

Article

NHC-Cu(I)-Catalyzed Friedländer-Type Annulation of Fluorinated *o*-Aminophenones with Alkynes on Water. Competitive Base-Catalyzed Dibenzo[*b,f*][1,5]diazocine Formation

Paweł Czerwiński, and Michał Michalak

J. Org. Chem., **Just Accepted Manuscript** • DOI: 10.1021/acs.joc.7b01235 • Publication Date (Web): 07 Jul 2017

Downloaded from <http://pubs.acs.org> on July 7, 2017

Just Accepted

“Just Accepted” manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides “Just Accepted” as a free service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. “Just Accepted” manuscripts appear in full in PDF format accompanied by an HTML abstract. “Just Accepted” manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are accessible to all readers and citable by the Digital Object Identifier (DOI®). “Just Accepted” is an optional service offered to authors. Therefore, the “Just Accepted” Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the “Just Accepted” Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these “Just Accepted” manuscripts.



1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

NHC-Cu(I)-catalyzed Friedländer-type annulation of fluorinated o-aminophenones with alkynes on Water. Competitive base-catalyzed dibenzo[*b,f*][1,5]diazocine formation

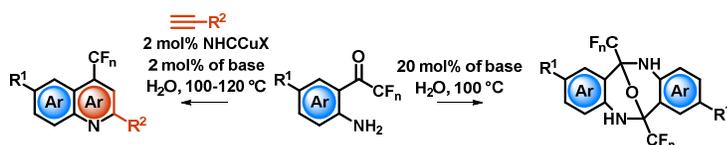
Paweł Czerwiński and Michał Michalak*

Institute of Organic Chemistry, Polish Academy of Sciences

Kasprzaka 44/52, 01-224 Warsaw, Poland

*michal.michalak@icho.edu.pl

Graphical abstract



Abstract: An efficient, easily scalable synthesis of 4-trifluoromethylquinolines and naphthydrines (as well as their difluoro- and perfluoro-analogues) as a result of tandem direct catalytic alkynylation/dehydrative condensation of o-aminofluoromethylketones (o-FMKs), for the first time catalyzed by NHC-copper(I) complexes on water, is reported. A wide range of terminal alkynes is tolerated under the reaction conditions, including β -lactam-, steroid-, and sugar-derived ones, leading to desired quinolines and naphthydrines with good yields. Further investigations proved that o-FMKs could be

1
2
3 efficiently transformed into a rare class of heterocyclic compounds –
4
5 dibenzo[*b,f*][1,5]diazocines – by a base-catalyzed condensation, also on water. The
6
7 developed method was applied for gram-scale synthesis of a fluorinated analogue of G
8
9 protein-coupled receptor antagonist (GPR91).
10
11

12
13
14 **Keywords:** o-aminofluorophenone, Friedländer reaction, copper, *N*-heterocyclic
15
16 carbene, quinoline, naphthydrine, on water
17
18
19

20 21 Introduction

22
23
24 Fluorine-containing organic molecules, in particular fluorinated heterocycles,
25
26 constitute a privileged structural motif in many areas of modern society, including
27
28 material science, pharmaceutical industry, agrochemicals, fine chemicals,¹ and most of
29
30 all medicinal chemistry.² More than 30% of all compounds present on the worldwide
31
32 agrochemical and pharmaceutical market contain fluorine or fluorinated groups.^{1b,3} The
33
34 incorporation of fluorine into organic compounds strongly affects their biological
35
36 properties, simultaneously allowing for structural elaboration, and hence has become
37
38 crucial for modern drug development. It is well-established that the presence of fluorine
39
40 improves bioavailability and increases lipophilicity and metabolic stability.^{2,3b,4} Iconic
41
42 examples of the beneficial properties bestowed upon fluorinated compounds are
43
44 Ezetimibe⁵ (inhibitor of cholesterol adsorption) and Efavirenz⁶ (non-nucleoside inhibitor
45
46 of reverse transcriptase of HIV), which have been commercialized in the last decade.
47
48
49
50
51

52
53 Among a plethora of fluorinated heterocycles, quinolines and quinolones have
54
55 retained long-standing interest of the synthetic community. Tremendous advances in
56
57
58
59
60

the field of fluorination⁷ have naturally led to the discovery of fluorinated quinolines or quinolones exhibiting remarkable biological properties, and some of them have been introduced to the pharmaceutical market. Representative examples include fluoroquine and mefloquine (antimalarial drugs), brequinar[®] (used in transplantation and for the treatment of psoriasis and rheumatic arthritis), flosequinan (used for the treatment of heart disease) and many derivatives of fluoroquinolones with a broad spectrum of antibacterial activity (Figure 1).⁸

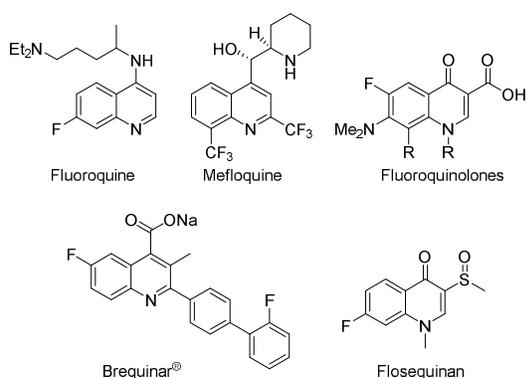
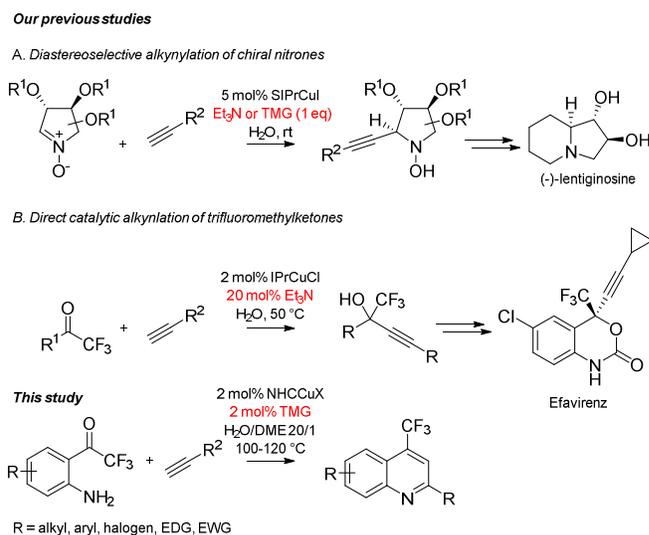


Figure 1. Representative structures of biologically active fluorine-containing quinolines and quinolones.

Due to the importance of quinolines in natural product synthesis and medicinal chemistry, many useful methods for their preparation have been developed to date. Among classical methods leading to this heterocyclic scaffold, including the Skraup reaction,⁹ the Doebner-von Miller reaction,¹⁰ the Gould-Jacobs reaction,¹¹ the Conrad-Limpach reaction,¹² the Combes synthesis,¹³ the Knorr synthesis,¹⁴ and the Niementowski synthesis,¹⁵ the Friedländer reaction¹⁶ constitutes an obvious choice due to its simplicity. Generally, the classical protocol for the latter involves an acid- or a base-catalyzed reaction between a 2-aminocarbonyl compound with an α -methylene

ketone. However, in most cases high temperature is required.^{16b} To avoid harsh conditions and expand the scope of substrates in terms of functional group compatibility, an alternative route based on tandem addition of terminal alkynes to 2-aminobenzaldehydes and spontaneous cyclization has been developed and serves as an excellent alternative. Gold(I) complexes¹⁷ have most often been used as catalysts, in addition to sparse reports devoted to silver(I),¹⁸ iron(III),¹⁹ and copper salts (which necessitate the preactivation of the carbonyl group via an iminium ion).²⁰ In this respect, the application of *o*-aminotrifluoromethylketones (*o*-TFMKs) for the synthesis of fluorinated quinolines is largely unexplored. To date, the application of a stoichiometric amount of $\text{Zn}(\text{OTf})_2$ ²¹ and catalytic pyrophoric Ag/PCy_3 ²² have been reported with an extremely narrow scope of substrates.

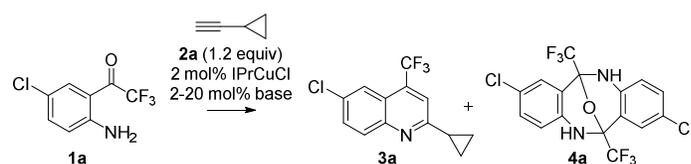


Scheme 1. Direct catalytic alkylation of different electrophiles on water catalyzed by NHCCuX complexes

Herein, we present the first application of *N*-heterocyclic carbene copper(I) complexes (NHCCuX) as efficient catalysts (2 mol%) for the synthesis of fluorine-containing

1
2
3 quinolines (CF_3 , CF_2H , $\text{C}_n\text{F}_{2n+1}$) and naphthydrines via direct catalytic
4 alkylation/dehydrative cyclization sequence on water (Friedländer-type reaction).
5
6 Recently, we have proved the effectiveness of *N*-heterocyclic carbene copper(I)
7
8 complexes in direct catalytic alkylation of chiral nitrones²³ and
9
10 trifluoromethylketones²⁴ (TFMKs) on water (Scheme 1). The key rationale for
11
12 a successful transformation was based on the observation that NHC-copper(I)
13
14 complexes can form stable, monomeric (as confirmed by X-ray crystallography²⁵),
15
16 stable, highly reactive (e.g in triazole formation²⁶) and moisture-insensitive acetylides, in
17
18 contrast to polymeric ones, generated from copper(I) salts themselves.²⁷ Moreover,
19
20 strong σ -donor properties of the NHC ligands enhance the nucleophilicity of the C_{sp}
21
22 carbon atom of the acetylide, facilitating the addition to the electrophile.²⁸ Bearing in
23
24 mind the successful application of NHCCuX complexes for the alkylation of nitrones
25
26 and TFMKs, we anticipated that the same catalytic system, namely $\text{NHCCuX}/\text{water}$,
27
28 should deliver a practical route to fluorinated quinolines as a result of direct catalytic
29
30 alkylation/dehydrative cyclization sequence.
31
32
33
34
35
36
37
38
39
40
41

42 Results and discussion



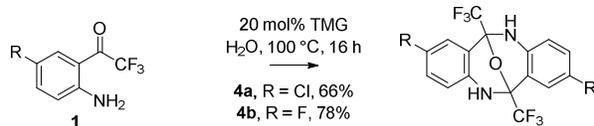
52 Scheme 2. Model reaction.

56 Table 1. Initial optimization of the reaction conditions^a

Entry	Base (mol %)	Solvent	Temperature (°C)	Reaction time (h)	Yield of 3a (%) ^b	Yield of 4a (%) ^b
1	Et ₃ N (20)	DME, DCE, toluene, DMF, DMSO, pyridine	50	48	-	-
2	Et ₃ N (20)	EtOH	50	48	10	n.d.
3	Et ₃ N (20)	H ₂ O/DME (20/1)	50	48	23	22
4	Et ₃ N (20)	H ₂ O/DME (20/1)	100	16	42	43
5	TMG (20)	H ₂ O/DME (20/1)	100	16	53	15
6	TMG (2)	H ₂ O/DME (20/1)	100	16	55	26
7	TMG (2)	H ₂ O/DME (20/1)	100	16	61	21

^a Reactions conditions: ketone **1a** (1.0 mmol), alkyne **2a** (1.2 mmol), NHCCuX (0.02 mmol), tetradecane (10 μ L), DME (0.1 mL), H₂O (2 mL) and appropriate amount of base; conversion was based on GC with tetradecane as internal standard; ^b isolated yield after chromatography; n.d. – not determined, TMG – 1,1,3,3-Tetramethylguanidine.

At the outset of our study, we selected the reaction of cyclopropylacetylene (**2a**) with o-TFMK **1a** (Scheme 2). First, we examined the effect of solvent on the model reaction. When substrates were mixed together in common organic solvents (Table 1, entry 1) in the presence of 2 mol% IPrCuCl and 20 mol% of Et₃N, only unreacted starting materials were detected, even after 48 h at 50 °C. However, when EtOH was used, the expected quinoline **3a** was isolated with poor 10% yield after chromatography (Table 1, entry 2). The best 42% yield was obtained when the reaction was conducted in a mixture of water and DME (v/v 20:1)²⁹ at a higher temperature, which is consistent with our previous findings.²³⁻²⁴ Surprisingly, dibenzo[*b,f*][1,5]diazocine **4a** was also isolated as a by-product with 43% yield. In order to improve the yield of quinoline **3a**, we screened carefully the influence of the type and the amount of base. After some experimentation, it was found that the yield of quinoline **3a** could be improved to 61% when 2 mol% of 1,1,3,3-tetramethylguanidine (TMG) were used. Dibenzocyclopropylidene [1,5]diazocine **4a** was isolated with 21% yield in this case. These results clearly suggested that the formation of dibenzo[*b,f*][1,5]diazocine **4a** is a base-catalyzed process and its formation can be suppressed by lowering the loading of TMG to an equimolar amount relative to the NHCCuX complex used.



9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32

Scheme 3. TMG-catalyzed formation of dibenzo[*b,f*][1,5]diazocine **4**.

To the best of our knowledge, this is the first example of a base-catalyzed synthesis of a dibenzo[*b,f*][1,5]diazocine on water. Similar compounds have been reported as by-products by Wang³⁰ during the synthesis of quinolines via the Friedländer reaction and by Warm³¹ in an attempt towards the large-scale preparation of efavirenz. To confirm the base-catalyzed mechanism, ketones **1** were treated with 20 mol% of TMG on water, furnishing heterocycles **4a** and **4b** cleanly in 66% and 58% yield, respectively (Scheme 3). Inquiries into the scope of formation of dibenzo[*b,f*][1,5]diazocines of type **4** are underway with promising results.

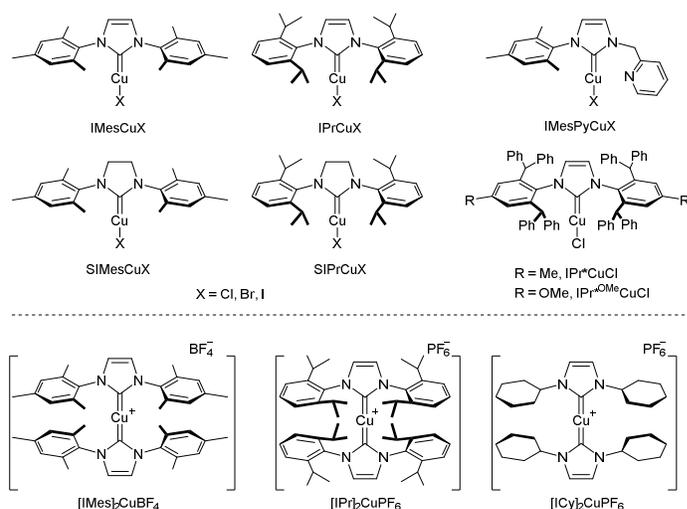


Figure 2. NHC metal complexes used for quinoline **3a** formation.

Table 2. The influence of the NHC carbene ligand and the counterion on the alkylation/cyclization sequence leading to quinoline **3a** and diazocine **4a**.

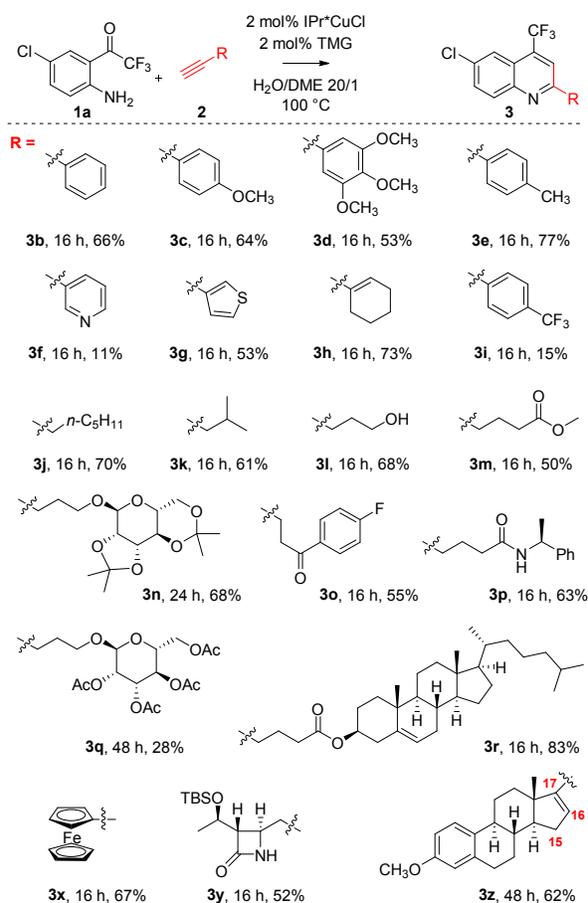
Entry	NHCCuX (2 mol%)	Conversion (%) ^a	Quinoline 3a (%) ^{b,c}	Diazocine 4a (%) ^{b,c}
1	IMesCuCl	94	42 (48)	39 (40)
2	IMesCuBr	91	51 (57)	28 (28)
3	SIMesCuCl	95	35 (45)	40 (40)
4	SIMesCuBr	91	33 (44)	35 (41)
5	IPrCuCl	91	44 (50)	16 (22)
6	IPrCuBr	92	46 (53)	31 (34)
7	IPrCuI	90	37 (45)	32 (33)
8	SIPrCuCl	87	32 (49)	22 (27)
9	SIPrCuBr	95	53 (54)	30 (33)
10	SIPrCuI	94	48 (52)	31 (33)
11	[IMes] ₂ CuBF ₄	91	26 (28)	61 (63)
12	[ICy] ₂ PF ₆	94	44 (47)	35 (36)
13	[IPr] ₂ PF ₆	90	37 (41)	47 (48)
14	PyIMesCuCl	96	49 (53)	33 (34)
15	IPr [*] CuCl	94	71 (78)	7 (8)

16	IPr*CuCl	n.d.	76 ^d	11 ^d
----	----------	------	-----------------	-----------------

^a Reactions conditions: ketone **1a** (1.0 mmol), alkyne **2a** (1.2 mmol), NHCCuX (0.02 mmol), TMG (0.02 mmol), tetradecane (10 μ L), DME (0.1 mL), H₂O (2 mL), 16h, 100 °C; conversion was based on GC with tetradecane as internal standard; ^b isolated yield after chromatography; ^c estimated yield based on GC from calibration curve is given in parentheses (see ESI); ^d reaction conducted on 5.0 mmol scale of **1a**; n.d. – not determined.

This unprecedented product distribution led us to examine the influence of the electronic and steric nature of the NHC carbene-copper(I) complex in more detail (for the structures of NHC complexes used, see Figure 2). Initially, we were pleased to find that copper(I) complexes bearing IMes or SIMes carbene ligands also catalyze the formation of quinoline **3a**. However, no clear conclusion could be given regarding the role of counterion in the copper complex (Table 2, entries 1-4). Quinoline **3a** was isolated with moderate 33–51% yield. In a series of more hindered unsaturated IPr or SIPr complexes, the yield of the reaction could be slightly improved by changing the counterion from iodide or chloride to bromide (Table 2, entry 6 and 9). Unfortunately, in all cases dibenzo[*b,f*][1,5]diazocine **4a** was also formed in significant quantities, up to 32%. Unexpectedly, dibenzo[*b,f*][1,5]diazocine **4a** was isolated in higher yield when the reaction was catalyzed by homoleptic ionic [ICy]₂CuBF₄. Other ionic complexes were less effective. The best results in terms of conversion and yield were obtained when sterically hindered complex IPr*CuCl was applied. Quinoline **3a** was isolated in 71% yield while dibenzo[*b,f*][1,5]diazocine was formed in low 7% yield (Table 2, Entry 15).

Pleasingly, the same reaction conducted on 5.0 mmol scale furnished cleanly product in 76% after chromatography (more than 1 g of **3a** was isolated, Table 2, entry 16).



Scheme 4. The scope of alkynes.

With the optimized conditions (2 mol% of IPr*CuCl and 2 mol% of TMG on water) in hand, we explored the scope of alkynes. It was found that phenylacetylene and its derivatives afforded products **3b-e** with good yields when electron-donating substituents were present, while electron-withdrawing groups such as CF₃ gave poor yields (Scheme 4, **3i**). Further investigations revealed that heterocyclic substituents were tolerated under the reaction conditions. 3-Ethynylthiophene furnished product **3g** cleanly in 53% yield, while 3-ethynylpyridine in only 11% yield. Next, we turned our attention to alkyl-

1
2
3 substituted alkynes. In those cases, quinolines **3j-p** were formed in good yields in the
4 presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
5 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
6 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
7 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
8 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
9 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
10 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
11 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
12 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
13 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
14 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
15 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
16 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
17 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
18 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
19 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
20 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
21 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
22 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
23 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
24 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
25 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
26 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
27 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
28 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
29 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
30 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
31 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
32 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
33 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
34 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
35 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
36 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
37 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
38 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
39 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
40 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
41 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
42 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
43 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
44 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
45 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
46 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
47 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
48 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
49 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
50 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
51 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
52 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
53 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
54 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
55 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
56 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
57 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
58 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate
59 the broad scope of substrates for the synthesis of trifluoromethylquinolines further,
60 the presence of alkene, ketone, ester, amide or free hydroxyl functional groups. To illustrate

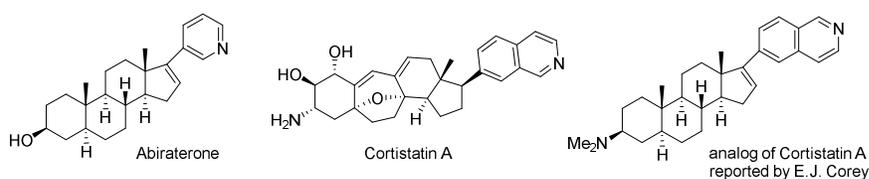
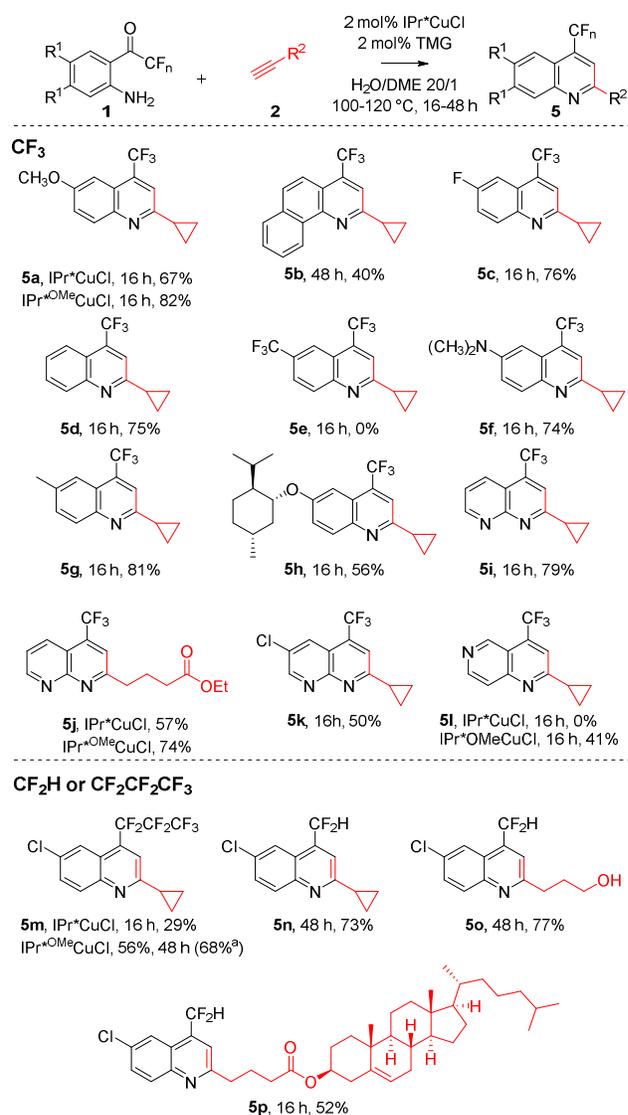


Figure 3. Representative examples of biologically active steroids bearing heterocyclic substituent.

Finally, the developed methodology was applied for the synthesis of structurally complex and pharmaceutically relevant estrone derivative **3z** which is of high importance from the perspective of medicinal applications. The class of compounds consisting of a heterocyclic subunit (usually quinolone, isoquinoline or pyridine) attached to the steroid core in position C17 (for numbering, see Scheme 4, structure **3z**) exhibit a broad spectrum of biological activities including antiangiogenic (e.g. natural products of the cortistatin family³²) or proven anticancer ones (e.g. semisynthetic analogue of cardenolides – abiraterone,³³ Figure 3). Recently, Corey has simplified the structure of naturally occurring cortistatins bearing an unusual steroid scaffold to a

1
2
3 simple androstane derivative and confirmed its prominent antiangiogenic activity.³⁴
4
5 Regarding the synthetic efforts towards cortistatin and its analogues,³⁴⁻³⁵ the key
6
7 transformation involved a palladium-catalyzed (10–50 mol% of Pd) Stille coupling of a
8
9 sterically congested steroid vinyl iodide or enol triflate located in a neopentyl position.
10
11 The application of palladium and tin compounds to the synthesis of biologically active
12
13 molecules can affect biological evaluation due to toxic heavy metal contamination. The
14
15 developed method offers an attractive alternative for the construction of heterocycle-
16
17 steroid hybrids via a palladium-free protocol. The respective estrone derivative **3z** was
18
19 isolated in 62% yield using only 2 mol% of the IPr*CuCl complex.
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

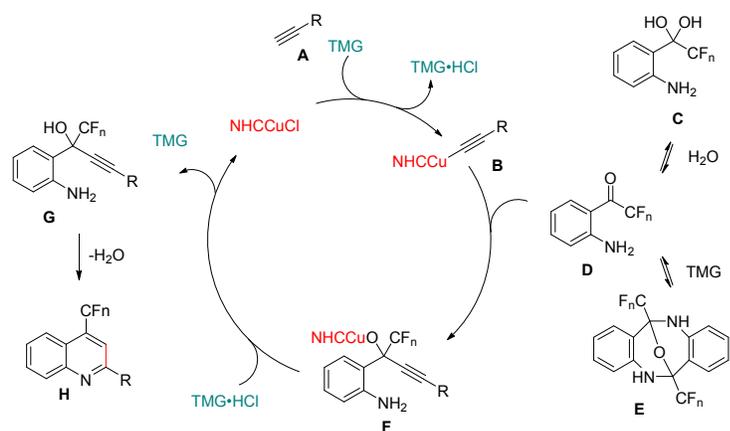


Scheme 5. The scope of fluoroketones. ^a 5 mol% of IPr*OMeCuCl was used.

