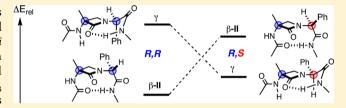


Chirality-Driven Folding of Short β -Lactam Pseudopeptides

Jesus M. Aizpurua,*,‡ Claudio Palomo,*,† Eva Balentová,‡ Azucena Jimenez,‡ Elena Andreieff,‡ Maialen Sagartzazu-Aizpurua,‡ José Ignacio Miranda,‡ and Anthony Linden§

Supporting Information

ABSTRACT: Novel enantiopure pseudopeptide models containing a central -(β-lactam)-(Aa)- scaffold characterized by the combined presence of an α -alkyl- α -amino- β -lactam (i +1) residue and a α -substituted (i + 2) amino acid have been readily synthesized from α -alkyl serines. The conformational analysis of such β -lactam pseudopeptides conducted in CDCl₃ and DMSO-d₆ solutions using 1D- and 2D-NMR techniques revealed an equilibrium between β -II turn and γ -turn



conformers, which was ultimately modulated by the relative configuration of the $-(\beta-\text{lactam})-(\text{Aa})$ - residues. Long-range chiral effects on the α -lactam pseudopeptide conformers were also found when two (i) and (i + 3) chiral residues were attached to the termini of a central -(β -lactam)-(Aib)- segment. In such mimetics, heterochiral (i) and (i + 3) residues reinforced a β -II turn conformer, whereas homochiral corner residues stabilized an overlapped β -II/ β -I double turn motif. No β -hairpin nucleation was observed in any instance. In good agreement with the conformers found in solution, β -turned and open structures were also characterized by X-ray crystallography. Relative stabilities of the different conformers were estimated computationally at a B3LYP/6-31++G** calculation level, and finally, a conformation equilibrium model based on steric inter-residual interactions around the $-(\beta$ -lactam)-(i + 2)- segment was proposed to account for the observed chiral effects.

■ INTRODUCTION

Chirality is at the origin of the regular folding patterns occurring in proteins and natural peptides. It is well-known that short "all-L" peptides may experience dramatic changes in their backbone conformation due to the replacement of a single residue by an achiral unit or a D-configuration amino acid. This phenomenon was first found in natural peptides and is now widely used as a strategy to design peptidomimetics stabilizing reverse turn motifs such as β -turns² 1 and γ -turns³ 2 (Figure 1, top). Indeed, changing the configuration of some backbone residues in functional short peptides may result in increased potency, altered selectivity,⁴ or improved proteolytic stability. Furthermore, these effects can become amplified when conformationally constrained lactam scaffolds 3 incorporating interresidual bridges within contiguous $C\alpha H$ and NH positions (Freidinger lactams)⁵ are embedded in the native peptide backbone (Figure 1, bottom). Formation of more complex structures, such as the doubly hydrogen-bonded β -hairpins 3 (antiparallel β -sheet nucleators), is also strongly dependent on the configurations of the terminal residues (i) and (i + 3).

Unfortunately, most of the (i + 1)-(i + 2) dipeptide lactam surrogates used routinely as turn mimetics⁷ possess comparatively large external fragments, and their restraint elements (cycles) and recognition groups (R¹ and R²) are often crammed in the scaffold.8 Alternatively, the studies carried out on proline

peptides⁹ were only of limited success because of their inherent ring-puckering. This makes it very difficult to interpret the stereoelectronic interactions arising from the chirality effect of residues neighboring the lactam moiety. Hence, a scaffold size reduction strategy intended to design "minimal" short peptide models becomes particularly attractive in terms of design simplicity and folding interpretation.

In our first accomplishment in this area, ¹⁰ we introduced the β -lactam peptides 4 containing a central α -alkyl- α -amino- β lactam ring placed as the (i+1) residue. 11 This approach, termed " β -lactam scaffold-assisted design" (β -LSAD), involved the formal insertion of a single carbon atom ($C\alpha$ -H + H-N \rightarrow $C\alpha$ - CH_2 -N) in the native peptide to form "minimal" β -lactam pseudopeptides 4, which displayed a set of interesting structural features. First, the linear disposition of the $C\alpha$, N and $C\alpha'$ atoms within the β -lactam ring confers to peptidomimetics 4 a high potential for β -hairpin nucleation. Second, they possess a $\psi_{(i+1)}$ dihedral angle fixed at 120° by the β -lactam ring, which prevents the formation of type-I and/or type-III β -turns (both with $\psi_{(i+1)}$ near -30°) and favors only the β -II-turn/ γ -turn conformational equilibrium. 12 Indeed, a preliminary conformational study of β -LSAD pseudopeptides containing a glycine (i

Received: October 28, 2012

[†]Departamento de Química Orgánica-I, Joxe Mari Korta R&D Center, Universidad del País Vasco UPV/EHU, Avda Tolosa-72, 20018 San Sebastian, Spain

[‡]Departamento de Química Orgánica-I, Facultad de Química, Universidad del País Vasco UPV/EHU, Paseo Manuel Lardizabal-3, 20018 San Sebastian, Spain

[§]Institute of Organic Chemistry, University of Zürich, Winterthurerstrasse 190, CH-8057 Zürich, Switzerland

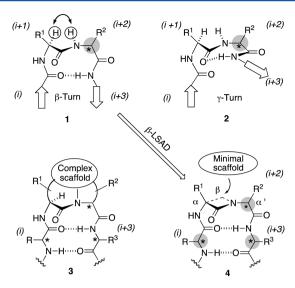


Figure 1. (Top) Peptides folded into β-turn (1) and γ-turn (2) motifs. (Bottom) Freidinger's lactam peptidomimetics (3) and β-LSAD concept (β-lactam scaffold-assisted design) to design minimally scaffolded β-lactam pseudopeptides 4, differing from the parent peptides in a single carbon atom.

+ 2) residue (4; $R^2 = H$) conducted in our laboratory confirmed their strong bias toward forming type-II β -turns which survive even in the presence of the highly coordinating solvent DMSO. Finally, α-alkyl-α-amino-β-lactam peptides are expected to be chemically much more stable than common α-amino-β-lactam antibiotics, which undergo easy ring-opening to β-amino acid derivatives. This transformation usually involves an Ad_N nucleophilic attack onto the C=O sp² carbon atom following the Bürgi-Dunitz trajectory. In β-lactams 4, such a trajectory is hindered by the α-substituents at both sides of the 2-azetidinone plane, thus making the ring-opening reaction more difficult with respect to other α-monosubstituted β-lactam analogues. Recently, the surprisingly high "amidicity" of monocyclic β-lactams has also been invoked to account for their chemical stability.

As mentioned above, only the solution conformation of short β -lactam pseudopeptides containing a central -(β -lactam)-(Gly)- core has been studied so far, and the possible β -turn stabilizing (or destabilizing) effect exerted by the chiral α -amino acid residues neighboring the β -lactam nucleus remains to be explored. Herein we report the synthesis and detailed conformational analysis of three families of β -lactam pseudopeptide models designed to show the chiral effects of residues (i), (i+2) and (i+3) on the folding of short β -lactam peptides 4 in solution and in the solid state. Finally, we further examine the ability of this motif to nucleate a β -hairpin, the shortest β -sheet structure.

RESULTS AND DISCUSSION

Design and Synthesis of *β***-Lactam Pseudopeptide Models.** Short pseudopeptide models **5**–7 (Figure 2) with a homogeneous (R) configuration at the C α stereocenter of the β -lactam ring were selected for development because they are reminiscent of constrained natural L α -amino acid residues. Configurationally related β -lactam peptides **5** and **6** containing a single additional stereocenter at residue (i + 2) were designed to carry terminal groups with different steric demands. Accordingly, mimetics **5** [Ac-(α -Me- β -lactam)-Aa-Me] repre-

5.7
$$\stackrel{\text{H. Me}}{\longrightarrow}$$
 $\stackrel{\text{N. Me}}{\longrightarrow}$ $\stackrel{\text{N. Me}}{\longrightarrow}$

Figure 2. (Top) β -Lactam pseudopeptide models 5–7 selected to study the chiral effect of residues (i), (i+2), and (i+3) on the peptide folding. (Bottom) Synthesis of the β -lactam dipeptide scaffolds 8 and 9.

sented the smallest capped β -lactam dipeptide structures containing three peripheral methyl groups, whereas mimetics $\mathbf{6} \ [\text{Boc-}(\alpha\text{-Bn-}\beta\text{-lactam})\text{-Aa-Aib-OH}] \ \text{had} \ \text{a larger central benzyl group and bulkier achiral N- and C-terminal groups} \ (\text{Boc and Aib-OH}). Finally, pseudopeptides 7 <math>\ [\text{Boc-Ala*-}(\alpha\text{-Bn-}\beta\text{-lactam})\text{-Aib-Ala*-NH}_2] \ \text{were designed to ascertain the chiral effect of two additional stereocenters at the } (i) \ \text{and } (i+3) \ \text{alanine residues on the possible } \beta\text{-hairpin nucleation}.$

The enantiopure scaffolds 8 and 9 required to form the central -(i + 1)-(i + 2)- pseudopeptide segments were easily obtainable from the corresponding α -alkylserinates¹⁷ 10 and 11 according to an operationally very simple method developed in our laboratory. Tables 1 and 2 show the details of the transformation of the dipeptide scaffolds 8, 9, and 14 into the target β -lactam pseudopeptide models.

As shown in Table 1, the synthesis of pseudopeptides 5 and 6 required the coupling of the C-terminal carboxylic acid group in **8** and **9** with methylamine and α -aminoisobutiric benzyl ester, respectively. Formation of N-methylamides 12a-e from nonhindered β -lactam carboxylic acid precursors was best performed using the mixed anhydride method assisted by isobutyl chloroformate and N-methylmorpholine, whereas couplings involving hindered Aib residues were most efficiently accomplished with EEDQ¹⁹ reagent (products 13, 15, 16, and 7a-d). Denosylation of compounds 12 and 13 proceeded smoothly with thiophenol²⁰ in the presence of potassium carbonate to afford the intermediate α -amino- β -lactams, which were trapped in situ with acetic anhydride or di-tert-butyl dicarbonate to provide the desired mimetic models 5 or the analogues 6 after hydrogenolytic deprotection of the C-terminal benzyl ester. Stereochemical integrity was maintained in all transformations and no epimerization of the α' stereocenter could be detected via NMR and HPLC analysis of the reaction crudes. Following a similar protocol (Table 2), mimetics 7 were obtained in good overall yields from the α',α' -dimethyl- β lactam 14 after consecutive coupling with N-protected alanines

Table 1. Preparation of β -Lactam Peptidomimetics 5 and 6

\mathbb{R}^1	\mathbb{R}^2	config $C\alpha'$	product	yield (%)	product	yield (%)
Н	Н		12a	80	5a	84
Me	H	R	12b	58	5b	72
Н	Me	S	12c	59	5c	81
Ph	Н	R	12d	78	5d	89
Н	Ph	S	12e	85	5e	94
Me	Me		12f	75 ^a	5f	82
Me	Н	R	13b	79	6b	79
Н	Me	S	13c	70	6c	80
Ph	Н	R	13d	55	6d	82
Н	Ph	S	13e	66	6e	80
Me	Me				$6f^b$	92

"EEDQ reagent was used as the amide formation reagent. ^bThe product was prepared from acid Boc-(β -lactam)-Aib-OH, which was coupled to H-Aib-OBn and submitted to hydrogenolysis.

Table 2. Preparation of β -Lactam Peptidomimetics 7

\mathbb{R}^1	\mathbb{R}^2	config C $lpha''$	\mathbb{R}^3	\mathbb{R}^4	config $C\alpha'''$	product	yield a (%)
Н	Me	R	Н	Me	R	7a	80
H	Me	R	Me	Н	S	7b	65
Me	Н	S	Me	Н	S	7c	73
Me	Н	S	Н	Me	R	7d	77

^aOverall yield from 15 and 16.

17 and alaninamides 18 using, respectively, the EEDQ and ${\rm HATU/HOAT}^{21}$ reagents.

Solution Conformation of β -Lactam Peptides 5 and 6 with a Chiral (i + 2) Residue. With the model β -lactam peptides in hand, we first addressed the NMR conformational analysis of the simplest families 5 [Ac- $(\alpha$ -Me- β -lactam)-Aa-Me] and **6** [Boc-(α -Bn- β -lactam)-Aa-Aib-OH] in order to characterize the presumably more populated β -II turn conformers in solution and to assess the possible stabilizing or destabilizing effects exerted by the chiral Aa residues on the conformer equilibrium. Although water was considered first as the most biologically meaningful medium for the study, finally DMSO- d_6 solvent was chosen for practical reasons to ensure solubility. In the case of peptides 5a-f, however, the recurrent overlapping of NH amide signals in dimethylsulfoxide precluded a straightforward interpretation of the results and CDCl3 was used as solvent. Both solvents (99.80% D) were used as purchased and were not subjected to any particular pretreatment.

Solutions of each peptide 5a-f and 6a-f (5 × 10^{-3} M) yielded sharp, well-resolved 1H NMR spectra consisting of single sets of signals in both solvents. All protons were

unambiguously assigned from DQF-COSY and HMBC spectra (Figures S1-S10, Supporting Information). Chemical shift variations upon solvent change and ¹H NMR thermal coefficients²² for the exchangeable amide protons of the pseudopeptides in DMSO-d₆ solutions were determined over a temperature range of 300-325 K at 5 K intervals (Table 3). An inspection of the data recorded immediately revealed a totally different deshielding for the β -lactam N₂H amide protons and the N_7H protons of the (i + 3) residue, suggesting the formation of a turned conformer in most of the cases with the N₇H amide protons participating in the intramolecular hydrogen bond network of the β -lactam peptides and the N₂H protons exposed to the solvent. This was evident from the analysis of thermal coefficients in DMSO- d_6 (see Table 3, two right-hand columns) showing in the range -2.3 to -4.9 ppb/K for N₇H protons, whereas amides and carbamates N₂H gave larger absolute thermal coefficient values (up to -8.9 ppb/K). Consistently, the solvent change from nonacceptor CDCl₃ to acceptor DMSO-d₆ resulted in a very small chemical shift variation for N₇H amides in all peptides (0.1-1.1 ppm) but in a significantly larger downfield shift (2.3 – 2.8 ppm) in N₂H β lactam amides. While seeking more insight into the conforma-

Table 3. Amide Proton ¹H NMR Chemical Shifts (δ) and Thermal Coefficients ($\Delta\delta/\Delta T$) for β -Lactam Peptidomimetic Models 5a–f and 6a–f

$$\begin{array}{c} \text{Me}^{1} \\ \text{Me}^{1} \\ \text{N} \\$$

entry	compd	R ⁵	R^6	config C α'	$N_2H(\delta)^a$	$N_7H(\delta)^a$	$N_2H(\Delta\delta/\Delta T)^b$	$N_7H(\Delta\delta/\Delta T)^b$
1	5a	Н	Н		5.63 (+2.77)	8.09 (+0.08)	-5.3	-3.2
2	5b	Me	H	R	5.60 (+2.72)	8.08 (+0.18)	-4.8	-3.4
3	5c	H	Me	S	5.57 (+2.66)	7.94 (+0.29)	-4.5	-3.9
4	5d	Ph	H	R	5.64 (+2.77)	8.41 (+0.16)	-5.0	-4.0
5	5e	H	Ph	S	5.62 (+2.52)	7.12 (+1.15)	-4.5	-4.9
6	5f	Me	Me		5.55 (+2.81)	8.19 (+0.16)	-6.5	-3.9
7	6a	H	H		4.89 (+2.21)	7.82 (+0.01)	-8.3	-2.3
8	6b	Me	H	R	4.76 (+2.84)	7.93 (+0.03)	-8.9	-2.8
9	6c	H	Me	S	4.90 (+2.29)	7.83 (+0.10)	-9.0	-3.0
10	6d	Ph	Н	R	4.85 (+2.69)	8.18 (+0.14)	-8.3	-3.1
11	6e	H	Ph	S	$4.94 (+2.32)^c$	7.85 (+0.56)	-8.9	-4.2
12	6f	Me	Me		4.84 (+2.73)	7.76 (+0.10)	-8.6	-2.5

"Chemical shifts (ppm) measured in CDCl₃ at 300 K. Values in parentheses represent variations of the chemical shift when the CDCl₃ solvent was changed to DMSO-d₆. "Thermal coefficients in ppb/K measured in DMSO-d₆ from 300 to 325 K at 5 K intervals. "A mixture of conformers was observed at room temperature in CDCl₃, which turned to a single set of peaks in DMSO-d₆.

tional differences between the β -lactam peptide diastereomers with (i + 2) chiral α -amino acid residues (Table 3, entries 2–5 and 8-11), we noticed that the thermal coefficients of the N₇H protons where routinely larger for (i + 2)-(S) diastereomers (5c, 5e and 6c, 6e) than for (i + 2)-(R) ones (5b, 5d and 6b, 6d), and furthermore, this effect was appreciably enhanced in pseudopeptides containing the bulkier phenylglycine residues (compare entries 4–5 and 10–11). This lower exposure of the (R) diastereomers to the DMSO- d_6 coordinating solvent was consistent with the tight intramolecular N₇H···O=C hydrogen bonding of β -turned β -lactam peptides 4 (Figure 1), whereas the weaker hydrogen bonds of the (S) diastereomers (entries 3, 5, 9, and 11) suggested a partial folding of the peptide backbone into γ-turned conformers or open conformers possessing more exposed amide protons. Finally, the β -lactam pseudopeptides with nonchiral Gly or Aib (i + 2) residues (entries 1, 6, 7, and 12) exhibited only very shielded N₇H protons in DMSO- d_6 , consistent with their likely participation in stable β -turns.

