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Corrigendum

Corrigendum to "Microwave-promoted one-pot three-component synthesis of 2,3-dihydroquinazolin-4(1H)-ones catalyzed by heteropolyanion-based ionic liquids under solvent-free conditions" [Tetrahedron Lett. 76/27 (2020) 131312]

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The authors regret the following errors:

1. In page 1, "Additionally, 2,3-dihydroquinazolin-4(1*H*). -one derivatives can easily…" should be changed to "Additionally, 2,3-dihydroquinazolin-4(1*H*)-one derivatives can easily…"
2. In page 1, "… numerous protocols have been developed recently for the synthesis using catalysts including acidic reagents [21], [−] [29] lewis acids [30−36],…" should be changed to "… numerous protocols have been developed recently for the synthesis using catalysts including acidic reagents [21−29], lewis acids [30−36],…"
3. In page 1, "… alternative methods to perform organic. synthesis under environmentally benign conditions is in…" should be changed to "… alternative methods to perform organic synthesis under environmentally benign conditions is in…"
4. In page 2, "… condensation, which were. readily available based on our previous publicationsn…" should be changed to "… condensation, which were readily available based on our previous publications…"
5. In page 3, "… while bringing down the catalyst loading to 1 mol % led. to a reduction in the yield to…" should be changed to "… while bringing down the catalyst loading to 1 mol % led to a reduction in the yield to…"
6. In page 3, "… transformation to different substituted isatoic. anhydrides was performed." should be changed to "… transformation to different substituted isatoic anhydrides was performed."
7. In page 3, "… easily separated from the. reaction mixture via simple centrifugation…" should be changed to "… easily separated from the reaction mixture via simple centrifugation…"
8. The chemical structure 4a-4g should be changed in Table 2 as below:

Table 2. HPAIL catalyzed three-component condensation of isatoic anhydrides, amines and aldehydes (or ketones).[a]

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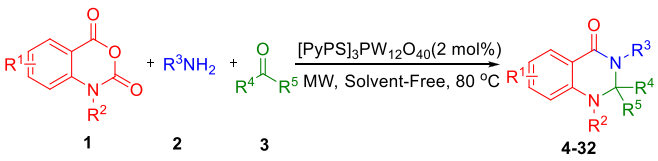
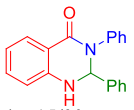
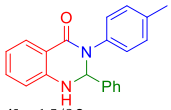
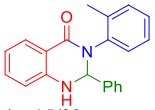
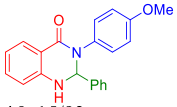
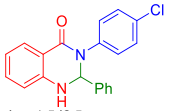
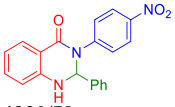
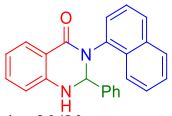
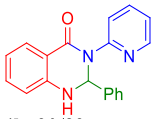
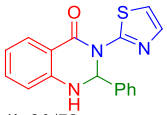
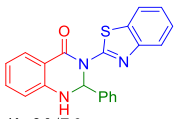
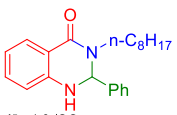
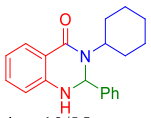
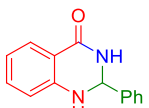
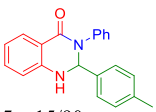
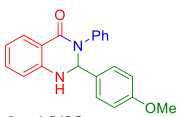
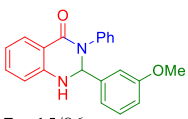
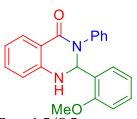
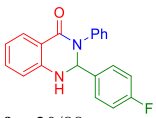
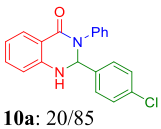
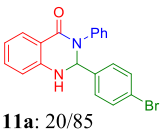
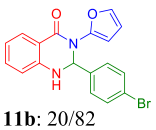
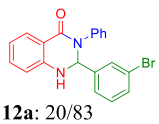
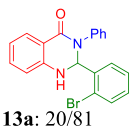


1 2 3 4-32

Product: time (min)/yield (%)[b]		
 **4a**: 15/90	 **4b**: 15/92	 **4c**: 15/89
 **4d**: 15/93	 **4e**: 15/85	 **4f** 20/78
 **4g**: 20/80	 **4h**: 20/82	 **4i**: 20/78
 **4j**: 30/76	 **4k**:10/87	 **4l**: 10/88
 **4m**: 10/85	 **4n**[c]: 10/92	 **5a**: 15/90
 **6a**: 15/88	 **7a**: 15/86	 **8a**: 15/85
 **9a**: 20/88	 **10a**: 20/85	 **11a**: 20/85
 **11b**: 20/82	 **12a**: 20/83	 **13a**: 20/81

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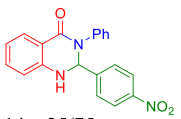
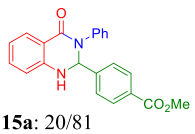
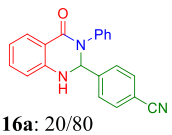
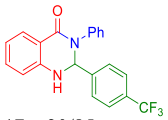
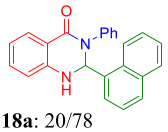

14a: 25/75


15a: 20/81


16a: 20/80


17a: 20/85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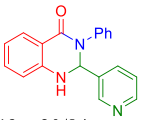
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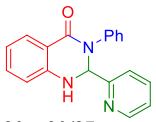
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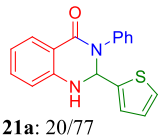
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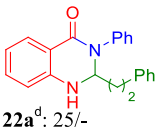
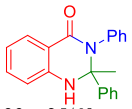

22a[d]: 25/-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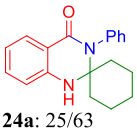
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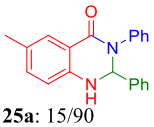
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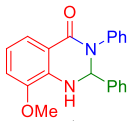
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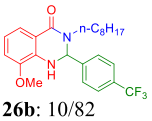
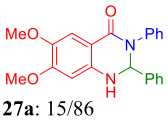

26b: 10/82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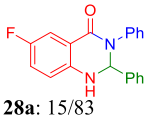
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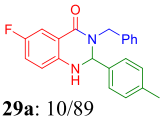
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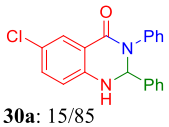
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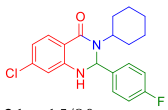
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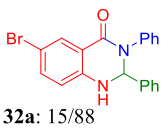
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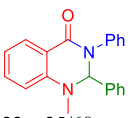
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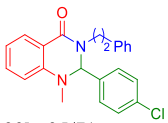
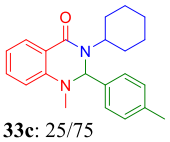

33b: 25/71


33c: 25/75

[a] Unless otherwise specified, all reactions were carried out with isatoic anhydrides **1** (2.0 mmol), amines **2** (2.4 mmol), aldehydes **3** (or ketones) (2.4 mmol) and $[PyPS]_3PW_{12}O_{40}$ (2 mol %) under MW (700 W) and solvent-free conditions at 80 °C in a sealed glass pressure tube.
[b] Isolated yields based on isatoic anhydrides.
[c] Ammonium acetate was used as ammonia source.
[d] 2-Amino-N-phenylbenzamide was the only product.

(***continued***).

The authors would like to apologise for any inconvenience caused.

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