Next, we explored variously substituted o-TFMKs for the synthesis of quinolines. Ketones bearing strongly and weakly electron-donating groups, such as NMe₂, OMe, Me, F, H as well as a naphthyl moiety furnished the corresponding quinolines **5a,c,d,f-h** and benzo[*h*]quinoline **5b** in moderate to good yields (Scheme 5). Further improvement of the formation of **5a** was accomplished by fine tuning the electronic properties of NHC ligand. Application of IPr*^{OMe}CuCl complex, introduced by Kobayashi,³⁶ in place of

1
2
3 IPr*CuCl resulted in increases in yield, up to 82%. Presumably, this can be attributed to
4
5 the electron-donating properties of the methoxy group attached to the NHC ligand which
6
7 enhance the nucleophilicity of the copper acetylide. Similarly, IPr*^{OMe}CuCl appeared to
8
9 be the complex of choice for the synthesis of fluorinated 1,6- and 1,8-naphthyridines
10
11 **5i-k** and **5l**, accessible by other methods only with difficulty.
12
13

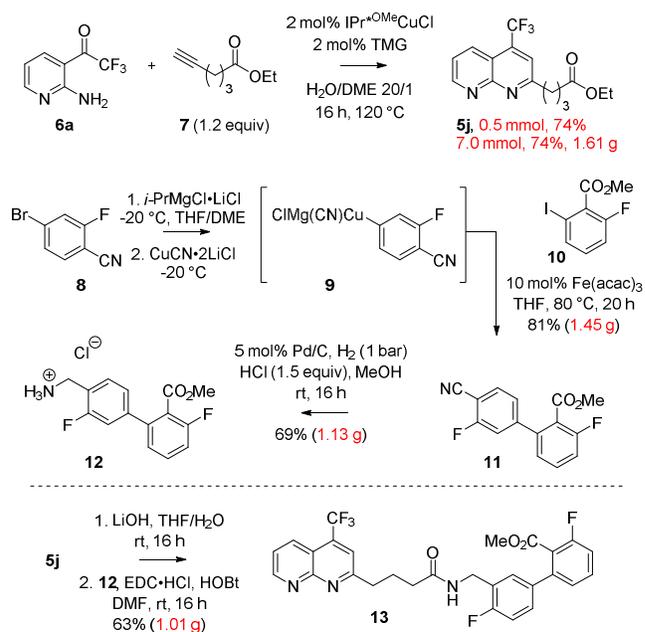
14
15 Further investigations were directed at the synthesis of difluoromethyl-containing
16
17 quinolines. It is well-established that CF₂H can act as a bioisostere of hydroxyl, *N*-
18
19 hydroxamic acid and thiol groups,³⁷ which offers new routes for drug design. CF₂H
20
21 constitutes a good platform for lipophilic interactions as a hydrogen bond donor³⁷⁻³⁸ and
22
23 may enhance membrane permeability, binding affinity and bioavailability. Although
24
25 direct introduction of CF₂H onto arenes has become the subject of intensive research in
26
27 recent years (including Pd,³⁹ Ni,⁴⁰ Cu⁴¹ or Zn⁴² catalyzed transformations), the
28
29 developed methods usually need harsh conditions and are incompatible with many
30
31 functional groups. Furthermore, the complexity of reagent mixtures used or catalytic
32
33 systems makes these approaches far from practical, especially on a large scale.
34
35 Bearing in mind the practical aspects, the developed protocol has revealed a simple
36
37 procedure of difluoromethylquinoline synthesis. Indeed, heating a mixture of
38
39 difluoromethylketone and a terminal alkyne on water in the presence of 2 mol% IPr*CuCl
40
41 and 2 mol% of TMG afforded the respective quinolones **5n-o** in good yields (52–77%),
42
43 also in the case of cholesterol derivative **5p**. The same conditions appeared compatible
44
45 with the synthesis of perfluoroalkyl-quinoline **5m**. However, IPr*^{OMe}CuCl had to be used
46
47 to maintain a reasonable 56% isolated yield.
48
49
50
51
52
53
54
55
56
57
58
59
60



Scheme 6. A plausible catalytic cycle; NHC – N-heterocyclic carbene, TMG – 1,1,3,3-Tetramethylguanidine

Based on our experiments and literature data,²² a plausible catalytic cycle of the NHCCuX-catalyzed Friedländer reaction leading to quinolines was proposed (Scheme 6). The NHCCuX complex interacts with terminal alkyne **A** (via universally accepted π -activation mode) forming the mononuclear copper acetylide **B** and releasing an amine hydrochloride simultaneously (if copper chloride complex is used). The structure of the NHC-copper acetylide was confirmed independently by NMR spectroscopy and X-ray crystallography (by Nolan^{26b} and Jones,²⁵ respectively). Next, the monomeric acetylide **B** undergoes an addition to the fluorinated ketone **D**, affording copper alkoxide **F** which is protonated by the strongest Brønsted acid present in the system – amine hydrochloride. The protonation step regenerates the NHCCuCl complex and the amine needed for the activation of terminal alkyne **A**, closing the catalytic cycle. The spontaneous dehydrative cyclization of the propargyl alcohol **G**, which may be enhanced by the NHCCuX complex via π -activation or by the Brønsted acid,²² affording quinoline **H**. It should be stressed that other side reactions, such as self-condensation or hydration could decrease the yield of quinoline **H**. For those reasons, an equivalent

loading of the base and the NHCCuX complex has to be used to prevent the formation of dibenzo[*b,f*][1,5]diazocine **E** (see, Table 1).



Scheme 7. The application of the Friedländer-type reaction to a gram-scale synthesis of a fluorinated analogue of a G protein-coupled receptor antagonist (GPR91); TMG – 1,1,3,3-tetramethylguanidine, HOBT – 1-hydroxybenzotriazole, EDC·HCl – 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide hydrochloride.

The utility of the developed method is further illustrated through a scalable synthesis of a fluorinated analogue **13** of G protein-coupled receptor antagonist (known as GPR91, Scheme 7). GPR91 belongs to a class of membrane proteins able to bind an extracellular succinate – an intermediate in the Krebs cycle.⁴³ The binding of succinate to GPR91 can affect a wide array of physiological and pathological processes such as inhibition of lipolysis, regulation of blood pressure, cardiac hypertrophy, angiogenesis in retinal tissues, etc.⁴⁴ Among non-peptide small organic molecules which serve as

1
2
3 ligands for GPR91, naphthydrines coupled with biphenyl-derived benzylamine, originally
4 synthesized and evaluated by Merck as inhibitors of Bradykinin B1 receptor⁴⁵ (BK₁R),
5 have witnessed a growing interest.⁴⁶ However, their fluorinated analogues have not
6 been investigated to date. Herein, we present a scalable convergent route to this class
7 of compounds. First, we synthesized fluorinated naphthydrine **5j** applying the developed
8 method on a 0.5 mmol scale which afforded the product in 74% yield. Pleasingly, we
9 found that a 14-fold increase in scale furnished the product cleanly with equally
10 excellent yield, further underlining the practical aspect of the developed method.
11
12

13 The intermediate benzylamine derivative **12** was synthesized by a two-step sequence.
14 The key biphenyl **11** was synthesized via Fe-catalyzed aryl-aryl cross-coupling under
15 conditions developed by Knochel.⁴⁷ Arylcopper reagent **9**, generated from the
16 respective nitrile **8**, was coupled with iodoester **10** to afford biaryl **11** in 81% yield on
17 large scale (1.45 g). It should be noted that homocoupling product was observed when
18 the arylcopper species was generated from iodoester **10**. Subsequent chemoselective
19 reduction of nitrile group to amine appeared to be challenging. BH₃ and common metal
20 hydrides (NaBH₄ in combination with CoCl₂, CuCl) appeared to be unsuccessful,
21 whereas the reduction in the presence of Raney nickel furnished secondary amine⁴⁸ (for
22 details, see ESI). Gratifyingly, Pd-catalyzed reduction in the presence of HCl in MeOH
23 cleanly afforded the respective amine hydrochloride **12** without chromatography on
24 gram scale. The obtained amine **12** was coupled with ester **5j** (after hydrolysis of the
25 latter), mediated by the mixture of EDC•HCl/HOBt to give amide **13** with 63% yield after
26 two steps. The evaluation of biological activity of naphthydrine **13** and its analogues are
27 in progress.
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

Conclusion

In summary, we developed an efficient synthesis of trifluoromethylquinolines (and their difluoromethyl- and perfluoroalkyl- analogues) and naphthydrines via tandem direct catalytic alkynylation of *o*-aminofluoromethylketones/condensation sequence, for the first time on water in the presence of a catalytic amount of NHC-copper(I) complexes. The established method allows to obtain a series of fluorinated quinolines with good to moderate yields. During the optimization of the reaction conditions, it was found that *o*-aminotrifluoromethylketones undergo base-catalyzed self-condensation leading to a rare example of dibenzo[*b,f*][1,5]diazocine synthesis. Further experiments confirmed that the formation of dibenzo[*b,f*][1,5]diazocines is catalyzed by a weak base on water. The studies on the scope of the formation of dibenzo[*b,f*][1,5]diazocines are underway with promising results.

Experimental section

General remarks

NMR spectra were recorded in CDCl₃ or DMSO-*d*₆ solutions (unless indicated otherwise); chemical shifts are quoted on the δ scale, ppm, with the solvent signal as the internal standard (CHCl₃, ¹H NMR 7.26 ppm; CDCl₃, ¹³C NMR 77.00 ppm, DMSO-*d*₆ 2.50 ppm, ¹³C NMR 39.40 ppm). High resolution mass spectra (HRMS) were taken using EI technique or electrospray ionization (ESI). Column chromatography was performed on Merck silica gel 60, 230-400 mesh. TLC was performed on aluminum sheets, Merck 60F 254. Anhydrous solvents were obtained by distillation over CaCl₂

(CH₂Cl₂) or Na/benzophenone (THF, hexane, MTBE). Air-sensitive reactions were performed in flame-dried glassware under atmosphere of argon. Organic extracts were dried and solvents were evaporated in a rotary evaporator. Reagents were used as they were purchased unless otherwise indicated (which is specified at beginning of each section). The name of compounds were generated using ACD Lab Name 12.0 software.

Synthesis of N-heterocyclic carbene copper(I) complexes

[IMes]₂CuPF₆, SIPrAgCl and IPrAuCl were commercially available from Aldrich and used as received. SIMesCuCl,⁴⁹ SIMesCuBr,^{26a} IMesCuCl,⁴⁹ IMesCuBr,⁵⁰ SIPrCuCl,⁴⁹ SIPrCuBr,⁵¹ SIPrCuI,⁵¹ SIPrCuOTf,⁵² SIPrOAc,⁵³ IPrCuCl,⁴⁹ IPrCuBr,⁵¹ IPrCuI,⁵¹ IPr^{*}CuCl,⁴⁹ [ICy]₂CuPF₆,^{26b} and [IPr]₂CuPF₆⁵⁴ were prepared followed by literature procedure.

{1,3-bis[2,6-bis(Diphenylmethyl)-4-methoxyphenyl]-1,3-dihydro-2H-imidazol-2-

ylidene}(chloro)copper IPr^{*OMe}CuCl was prepared applying literature procedure developed by Cazin *et. al.*⁴⁹ In a glove box, a 20-mL screw-cap vial was charged with K₂CO₃ (281.9 mg, 2.04 mmol, 2.0 equiv), IPr^{*OMe}Cl (1.0 g, 1.02 mmol), CuCl (100.8 mg, 1.02 mmol, 1.0 equiv) and anhydrous acetone (10 mL). Next the vial was capped, removed from the glove box and the resulting suspension was vigorously stirred for 20h at 60 °C (temp. of oil bath). Then solvent was evaporated, DCM (20 mL) was added and the resulting mixture was filtered through a pad of Celite (washing with DCM). The crude copper(I) complex was dissolved in minimum volume of DCM (6 mL) and crashed with *n*-pentane (28 mL). The obtained solid was filtered, washed with *n*-pentane (3 x 10 mL), and dried in vacuo to give IPr^{*OMe}CuCl as a white solid (933.3 mg, 88%). ¹H NMR (400 MHz, CD₂Cl₂) δ 7.28–7.18 (m, 24H), 7.13–7.08 (m, 8H), 6.98–6.91 (m, 8H), 6.62

(s, 4H), 5.86 (s, 2H), 5.25 (s, 4H), 3.62 (s, 6H). ¹H NMR data are in agreement with those reported.³⁶

Synthesis of alkynes

Cyclopropylethyne (**2a**), ethynylbenzene (**2b**), 1-ethynyl-4-methoxybenzene (**2c**), 5-ethynyl-1,2,3-trimethoxybenzene (**2d**), 1-ethynyl-4-methylbenzene (**2e**), 3-ethynylpyridine (**2f**), 3-ethynylthiophene (**2g**), 1-ethynylcyclohexene (**2h**), 1-ethynyl-4-trifluoromethylbenzene (**2i**), octy-1-yne (**2j**), 4-methylpent-1-yne (**2k**), pent-4-yne-1-ol (**2l**), methyl hex-5-ynoate (**2m**) and ethynylferrocene (**2x**) are commercially available and used as received without further purification. 1-(4-Fluorophenyl)pent-4-yn-1-one (**2o**),²⁴ ethyl hex-5-ynoate (**7**),⁵⁵ (3 β)-cholest-4-en-3-yl hex-5-ynoate (**2r**),⁵⁶ 17-ethynyl-3-methoxyestra-1(10),2,4,16-tetraene (**2z**)⁵⁷ and (3*S*,4*R*)-3-((1*R*)-[[*tert*-butyl(dimethyl)silyl]oxo]ethyl-4-prop-2-yn-1-yl)azetidine-2-one (**2y**)⁵⁸ were prepared followed by literature procedure.

Pent-4-yn-1-yl 2,3,4,6-tetra-O-acetyl- α -D-mannopyranoside (2q) To a solution of 1,2,3,4,6-penta-O-acetyl-D-mannopyranose (4.37 g, 11.2 mmol), MS 3Å (4 g) in anhydrous DCM (80 mL) and 4-pentyn-1-ol (**2l**) (2.08 mL, 22.4 mmol, 2.0 equiv), cooled to -10 °C, BF₃·Et₂O (11.2 mL, 89.6 mmol, 8.0 equiv) was added dropwise. The resulting mixture was slowly allowed to reach rt and stirred for 16 h. The reaction mixture was then quenched by addition of solid NaHCO₃ (10 g) and stirred at rt. After 30 min. the reaction mixture was washed with sat. aqueous NaHCO₃ (3 x 50 mL), dried over MgSO₄, and solvent was evaporated. The residue was chromatographed on silica FCC (15-30% EtOAc/hexanes) to give a light yellow oil (1.55 g, 33%, CAS: 1327252-78-7, no spectral data available for comparison were reported). $[\alpha]_D^{23} = 46.4$ (c = 1.1, CHCl₃); IR

(film) 3283, 2958, 2939, 2117, 1751 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 5.32 (dd, J = 10.0, 3.2 Hz, 1H), 5.29 (t, J = 9.6 Hz, 1H), 5.22 (dd, J = 3.2, 1.6 Hz, 1H), 4.80 (d, J = 1.6 Hz, 1H), 4.27 (dd, J = 12.2, 5.2 Hz, 1H), 4.10 (dd, J = 12.3, 2.4 Hz, 1H), 4.00 (ddd, J = 9.6, 5.2, 2.4 Hz, 1H), 3.83 (ddd, J = 9.6, 7.2, 5.6 Hz, 1H), 3.55 (ddd, J = 9.6, 5.6, 5.6 Hz, 1H), 2.35–2.27 (m, 2H), 2.14 (s, 3H), 2.09 (s, 3H), 2.03 (s, 3H), 1.98 (s, 3H), 1.95 (t, J = 2.6 Hz, 1H), 1.89–1.73 (m, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 170.6, 170.0, 169.8, 169.7, 97.6, 83.2, 69.6, 69.1, 69.0, 68.5, 66.5, 66.2, 62.4, 27.9, 20.8, 20.7, 20.6, 20.6, 15.2; HRMS (ESI-TOF) m/z : $[\text{M}+\text{Na}]^+$ Calcd for $\text{C}_{19}\text{H}_{26}\text{O}_{10}\text{Na}$ 437.1424; Found: 437.1416.

Pent-4-yn-1-yl α -D-mannopyranoside (14) To a stirred solution of pent-4-yn-1-yl 2,3,4,6-tetra-O-acetyl- α -D-mannopyranoside (**2q**) (875.0 mg, 2.24 mmol) in dry MeOH (20 mL), KOMe (15.7 mg, 0.224 mmol, 0.1 equiv) was added at rt and stirred for 3h. Then solvent was evaporated and residue was passed through pad of silica (10% MeOH/DCM) to give a colourless oil (474.4 mg, 86%). The obtained pent-4-yn-1-yl α -D-mannopyranoside (**14**) was used without further purification in the next step. $[\alpha]_D^{23} = -70.7$ ($c = 1.9$, MeOH); ^{13}C NMR (100 MHz, CD_3OD) δ 101.6, 84.3, 74.5, 72.7, 72.2, 69.9, 68.5, 66.9, 62.8, 29.6, 15.9; HRMS (ESI-TOF) m/z : $[\text{M}+\text{Na}]^+$ Calcd for $\text{C}_{11}\text{H}_{18}\text{O}_6\text{Na}$ 269.1001; Found: 269.0991.

(3aS,4S,5aR,9aR,9bS)-2,2,8,8-Tetramethyl-4-(pent-4-yn-1-

oxy)hexahydro[1,3]dioxolo[4,5]pyrano [3,2- d][1,3]dioxine (2n) To a stirred rt solution of the tetraol **14** (0.397 g, 1.71 mmol) in dry DMF (13 mL) under atmosphere of argon, $\text{Me}_2\text{C}(\text{OMe})_2$ (1.32 mL, 10.9 mmol, 10.0 equiv) and p -TsOH (20.9 mg, 0.11 mmol, 0.1 equiv) were added at rt. The reaction mixture was stirred for 3 h before

1
2
3 acetone (2.7 mL) was introduced. After 24 h the reaction was diluted with Et₂O (20 mL)
4
5 and washed with sat. NaHCO₃ solution (2 x 20 mL). The organic layer was separated,
6
7 washed with H₂O (3 x 20 mL), brine (2 x 30 mL), dried over Na₂SO₄, filtered and
8
9 concentrated in *vacuo*. The residue was purified by chromatography on silica (5%
10
11 EtOAc/hexanes) to give diacetal **2n** as a colourless oil (441.0 mg, 79%). $\alpha = -70.7$ (c =
12
13 1.9, MeOH); IR (film) 3278, 2990, 2939, 2915, 2876, 2117 cm⁻¹; ¹H NMR (400 MHz,
14
15 CDCl₃) δ 5.00 (s, 1H), 4.18–4.09 (m, 2H), 3.87 (dd, *J* = 10.8, 5.6 Hz, 1H), 3.84–3.69 (m,
16
17 3H), 3.63–3.44 (m, 2H), 2.30 (dt, *J* = 7.2, 2.8 Hz, 2H), 1.95 (t, *J* = 2.8 Hz, 1H), 1.86–
18
19 1.74 (m, 2H), 1.54 (s, 3H), 1.51 (s, 3H), 1.42 (s, 3H), 1.34 (s, 3H); ¹³C NMR (100 MHz,
20
21 CDCl₃) δ 109.4, 99.7, 97.8, 83.3, 76.1, 74.9, 72.7, 68.8, 65.9, 62.1, 61.4, 29.0, 28.2,
22
23 28.1, 26.1, 18.8, 15.2; HRMS (ESI-TOF) *m/z*: [M+Na]⁺ Calcd for C₁₇H₂₆O₆Na 349.1627;
24
25 Found: 349.1629.
26
27
28
29
30
31

32 ***N*-[(1*S*)-1-Phenylethyl]hex-5-ynamide (2p)** To a solution of 5-hexynoic acid (1 g, 8.91
33
34 mmol) in DCM (20 mL), cooled to 0 °C, EDC (2.05 g, 10.7 mmol, 1.2 equiv) and (*S*-
35
36 methylbenzylamine (1 g, 8.91 mmol, 1.2 equiv) were added, and stirred for 3h at rt.
37
38 Then reaction mixture was diluted with water (40 mL) and organic phase was
39
40 separated, washed with 5% HCl_{aq} (2 x 20 mL), dried over MgSO₄ and evaporated. The
41
42 residue was passed through a pad of silica (eluted with 40% EtOAc/hexanes) to give a
43
44 waxy solid (1.83 g, 95%). $[\alpha]_D^{23} = -92.7$ (c = 0.94, CHCl₃); IR (film) 3292, 3063, 2972,
45
46 2934, 2116, 1614, 1545 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.37–7.23 (m, 5H), 5.83 (br
47
48 s, *J* = 6.3 Hz, 1H), 5.19–5.06 (m, 1H), 2.37–2.19 (m, 4H), 1.96 (t, *J* = 2.6 Hz, 1H), 1.91–
49
50 1.80 (m, 2H), 1.48 (d, *J* = 6.9 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 171.2, 143.2,
51
52
53
54
55
56
57
58
59
60

1
2
3 128.6, 127.3, 126.1, 83.5, 69.2, 48.7, 35.1, 24.1, 21.7, 17.8; HRMS (ESI-TOF) m/z :
4
5 [M+Na]⁺ Calcd for C₁₄H₁₇NONa 238.1208; Found: 238.1204.
6
7

8 **Synthesis of ortho-aminotrifluoromethylketones (o-TFMK's) and their derivatives**
9
10 **containing difluoromethyl group and perfluoroalkyl chain**
11

12
13 1-(2-Amino-5-chlorophenyl)-2,2,2-trifluoroethanone (**1a**),^{6a} 2,2-difluoro-1-(morpholin-4-
14 yl)ethanone⁵⁹ were prepared according to literature procedure. 1-(2-Amino-5-
15 chloropyridin-3-yl)-2,2-difluoroethanone is commercially available and was used as
16 received.
17
18
19
20
21

22
23 **General procedure** To a solution of amide (x mmol) and TMEDA (1.0 equiv), cooled to
24 -20 °C (or lower temp.) in MTBE (or THF), a solution of *n*-BuLi (2.2 equiv) was added
25 dropwise by means of syringe pump within 30 min. - 2h while the temperature was kept
26 below -10 °C (or lower as indicated). The mixture was aged at -5 – 5 °C for 2-4 h,
27 cooled below -30 °C (or -65 °) and CF₃CO₂Et (1.4 equiv) or other source of fluorinated
28 group was added. Then reaction mixture was stirred for 0.5 - 1h at rt, quenched with 5%
29 HCl (or 10% citric acid). Then solvents were evaporated. In some cases product was
30 separated by chromatography or crystallization from unreacted substrate. Then residue
31 was treated with 36% aqueous HCl or its solution in dioxane (4M in dioxane), and
32 stirred for 2-16h at 70-90 °C (temp. of oil bath). Then reaction mixture was neutralized
33 with sat. solution of K₂CO₃ (or NaHCO₃) and extracted with EtOAc. The combined
34 organic extracts were washed with brine, dried over MgSO₄ and evaporated. The
35 residue was chromatographed on silica and further purified by crystallization (usually
36 from *n*-heptane) to give a pure *ortho*-aminofluorophenones and its derivatives.
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 **1-(2-Amino-5-methoxyphenyl)-2,2,2-trifluoroethanone (1b)** To a solution of amide *N*-
4 (4-methoxyphenyl)-2,2-dimethylpropanamide (11.41 g, 55.1 mmol) in Et₂O (60 mL) and
5
6 TMEDA (8.5 mL, 55.1 mmol, 1.0 equiv), cooled to -20 °C, a solution of *n*-BuLi (48.5 mL,
7
8 121.2 mmol, 2.2 equiv, *n*-BuLi 2.5 M in hexanes) was added dropwise while the
9
10 temperature was kept below 0 °C. The mixture was aged at 0-5 °C for 4 h and cooled
11
12 below -20 °C, and CF₃CO₂Et (8.0 mL, 67.2 mmol, 1.4 equiv) was added rapidly (internal
13
14 temp. reached 10 °C). Then reaction mixture was stirred for 1h at rt, quenched with 5%
15
16 HCl (40 mL). The aqueous phase was separated and extracted with MTBE (3 x 50 mL).
17
18 The residue was treated with 36% HCl (20 mL), and stirred for 16h at 90 °C (temp. of oil
19
20 bath). Then reaction mixture was cooled to 0 °C and washed with EtOAc to give a white
21
22 solid. The obtained solid was suspended in MTBE (40 mL) and treated with aq sat.
23
24 solution of NaOAc (80 mL). After 30 min. of vigorous stirring, organic phase was
25
26 separated, dried over MgSO₄, and evaporated. The residue was chromatographed on
27
28 silica (5-15% EtOAc/hexanes) to give an orange solid (1.64 g, 14%). ¹H NMR (400 MHz,
29
30 CDCl₃) δ 7.18–7.13 (m, 1H), 7.09 (dd, *J* = 9.1, 2.8 Hz, 1H), 6.69 (d, *J* = 9.1 Hz, 1H),
31
32 6.23 (br s, 2H), 3.78 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 180.1 (q, *J*_{CF} = 33.0 Hz),
33
34 150.0, 148.7, 127.4, 118.9, 117.1 (q, *J*_{CF} = 289.8 Hz), 111.4 (q, *J*_{CF} = 4.3 Hz), 110.4,
35
36 55.8; ¹⁹F NMR (376 MHz, CDCl₃) δ -69.9. ¹H NMR data are in agreement with those
37
38 reported.⁶⁰ ¹³C NMR spectra is described by authors without C-F coupling constants
39
40 and the accurate description and the copy of spectra are included in this experimental
41
42 section.
43
44
45
46
47
48
49
50
51
52

53 **1-(1-Aminonaphthalene-2-yl)-2,2,2-trifluoroethanone (1c)** To a solution of 2,2-dimethyl-
54
55 *N*-(naphthalen-1-yl)propanamide (10.0 g, 44.0 mmol) in MTBE (80 mL) and TMEDA (6.8
56
57
58
59
60