Next, relevant noncontiguous interproton distances for peptides **5** and **6** were calculated following the ISPA (isolated spin-pair approximation) method²³ from the integration of key NOESY or ROESY cross-peak signals recorded at 300 K (500 MHz) at mixing times of 200–400 ms (Tables S1 and S2, Figures S15 and S16, Supporting Information).

Some expansions of the NOESY spectra of $\mathbf{5d}$, \mathbf{e} (in CDCl₃) and $\mathbf{6d}$, \mathbf{e} (in DMSO- d_6), incorporating a chiral phenylglycine (i+2) residue, are shown in Figure 3 as an example. The spectra gave evidence for a number of NOE contacts, including the excellent diagnostic correlation peaks provided by the pro-R and pro-R diastereotopic R0 protons placed, respectively, at the upper and lower faces of the R1-lactam ring, as depicted in Figure 3. Only the R2-lactam ring, as depicted in Figure 3. Only the R3-lactam ring, as depicted in Figure 3. Only the R4-R5 proton gave a characteristic strong NOE contact (cross-peak a) with the 3Me singlet in peptides 5 and with the benzylic moiety R3-R4-R5 proton oriented to the lower face of the azetidin-2-one

ring plane was used to monitor the 4H^S-N₇H distances (crosspeak **b**) and to estimate the preferred formation of canonical β -II turns ($\approx 2.5-2.8$ Å), distorted γ -turns ($\approx 3.8-4.0$ Å) or open conformers (>4 Å). Additionally, the orientation of the R⁵ and R^6 substituents of the chiral (i + 2) residue relative to the β -lactam ring could be established on the basis of the distances of some of their key protons to 4HR and 4HS (cross-peaks c and d). In the case of 6d, the abnormal shielding of 5Ar ortho protons (6.52 ppm) arising from the diamagnetic effect of the neighboring benzylic 3Ar ring provided a further indirect sign of β -turn conformation. Finally, as none of the peptides 5 and 6 gave measurable 4H^R/N₇H correlation cross-peaks, a significant participation in the conformational equilibrium of inverse γ^i turn β -lactam pseudopeptides with the (i + 2)-(i + 3) branch oriented toward the top side of the azetidinone ring was ruled out (see empty boxes in Figure 3).

From the interproton distances collected in Figure 3, a regular chiral effect trend could be observed for the β -turn/ γ turn equilibrium as the α' -(R) configuration of peptides 5d/6d changed to α' -(S) in their diastereomers **5e/6e**. Indeed, the average 4H^S -N₇H distances (cross-peak b) were significantly shorter for 5d and 6d (2.6 and 3.0 Å) than for their respective isomers 5e and 6e (3.8 Å and >4 Å), suggesting that in the latter cases β -turns were weakened or broken in favor of γ turns. This assumption was further supported by the close distances of the $C\alpha H^5$ protons of the (S)-phenylglycine residue to 4HR and 4HS in 5e/6e and also by the absence of a comparable interaction for $C\alpha H^6$ in 5d/6d. In line with these observations, β -lactam pseudopeptides 5b/6b and 5c/6cderived, respectively from (R)-alanine and (S)-alanine, showed similar β -turn destabilization trends upon $(i + 2) \alpha'$ stereocenter configuration inversion, albeit in these cases the chiral effect was less pronounced (Figures S15 and S16, Supporting Information). Finally, all the β -lactam pseudopeptides with achiral (i + 2) residues of glycine (5a/6a) or α aminoisobutyric acid Aib (5f/6f) populated exclusively very

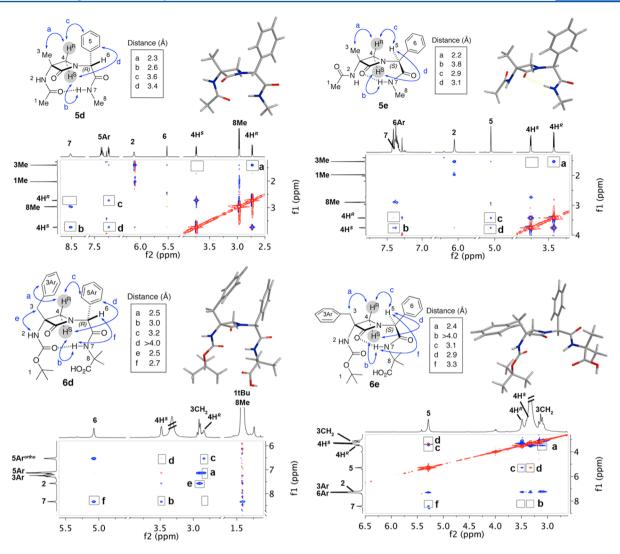


Figure 3. Expansion of the NOESY spectra of pseudopeptides 5d-e in CDCl₃ (top) and 6d-e in DMSO- d_6 (bottom) showing a preferred β-II turn conformation (left-hand) and a γ -turn conformation (right-hand). Key NOE correlations with the upper $(4H^R)$ and lower $(4H^S)$ sides of the β -lactam ring plane are highlighted inside boxes. Empty boxes show the expected position of alternative **b** cross-peaks with $4H^R$ protons. Model figures show energy-minimized SYBYL molecular mechanics structures restrained with NOESY interproton distances $(\pm 0.3 \text{ Å})$ (Tables S1 and S2, Supporting Information).

stable β -turn conformations and showed no tendency to form γ -turns.

Solution Conformation of β -Lactam Peptides 7 with Chiral (i) and (i + 3) Residues. Following the same NMR methodology disclosed above, we next studied the conformational behavior of β -lactam peptidomimetics 7a-d in DMSO- d_6 $(2 \times 10^{-3} \text{ M})$ to check the chiral effect of (i) and (i + 3) alanine residues on the nucleation of β -sheets. The thermal coefficient data collected in Table 4 show that all the (i + 3)N₈H amide protons participate in strong intramolecular hydrogen bonds, likely forming stable β -II turns. Interestingly, unlike the "heterochiral" diastereomers 7b and 7d, "homochiral" diastereomers 7a (R,R) and 7c (S,S) showed solventshielded N₁₀H terminal trans-amide protons which were expected to be accepted by the Boc groups to form β -hairpin type secondary structures. A more detailed examination of the interproton distances from ROESY spectra of 7a, however, revealed that the actual acceptor of the N₁₀H amide proton was the β -lactam carbonyl oxygen, thus forming two overlapped β -II/ β -I turns (Figure 4). In particular, this assumption was

Table 4. Amide Proton ¹H NMR Thermal Coefficients $(\Delta \delta / \Delta T)^a$ for β-Lactam Peptidomimetic 7

entry	compd	config (C_3, C_9)	N_2H	N ₄ H	N ₈ H	$N_{10}H$	N ₁₁ H
1	7a	R, R	-7.4	-6.6	-2.3	-2.4	-5.2
2	7b	R, S	-7.3	-6.7	-2.8	-3.8	-4.1
3	7c	S,S	-6.3	-5.8	-3.5	-2.5	-5.2
4	7d	S.R	-7.0	-5.8	-2.5	-4.3	-4.5

 $^a{\rm Thermal}$ coefficients in ppb/K measured in DMSO- d_6 from 300 to 325 K at 5 K intervals.

further supported by the long-range $N_8H-N_{10}H$ NOE diagnostic cross-peak h observed for the homochiral peptide

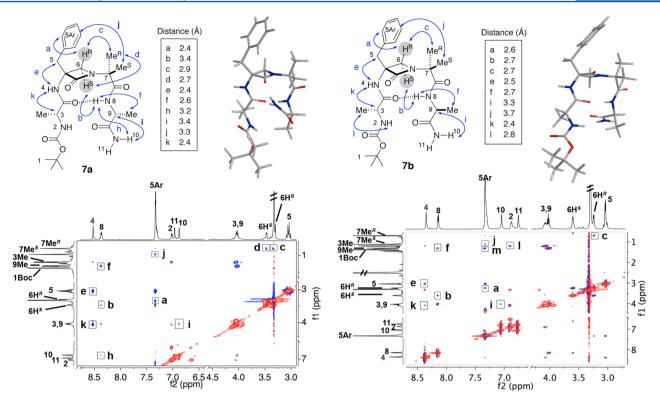


Figure 4. Expansion of the ROESY spectra of pseudopeptides 7a and 7b in DMSO- d_6 showing two overlapped β-turns sharing the β-lactam ring and the N₈H-N₁₀H close distance NOE correlation (h). The model figure shows the energy-minimized structure restrained with ROESY interproton distances (±0.3 Å) (Table S3, Supporting Information).

7a, but not for the heterochiral peptide 7b (for an analogous behavior for diastereomers 7c/7d, see Figure S17, Supporting Information).

Solid-State Conformation of β -Lactam Peptidomimetics 5–7. Single crystals of the β -lactam pseudopeptides 5e, 5f, 7a, and 7d were grown from their solutions in MeOH and were used for X-ray crystal structure determination at 160 K (Figure 5). While 5e and 5f are ansolvates and there is no free space within the structures that might accommodate solvent molecules, 7a crystallizes as a 1:1 methanol solvate and 7d crystallizes as its monohydrate. A detailed analysis of some key structural parameters, Table 5, revealed the formation of welldefined β -II turn conformers in all instances, with the exception of the (S)-phenylglycine-derived mimetic **5e**, which showed an extended backbone structure with the (i) acetamido group and the (i + 3) phenylglycine N-methylcarbamoyl group lying on opposite sides of the plane of the β -lactam ring. Consistent with the extremely destabilized γ -turn conformation observed for this pseudopeptide in solution by NMR, no intramolecular hydrogen bond was detected in the solid state. Instead, each NH amide group acted as a donor to form intermolecular bifurcated hydrogen bonds with the C=O oxygen atom of the β -lactam ring.

Mimetic models **5f**, 7a, and 7d containing the (i + 2) Aib residue fulfilled all the structural criteria of canonical type-II β -turns in terms of $_{(i)}C=O^{\cdots}HN_{(i+3)}$ hydrogen bond distances (1.9-2.1 Å), $_{(i)}O^{\cdots}H-N_{(i+3)}$ angles $(159-167^{\circ})$ and the δ pseudodihedral angle formed by the four $C\alpha$ atoms $(5-14^{\circ})$, which was very close to the ideal null value. Only a slight distortion $(15-20^{\circ})$ caused by the coplanar arrangement of $N_{(i+1)}$, $C\alpha_{(i+1)}$, $N_{(i+2)}$ and $C\alpha_{(i+2)}$ atoms around the central β -lactam ring, was observed for the torsion angles $\phi-\psi$ in

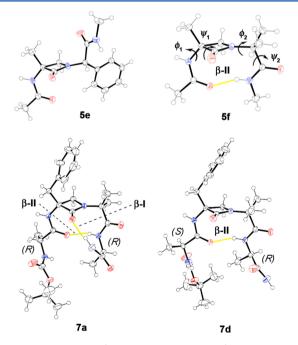


Figure 5. ORTEP plots (50% probability ellipsoids) of the molecular structures of β-lactam peptidomimetics **5e**, **5f**, **7a**, and **7d** in the solid state. Compound **7a** shows overlapping β-II/β-I turns.

residues (i + 1)-(i + 2) with respect to the ideal values for a type-II β -turn. On the other hand, none of the peptides 7a and 7d showed a second hydrogen bond between the terminal Boc and NH₂ groups denoting the formation of β -sheet structures. In spite of that, the NH_{trans} terminal amide proton of peptide 7a acted as a donor to the β -lactam carbonyl oxygen atom to form

Table 5. Selected Geometric Parameters for β -Lactam Peptidomimetic Models 5e, 5f, 7a, and 7d^a

compd	$_{(i)}$ C=O"HN $_{(i+3)}$ (Å)	$\phi_{(i+1)}; \psi_{(i+1)} ext{ (deg)}$	$\phi_{(i+2)}$; $\psi_{(i+2)}$ (deg)	δ^b (deg)	$_{(i)}C = O - H - N_{(i+3)}^{c} (deg)$
5e	>4.0	-39.0(4); 126.6(3)	-95.2(4); 134.9(3)	-157.0	
5f	1.92(3)	-38.3(2); 125.4(2)	90.8(2); -2.8(2)	14.4	160(3)
7a (β-II)	2.07(3)	-39.7(3); 118.9(2)	47.0(4); 30.4(3)	-31.3	167(2)
7a (β-I)	1.87(4)	47.0(4); 30.4(3)	67.2(3); 24.3(3)	-77.3	168(5)
7 d	1.97(3)	-40.6(3); 123.9(2)	95.4(3); -2.5(3)	15.9	159(3)
β -II (ideal)	1.8-2.5	-60; 120	80; 0	0	160
β -I (ideal)	1.8-2.5	60; 30	90; 0	0	160
γ (ideal)	1.8-2.5	-; -	50; -60	0	160

"Measured at 170 K. Solvates: 7a·MeOH and 7d·H₂O. ^bPseudodihedral angle formed by the four Cα atoms of the peptide models. ^cO···H−N angle.

a quasi-canonical β -I turn partially overlapping the β -II turn. To the best of our knowledge, this is the first example of such a type of double motif in solid pseudopeptide structures.²⁴

Finally, a rationale for the chiral effect caused by the (i+2) residue on the relative stability of β -II/ γ conformers in pseudopeptides 5 was established on the basis of the steric interactions of R^1 and R^2 groups with the carbonyl and methylene groups of the β -lactam ring (i+1). Accordingly, we submitted structures $\mathbf{5a-f}$ to a DFT conformational optimization $\mathbf{5a-f}$ consisting of a Monte Carlo search $(6\text{-}31\text{G}^*)$, followed by structure minimization at the $\mathbf{B3LYP/6-31++G^{**}}$ level $\mathbf{6a-f}$ (for details, see the Supporting Information, S39). Energy values collected in Figure 6 show a general trend for the preferred stabilization (2-3 kcal/mol) of type II β -turns, over the normal γ -turns. As expected, no type-I β -turn could be characterized as a stable energy minimum. In addition, a clear chiral-effect destabilizing the β -II turn was apparent in the case

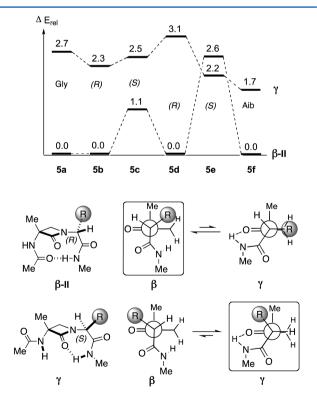


Figure 6. (Top) Relative zero energies (kcal·mol⁻¹) of β -II turn and γ -turn conformers for pseudopeptides 5a-f. (Bottom) Steric interactions accounting for the chiral-effect and relative conformer stabilities. The energy values for the isomer pairs 5b-c and 5d-e are assigned to the lower energy conformer in each case.

of heterochiral (R,S) mimetics $\mathbf{5c}$ and $\mathbf{5e}$ with respect to the homochiral (R,R) mimetics $\mathbf{5b}$ and $\mathbf{5d}$, and this energy-difference increased when the size of R group was increased from methyl to phenyl. In the latter case, a dramatic inversion in the stability of the γ -turn is also observed. Finally, models $\mathbf{5a}$ and $\mathbf{5f}$ bearing respectively achiral (i+2) residues of Gly and Aib displayed very similar relative stability patterns favoring the type β -II conformer over the γ -turns, albeit in the latter case the energy difference was lower.

