue was extracted in EtOAc. The aqueous layer was washed with EtOAc. The solvent was removed in vacuum from the combined extracts and the residue was subjected to column chromatography on silica gel to give **9b**, Yield: 1.15 g (73.80%), liquid; $[\alpha]_{D}^{20} = -12.8$ (c 0.45, CHCl₃); lit.^{11a} ethyl ester $[\alpha]_{D}^{20} = -12.5$ (c 1, CHCl₃); Chiral GC analysis: $t_R(minor) = 26.4 min, t_R(major)$ = 27.5 min; IR v_{max}: 3428, 2932, 1730, 1714, 1603, 1495, 1454, 1370, 1258, 1154, 1030, 843, 754 cm⁻¹; ¹H NMR (500 MHz, CDCl₃): δ 7.33–7.18 (m, 5H), 4.16 (t, J = 4.45 Hz, 2H), 4.06–3.99 (m, 1H), 2.86–2.67 (m, 2H), 2.50 (dd, J = 3.45, 16.45 Hz, 1H), 2.46 (dd, J = 8.7, 16.4 Hz, 1H),3.16 (bs, 1H),1.90–1.70 (m 2H), 1.48 (s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 192.2, 173.2, 141.8, 128.9, 128.4, 126.6, 126.4, 67.3, 64.7, 49.4, 38.3, 31.9, 30.7, 28.2, 19.2, 14.1. HRMS (*m*/*z*) calcd for C₁₇H₂₄O₄; 292.3742; found: 292.3736.

4.10. Synthesis of *t*-Butyl (3*R*,5*R*)-3,5-dihydroxy-7-phenylheptanoate 10b

To a stirred solution of compound **9b** (0.668 g, 2.28 mmol) in dry THF (15 mL), cooled at -10 °C under an N₂ atmosphere was added triethylborane (0.278 mL, 2.7 mmol). The reaction was stirred for 15 min. To this mixture was added dropwise a solution of NaBH₄ (0.180 g, 4.6 mmol) in dry THF (5 mL). The reaction mixture was allowed to stir for 4 h. The reaction was monitored on TLC. It was then quenched with water and extracted with EtOAc, washed with brine and dried over anhydrous Na₂SO₄. The solvent was removed in *vacuum* and the residue was subjected to column chromatography on silica gel to give **10b**, Yield: 0.577 g (85.80%), viscous liquid; $[\alpha]_D^{20} = -5.5$ (*c* 0.51, CHCl₃); lit.^{11a} ethyl ester $[\alpha]_D^{20} = -4.8$ (*c* 1, CHCl₃); Chiral GC analysis: $t_R(\text{minor}) = 29.8 \text{ min}$, $t_R(\text{major}) = 30.4 \text{ min}$; IR v_{max} : 3435, 3018, 2981, 1713, 1495, 1454, 1394, 1369, 1216, 1152, 1085, 740; ¹H NMR (500 MHz,

Please cite this article in press as: Gaikwad, R. D.; et al. Tetrahedron: Asymmetry (2016), http://dx.doi.org/10.1016/j.tetasy.2016.12.006

CDCl₃): δ 7.30–7.15 (m, 5H), 4.25–4.19 (m, 1H), 3.92–3.86 (m, 1H), 2.84-2.60 (m, 2H), 2.41-2.38 (m, 2H), 1.87-1.53 (m, 4H), 1.48 (s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 172.2, 142.2, 128.9, 128.6, 128.5, 128.4, 125.9, 81.6, 71.4, 69.3, 42.7, 39.5, 31.8, 28.1; GCMS (*m*/*z*): 294 (M⁺), 281, 257, 234, 204, 189, 177, 159, 147, 131, 117, 104, 91, 78, 65, 45; HRMS (*m*/*z*) calcd for C₁₇H₂₆O₄; 294.3900; found: 294.3894.

4.11. Synthesis of (4R,6R)-tetrahydro-4-hydroxy-6-phenethylpyran-2-one 6b

A catalytic amount of *p*-toluenesulfonic acid (15 mg) was added to a stirred solution of **10b** (0.150 g, 0.51 mmol) in dry CH₂Cl₂ (5 mL) under an N₂ atmosphere and stirring was continued at rt for 3 h. After completion of the reaction, the reaction mixture was quenched by the addition of NaHCO₃ (0.092 g). The mixture was filtered, and the solvent was removed in *vacuum*. The crude residue was subjected to column chromatography to give lactone **6b**, Yield: 0.082 g (73.0%); $[\alpha]_{D}^{20}$ = +37.1 (*c* 0.10, CHCl₃); lit^{11a} $[\alpha]_{D}^{20}$ = +47 (*c* 0.3, CHCl₃); Chiral GC analysis: $t_{\rm R}$ (major) = 34.56 min; IR $v_{\rm max}$: 3399, 3019, 2927, 1722, 1407, 1215, 1022, 720, 668 cm⁻¹; ¹H NMR (500 MHz, CDCl₃): δ 7.45-7.30 (m, 5H), 4.80-4.55 (m, 1H), 4.20 (m, 1H), 3.10–2.40 (m, 4H), 2.30–1.80 (m, 4H), 1.30 (br s, 1H). ¹³C NMR (125 MHz, CDCl₃): δ 171.4, 140.9, 128.7, 128.7, 128.6, 128.5, 128.4, 76.8, 74.2, 38.4, 37.5, 36.0, 32.0; HRMS (m/z) calcd for C₁₃H₁₆O₃; 220.2676; found: 220.2668.