1
2
3 mL, 44.0 mmol, 1.0 equiv), cooled to -20 °C, a solution of *n*-BuLi (38.7 mL, 96.8 mmol,
4
5 2.2 equiv, BuLi 2.5 M in hexane) was added dropwise by means of syringe pump within
6
7 2h while the temperature was kept below -10 °C. The mixture was aged at -5 - 0 °C for
8
9 2 h and cooled below -30 °C, and CF₃CO₂Et (7.35 mL, 61.8 mmol, 1.4 equiv) was
10
11 added rapidly (internal temp. reached 20 °C). Then reaction mixture was stirred for 1h at
12
13 rt, quenched with 5% HCl (40 mL) and solvents were evaporated. The residue was
14
15 treated with 36% HCl (20 mL), and stirred for 5h at 70 °C (temp. of oil bath). Then
16
17 reaction mixture was neutralized with sat. solution of K₂CO₃ and extracted with EtOAc
18
19 (4 x 30 mL). The combined organic extracts were washed with brine (1 x 30 mL), dried
20
21 over MgSO₄, and evaporated. The residue was chromatographed on silica (toluene)
22
23 and further purified by crystallization (*n*-heptane) to give a bright-yellow solid (3.03 g,
24
25 29%). M.p. 137.5–138.6 °C (*n*-heptane); IR (KBr) 3461, 3313, 1647, 1614 cm⁻¹; ¹H NMR
26
27 (400 MHz, CDCl₃) δ 8.01-7.70 (br s, 2H) overlapping 7.94-7.90 (pseudo dd, 1H) and
28
29 7.75–7.71 (m, 1H), 7.69–7.71 (m, 2H), 7.55–7.48 (m, 1H), 7.05 (d, *J* = 9.1 Hz, 1H); ¹³C
30
31 NMR (100 MHz, CDCl₃) δ 179.9 (*J*_{CF} = 32.9 Hz), 153.3, 136.8, 130.6, 128.7, 126.1,
32
33 125.1 (*J*_{CF} = 4.3 Hz), 122.6, 121.9, 117.5 (*J*_{CF} = 289.4 Hz), 116.4, 105.3; ¹⁹F NMR (376
34
35 MHz, CDCl₃) δ -69.2; HRMS (ESI-TOF) *m/z*: [M+H]⁺ Calcd for C₁₂H₉F₃NO 240.0636;
36
37 Found: 240.0632.
38
39
40
41
42
43
44
45

46 **1-(2-Amino-5-fluorophenyl)-2,2,2-trifluoroethanone (1d)** To a solution of *N*-(4-
47
48 fluorophenyl)-2,2-dimethylpropanamide (9.37 g, 48.0 mmol) in MTBE (150 mL) and
49
50 TMEDA (7.4 mL, 48.0 mmol, 1.0 equiv), cooled to -20 °C, a solution of *n*-BuLi (45.8 mL,
51
52 109.0 mmol, 2.27 equiv, BuLi 2.38 M in hexanes) was added dropwise while the
53
54 temperature was kept below 0 °C. The mixture was aged at 0-5 °C for 2 h and cooled
55
56
57
58
59
60

1
2
3 below -20 °C, and CF₃CO₂Et (8.0 mL, 67.2 mmol, 1.4 equiv) was added rapidly (internal
4 temp. reached 10 °C). Then reaction mixture was stirred for 1h at rt, quenched with 5%
5 HCl (40 mL) and solvents were evaporated. The residue was treated with 36% HCl (20
6 mL), and stirred for 3h at 70 °C (temp. of oil bath). Then reaction mixture was cooled to
7 0 °C and washed with EtOAc to give a white solid. The obtained solid was suspended in
8 MTBE (40 mL) and treated with aq sat. solution of NaOAc (80 mL). After 30 min. of
9 vigorous stirring, organic phase was separated, dried over MgSO₄, and evaporated. The
10 residue was crystallized from *n*-heptane to give a bright-yellow solid (7.65 g, 77%). ¹H
11 NMR (400 MHz, CDCl₃) δ 7.45–7.39 (m, 1H), 7.21–7.14 (m, 1H), 6.70 (dd, *J* = 9.2, 4.5
12 Hz, 1H), 6.36 (br s, 2H, NH); ¹³C NMR (100 MHz, CDCl₃) δ 180.7 (dq, *J*_{CF} = 34.0, 3.3
13 Hz, C=O), 153.2 (d, *J*_{CF} = 238.4 Hz), 150.0, 125.7 (d, *J*_{CF} = 24.3 Hz), 118.9 (d, *J*_{CF} = 7.1
14 Hz), 116.8 (q, *J*_{CF} = 289.5 Hz), 115.3 (dq, *J*_{CF} = 23.4, 4.3 Hz), 110.0 (d, *J*_{CF} = 6.6 Hz);
15 ¹⁹F NMR (376 MHz, CDCl₃) δ -70.1, -127.1. CAS registry number - 214288-07-0, no
16 spectroscopic data are available for comparison.

17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37 **1-(2-Aminophenyl)-2,2,2-trifluoroethanone (1e)** To a solution of 2,2-dimethyl-*N*-
38 phenylpropanamide (7.80 g, 44.0 mmol) in MTBE (100 mL) and TMEDA (6.8 mL, 44.0
39 mmol, 1.0 equiv), cooled to -20 °C, a solution of *n*-BuLi (38.7 mL, 96.8 mmol, 2.2 equiv,
40 *n*-BuLi 2.5 M in hexane) was added dropwise by means of syringe pump within 2h while
41 the temperature was kept below -10 °C. The mixture was aged at 25 °C for 3 h (yellow
42 solid has formed), cooled below -25 °C, and CF₃CO₂Et (7.35 mL, 61.8 mmol, 1.4 equiv)
43 was added rapidly (internal temp. reached 0 °C). Then reaction mixture was stirred for
44 16h at rt and quenched with 5% HCl (100 mL). The organic phase was separated and
45 solvents were evaporated. The residue was chromatographed on silica (5-10%
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 EtOAc/hexanes) to give a yellow oil (5.1 g) and unreacted substrate (1.6 g). The crude
4
5 TFMK was treated with 36% HCl (20 mL) at rt (white solid was formed immediately) and
6
7 stirred for 3h at 90 °C (temp. of oil bath, yellow solution has been formed at 70 °C).
8
9 Then reaction mixture was neutralized with sat. solution of Na₂CO₃ and extracted with
10
11 EtOAc (4 x 30 mL). The combined organic extracts were washed with brine (1 x 30 mL),
12
13 dried over MgSO₄, and evaporated. The residue was purified by crystallization (*n*-
14
15 heptane, 8 mL, -40 °C, 2h) to give a bright yellow solid (2.66 g, 32%). ¹H NMR (400
16
17 MHz, CDCl₃) δ 7.82–7.73 (m, 1H), 7.43–7.36 (m, 1H), 6.77–6.68 (m, 2H), 6.45 (br s, 2H,
18
19 NH); ¹³C NMR (100 MHz, CDCl₃) δ 180.8 (q, *J*_{CF} = 33.1 Hz), 151.1, 136.6, 131.3 (q, *J*_{CF}
20
21 = 4.1 Hz), 117.4, 117.1 (q, *J*_{CF} = 289.7 Hz), 116.4, 111.2; ¹⁹F NMR (376 MHz, CDCl₃) δ
22
23 -69.6. ¹H NMR data are in agreement with those reported.⁶⁰ ¹³C NMR spectra is
24
25 described by authors without C-F coupling constants and the accurate description and
26
27 the copy of spectra are included in this experimental section.
28
29
30
31
32
33

34
35 **1-[2-Amino-5-(trifluoromethyl)phenyl]-2,2,2-trifluoroethanone (1f)** To a solution of
36
37 2,2-dimethyl-*N*-[4-(trifluoromethyl)phenyl]propanamide (10.8 g, 44.0 mmol) in MTBE
38
39 (130 mL) and TMEDA (6.8 mL, 44.0 mmol, 1.0 equiv), cooled to -20 °C, a solution of *n*-
40
41 BuLi (38.7 mL, 96.8 mmol, 2.2 equiv, *n*-BuLi 2.5 M in hexane) was added dropwise by
42
43 means of syringe pump within 1h while the temperature was kept below -20 °C. The
44
45 mixture was aged at 0 - 5 °C for 2 h (a brown-black solid has been formed), cooled
46
47 below -25 °C, and CF₃CO₂Et (7.35 mL, 61.8 mmol, 1.4 equiv) was added rapidly
48
49 (internal temp. reached 0 °C). Then reaction mixture was stirred for 1h at rt, quenched
50
51 with 5% HCl (100 mL) and organic phase was separated and solvents were evaporated.
52
53
54
55
56 The yellow-green residue was treated with 36% HCl (100 mL), and stirred for 16h at 90
57
58
59
60

1
2
3 °C (temp. of oil bath). Then reaction mixture was neutralized with solid Na₂CO₃ and
4
5 extracted with EtOAc (4 x 30 mL). The combined organic extracts were washed with
6
7 brine (1 x 30 mL), dried over MgSO₄, and evaporated. The residue was
8
9 chromatographed on silica (10% EtOAc/hexanes) to give a bright yellow solid (4.44 g,
10
11 pure fraction). Additional 1.89 g of product was obtained by crystallization (*n*-heptane) of
12
13 mixed fraction after chromatography. (4.44 g + 1.89 g, 56%). ¹H NMR (400 MHz, CDCl₃)
14
15 δ 8.05 (s, 1H), 7.58 (dd *J* = 8.9, 1.8 Hz, 1H), 6.83 (d, *J* = 2.2 Hz, 1H), 6.75 (br s, NH,
16
17 2H); ¹³C NMR (100 MHz, CDCl₃) δ 180.7 (q, *J*_{CF} = 34.0 Hz), 154.7, 132.6 (q, *J*_{CF} = 3.0
18
19 Hz), 129.3 – 129.0 (m), 123.8 (q, *J*_{CF} = 269.3 Hz), 118.6 (q, *J*_{CF} = 33.4 Hz), 118.1, 116.7
20
21 (q, *J*_{CF} = 289.4 Hz), 109.9; ¹⁹F NMR (376 MHz, CDCl₃) δ -62.3, -69.8. ¹H NMR data are
22
23 in agreement with those reported.⁶⁰ ¹³C NMR spectra is described by authors without C-
24
25 F coupling constants and the accurate description and the copy of spectra are included
26
27 in this experimental section.
28
29
30
31
32
33

34 **1-[2-Amino-5-(dimethylamino)phenyl]-2,2,2-trifluoroethanone (1g)** To a solution of
35
36 *N*-[4-(dimethylamino)phenyl]-2,2-dimethylpropanamide (4.99 g, 22.66 mmol) and
37
38 TMEDA (45.34 mmol, 6.95 mL, 2.0 eq) in THF, cooled to -60 °C, *n*-BuLi (19.9 mL, 45.34
39
40 mmol, 2.2 equiv, 2.28 M in hexane) was added dropwise by means of syringe pump
41
42 within 30 min. (temp. has increased to -50 °C), and the reaction was stirred for 4h at 0 –
43
44 -5 °C degree (internal temperature; reaction mixture was stirred by means of
45
46 mechanical stirrer due to lithium dianion precipitation). Then reaction mixture was
47
48 cooled to -65 °C and CF₃CO₂Et (3.77 mL, 31.73 mmol, 1.4 equiv) was added within 5
49
50 min. (the internal temp. has risen to -35 °C). Then cooling bath was removed, and the
51
52 reaction mixture was allowed to warm up to 0 °C and stirred for additional 1h at this
53
54
55
56
57
58
59
60

1
2
3 temp. and quenched with 10% citric acid (100 mL). Then mixture was diluted with
4 EtOAc (100 mL), and aqueous phase was extracted with EtOAc (2 x 50 mL). The
5
6 combined organic phases were washed with brine (2 x 100 mL), dried over MgSO₄,
7
8 evaporated and dried in vacuo to give an orange foam. The residue was
9
10 chromatographed on silica (20% EtOAc/hexanes) to give a brown oil contains some
11
12 impurities. Protected aminofluoroketone was used in the next step without further
13
14 purification.
15
16
17
18

19
20 The crude aminofluoroketone derivative was treated with HCl in dioxane (20 mL, 4M in
21
22 dioxane) and water (2 mL) and resulting mixture was heated at 90 °C for 16h. Then
23
24 reaction mixture was neutralized with sat. solution of Na₂CO₃ (100 mL), and aqueous
25
26 phase was extracted with EtOAc (3 x 50 mL). The combined organic phases were
27
28 washed with brine (2 x 100 mL), dried over MgSO₄, and evaporated. The residue
29
30 purified by chromatography (25% EtOAc/hexanes) to give a brown-black oil which
31
32 solidified upon solid in the fridge (720.2 mg, 14%). M.p. 72.2–73.0 °C (*n*-heptane); ¹H
33
34 NMR (400 MHz, CDCl₃) δ 7.13 (dd, *J* = 8.9, 2.8 Hz, 1H), 7.08–7.03 (m, 1H), 6.69 (d, *J* =
35
36 9.1 Hz, 1H), 6.08 (br s, NH, 2H), 2.84 (s, CH₃, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 180.4
37
38 (q, *J*_{CF} = 32.7 Hz), 146.9, 142.3, 127.2, 118.5, 117.2 (q, *J*_{CF} = 289.9 Hz), 113.3 (q, *J*_{CF} =
39
40 4.3 Hz), 111.3, 41.9; ¹⁹F NMR (376 MHz, CDCl₃) δ -69.8. The structure of ketone **1g**
41
42 has reported by Patel,⁶¹ however no spectroscopic data are available for comparison
43
44 (no CAS registry number). Copy of spectra are included in this experimental section.
45
46
47
48
49

50
51 **1-(2-Amino-5-methylphenyl)-2,2,2-trifluoroethanone (1h)** To a solution of 2,2-
52
53 dimethyl-*N*-(4-methylphenyl)propanamide (8.42 g, 44.0 mmol) in MTBE (100 mL) and
54
55 TMEDA (6.8 mL, 44.0 mmol, 1.0 equiv), cooled to -20 °C, a solution of *n*-BuLi (38.7 mL,
56
57
58
59
60

1
2
3 96.8 mmol, 2.2 equiv, *n*-BuLi 2.5 M in hexane) was added dropwise by means of
4 syringe pump within 1h while the temperature was kept below -10 °C. The mixture was
5
6 aged at 10-15 °C for 5 h (a light-yellow solid has been formed), cooled below -25 °C,
7
8 and CF₃CO₂Et (7.35 mL, 61.8 mmol, 1.4 equiv) was added rapidly (internal temp.
9
10 reached 0 °C). Then reaction mixture was stirred for 16h at rt and quenched with 5%
11
12 HCl (100 mL). The organic phase was separated and the aqueous phase was extracted
13
14 with EtOAc (3 x 20 mL). Then combined organic phases were washed with brine (1 x
15
16 100 mL), dried over MgSO₄ and evaporated. The residue was chromatographed on
17
18 silica (5% EtOAc/hexanes) to give a yellow oil (8.31 g) and unreacted substrate (**S45**).
19
20 The crude TFMK was treated with 36% HCl (30 mL) at rt (white solid was formed
21
22 immediately) and stirred for 16h at 90 °C (temp. of oil bath, grey solid has been formed).
23
24 Then reaction mixture was cooled to rt, neutralized with sat. solution of Na₂CO₃ (pH = 9)
25
26 and extracted with EtOAc (4 x 30 mL). The combined organic extracts were washed
27
28 with brine (1 x 30 mL), dried over MgSO₄, and evaporated. The residue was purified by
29
30 crystallization (*n*-heptane, 12 mL, -45 °C, 2h) to give an orange solid (4.59 g, 51%). ¹H
31
32 NMR (400 MHz, CDCl₃) δ 7.51 (br s, 1H), 7.21 (dd, *J* = 8.6, 2.0 Hz, 1H), 6.65 (d, *J* = 8.6
33
34 Hz, 1H), 6.30 (br s, 2H), 2.26 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ 180.6 (q, *J*_{CF} =
35
36 33.1 Hz), 151.3, 138.2, 130.3 (q, *J*_{CF} = 4.0 Hz), 125.4, 117.5, 117.1 (q, *J*_{CF} = 289.8 Hz),
37
38 111.0, 20.4; ¹⁹F NMR (376 MHz, CDCl₃) δ -69.5. CAS registry number - 205756-35-0;
39
40 no spectroscopic data are available for comparison. Copy of spectra are included in this
41
42 experimental section.
43
44
45
46
47
48
49
50
51
52

53 **(1*R*,2*S*,5*R*)-5-Methyl-2-(propan-2-yl)cyclohexyl 4-nitrophenyl ether (15)** was
54 prepared by literature procedure,⁶² however 75% isolated reported in the literature has
55
56
57
58
59
60

1
2
3 not been achieved. To a solution of *p*-chloronitrobenzene (11.03 g, 70 mmol) and (-)-
4 mentol (10.93 g, 70 mmol) in dry DMSO (120 mL), a suspension of NaH (3.1 g, 77.0
5 mmol, 1.1 equiv) was added in portions at rt (evolution of hydrogen has not been
6 observed). The reaction mixture was heated at 70-75 °C (bath temp.; **strong evolution**
7 **of hydrogen !!!**) and the reaction mixture was kept at this temp. for 16 h. Then reaction
8 mixture cooled to rt, diluted with 5% HCl (300 mL) and extracted with EtOAc (3 x 100
9 mL). The combined organic extracts were washed brine (3 x 100 mL), dried over MgSO₄
10 and evaporated. The obtained black oil was adsorbed on silica (50 g) and was washed
11 with a mixture of EtOAc and hexanes (5% EtOAc/hexanes) to give an orange oil which
12 was further purified by crystallisation from MeOH (70 mL) to give a light-orange solid
13 (6.21 g, 32%). CAS registry number - 85002-82-0, 94730-54-8, no spectroscopic data
14 are available for comparison. IR (KBr) 3298, 2956, 2921, 2871, 1647, 1510 cm⁻¹; ¹H
15 NMR (400 MHz, CDCl₃) δ 8.21–8.14 (m, 2H, =C-H), 6.96–6.90 (m, 2H, =C-H), 4.16
16 (ddd, *J* = 10.6, 10.6, 4.2 Hz, 1H, CHOAr), 2.16–2.05 (m, 2H), 1.79–1.70 (m, 2H), 1.61–
17 1.45 (m, 2H), 1.18–0.96 (m, 3H), 0.94 (d, *J* = 7.7 Hz, 3H), 0.92 (d, *J* = 8.1 Hz, 3H), 0.75
18 (d, *J* = 7.0 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 163.70, 141.0, 126.0, 115.00, 78.4,
19 47.8, 39.9, 34.3, 31.4, 26.3, 23.8, 22.0, 20.6, 16.6.

20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44 **4-[[*(1R,2S,5R)*-5-Methyl-2-(propan-2-yl)cyclohexyl]oxy]aniline (16)** To a solution of
45 nitrocompound **15** (6.21 g, 22.4 mmol) in reagent-grade EtOAc (100 mL), 10% Pd/C
46 (238.3 mg, 0.22 mmol, 1 mol%) was added and the resulting mixture was shaken in
47 Parr apparat for 4h at rt (H₂ pressure - 4 bar; in some cases reduction of nitro group has
48 appeared strongly exothermic and the respective caution should be maintained; temp.
49 of the reaction mixture has increased from 17°C to 24 °C within 30 min.). Then argon
50
51
52
53
54
55
56
57
58
59
60

1
2
3 was bubbled through the reaction mixture for 5 min. and palladium catalyst was filtered
4 through a pad of Celite (washed with EtOAc), and solvent was evaporated to give a
5 light-yellow oil (5.5 g). The crude amine **S35** product was used in the next step without
6 further purification. CAS registry number - 94730-56-0, 94661-07-1.
7
8
9
10
11

12
13 **2,2-dimethyl-N-(4-[[[(1R,2S,5R)-5-methyl-2-(propan-2-**

14 **yl)cyclohexyl]oxy}phenyl)propanamide (17)** To a solution of aniline **16** (5.50 g, 22.4
15 mmol) and Et₃N (4.7 mL, 33.6 mmol) in DCM, cooled to 0 °C, PivCl (3.5 mL, 29.1 mmol,
16 1.3 equiv) was added dropwise. After 10 min. cooling bath was removed and reaction
17 mixture was stirred for 16h at rt. Then reaction mixture was diluted with 5% HCl (100
18 mL), organic phase was separated, dried over MgSO₄, and evaporated. The residue
19 was purified by crystallisation (*n*-heptane, 90 mL) to give a white crystalline amide **17**
20 (6.68 g, 90%). M.p. 168.6–169.2 °C (*n*-heptane); [α]_D²³ = -45.3 (c = 1.0, CHCl₃); ¹H NMR
21 (400 MHz, CDCl₃) δ 7.41–7.35 (m, 2H), 7.21 (br, 1H, NH) 6.88–6.82 (m, 2H), 3.95 (ddd,
22 *J* = 10.6, 10.6, 4.1 Hz, 1H, CHO), 2.35–2.16 (m, 1H), 2.15–2.07 (m, 1H), 1.76–1.65 (m,
23 2H), 1.54–1.36 (m, 2H), 1.30 [s, 9H, NHC(CH₃)₃], 1.17–0.86 (m, 4H) overlapping 0.92
24 (d, *J* = 7.0 Hz, 3H) and 0.91 (d, *J* = 6.5 Hz, 3H), 0.76 (d, *J* = 7.0 Hz, 2H); ¹³C NMR (100
25 MHz, CDCl₃) δ 176.3 (C=O), 163.7, 141.0, 126.0, 115.0, 78.4, 47.8, 39.9, 34.3, 31.4,
26 26.4, 23.8, 22.0, 20.6, 16.6; HRMS (ESI-TOF) *m/z*: [M+Na]⁺ Calcd for C₂₁H₃₃NO₂Na
27 354.2409; Found: 354.2409.
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48

49 **2,2-Dimethyl-N-[4-[[[(1R,2S,5R)-5-methyl-2-(propan-2-yl)cyclohexyl]oxy}-2-**

50 **(trifluoroacetyl)phenyl]propanamide (18)** To a solution of amide **17** (6.50 g, 19.6
51 mmol) and TMEDA (3.02 mL, 19.6 mmol, 1.0 equiv) in anhydrous MTBE (50 mL),
52 cooled to -20 °C, *n*-BuLi (18.7 mL, 43.1 mmol, 2.2 equiv, *n*-BuLi 2.31 M in hexane) was
53
54
55
56
57
58
59
60

1
2
3 added dropwise by means of syringe pump within 40 min. (the internal temp. has
4 increased to -8 °C). Then cooling bath was removed and the resulting reaction mixture
5 was stirred for 4h at 20 °C (bright-yellow precipitate has formed). Then reaction mixture
6 was cooled to -25 °C, and CF₃CO₂Et (3.3 mL, 27.4 mmol, 1.4 equiv) was added rapidly
7 (internal temp. has increased to 10 °C) and reaction mixture was stirred at rt for 1.5h.
8 The resulting reaction mixture was quenched with 5% HCl (200 mL), and aqueous
9 phase was extracted with EtOAc (3 x 50 mL). The combined organic extracts were
10 washed with brine (2 x 100 mL), dried over MgSO₄, and evaporated. The residue was
11 chromatographed on silica (5% EtOAc/hexanes) to give a bright orange oil (4.68 g,
12 56%). The resulting amide **18** was used in the next step without further purification.
13
14
15
16
17
18
19
20
21
22
23
24
25
26

27
28 **1-(2-Amino-5-[(1*R*,2*S*,5*R*)-5-methyl-2-(propan-2-yl)cyclohexyl]oxy}phenyl)-2,2,2-**
29 **trifluoroethanone (1i)** To round-bottom flask containing amide **18** (4.2 g, 9.82 mmol),
30 HCl in dioxane (35 mL, 4M in dioxane) and water (5 mL) were added and stirred for 2h
31 at 90 °C (TLC analysis indicated absence of substrate, 5% EtOAc/hexanes). Then
32 reaction mixture was cooled to rt, diluted with DCM (50 mL), neutralized with solid
33 NaHCO₃ (ca. 12 g) and stirred for additional 15 min. Then solid was filtered, washed
34 with DCM (5 times), and solvents were evaporated. The residue was chromatographed
35 on silica (5% EtOAc/hexanes) to give an deep-orange oil (2.98 g, 88%). $\alpha = -113.6$ ($c =$
36 0.69 , CHCl₃); IR (film) 3483, 3367, 2957, 2927, 2871, 1668, 1592, 1547, 1489 cm⁻¹; ¹H
37 NMR (400 MHz, CDCl₃) δ 7.25–7.21 (m, 1H), 7.09 (dd, $J = 9.1, 2.8$ Hz, 1H), 6.67 (d, $J =$
38 9.1 Hz, 1H), 6.19 (br s, 2H), 3.08 (ddd, $J = 10.5, 10.5, 4.2$ Hz, 1H), 2.32–2.20 (m, 1H),
39 2.12–2.02 (m, 1H), 1.76–1.63 (m, 2H), 1.52–1.34 (m, 2H), 1.14–0.86 (m, 3H)
40 overlapping 0.94 (d, $J = 6.8$ Hz, 3H) and 0.91 (d, $J = 7.0$ Hz, 3H), 0.81 (d, $J = 7.0$ Hz,
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 180.2 (q, $J_{\text{CF}} = 33.0$ Hz), 148.6, 148.5, 129.4, 118.7,
4
5 117.1 (q, $J_{\text{CF}} = 289.8$ Hz), 116.5 (q, $J_{\text{CF}} = 4.2$ Hz), 110.8, 79.8, 48.1, 40.4, 34.5, 31.4,
6
7 25.9, 23.5, 22.1, 20.8, 16.3; ^{19}F NMR (376 MHz, CDCl_3) δ -69.8; HRMS (ESI-TOF) m/z:
8
9 $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{18}\text{H}_{25}\text{F}_3\text{NO}_2$ 344.1837; Found: 344.1830.
10
11

12
13
14 **1-(2-Amino-5-chlorophenyl)-2,2,3,3,4,4,4-heptafluorobutan-1-on (1j)** To a solution of
15
16 *N*-(4-chlorophenyl)-2,2-dimethylpropanamide (5 g, 23.62 mmol) and TMEDA (23.62
17
18 mmol, 3.6 mL) in anhydrous THF (50 mL), cooled to -20 °C, *n*-BuLi (22.8 mL, 51.96
19
20 mmol, 2.2 equiv, 2.28 M) was added dropwise by means of syringe pump within 30 min.
21
22 (the internal temp. has increased to -8 °C; a white precipitated which was formed has
23
24 been dissolved to give an orange solution). The resulting reaction mixture was stirred
25
26 for 4h at 0 °C (bright-yellow precipitate has formed). Then reaction mixture was cooled
27
28 to -30 °C, and ethyl heptafluorobutanoate (5.7 mL, 33.1 mmol, 1.4 equiv) was added
29
30 rapidly (internal temp. has increased to 10 °C), cooling bath was removed, stirring was
31
32 continued for 1h and the reaction mixture was quenched with water (100 mL). The
33
34 aqueous phase was separated and extracted with EtOAc (3 x 30 mL). The combined
35
36 organic extracts were washed with brine (2 x 30 mL), dried over MgSO_4 , and
37
38 evaporated. The residue was chromatographed on silica (200 g, 10% EtOAc/hexanes)
39
40 to give a yellow oil (2.11 g) which was used in the next step without further purification.
41
42
43
44
45
46
47 The obtained partially purified ketone was treated with solution of HCl in dioxane (10
48
49 mL, 4M) and water (1 mL), and reaction mixture was heated at 90 °C for 2h (TLC
50
51 indicated absence of substrate). Then solvent was evaporated and crude amine
52
53 hydrochloride (*attempts to separate of free aminoketone from by-products has failed*)
54
55 was purified by chromatography on silica (5-25% EtOAc/hexanes) to give a bright-
56
57
58
59
60

1
2
3 yellow oil. The resulting hydrochloride was dissolved in CHCl₃ (10 mL) and sat. aq.
4 solution of NaHCO₃ (10 mL) was added. The biphasic mixture was vigorously stirred for
5
6 30 min. Then organic phase was separated, dried over MgSO₄, and evaporated to give
7
8 a bright-orange solid (951.6 mg, 10%). M.p. 57.8–59.1 °C (*n*-heptane); IR (film) 3494,
9
10 3370, 1664, 1622, 1587, 1539 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.78 (s, 1H), 7.32 (dd,
11
12 *J* = 9.0, 2.4 Hz, 1H), 6.69 (d, *J* = 9.0 Hz, 1H), 6.50 (br s, 2H); ¹³C NMR (100 MHz,
13
14 CDCl₃) δ 182.2 (t, *J*_{CF} = 24.2 Hz), 151.8, 136.9, 130.2–129.8 (m), 120.9, 119.2, signals
15
16 in the region of 120-102 ppm has been omitted in description of spectra for clarity due to
17
18 complicated multiplicity; ¹⁹F NMR (376 MHz, CDCl₃) δ -79.8 (t, *J* = 9.5 Hz), -110.9 – -
19
20 111.1 (m), -125.0 – - 125.2 (m); HRMS (ESI-TOF) *m/z*: [M+H]⁺ Calcd for C₁₀H₆F₇CINO
21
22 324.0026; Found: 324.0019.