Figure 6 also depicts the Newman projections of models 5 with the β -lactam nitrogen atom and $C\alpha'$ eclipsed. In homochiral (R,R) isomers, the group R can adopt an alternative disposition only when the β -turn-II conformation is reached, leading to the more stable conformations. The exception to this general bias only occurs when the (i+2) residue is heterochiral with respect to the β -lactam (i+1) stereocenter and the size-difference between R and the β -lactam α - group is large. For instance, in model **5e** with a (S)-phenylglycine (i+2) residue, conformer β -II is destabilized by the steric repulsion of the phenyl group and the β -lactam carbonyl oxygen atom, resulting in the stabilization of the γ -turn conformer.

CONCLUSIONS

The present work illustrates the intrinsic ability of short β lactam peptide chains to form β -II and γ -turn structures in solution. It also emphasizes that "one carbon-scaffolded" β lactam pseudopeptides provide very precise information on the H-bonding pattern in short peptide models, which complements prior quantum chemistry and gas-phase laser-desorption experiments. 25c 2D NMR experiments conducted in solution have provided compelling evidence that β -lactam pseudopeptides 5 and 6 containing a central homochiral - $[(R)-\beta$ -Lactam-(R)-Aa]- segment behave as betagenic, whereas the heterochiral analogues - $\lceil (R) - \beta$ -lactam-(S)-Aa \rceil - destabilize β -turn conformations and favor γ -turned peptides, especially when the (i + 2)Aa residue carries sterically demanding $C\alpha$ backbone substituent groups. In addition, pseudopeptides 7 containing a central betagenic $-[(R)-\beta$ -lactam-Aib]- nucleus flanked by heterochiral (i) and (i + 3) α -amino acid residues do not show a straightforward ability to stabilize β -hairpins. Instead, the homochiral counterparts show a strong tendency to form unprecedented β -II/ β -I overlapped double turns around the central β -lactam core.

In the light of the current results, it can be stated that chirality-driven conformational effect may ultimately determine the preferred type of turn stabilized by β -lactam pseudopeptides. Therefore, the β -LASD design model, consisting in the incorporation of an intraresidual methylene bridge to frozen type-II β -turns, must be applied cautiously to "all-L" peptide natural segments when chiral residues neighboring the central

 β -lactam core are involved. Despite this limitation, the β -LASD design method provides a simple steric model that opens up the route for a better understanding of the steric interactions driving the shallow β -/ γ - transition in larger peptides.

EXPERIMENTAL SECTION

General Experimental Procedures. All reactions were carried out under an atmosphere of nitrogen in oven- or flame-dried glassware with magnetic stirring. Solvents were distilled prior to use. Anhydrous tetrahydrofuran (THF) and dichloromethane were distilled, respectively, from sodium metal/benzophenone ketyl and from calcium hydride. Reaction products were purified by flash chromatography using silica gel 60 (230-400 mesh). Analytical TLC was performed on 0.25 mm silica gel 60-F plates and spots were visualized, either with UV light or with the spray reagent phosphomolybdic acid-ammonium cerium(IV) nitric-sulfuric acid-water, followed by heating. Melting points are uncorrected. ¹H NMR and ¹³C NMR spectra were recorded at 500 and 125.7 MHz, respectively, and are reported as δ values (ppm) relative to residual CDCl₃ (δ 7.26), DMSO (δ 2.49), and CDCl₃ (δ 77.0), DMSO (δ 39.5) as internal standards. Mass spectra were obtained from samples ionized under EI (70 eV) or CI conditions and detected using a quadrupole or a TOC mass analyzer, after direct injection (HRMS) or GC-MS coupling (column: fused silica gel, 15 m, 0.25 mm, 0.25 nm phase SPB-5). The preparations and spectroscopic data of compounds 6a, 8a, 9a-f, and 14 were reported previously. 10a,18

methyl-3-(2-nitrobenzenesulfonylamino)azetidin-2-ones 8b-f. To a solution of (P) mothyl 2 General Procedure for the Preparation of 1-Carboxyalkyl-3-To a solution of (R)-methyl 2-methylserinate 10 (1.90 mmol, 0.32 g) and 2-nitrobenzenesulfonyl chloride (NsCl, 4.18 mmol, 0.935 g) in acetonitrile (30 mL) was added KHCO₃ (9.50 mmol, 0.951 g), and the suspension was stirred at reflux for 16 h. Then saturated aqueous NaHCO₃ (15 mL) was added, and the mixture was extracted with EtOAc (3 × 15 mL). The organic layer was dried (MgSO₄), and the solvents were evaporated at reduced pressure to give pure (2R)-2methyl-2-methoxycarbonyl-1-(2-nitrobenzenesulfonyl)aziridine. A solution of the aziridine (1.51 mmol) and the corresponding β -(tertbutyldimethylsilyloxy)ethylamine (1.51 mmol) in dry acetonitrile (30 mL) was stirred at room temperature for 20 h. Then saturated aqueous NaHCO₃ (15 mL) was added, the mixture was extracted with EtOAc $(3 \times 15 \text{ mL})$, the organic fractions were dried (MgSO₄) and filtered, and solvents were evaporated in vacuo to give the corresponding β azaalanine, which was purified by column chromatography (hexanes/ EtOAc 4:1). To a stirred solution of the β -azaalanine (1 mmol) in dry THF (15 mL) was added 2.5 mmol of LiHMDS in THF (1 M). After being stirred for 12 h, the mixture was washed with a saturated aqueous solution of NaHCO₃ (3 \times 10 mL), and the aqueous layer was extracted with CH2Cl2. The solvent was evaporated under reduced pressure, and the crude product was purified by column chromatography (hexanes/EtOAc, 1:7). The intermediate obtained was oxidized following two different procedures.

Oxidation Method A.²⁷ BAIB (1.069 mmol, 0.344 g), TEMPO (0.243 mmol, 0.038 g), and the corresponding N-hydroxyethyl-β-lactam (0.486 mmol) were combined in the reaction vessel, and to this mixture was added 1:1 acetone—water solution (2 mL). The reaction mixture was stirred for 3 h before evaporation of the solvents under reduced pressure. The resulting product was purified by column chromatography (hexanes/EtOAc, 1:10 or CH₂Cl₂/MeOH, 10:1).

Oxidation Method B. The Jones reagent 28 (1.48 mmol) was added dropwise to the solution of the corresponding N-hydroxyphenylethyl-β-lactam (0.74 mmol) in acetone (10 mL) at 0 °C, and the mixture was stirred at the same temperature for 2 h. The oxidant excess was quenched with $^{\rm i}$ PrOH (10 mL), and the dark solid formed was dissolved with H₂O (4 mL). The mixture was extracted with EtOAc (3 × 10 mL), and the combined organic fractions were washed with brine (5 mL). The organic solution was dried (MgSO₄) and evaporated under reduced pressure, and the resulting product was purified by column chromatography (MeOH/CH₂Cl₂, 1:10).

(3R)-1-[(R)-1-Carboxyethyl]-3-methyl-3-(2-nitrobenzenesulfonylamino)azetidin-2-one (**8b**). The general procedure was followed (oxidation method A) from N-[(2R)-2-hydroxy-1-methylethyl]-β-lactam (0.524 mmol, 0.180 g): yield 0.160 g (87%); yellowish solid; mp 55 °C; $[\alpha]^{25}_{\rm D}$ +7.7 (0.45, MeOH); IR (KBr) $\nu_{\rm max}$ 3320, 2944, 1747, 1718, 1545, 1169, 832 cm⁻¹; ¹H NMR (500 MHz, CD₃OD) $\delta_{\rm H}$ 8.22–7.78 (m, 4H), 4.09 (q, 1H, J = 7.2 Hz), 3.72 (d, 1H, J = 5.8 Hz), 3.35 (d, 1H, J = 5.8 Hz), 1.52 (s, 3H), 1.38 (d, 3H, J = 7.3 Hz); ¹³C NMR (75 MHz, CD₃OD) δ 177.7, 169.4, 149.4, 136.4, 135.0, 133.7, 131.9, 125.9, 66.5, 54.8, 54.3, 20.5, 16.5; MS m/z (ESI, negative polarity) MS-1 355.9; MS2(355.9) 308.8, 200.7, 185.7, 137.8. Anal. Calcd for C₁₃H₁₅N₃O₇S: C, 43.69; H, 4.23; N, 11.76. Found: C, 43.50; H, 4.17; N, 11.49.

(3R)-1-[(S)-1-Carboxyethyl]-3-methyl-(2-nitrobenzenesulfonylamino)azetidin-2-one (8c). The general procedure was followed (oxidation method A) from N-hydroxy-(2S)-methylethyl-β-lactam (0.350 mmol, 0.120 g): yield 0.110 g (88%); white solid; mp 58–60 °C; [α] $^{25}_{\rm D}$ –21.3 (0.5, MeOH); IR (KBr) $\nu_{\rm max}$ 3260, 2901, 1742, 1728, 1543, 1170, 820 cm $^{-1}$; 1 H NMR (500 MHz, CD₃OD) $\delta_{\rm H}$ 8.17–7.83 (m, 4H), 4.23 (q, 1H, J = 7.5 Hz), 3.63 (d, 1H, J = 5.5 Hz), 3.50 (d, 1H, J = 5.5 Hz), 1.58 (s, 3H), 1.38 (d, 3H, J = 7.5 Hz); 13 C NMR (75 MHz, CD₃OD) δ 176.1, 168.1, 147.9, 135.1, 133.6, 132.2, 130.1, 124.4, 65.0, 52.5, 51.5, 19.3, 14.9; MS m/z (ESI, negative polarity) MS-1 356.0; MS2(356.0) 200.7, 137.8; MS3(200.7) 136.8. Anal. Calcd for C₁₃H₁₅N₃O₇S: C, 43.69; H, 4.23; N, 11.76. Found: C, 43.54; H, 4.46; N, 11.47.

(3R)-3-Methyl-3-(2-nitrobenzenesulfonylamino)-1-[(R)-1-phenylcarboxymethyl]azetidin-2-one (**8d**). The general procedure was followed (oxidation method B) from N-hydroxy-(2R)-phenylethyl- β -lactam (1.43 mmol, 0.58 g): yield 0.330 g (55%); yellowish powder; mp 115 °C; [α]²⁵_D 56.1 (0.7, MeOH); IR (KBr) $\nu_{\rm max}$ 3050, 2934, 1749, 1725, 1530, 1156, 802 cm⁻¹; ¹H NMR (500 MHz, CD₃OD) $\delta_{\rm H}$ 8.24–7.32 (m, 9H), 5.37 (s, 1H), 3.97 (d, 1H, J = 6.0 Hz), 3.08 (d, 1H, J = 6.0 Hz), 1.41 (s, 3H); ¹³C NMR (75 MHz, CD₃OD) δ 170.9, 166.4, 146.4, 134.2, 132.2, 130.9, 129.1, 126.9, 126.5, 126.4, 123.1, 63.8, 58.4, 52.5, 17.8; MS m/z (ESI, positive polarity) MS+23: 442.0, 420.0; MS2(420.1) 392.0; 204.9; MS3(392.0) 346.0, 205.0, 158.9. Anal. Calcd for C₁₈H₁₇N₃O₇S: C, 51.55; H, 4.09; N, 10.02. Found: C, 51.78; H, 4.28; N, 9.76.

(3R)-3-Methyl-3-(2-nitrobenzenesulfonylamino)-1-[(S)-1-phenylcarboxymethyl]azetidin-2-one (8e). The general procedure was followed (oxidation method B) from N-hydroxy-(2S)-phenylethyl-β-lactam (0.740 mmol, 0.300 g): yield 0.180 g (58%); yellowish powder; mp 188–190 °C; $[\alpha]^{25}_{\rm D}$ +66.3 (1.25, MeOH); IR (KBr) $\nu_{\rm max}$ 3360, 2944, 1741, 1736, 1536, 1167, 826 cm⁻¹; ¹H NMR (500 MHz, CD₃OD) $\delta_{\rm H}$ 7.83–7.33 (m, 9H), 5.38 (s, 1H), 3.56 (d, 1H, J = 5.6 Hz), 3.23 (d, 1H, J = 5.6 Hz), 1.60 (s, 3H); ¹³C NMR (75 MHz, CD₃OD) δ 173.4, 168.1, 147.8, 135.6, 134.9, 133.6, 132.2, 130.0, 128.4, 128.3, 127.7, 124.5, 65.0, 59.8, 52.8, 19.7; MS m/z (ESI, positive polarity) MS+23 442.0, 420.1; MS2(420.1) 392.0; MS3(392.0) 346.0, 205.0, 159.0. Anal. Calcd for C₁₈H₁₇N₃O₇S: C, 51.55; H, 4.09; N, 10.02. Found: C, 51.32; H, 3.82; N, 9.82.