Acknowledgments

The authors are grateful to the Kelkar Education Trust, Mumbai for encouragement and DST, India, for FIST support. We are also thankful to the Department of Chemistry and Sophisticated Analytical Instrumentation Facility, Indian Institute of Technology, Bombay and Tata Institute of Fundamental Research, Mumbai for NMR spectroscopic data.

A. Supplementary data

Supplementary data associated with this article can be found, in the online version, at http://dx.doi.org/10.1016/j.tetasy.2016.12. 006.

References

- 1. Lewington, S.; Whitlock, G.; Clarke, R.; Sherliker, P.; Emberson, J.; Halsey, J.; Qizilbash, N.; Peto, R.; Collins, R. Lancet 2007, 370, 1829-1839.
- Shepherd, J.; Hunninghake, D. B.; Barter, P.; McKenney, J. M.; Hutchinson, H. G. *Am. J. Cardiol.* **2003**, *91*, 11C–17C. discussion 17C–19C. 2
- Stokker, G. E.; Alberts, A. W.; Gilfillan, J. L.; Huff, J. W.; Smith, R. L. J. Med. Chem. 1986, 29, 852-855.
- (a) Lynch, J. E.; Volante, R. P.; Wattley, R. V.; Shinkai, I. Tetrahedron Lett. 1987, 4 28, 1385–1387; (b) Heathcock, C. H.; Davis, B. R.; Hadly, C. R. J. Med. Chem. 1989, 32, 197-202.
- Clive, D. L. J.; Murthy, K. S. K.; Wee, A. G. H.; Prasad, J. S.; Da Silva, G. V. J.; Majewski, M.; Anderson, P. C.; Evans, C. F.; Haugen, R. D.; Heerze, L. D.; Barrie, J. R. J. Am. Chem. Soc. 1990, 112, 3018–3028.
- Reddy, K. D.; Shekhar, V.; Reddy, S. T.; Reddy, S. P.; Venkateswarlu, Y. Tetrahedron: Asymmetry 2009, 20, 2315–2319.
- 7. Reddy, T. S.; Reddy, D. K.; Narasimhulu, M.; Ramesh, D.; Venkateswarlu, Y. Helv. Chim. Acta 2010, 93, 2158-2163.
- Sabitha, G.; Padmaja, P.; Yadav, J. S. *Helv. Chem. Acta* **2008**, *91*, 2235–2239. Liu, J.; Hsu, C.-C.; Wong, C.-H. *Tetrahedron Lett.* **2004**, *45*, 2439–2441. 8
- 9
- (a) Mohma, H.; Lajis, N. H.; Abas, F.; Ali, A. M.; Sukari, M. A.; Kikuzaki, H.; Nakatani, N. *J. Nat. Prod.* **2005**, *68*, 285; (b) Ali, M. S.; Tezuka, Y.; Awale, S.; 10. Banskota, A. H.; Kadota, S. J. Nat. Prod. 2001, 64, 289.
- 11. (a) Sabitha, O.; Sudhakar, K.; Srinivas, C.; Yadav, J. S. Synthesis 2007, 5, 705-708; (b) Kamal, A.; Krishnaji, T.; Khanna, G. B. R. Tetrahedron Lett. 2006, 47, 8657-8660; (c) Santosh Kumar, P.; Chadha, A. Tetrahedron: Asymmetry 2005, 16, 2790; (d) Solladie, G.; Bauder, C.; Rossi, L. J. Org. Chem. 1995, 60, 7774-7777.
- 12. (a) Heathcock, C. H.; Kleinman, E.; Binley, E. S. J. Am. Chem. Soc. 1978, 100, 8036; (b) Ramachandran, P. V.; Prabhudas, B.; Chandra, J. S.; Reddy, M. V. R. *J. Org. Chem.* **2004**, 69, 6294–6304; (c) Gruttadauria, M.; Aprile, C.; Noto, R. *Tetrahedron Lett.* **2002**, 43, 1669–1672; (d) Xu, C.; Yuan, C. *Tetrahedron* **2005**, 61, 2169–2186; (e) Chu, C.; Morishita, K.; Tanaka, T.; Hayashi, M. Tetrahedron: Asymmetry 2006, 17, 2672–2677.
- (a) Moreno, R. M.; Moyano, A. Tetrahedron: Asymmetry 2006, 17, 1104-1110; 13 (b) Reddy, M. v.; Brown, H. C.; Ramachandran, P. V. J. Organometallic Chem. 2001, 624, 239–243; (c) Gadakh, S. K.; Sudalai, A. Tetrahedron: Asymmetry 2015, 28, 118-123.
- 14. Sawant, K. B.; Jennings, M. P. J. Org. Chem. 2006, 71, 7911-7914.
- (a) Periasamy, M.; Seenivasaperumal, M.; Rao, V. D. Synthesis 2003, 2507-15. 2510; (b) Xu, J.; Wei, T.; Zhang, Q. J. Org. Chem. 2003, 68, 10146-10151; (c) Chen, K. M.; Hardtman, G. E.; Prasad, K.; Rapič, O.; Shapiro, M. J. Tetrahedron Lett. 1987, 28, 155-158.
- 16. Cui, J.; Zhang, M. J.; Wang, X. Q.; Cui, N. J. Chin. Chem. Lett. 2008, 19, 311-316.
- 17. Takushi, S. Chem. Commun. 2009, 5987-5989.