23
24
25 **1-(2-Amino-5-chlorophenyl)-2,2-difluoroethanone (1k)** To a solution of *N*-(4-
26
27 chlorophenyl)-2,2-dimethylpropanamide (10.0 g, 44.7 mmol) and TMEDA (44.7 mmol,
28
29 6.8 mL, 2.0 equiv) in THF (100 mL), cooled to -40 °C, *n*-BuLi (39.3 mL, 98.3 mmol, 2.2
30
31 equiv, 2.5 M in hexane) was added dropwise by means of syringe pump within 30 min.
32
33 (temp. has increased to -25 °C), and the reaction was stirred for 2h at 0 - -5 °C degree
34
35 (internal temperature; reaction mixture was stirred by means of mechanical stirrer due
36
37 to lithium dianion precipitation). Then reaction mixture was cooled to -40 °C and 2,2-
38
39 difluoro-1-(morpholin-4-yl)ethanone (10.3 g, 62.6 mmol, 1.4 equiv) was added (the
40
41 internal temp. has risen to -15 °C). Then reaction mixture was allowed to stir below 0 °C
42
43 for additional 1h and quenched with 10% citric acid (100 mL). Then mixture was diluted
44
45 with EtOAc (100 mL), and aqueous phase was separated and extracted with EtOAc (2 x
46
47 50 mL). The combined organic phases were washed with brine (2 x 100 mL), dried over
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 MgSO₄, evaporated. The residue was chromatographed on silica (5% EtOAc/hexanes)
4
5 to give amide containing some impurities which was used in the next step without
6
7 further purification.
8
9

10 The obtained *o*-TFMK derivative was treated with 36% HCl (40 mL) and the resulting
11
12 mixture was heated at 90 °C for 16h. Then reaction mixture was quenched with sat.
13
14 solution of Na₂CO₃ (100 mL), and aqueous phase was extracted with MTBE (3 x 30
15
16 mL). The combined organic phases were washed with brine (2 x 100 mL), dried over
17
18 MgSO₄, and evaporated. The residue crystallized from
19
20 *n*-heptane (15 mL) to give an orange solid (3.13 g). The mother liquid was evaporated
21
22 and the resulting solid was recrystallized from *n*-heptane to give additional portion of
23
24 ketone **1k** (567.7 mg, summary yield 39%). ¹H NMR (400 MHz, CDCl₃) δ 7.80–7.40 (m,
25
26 1H), 7.29 (dd, *J*_{CF} = 9.0, 2.4 Hz, 1H), 6.57 (d, *J*_{CF} = 9.0 Hz, 1H), 6.39 (br s, 2H, NH)
27
28 overlapping 6.28 (t, *J*_{CF} = 53.6 Hz, 1H, CF₂H). Spectral data are in agreement with
29
30 those reported.⁵⁹
31
32
33
34
35

36 **1-(2-Aminopyridin-3-yl)-2,2,2-trifluoroethanone (6a)** To a solution of 2,2-dimethyl-*N*-
37
38 (pyridin-2-yl)propanamide amide (5 g, 28.05 mmol) and TMEDA (56.11 mmol, 8.6 mL,
39
40 2.0 equiv) in THF (50 mL), cooled to -60 °C, *n*-BuLi (24.6 mL, 56.11 mmol, 2.2 equiv,
41
42 2.28 M in hexane) was added dropwise by means of syringe pump within 30 min. (temp.
43
44 has increased to -50 °C), and the reaction was stirred for 4h at 0 - -5 °C degree (internal
45
46 temperature; reaction mixture was stirred by means of mechanical stirrer due to lithium
47
48 dianion precipitation). Then reaction mixture was cooled to -65 °C and CF₃CO₂Et (4.67
49
50 mL, 39.27 mmol, 1.4 equiv) was added (the internal temp. has risen to -45 °C). Then
51
52 cooling bath was removed, and the reaction mixture was allowed to warm up to -20 °C
53
54
55
56
57
58
59
60

1
2
3 (exothermic reaction begins at -35 °C) and stirred for additional 30 min. and quenched
4
5 with sat. NH₄Cl (100 mL). Then mixture was diluted with EtOAc (150 mL), and organic
6
7 phase was separated, washed with citric acid (1 x 100 mL), brine (2 x 100 mL), dried
8
9 over MgSO₄ and evaporated. The residue was chromatographed on silica (25-60%
10
11 EtOAc/hexanes) to give yellow solid (4.68 g). ¹H NMR indicated some impurities and
12
13 thus obtained pyridine derivative was used in the next step without further purification.
14
15

16
17 The crude pyridine derivative was treated with HCl in dioxane (20 mL, 4M in dioxane)
18
19 and water (2 mL) and the resulting mixture was heated at 90 °C for 16h. Then reaction
20
21 mixture was quenched with sat. solution of NaHCO₃ (100 mL), and the aqueous phase
22
23 was extracted with EtOAc (3 x 50 mL). The combined organic phases were washed with
24
25 brine (2 x 100 mL), dried over MgSO₄, and evaporated. The residue was crystallized (*n*-
26
27 heptane, to remove partially contaminations) and then was passed through pad of silica
28
29 (10% MeOH/DCM) to give a yellow solid (2.45 g, 46%). ¹H NMR (400 MHz, CDCl₃) δ
30
31 8.35 (dd, *J* = 4.6, 1.8 Hz, 1H), 8.10–8.04 (m, 1H), 7.17 (br s, 2H), 6.70 (dd, *J* = 20.2, 4.6
32
33 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 180.1 (q, *J*_{CF} = 33.8 Hz), 160.4, 156.8, 140.5 (q,
34
35 *J*_{CF} = 4.0 Hz), 114.8 (q, *J*_{CF} = 289.5 Hz), 112.8, 106.4; ¹⁹F NMR (376 MHz, CDCl₃) δ -
36
37 69.2. CAS registry number - 1060801-31-1; no spectroscopic data are available for
38
39 comparison. Copy of spectra are included in this experimental section.
40
41
42
43
44

45
46 **1-(4-Aminopyridin-3-yl)-2,2,2-trifluoroethanone (6b)** To a solution 2,2-dimethyl-*N*-
47
48 (pyridin-4-yl)propanamide (4.04 g, 22.66 mmol) TMEDA (45.34 mmol, 6.95 mL) amide
49
50 in THF (40), cooled to -60 °C, *n*-BuLi (19.9 mL, 45.34 mmol, 2.2 equiv, 2.28 M in
51
52 hexane) was added dropwise by means of syringe pump within 30 min. (temp. has
53
54 increased to -50 °C), and the reaction was stirred for 4h at 0 - -5 °C degree (internal
55
56
57
58
59
60

1
2
3 temperature; reaction mixture was stirred by means of mechanical stirrer due to lithium
4 dianion precipitation). Then reaction mixture was cooled to -65 °C and CF₃CO₂Et (3.77
5 mL, 31.73 mmol, 1.4 equiv) was added (the internal temp. has risen to -35 °C). Then
6 cooling bath was removed, and the reaction mixture was allowed to warm up to -10 °C
7 and stirred for additional 1h and quenched with 10% citric acid (100 mL). Then mixture
8 was diluted with EtOAc (100 mL), and aqueous phase was extracted with EtOAc (2 x 50
9 mL). The combined organic phases were washed with brine (2 x 100 mL), dried over
10 MgSO₄, evaporated and dried in vacuo to give a yellow foam (6.1 g) which was used in
11 the next step without further purification.
12
13
14
15
16
17
18
19
20
21
22
23

24 The crude pyridine derivative was treated with HCl in dioxane (20 mL, 4M in dioxane)
25 and water (2 mL) and the resulting mixture was heated at 90 °C for 16h. Then reaction
26 mixture was quenched with sat. solution of Na₂CO₃ (100 mL), and aqueous phase was
27 extracted with EtOAc (3 x 50 mL), The combined organic phases were washed with
28 brine (2 x 100 mL), dried over MgSO₄, and evaporated. The residue purified by
29 chromatography (5-10% MeOH/DCM) and further crystallized (toluene/*n*-heptane =
30 50:50) to give a brown-orange solid (2.99 g, 69%). ¹H NMR (400 MHz, CDCl₃) δ 8.93–
31 8.86 (m, 1H), 8.25 (d, *J* = 6.0 Hz, 1H), 6.61 (d, *J* = 6.0 Hz, 1H), (NH₂ group was
32 exchange with deuterium); ¹³C NMR (100 MHz, CDCl₃) δ 181.3 (q, *J*_{CF} = 35.2 Hz),
33 156.4, 154.1 (q, *J*_{CF} = 5.2 Hz), 153.0, 116.5 (q, *J*_{CF} = 289.0 Hz), 111.5, 109.3; ¹⁹F NMR
34 (376 MHz, CDCl₃) δ -70.1 CAS registry number: 1343447-95-9; no spectra available for
35 comparison. Copy of spectra are included in this experimental section.
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52

53 **1-(2-Amino-5-chloropyridin-3-yl)-2,2,2-trifluoroethanone (6c)** To a solution of *N*-(5-
54 chloropyridin-2-yl)-2,2-dimethylpropanamide (5.0 g, 23.51 mmol) in MTBE (50 mL) and
55
56
57
58
59
60

1
2
3 TMEDA (3.6 mL, 23.51 mmol, 1.0 equiv), cooled to -35 °C, a solution of *n*-BuLi (20.7
4 mL, 51.72 mmol, 2.2 equiv, 2.5 M in hexanes) was added dropwise by means of syringe
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

TMEDA (3.6 mL, 23.51 mmol, 1.0 equiv), cooled to -35 °C, a solution of *n*-BuLi (20.7 mL, 51.72 mmol, 2.2 equiv, 2.5 M in hexanes) was added dropwise by means of syringe pump within 1h while the temperature was kept below -20 °C. The mixture was aged at 0-5 °C for 2 h (a light-orange solid has been formed), cooled to -45 °C, and CF₃CO₂Et (3.92 mL, 32.91 mmol, 1.4 equiv) was added rapidly (internal temp. reached -10 °C). Then reaction mixture was stirred for 16h at rt and quenched with 5% HCl (100 mL). Then reaction mixture was stirred for 16h at rt and quenched with 5% HCl (50 mL). The organic phase was separated and the aqueous phase was extracted with EtOAc (3 x 20 mL). Then combined organic phases were washed with brine (1 x 100 mL), dried over MgSO₄ and evaporated. The residue was chromatographed on silica (25% EtOAc/hexanes - EtOAc) to give an orange solid (3.96 g). ¹H NMR indicated some impurities and thus obtained aminofluoroketone was used in the next step without further purification.

The crude aminopyridine was treated with 36% HCl (30 mL) at rt and stirred for 16h at 90 °C (temp. of oil bath, a white solid has been formed). Then reaction mixture was cooled to rt, neutralized with sat. solution of Na₂CO₃ (pH = 9) and extracted with EtOAc (4 x 30 mL). The combined organic extracts were washed with brine (1 x 30 mL), dried over MgSO₄, and evaporated. The residue was chromatographed on silica to give a yellow solid (1.15 g, 22%). ¹H NMR (400 MHz, CDCl₃) δ 8.31 (d, *J* = 2.5 Hz, 1H), 8.02–7.99 (m, 1H), 7.09 (br s, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 179.5 (q, *J*_{CF} = 34.8 Hz), 158.5, 155.7, 137.9 (q, *J*_{CF} = 4.2 Hz), 119.2, 116.6 (q, *J*_{CF} = 289.3 Hz), 106.4; ¹⁹F NMR (376 MHz, CDCl₃) δ -70.1. Spectral data are in agreement with those reported.⁶³

1
2
3 **General procedure for synthesis of quinoline and naphthydrine derivatives:** To a
4 screw-cap vial, trifluoromethylketone (TFMK), alkyne (1.2 equiv), copper catalyst (2
5 mol%) and DME (100 μ L) were added. The obtained mixture was stirred at rt (2-3 min.)
6 until complete dissolution of the substrates. Then, a solution of TMG (*N,N,N',N'*-
7 tetramethylguanidine, 0.01 mmol, 2 mol%) in degassed water (2 mL) was added. The
8 biphasic mixture was stirred for the indicated time at 100-120 $^{\circ}$ C (oil bath temp.) with
9 vigorous stirring. Next, the reaction mixture was diluted with EtOAc (2-3 mL), and the
10 aqueous phase was separated and extracted with two additional portions of EtOAc. The
11 combined extracts were washed with brine (1 x 10 mL), dried over $MgSO_4$ and
12 evaporated. The residue was chromatographed on silica (silica : crude reaction mixture
13 = 30 : 1 w/w) using an appropriate eluting system to afford the product.
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28

29 **6-Chloro-2-cyclopropyl-4-(trifluoromethyl)quinoline (3a)** was obtained according to
30 the general procedure from ketone **1a** (223.6 mg, 1.0 mmol), TMG (2.5 μ L, 0.02 mmol,
31 2 mol%), ethynylcyclopropane (**2a**) (134.0 μ L, 1.2 mmol, 1.2 eq.), IPr^*CuCl (20.2 mg,
32 0.02 mmol, 2 mol%), water (2 mL) and DME (0.1 mL). The crude product was
33 chromatographed on silica (2% EtOAc/hexanes) to give a colourless oil which
34 spontaneously solidified upon standing in the fridge (96.4 mg, 71%). 1H NMR (400
35 MHz, $CDCl_3$) δ 8.03-7.99 (m, 1H), 7.94 (dd, $J = 9.0, 0.3$ Hz, 1H), 7.65 (dd, $J = 9.0, 2.2$
36 Hz, 1H), 7.54 (s, 1H), 2.29–2.21 (m, 1H), 1.27–1.21 (m, 2H), 1.18–1.11 (m, 2H). 1H
37 NMR data are in agreement with those reported.⁶⁴
38
39
40
41
42
43
44
45
46
47
48
49

50 **Gram –scale preparation of 3a** was conducted according to the general procedure
51 with ketone **1a** (1.12 g, 5.0 mmol), TMG (12.5 μ L, 0.1 mmol, 2 mol%),
52 ethynylcyclopropane (**2a**) (0.5 mL, 6.0 mmol, 1.2 equiv), IPr^*CuCl (101.2 mg, 0.1 mmol,
53
54
55
56
57
58
59
60

1
2
3 2 mol%), water (10 mL) and DME (0.5 mL). The crude product was chromatographed
4
5 on silica (2% EtOAc/hexanes) to give quinolone 3a (1.032 g, 76%) and diazocine 4a
6
7 (75.1 mg, 7%)
8
9

10 **6-Chloro-2-phenyl-4-(trifluoromethyl)quinoline (3b)** was obtained according to the
11
12 general procedure from ketone 1a (111.8 mg, 0.5 mmol), TMG (1.25 μ L, 0.01 mmol, 2
13
14 mol%), ethynylbenzene (2b) (67.0 μ L, 0.6 mmol, 1.2 equiv), IPr*CuCl (10.2 mg, 0.01
15
16 mmol, 2 mol%), DME (100 μ L) and water (2 mL). The reaction mixture was heated at
17
18 100 °C for 16h. The crude product was chromatographed on silica (2% EtOAc/hexanes)
19
20 to give a colourless oil (66.6 mg, 43%). IR (KBr) 3094, 3063, 3038, 1610, 1550, 1152,
21
22 1121 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 8.22–8.15 (m, 4H), 8.13–8.09 (m, 1H), 7.75
23
24 (dd, $J = 9.0, 2.2$ Hz, 1H), 7.60–7.49 (m, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 156.8, 147.5,
25
26 137.9, 134.3 (q, $J_{\text{CF}} = 31.7$ Hz), 134.0, 132.1, 131.4, 130.3, 129.1, 127.4, 123.3 (q, $J_{\text{CF}} =$
27
28 273.1 Hz), 122.9 (q, $J_{\text{CF}} = 2.3$ Hz), 122.4, 116.7 (q, $J_{\text{CF}} = 5.2$ Hz); ^{19}F NMR (376 MHz) δ
29
30 -61.7; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{16}\text{H}_{10}\text{ClF}_3\text{N}$ 308.0454; Found
31
32 308.0455.
33
34
35
36
37
38

39 **6-Chloro-2-(4-methoxyphenyl)-4-(trifluoromethyl)quinoline (3c)** was obtained
40
41 according to the general procedure from ketone 1a (111.8 mg, 0.5 mmol), TMG (1.25
42
43 μ L, 0.01 mmol, 2 mol%), 1-ethynyl-4-methoxybenzene (2c) (78.0 μ L, 0.6 mmol, 1.2
44
45 equiv), IPr*CuCl (10.2 mg, 0.01 mmol, 2 mol%), DME (100 μ L) and water (2 mL). The
46
47 reaction mixture was heated at 100 °C for 16h. The crude product was
48
49 chromatographed on silica (2% EtOAc/hexanes) to give a colourless oil (107.6 mg,
50
51 64%). IR (KBr) 3070, 2967, 2938, 2917, 2840, 1610, 1546, 1148, 1121 cm^{-1} ; ^1H NMR
52
53 (400 MHz, CDCl_3) δ 8.18–8.10 (m, 4H), 8.09–8.04 (m, 1H), 7.71 (dd, $J = 9.0, 2.2$ Hz,
54
55
56
57
58
59
60

1
2
3 1H), 7.09–7.02 (m, 2H), 3.90 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 161.6, 156.3, 147.5,
4
5
6 134.1 (q, *J*_{CF} = 31.6 Hz), 133.4, 131.8, 131.3, 130.5, 128.9, 123.3 (q, *J*_{CF} = 273.2 Hz),
7
8 122.9 (q, *J*_{CF} = 2.2 Hz), 122.0, 116.2 (q, *J*_{CF} = 5.3 Hz), 114.5, 55.4; ¹⁹F NMR (376 MHz)
9
10 δ -61.7; HRMS (ESI-TOF) *m/z*: [M+H]⁺ Calcd for C₁₇H₁₂ClF₃NO 338.0560; Found:
11
12 338.0561.
13

14
15 **6-Chloro-4-(trifluoromethyl)-2-(3,4,5-trimethoxyphenyl)quinoline (3d)** was obtained
16 according to the general procedure from ketone **1a** (111.8 mg, 0.5 mmol), TMG (1.25
17 μL, 0.01 mmol, 2 mol%), 1-ethynyl-3,4,5-trimethoxybenzene (**2d**) (115.0 mg, 0.6 mmol,
18 1.2 equiv), IPr^{*}CuCl (10.2 mg, 0.01 mmol, 2 mol%), DME (100 μL) and water (2 mL).
19
20 The reaction mixture was heated at 100 °C for 16h. The crude product was
21 chromatographed on silica (5% EtOAc/toluene) to give a colourless oil which
22 spontaneously solidified upon standing in the fridge (106.0 mg, 53%). M.p. 134.5–136.5
23 °C; IR (KBr) 3097, 2937, 2842, 1581, 1167, 1126 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ
24 8.18 (d, *J* = 8.8 Hz, 1H), 8.13–8.05 (m, 2H), (dd, *J* = 9.0, 2.2 Hz, 1H), 7.40 (br s, 2H),
25 4.02 (s, 6H), 3.94 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 156.4, 153.8, 147.4, 140.5,
26 134.4 (q, *J*_{CF} = 31.7 Hz), 133.9, 133.4, 132.0, 131.5, 123.3 (q, *J*_{CF} = 273.1 Hz), 122.9 (q,
27 *J*_{CF} = 2.4 Hz), 122.3 (q, *J*_{CF} = 1.1 Hz), 116.7 (q, *J*_{CF} = 5.2 Hz), 104.9, 61.0, 56.4; ¹⁹F
28 NMR (376 MHz) δ -61.6; HRMS (ESI-TOF) *m/z*: [M+H]⁺ Calcd for C₁₉H₁₆ClF₃NO₃
29 398.0771; Found: 398.0769.
30
31

32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49 **6-Chloro-2-(4-methylphenyl)-4-(trifluoromethyl)quinoline (3e)** was obtained
50 according to the general procedure from ketone **1a** (111.8 mg, 0.5 mmol), TMG (1.25
51 μL, 0.01 mmol, 2 mol%), 1-ethynyl-4-methylbenzene (**2e**) (76.1 μL, 0.6 mmol, 1.2
52 equiv), IPr^{*}CuCl (10.2 mg, 0.01 mmol, 2 mol%), DME (100 μL) and water (2 mL). The
53
54
55
56
57
58
59
60

1
2
3 reaction mixture was heated at 100 °C for 16h. The crude product was
4 chromatographed on silica (2.5 % EtOAc/hexanes) to give a colourless oil which
5 spontaneously solidified upon standing in the fridge (109.8 mg, 77%). ¹H NMR (400
6 MHz, CDCl₃) δ 8.19–8.15 (m, 2H), 8.11–8.06 (m, 3H), 7.74 (dd, *J* = 9.0, 2.2 Hz, 1H),
7 7.38–7.33 (m, 2H), 2.45 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 156.7, 147.5, 140.6,
8 135.1, 134.1 (q, *J*_{CF} = 31.6 Hz), 133.7, 132.0, 131.2, 129.8, 127.3, 123.3 (q, *J*_{CF} = 273.1
9 Hz), 129.9 (q, *J*_{CF} = 2.3 Hz), 122.2, 116.5 (q, *J*_{CF} = 5.2 Hz), 21.4; ¹⁹F NMR (376 MHz) δ -
10 61.7. ¹H NMR data are in agreement with those reported.²² ¹³C NMR spectra are
11 described by the authors without C-F coupling constants and the accurate description is
12 provided above. A copy of the spectra is included in the ESI.

13
14
15
16
17
18
19
20
21
22
23
24
25
26
27 **6-Chloro-2-(pyridin-3-yl)-4-(trifluoromethyl)quinoline (3f)** was obtained according to
28 the general procedure from ketone **1a** (111.8 mg, 0.5 mmol), TMG (1.25 μL, 0.01 mmol,
29 2 mol%), 3-ethynylpyridine (**2f**) (62.0 mg, 0.6 mmol, 1.2 equiv), IPr*CuCl (10.2 mg, 0.01
30 mmol, 2 mol%), DME (100 μL) and water (2 mL). The reaction mixture was heated at
31 100 °C for 16h. The crude product was chromatographed on silica (25%
32 EtOAc/hexanes) to give a colourless oil which spontaneously solidified upon standing in
33 the fridge (17.8 mg, 11%). M.p. 157.2–157.5 °C; IR (film) 3059, 1613, 1126 cm⁻¹; ¹H
34 NMR (400 MHz, CDCl₃) δ 9.41 (br s, 1H), 8.78 (br s, 1H), 8.50 (d, *J* = 8.0 Hz, 1H), 8.24–
35 8.16 (m, 2H), 8.15–8.10 (m, 1H), 7.78 (dd, *J* = 9.2, 2.0 Hz, 1H), 7.55–7.44 (m, 1H); ¹³C
36 NMR (100 MHz, CDCl₃) δ 154.3, 151.1, 148.7, 147.6, 134.8 (q, *J*_{CF} = 31.9 Hz), 134.7,
37 134.7, 132.2, 131.8, 123.9 (m), 123.1 (q, *J*_{CF} = 273.2 Hz), 123.0 (q, *J*_{CF} = 2.3 Hz), 122.6,
38 116.2 (q, *J*_{CF} = 5.2 Hz); ¹⁹F NMR (376 MHz) δ -61.7; HRMS (EI) *m/z*: [M⁺] Calcd for
39 C₁₅H₈ClF₃N₂ 308.0328; Found: 308.0321.
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 **6-Chloro-2-(thiophen-3-yl)-4-(trifluoromethyl)quinoline (3g)** was obtained according
4
5 to the general procedure from ketone **1a** (111.8 mg, 0.5 mmol), TMG (1.25 μ L, 0.01
6
7 mmol, 2 mol%), 3-ethynylthiophene (**2g**) (59.0 μ L, 0.6 mmol, 1.2 equiv), IPr*CuCl (10.2
8
9 mg, 0.01 mmol, 2 mol%), DME (100 μ L) and water (2 mL). The reaction mixture was
10
11 heated at 100 °C for 16h. The crude product was chromatographed on silica (3%
12
13 EtOAc/hexanes) to give a colourless oil which spontaneously solidified upon standing in
14
15 the fridge (84.1 mg, 53%). M.p. 145.0–146.0 °C; IR (KBr) 3095, 1641, 1554, 1148, 1119
16
17 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 8.19–8.00 (m, 4H), 7.86 (dd, J = 5.1, 1.2 Hz, 1H),
18
19 7.71 (dd, J = 9.0, 2.2 Hz, 1H), 7.47 (q, J = 5.1, 2.2 Hz, 1H); ^{13}C NMR (100 MHz, CDCl_3)
20
21 δ 152.7, 147.5, 141.2, 134.2 (q, J_{CF} = 31.7 Hz), 133.7, 131.8, 131.4, 127.0, 126.5,
22
23 125.8, 123.2 (q, J_{CF} = 273.2 Hz), 123.0 (q, J_{CF} = 2.2 Hz), 122.2, 116.9 (q, J_{CF} = 5.3 Hz);
24
25 ^{19}F NMR (376 MHz) δ -61.8; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{14}\text{H}_8\text{ClF}_3\text{SNa}$
26
27 314.0018; Found: 314.0017.