(3R) 1-(1-Carboxyisopropyl)-3-methyl-3-(2-nitrobenzenesulfonylamino)azetidin-2-one (8f). To the solution of N-(methoxycarbonylisopropyl)- β -lactam (0.259 mmol, 0.110 g) in THF (1 mL) was added a solution of LiOH·H₂O (1.299 mmol, 0.055 g) in H₂O (1 mL). The reaction mixture was stirred at room temperature for 2 h and then was acidified with 2 M HCl and extracted with EtOAc (3 × 5 mL). The organic layer was dried over MgSO₄ and evaporated to give a product which was purified by column chromatography (hexanes/EtOAc, 2:1): yield 0.090 g (93%); yellow oil; [α] $^{25}_{\rm D}$ -33.0 (1.2, MeOH); IR (KBr) $\nu_{\rm max}$ 3260, 2989, 1785, 1764, 1532, 1179, 816 cm⁻¹; ¹H NMR (500 MHz, CD₃OD) $\delta_{\rm H}$ 8.16–7.78 (m, 4H), 3.68 (d, 1H, J = 5.0 Hz), 3.34 (d, 1H, J = 5.0 Hz), 1.61 (s, 3H), 1.52 (s, 3H), 1.44 (s, 3H). ¹³C NMR (75 MHz, CD₃OD) δ 175.9, 169.1, 149.3, 136.4, 135.0, 133.6, 131.7, 125.9, 65.6, 60.2, 54.5, 24.3, 24.2, 20.9; MS m/z (ESI, negative polarity) MS-1: 383.0; MS2(383.0) 200.7, 185.7, 137.8; MS3(137.8) 117.9. Anal. Calcd for C₁₄H₁₇N₃O₇S: C, 45.28; H, 4.61; N, 11.32. Found: C, 44.97; H, 4.77; N, 11.53.

General Procedure for the Preparation of 3-Methyl-1-[(*N*-methylcarbamoyl)alkyl]-3-(2-nitrobenzenesulfonylamino)-azetidin-2-ones (12a–f). To the solution of the corresponding β -lactam 8 (0.175 mmol, 0.060 g) in THF (1.5 mL) were successively added NMM (0.175 mmol, 0.020 mL) and IBCF²⁹ (0.175 mmol, 0.022 mL) at -20 °C. After an activation period of 3 min, 40% aqueous methylamine (0.350 mmol, 0.012 mL) was added, and the resulting solution was stirred for 45 min at -20 °C. After the solvent was evaporated under reduced pressure, the residue was directly loaded on silica gel (hexanes/EtOAc, 1:7) to give the corresponding *N*-methylcarbamoyl- β -lactam.

(3R)-3-Methyl-1-(N-methylcarbamoylmethyl)-3-(2-nitrobenzenesulfonylamino)azetidin-2-one (12a). The general procedure was followed from β-lactam (8a) (0.175 mmol, 0.060 g). The obtained crude product was purified by preparative TLC (hexanes/EtOAc, 1:7): yield 0.050 g (80%); white powder; mp 145–147 °C; $[\alpha]^{25}_{\rm D}$ –87.3 (1.5, CH₂Cl₂); IR (KBr) $\nu_{\rm max}$ 3108, 2990, 1760, 1732, 1526, 1140, 825 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) $\delta_{\rm H}$ 8.10–7.26 (m, 4H), 6.98 (br s, 1H), 5.95 (s, 1H), 4.35 (d, 1H, J = 17.0 Hz), 3.89 (d, 1H, J = 5.5 Hz), 3.57 (d, 1H, J = 17.0 Hz), 3.37 (d, 1H, J = 5.5 Hz), 2.90 (d, 3H, J = 5.5 Hz), 1.62 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) $\delta_{\rm C}$ 167.4, 167.1, 147.6, 134.9, 133.9, 133.2, 130.8, 125.5, 67.9, 57.4, 45.6, 26.3, 21.8; MS m/z (ESI, negative polarity) MS-1 354.9; MS2(354.9) 135.8 137.7 185.7, 193.7. Anal. Calcd for C₁₃H₁₆N₄O₆S: C, 43.82; H, 4.53; N, 15.72. Found: C, 43.80; H, 4.69; N, 15.47.

(3R)-3-Methyl-1-[(1R)-1-(N-methylcarbamoyl)ethyl]-3-(2-nitrobenzenesulfonylamino)azetidin-2-one (12b). The general procedure was followed from β-lactam (8b) (0.08 mmol, 0.030 g). The crude product was purified by preparative TLC (hexanes/EtOAc, 1:7): yield 0.018 g (58%); white solid; mp 152 °C; [α]²⁵_D -67.3 (0.75, CH₂Cl₂); IR (KBr) $\nu_{\rm max}$ 3150, 2963, 1750, 1735, 1520, 1160, 825 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) $\delta_{\rm H}$ 8.09–7.74 (m, 4H), 7.08 (br s, 1H), 5.92 (s, 1H), 4.47 (q, 1H, J = 7.3 Hz), 3.74 (d, 1H, J = 5.5 Hz), 3.40 (d, 1H, J = 5.5 Hz), 2.88 (d, 3H, J = 4.7 Hz), 1.60 (s, 3H), 1.42 (d, 3H, J = 7.3 Hz); ¹³C NMR (125 MHz, CDCl₃) $\delta_{\rm C}$ 170.1, 166.9, 147.6, 135.0, 133.9, 133.2, 130.8, 125.4, 66.5, 53.9, 50.3, 26.4, 21.9, 14.5; MS m/z (ESI, negative polarity) MS-1 369.0; MS2(369.0) 185.7, 137.8, 126.8. Anal. Calcd for C₁₄H₁₈N₄O₆S: C, 45.40; H, 4.90; N, 15.11. Found: C, 45.41; H, 4.78; N, 15.20.

(3R)-3-Methyl-1-[(1S)-1-(N-methylcarbamoyl)ethyl]-3-(2-nitrobenzenesulfonylamino)azetidin-2-one (12c). The general procedure was followed from β-lactam (8c) (0.078 mmol, 0.028 g). The crude product was purified by preparative TLC (hexanes/EtOAc, 1:7): yield 0.017 g (59%); white solid; mp 163–164 °C; [α] $^{25}_{\rm D}$ –69.3 (0.67, CH₂Cl₂); IR (KBr) $\nu_{\rm max}$ 3120, 2959, 1765, 1745, 1540, 1100, 808 cm $^{-1}$; 1 H NMR (500 MHz, CDCl₃) $\delta_{\rm H}$ 8.12–7.72 (m, 4H), 6.56 (br s, 1H), 5.96 (s, 1H), 4.06 (q, 1H, J = 7.3 Hz), 3.84 (d, 1H, J = 5.5 Hz), 3.39 (d, 1H, J = 5.5 Hz), 2.84 (d, 3H, J = 4.8 Hz), 1.58 (s, 3H), 1.57 (d, 3H, J = 7.3 Hz); 13 C NMR (125 MHz, CDCl₃) $\delta_{\rm C}$ 171.1, 167.4, 148.0, 135.8, 134.2, 133.5, 131.2, 125.8, 66.5, 55.9, 53.9, 26.8, 21.9, 15.9; MS m/z (ESI, negative polarity) MS-1 369.0; MS2(369.0) 185.7, 137.8, 126.9; MS3(137.0) 121.8, 90.0. Anal. Calcd for C₁₄H₁₈N₄O₆S: C, 45.40; H, 4.90; N, 15.11. Found: C, 45.04; H, 4.88; N, 15.23.

(3R)-3-Methyl-1-[(1R)-1-(N-methylcarbamoyl)-1-phenylmethyl]-3-(2-nitrobenzenesulfonylamino)azetidin-2-one (12d). To a solution of β -lactam (8d) (0.24 mmol, 0.100 g) and methylamine (2 M in THF; 0.357 mmol, 0.179 mL) in CH₂Cl₂ (5 mL) were added sequentially DCC (0.334 mmol, 0.069 g) and HOBt (0.286 mmol, 0.039 g) at 0 $^{\circ}$ C. The reaction mixture was allowed to warm to rt and stirred for 12 h. Then solvent was removed under reduced pressure, and the resulting crude product was purified by preparative TLC (hexanes/EtOAc, 1:5): yield 0.080 g (78%); yellowish powder; mp 145 °C; $[\alpha]^{25}_{D}$ –144.4 (0.75, CH₂Cl₂); IR (KBr) ν_{max} 3320, 2984, 1768, 1732, 1540, 1158, 828 cm $^{-1}$; ^{1}H NMR (500 MHz, CDCl $_{3}$) $\delta_{\rm H}$ 8.14-7.25 (m, 10H), 5.52 (s, 1H), 3.80 (d, 1H, J = 5.8 Hz), 2.99 (d, 3H, J = 4.7 Hz), 2.94 (d, 1H, J = 5.8 Hz), 1.46 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ_C 169.1, 167.2, 147.9, 135.4, 134.3, 133.7, 131.3, 129.6, 129.2, 125.9, 66.7, 60.1, 55.3, 27.0, 21.8; MS m/z (ESI, negative polarity) MS+1 431.0; MS2(431.0) 193.7, 188.8, 137.8; MS3(188.8)

137.8, 131.8. Anal. Calcd for $C_{19}H_{20}N_4O_6S$: C, 52.77; H, 4.66; N, 12.96. Found: C, 52.91; H, 4.70; N, 12.61.

(3R)-3-Methyl-1-[(1S)-1-(N-methylcarbamoyl)-1-phenylmethyl]-3-(2-nitrobenzenesulfonylamino)azetidin-2-one (12e). According to the aforementioned procedure (compound 12d) product 12e was prepared starting from β -lactam (8e) (0.31 mmol, 0.130 g) and methylamine (2 M in THF; 0.465 mmol, 0.233 mL). An epimerization took place, and the mixture of diastereisomers 12e and 12d (ratio 96:4) was separated by preparative TLC (hexanes/EtOAc, 1:5): yield 0.113 g (85%); yellowish powder; mp 148–150 °C; $[\alpha]^{25}_{D}$ +38.9 (0.7, CH_2Cl_2); IR (KBr) ν_{max} 3210, 2943, 1745, 1725, 1544, 1164, 821 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) $\delta_{\rm H}$ 7.92–7.35 (m, 10H), 5.99 (br s, 1H), 5.93 (br s, 1H), 5.38 (s, 1H), 3.62 (s, 2H), 2.79 (d, 3H, J = 4.7 Hz), 1.62 (s, 3H); 13 C NMR (125 MHz, CDCl₃) $\delta_{\rm C}$ 168.7, 167.1, 147.5, 135.3, 133.9, 133.5, 132.9, 130.7, 129.2, 128.9, 128.4, 125.3, 65.9, 59.2, 54.8, 26.4, 21.3; MS m/z (ESI, negative polarity) MS-1: 431.0; MS2(431.0) 137.8, 188.8, 185.7. Anal. Calcd for C₁₉H₂₀N₄O₆S: C, 52.77; H, 4.66; N, 12.96. Found: C, 53.00; H, 4.59; N, 12.58.

(3R)-3-Methyl-1-[1-(N-methylcarbamoyl)isopropyl]-3-(2-nitrobenzenesulfonylamino)azetidin-2-one (12f). The β -lactam (8f) (0.162 mmol, 0.160 g) was dissolved in CH₂Cl₂ (2.5 mL) and cooled to 0 °C. To this solution was added methylamine (2 M THF) (0.323 mmol, 0.162 mL), and then the reaction mixture was cooled to -12°C (ice/NaCl) and EEDQ (0.072g, 0.292 mmol) was added as a CH₂Cl₂ solution (2.5 mL). The reaction was slowly warmed to room temperature and stirred overnight. After this time, the reaction mixture was concentrated under reduced pressure, and the crude product was purified by preparative TLC (hexanes/EtOAc, 1:7) to afford pure Nmethylamide 12f as a white powder: yield 0.050 g (75%); mp 135 °C; $[\alpha]^{25}$ –46.2 (1.75, CH₂Cl₂); IR (KBr) ν_{max} 3315, 2934, 1758, 1728, 1540, 1166, 798 cm⁻¹; ¹H NMR (500 MHz, CD₃OD) $\delta_{\rm H}$ 8.14–7.84 (m, 4H), 3.63 (d, 1H, J = 5.0 Hz), 3.41 (d, 1H, J = 5.0 Hz), 2.79 (s, 3H,), 1.57 (s, 3H), 1.54 (s, 3H), 1.45 (s, 3H); ¹³C NMR (125 MHz, CD₃OD) δ_C 174.5, 168.3, 148.2, 135.4, 134.1, 132.6, 130.2, 124.8, 64.8, 60.2, 53.8, 25.8, 23.4, 23.3, 20.1; MS *m/z* (ESI, negative polarity) MS-1 383.0; MS2(383.0) 200.7, 185.7, 137.8; MS3(137.8) 117.9. Anal. Calcd for C₁₅H₂₀N₄O₆S: C, 46.87; H, 5.24; N, 14.57. Found: C, 47.02; H, 5.33; N, 14.36.

General Procedure for the Preparation 3-Acetamido-3-methyl-1-[(N-methylcarbamoyl)alkyl]azetidin-2-ones (5a–f). To a stirred solution of the corresponding β-lactam 12 (0.067 mmol, 0.045 g) in dry CH₃CN (5 mL) were added thiophenol (0.335 mmol, 0.034 g) and K_2 CO₃ (0.268 mmol, 0.037 g). After the solution was stirred at room temperature for 1 h (monitored by TLC and 1 H NMR), acetic anhydride (0.402 mmol, 0.088 g) was added, and the reaction was stirred overnight. The solvent was evaporated under reduced pressure, and the crude was purified by preparative TLC (hexanes/EtOAc, 1:7).

(R)-3-Acetamido-3-methyl-1-[1-(N-methylcarbamoyl)methyl]-azetidin-2-one (5a). The general procedure was followed from α-(2-nosyl)-β-lactam 12a (0.084 mmol, 0.030 g). The crude product was purified by preparative TLC (hexanes/EtOAc, 1:7): yield 0.015 g (84%); yellowish oil; $[\alpha]^{25}_{\rm D}$ +94.8 (0.5, CH₂Cl₂); IR (KBr) $\nu_{\rm max}$ 3210, 2965, 1740, 1728 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) $\delta_{\rm H}$ 8.19 (br s, 1H), 6.16 (s, 1H), 4.44 (d, 1H, J = 17.6 Hz), 3.89 (d, 1H, J = 5.0 Hz), 3.57 (d, 1H, J = 17.6 Hz), 3.37 (d, 1H, J = 5.0 Hz), 2.90 (d, 3H, J = 4.6 Hz), 2.06 (s, 3H), 1.58 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) $\delta_{\rm C}$ 170.7, 168.8, 168.0, 65.0, 53.5, 45.3, 26.3, 22.9, 20.5; MS m/z (ESI, negative polarity) MS-1 211.8; MS2(212.0) 181.8, 169.8, MS3(169.8) 126.8, 112.8, 100.8, 96.0, 82.9. Anal. Calcd for C₉H₁₅N₃O₃: C, 50.69; H, 7.09; N, 19.71. Found: C, 50.72; H, 7.35; N, 19.62.

(3R)-3-Acetamido-3-methyl-1-[1-(1R)-(N-methylcarbamoyl)-ethyl]azetidin-2-one (5b). The general procedure was followed from α-(2-nosyl)-β-lactam 12b (0.064 mmol, 0.024 g). The crude product was purified by preparative TLC (hexanes/EtOAc, 1:7): yield 0.010 g (72%); yellowish solid; mp 159–160 °C; $[\alpha]^{25}_{\rm D}$ +58.2 (0.5, CH₂Cl₂); IR (KBr) $\nu_{\rm max}$ 3125, 3060, 1782, 1768 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) $\delta_{\rm H}$ 8.13 (br s, 1H), 5.96 (s, 1H), 4.47 (q, 1H, J = 7.3 Hz), 3.73 (d, 1H, J = 5.1 Hz), 3.23 (d, 1H, J = 5.1 Hz), 2.84 (d, 3H, J = 4.6 Hz), 2.04 (s, 3H), 1.55 (s, 3H), 1.40 (d, 3H, J = 7.3 Hz); ¹³C NMR (125

I

MHz, CDCl₃) $\delta_{\rm C}$ 170.6, 168.6, 63.5, 50.2, 49.8, 26.5, 23.0, 20.7, 14.4; MS m/z (ESI, negative polarity) MS-1 225.9; MS2(225.9) 183.8; MS3(183.8) 126.8, 96.9. Anal. Calcd for $\rm C_{10}H_{17}N_3O_3$: C, 52.85; H, 7.54; N, 18.49. Found: C, 52.86; H, 7.60; N, 18.53.