28
29
30
31
32
33 **6-Chloro-2-(cyclohex-1-en-1-yl)-4-(trifluoromethyl)quinoline (3h)** was obtained
34
35 according to the general procedure from ketone **1a** (111.8 mg, 0.5 mmol), TMG (1.25
36
37 μ L, 0.01 mmol, 2 mol%), 1-ethynylcyclohexene (**2h**) (70.5 μ L, 0.6 mmol, 1.2 equiv),
38
39 IPr*CuCl (10.2 mg, 0.01 mmol, 2 mol%), DME (100 μ L) and water (2 mL). The reaction
40
41 mixture was heated at 100 °C for 16h. The crude product was chromatographed on
42
43 silica (3% EtOAc/hexanes) to give a colourless oil which spontaneously solidified upon
44
45 standing in the fridge (114.1 mg, 73%). M.p. 101.8–105.2 °C; IR (KBr) 3048, 2942,
46
47 2910, 2832, 1632, 1612, 1550, 1153, 1125 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 8.08–
48
49 8.00 (m, 2H), 7.90 (br s, 1H), 7.70–7.64 (m, 1H), 6.88–6.83 (m, 1H), 2.73–2.65 (m, 2H),
50
51 2.39–2.31 (m, 2H), 1.89–1.80 (m, 2H), 1.78–1.69 (m, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ
52
53
54
55
56
57
58
59
60

1
2
3 158.4, 147.0, 137.0, 133.4 (q, $J_{CF} = 31.4$ Hz), 133.2, 132.4, 131.8, 130.9, 123.4 (q, $J_{CF} =$
4 272.2 Hz), 122.8 (q, $J_{CF} = 2.3$ Hz), 122.0, 115.8 (q, $J_{CF} = 5.3$ Hz), 26.3, 25.6, 22.6, 22.0;
5
6
7
8 ^{19}F NMR (376 MHz) δ -61.8; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{16}\text{H}_{14}\text{ClF}_3\text{N}$
9 312.0767; Found: 312.0768.

10
11
12 **6-Chloro-4-(trifluoromethyl)-2-[4-(trifluoromethyl)phenyl] (3i)** was obtained
13 according to the general procedure from ketone **1a** (111.8 mg, 0.5 mmol), TMG (1.25
14 μL , 0.01 mmol, 2 mol%), 1-ethynyl-4-trifluoromethylbenzene (**2i**) (98.0 μL , 0.6 mmol, 1.2
15 equiv), IPr^*CuCl (10.2 mg, 0.01 mmol, 2 mol%), DME (100 μL) and water (2 mL). The
16 reaction mixture was heated at 100 °C for 16h. The crude product was
17 chromatographed on silica (2% EtOAc/hexanes) to give a colourless oil which
18 spontaneously solidified upon standing in the fridge (29.0 mg, 15%). M.p. 165.5–166.8
19 °C; IR (KBr) 3070, 2967, 2938, 2916, 2840, 1610, 1546, 1149, 1121 cm^{-1} ; ^1H NMR (400
20 MHz, CDCl_3) δ 8.30 (br d, $J = 8.2$ Hz, 2H), 8.24–8.16 (m, 2H), 8.15–8.11 (m, 1H), 7.84–
21 7.75 (m, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 155.1, 147.5, 141.1, 134.8, 134.7 (q, $J_{CF} =$
22 31.9 Hz), 132.5, 132.1 (q, $J_{CF} = 32.4$ Hz), 131.8, 127.8, 126.0 (q, $J_{CF} = 3.8$ Hz), 124.0
23 (q, $J_{CF} = 270.1$ Hz), 123.2 (q, $J_{CF} = 273.1$ Hz), 123.0 (q, $J_{CF} = 2.4$ Hz), 122.7, 116.5 (q,
24 $J_{CF} = 5.3$ Hz); ^{19}F NMR (376 MHz) δ -61.7, -62.8; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd
25 for $\text{C}_{17}\text{H}_8\text{ClF}_6\text{N}$ 376.0328; Found: 376.0323.

26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46 **6-Chloro-2-hexyl-4-(trifluoromethyl)quinoline (3j)** was obtained according to the
47 general procedure from ketone **1a** (111.8 mg, 0.5 mmol), TMG (1.25 μL , 0.01 mmol, 2
48 mol%), 1-octyne (**2j**) (88.5 μL , 0.6 mmol, 1.2 equiv), IPr^*CuCl (10.2 mg, 0.01 mmol, 2
49 mol%), DME (100 μL) and water (2 mL). The reaction mixture was heated at 100 °C for
50 16h. The crude product was chromatographed on silica (2% EtOAc/hexanes) to give a
51
52
53
54
55
56
57
58
59
60

1
2
3 colourless oil (110.6 mg, 70%). IR (film) 2957, 2929, 2858, 1613, 1164, 1137 cm^{-1} ; ^1H
4
5 NMR (400 MHz, CDCl_3) δ 8.09–8.03 (m, 2H), 7.70 (dd, $J = 9.0, 2.4$ Hz, 1H), 7.59 (br s,
6
7 1H), 3.01 (t, $J = 7.6$ Hz, 2H), 1.91–1.74 (m, 2H), 1.52–1.21 (m, 6H), 0.89 (t, $J = 6.8$ Hz,
8
9 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 162.8, 147.2, 133.6 (q, $J_{\text{CF}} = 31.7$ Hz), 133.4, 131.3,
10
11 131.0, 123.3 (q, $J_{\text{CF}} = 273.0$ Hz), 122.9 (q, $J_{\text{CF}} = 2.3$ Hz), 122.0, 119.2 (q, $J_{\text{CF}} = 5.1$ Hz),
12
13 39.2, 31.6, 29.5, 29.1, 22.5, 14.0; ^{19}F NMR (376 MHz) δ -61.7; HRMS (ESI-TOF) m/z :
14
15
16
17
18 $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{16}\text{H}_{18}\text{ClF}_3\text{N}$ 316.1080; Found: 316.1080.

19
20 **6-Chloro-2-(2-methylpropyl)-4-(trifluoromethyl)quinoline (3k)** was obtained
21
22 according to the general procedure from ketone **1a** (111.8 mg, 0.5 mmol), TMG (1.25
23
24 μL , 0.01 mmol, 2 mol%), 4-methylpent-1-yne (**2k**) (70.5 μL , 0.6 mmol, 1.2 equiv),
25
26 IPr^*CuCl (10.2 mg, 0.01 mmol, 2 mol%), DME (100 μL) and water (2 mL). The reaction
27
28 mixture was heated at 100 $^\circ\text{C}$ for 16h. The crude product was chromatographed on
29
30 silica (5% EtOAc/hexanes) to give a colourless oil which spontaneously solidified upon
31
32 standing in the fridge (87.4 mg, 61%). M.p. 56.0–58.0 $^\circ\text{C}$; IR (KBr) 3107, 3071, 2958,
33
34 2872, 1608, 1154, 1126 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 8.11–8.02 (m, 2H), 7.70 (dd,
35
36 $J = 9.0, 2.2$ Hz, 1H), 7.56 (s, 1H), 2.89 (d, $J = 7.6$ Hz, 2H), 2.24 (hept, $J = 6.8$ Hz, 1H),
37
38 0.99 (d, $J = 6.4$ Hz, 6H); ^{13}C NMR (100 MHz, CDCl_3) δ 162.0, 147.2, 133.4 (q, $J_{\text{CF}} =$
39
40 31.7 Hz), 134.4, 131.4, 131.0, 123.2 (q, $J_{\text{CF}} = 273.0$ Hz), 122.9 (q, $J_{\text{CF}} = 2.3$ Hz), 122.0,
41
42 119.6 (q, $J_{\text{CF}} = 5.1$ Hz), 48.1, 29.2, 22.5; ^{19}F NMR (376 MHz) δ -61.7; HRMS (ESI-TOF)
43
44 m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{14}\text{H}_{14}\text{ClF}_3\text{N}$ 288.0767; Found: 288.0768.

45
46
47
48
49
50 **3-[6-Chloro-4-(trifluoromethyl)quinolin-2-yl]propan-1-ol (3l)** was obtained according
51
52 to the general procedure from ketone **1a** (111.8 mg, 0.5 mmol), TMG (1.25 μL , 0.01
53
54 mmol, 2 mol%), pent-4-yne-1-ol (**2l**) (50.4 mg, 0.6 mmol, 1.2 equiv), IPr^*CuCl (10.2 mg,
55
56
57
58
59
60

0.01 mmol, 2 mol%), DME (100 μ L) and water (2 mL). The reaction mixture was heated at 100 $^{\circ}$ C for 16h. The crude product was chromatographed on silica (40–50% EtOAc/hexanes) to give a colourless oil which spontaneously solidified upon standing in the fridge (104.9 mg, 68%). M.p. 94.0–95.0 $^{\circ}$ C; IR (KBr) 3374, 3296, 3097, 3067, 3039, 2964, 2893, 1617, 1557, 1152, 1124 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 8.08–8.01 (m, 2H), 7.71 (dd, J = 9.0, 2.2 Hz, 1H), 7.63 (br s, 1H), 3.77 (t, J = 6.0 Hz, 2H), 3.19 (t, J = 7.1 Hz, 2H), 2.18–2.08 (m, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 162.0, 146.7, 134.0 (q, J_{CF} = 31.8 Hz), 133.7, 131.3, 131.0, 123.1 (q, J_{CF} = 273.4 Hz), 122.9 (q, J_{CF} = 2.3 Hz), 122.0, 119.5 (q, J_{CF} = 5.0 Hz), 62.0, 35.8, 31.1; ^{19}F NMR (376 MHz) δ -61.7; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ calcd for $\text{C}_{13}\text{H}_{12}\text{ClF}_3\text{NO}$ 290.0560; Found: 290.0555.

Methyl 4-[6-chloro-4-(trifluoromethyl)quinolin-2-yl]butanoate (3m) was obtained according to the general procedure from ketone **1a** (111.8 mg, 0.5 mmol), TMG (1.25 μ L, 0.01 mmol, 2 mol%), methyl hex-5-ynoate (**2m**) (74.0 μ L, 0.6 mmol, 1.2 equiv), IPr^*CuCl (10.2 mg, 0.01 mmol, 2 mol%), DME (100 μ L) and water (2 mL). The reaction mixture was heated at 100 $^{\circ}$ C for 16h. The crude product was chromatographed on silica (5% EtOAc/toluene) to give a colourless oil which spontaneously solidified upon standing in the fridge (82.7 mg, 50%). M.p. 54.0–56.0 $^{\circ}$ C; IR (KBr) 3077, 2973, 2954, 2921, 2898, 2852, 1732, 1620, 1560, 1146, 1125 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 8.10–8.02 (m, 2H), 7.71 (dd, J = 9.0, 2.2 Hz, 1H); 7.60 (s, 1H), 3.67 (s, 3H), 3.07 (t, J = 7.6 Hz, 2H), 2.46 (t, J = 7.4 Hz, 2H), 2.26–2.14 (m, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 173.5, 161.4, 147.1, 133.8 (q, J_{CF} = 31.8 Hz), 133.6, 131.4, 131.2, 123.2 (q, J_{CF} = 273.1 Hz), 122.9 (q, J_{CF} = 2.4 Hz), 122.1 (q, J_{CF} = 1.2 Hz), 119.2 (q, J_{CF} = 5.1 Hz), 51.6, 38.0,

33.3, 24.1; ^{19}F NMR (376 MHz) δ -61.7; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{15}\text{H}_{14}\text{ClF}_3\text{NO}_2$ 332.0665; Found: 332.0666.

6-Chloro-2-(3-([(3a*S*,4*S*,5a*R*,9a*R*,9b*S*)-2,2,8,8-

tetramethylhexahydro[1,3]dioxolo[4,5]pyrano[3,2-*d*][1,3]dioxin-4-yl]oxy}propyl)-4-
(trifluoromethyl)quinoline (3n) was prepared according to the general procedure from

o-TFMK **1a** (111.8 mg, 0.5 mmol), (3a*S*,4*S*,5a*R*,9a*R*,9b*S*)-2,2,8,8-Tetramethyl-4-(pent-4-yn-1-yloxy)hexahydro[1,3]dioxolo[4,5]pyrano [3,2-*d*][1,3]dioxine (**2n**) (163.1 mg, 0.5 mmol), IPr*CuCl (10.1 mg, 0.00005 mmol, 2 mol%), TMG (1.25 μL , 0.00005 mmol, 2 mol%), DME (100 μL) and water (2 mL). Reaction mixture was heated at 100 $^\circ\text{C}$ for 24 h. Then reaction mixture was extracted with EtOAc (3 x 10 mL), and the combined organic phases were washed with brine (1 x 20 mL), dried over MgSO_4 and evaporated. The residue was chromatographed on silica (20–30% MTBE/hexanes) to give a light-yellow oil (181.4 mg, 68%). Purity based on HPLC: 99% (major peak 8.4 min., LiChrospher $^\circledR$ Si60 column, 250 mm x 4.6 mm, 5 μm , DAD detector 209 nm, MTBE/hexane = 40:60; 1.0 mL/min); M.p. 95.5–96.4 $^\circ\text{C}$; $[\alpha]_D^{23} = 5.8$ ($c = 1.1$, CHCl_3); IR (film) 2991, 2938, 1613, 1496 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 8.10–8.03 (m, 2H), 7.71 (dd, $J = 9.2, 2.4$ Hz, 1H), 7.61 (s, 1H), 5.0 (s, 1H), 4.11–4.05 (m, 2H), 3.86–3.77 (m, 2H), 3.77–3.68 (m, 2H), 3.61–3.48 (m, 2H), 3.12 (t, $J = 7.6$ Hz, 2H), 2.25–2.08 (m, 2H), 1.54 (s, 3H), 1.51 (s, 3H), 1.43 (s, 3H), 1.32 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 161.7, 147.2, 133.8 (q, $J_{\text{CF}} = 31.7$ Hz), 133.6, 131.4, 131.2, 123.2 (q, $J_{\text{CF}} = 273.1$ Hz), 122.9 (q, $J_{\text{CF}} = 2.3$ Hz), 122.0, 119.2 (q, $J_{\text{CF}} = 5.1$ Hz), 109.4, 99.7, 97.9, 76.1, 74.9, 72.7, 66.9, 35.6, 29.0, 28.7, 28.2, 26.1, 18.8; ^{19}F NMR (376 MHz, CDCl_3) δ -61.7; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{25}\text{H}_{30}\text{ClF}_3\text{NO}_6$ 532.1714; Found: 532.1703.

3-[6-Chloro-4-(trifluoromethyl)quinolin-2-yl]-1-(4-fluorophenyl)propan-1-one (3o)

was obtained according to the general procedure from ketone **1a** (111.8 mg, 0.5 mmol), TMG (1.25 μ L, 0.01 mmol, 2 mol%), 1-(4-Fluorophenyl)pent-4-yl-1-one (**2o**) (105.7 mg, 0.6 mmol, 1.2 equiv), IPr*CuCl (10.2 mg, 0.01 mmol, 2 mol%), DME (100 μ L) and water (2 mL). The reaction mixture was heated at 100 °C for 16h. The crude product was chromatographed on silica (5% EtOAc/hexanes) to give a colourless oil which spontaneously solidified upon standing in the fridge (105.0 mg, 55%). M.p. 141.5–142.5 °C; IR (film) 3357, 3038, 3058, 2926, 1688, 1616, 1600, 1561 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 8.11–8.01 (m, 3H), 7.93 (br d, $J = 9.2$ Hz, 1H), 7.71 (br s, 1H), 7.67 (dd, $J = 9.0, 2.0$ Hz, 1H), 7.19–7.10 (m, 2H), 3.63 (t, $J = 6.6$ Hz, 2H), 3.49 (t, $J = 6.6$ Hz, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 197.3, 165.8 (d, $J_{\text{CF}} = 253.1$ Hz), 160.7, 147.0, 133.6 (q, $J_{\text{CF}} = 31.8$ Hz), 133.5, 133.4 (d, $J_{\text{CF}} = 3.0$ Hz), 131.1 (d, $J_{\text{CF}} = 20.7$ Hz), 130.7 (d, $J_{\text{CF}} = 9.2$ Hz), 123.2 (q, $J_{\text{CF}} = 273.1$ Hz), 123.0 (q, $J_{\text{CF}} = 2.4$ Hz), 122.2, 119.9 (q, $J_{\text{CF}} = 5.2$ Hz), 115.7 (d, $J_{\text{CF}} = 21.7$ Hz), 36.2, 32.4; ^{19}F NMR (376 MHz) δ -61.7, -105.2; HR MS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{19}\text{H}_{13}\text{ClF}_4\text{NO}$ 382.0619; Found: 382.0622.

4-[6-Chloro-4-(trifluoromethyl)quinolin-2-yl]-N-[(1S)-1-phenylethyl]butanamide (3p)

was prepared according to the general procedure from *o*-TFMK **1a** (111.8 mg, 0.5 mmol), *N*-[(1S)-1-phenylethyl]hex-5-ynamide (**2p**) (129.2 mg, 0.6 mmol, 1.2 equiv), IPr*CuCl (10.1 mg, 0.00005 mmol, 2 mol%), TMG (1.25 μ L, 0.00005 mmol, 2 mol%), DME (100 μ L) and water (2 mL). The reaction mixture was heated at 100 °C for 16 h. Then reaction mixture was extracted with EtOAc (3 x 10 mL), and combined organic phases were washed with brine (1 x 20 mL), dried over MgSO_4 and evaporated. The residue was chromatographed on silica (30–50% EtOAc/hexanes) to give a white solid

(132.4 mg, 63%). M.p. 161.7–162.8 °C (*n*-heptane); $[\alpha]_D^{23} = -26.2$ ($c = 0.22$, CHCl_3); IR (KBr) 3282, 3062, 2971, 2925, 2872, 1636, 1543 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 8.07–8.03 (m, 1H), 7.93 (d, $J = 9.0$ Hz, 1H), 7.68 (dd, $J = 9.0, 2.2$ Hz, 1H), 7.59 (s, 1H), 7.37–7.22 (m, 5H), 6.25 (br d, $J = 7.4$ Hz, 1H), 5.20–5.10 (m, 1H), 3.05 (t, $J = 7.3$ Hz, 2H), 2.29–2.11 (m, 4H), 1.51 (d, $J = 6.9$ Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 171.3, 161.6, 146.9, 143.3, 134.0 (q, $J_{\text{CF}} = 31.8$ Hz), 133.5, 131.2, 131.1, 128.6, 127.3, 126.2, 123.1 (q, $J_{\text{CF}} = 273.2$ Hz), 122.9 (q, $J_{\text{CF}} = 2.4$ Hz), 122.1, 119.4 (q, $J_{\text{CF}} = 5.1$ Hz), 48.8, 37.5, 35.4, 25.0, 21.7; ^{19}F NMR (376 MHz, CDCl_3) δ -61.7; HRMS (ESI-TOF) m/z : $[\text{M}+\text{Na}]^+$ Calcd for $\text{C}_{22}\text{H}_{20}\text{ClF}_3\text{N}_2\text{ONa}$ 443.1114; Found: 443.1108.

3-[6-Chloro-4-(trifluoromethyl)quinolin-2-yl]propyl 2,3,4,6-tetra-O-acetyl- α -D-

mannopyranoside (3q) was prepared according to the general procedure from *o*-TFMK **1a** (111.8 mg, 0.5 mmol), pent-4-yn-1-yl 2,3,4,6-tetra-O-acetyl- α -D-mannopyranoside (**2q**) (207.1 mg, 0.5 mmol), IPr^*CuCl (10.1 mg, 0.00005 mmol, 2 mol%), TMG (1.25 μL , 0.00005 mmol, 2 mol%), DME (100 μL) and water (2 mL). Reaction mixture was heated at 100 °C for 48 h. Then reaction mixture was extracted with EtOAc (3 x 10 mL), and the combined organic phases were washed with brine (1 x 20 mL), dried over MgSO_4 and evaporated. The residue was chromatographed on silica (40% MTBE/hexanes) to give a yellow oil (85.5 mg, 28%). Purity based on HPLC: 95% (major peak 6.8 min., LiChrospher® Si60 column, 250 mm x 4.6 mm, 5 μm , DAD detector 254 nm, MTBE/hexane = 20:80; 1.0 mL/min); $[\alpha]_D^{23} = 32.7$ ($c = 1.2$, CHCl_3); IR (film) 2957, 2937, 1751, 1613 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 8.08–8.02 (m, 2H), 7.71 (dd, $J = 9.1, 2.2$ Hz, 1H), 7.63 (br s, 1H), 5.34 (dd, $J = 10.0, 3.2$ Hz, 1H), 5.29 (dd, $J = 11.6, 9.6$ Hz, 1H), 5.25 (dd, $J = 3.2, 1.6$ Hz, 1H), 4.82 (d, $J = 1.6$ Hz, 1H), 4.25 (dd, J

1
2
3 = 12.0, 5.2 Hz, 1H), 4.06 (dd, $J = 12.0, 2.4$ Hz, 1H), 3.97 (ddd, $J = 9.6, 5.2, 2.4$ Hz, 1H),
4
5 3.81 (dt, $J = 9.8, 6.4$ Hz, 1H), 3.57 (dt, $J = 9.8, 6.4$ Hz, 1H), 3.13 (t, $J = 7.6$ Hz, 2H),
6
7 2.27–2.16 (m, 2H), 2.15 (s, 3H), 2.04 (s, 3H), 2.02 (s, 3H), 1.99 (s, 3H); ^{13}C NMR (100
8
9 MHz, CDCl_3) δ 170.5, 170.0, 169.8, 169.7, 161.3, 147.1, 133.8 (q, $J_{\text{CF}} = 31.8$ Hz), 133.6,
10
11 131.3, 131.2, 123.1 (q, $J_{\text{CF}} = 273.2$ Hz), 122.9 (q, $J = 2.3$ Hz), 122.1, 119.3 (q, $J = 5.1$
12
13 Hz), 97.7, 69.6, 69.1, 68.6, 67.5, 66.2, 62.5, 35.2, 28.4, 20.8, 20.6, 20.6, 20.6; ^{19}F NMR
14
15 (376 MHz, CDCl_3) δ -61.7; HRMS (ESI-TOF) m/z : $[\text{M}+\text{Na}]^+$ Calcd for $\text{C}_{27}\text{H}_{29}\text{ClF}_3\text{NO}_{10}\text{Na}$
16
17 642.1330; Found: 642.1315.

21
22
23 **(3 β)-Cholest-5-en-3-yl 4-[6-chloro-4-(trifluoromethyl)quinolin-2-yl]butanoate (3r)**

24
25 was prepared according to the general procedure from *o*-TFMK **1a** (111.8 mg, 0.5
26
27 mmol), (3 β)-cholest-5-en-3-yl hex-5-ynoate (**2r**) (240.4 mg, 0.5 mmol), IPr^*CuCl (10.1
28
29 mg, 0.00005 mmol, 2 mol%), TMG (1.25 μL , 0.00005 mmol, 2 mol%), DME (100 μL)
30
31 and water (2 mL). Reaction mixture was heated at 100 $^\circ\text{C}$ for 16 h. Then reaction
32
33 mixture was extracted with EtOAc (3 x 10 mL), and the combined organic phases were
34
35 washed with brine (1 x 20 mL), dried over MgSO_4 and evaporated. The residue was
36
37 chromatographed on silica (10% MTBE/hexanes) to give a yellow oil (286.3 mg, 83%).
38
39 $[\alpha]_D^{23} = -15.9$ ($c = 1.5$, CHCl_3); Purity based on HPLC: 98% (major peak 8.6 min.,
40
41 LiChrospher $^{\text{®}}$ Si60 column, 250 mm x 4.6 mm, 5 μm , DAD detector 208 nm,
42
43 MTBE/hexane = 5:95; 1.0 mL/min); IR (film) 2948, 2868, 1732 cm^{-1} ; ^1H NMR (400 MHz,
44
45 CDCl_3) δ 8.11–8.05 (m, 2H), 7.71 (dd, $J = 9.0, 2.1$ Hz, 1H), 7.61 (s, 1H), 5.39–5.33 (m,
46
47 1H), 4.70–4.56 (m, 1H), 3.11–3.02 (m, 2H), 2.42 (t, $J = 7.3$ Hz, 2H), 2.35–2.27 (m, 2H),
48
49 2.24–2.12 (m, 2H), 2.06–1.76 (m, 5H), 1.65–1.08 (m, 21H) overlapping 1.01 (s, 3H),
50
51 0.91 (d, $J = 6.5$ Hz, 3H), 0.87 (d, $J = 1.7$ Hz, 3H), 0.87 (d, $J = 1.7$ Hz, 3H), 0.68 (s, 3H);
52
53
54
55
56
57
58
59
60

¹³C NMR (100 MHz, CDCl₃) δ 172.5, 161.5, 147.2, 139.6, 133.8 (q, *J*_{CF} = 31.7 Hz), 133.6, 131.4, 131.2, 123.2 (q, *J*_{CF} = 273.1 Hz), 122.9 (q, *J*_{CF} = 2.3 Hz), 122.7, 122.1, 119.3 (q, *J*_{CF} = 5.1 Hz), 74.1, 56.7, 56.2, 50.0, 42.3, 39.7, 39.5, 38.2, 38.0, 37.0, 36.6, 36.2, 35.8, 33.9, 31.9, 31.9, 28.2, 28.0, 27.8, 24.3, 24.3, 23.8, 22.8, 22.6, 21.0, 19.3, 18.7, 11.9; ¹⁹F NMR (376 MHz, CDCl₃) δ -61.6; HRMS (ESI-TOF) *m/z*: [M+H]⁺ Calcd for C₄₁H₅₆ClF₃NO₂ 686.3952; Found: 686.3958.