(3R)-3-Acetamido-3-methyl-1-[1-(1R)-(N-methylcarbamoyl)-ethyl]azetidin-2-one (**5c**). The general procedure was followed from α-(2-nosyl)-β-lactam **12c** (0.060 mmol, 0.022 g). The crude product was purified by preparative TLC (hexanes/EtOAc, 1:7): yield 0.011 g (81%); white powder; mp 115 °C; [α] $^{25}_{\rm D}$ +58.6 (0.5, CH₂Cl₂); IR (KBr) $\nu_{\rm max}$ 3310, 3140, 1785, 1758 cm $^{-1}$; ¹H NMR (500 MHz, CDCl₃) $\delta_{\rm H}$ 8.01 (br s, 1H), 5.96 (s, 1H), 3.80 (d, 1H, J = 5.0 Hz), 3.75 (q, 1H, J = 7.4 Hz), 3.17 (d, 1H, J = 5.0 Hz), 2.86 (d, 3H, J = 4.7 Hz), 2.03 (s, 3H), 1.73 (d, 3H, J = 7.5 Hz), 1.55 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) $\delta_{\rm C}$ 171.4, 170.5, 168.6, 63.4, 56.2, 53.5, 26.4, 23.0, 20.6, 16.3; MS m/z (ESI, negative polarity) MS-1 225.9; MS2(225.9) 195.8, 183.8; MS3(183.8) 126.8, 96.9. Anal. Calcd for C₁₀H₁₇N₃O₃: C, 52.85; H, 7.54; N, 18.49. Found: C, 52.59; H, 7.68; N, 18.84.

(3R)-3-Acetamido-3-methyl-1-[1-(1R)-(N-methylcarbamoyl)-phenylmethyl]azetidin-2-one (5d). The general procedure was followed from α-(2-nosyl)-β-lactam 12d (0.108 mmol, 0.047 g). The crude product was purified by preparative TLC (CH₂Cl₂/MeOH 10:1): yield 0.028 g (89%); white powder; mp 99 °C; $[\alpha]^{25}_D$ +21.2 (0.5, CH₂Cl₂); IR (KBr) ν_{max} 3125, 3094, 1760, 1725 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ_{H} 8.44 (br s, 1H), 7.37–7.22 (m, 5H), 5.99 (br s, 1H), 5.50 (s, 1H), 3.73 (d, 1H, J = 5.0 Hz), 2.95 (d, 3H, J = 4.6 Hz), 2.72 (d, 1H, J = 5.0 Hz), 2.00 (s, 3H), 1.38 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ_{C} 170.9, 169.2, 168.3, 134.3, 129.0, 128.9, 128.4, 63.4, 59.4, 51.3, 26.6, 22.9, 20.2; MS m/z (ESI, negative polarity) MS-1 288.1; MS2(288.0) 245.8, 230.8, 171.8, 126.8, 111.9; MS3(245.8) 126.8; MS4(126.8) 111.8, 98.9. Anal. Calcd for C₁₅H₁₉N₃O₃: C, 62.27; H, 6.62; N, 14.52. Found: C, 62.40; H, 6.51; N, 14.75.

(3R)-3-Acetamido-3-methyl-1-[1-(1S)-(N-methylcarbamoyl)-phenylmethyl]azetidin-2-one (**5e**). The general procedure was followed from α-(2-nosyl)-β-lactam **12e** (0.144 mmol, 0.062 g). The crude product was purified by preparative TLC (CH₂Cl₂/MeOH 10:1): yield 0.039 g (94%); white crystals; mp 198–199 °C; [α]²⁵_D +133.5 (0.75, MeOH); IR (KBr) ν_{max} 3210, 2998, 1744, 1725 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ_{H} 7.43–7.39 (m, 5H), 7.14 (br s, 1H), 5.66 (s, 1H), 5.77 (s, 1H), 3.77 (d, 1H, J = 5.0 Hz), 3.46 (d, 1H, J = 5.0 Hz), 2.93 (d, 3H, J = 4.7 Hz), 2.02 (s, 3H), 1.58 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ_{C} 170.1, 168.9, 168.3, 134.6, 128.8, 128.7, 128.5, 76.6, 63.4, 62.3, 53.9, 26.5, 23.0, 20.4; MS m/z (ESI, negative polarity) MS-1: 288.1; MS2(288.0) 245.8, 230.8, 188.8, 171.8, 126.8; MS3(245.8) 126.8; MS4(126.8) 111.8, 58.3. Anal. Calcd for C₁₅H₁₉N₃O₃: C, 62.27; H, 6.62; N, 14.52. Found: C, 62.67; H, 6.70; N, 14.30.

(3R)-3-Acetamido-3-methyl-1-[1-(N-methylcarbamoyl)-isopropyl]azetidin-2-one (5f). The general procedure was followed from α-(2-nosyl)-β-lactam 12f (0.091 mmol, 0.035 g). The crude product was purified by preparative TLC (hexanes/EtOAc, 1:7): yield 0.018 g (82%); yellowish powder; mp 216–218 °C; $[a]^{25}_{\rm D}$ +57.7 (0.75, CH₂Cl₂); IR (KBr) $\nu_{\rm max}$ 3320, 3180, 1769, 1748 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) $\delta_{\rm H}$ 8.28 (br s, 1H), 6.30 (s, 1H), 3.73 (d, 1H, J = 5.0 Hz), 3.20 (d, 1H, J = 5.0 Hz), 2.82 (d, 3H, J = 4.6 Hz), 2.02 (s, 3H), 1.71 (s, 3H), 1.52 (s, 3H), 1.36 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) $\delta_{\rm C}$ 173.7, 170.7, 168.3, 62.2, 60.3, 51.0, 26.6, 24.8, 24.3, 22.9, 20.7; MS m/z (ESI, positive polarity) MS+23 264.0, 242.0; MS2(242.0) 129.0; MS3(129.0) 101.1, 58.4. Anal. Calcd for C₁₁H₁₉N₃O₃: C, 54.76; H, 7.94; N, 17.42. Found: C, 54.50; H, 8.00; N, 17.09.

General Procedure for the Preparation of Ns-(β -lactam)-Aa-Aib-OBn Compounds (13b-e). To a stirred solution of the corresponding N-carboxymethyl β -lactam 9b-e (1 mmol) in CH₂Cl₂ (20 mL) cooled to -0 °C was added H-Aib-OBn (1 mmol, 0.193 g) dissolved in CH₂Cl₂ (9 mL). The solution was cooled to -12 °C, EEDQ (1.2 mmol, 0.310 g) was added, and the reaction mixture was slowly warmed to room temperature and stirred overnight. The resulting solution was washed with 1 M HCl (3 × 10 mL), the aqueous layer was extracted with CH₂Cl₂, and the combined organic

phases were dried (MgSO₄) and evaporated. The product was purified by flash column chromatography (hexanes/EtOAc, 2:1).

Compound 13b. The general procedure was followed from the Ncarboxyalkyl-β-lactam **9b** (1 mmol, 0.433 g): yield 0.486 g (79%); white solid; mp 216–218 °C; $[\alpha]_{D}^{25}$ +54.7 (0.42, CH₂Cl₂); IR (KBr) $\nu_{\rm max}$ 3396, 3085, 1762, 1681, 1539, 1347, 1153 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) $\delta_{\rm H}$ 8.15–7.72 (m, 4H, o-Ns), 7.37–7.21 (m, 10H, Ar), 7.17 (s, 1H, NHCMe₂), 5.89 (s, 1H, NH-Ns), 5.21 (d, 1H, I = 12.3Hz, OC H_2 Ph), 5.09 (d, 1H, J = 12.6 Hz, OC H_2 Ph), 4.32 (m, 1H, NCHMe), 3.55 (d, 1H, J = 5.7 Hz, CH_2NCO), 3.34 (d, 1H, J = 5.7Hz, CH₂NCO), 3.14 (d, 1H, J = 13.5 Hz, NCCH₂Ph), 3.03 (d, 1H, J = 13.5 Hz, NCCH₂Ph) 13.5 Hz NCCH₂Ph), 1.59 (s, 3H, Me₂C), 1.56 (s, 3H, Me₂C), 0.83 (d, 3H, I = 6.9 Hz, MeCHNH); ¹³C NMR (125 MHz, CDCl₃) δ_C 173.8, 168.8, 166.1, 147.4, 134.9, 133.8, 133.2, 132.6, 130.9, 130.2, 128.9, 128.5, 128.2, 128.1, 125.3, 70.1, 67.0, 56.6, 50.5, 49.5, 40.3, 25.4, 24.7, 13.4; MS m/z (ESI, positive polarity) MS+23: 264.0, 242.0; MS2(242.0) 129.0; MS3(129.0) 101.1, 58.4. Anal. Calcd for C₃₀H₃₂N₄O₈S: C, 59.20; H, 5.30; N, 9.20. Found: C, 58.86; H, 5.64; N, 9.01.

Compound 13c. The general procedure was followed from the *N*-carboxyalkyl-β-lactam 9b (1 mmol, 0.433 g): yield 0.426 g (70%); white solid; mp 216–218 °C; $[\alpha]^{25}_{\rm D}$ +106.7 (0.3, CH₂Cl₂); IR (KBr) $\nu_{\rm max}$ 3378, 3105, 1748, 1539, 1356, 1165 cm⁻¹. ¹H NMR (500 MHz, CDCl₃) $\delta_{\rm H}$ 8.14–7.74 (m, 4H, o-Ns), 7.40–7.25 (m, 10H, Ar), 6.93 (s, 1H, NHCMe₂), 5.86 (s, 1H, NH-Ns), 5.18 (s, 2H, OCH₂Ph), 3.89 (q, 1H, J = 1.6 Hz, NCH(Me)CO), 3.72 (d, 1H, J = 5.7 Hz, CH₂NCO), 3.47 (d, 1H, J = 5.7 Hz, CH₂NCO), 3.14 (q, 2H, J = 1.6 Hz, CqCH₂Ph), 1.59 (s, 3H, Me_2 C), 1.58 (s, 3H, Me_2 C), 1.46 (d, 3H, J = 7.25 Hz, MeCHNH); ¹³C NMR (125 MHz, CDCl₃) $\delta_{\rm C}$ 173.9, 169.5, 166.5, 147.4, 133.6, 133.2, 133.1, 130.9, 130.0, 129.0, 128.5, 128.2, 128.1, 127.9, 125.3, 69.5, 67.2, 56.6, 53.8, 53.5, 40.6, 29.7, 24.9, 15.1; MS m/z (TOF MS ES+) MS[+Na] 630.9, MS[+H] 608.8, MS: 562.8, 500.9, 472.9, 394.0, 359.9, 115.9. Anal. Calcd for C₃₀H₃₂N₄O₈S: C, 59.20; H, 5.30; N, 9.20. Found: C, 59.14; H, 5.55; N, 8.92.

Compound 13d. The general procedure was followed from the *N*-carboxyalkyl-β-lactam 9d (1 mmol, 0.497 g): yield 0.368 g (55%); white solid; mp 72–74 °C; [α]²⁵_D –83.8 (0.65, CH₂Cl₂); IR (KBr) $\nu_{\rm max}$ 3378, 3028, 1768, 1742, 1677, 1534, 1148 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) $\delta_{\rm H}$ 8.15–7.68 (m, 4H), 7.50–6.85 (m, 15H), 5.90 (br s, 1H), 5.32 (s, 1H), 5.20 (d, 1H, J = 12.5 Hz), 5.16 (d, 1H, J = 12.5 Hz), 3.69 (d, 1H, J = 6.0 Hz), 3.04 (d, 1H, J = 6.0 Hz), 3.00 (d, 1H, J = 14.0 Hz), 2.95 (d, 1H, J = 14.0 Hz), 1.67 (s, 3H), 1.64 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) $\delta_{\rm C}$ 173.8, 167.8, 165.9, 147.5, 136.0, 134.8, 134.0, 133.4, 133.3, 132.5, 131.1, 130.1, 129.1, 128.9, 128.7, 128.6, 128.5, 128.2, 128.1, 128.0, 125.5, 69.9, 67.2, 59.2, 55.9, 52.3, 40.3, 25.5, 24.9; MS m/z (ESI, positive polarity) MS 693.2; MS2(693.2) 671.2, 666.1, 665.2, 478.3; MS3(665.2) 479.2, 478.2, 463.1. Anal. Calcd for C₃₅H₃₄N₄O₈S: C, 62.67; H, 5.11; N, 8.35. Found: C, 62.60; H, 4.95; N, 8.07.

Compound 13e. The general was followed from the N-carboxyalkyl-β-lactam 9e (0.42 mmol, 0.2 g) and α-aminoisobutyric acid benzyl ester (0.44 mmol; 0.085 g): yield 0.023 g (66%); mp 64–65 °C; [α] $^{25}_{\rm D}$ +9.4 (0.65, CH₂Cl₂); IR (KBr) $\nu_{\rm max}$ 3350, 2928, 1744, 1726, 1682, 1536, 1148 cm $^{-1}$; 1 H NMR (500 MHz, CDCl₃) $\delta_{\rm H}$ 7.95–7.61 (m, 4H), 7.37–7.26 (m, 15H), 6.44 (br s, 1H), 5.81 (br s, 1H), 5.33 (s, 1H), 5.19 (d, 1H, J = 12.5 Hz), 5.15 (d, 1H, J = 12.5 Hz), 3.76 (d, 1H, J = 5.5 Hz), 3.52 (d, 1H, J = 5.5 Hz), 3.25 (dd, 2H, J = 11.9 Hz), 1.59 (s, 3H), 1.56 (s, 3H); 13 C NMR (125 MHz, CDCl₃) $\delta_{\rm C}$ 173.9, 167.4, 166.7, 147.3, 135.6, 135.3, 133.6, 133.5, 133.0, 131.1, 129.9, 129.2, 129.1, 129.0, 128.7, 128.5, 128.2, 127.9, 125.4, 69.3, 67.5, 59.7, 57.1, 53.6, 41.1, 25.0, 24.5; MS m/z (ESI, positive polarity) MS 693.3; MS2(693.2) 666.1, 665.2, 478.2, 361.2; MS3(665.2) 479.1, 478.2, 463.2. Anal. Calcd for C₃₅H₃₄N₄O₈S: C, 62.67; H, 5.11; N, 8.35. Found C, 62.32; H, 5.06; N, 7.98.

General Procedure for the Preparation of Boc-(β -lactam)-Aa-Aib-OH Compounds 6b–e. Thiophenol (5.00 mmol, 0.469 mL) and K₂CO₃ (4.00 mmol, 0.567 g) were added to a solution of the corresponding Ns-(β -Lactam)-Aa-Aib-OBn 13b–e (1 mmol) in dry MeCN (15 mL), and the suspension was stirred at room temperature for 2 h (monitoried by TLC and ¹H NMR). Upon completion, the

reaction mixture was filtered through a pad of Celite, evaporated in vacuo, and quickly purified by column chromatography (CH₂Cl₂, then MeOH). The corresponding intermediate amine H-(β -lactam)-Aa-Aib-OBn was redissolved in MeCN (15 mL), di-tert-butyl dicarbonate (0.402 mmol; 0.088 g) was added, and the solution was stirred overnight at room temperature. The solution was evaporated, treated with saturated aqueous NaHCO3, and extracted with CH_2Cl_2 (3 × 10 mL). The combined organic solutions were dried (MgSO₄), and the solvents were evaporated under reduced pressure to afford the corresponding intermediate Boc-(β-lactam)-Aa-Aib-OBn, which was purified by preparative TLC (hexanes/EtOAc, 2:1). This benzyl ester was dissolved in ethanol (15 mL) containing 10% Pd/C (0.050 g), and the suspension was stirred at room temperature under hydrogen for 16 h. After this time, the reaction mixture was filtered through a pad of Celite, washed with ethanol ($2 \times 10 \text{ mL}$), and the combined organic solution was evaporated under reduced pressure to afford the corresponding carboxyl peptide product 6b-e.