6-Chloro-2-ferrocenyl-4-(trifluoromethyl)quinoline (3x) was prepared according to the general procedure from *o*-TFMK **1a** (111.8 mg, 0.5 mmol), ethynylferrocene (**2x**) (126.0 mg, 0.6 mmol, 1.2 equiv), IPr*CuCl (10.1 mg, 0.00005 mmol, 2 mol%), TMG (1.25 μL, 0.00005 mmol, 2 mol%), DME (100 μL) and water (2 mL). The reaction mixture was heated at 100 °C for 16h. Then reaction mixture was extracted with EtOAc (3 x 10 mL), and the combined organic phases were washed with brine (1 x 20 mL), dried over MgSO₄ and evaporated. The residue was chromatographed on silica (5% Et₂O/hexanes) to give a red solid (138.2 mg, 67%). M.p. 183.2–184.6 (CHCl₃, by slow evaporation); IR (KBr) 3103, 3068, 1908, 1792, 1612, 1550, 1494 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 8.04 (d, *J* = 9.0 Hz, 1H), 8.03–8.01 (m, 1H), 7.78 (s, 1H), 7.68 (dd, *J* = 9.0, 2.2 Hz, 1H), 5.08 (t, *J* = 2.0 Hz, 2H), 4.55 (t, *J* = 2.0 Hz, 2H), 4.09 (s, 5H); ¹³C NMR (100 MHz, CDCl₃) δ 159.8, 147.6, 133.0 (q, *J*_{CF} = 32.0 Hz), 132.7, 131.3, 131.1., 123.3 (q, *J*_{CF} = 273.0 Hz), 123.1 (q, *J*_{CF} = 2.3 Hz), 121.7, 117.0 (q, *J*_{CF} = 5.3 Hz), 82.3, 71.2, 69.8, 68.1; ¹⁹F NMR (376 MHz, CDCl₃) δ -61.8; HRMS (ESI-TOF) *m/z*: [M+H]⁺ Calcd for C₂₀H₁₄ClF₃NFe 416.0116; Found: 416.0100.

(3*S*,4*R*)-3-[(1*R*)-1-[[*tert*-Butyl(dimethyl)silyl]oxy]ethyl]-4-[[6-chloro-4-(trifluoromethyl)quinolin-2-yl]methyl]azetidin-2-one (3y) was prepared according to

1
2
3 the general procedure from *o*-TFMK **1a** (111.8 mg, 0.5 mmol), (3*S*,4*R*)-3-*[(1*R*)-[tert-*
4 *butyl(dimethyl)silyl]oxo]ethyl-4-prop-2-yn-1yl*azetidine-2-one (**2y**) (160.5 mg, 0.6 mmol,
5 1.1 equiv), IPr*CuCl (10.1 mg, 0.00005 mmol, 2 mol%), TMG (1.25 μ L, 0.00005 mmol, 2
6 mol%), DME (100 μ L) and water (2 mL). The reaction mixture was heated at 100 °C for
7 16h. Then reaction mixture was extracted with EtOAc (3 x 3 mL), and the combined
8 organic phases were washed with brine (1 x 20 mL), dried over MgSO₄ and evaporated.
9 The residue was chromatographed on silica (20% EtOAc/toluene) to give a white solid
10 (123.1 mg, 52%). M.p. 129.4–130.3 °C (*n*-heptane); $[a]_D^{23} = 41.3$ (*c* = 1.01, CHCl₃); IR
11 (film) 3244, 2954, 2930, 2857, 1756, 1612 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 8.11–8.04
12 (m, 1H) overlapping 8.06 (d, *J* = 9.0 Hz, 1H), 7.75 (dd, *J* = 9.0, 2.2 Hz, 1H), 7.58 (s, 1H),
13 6.17 (br s, 1H), 4.27–4.18 (m, 2H), 3.41 (dd, *J* = 15.1, 4.4 Hz, 1H), 3.29 (dd, *J* = 15.1,
14 9.0 Hz, 1H), 3.03–2.97 (m, 1H), 1.15 (d, *J* = 6.2 Hz, 3H), 0.87 (s, 9H), 0.07 (s, 6H); ¹³C
15 NMR (100 MHz, CDCl₃) δ 169.0, 158.7, 147.2, 134.2 (q, *J*_{CF} = 31.9 Hz), 134.1, 131.5,
16 123.0 (q, *J*_{CF} = 273.2 Hz), 123.0 (q, *J*_{CF} = 2.3 Hz), 122.2, 119.4 (q, *J*_{CF} = 5.1 Hz), 65.6,
17 64.4, 49.6, 43.5, 25.7, 22.5, 17.9, -4.3, -5.0; ¹⁹F NMR (376 MHz, CDCl₃) δ -61.7; HRMS
18 (ESI-TOF) *m/z*: [M+H]⁺ Calcd for C₂₂H₂₉ClF₃N₂O₂Si 473.1639; Found: 473.1638.

19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42 **17-[6-Chloro-4-(trifluoromethyl)quinolin-2-yl]-3-methoxyestra-1(10),2,4,16-tetraene**
43 (**3z**) was prepared according to the general procedure from *o*-TFMK **1a** (72.1 mg, 0.322
44 mmol, 1.2 eqiuv), 17-ethynyl-3-methoxyestra-1(10),2,4,16-tetraene (**2z**) (78.6 mg, 0.269
45 mmol), IPr*CuCl (5.4 mg, 5.38 10⁻⁶ mmol, 2 mol%), TMG (0.67 μ L, 0.619 mg, 5.38 10⁻⁶
46 mmol, 2 mol%), DME (100 μ L) and water (1.1 mL). The reaction mixture was heated at
47 100 °C for 48 h. Then reaction mixture was extracted with EtOAc (3 x 10 mL), and the
48 combined organic phases were washed with brine (1 x 20 mL), dried over MgSO₄ and
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 evaporated. The residue was chromatographed on silica (5% MTBE/hexanes, the
4 unreacted *o*-TFMK was separated) to give a mixture of starting alkyne and quinoline
5 derivative **3z**. This mixture of was further separated by reversed-phase column
6 chromatography (MeOH) to give **3z** as a white solid (86.9 mg, 62%). M.p. 96.6–97.7
7 (crystals were obtained by slow evaporation of DCM solution); $[\alpha]_D^{23} = 49.9$ ($c = 0.72$,
8 CHCl_3); Purity based on HPLC: 99% (major peak 21.0 min., YMC-Pack Pro C18
9 column, 250 mm x 4.6 mm, 5 μm , UV detector 254 nm, MeOH; 1.0 mL/min); IR (KBr)
10 2923, 2851, 1610, 1550 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 8.08 (d, $J = 9.0$ Hz, 1H),
11 8.05–8.02 (m, 1H), 7.91 (s, 1H), 7.69 (dd, $J = 9.0, 2.2$ Hz, 1H), 7.25 (d, $J = 9.0$ Hz, 1H),
12 6.75 (dd, $J = 8.5, 2.7$ Hz, 1H), 6.69–6.64 (m, 2H), 3.80 (s, 3H), 3.04–2.86 (m, 3H), 2.47
13 (ddd, $J = 16.4, 6.3, 3.4$ Hz, 1H), 2.45–2.29 (m, 2H), 2.25 (ddd, $J = 16.4, 11.5, 1.2$ Hz,
14 1H), 2.04–1.95 (m, 1H), 1.91–1.40 (m, 5H), 1.23 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ
15 157.5, 155.1, 154.2, 147.1, 137.9, 135.3, 133.3, 133.0 (q, $J_{\text{CF}} = 31.5$ Hz), 132.9, 132.0,
16 130.9, 126.1, 123.3 (q, $J_{\text{CF}} = 273.0$ Hz), 122.9 (q, $J_{\text{CF}} = 2.3$ Hz), 121.8, 117.5 (q, $J_{\text{CF}} =$
17 5.2 Hz), 113.9, 111.5, 56.4, 55.2, 47.9, 44.3, 37.2, 35.4, 32.0, 29.7, 27.8, 26.7, 16.2; ^{19}F
18 NMR (376 MHz, CDCl_3) δ -61.8; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{29}\text{H}_{28}\text{ClF}_3\text{NO}$
19 498.1812; Found: 498.1816.

20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44 **2-Cyclopropyl-6-methoxy-4-(trifluoromethyl)quinoline (5a)** was prepared according
45 to the general procedure from 1-(2-amino-5-methoxyphenyl)-2,2,2-trifluoroethanone
46 (**1b**) (109.6 mg, 0.5 mmol), cyclopropylacetylene (**2a**) (51.0 μL , 0.6 mmol, 1.2 equiv),
47 IPr*CuCl (10.1 mg, 0.00005 mmol, 2 mol%), TMG (1.25 μL , 0.00005 mmol, 2 mol%),
48 DME (100 μL) and water (2 mL). Reaction mixture was heated at 120 $^\circ\text{C}$ for 16h. Then
49 reaction mixture was extracted with EtOAc (3 x 10 mL), and the combined organic
50
51
52
53
54
55
56
57
58
59
60

1
2
3 phases were washed with brine (1 x 20 mL), dried over MgSO₄ and evaporated. The
4
5 residue was chromatographed on silica (2% EtOAc/hexanes) to give a light-yellow oil
6
7 (89.8 mg, 67%). The reaction conducted on 0.5 mmol scale with IPr^{*OMe}CuCl (10.4 mg,
8
9 0.00005 mmol, 2 mol%) afforded product in 82% yield (109.4 mg, 82%). ¹H NMR (400
10
11 MHz, CDCl₃) δ 7.93 (d, *J* = 9.2 Hz, 1H), 7.47 (s, 1H), 7.37 (dd, *J* = 9.2, 2.7 Hz, 1H),
12
13 7.32–7.27 (m, 1H), 3.93 (s, 3H), 2.27–2.18 (m, 1H), 1.22–1.06 (m, 4H); ¹³C NMR (100
14
15 MHz, CDCl₃) δ 160.0, 157.8, 145.1, 132.6 (q, *J*_{CF} = 30.9 Hz), 131.0, 127.2 (q, *J*_{CF} =
16
17 277.1 Hz), 122.5, 117.2 (q, *J*_{CF} = 5.3 Hz), 102.0 (q, *J*_{CF} = 2.3 Hz), 55.5, 17.8, 10.3; ¹⁹F
18
19 NMR (376 MHz, CDCl₃) δ -62.3. ¹H NMR spectral data are in agreement with those
20
21 reported.⁶⁴ ¹³C NMR spectra are described by the authors without C-F coupling
22
23 constants and the accurate description is provided above. A copy of the spectra is
24
25 included in the ESI.

26
27
28
29
30
31
32 **2-Cyclopropyl-4-(trifluoromethyl)benzo[*h*]quinoline (5b)** was prepared according to
33
34 the general procedure from 1-(1-aminonaphthalene-2-yl)-2,2,2-trifluoroethanone (**1c**)
35
36 (119.6 mg, 0.5 mmol), cyclopropylacetylene (**2a**) (51.0 μL, 0.6 mmol, 1.2 equiv),
37
38 IPr^{*}CuCl (10.1 mg, 0.00005 mmol, 2 mol%), TMG (1.25 μL, 0.00005 mmol, 2 mol%)
39
40 and water (2 mL). Reaction mixture was heated at 120 °C for 48 h. Then reaction
41
42 mixture was extracted with EtOAc (3 x 10 mL), and the combined organic phases were
43
44 washed with brine (1 x 20 mL), dried over MgSO₄ and evaporated. The residue was
45
46 chromatographed on silica (5% EtOAc/hexanes) to give a colourless oil (58.1 mg, 40%).
47
48 IR (KBr) 3092, 3000, 1603, 1560 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 9.30–9.23 (m, 1H),
49
50 7.98–7.78 (m, 3H), 7.76–7.66 (m, 3H), 2.37–2.28 (m, 1H), 1.43–1.36 (m, 2H), 1.23–1.15
51
52 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 161.7, 147.2, 133.8 (q, *J*_{CF} = 31.0 Hz), 133.4,
53
54
55
56
57
58
59
60

1
2
3 131.1, 128.6, 127.9, 127.6, 127.1, 124.9, 123.8 (q, $J_{CF} = 273.1$ Hz), 120.8 (q, $J_{CF} = 2.5$
4 Hz), 119.5 (q, $J_{CF} = 1.4$ Hz), 117.3 (q, $J_{CF} = 5.4$ Hz), 18.1, 11.5; ^{19}F NMR (376 MHz,
5 CDCl_3) δ -60.9; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{17}\text{H}_{13}\text{F}_3\text{N}$ 288.1000; Found:
6 288.0992.
7
8
9

10
11
12 **2-Cyclopropyl-6-fluoro-4-(trifluoromethyl)quinoline (5c)** was prepared according to
13 the general procedure from 1-(2-amino-5-fluorophenyl)-2,2,2-trifluoroethanone (**1d**)
14 (103.4 mg, 0.5 mmol), cyclopropylacetylene (**2a**) (51.0 μL , 0.6 mmol, 1.2 equiv),
15 IPr*CuCl (10.1 mg, 0.00005 mmol, 2 mol%), TMG (1.25 μL , 0.00005 mmol, 2 mol%),
16 DME (100 μL) and water (2 mL). Reaction mixture was heated at 100 $^\circ\text{C}$ for 16 h. Then
17 reaction mixture was extracted with EtOAc (3 x 10 mL), and the combined organic
18 phases were washed with brine (1 x 20 mL), dried over MgSO_4 and evaporated. The
19 residue was chromatographed on silica (2% EtOAc/hexanes) to give a colourless oil
20 (97.5 mg, 76%). IR (KBr) 3094, 3013, 1931, 1627, 1566, 1514 cm^{-1} ; ^1H NMR (400 MHz,
21 CDCl_3) δ 8.02 (dd, $J = 9.2, 5.6$ Hz, 1H), 7.71–7.62 (m, 1H), 7.54 (br s, 1H), 7.52–7.44
22 (m, 1H), 2.30–2.19 (m, 1H), 1.28–1.04 (m, 4H); ^{13}C NMR (100 MHz, CDCl_3) δ 162.2 (d,
23 $J_{CF} = 2.8$ Hz), 160.4 (d, $J_{CF} = 247.0$ Hz), 146.1, 133.6 (dq, $J_{CF} = 31.5, 5.6$ Hz), 132.0 (d,
24 $J_{CF} = 9.2$ Hz), 123.4 (q, $J_{CF} = 272.8$ Hz), 121.9 (d, $J_{CF} = 9.4$ Hz), 120.1 (d, $J_{CF} = 25.4$
25 Hz), 118.0 (q, $J_{CF} = 5.2$ Hz), 107.9 (dq, $J_{CF} = 24.3, 2.2$ Hz), 18.0, 10.8; ^{19}F NMR (376
26 MHz, CDCl_3) δ -62.2, -111.9; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{13}\text{H}_{10}\text{F}_4\text{N}$
27 256.0749; Found: 256.0750.
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50

51 **2-Cyclopropyl-4-(trifluoromethyl)quinoline (5d)** was prepared according to the
52 general procedure from 1-(2-Aminophenyl)-2,2,2-trifluoroethanone (**1e**) (94.6 mg, 0.5
53 mmol), cyclopropylacetylene (**2a**) (51.0 μL , 0.6 mmol, 1.2 equiv), IPr*CuCl (10.1 mg,
54
55
56
57
58
59
60

0.00005 mmol, 2 mol%), TMG (1.25 μ L, 0.00005 mmol, 2 mol%), DME (100 μ L) and water (2 mL). Reaction mixture was heated at 120 $^{\circ}$ C for 16 h. Then reaction mixture was extracted with EtOAc (3 x 10 mL), and the combined organic phases were washed with brine (1 x 20 mL), dried over MgSO₄ and evaporated. The residue was chromatographed on silica (2% EtOAc/hexanes) to give a colourless oil (89.0 mg, 75%). ¹H NMR (400 MHz, CDCl₃) δ 8.10–8.01 (m, 2H), 7.75–7.68 (m, 1H), 7.57–7.50 (m, 2H), 2.32–2.25 (m, 1H), 1.28–1.10 (m, 4H); ¹⁹F NMR (376 MHz, CDCl₃) δ -61.6. Spectral data are in agreement with those reported.²¹

2-Cyclopropyl-*N,N*-dimethyl-4-(trifluoromethyl)quinolin-6-amine (5f) was prepared according to the general procedure from 1-[2-Amino-5-(dimethylamino)phenyl]-2,2,2-trifluoroethanone (**1g**) (116.1 mg, 0.5 mmol), cyclopropylacetylene (**2a**) (51.0 μ L, 0.6 mmol, 1.2 equiv), IPr^{*}CuCl (10.1 mg, 0.00005 mmol, 2 mol%), TMG (1.25 μ L, 0.00005 mmol, 2 mol%), DME (100 μ L) and water (2 mL). Reaction mixture was heated at 100 $^{\circ}$ C for 16 h. Then reaction mixture was extracted with EtOAc (4 x 2 mL), and the combined organic phases were washed with brine (1 x 20 mL), dried over Na₂SO₄ and evaporated. The residue was chromatographed on silica (10–30% EtOAc/hexanes) to give a yellow solid (103.6 mg, 74%). M.p. 75.0–75.4 (*n*-pentane); IR (KBr) 2923, 2852, 1620, 1609, 1508 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.90 (d, *J* = 9.4 Hz, 1H), 7.37 (s, 1H), 7.39–7.32 (dd, *J* = 9.4, 2.8 Hz, 1H), 7.02–6.97 (m, 1H), 3.08 (s, 6H), 2.25–2.15 (m, 1H), 1.17–1.09 (m, 2H), 1.09–1.02 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 157.6, 148.5, 143.0, 131.5 (q, *J*_{CF} = 30.5 Hz), 130.1, 124.1 (q, *J*_{CF} = 272.5 Hz), 123.0, 119.5, 116.9 (q, *J*_{CF} = 5.4 Hz), 101.2 (q, *J*_{CF} = 2.2 Hz), 40.5, 17.7, 9.7; ¹⁹F NMR (376 MHz, CDCl₃) δ -62.8; HRMS (ESI-TOF) *m/z*: [M+H]⁺ Calcd for C₁₅H₁₆F₃N₂ 281.1266; Found: 281.1268.

1
2
3 **2-Cyclopropyl-6-methyl-4-(trifluoromethyl)quinolone (5g)** was prepared according
4
5 to the general procedure from 1-(2-amino-5-methylphenyl)-2,2,2-trifluoroethanone (**1h**)
6
7 (101.6 mg, 0.5 mmol), cyclopropylacetylene (**2a**) (51.0 μ L, 0.6 mmol, 1.2 equiv),
8
9 IPr*CuCl (10.1 mg, 0.00005 mmol, 2 mol%), TMG (1.25 μ L, 0.00005 mmol, 2 mol%),
10
11 DME (100 μ L) and water (2 mL). Reaction mixture was heated at 120 °C for 16 h. Then
12
13 reaction mixture was extracted with EtOAc (3 x 10 mL), and combined organic phases
14
15 were washed with brine (1 x 20 mL), dried over MgSO₄ and evaporated. The residue
16
17 was chromatographed on silica (2% EtOAc/hexanes) to give a colourless oil which
18
19 solidified upon standing (102.3 mg, 81%). M.p. 64.3–64.8 (*n*-heptane, -78 °C); IR (KBr)
20
21 3092, 3006, 2922, 2867, 1611, 1561, 1506 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.93 (br
22
23 d, *J* = 8.6 Hz, 1H), 7.81 (br s, 1H), 7.55 (dd, *J* = 8.6, 1.6 Hz, 1H), 7.47 (s, 1H), 2.55 (s,
24
25 3H), 2.30–2.20 (m, 1H), 1.26–1.07 (m, 4H); ¹³C NMR (100 MHz, CDCl₃) δ 161.7, 147.5,
26
27 136.7, 133.3 (q, *J*_{CF} = 31.1 Hz), 132.1, 129.3, 123.7 (q, *J*_{CF} = 272.9 Hz), 122.7 (q, *J*_{CF} =
28
29 2.1 Hz), 121.4, 116.9 (q, *J*_{CF} = 5.3 Hz), 21.9, 18.0, 10.5; ¹⁹F NMR (376 MHz, CDCl₃) δ -
30
31 61.6; HRMS (ESI-TOF) *m/z*: [M+H]⁺ Calcd for C₁₄H₁₃F₃N 252.1000; Found: 252.1001.

32
33 **2-Cyclopropyl-6-[(1*R*,2*S*,5*R*)-5-methyl-2-(propan-2-yl)cyclohexyl]oxy-4-**
34
35 **(trifluoromethyl)quinoline (5h)** was prepared according to the general procedure from
36
37 1-(2-amino-5-[(1*R*,2*S*,5*R*)-5-methyl-2-(propan-2-yl)cyclohexyl]oxy)phenyl)-2,2,2-
38
39 trifluoroethanone (**1i**) (171.7 mg, 0.5 mmol), cyclopropylacetylene (**2a**) (51.0 μ L, 0.6
40
41 mmol, 1.2 equiv), IPr*CuCl (10.1 mg, 0.00005 mmol, 2 mol%), TMG (1.25 μ L, 0.00005
42
43 mmol, 2 mol%), DME (100 μ L) and water (2 mL). Reaction mixture was heated at 100
44
45 °C for 16 h. Then reaction mixture was extracted with EtOAc (4 x 2 mL), and the
46
47 combined organic phases were washed with brine (1 x 20 mL), dried over Na₂SO₄ and
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 evaporated. The residue was chromatographed on silica (30–50% EtOAc/hexanes) to
4 give a yellow oil (109.8 mg, 56%). IR (KBr) 2956, 2927, 2871, 1620, 1502 cm^{-1} ; ^1H NMR
5 (400 MHz, CDCl_3) δ 7.93 (d, $J = 9.1$ Hz, 1H), 7.45 (s, 1H), 7.40–7.32 (m, 2H), 4.18 (dd,
6 $J = 10.5, 10.5, 4.1$ Hz, 1H), 2.28–2.15 (m, 3H), 1.81–1.69 (m, 2H), 1.65–1.46 (m, 2H),
7 1.19–1.13 (m, 3H), 1.13–1.05 (m, 3H), 1.00–0.97 (m, 1H), 0.95 (d, $J = 7.0$ Hz, 3H), 0.94
8 (d, $J = 6.6$ Hz, 3H) 0.79 (d, $J = 7.0$ Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 159.8, 156.6,
9 144.9, 132.5 (q, $J_{\text{CF}} = 30.8$ Hz), 131.0, 123.8 (q, $J_{\text{CF}} = 272.9$ Hz), 123.1, 122.6, 117.1 (q,
10 $J_{\text{CF}} = 5.3$ Hz), 104.8 (q, $J_{\text{CF}} = 2.2$ Hz), 78.2, 48.1, 40.1, 34.5, 31.4, 26.2, 23.7, 22.1,
11 20.7, 17.8, 16.6, 10.2, 10.2; ^{19}F NMR (376 MHz, CDCl_3) δ -62.3; HRMS (ESI-TOF)
12 $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{23}\text{H}_{29}\text{F}_3\text{NO}$ 392.2201; Found: 392.2198.