2-[2-[(3R)-3-Benzyl-3-(tert-butoxycarbonylamino)-2-oxoazetidin-1-yl]-(R)-propanamido]-2-methylpropanoic Acid (6b). The general procedure was followed from compound 13b (1 mmol, 0.608 g): yield 0.342 g (79%); oil; [α] $^{25}_{D}$ –15.0 (0.2, CH₂Cl₂); IR (KBr) ν_{max} 3320, 3180, 1769, 1748 cm $^{-1}$. H NMR (500 MHz, CD₃OD) δ_{H} 7.31–7.29 (m, 5H, Ar), 4.0 (m, 1H, CHMe), 3.67 (d, 1H, J = 4.8 Hz, CH₂NCO), 3.35 (d, 1H, J = 4.8 Hz, CH₂NCO), 3.07 (d, 1H, J = 13.3 Hz, CH₂Ph), 2.97 (d, 1H, J = 13.3 Hz, CH₂Ph), 1.47 (s, 9H, tBuO), 1.46 (s, 6H, Me₂CH), 0.58 (d, 3H, J = 7.1 Hz, MeCH); 13 C NMR (125 MHz, CD₃OD) δ_{C} 181.2, 171.2, 171.0, 156.6, 135.6, 131.8, 129.4, 128.4, 81.0, 68.8, 59.5, 50.8, 39.1, 28.7, 26.9, 25.8, 13.9; HPLC-MS, MeOH/HCO₂H m/z (ESI, positive polarity) MS+23 264.0, 242.0; MS2(242.0) 129.0; MS3(129.0) 101.1, 58.4. Anal. Calcd for C₂₂H₃₁N₃O₆: C, 60.95; H, 7.21; N, 9.69. Found C, 61.33; H, 6.95; N, 9.60.

2-[2-[(3R)-3-Benzyl-3-(tert-butoxycarbonylamino)-2-oxoazetidin-1-yl]-(S)-propanamido]-2-methylpropanoic acid (6c). The general procedure was followed from compound 13c (1 mmol, 0.608 g): overall yield 0.346 g (80%); white solid; mp 216–218 °C; [α]²⁵_D –28.1 (0.36, CH₂Cl₂); IR (KBr) ν_{max} 2974, 1739, 1698, 1583, 1363, 1166 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ_{H} 7.77 (s, 1H, NHCMe₂), 7.39–7.19 (m, 5H, Ar), 4.9 (s, 1H, BocHN), 4.34 (m, 1H, CHMe), 3.65 (d, 1H, J = 4.4 Hz, CH₂NCO), 3.38 (d, 1H, J = 5.4 Hz, CH₂NCO), 3.19 (d, 1H, J = 13.5 Hz, CH₂Ph), 2.94 (d, 1H, J = 13.5 Hz, CH₂Ph), 1.55 (s, 3H, Me_2 C), 1.55 (s, 3H, Me_2 C), 1.45 (s, 9H, tBuO), 0.85 (d, 3H, J = 7.25 Hz, MeCH); ¹³C NMR (125 MHz, CDCl₃) δ_{C} 180.6, 170.2, 169.8, 155.5, 134.6, 130.7, 128.4, 127.4, 79.9, 67.7, 58.4, 49.8, 38.1, 29.7, 27.7, 25.8, 24.8, 12.9. HRMS (TOF CI) m/z 433.2221, C₂₂H₃₁N₃O₆ requires 433.2216.

2-[[(3R)-3-Benzyl-3-(tert-butoxycarbonylamino)-2-oxoazetidin1yl]-(R)-phenylacetamido]-2-methylpropanoic Acid (**6d**). The general procedure was applied from β -lactam 13d (1.00 mmol, 0.670 g): yield 0.355 g (82%); white solid; mp = 238 °C; $[\alpha]^{25}$ D +35.8 (0.65, MeOH); IR (KBr) ν_{max} 3320, 2928, 2851, 1744, 1686, 1670, 1651 cm⁻¹; ¹H NMR (500 MHz, CD₃OD) $\delta_{\rm H}$ 7.32–6.64 (m, 10H), 5.03 (s, 1H), 3.58 (d, 1H, J = 5.5 Hz), 3.02 (d, 1H, J = 13.5 Hz), 2.85 (d, 1H, J= 13.5 Hz), 2.83 (d, 1H, J = 5.5 Hz), 1.61 (s, 3H), 1.57 (s, 3H), 1.50 (s, 9H); 13 C NMR (125 MHz, CD₃OD) $\delta_{\rm C}$ 176.8, 169.7, 168.9, 155.4, 133.7, 133.6, 130.3, 128.7, 128.3, 127.8, 127.2, 79.8, 67.6, 58.6, 56.8, 48.5, 27.3, 25.4, 23.7. HPLC-MS, MeOH/HCO₂H m/z (ESI, positive polarity) 566.4; MS2(566.4) 436.1, 435.0, 434.2, 374.1, 202.1; MS3(434.2) 375.1, 374.1, 188.0, 187.0, 178.0, 173.0, 172.0, 159.0, 131.0, 130.0; MS4(374.1) 188.0, 187.0, 172.0, 170.0, 159.0, 130.9. Anal. Calcd for C₂₇H₃₃N₃O₆: C, 65.44; H, 6.71; N, 8.48. Found C, 65.30; H, 6.49; N, 8.13.

2-[[(3R)-3-Benzyl-3-(tert-butoxycarbonylamino)-2-oxoazetidin1-yl](S)-phenylacetamido]-2-methylpropanoic Acid (6e). The general procedure was applied from β-lactam 13e (1.00 mmol, 0.670 g): yield 0.346 g (80%); mp 113–115 °C; [α] $^{25}_{\rm D}$ +44.5 (0.3, MeOH); IR (KBr) $\nu_{\rm max}$ 3326, 2980, 2916, 2853, 1744, 1698, 1674, 1660 cm $^{-1}$; 1 H NMR (500 MHz, DMSO- 4 6) $\delta_{\rm H}$ 8.41 (s, 1H), 7.30–7.22 (m, 10H), 5.29 (s, 1H), 3.48 (br s, 1H), 3.32 (br s, 1H), 3.17–3.10 (br s, 2H), 1.36 (br s, 1SH). 13 C NMR (75 MHz, DMSO- 4 6) δ 176.8, 168.1,

167.3, 154.8, 136.3, 136.1, 130.5, 128.7, 128.5, 128.4, 128.1, 127.0, 79.2, 67.5, 58.4, 56.0, 51.4, 38.8, 28.6, 25.2, 25.0. HPLC-MS m/z (ESI, negative polarity) 494.3; MS2(494.3) 420.0; MS3(420.0) 377.0, 334.0, 333.1, 291.0, 273.0, 260.9, 249.9, 248.0, 174.8, 127.9; MS4(377.0) 333.0, 275.9, 249.9, 247.9, 207.0, 159.9, 127.9 188.0, 187.0, 172.0, 170.0, 159.0, 130.9. Anal. Calcd for $C_{27}H_{33}N_3O_6$: C, 65.44; H, 6.71; N, 8.48. Found C, 65.14; H, 6.98; N, 8.20.

Preparation of 2-[[(3R)-3-Benzyl-3-(tert-butoxycarbonylamino)2-oxoazetidin1-yl]dimethylacetamido]-2-methylpropa**noic Acid (6f).** (3R)-3-Benzyl-3-tert-butoxycarbonyl-1-[1-(methoxycarbonyl)isopropyl]azetidin-2-one. Thiophenol (5 mmol, 0.469 mL) and $\ensuremath{K_2 \text{CO}_3}$ (4 mmol, 0.567 g) were added to a solution of (3R)-3-benzyl-3-(2-nitrobenzenesulfonylamino)-1-[1-(methoxycarbonyl)isopropyl]-azetidin-2-one¹ (1 mmol, 0.461 g) in MeCN (5 mL). The suspension was stirred for 2 h at rt, the solvent was evaporated in vacuo, and the residue was purified by column chromatography (EtOAc, then MeOH) to afford (3R)-3-benzyl-3amino-1-[1-(methoxycarbonyl)isopropyl]azetidin-2-one in 90% yield (0.248 g). To a solution of this material in CH₂Cl₂ (5 mL) was added (Boc)₂O (3 mmol, 0.654g, and the mixture was stirred at rt for 16 h. After evaporation of the solvent, the crude was purified by column chromatography hexanes/EtOAc, 5:1): overall yield 0.188 g (50%); yellow oil; $[\alpha]^{25}_{\rm D}$ +10.2 (0.6, CH₂Cl₂); IR (KBr) $\nu_{\rm max}$ 3318, 1760, 1742, 1714 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) $\delta_{\rm H}$ 7.38–7.25 (m, 5H, Ph), 4.78 (s, 1H, NH), 3.66 (s, 3H, OCH₃), 3.58 (d, 1H, J = 5.2 Hz, CH_2NCO), 3.38 (d, 1H, J = 5.2 Hz, CH_2NCO), 3.11 (s, 2H, CH CH₂Ph), 1.48 (s, 9H, tBuO), 1.35 (s, 3H, CMe₂), 1.21 (s, 3H, CMe₂). ¹³C NMR (75 MHz, CDCl₃) 173.2, 166.9, 153.4, 135.3, 130.3, 128.4, 127.1, 65.8, 58.8, 52.6, 49.9, 39.1, 28.2, 23.7, 23.6; MS m/z (ESI, positive polarity) +MS2 343.1, +MS4 283.0. Anal. Calcd for C₂₀H₂₈N₂O₅: C, 63.81; H, 7.50; N, 7.44. Found: C, 63.64; H, 7.36; N, 7.19.

Compound 6f. To a solution of (3R)-3-benzyl-3-tert-butoxycarbonyl-1-[1-(methoxycarbonyl)isopropyl]azetidin-2-one (1 mmol, 0.376 g) in MeOH (20 mL) was added LiOH·.H₂O (10 mmol, 0.485 g) in 10 mL of H₂O, and the mixture was stirred at rt for 4 h. Acidification of the resulting mixture with 1 M HCl and extraction with CH_2Cl_2 (3 × 8 mL) provided an organic solution of (3R)-3benzyl-3-tert-butoxycarbonyl-1-[1-(carboxy)isopropyl]azetidin-2-one, which was dried (Na₂SO₄) and filtered. H-Aib-OBn (1.3 mmol, 0.248 g) was added, the mixture was cooled to -12 °C, and EEDQ (1.3 mmol, 0.331 g) was added. The reaction mixture was stirred during 16 h, while rt was slowly reached. The resulting solution was washed with 1 M HCl (3 \times 10 mL), the aqueous layer was extracted with CH₂Cl₂, and the combined organic phases were dried (MgSO₄) and evaporated. The crude was purified by flash column chromatography (hexanes/EtOAc, 5:1) to afford benzyl 2-[[(3R)-3-benzyl-3-(tertbutoxycarbonylamino)2-oxoazetidin1-yl]dimethylacetamido]-2-methylpropanoate: white solid; 0.494 g (92%). A suspension of this benzyl ester (0.92 mmol) and 20% Pd/C in EtOAc (35 mL) was stirred for 1 h at rt under H2 atmosphere. The mixture was filtered through a Celite pad, and the filtrate was evaporated: yield 0.411 g (100%); white solid; mp 213–215 °C; $[\alpha]^{25}_{D}$ +65.2 (0.48, CH₂Cl₂); IR (KBr) 3346, 1736, 1702, 1685 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ_{H} 7.87 (s, 1H, NHCMe₂), 7.38–7.27 (m, 5H, Ar), 4.84 (s, 1H, BocNH), 3.62 (d, 1H, $I = 5.1 \text{ Hz}, \text{CH}_2\text{NCO}$, 3.35 (d, 1H, $I = 5.1 \text{ Hz}, \text{CH}_2\text{NCO}$), 3.14 (d, 1H, J = 13.4 Hz, CH₂Ph), 2.89 (d, 1H, J = 13.4 Hz, CH₂Ph), 1.58 (s, 3H, Me₂C), 1.57 (s, 3H, Me₂C), 1.45 (s, 9H, tBuO), 1.32 (s, 3H, Me₂C), 0.95 (s, 3H, Me₂C); ¹³C NMR (75 MHz, CDCl₃) 175.8, 174.5, 168.3, 154.7, 133.5, 130.8, 129.1, 128.2, 81.6, 66.4, 59.9, 57.9, 48.6, 39.5, 28.7, 25.9, 25.3, 24.4, 23.8; MS m/z (ESI, positive polarity) MS 446.2, MS2(446.2) 372.8, 372.1, MS3(372.1) 329.0, MS4(329.0) 329.1, 242.9, 201.9, 199.9, 172.0, 168.9, 153.9, 127.9, 111.0. Anal. Calcd for C₂₃H₃₃N₃O₆: C, 61.73; H, 7.43; N, 9.39. Found: C, 61.90; H, 7.26; N, 9.34.

General Procedure for the Preparation of Boc-Ala-(α -Bn- β -lactam)-Aib-OMe Compounds 15 and 16. Thiophenol (5 mmol, 0.469 mL) and K₂CO₃ (4 mmol, 0.567 g) were added to a stirred solution of (3R)-3-benzyl-1-[1-(methoxycarbonyl)isopropyl]-3-(2-nitrobenzenesulfonylamino)azetidin-2-one 14 (1 mmol, 0.461 g) in

MeCN (5 mL). The mixture was stirred over 2 h, the solvent was evaporated in vacuo, and the crude was purified by flash column chromatography (EtOAc, then MeOH): yield 90%. A solution of the resulting amine and the corresponding Boc-Ala-OH (1.3 mmol, 0.248 g) in anhydrous $\rm CH_2Cl_2$ (20 mL) was cooled to $\rm -12$ °C, and EEDQ (1.3 mmol, 0.331 g) was added. The mixture was stirred during 16 h, while rt was slowly reached. The resulting solution was washed with 1 M HCl (3 \times 10 mL), the aqueous layer was extracted with $\rm CH_2Cl_2$ and the combined organic phases were dried (MgSO₄) and evaporated. The product was purified by flash column chromatography (hexanes/EtOAc, 2:1).