13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28 **2-Cyclopropyl-4-(trifluoromethyl)-1,8-naphthyridine (5i)** was prepared according to
29 the general procedure 1-(2-aminopyridin-3-yl)-2,2,2-trifluoroethanone (**6a**) (95.0 mg, 0.5
30 mmol), cyclopropylacetylene (**2a**) (51.0 μL , 0.6 mmol, 1.2 equiv), $\text{IPr}^{\text{OMe}}\text{CuCl}$ (10.4 mg,
31 0.00005 mmol, 2 mol%), TMG (1.25 μL , 0.00005 mmol, 2 mol%), DME (100 μL) and
32 water (2 mL). Reaction mixture was heated at 120 $^\circ\text{C}$ for 16 h. Then reaction mixture
33 was extracted with EtOAc (4 x 2 mL), and the combined organic phases were washed
34 with brine (1 x 20 mL), dried over Na_2SO_4 and evaporated. The residue was
35 chromatographed on silica (30% EtOAc/hexanes) to give an orange solid (93.7 mg;
36 79%). M.p. 71.3–71.8 $^\circ\text{C}$ (*n*-pentane); IR (KBr) 3078, 3004, 1622, 1599, 1507 cm^{-1} ; ^1H
37 NMR (400 MHz, CDCl_3) δ 9.10 (dd, $J = 2.4$ Hz, $J = 1.7$ Hz, 1H), 8.45–8.38 (m, 1H), 7.68
38 (s, 1H), 7.49 (dd, $J = 8.4, 4.2$ Hz, 1H), 2.34–2.25 (m, 1H), 1.48–1.42 (m, 2H), 1.24–1.17
39 (m, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 167.7, 156.5, 153.9, 134.9 (q, $J_{\text{CF}} = 31.8$ Hz),
40 133.3 (q, $J_{\text{CF}} = 2.0$ Hz), 123.0 (q, $J_{\text{CF}} = 273.4$ Hz), 121.9, 119.0 (q, $J_{\text{CF}} = 5.0$ Hz), 116.2,
41 42 43 44 45 46 47 48 49 50 51 52 53 54 55 56 57 58 59 60

1
2
3 18.6, 12.5; ^{19}F NMR (376 MHz, CDCl_3) δ -61.0; HRMS (ESI-TOF) m/z : $[\text{M}+\text{Na}]^+$ Calcd
4 for $\text{C}_{12}\text{H}_9\text{F}_3\text{N}_2\text{Na}$ 261.0616; Found: 261.0614.
5
6
7

8
9 **6-Chloro-2-Cyclopropyl-4-(trifluoromethyl)-1,8-naphthydrine (5k)** was prepared
10 according to the general procedure from 1-(2-amino-5-chloropyridin-3-yl)-2,2,2-
11 trifluoroethanone (**6c**) (101.5 mg, 0.5 mmol), cyclopropylacetylene (**2a**) (51.0 μL , 0.6
12 mmol, 1.2 equiv), IPr^*CuCl (10.1 mg, 0.00005 mmol, 2 mol%), TMG (1.25 μL , 0.00005
13 mmol, 2 mol%) and water (2 mL). Reaction mixture was heated at 100 $^\circ\text{C}$ for 16 h. Then
14 reaction mixture was extracted with EtOAc (3 x 10 mL), and the combined organic
15 phases were washed with brine (1 x 20 mL), dried over MgSO_4 and evaporated. The
16 residue was chromatographed on silica (10% $\text{Et}_2\text{O}/n$ -pentane) to give a colourless oil
17 which solidified upon standing in the fridge (68.2 mg, 50%). M.p. 93.0–94.4 $^\circ\text{C}$ (crystals
18 were obtained by slow evaporation of DCM solution; attempts to crystallize quinoline **5k**
19 from n -heptane, n -pentane, or its mixture with EtOH, Et_2O has failed); IR (film) 3088,
20 3069, 3011, 1611, 1478 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 9.00 (d, J = 2.5 Hz, 1H),
21 8.37–8.32 (m, 1H), 7.70 (s, 1H), 2.34–2.23 (m, 1H), 1.48–1.38 (m, 2H), 1.27–1.17 (m,
22 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 168.0, 154.5, 153.2, 134.2 (q, J_{CF} = 32.1 Hz), 131.6
23 (q, J_{CF} = 2.2 Hz), 129.5, 122.7 (q, J_{CF} = 273.2 Hz), 119.8 (q, J_{CF} = 5.0 Hz), 116.4, 18.7,
24 12.7; ^{19}F NMR (376 MHz, CDCl_3) δ -61.2; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for
25 $\text{C}_{12}\text{H}_9\text{F}_3\text{N}_2\text{Cl}$ 273.0406; Found: 273.0407.
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48

49 **2-Cyclopropyl-4-(trifluoromethyl)-1,6-naphthyridine (5l)** was prepared according to
50 the general procedure from 1-(4-aminopyridin-3-yl)-2,2,2-trifluoroethanone (**6b**) (95.0
51 mg, 0.5 mmol), cyclopropylacetylene (**2a**) (51.0 μL , 0.6 mmol, 1.2 equiv), $\text{IPr}^{*\text{OMe}}\text{CuCl}$
52 (10.4 mg, 0.00005 mmol, 2 mol%), TMG (1.25 μL , 0.00005 mmol, 2 mol%), DME (100
53
54
55
56
57
58
59
60

1
2
3 μL) and water (2 mL). Reaction mixture was heated at 120 °C for 16 h. Then reaction
4
5
6 mixture was extracted with EtOAc (4 x 2 mL), and the combined organic phases were
7
8 washed with brine (1 x 20 mL), dried over Na_2SO_4 and evaporated. The residue was
9
10 chromatographed on silica (1.5–3% acetone/DCM) to give an orange solid (49.2 mg,
11
12 41%). M.p. 118.2–118.9 °C (*n*-pentane); IR (KBr) 3015, 1613, 1599 cm^{-1} ; ^1H NMR (400
13
14 MHz, CDCl_3) δ 9.47 (s, 1H), 8.78 (d, $J = 5.8$ Hz, 1H), 7.83 (d, $J = 5.8$ Hz, 1H), 7.62 (s,
15
16 1H), 2.33–2.26 (m, 1H), 1.41–1.29 (m, 2H), 1.29–1.17 (m, 2H); ^{13}C NMR (100 MHz,
17
18 CDCl_3) δ 168.6, 151.2, 148.9 (q, $J_{\text{CF}} = 3.0$ Hz), 147.6, 134.2 (q, $J_{\text{CF}} = 32.9$ Hz), 122.9 (q,
19
20 $J_{\text{CF}} = 273.3$ Hz), 122.0, 118.8 (q, $J_{\text{CF}} = 5.2$ Hz), 117.0, 18.8, 12.4; ^{19}F NMR (376 MHz,
21
22 CDCl_3) δ -60.7 ; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{12}\text{H}_{10}\text{F}_3\text{N}_2$ 239.0796; Found:
23
24 239.0794.
25
26
27
28
29

30 **6-Chloro-2-cyclopropyl-4-(heptafluoropropyl)quinolone (5m)** was obtained
31
32 according to the general procedure from ketone 1-(2-amino-5-chlorophenyl)-
33
34 2,2,3,3,4,4,4-heptafluorobutan-1-on (**1j**) (161.8 mg, 0.5 mmol), TMG (1.25 μL , 0.01
35
36 mmol, 2 mol%), cyclopropylacetylene **2a** (55.0 μL , 0.6 mmol, 1.2 equiv), IPr^*CuCl (10.2
37
38 mg, 0.01 mmol, 2 mol%), water (2 mL) and DME (0.1 mL). The reaction mixture was
39
40 heated at 120 °C for 16h. The crude product was chromatographed on silica (5%
41
42 $\text{Et}_2\text{O}/n$ -pentane) to give quinoline **5m** as a colourless oil (54.3 mg, 29%, 92% based on
43
44 recovered starting ketone) and unreacted substrate (72.3 mg). Additional experiments
45
46 were conducted with 2 mol% of $\text{IPr}^{*\text{OMe}}\text{CuCl}$ to afford product **5m** with 51% (94.6 mg,
47
48 after 16h at 120 °C) and 56% (103.6 mg, after 48h at 120 °C) and 5 mol% of
49
50 $\text{IPr}^{*\text{OMe}}\text{CuCl}$ (126.3 mg of **5m**, 68%). IR (film) 3011, 1604, 1557, 1496 cm^{-1} ; ^1H NMR
51
52 (400 MHz, CDCl_3) δ 8.07–8.01 (m, 1H), 7.96 (d, $J = 9.0$ Hz, 1H), 7.65 (dd, $J = 9.0, 2.2$
53
54
55
56
57
58
59
60

1
2
3 Hz, 1H), 7.50 (s, 1H), 2.30–2.20 (m, 1H), 1.28–1.11 (m, 4H); ^{13}C NMR (100 MHz,
4
5 CDCl_3) δ 163.0, 147.6, 132.8, 132.0 (t, $J_{\text{CF}} = 22.8$ Hz), 131.4, 130.8, 123.3–123.6 (m),
6
7 123.0, 121.0 (t, $J_{\text{CF}} = 8.6$ Hz), 18.0, 11.2. Signals in the region of 120–102 ppm has
8
9 been omitted in description of spectra for clarity due to complicated multiplicity; ^{19}F NMR
10
11 (376 MHz, CDCl_3) δ -77.8 (t, $J = 10.2$ Hz), -107.8 — -108.0 (m), -124.7 – -124.8 (m);
12
13 HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{15}\text{H}_{10}\text{ClF}_7\text{N}$ 372.0390; Found: 372.0380.
14
15

16
17 **6-Chloro-2-cyclopropyl-4-(difluoromethyl)quinolone (5n)** was prepared according to
18
19 the general procedure from 1-(2-amino-5-chlorophenyl)-2,2-difluoroethanone (**1k**)
20
21 (102.8 mg, 0.5 mmol), cyclopropylacetylene (**2a**) (51.0 μL , 0.6 mmol, 1.2 equiv),
22
23 IPr^*CuCl (10.1 mg, 0.00005 mmol, 2 mol%), TMG (1.25 μL , 0.00005 mmol, 2 mol%),
24
25 water (2 mL) and DME (0.1 mL). Reaction mixture was heated at 100 °C for 16 h. Then
26
27 reaction mixture was extracted with EtOAc (3 x 10 mL), and combined organic phases
28
29 were washed with brine (1 x 20 mL), dried over MgSO_4 and evaporated. The residue
30
31 was chromatographed on silica (5% EtOAc/hexanes) to give a colourless oil (68.7 mg,
32
33 54%). Additional experiment was conducted on the same scale at 100 °C for 48 h to
34
35 give a quinoline **5n** with 73% (92.6 mg). M.p. 63.7–64.1; IR (film) 3090, 3013, 2972,
36
37 1743, 1613 1557 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 8.00–7.91 (m, 2H), 7.63 (dd, $J =$
38
39 9.0, 2.2 Hz, 1H), 7.39 (br s, 1H), 7.02 (t, $J_{\text{HF}} = 54.5$ Hz, 1H), 2.29–2.19 (m, 1H), 1.27–
40
41 1.08 (m, 4H); ^{13}C NMR (100 MHz, CDCl_3) δ 163.6, 147.0, 136.7 (t, $J_{\text{CF}} = 21.6$ Hz),
42
43 132.1, 131.1, 130.6, 123.1 (t, $J_{\text{CF}} = 2.8$ Hz), 122.3, 117.9 (t, $J_{\text{CF}} = 7.7$ Hz), 113.4 (t, $J_{\text{CF}} =$
44
45 239.2 Hz), 18.1, 10.7; ^{19}F NMR (376 MHz, CDCl_3) δ -114.9; HRMS (EI) m/z : $[\text{M}^{++}]$ Calcd
46
47 for $\text{C}_{13}\text{H}_{10}\text{F}_2\text{NCl}$ 253.0470; Found: 253.0491.
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 **3-[6-Chloro-4-(difluoromethyl)quinolin-2-yl]propan-1-ol (5o)** was prepared
4
5 according to the general procedure from 1-(2-Amino-5-chlorophenyl)-2,2-
6 difluoroethanone (**1k**) (102.8 mg, 0.5 mmol), pent-4-yn-1-ol (**2l**) (56.0 μ L, 0.6 mmol, 1.2
7 equiv), IPr*CuCl (10.1 mg, 0.00005 mmol, 2 mol%), TMG (1.25 μ L, 0.00005 mmol, 2
8 mol%), water (2 mL) and DME (0.1 mL). Reaction mixture was heated at 100 °C for 16
9 h. Then reaction mixture was extracted with EtOAc (3 x 10 mL), and combined organic
10 phases were washed with brine (1 x 20 mL), dried over MgSO₄ and evaporated. The
11 residue was chromatographed on silica (50–75% EtOAc/hexanes) to give a light-yellow
12 oil which solidified upon standing (96.4 mg, 71%). The experiment was conducted on
13 the same scale at 100 °C for 48 h, affording product **5o** with 77% (104.1 mg) yield. M.p.
14 75.5–76.8 (*n*-pentane); IR (KBr) 3252, 2950, 2925, 2858, 1620, 1567 cm⁻¹; ¹H NMR
15 (400 MHz, CDCl₃) δ 8.06–7.99 (m, 2H), 7.70 (dd, *J* = 9.0, 2.3 Hz, 1H), 7.52 (br s, 1H),
16 7.05 (t, *J*_{HF} = 54.4 Hz, 1H), 3.77 (t, *J* = 7.7 Hz, 2H), 3.23 (br s, 1H, OH) overlapping 3.19
17 (t, *J* = 7.0 Hz, 2H), 2.18–2.06 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 162.2, 146.4, 137.6
18 (t, *J*_{CF} = 21.8 Hz), 133.2, 131.0, 131.0, 123.2 (t, *J*_{CF} = 2.7 Hz), 122.4 (t, *J*_{CF} = 1.4 Hz),
19 119.6 (t, *J*_{CF} = 7.6 Hz), 113.2 (t, *J*_{CF} = 239.5 Hz), 62.3, 36.0, 31.1; ¹⁹F NMR (376 MHz,
20 CDCl₃) δ -115.0; HRMS (ESI-TOF) *m/z*: [M+H]⁺ Calcd for C₁₃H₁₃F₂NOCl 272.0654;
21 Found: 272.0646.
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45

46 **3(β)-Cholest-5-en-3-yl [4-(6-Chloro-4-(difluoromethyl)quinolin-2-yl]butanoate (5p)**
47 was prepared according to the general procedure from 1-(2-amino-5-chlorophenyl)-2,2-
48 difluoroethanone (**1k**) (102.8 mg, 0.5 mmol), (3 β)-cholest-4-en-3-yl hex-5-ynoate (**2r**)
49 (280.1 mg, 0.58 mmol, 1.2 equiv), IPr*CuCl (10.1 mg, 0.00005 mmol, 2 mol%), TMG
50 (1.25 μ L, 0.00005 mmol, 2 mol%), water (2 mL) and DME (0.1 mL). Reaction mixture
51
52
53
54
55
56
57
58
59
60

1
2
3 was heated at 100 °C for 16 h. Then reaction mixture was extracted with EtOAc (3 x 10
4 mL), and combined organic phases were washed with brine (1 x 20 mL), dried over
5
6 MgSO₄ and evaporated. The residue was chromatographed on silica (10%
7
8 MTBE/hexanes). Thus obtained mixture of steroidal alkyne and product **5p** was
9
10 subjected to chromatography on RP-18 silica (MeOH) to give a white solid (173.9 mg,
11
12 52%). $[\alpha]_D^{23} = -13.1$ (c = 1.32, MeOH/CHCl₃ 1/1); M.p. 110.5–111.7 (attempts to
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
crystalize quinoline **5p** failed; compound does not dissolved in common organic
solvents – DCM, EtOH, MeOH, DMSO, DMF, hexanes, MeCN, MTBE, Et₂O; slightly
soluble in CHCl₃ – ca. 2 mg/5 mL); IR (film) 3439, 2947, 2868, 1729, 1611 cm⁻¹; ¹H
NMR (400 MHz, CDCl₃) δ 8.05 (d, *J* = 9.1 Hz, 1H), 8.04–8.01 (m, 1H), 7.69 (dd, *J* = 9.1,
2.2 Hz, 1H), 7.49 (s, 1H), 7.05 (t, *J* = 54.4 Hz, 1H), 5.39–5.34 (m, 1H), 4.69–4.55 (m,
1H), 3.05 (t, *J* = 7.5 Hz, 2H), 2.41 (t, *J* = 7.4 Hz, 2H), 2.34–2.26 (m, 2H), 2.23–2.13 (m,
2H), 2.05–1.92 (m, 5H), 1.90–1.77 (m, 5H), 1.62–0.86 (m, 16H) overlapping 1.01 (s,
3H), 0.92 (d, *J* = 6.5 Hz, 3H), 0.87 (d, *J* = 6.6 Hz, 3H), 0.86 (d, *J* = 6.6 Hz, 3H), 0.68 (s,
3H); ¹³C NMR (100 MHz, CDCl₃) δ 172.6, 161.8, 146.9, 139.6, 137.3 (t, *J*_{CF} = 22.0 Hz),
131.4, 130.8, 123.3 (t, *J*_{CF} = 2.5 Hz), 122.7, 122.4, 119.4 (t, *J*_{CF} = 7.6 Hz), 113.3 (t, *J*_{CF} =
239.4 Hz), 74.1, 56.7, 56.2, 50.1, 42.3, 39.8, 39.5, 38.2, 38.1, 37.0, 36.6, 36.2, 35.8,
34.0, 31.9, 31.9, 29.7, 28.2, 28.0, 27.8, 24.5, 24.3, 23.8, 22.8, 22.6, 21.0, 19.3, 18.7,
11.7; ¹⁹F NMR (376 MHz, CDCl₃) δ -114.8; HRMS (ESI-TOF) *m/z*: [M+H]⁺ Calcd for
C₄₁H₅₇F₂NO₂Cl 668.4046; Found: 668.4043.

Synthesis of dibenzo[*b,f*][1,5]diazocines

2,8-Dichloro-6,12-bis(trifluoromethyl)-5,6,11,12-tetrahydro-6,12-

epoxydibenzo[*b,f*][1,5]diazocine (4a) 1-(2-Amino-5-chlorophenyl)-2,2,2-

1
2
3 trifluoroethanone (**1a**) (223.6 mg, 1.0 mmol), TMG (25.1 μ L, 0.2 mmol, 20 mol%) and
4
5 water (2 mL) were placed in a screw-cap 4 mL vial and heated at 100 °C (temp. of oil
6
7 bath) for 16 h. Then reaction mixture was extracted with EtOAc (3 x 10 mL). The
8
9 combined organic extracts were dried over MgSO₄, and evaporated. The residue was
10
11 chromatographed on silica (5% EtOAc/hexanes). Compound **4a** was isolated as a
12
13 yellow oil (141.2 mg, 66%) which spontaneously solidified upon standing in the fridge.
14
15 M.p. 217–220 °C (EtOAc; by slow evaporation); IR (film) 3379, 3350, 1610 cm⁻¹; ¹H
16
17 NMR (600 MHz, CDCl₃) δ 7.43 (s, 2H), 7.22 (dd, *J* = 8.6, 2.3 Hz, 2H), 6.81 (d, *J* = 8.6
18
19 Hz, 2H), 4.90 (br s, NH, 2H); ¹³C NMR (150 MHz, CDCl₃) δ 138.4, 130.9, 127.6, 125.6
20
21 (q, *J*_{C-F} = 32.0 Hz), 122.4 (q, *J*_{C-F} = 282.5 Hz), 121.4, 120.8, 82.9 (q, *J* = 32.3 Hz); ¹⁹F
22
23 NMR (470 MHz, CDCl₃) δ -79.1. HRMS (ESI-TOF) *m/z*: [M-H]⁻ Calcd for C₁₆H₇Cl₂F₆N₂O
24
25 426.9840; Found: 426.9845. Spectral data are in agreement with those reported.³¹
26
27
28
29
30
31

32 **2,8-Fluoro-6,12-bis(trifluoromethyl)-5,6,11,12-tetrahydro-6,12-**

33 **epoxydibenzo[*b,f*][1,5]diazocine** (**4b**) 1-(2-Amino-5-fluorophenyl)-2,2,2-
34
35 trifluoroethanone (**1d**) (207.1 mg, 1.0 mmol), Et₃N (28 μ L, 0.2 mmol), TMG (25.1 μ L, 0.2
36
37 mmol, 20 mol%) and water (2 mL) were placed in a screw-cap 4 mL vial and heated at
38
39 100 °C (temp. of oil bath) for 16 h. Then reaction mixture was diluted with water (10 mL)
40
41 and extracted with EtOAc (3 x 10 mL). The combined organic extracts were dried over
42
43 MgSO₄ and evaporated. The residue was chromatographed on silica (5%
44
45 EtOAc/hexanes). Compound **4b** was isolated as a yellow oil (154.7 mg, 78%) which
46
47 spontaneously solidified upon standing in the fridge. M.p. 169.1–169.5 °C (DCM; by
48
49 slow evaporation); IR (film) 3426, 3370, 3328, 1891, 1725, 1499 cm⁻¹; ¹H NMR (400
50
51 MHz, CDCl₃) δ 7.23–7.17 (m, 2H), (ddd, *J* = 8.9, 7.9, 2.8 Hz, 2H), 6.86 (dd, *J* = 8.9, 4.9
52
53
54
55
56
57
58
59
60

1
2
3 Hz, 2H), 4.77 (s, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 158.2 (d, $J_{\text{CF}} = 241.1$ Hz), 135.9 (d,
4
5 $J_{\text{CF}} = 2.4$ Hz), 122.5 (q, $J_{\text{CF}} = 282.3$ Hz), 121.7 (d, $J_{\text{CF}} = 7.8$ Hz), 121.6 (d, $J_{\text{CF}} = 7.2$ Hz),
6
7 118.2 (d, $J_{\text{CF}} = 23.0$ Hz), 112.6 (dq, $J_{\text{CF}} = 24.5, 3.2$ Hz), 83.4 (dq, $J_{\text{CF}} = 32.0, 2.1$ Hz);
8
9 ^{19}F NMR (376 MHz, CDCl_3) δ -79.3, -118.8; HRMS (ESI-TOF) m/z : $[\text{M}-\text{H}]^-$ Calcd for
10
11 $\text{C}_{16}\text{H}_7\text{F}_8\text{N}_2\text{O}$ 395.0431; Found: 395.0439.
12
13
14

15 16 **Synthesis of fluorinated analogue of G protein-coupled receptor antagonist** 17 18 **(GPR91)**

19
20 **Ethyl 4-[4-(trifluoromethyl)-1,8-naphthyridin-2-yl]butanoate (5j)** was prepared
21
22 according to the general procedure 1-(2-aminopyridin-3-yl)-2,2,2-trifluoroethanone (**6a**)
23
24 (1.33 g, 7.0 mmol), ethyl hex-5-ynoate (**7**) (1.18 g, 8.4 mmol, 1.2 equiv), $\text{IPr}^{\text{OM}}\text{CuCl}$
25
26 (145.6 mg, 0.14 mmol, 2 mol%), TMG (17.5 μL , 0.14 mmol, 2 mol%), water (28 mL) and
27
28 DME (1.4 mL). Reaction mixture was heated at 120 $^\circ\text{C}$ for 20 h. Then reaction mixture
29
30 was extracted with EtOAc (4 x 20 mL), and combined organic phases were washed with
31
32 brine (1 x 20 mL), dried over Na_2SO_4 and evaporated. The residue was
33
34 chromatographed on silica (30–50% EtOAc/hexanes) to give an orange oil (1.61 g,
35
36 74%). IR (film) 2981, 2938, 1732, 1621, 1601 1555, 1504 cm^{-1} ; ^1H NMR (400 MHz,
37
38 CDCl_3) δ 9.20–9.10 (m, 1H), 8.44 (d, $J = 8.3$ Hz, 1H), 7.66 (s, 1H), 7.58–7.52 (dd, $J =$
39
40 8.4, 4.2 Hz, 1H), 4.10 (q, $J = 7.1$ Hz, 2H), 3.14 (t, $J = 7.4$ Hz, 2H), 2.43 (t, $J = 7.2$ Hz,
41
42 2H), 2.31–2.19 (m, 2H), 1.22 (t, $J = 7.2$ Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 173.0,
43
44 165.2, 156.2, 154.1, 135.6 (q, $J_{\text{CF}} = 31.9$ Hz), 133.2 (q, $J_{\text{CF}} = 2.0$ Hz), 122.9 (q, $J_{\text{CF}} =$
45
46 273.3 Hz), 122.6, 119.5 (q, $J_{\text{CF}} = 4.9$ Hz), 116.4, 60.3, 38.2, 33.5, 23.9, 14.1; ^{19}F NMR
47
48 (376 MHz, CDCl_3) δ -60.9; HRMS (ESI-TOF) m/z : $[\text{M}+\text{Na}]^+$ Calcd for $\text{C}_{15}\text{H}_{15}\text{F}_3\text{N}_2\text{O}_2\text{Na}$
49
50 335.0983; Found: 335.0976.
51
52
53
54
55
56
57
58
59
60

1
2
3 **Methyl 4'-cyano-3,3'-difluorobiphenyl-2-carboxylate (11)** To a solution of nitrile **8**
4 (4.0 g, 20.0 mmol) in anhydrous DME (25 mL), cooled to -20 °C, commercially available
5 solution of *i*-PrMgCl•LiCl (14.6 mL, 19.0 mmol, 1.3M in THF) was added (the colour of
6 the reaction mixture changed from light-yellow to deep orange). After 15 min, a freshly
7 prepared solution of CuCN•2LiCl in THF (by mixing of CuCN - 1.62 g, 18.0 mmol and
8 LiCl - 1.53 g, 36 mmol in THF – 25 mL) was added dropwise (the colour of the reaction
9 mixture changed from deep orange to green) and stirred for additional 15 min at -20 °C.
10 In a separate Schlenk tube, a solution of iodoester **10** (1.82 g, 6.5 mmol) and Fe(acac)₃
11 (230.0 mg, 0.65 mmol, 10 mol%) in anhydrous DME (15 mL) was prepared and slowly
12 added to the organocopper compound **9**. The resulting mixture was stirred at 80 °C
13 (temp. of oil bath) for 20 h. Then reaction mixture was quenched with a saturated
14 aqueous solution of NH₄Cl (30 mL) and extracted with EtOAc (4 x 50 mL). The
15 combined organic extracts were washed with brine (1 x 100 mL), dried over Na₂SO₄,
16 and evaporated. The residue was chromatographed on silica (15–30% MTBE/hexanes)
17 giving a yellow solid (1.45 g, 81%). ¹H NMR (400 MHz, CDCl₃) δ 7.72–7.65 (m, 1H),
18 7.58–7.50 (m, 1H), 7.30–7.14 (m, 4H), 3.76 (s, 3H); CAS registry number 1026635-99-
19 3; no spectroscopic data are available for comparison.
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43

44 **[3,3'-Difluoro-2'-(methoxycarbonyl)biphenyl-4-yl]methanaminium chloride (12)** To
45 a solution of nitrile **11** (1.43 g, 5.23 mmol) in anhydrous MeOH (100 mL), a solution of
46 HCl in MeOH (6.25 mL, 1.5 equiv, 1.25 M in MeOH) and 10% Pd/C (278 mg, 0.26
47 mmol, 5 mol%) were added. The flask was evacuated (water aspirator) and back-filled
48 with hydrogen (3 times) and the reaction mixture was vigorously stirred at rt under
49 atmosphere of hydrogen (balloon). After 16h, argon was purged through the reaction
50
51
52
53
54
55
56
57
58
59
60

1
2
3 mixture for 10 min. and then it was filtered through a pad of celite (washing with MeOH).
4
5 Then, the solvent was evaporated and the residue was treated with MTBE (50 mL),
6
7 heated to reflux for 15 min. and filtered while hot. The obtained solid was washed with
8
9 MTBE (2 x 10 mL) and dried in vacuo to afford hydrochloride **12** as an off-white solid
10
11 (1.13 g, 69%). M.p. 210.1–210.5 °C; IR (KBr) 2970, 2951, 2909, 2875, 2707, 2627,
12
13 2021, 1741, 1627, 1609, 1568, 1508 cm⁻¹; ¹H NMR (400 MHz, DMSO-*d*₆) δ 8.72 (br s,
14
15 3H), 7.79–7.58 (m, 2H), 7.48–7.16 (m, 4H), 4.09 (br s, 2H), 3.70 (s, 3H); ¹³C NMR (100
16
17 MHz, DMSO-*d*₆) δ 165.1, 160.0 (d, *J*_{CF} = 246.3 Hz), 158.8 (d, *J*_{CF} = 247.1 Hz), 141.0
18
19 (dd, *J*_{CF} = 8.0, 2.0 Hz), 139.6, 132.4 (d, *J*_{CF} = 9.1 Hz), 131.6 (d, *J*_{CF} = 3.7 Hz), 125.9 (d,
20
21 *J*_{CF} = 2.8 Hz), 124.1 (d, *J*_{CF} = 3.0 Hz), 121.0 (d, *J*_{CF} = 32.5 Hz), 120.8 (d, *J*_{CF} = 35.1 Hz),
22
23 115.5 (d, *J*_{CF} = 21.2 Hz), 114.9 (d, *J*_{CF} = 22.7 Hz), 52.7, 35.3 (d, *J*_{CF} = 4.0 Hz); ¹⁹F NMR
24
25 (376 MHz, DMSO-*d*₆) δ -115.7, -116.3; Anal. Calcd for C₁₅H₁₄ClF₂NO₂•1/2H₂O: C,
26
27 55.82; H, 4.68; N, 4.34 Found: C, 55.84; H, 4.74; N, 4.67; HRMS (ESI-TOF) *m/z*: [M-Cl]⁺
28
29 Calcd for C₁₅H₁₄F₂NO₂: 278.0993; Found: 278.0984.
30
31
32
33
34
35
36