Methyl 2-[[(3R)-3-Benzyl-3-[2-(tert-butoxycarbonylamino)-(2R)propanamido]2-oxoazetidin1-yl]-2-methylpropanoate (15). The general procedure was followed from compound 17 (1 mmol, 0.461 g) and Boc-D-Ala-OH (1.3 mmol, 0.248 g): yield 0.371 g (83%); white solid; mp 135–137 °C; $[\alpha]^{25}_{D}$ +28.5 (0.2, CH₂Cl₂); IR (KBr) ν 3309, 3000, 1763, 1752, 1704, 1656, 1160, 672 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) $\delta_{\rm H}$ 7.28–7.32 (m, 5H, Ar), 6.62 (bs, 1H, CONHCCH₂Ph), 4.9 (bs, 1H, CONHCH(CH₃)CO), 4.14 (m, 1H, NHCH(Me)CO), 3.71 (s, 3H, CO_2Me), 3.63 (d, 1H, J = 5.5 Hz, NCH_2CCH_2Ph), 3.45 (d, 1H, J = 5.4 Hz, NCH_2CCH_2Ph), 3.21 (d, 1H, J = 14.2 Hz, CH_2Ph), 3.18 (d, 1H, J = 14.0 Hz, CH_2Ph), 1.45 (s, 9H, tBuO), 1.43 (s, 3H, Me_2CCO_2Me), 1.36 (s, 3H, Me_2CCO_2Me), 1.31 (d, 3H, J = 7.1Hz, CH(Me)CONH); ¹³C NMR (75 MHz, CDCl₃) 173.1, 172.7, 166.8, 155.3, 134.8, 130.2, 128.4, 127.2, 80.2, 65.7, 58.9, 52.5, 49.4, 38.7, 28.3, 23.6, 18.3, 16.8; HPLC-MS, MeOH/HCO₂H m/z (ESI, positive polarity) 187.1 (35), 133.1 (99), 126.1 (62), 112.1 (41), 97.1 (41), 72.1 (85), 59.1 (100), 57.1 (48), 55.1 (53); HRMS (TOF CI) m/z 447.2372, $C_{23}H_{33}N_3O_6$ requires 447.2369.

Methyl 2-[(3R)-3-Benzyl-3-[2-(tert-butoxycarbonylamino)(2S)propanamido]2-oxoazetidin-1-yl]-2-methylpropanoate (16). The general procedure was followed from compound 17 (1 mmol, 0.461 g) and Boc-Ala-OH (1.3 mmol, 0.248 g): yield 0.380 g (85%); yellowish solid; $[\alpha]^{25}_{D}$ –22.4 (0.5, CH₂Cl₂); IR (KBr) ν 3271, 2993, 1753, 1721, 1716, 1673, 1167, 740 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) $\delta_{\rm H}$ 7.32 (m, 5H, Ar), 4.08 (m, 1H, COCH(Me)NH), 3.65 (s, 3H, CO₂Me), 3.59 (d, 1H, J = 5.5 Hz, NCH₂C), 3.41 (d, 1H, J = 5.3Hz, NC H_2 C), 3.15 (dd, 2H, J = 13.7 Hz, CH $_2$ Ph), 1.45 (s, 9H, tBuO), 1.33 (s, 3H, CMe₂CO₂Me), 1.29 (d, 1H, I = 6.9 Hz, CHMeNH), 1.23 (s, 3H, CMe₂CO₂Me); ¹³C NMR (75 MHz, CDCl₃) 173.2, 172.9, 155.3, 134.9, 130.3, 128.4, 127.1, 79.6, 65.9, 58.9, 53.5, 52.5, 49.8, 38.5, 28.3, 23.4, 18.3; HPLC-MS, MeOH/HCO₂H m/z (ESI, positive polarity) 187.1 (32), 134.1 (28), 133.1 (100), 115.6 (27), 102.5 (21), 92.1 (33), 91.1 (41), 58 (56); HRMS (TOF CI) m/z 447.2348, C₂₃H₃₃N₃O₆ requires 447.2369.

General Procedure for the Preparation of Boc-Ala-(β-Lactam)-Aib-Ala-NH₂ Compounds 7a-d. To a solution of the corresponding methyl 2-[(3R)-3-benzyl-3-[2-(tert-butoxycarbonylamino)2R-propanamido]2-oxoazetidin-1-yl]-2-dimethylpropanoate 15 or 16 (1 mmol, 0.447 g) in MeOH (20 mL) was added a solution of LiOH·H₂O (10 mmol, 0.485 g) in MeOH/H₂O 10:5 mL. The mixture was stirred for 2 h and acidified with 1 M HCl (5 mL), and the aqueous phase was extracted with CH2Cl2 (3 × 15 mL) to afford the intermediate carboxylic acid in 85% yield. To this crude material dissolved in anhydrous DMF (15 mL) cooled to 0 $^{\circ}\text{C}$ was added successively the corresponding alaninamide (1.5 mmol, 0.131 g), HATU (1.5 mmol, 0.570 g), HOAT (1.4 mmol, 0.191 g), and KHCO₃ (15 mmol, 1.5 g). The mixture was stirred during 16 h while slowly reaching rt. The solvent was evaporated at low pressure, and the residue was redissolved in CH_2Cl_2 and washed with 1 M HCl (3 × 10 mL) and satd NaHCO₃ (3 \times 10 mL). After evaporation of the organic solvent, the crude product was purified by flash column chromatography (hexanes/EtOAc, 5:1 or CH₂Cl₂/MeOH, 15:1).

2-[2-[(3R)-Benzyl-3-[2-(tert-butoxycarbonylamino)-(2R)-propanamido]2-oxoazetidin1-yl]-2-methylpro panamido]-(2R)-propanamide (7a). The general procedure was followed from methyl 2-[(3R)-3-benzyl-3-[2-(tert-butoxycarbonylamino)(2R)-propanamido]2-oxoazetidin1-yl]-2-dimethylpropanoate 15 (1 mmol, 0.447 g) and (R)-alaninamide 18 (1.5 mmol, 0.131 g): yield 0.629 g (80%); white solid; mp 190–192 °C; $[\alpha]^{25}_{\rm D}$ +36.0 (0.30, CH₂Cl₂); IR (KBr)

 $\nu_{\rm max}$ 3416, 3308, 2965, 1741, 1726, 1694, 1536, 691 cm $^{-1}$; $^{1}{\rm H}$ NMR (500 MHz, DMSO- d_6) δ_H 8.3 (s, 1H, NHCCH,Ph), 8.16 (d, 1H, J =6.9 Hz, NHCH(Me)CONH₂), 7.33-7.28 (m, 5H, Ph), 7.1 (bs, 1H, NH_2), 6.87 (d, 1H, J = 8.1 Hz, BocNH), 6.83 (bs, 1H, NH_2), 4.13 (m, 1H, BocHNCH(Me), 4.06 (m, 1H, CH(Me)CONH₂), 3.58 (d, 1H, I = 5.1 Hz, CH_2NCMe_2), 3.26 (d, 1H, J = 5.0 Hz, CH_2NCMe_2), 3.05 (d, 1H, J = 13.4 Hz, CH₂Ph), 3.02 (d, 1H, J = 13.5 Hz, CH₂Ph), 1.39 (s, 9H, tBuO), 1.29 (d, 3H, I = 7.2 Hz, $CH(Me)CONH_2$), 1.20 (d, 3H, I= 6.9 Hz, BocHNCH(Me), 1.12 (s, 3H, Me₂CCO), 0.7 (s, 3H, Me₂CCO); ¹³C NMR (75 MHz, CDCl₃) 175.0, 173.6, 173.2, 166.8, 156.1, 133.1, 130.3, 128.7, 127.8, 80.9, 77.2, 65.8, 59.8, 49.6, 38.9, 28.3, 24.3, 23.6, 15.9, 16.5; HPLC-MS, MeOH/HCO₂H m/z (ESI, positive polarity) 217.2 (19), 167.1 (33), 149.0 (100), 119.1 (26), 85.9 (29); HRMS (TOF CI) m/z 503.2763, requires 503.2744. Anal. Calcd for C₂₅H₃₇N₅O₆: C, 59.63; H, 7.41; N, 13.91. Found: C, 59.70; H, 7.01; N, 13.91.

2-[2-[(3R)-Benzyl-3-[2-(tert-butoxycarbonylamino)-(2R)propanamido]2-oxoazetidin1-yl]2-methylpro panamido]-(2S)propanamide (7b). The general procedure was followed from methyl 2-[(3R)-3-benzyl-3-[2-(tert-butoxycarbonylamino)(2R)-propanamido]2-oxoazetidin1-yl]-2-dimethylpropanoate 15 (1 mmol, 0.447 g) and (S)-alaninamide 18 (1.5 mmol, 0.131 g): yield 0.327 g (65%); white solid; mp 202–204 °C; $[\alpha]_{D}^{25}$ –80.0 (0.50, CH₂Cl₂); IR (KBr) $\nu_{\rm max}$ 3412, 3213, 1733, 1730, 1681, 1629, 696, 734 cm $^{-1}$; ¹H NMR $(500 \text{ MHz}, DMSO-d_6) 8.41 \text{ (s, 1H, NHCCH}_2\text{Ph)}, 8.30 \text{ (d, 1H, } J = 6.7)$ Hz, NHCH(Me)CONH₂), 7.33-7.28 (m, 5H, Ph), 7.10 (s, 1H, NH_2), 6.88 (d, 1H, J = 7.8 Hz, NHBoc), 6.84 (s, 1H, NH_2), 4.13 (m, 1H, tBuOCONHCHMe), 4.06 (m, 1H, CH(Me)CONH₂), 3.58 (d, 1H, J = 5.3 Hz, CH_2NCMe_2), 3.26 (d, 1H, J = 5.0 Hz, CH_2NCMe_2), 3.06 (d, 1H, J = 13.4 Hz, CH₂Ph), 3.02 (d, 1H, J = 13.6 Hz, CH₂Ph), 1.39 (s, 9H, tBuO), 1.29 (d, 3H, J = 7.1 Hz, $CH(Me)CONH_2$), 1.20 (d, 3H, J = 6.9 Hz, BocHNCH(Me), 1.09 (s, 3H, Me₂CCO), 0.75 (s, 3H, Me₂CCO); ¹³C NMR (75 MHz, CDCl₃) 175.2, 175.1, 173.5, 173.3, 167.0, 156.1, 133.1, 130.3, 128.7, 127.8, 80.8, 77.2, 65.8, 59.8, 49.7, 47.5, 38.8, 28.3, 24.3, 16.6; HPLC-MS, MeOH/HCO₂H m/z (ESI, positive polarity) 119.1 (88), 97.1 (60), 91.1 (72), 71.1 (67), 69.1 (63), 57.1 (100), 55.1 (90); HRMS (TOF CI) m/z 503.2727, $C_{25}H_{37}N_5O_6$ requires 503.2744.

2-[2-[(3R)-Benzyl-3-[2-(tert-butoxycarbonylamino)(2S)propanamido]2-oxoazetidin1-yl]2-methylpro panamido]-(2S)propanamide (7c). The general procedure was followed from methyl 2-[(3R)-3-benzyl-3-[2-(tert-butoxycarbonylamino)(2S)-propanamido]2-oxoazetidin-1-yl]-2-dimethylpropanoate 16 (1 mmol, 0.447 g) and (S)-alaninamide 18 (1.5 mmol, 0.131 g): yield 0.326 g (73%); yellowish solid; mp 160–162 °C; $[\alpha]^{25}_{D}$ +66.0 (0.20, CH₂Cl₂); IR (KBr) ν_{max} 3334, 2969, 1748, 1682, 1536, 1165 cm⁻¹; ¹H NMR (500 MHz, DMSO- d_6) δ_H 8.32 (s, 1H, NHCCH₂Ph), 8.16 (d, 1H, J = 6.5Hz, NHCH(Me)CONH2), 7.33-7.28 (m, 5H, Ph), 7.06 (bs, 1H, NH_2), 6.91 (d, 1H, I = 7.0 Hz, NHBoc), 6.77 (bs, 1H, NH_2), 4.08 (m, 1H, BocNHCHMe), 4.02 (m, 1H, $CH(Me)CONH_2$), 3.61 (d, 1H, J =4.7 Hz, CH_2NCMe_2), 3.24 (d, 1H, J = 4.8 Hz, CH_2NCMe_2), 3.05 (d, 1H, J = 13.4 Hz, CH₂Ph), 3.01 (d, 1H, J = 13.5 Hz, CH₂Ph), 1.38 (s, 9H, tBuO), 1.31 (d, 3H, J = 7.2 Hz, $CH(Me)CONH_2$), 1.20 (d, 3H, J= 7.2 Hz, BocNHCHMe), 1.12 (s, 3H, Me₂CCO), 0.7 (s, 3H, Me₂CCO); ¹³C NMR (75 MHz, CDCl₃) 175.0, 173.6, 173.0, 166.7, 155.9, 133.0, 130.3, 128.7, 127.8, 80.9, 75.8, 65.9, 59.8, 49.6, 47.6, 38.9, 28.3, 24.3, 23.6, 16.5; HPLC-MS, MeOH/HCO₂H *m/z* (ESI, positive polarity) 187.1 (49), 133.1 (37), 91.1 (100), 70.1 (60), 57.1 (42). HRMS (TOF CI) m/z 503.2771, requires 503.2744. Anal. Calcd for C₂₅H₃₇N₅O₆: C, 59.63; H, 7.41; N, 13.91. Found: C, 59.85; H, 7.34; N. 13.92

2-[2-[(3R)-Benzyl-3-[2-(tert-butoxycarbonylamino)(2S)-propanamido]2-oxoazetidin1-yl]2-methylpro panamido]-(2R)-propanamide (7**d**). The general procedure was followed from methyl 2-[(3R)-3-benzyl-3-[2-(tert-butoxycarbonylamino)(2S)-propanamido]2-oxoazetidin1-yl]-2-dimethylpropanoate 16 (1 mmol, 0.447 g) and (R)-alaninamide 18 (1.5 mmol, 0.131 g): yield 0.392 g (77%); yellowish solid; mp 219–220 °C; [α] 2S _D –15.9 (0.20, CH₂Cl₂); IR (KBr) ν _{max} 3413, 3303, 2978, 2914, 1744, 1663, 1542 cm $^{-1}$; 1 H NMR (500 MHz, DMSO-d₆) δ _H 8.53 (s, 1H, NHCCH₂Ph), 8.37 (d, 1H, J =

7.3 Hz, NHCH(Me)CONH₂), 7.33–7.28 (m, 5H, Ar), 7.03 (d, 1H, *J* = 6.7 Hz, BocNH), 6.97 (s, 1H, NH₂), 6.88 (s, 1H, NH₂), 4.04 (m, 1H, BocNHCHMe), 4.03 (m, 1H, CH(Me)CONH₂), 3.46 (d, 1H, *J* = 4.8 Hz, CH₂NCMe₂), 3.34 (d, 1H, *J* = 5.0 Hz, CH₂NCMe₂), 3.06 (d, 1H, *J* = 13.4 Hz, CH₂Ph), 1.38 (s, 9H, tBuO), 1.31 (d, 3H, *J* = 7.2 Hz, CH(Me)CONH₂), 1.19 (d, 3H, *J* = 7.0 Hz, BocHNCHMe), 0.97 (s, 3H, Me₂CCO), 0.79 (s, 3H, Me₂CCO); ¹³C NMR (75 MHz, CDCl₃) 175.9, 173.4, 172.5, 167.7, 155.9, 132.7, 130.3, 128.8, 127.9, 80.8, 77.2, 65.6, 58.9, 49.5, 48.1, 38.8, 28.3, 23.7, 17.3; HPLC-MS, MeOH/HCO₂H *m/z* (ESI, positive polarity) 187.1 (44), 173.1 (28), 172.1 (26), 117.1 (37), 116 (25), 115 (26), 113.1 (35), 91 (100), 70.0 (47); HRMS (TOF CI) *m/z* 503.2742, requires 503.2744. Anal. Calcd for C₂₅H₃₇N₅O₆: C, 59.63; H, 7.41; N, 13.91. Found: C, 59.67; H, 7.62; N, 13.88.