37
38 **Lithium 4-[4-(trifluoromethyl)-1,8-naphthyridin-2-yl]butanoate** To a solution of ester
39
40 **5j** (1.0 g, 3.2 mmol) in a mixture of THF (50 mL) and water (2 mL), LiOH•H₂O (268.5
41
42 mg, 6.4 mmol, 2.0 equiv) was added at rt and stirred for 16 h. Then, the solvent was
43
44 evaporated and the residue was dried in vacuo for 16 h to give a yellow solid (928.5
45
46 mg). The resulting lithium salt was used in the next step without further purification and
47
48 can be stored for a long time without decomposition under argon atmosphere.
49
50

51
52 **Methyl 3,3'-difluoro-4'-[4-[4-(trifluoromethyl)-1,8-naphthyridin-2-**
53
54 **yl]butanoyl]amino)methyl]biphenyl-2-carboxylate (13)** To a solution of lithium 4-[4-
55
56 (trifluoromethyl)-1,8-naphthyridin-2-yl]butanoate (863.0 mg, 2.97 mmol) in anhydrous
57
58
59
60

1
2
3 DMF (60 mL), HOBt (683.3 mg, 4.46 mmol, 1.5 equiv) was added and stirred for 15 min.
4
5 (until all of the solid has dissolved). Then, amine hydrochloride **12** (977.4 mg, 3.12
6
7 mmol, 1.05 eq) was added and stirred at rt. After 15 min., the reaction mixture was
8
9 cooled to 0 °C and EDC•HCl (854.0 mg, 4.46 mmol, 1.5 equiv) was added in one
10
11 portions (colour of the reaction mixture changed from yellow to light-green). After
12
13 additional 15 min. DIPEA (1.54 mL, 8.91 mmol, 3.0 equiv) was added, the cooling bath
14
15 was removed and the resulting reaction mixture was stirred at rt for 16 h. The reaction
16
17 mixture was cooled to 0 °C and water was slowly added (50 mL). The resulting mixture
18
19 was diluted with EtOAc (30 mL), and the organic phase was separated, washed with
20
21 water (2 x 75 mL), brine (7 x 50 mL), dried over Na₂SO₄ and the solvents were
22
23 evaporated. The residue was chromatographed on silica (1–3% MeOH/DCM) to give an
24
25 off-white gummy solid (1.013 g, 63%, after 2 steps). IR (film) 3291, 3060, 2952, 1736,
26
27 1657, 1555 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 9.05 (br d, *J* = 2.6 Hz, 1H), 8.42 (d, *J* =
28
29 8.3 Hz, 1H), 7.68 (s, 1H), 7.53 (dd, *J* = 8.4, 4.2 Hz, 1H), 7.43–7.32 (m, 2H), 7.12–7.96
30
31 (m, 4H), 6.89–6.82 (m, 1H), 4.50 (d, *J* = 5.7 Hz, 2H), 3.66 (s, 3H), 3.14 (t, *J* = 7.3 Hz,
32
33 2H), 2.42 (t, *J* = 7.1 Hz, 2H), 2.32–2.21 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 172.5,
34
35 165.8, 165.4, 160.5 (d, *J*_{CF} = 245.8 Hz), 159.6 (d, *J*_{CF} = 250.0 Hz), 155.9, 153.9, 140.7
36
37 (dd, *J*_{CF} = 2.0, 2.0 Hz), 140.3 (dd, *J*_{CF} = 8.1, 2.2 Hz), 135.6 (q, *J*_{CF} = 31.2 Hz), 133.3,
38
39 131.3 (d, *J*_{CF} = 8.9 Hz), 130.1 (d, *J*_{CF} = 4.8 Hz), 126.3 (d, *J*_{CF} = 1.3 Hz), 125.2 (d, *J*_{CF} =
40
41 10.3 Hz), 124.0 (d, *J*_{CF} = 3.2 Hz), 122.6, 122.8 (q, *J*_{CF} = 273.5 Hz), 121.3 (d, *J*_{CF} = 17.0
42
43 Hz), 119.7 (q, *J*_{CF} = 4.9 Hz), 116.4, 115.2 (d, *J*_{CF} = 4.7 Hz), 114.9 (dd, *J*_{CF} = 3.5 Hz),
44
45 52.4, 38.0, 37.0 (d, *J*_{CF} = 3.7 Hz), 35.3, 24.5; ¹⁹F NMR (376 MHz, CDCl₃) δ -60.9, -
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 114.8, -118.5; HRMS (EI) m/z : $[M+Na]^+$ Calcd for $C_{28}H_{22}F_5N_3O_3Na$ 566.1479, Found:
4
5 566.1477.
6
7

8 **Dimethyl 4',4''-(iminodimethanediyl)bis(3,3'-difluorobiphenyl-2-carboxylate) (19)**
9

10 To a solution of nitrile **11** (1.25 g, 4.57 mmol) in MeOH (15 mL), ammonia (5.5 mL) and
11
12 freshly prepared Raney Nickel (2.5 g) were added and vigorously stirred for 16h at rt.
13
14

15 Then reaction mixture was filtered through pad of celite (washing with MeOH), solvent
16
17 was evaporated and the residue was chromatographed on silica (40-50%
18
19 EtOAc/hexanes) to give a colourless oil (432.8 mg, 35 %). IR (film) 3348, 3003, 2952,
20
21 2845, 1736, 1612, 1567, 1511 cm^{-1} ; 1H NMR (400 MHz, $CDCl_3$) δ 7.48–7.40 (m, 4H),
22
23 7.20–7.04 (m, 8H), 3.92 (s, 4H), 3.72 (s, 6H); ^{13}C NMR (100 MHz, $CDCl_3$) δ 165.8,
24
25 160.9 (d, J_{CF} = 245.2 Hz), 159.6 (d, J_{CF} = 250.0 Hz), 140.9 (dd, J_{CF} = 2.3, 2.3 Hz), 140.0
26
27 (d, J_{CF} = 8.2, 2.3 Hz), 131.3 (d, J_{CF} = 8.9 Hz), 130.3 (d, J_{CF} = 5.3 Hz), 126.6 (d, J_{CF} =
28
29 14.9 Hz), 125.3 (d, J_{CF} = 3.1 Hz), 123.9 (d, J_{CF} = 31.8 Hz), 115.1 (d, J_{CF} = 23.0 Hz),
30
31 114.9 (d, J_{CF} = 9.1 Hz), 52.4, 46.3 (d, J_{CF} = 2.7 Hz); ^{19}F NMR (376 MHz, $CDCl_3$) δ -
32
33 114.8, -118.8; HRMS (ESI-TOF) m/z : $[M+H]^+$ Calcd for $C_{30}H_{24}NO_4F_4$ 538.1641; Found:
34
35 538.1636.
36
37
38
39
40
41
42
43
44

45 **Associated content**
46
47

48 **Supporting Information Available**
49
50

51 Detailed experimental procedures for the synthesis of substrates, copies of 1H , ^{13}C and
52
53 ^{19}F NMR spectra for all compounds and HPLC data for selected compounds. This
54
55 material is available free of charge via the Internet at <http://pubs.acs.org>.
56
57
58
59
60

Author Contributions

The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

ORCID

Michał Michalak: 0000-0002-1193-1551

Paweł Czerwiński: 0000-0003-0022-3813

Notes

The author declare no competing financial interest.

Acknowledgment

This research was supported by Polish Ministry of Science and Higher Education (grant Iuventus Plus IP2012 064172). We are grateful to Dr. Wioletta Kośnik for a helpful and critical discussion during the preparation of the manuscript.

References

- (1) (a) Wang, J.; Sánchez-Roselló, M.; Aceña, J. L.; del Pozo, C.; Sorochinsky, A. E.; Fustero, S.; Soloshonok, V. A.; Liu, H. *Chem. Rev.* **2014**, *114*, 2432-2506. (b) O'Hagan, D. *J. Fluorine Chem.* **2010**, *131*, 1071-1081.
- (2) Purser, S.; Moore, P. R.; Swallow, S.; Gouverneur, V. *Chem. Soc. Rev.* **2008**, *37*, 320-330.
- (3) (a) Wang, J.; Sánchez-Roselló, M.; Aceña, J. L.; del Pozo, C.; Sorochinsky, A. E.; Fustero, S.; Soloshonok, V. A.; Liu, H. *Chem. Rev.* **2013**, *114*, 2432-2506. (b)

- 1
2
3 Zhou, Y.; Wang, J.; Gu, Z.; Wang, S.; Zhu, W.; Aceña, J. L.; Soloshonok, V. A.;
4
5 Izawa, K.; Liu, H. *Chem. Rev.* **2016**, *116*, 422-518.
6
7
8 (4) (a) Böhm, H.-J.; Banner, D.; Bendels, S.; Kansy, M.; Kuhn, B.; Müller, K.; Obst-
9
10 Sander, U.; Stahl, M. *ChemBioChem* **2004**, *5*, 637-643. (b) Kirk, K. L. *J. Fluorine*
11
12 *Chem.* **2006**, *127*, 1013-1029.
13
14
15 (5) (a) Wu, G.; Wong, Y.; Chen, X.; Ding, Z. *J. Org. Chem.* **1999**, *64*, 3714-3718. (b)
16
17 Earl, J.; Kirkpatrick, P. *Nat Rev Drug Discov* **2003**, *2*, 97-98.
18
19
20 (6) (a) Pierce, M. E.; Parsons, R. L.; Radesca, L. A.; Lo, Y. S.; Silverman, S.; Moore,
21
22 J. R.; Islam, Q.; Choudhury, A.; Fortunak, J. M. D.; Nguyen, D.; Luo, C.; Morgan,
23
24 S. J.; Davis, W. P.; Confalone, P. N.; Chen, C.-y.; Tillyer, R. D.; Frey, L.; Tan, L.;
25
26 Xu, F.; Zhao, D.; Thompson, A. S.; Corley, E. G.; Grabowski, E. J. J.; Reamer,
27
28 R.; Reider, P. J. *J. Org. Chem.* **1998**, *63*, 8536-8543. (b) Chinkov, N.; Warm, A.;
29
30 Carreira, E. M. *Angew. Chem. Int. Ed.* **2011**, *50*, 2957-2961.
31
32
33
34 (7) (a) Liang, T.; Neumann, C. N.; Ritter, T. *Angew. Chem. Int. Ed.* **2013**, *52*, 8214-
35
36 8264. (b) Campbell, M. G.; Ritter, T. *Chem. Rev.* **2015**, *115*, 612-633. (c)
37
38 Charpentier, J.; Früh, N.; Togni, A. *Chem. Rev.* **2015**, *115*, 650-682. (d) Liu, X.;
39
40 Xu, C.; Wang, M.; Liu, Q. *Chem. Rev.* **2015**, *115*, 683-730. (e) Yang, X.; Wu, T.;
41
42 Phipps, R. J.; Toste, F. D. *Chem. Rev.* **2015**, *115*, 826-870. (f) In *Modern*
43
44 *Synthesis Processes and Reactivity of Fluorinated Compounds*; Leroux, F. R.,
45
46 Tressaud, A., Eds.; Elsevier, 2017.
47
48
49
50 (8) Nenajdenko, V. *Fluorine in Heterocyclic Chemistry: Volume 2*; Springer
51
52 International Publishing, 2014.
53
54
55 (9) Skraup, Z. *Ber. Dtsch. Chem. Ges.* **1880**, *13*, 2086-2087.
56
57
58
59
60

- 1
2
3 (10) Doebner, O.; v. Miller, W. *Ber. Dtsch. Chem. Ges.* **1881**, *14*, 2812-2817.
4
5 (11) Gould, R. G.; Jacobs, W. A. *J. Am. Chem. Soc.* **1939**, *61*, 2890-2895.
6
7 (12) Conrad, M.; Limpach, L. *Ber. Dtsch. Chem. Ges.* **1887**, *20*, 944-948.
8
9 (13) Combes, A. *Bull. Soc. Chim. Fr.* **1883**, *49*, 89.
10
11 (14) Knorr, L. *Justus Liebigs Ann. Chem.* **1886**, *236*, 69-115.
12
13 (15) Niementowski, S. *Ber. Dtsch. Chem. Ges.* **1894**, *27*, 1394-1403.
14
15 (16) (a) Friedlaender, P. *Ber. Dtsch. Chem. Ges.* **1882**, *15*, 2572-2575. (b) Marco-
16 Contelles, J.; Pérez-Mayoral, E.; Samadi, A.; Carreiras, M. d. C.; Soriano, E.
17 *Chem. Rev.* **2009**, *109*, 2652-2671.
18
19 (17) (a) Liu, X.-Y.; Xiao, Y.-P.; Siu, F.-M.; Ni, L.-C.; Chen, Y.; Wang, L.; Che, C.-M.
20 *Org. Biomol. Chem.* **2012**, *10*, 7208-7219. (b) Patil, N. T.; Raut, V. S.; Shinde, V.
21 S.; Gayatri, G.; Sastry, G. N. *Chem. – Eur. J.* **2012**, *18*, 5530-5535. (c) Patil, N.
22 T.; Raut, V. S.; Tella, R. B. *Chem. Commun.* **2013**, *49*, 570-572. (d) Shaikh, A.
23 C.; Ranade, D. S.; Thorat, S.; Maity, A.; Kulkarni, P. P.; Gonnade, R. G.; Munshi,
24 P.; Patil, N. T. *Chem. Commun.* **2015**, *51*, 16115-16118.
25
26 (18) Li, H.; Wang, C.; Huang, H.; Xu, X.; Li, Y. *Tetrahedron Lett.* **2011**, *52*, 1108-1111.
27
28 (19) Li, H.; Xu, X.; Yang, J.; Xie, X.; Huang, H.; Li, Y. *Tetrahedron Lett.* **2011**, *52*, 530-
29 533.
30
31 (20) Patil, N. T.; Raut, V. S. *J. Org. Chem.* **2010**, *75*, 6961-6964.
32
33 (21) Jiang, B.; Si, Y.-G. *J. Org. Chem.* **2002**, *67*, 9449-9451.
34
35 (22) Zhao, B.-C.; Zhang, Q.-Z.; Zhou, W.-Y.; Tao, H.-C.; Li, Z.-G. *RSC Adv.* **2013**, *3*,
36 13106-13109.
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

- 1
2
3 (23) Wozniak, L.; Staszewska-Krajewska, O.; Michalak, M. *Chem. Commun.* **2015**,
4
5 51, 1933-1936.
6
7
8 (24) Czerwiński, P.; Molga, E.; Cavallo, L.; Poater, A.; Michalak, M. *Chem. – Eur. J.*
9
10 **2016**, 22, 8089-8094.
11
12 (25) Jones, C.; Mills, D. P.; Rose, R. P.; Stasch, A.; Woodul, W. D. *J. Organomet.*
13
14 *Chem.* **2010**, 695, 2410-2417.
15
16
17 (26) (a) Díez-González, S.; Correa, A.; Cavallo, L.; Nolan, S. P. *Chem. – Eur. J.* **2006**,
18
19 12, 7558-7564. (b) Díez-González, S.; Nolan, S. P. *Angew. Chem. Int. Ed.* **2008**,
20
21 47, 8881-8884.
22
23
24 (27) van Koten, G.; James, S. L.; Jastrzebski, J. T. B. H. In *Comprehensive*
25
26 *Organometallic Chemistry II*; Stone, F. G. A., Wilkinson, G., Eds.; Pergamon:
27
28 Oxford, 1995.
29
30
31 (28) (a) Yu, D.; Zhang, Y. *Proc. Natl. Acad. Sci.* **2010**, 107, 20184-20189. (b) Nelson,
32
33 D. J.; Nolan, S. P. *Chem. Soc. Rev.* **2013**, 42, 6723-6753.
34
35
36 (29) The term "on water", introduced by Sharpless, refers to a reaction in which one of
37
38 the substrates is a liquid. In the case when all substrates are solid, it is possible
39
40 to recreate this effect if they melt under the reaction conditions, creating an
41
42 organic phase or when they have been melted together before the reaction. In
43
44 the case of the test reaction, all of the substrates were solid and due to the
45
46 sensitivity of the NHCCuX complex could not be melted. For this reason, the
47
48 mixture of substrates was dissolved in a minimum volume of DME and then
49
50 water was added in order to preserve the unique reactivity; Narayan, S.;
51
52
53
54
55
56
57
58
59
60

- 1
2
3 Muldoon, J.; Finn, M. G.; Fokin, V. V.; Kolb, H. C.; Sharpless, K. B. *Angew.*
4 *Chem. Int. Ed.* **2005**, *44*, 3275-3279.
5
6
7
8 (30) Wang, Y.; Ai, J.; Liu, G.; Geng, M.; Zhang, A. *Org. Biomol. Chem.* **2011**, *9*, 5930-
9 5933.
10
11
12 (31) Griffiths, G. J.; Warm, A. *Org. Process Res. Dev.* **2016**, *20*, 803-813.
13
14
15 (32) Aoki, S.; Watanabe, Y.; Sanagawa, M.; Setiawan, A.; Kotoku, N.; Kobayashi, M.
16 *J. Am. Chem. Soc.* **2006**, *128*, 3148-3149.
17
18
19 (33) (a) Bryce, A.; Ryan, C. J. *Clin. Pharmacol. Ther.* **2012**, *91*, 101-108. (b) Salvador,
20 J. A. R.; Carvalho, J. F. S.; Neves, M. A. C.; Silvestre, S. M.; Leitao, A. J.; Silva,
21 M. M. C.; Sa e Melo, M. L. *Nat. Prod. Rep.* **2013**, *30*, 324-374.
22
23
24
25
26
27 (34) Czakó, B.; Kürti, L.; Mammoto, A.; Ingber, D. E.; Corey, E. J. *J. Am. Chem. Soc.*
28 **2009**, *131*, 9014-9019.
29
30
31
32 (35) (a) Lee, H. M.; Nieto-Oberhuber, C.; Shair, M. D. *J. Am. Chem. Soc.* **2008**, *130*,
33 16864-16866. (b) Nilson, M. G.; Funk, R. L. *J. Am. Chem. Soc.* **2011**, *133*,
34 12451-12453. (c) Shenvi, R. A.; Guerrero, C. A.; Shi, J.; Li, C.-C.; Baran, P. S. *J.*
35 *Am. Chem. Soc.* **2008**, *130*, 7241-7243. (d) Shi, J.; Manolikakes, G.; Yeh, C.-H.;
36 Guerrero, C. A.; Shenvi, R. A.; Shigehisa, H.; Baran, P. S. *J. Am. Chem. Soc.*
37 **2011**, *133*, 8014-8027.
38
39
40
41
42
43
44
45
46 (36) Yoo, W.-J.; Nguyen, T. V. Q.; Kobayashi, S. *Angew. Chem. Int. Ed.* **2014**, *53*,
47 10213-10217.
48
49
50
51 (37) Meanwell, N. A. *J. Med. Chem.* **2011**, *54*, 2529-2591.
52
53
54 (38) Zafrani, Y.; Yeffet, D.; Sod-Moriah, G.; Berliner, A.; Amir, D.; Marciano, D.;
55 Gershonov, E.; Saphier, S. *J. Med. Chem.* **2017**, *60*, 797-804.
56
57
58
59
60

- 1
2
3 (39) (a) Feng, Z.; Min, Q.-Q.; Zhang, X. *Org. Lett.* **2016**, *18*, 44-47. (b) Ge, S.;
4
5
6
7
8
9
10
11 (41) (a) Fujikawa, K.; Fujioka, Y.; Kobayashi, A.; Amii, H. *Org. Lett.* **2011**, *13*, 5560-
12
13 5563. (b) Fier, P. S.; Hartwig, J. F. *J. Am. Chem. Soc.* **2012**, *134*, 5524-5527.
14
15
16 (42) (a) Prakash, G. K. S.; Ganesh, S. K.; Jones, J.-P.; Kulkarni, A.; Masood, K.;
17
18 Swabeck, J. K.; Olah, G. A. *Angew. Chem. Int. Ed.* **2012**, *51*, 12090-12094. (b)
19
20 Fujiwara, Y.; Dixon, J. A.; Rodriguez, R. A.; Baxter, R. D.; Dixon, D. D.; Collins,
21
22 M. R.; Blackmond, D. G.; Baran, P. S. *J. Am. Chem. Soc.* **2012**, *134*, 1494-1497.
23
24 (43) Rubic, T.; Lametschwandtner, G.; Jost, S.; Hinteregger, S.; Kund, J.; Carballido-
25
26
27 Perrig, N.; Schwarzler, C.; Junt, T.; Voshol, H.; Meingassner, J. G.; Mao, X.;
28
29
30 Werner, G.; Rot, A.; Carballido, J. M. *Nat. Immunol.* **2008**, *9*, 1261-1269.
31
32 (44) (a) Blakeney, J. S.; Reid, R. C.; Le, G. T.; Fairlie, D. P. *Chem. Rev.* **2007**, *107*,
33
34 2960-3041. (b) Congreve, M.; Langmead, C. J.; Mason, J. S.; Marshall, F. H. *J.*
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51 (46) (a) Bhuniya, D.; Umrani, D.; Dave, B.; Salunke, D.; Kukreja, G.; Gundu, J.;
52
53
54
55
56
57
58
59
60
60
Reddy, S. B.; Tambe, S.; Shejul, Y.; Chugh, A.; Palle, V. P.; Mookhtiar, K. A.;

- 1
2
3 Cully, D.; Vacca, J.; Chakravarty, P. K.; Nargund, R. P.; Wright, S. D.; Graziano,
4 M. P.; Singh, S. B.; Roy, S.; Cai, T.-Q. *Bioorg. Med. Chem. Lett.* **2011**, *21*, 3596-
5 3602. (b) Klenc, J.; Lipowska, M.; Taylor, A. T. *Bioorg. Med. Chem. Lett.* **2015**,
6 25, 2335-2339.
7
8
9
10
11
12 (47) Sapountzis, I.; Lin, W.; Kofink, C. C.; Despotopoulou, C.; Knochel, P. *Angew.*
13 *Chem. Int. Ed.* **2005**, *44*, 1654-1658.
14
15
16
17 (48) Hydrogenation of nitriles catalyzed by Raney nickel often leads to mixtures of
18 primary, secondary and even tertiary amines as a result of side reaction of the
19 primary amine with the imine intermediate; Nishimura, S. In *Handbook Of*
20 *Heterogeneous Catalytic Hydrogenation For Organic Synthesis*; John Wiley:
21 John Wiley & Sons, 2001.
22
23
24
25
26
27
28
29 (49) Santoro, O.; Collado, A.; Slawin, A. M. Z.; Nolan, S. P.; Cazin, C. S. J. *Chem.*
30 *Commun.* **2013**, *49*, 10483-10485.
31
32
33
34 (50) Broggi, J.; Díez-González, S.; Petersen, J. L.; Berteina-Raboin, S.; Nolan, S. P.;
35 Agrofoglio, L. A. *Synthesis* **2008**, 141-148.
36
37
38
39 (51) Díez-González, S.; Escudero-Adan, E. C.; Benet-Buchholz, J.; Stevens, E. D.;
40 Slawin, A. M. Z.; Nolan, S. P. *Dalton Trans.* **2010**, *39*, 7595-7606.
41
42
43
44 (52) Uehling, M. R.; Suess, A. M.; Lalic, G. *J. Am. Chem. Soc.* **2015**, *137*, 1424-1427.
45
46 (53) Nolte, C.; Mayer, P.; Straub, B. F. *Angew. Chem. Int. Ed.* **2007**, *46*, 2101-2103.
47
48 (54) Díez-González, S.; Scott, N. M.; Nolan, S. P. *Organometallics* **2006**, *25*, 2355-
49 2358.
50
51
52
53 (55) Duclos, S.; Stoeckli-Evans, H.; Ward, T. R. *Helv. Chim. Acta* **2001**, *84*, 3148-
54 3161.
55
56
57
58
59
60

- 1
2
3 (56) Boeck, F.; Kribber, T.; Xiao, L.; Hintermann, L. *J. Am. Chem. Soc.* **2011**, *133*,
4 8138-8141.
5
6
7
8 (57) Kovács, D.; Kádár, Z.; Mótyán, G.; Schneider, G.; Wölfling, J.; Zupkó, I.; Frank,
9
10 É. *Steroids* **2012**, *77*, 1075-1085.
11
12 (58) Lee, P. H.; Kim, H.; Lee, K.; Kim, M.; Noh, K.; Kim, H.; Seomoon, D. *Angew.*
13 *Chem. Int. Ed.* **2005**, *44*, 1840-1843.
14
15
16
17 (59) Petasis, N.; Myslinska, M.; US2009247766, 2009.
18
19 (60) Zhu, L.; Miao, Z.; Sheng, C.; Yao, J.; Zhuang, C.; Zhang, W. *J. Fluorine Chem.*
20 **2010**, *131*, 800-804.
21
22
23 (61) Patel, M.; Ko, S. S.; McHugh Jr, R. J.; Markwalder, J. A.; Srivastava, A. S.;
24
25 Cordova, B. C.; Klabe, R. M.; Erickson-Viitanen, S.; Trainor, G. L.; Seitz, S. P.
26
27 *Bioorg. Med. Chem. Lett.* **1999**, *9*, 2805-2810.
28
29 (62) Sohda, T.; Mizuno, K.; Imamiya, E.; Sugiyama, Y.; Fujita, T.; Kawamatsu, Y.
30
31 *Chem. Pharm. Bull.* **1982**, *30*, 3580-3600.
32
33
34 (63) Ishikawa, S.; Mizutani, T.; Nagase, T.; Sato, N.; Takahashi, H.; EP2210880,
35
36 2010.
37
38
39 (64) Cheng, J.; Zhai, H.; Bai, J.; Tang, J.; Lv, L.; Sun, B. *Tetrahedron Lett.* **2014**, *55*,
40
41 4044-4046.
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60