ASSOCIATED CONTENT

S Supporting Information

¹H NMR spectra of compounds 5–7 and 13d–e. ¹³C NMR spectra of compounds 5–13b,c, 15, and 16. NMR/ROESY spectra and interproton distances of compounds 5–7. Gaussian output data of structures 5a–f. X-ray data for 5e, 5f, 7a, and 7d (CIF). This material is available free of charge via the Internet at http://pubs.acs.org.

AUTHOR INFORMATION

Corresponding Author

*E-mail: jesusmaria.aizpurua@ehu.es, claudio.palomo@ehu.es.

Notes

The authors declare no competing financial interest. X-ray crystallography was performed at the University of Zurich.

ACKNOWLEDGMENTS

We thank the Ministerio de Ciencia y Competitividad (MEC, Spain) (Project No. CTQ2010-21625-C02-01), Gobierno Vasco (ETORTEK-nanoIKER IE-11/304), and Universidad del País Vasco UPV/EHU (UFI QOSYC 11/22) for financial support and SGIker UPV/EHU for NMR facilities. Grants from Gobierno Vasco to M.S.-A. and from UPV/EHU to E.A. are acknowledged.

REFERENCES

- (1) (a) Mahalakshmi, R.; Balaram, P. The Use of D-Amino Acids in Peptide Design. In D-Amino Acids: A New Frontier in Amino Acid and Protein Research Practical Methods and Protocols; Konno, R., Brückner, H., D'Aniello, A., Fischer, G. H., Fujii, N., Homma, H., Eds.; Nova Science Publishers: New York, 2007; pp 415–430. (b) Brückner, H.; Fuji, N. D-Amino Acids in Chemistry, Life Sciences, and Biotechnology; Wiley: New York, 2010.
- (2) (a) Venkatachalam, M. Biopolymers 1968, 6, 1425–1436. (b) Marcelino, A. M. C.; Gierasch, L. M. Biopolymers 2008, 89, 380–391 and references cited therein. (c) Rotondi, K. S.; Gierasch, L. M. Biopolym. (Peptide Sci.) 2006, 84, 13–22. (d) Tyndall, J. D. A.; Pfeiffer, B.; Abbenante, G.; Fairlie, D. P. Chem. Rev. 2005, 105, 793–826. (e) Vass, E.; Hollósi, M.; Besson, F.; Buchet, R. Chem. Rev. 2003, 103, 1917–1954. (f) Ball, J. B.; Hughes, R. A.; Alewood, P. F.; Andrews, P. R. Tetrahedron 1993, 49, 3467–3478.
- (3) (a) Bystrov, V. F.; Portnova, S. L.; Tsetlin, V. I.; Ivanov, V. T.; Ovchinnikov, Y. A. *Tetrahedron* 1969, 25, 493–515. (b) Zhu, Y.; Li, T.; Li, D.; Zhang, Y.; Xiong, W.; Sun, J.; Tang, Z.; Chen, G. *Amino Acids* 2012, 42, 1749–1755 and references cited therein. For selected γ-turn mimetics, see: (c) Saavedra, C.; Boto, A.; Hernandez, R.; Miranda, J. I.; Aizpurua, J. M. *J. Org. Chem.* 2012, 77, 5907–5913. (d) Rosenstroem, U.; Skoeld, C.; Plouffe, B.; Beaudry, H.; Lindeberg, G.; Botros, M.; Nyberg, F.; Wolf, G.; Karlen, A.; Gallo-Payet, N.; Hallberg, A. *J. Med. Chem.* 2005, 48, 4009–4024. (e) Yuan, Z.;

- Blomberg, D.; Sethson, I.; Brickmann, K.; Ekholm, K.; Johansson, B.; Nilsson, A.; Kihlberg, J. *J. Med. Chem.* **2002**, *45*, 2512–2519. (f) Chou, K. C. *Anal. Biochem.* **2000**, *286*, 1–16.
- (4) See, for example: (a) Conti, P.; Tamborini, L.; Pinto, A.; Blondel, A.; Minoprio, P.; Mozzarelli, A.; De Micheli, C. Chem. Rev. 2011, 111, 6919–6946. (b) Nickl, C. K.; Raidas, S. K.; Zhao, H.; Sausbier, M.; Ruth, P.; Tegge, W.; Brayden, J. E.; Dostmann, W. R. Biochim. Biophys. Acta 2010, 1804, 524–532. (c) Schumacher, T. N. M.; Mayr, L. M., Jr.; D. L., M.; Milhollen, M. A.; Burgess, M. W.; Kim, P. S. Science 1996, 271, 1854–1857.
- (5) (a) Freidinger, R. M. J. Med. Chem. 2003, 46, 5553–5566. (b) Veber, D. F.; Freidinger, R. M.; Perlow, D. S.; Paleveda, W. J., Jr.; Holly, F.; Strachan, R. G.; Nutt, R. F.; Arison, B. H.; Homnick, C.; Randall, W. C.; Glitzer, M. S.; Saperstein, R.; Hirshmann, R. A. Nature 1981, 292, 55–58.
- (6) (a) Robinson, J. A. Acc. Chem. Res. 2008, 41, 1278-1288.
 (b) Haque, T. S.; Little, J. C.; Gellman, S. H. J. Am. Chem. Soc. 1996, 118, 6975-6985.
- (7) (a) Synthesis of Peptides and Peptidomimetics. In Houben-Weyl, Methods of Organic Chemistry; Felix, A., Moroder, L., Toniolo, C., Eds.; Thieme: Stuttgart, 2003; Vol. E22c. (b) Burgess, K. Acc. Chem. Res. 2001, 34, 826–835. (c) MacDonald, M.; Aube, J. Curr. Org. Chem. 2001, 5, 417–438. (d) Souers, A. J.; Ellman, J. A. Tetrahedron 2001, 57, 7431–7448. (e) Peczuh, M. W.; Hamilton, A. D. Chem. Rev. 2000, 100, 2479–2494. (f) Hanessian, S.; McNaughton-Smith, G.; Lombart, H.-G.; Lubell, W. D. Tetrahedron 1997, 53, 12789–12854. (g) Gante, J. Angew. Chem., Int. Ed. Engl. 1994, 33, 1699–1720. (h) Giannis, A.; Kolter, T. Angew. Chem., Int. Ed. Engl. 1993, 32, 1244–1267.
- (8) For some recent selected examples, see: (a) Sacchetti, A.; Silvani, A.; Lesma, G.; Pilati, T. J. Org. Chem. 2011, 76, 833–839. (b) Larregola, M.; Lequin, O.; Karoyan, P.; Guianvarc'h, D.; Laville, S. J. Pept. Sci. 2011, 17, 632–643. (c) Fustero, S.; Mateu, N.; Albert, L.; Aceña, J. L. J. Org. Chem. 2009, 74, 4429–4432. (d) De Wachter, R.; Brans, L.; Ballet, S.; Van den Eynde, I.; Feytens, D.; Keresztes, A.; Toth, G.; Urbanczyk-Lipkowska, Z.; Tourwé, D. Tetrahedron 2009, 65, 2266–2278. (e) Bittermann, H.; Boeckler, F.; Einsiedel, J.; Gmeiner, P. Chem.—Eur. J. 2006, 12, 6315–6322. (f) Duggan, H. M. E.; Hitchcock, P. B.; Young, D. W. Org. Biomol. Chem. 2005, 3, 2287–2295.
- (9) (a) Lessing, J.; Roy, S.; Reppert, M.; Baer, M.; Marx, D.; La Cour Jansen, T.; Knoester, J.; Tokmakoff, A. J. Am. Chem. Soc. 2012, 134, 5032–5035. (b) De Poli, M.; Moretto, A.; Crisma, M.; Peggion, C.; Formaggio, F.; Kaptein, B.; Broxterman, Q. B.; Toniolo, C. Chem.—Eur. J. 2009, 15, 8015–8025. (c) Liang, G.-B.; Rito, C. J.; Gellman, S. H. J. Am. Chem. Soc. 1992, 114, 4440–4442.
- (10) (a) Palomo, C.; Aizpurua, J. M.; Benito, A.; Miranda, J. I.; Fratila, R. M.; Matute, C.; Domercq, M.; Gago, F.; Martin-Santamaria, S.; Linden, A. *J. Am. Chem. Soc.* 2003, 125, 16243–16260. (b) Palomo, C.; Aizpurua, J. M.; Benito, A.; Galarza, R.; Khamrai, U. K.; Vazquez, J.; de Pascual-Teresa, B.; Nieto, P. M.; Linden, A. Angew. Chem., Int. Ed. Engl. 1996, 35, 1239–1241.
- (11) For alternative β-lactams as reverse turn nucleators, see:
 (a) Bari, S.; Bhalla, A. Top. Heterocycl. Chem. 2010, 22, 49–99.
 (b) Bittermann, H.; Gmeiner., P. J. Org. Chem. 2006, 71, 97–102.
 (c) Khasanov, A. B.; Ramirez-Weinhouse, M. M.; Webb, T. R.; Thiruvazhi, M. J. Org. Chem. 2004, 69, 5766–5769. (d) Alonso, E.; Lopez-Ortiz, F.; del-Pozo, C.; Peralta, E.; Macia, A.; Gonzalez, J. J. Org. Chem. 2001, 66, 6333–6338.
- (12) The most abundant natural form of β -turn is type III ($\phi_{(i+1)} = -60/\ \psi_{(i+1)} = -30/\ \phi_{(i+2)} = -60/\ \psi_{(i+2)} = -30)$, as part of 3_{10} -helices. Type-I ($\phi_{(i+1)} = -60/\ \psi_{(i+1)} = -30/\ \phi_{(i+2)} = -90/\ \psi_{(i+2)} = 0$) is the second and type-II ($\phi_{(i+1)} = -60/\ \psi_{(i+1)} = 120/\ \phi_{(i+2)} = 80/\ \psi_{(i+2)} = 0$) is the third one. The sum of types I, II, and III accounts for more that 95% of all the β -turns assigned in globular proteins. As types III, I, and II differ from each other by some 30° in the torsional angles $\phi_{(i+2)}$ and $\psi_{(i+2)}$, they are often not distinguished. Thus, when type preference of β -turns is investigated as a function of their amino acid sequence, most commonly only the ratio of type (I + III) to type II is determined. See:

- Perczel, A.; Jákli, I.; Foxman, B. M.; Fasman, G. D. *Biopolymers* **1996**, 38, 723–732.
- (13) Bürgi, H. B.; Dunitz, J. D.; E. Shefter, E. J. Am. Chem. Soc. 1973, 95, 5065–5067.
- (14) Tarbe, M.; Azcune, I.; Balentova, E.; Miles, J. J.; Edwards, E. E.; Miles, K. M.; Do, P.; Baker, B. M.; Sewell, A. K.; Aizpurua, J. M.; Douat-Casassus, C.; Quideau, S. Org. Biomol. Chem. 2010, 8, 5345–5353.
- (15) Glover, S. A.; Rosser, A. A. J. Org. Chem. 2012, 77, 5492-5502.
- (16) The apparent configuration inversion at the azetidinone $C\alpha$ stereocenter was caused by a change in the group preference, according to the Cahn–Ingold–Prelog rules.
- (17) To prepare enantiopure α -alkylserinates from D-serine, see: (a) Di Giacomo, M.; Vinci, V.; Serra, M.; Colombo, L. *Tetrahedron: Asymmetry* **2008**, *19*, 247–257. (b) Seebach, D.; Aebi, J. D.; Gander-Coquoz, M.; Naef, R. *Helv. Chim. Acta* **1987**, *70*, 1194–1216.
- (18) Palomo, C.; Aizpurua, J. M.; Balentová, E.; Jiménez, A.; Oyarbide, J.; Fratila, R.; Miranda, J. I. Org. Lett. 2007, 9, 101–104.
- (19) (a) Chan, T. Y.; Chen, A.; Allason, N.; Chen, R.; Liu, D.; Sofia, M. J. Tetrahedron Lett. 1996, 37, 8097–8100. (b) Sipos, F.; Gaston, D. W. Synthesis 1971, 321–322.
- (20) Maligres, P. E.; See, M. M.; Askin, D.; Reider, P. J. Tetrahedron Lett. 1997, 38, 5253-5256.
- (21) (a) Tulla-Puche, J.; Marcucci, E.; Prats-Alfonso, E.; Bayo-Puxan, N.; Albericio, F. *J. Med. Chem.* **2009**, *52*, 834–839. (b) Ghosh, A. K.; Kulmaki, S. *Org. Lett.* **2008**, *10*, 3907–3909.
- (22) For methods, see: (a) Troganis, A.; Gerothanassis, I. P.; Athanassiou, Z.; Mavromoustakos, T.; Hawkes, G. E.; Sakarellos, C. Biopolymers 2000, 53, 72–83. (b) Kessler, H. Angew. Chem., Int. Ed. Engl. 1982, 21, 512–523. (c) Andersen, N. H.; Neidigh, J. W.; Harris, S. M.; Lee, G. M.; Liu, Z.; Tong, Hui J. Am. Chem. Soc. 1997, 119, 8547–8561.
- (23) (a) Neuhaus, D.; Williamson, M. The Nuclear Overhauser Effect in Structural and Conformational Analysis; VCH Publishers: New York, 1989. (b) Thomas, P. D.; Basus, V. J.; James, T. L. Proc. Nat. Acad. Sci. U.S.A. 1991, 88, 1237–1241.
- (24) For related β -II/ β -III' overlapped peptides, see: (a) Lisowski, M.; Jaremko, L.; Jaremko, M.; Mazur, A.; Latajka, R.; Makowski, M. *Biopolymers* **2010**, *93*, 1055–1064. (b) Inai, Y.; Oshikawa, T.; Yamashita, M.; Kato, I.; Hirabayashi, T. *Chem. Lett.* **2001**, 472–473. Two β -I overlapped peptides: (c) Lo, S.-C.; Li, X.; Henzl, M. T.; Beamer, L. J.; Hannink, M. *EMBO J.* **2006**, 25, 3605–3617. For a review on multiple β -turns (not all overlapped), see: (d) Guruprasad, K.; Prasad, M. S.; Kumar, G. R. *J. Peptide. Res.* **2000**, 56, 250–263.
- (25) For theoretical calculations on peptide conformation, see: (a) Perczel, A.; Jákli, I.; McAllister, M. A.; Csizmadia, I. G. Chem.—Eur. J. 2003, 9, 2551–2556 and references cited therein. (b) Sapse, A. M. Molecular Orbital Calculations for Amino Acids and Peptides; Birkhausen: New York, 2000. For gas-phase experimental studies conducted on configurationally altered short peptides, see: (c) Brenner, V.; Piuzzi, F.; Dimicoli, I.; Tardivel, B.; Mons, M. Angew. Chem., Int. Ed. 2007, 46, 2463–2466. (d) Chin, W.; Dognon, J. P.; Piuzzi, F.; Tardivel, B.; Dimicoli, I.; Mons, M. J. Am. Chem. Soc. 2005, 127, 707–712
- (26) For hydrogen delocalization from the lone pair (n) of the C=O acceptor over the antibonding orbital (σ^*) of the N-H donor, see: Khaliullin, R. Z.; Cobar, E. A.; Lochan, R. C.; Bell, A. T.; Head-Gordon, M. *J. Phys. Chem. A* **2007**, *111*, 8753–8765.
- (27) Epp, J. E.; Widlanski, T. S J. Org. Chem. 1999, 64, 293-295.
- (28) Rychnovsky, S. D.; Skalitzky, D. J. J. Am. Chem. Soc. **1992**, 114, 671–677.
- (29) Shendage, D. M.; Fröhlich, R.; Haufe, G. Org. Lett. 2004, 6, 3675–3678.